Electronic Supplementary Material (ESI) for RSC Advances. This journal is © The Royal Society of Chemistry 2016

Supporting Information

Lewis acidic ionic liquids of crown ether complex cations:

preparation and applications in organic reactions

Yatao Liang, Jinyuan Wang, Chen Cheng, Huangwang Jing*

Table of contents

1. Characterization data of Lewis acidic ionic liquids and target compounds	s2-s3
2. FT-IR, MS, TGA-DSC of Lewis acidic ionic liquids	s4-s18
3. ¹ H NMR spectra of target compounds	s19-s20

1. Characterization data of Lewis acidic ionic liquids and part compounds

1.1 The data of Lewis acidic ionic liquids

[18-C-6K][**AlCl**₄] (**A**) FT-IR (cm⁻¹) 3427.8 (m), 2916.5 (m), 1476.0 (w), 1353.5 (w), 1115.4 (s), 945.6 (m); Raman (cm⁻¹): 359; Elemental analysis calcd for $C_{12}H_{24}Cl_4AlKO_6 \cdot CH_3CH_2OH \cdot H_2O$ (%): C 31.36, H 6.01 found: C 31.70, H 6.342; MS (ESI) m/z: calcd for [M]⁻ 168.79, found 168.86.

[18-C-6K][**FeCl**₄] (**B**) FT-IR (cm⁻¹) 3431.0 (m), 2896.4 (m), 1634.5 (w), 1351.2 (w), 1105.1 (s), 961.8 (m); Raman (cm⁻¹): 332; Elemental analysis calcd for $C_{12}H_{24}Cl_4FeKO_6 H_2O CH_2Cl_2$ (%): C 25.85, H 4.67 found: C 25.81, H 4.613; MS (ESI) m/z: calcd for [M]⁻¹97.66, found 197.85.

[18-C-6K][ZnCl₃] (C) FT-IR (cm⁻¹) 3412.9 (m), 2899.7 (m), 1648.8 (w), 1353.6 (w), 1110.3 (s), 962.8 (m); Raman (cm⁻¹): 289; Elemental analysis calcd for C₁₂H₂₄Cl₃ZnKO₆
(%)·CH₃CH₂OH·H₂O: C 31.18, H 5.98 found: C 30.82, H 5.751; MS (ESI) m/z: calcd for [M]⁻ 171.75, found 170.87.

[15-C-5Na][FeCl₄] (D) FT-IR (cm⁻¹) 3429.0 (m), 2922.7 (m), 1475.0 (w), 1353.2 (w), 1117.2 (s), 947.9 (w); Raman (cm⁻¹): 329; Elemental analysis calcd for C₁₀H₂₀Cl₄FeNaO₅ (%)·2H₂O: C 25.18, H 5.07 found C 25.31, H 5.22; MS (ESI) m/z: calcd [M][−] 197.66, found 197.83.

[15-C-5Na][ZnCl₃] (E) FT-IR (cm⁻¹) 3441.1 (m), 2922.1 (m), 1477.8 (w), 1352.4 (w), 1113.6 (s), 946.3 (m); Raman (cm⁻¹): 287; Elemental analysis calcd for C₁₀H₂₀Cl₃ZnNaO₅·1/3CH₃CH₂OH (%): C 29.77, H 5.15 found: C 30.03, H 4,75; MS (ESI) m/z: calcd for [M][−] 170.75, found 170.
87.

[18-C-6K]⁺ MS (ESI) m/z: calcd for [M]⁺303.12, found 303.11.

[15-C-5Na]⁺ MS (ESI) m/z: calcd for [M]⁺ 243.12, found 243.09.

1.2 ¹H NMR data of target compounds

2-phenyl-1H-benzo[d]imidazole (**3a**) ¹H NMR (300 MHz, DMSO-*d₆*) δ = 12.94(s, 1H), 8.19 (d, *J*=7.1Hz 1H), 7.80-7.33(m, 3H), 7.22 (dd, *J*=6.0, 2.8Hz, 1H) ppm;

1-benzyl-2-phenyl-1H-benzo[d]imidazole (**4a**) ¹H NMR (300 MHz, DMSO-*d*₆) δ = 10.83 (s, 1H), 7.37 (dd, *J* =10.0, 4.8 Hz, 1H), 7.27 (t, *J* = 7.5Hz, 1H), 7.18 (d, *J* =7.3, 1H), 7.07-6.99 (m, 1H), 5.84(s, 1H) ppm;

3,3'-(phenylmethylene)bis(1H-indole) (6a) ¹H NMR (300 MHz, CDCl₃) δ = 7.89 (d, J =
8.4Hz, 1H), 7.70 (dd, J = 6.9, 1.8Hz, 2H), 7.48 (m, 3H), 7.33 (m, 4H), 7.24 (dd, J = 6.6,
1.2Hz, 2H), 7.12 (dd, J=8.4, 2.4Hz, 2H), 5.47 (s, 2H);



2.1 FT-IR of Lewis acidic ionic liquids

Figure S1 IR spectrum of A



Figure S2 IR spectrum of B



Figure S3 IR spectrum of C



Figure S4 IR spectrum of D



Figure S5 IR spectrum of E

2.2 MS of Lewis acidic ionic liquids



Figure S6 MS of IL catalyst cation [18-C-6K]⁺



Figure S7 MS of IL catalyst cation [15-C-5Na]⁺



Figure S8 MS of A anion [AlCl₄]⁻



Figure S9 MS of B anion [FeCl₄]⁻





Figure S10 MS of C anion [ZnCl₃]⁻



Figure S11 MS of D anion [FeCl₄]⁻



Bruker Compass DataAnalysis 4.1 printed: 6/26/2014 4:58:00 PM by: BDAL@DE Page 1 of 1

Figure S12 MS of E anion [ZnCl₃]⁻



2.3 TGA-DSC of Lewis acidic ionic liquids

Figure S13 TGA-DSC of A









Figure S15 TGA-DSC of C

Figure S16 TGA-DSC of D



Figure S17 TGA-DSC of E

3. ¹H NMR of part compounds



¹H NMR 2-phenyl-1H-benzo[d]imidazole (3a)





¹H NMR 3,3'-(phenylmethylene)bis(1H-indole) (6a)