

Supporting Information

Preparation of Ultrathin Perovskite Nanosheets by Exfoliation of $\text{H}_2\text{CaTa}_2\text{O}_7$ for High-Performance Lead Removal from Waters

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Article

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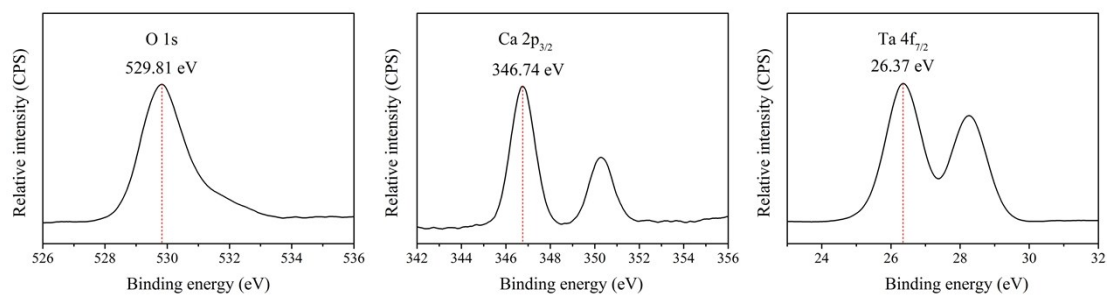


Figure S1. XPS spectra of sample U-HCT-Pb after washing with HCl.

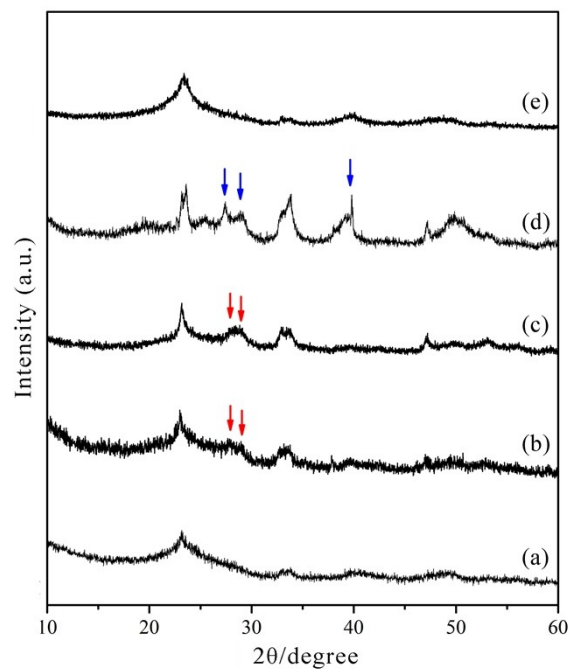


Figure S2. The XRD patterns of U-HCT sample in different state: (a) U-HCT; (b) U-HCT after uptake Pb(II) at pH 1.3; (c) U-HCT after uptake Pb(II) at pH 5.0; (d) U-HCT after uptake Pb(II) at pH 6.8; (e) U-HCT after uptake Pb(II), 1 M HCl solution washed repeatedly.

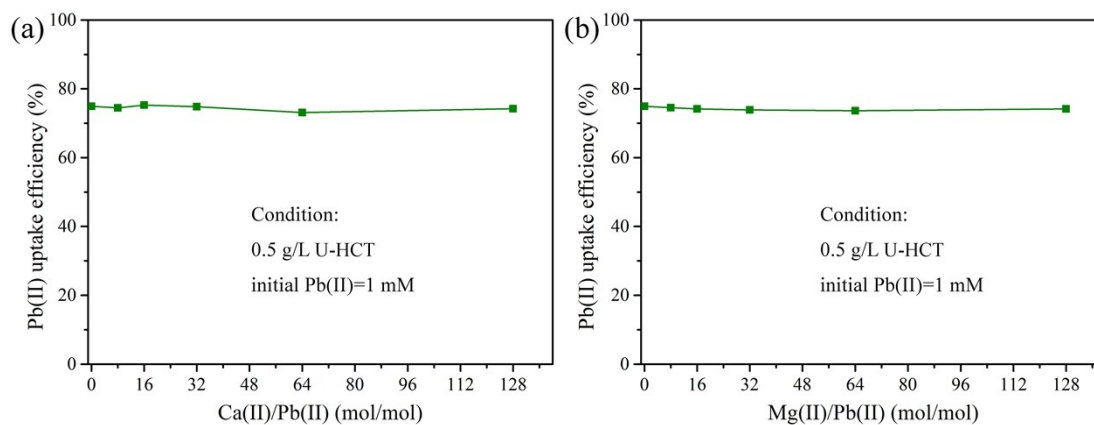


Figure S3. Effect of competitive ions on uptake of lead ions onto U-HCT (a) calcium ions, (b) magnesium ions.

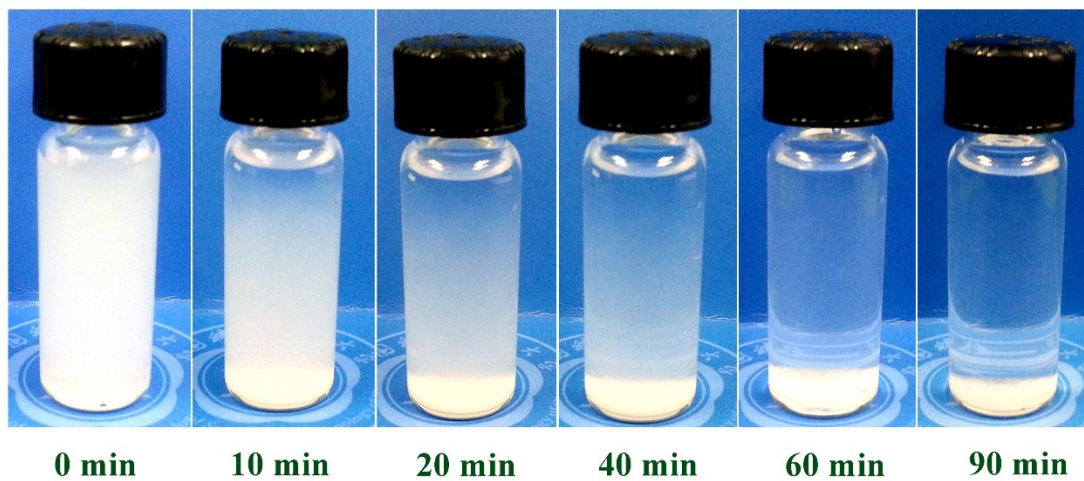


Figure S4. Static settling properties in various time intervals of U-HCT with adsorbed lead ions (initial Pb(II) concentration is 300 m/L).