## **Supporting Information**

## Breathing catalyst-supports: CO<sub>2</sub> adjustable and magnetic recyclable "smart" hybrid nanoparticles

Anchao Feng,<sup>a</sup> Yun Wang,<sup>a</sup> Liao Peng,<sup>a</sup> Xiaosong Wang<sup>b</sup> and Jinying Yuan<sup>\*a,c</sup>

<sup>a</sup> Key Lab of Organic Optoelectronics & Molecular Engineering of Ministry of Education

Department of Chemistry, Tsinghua University, Beijing 100084, P.R. China

E-mail: yuanjy@mail.tsinghua.edu.cn

<sup>b</sup> Department of Chemistry, Waterloo Institute of Nanotechnology, University of Waterloo, 200 University Avenue West, Waterloo, ON N2L 3G1, Canada

<sup>c</sup> State Key Laboratory of Polymer Materials Engineering (Sichuan University)



**Fig. S1** a) FT-IR spectra of  $Fe_3O_4@OA$ ,  $Fe_3O_4@SiO_2$ ,  $Fe_3O_4@SiO_2$ -MPS and  $Fe_3O_4@SiO_2@PDEAEMA$ . b) TGA curves of  $Fe_3O_4@SiO_2$ -MPS and  $Fe_3O_4@SiO_2@PDEAEMA$ .



Fig. S2 SEM image of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>-PDEAEMA hybrid microspheres.



**Fig. S3** Zeta potential of a)  $Fe_3O_4@SiO_2@PDEAEMA$  nanoparticles after treatment of  $CO_2$ ; b)  $Fe_3O_4@SiO_2@PDEAEMA-Au$  nanocomposites after treatment of  $CO_2$ ; c)  $Fe_3O_4@SiO_2@PDEAEMA-Au$  nanocomposites after treatment of  $N_2$ .



Fig. S4 The average diameter of  $Fe_3O_4$  ( $@SiO_2$ -PDEAEMA-Au nanocomposites by DLS after CO<sub>2</sub> and N<sub>2</sub> purging for 15 min.

## Determination of mass ratio of polymer through TGA data

Based on the TGA results, we can assume that there was 80.98 g residuum in 100 g of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>. Same component of residuum will be kept for the sample of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@PDEAEMA, as the polymer component should be removed totally at 800 °C. The mass of PDEAEMA can be calculated as 80.98 g/(1-40.91 %)-100 g, resulted in 37.05 g. Therefore, the mass ratio of polymer can be decided as 37.05 g/ (100 g+37.05g) = 27.04 wt %.