Electronic Supplementary Information

Room temperature synthesis of CuInS₂ nanocrystals

Christine Buchmaier,^a Thomas Rath,^a,* Franz Pirolt,^a Astrid-Caroline Knall,^a Petra Kaschnitz, ^a Otto Glatter,^a Karin Wewerka,^b Ferdinand Hofer,^b Birgit Kunert,^c Kurt Krenn^d and Gregor Trimmel^a

a) Institute for Chemistry and Technology of Materials (ICTM), NAWI Graz, Graz University of Technology, Stremayrgasse 9, 8010 Graz, Austria

b) Institute for Electron Microscopy and Nanoanalysis, Graz University of Technology & Centre for Electron Microscopy Graz, Steyrergasse 17, 8010 Graz, Austria

c) Institute of Solid State Physics, Graz University of Technology, Petersgasse 16, 8010 Graz, Austria

d) Institute of Earth Sciences, University of Graz, Universitätsplatz 2, 8010 Graz, Austria



Fig. S1 TEM image of the synthesized CuInS₂ nanocrystals and corresponding particle sizes.



Fig. S2 SAED pattern and the corresponding radial intensity profile of the CuInS₂ nanocrystals.



Fig. S3 SAXS data of a time resolved measurement during a synthesis performed directly in the measurement capillary.



Fig. S4 Chemical structures auf educts (indium xanthate A, hexylamine B) and reaction products (thiocarbamate C, dithiocarbamate D, alcohol E) and labelling of the carbon/hydrogen atoms used in Tables S1 and S2.

 Table S1 ¹H NMR spectroscopic data – indium xanthate/hexylamine model reaction

In xanthate	hexylamine	thiocarbamate	dithiocarbamate	alcohol
(A) [ppm]	(B) [ppm]	isomers (C) [ppm]	(D) [ppm]	(E) [ppm]
A2: 4.75 (t ; 1H)	B1: 2.68 (t ; 2H)	C2: 5.36 (dd)	D2: 3.45 (t)	E1: 3.10 (dd; 1H)
A3: 1.76 (m ; 2H)	B2: 1.44 (m ; 2H)	C3: 1.72; 1.53	D3: 1.64	E2: 1.22 (m ; 1H)
A4: 1.02 (t; 3H)	B3: 1.29 (br; 2H)	C4: 0.98	D4: 1.30	1.56 (m; 1H)
A6: 0.99 (s ; 9H)	B4: 1.29 (br; 2H)	C6: 0.98; 0.92	D5: 1.30	E3: 0.99 (t; 3H)
	B5: 1.29 (br ; 2H)	C7: 3.30; 3.58	D6: 1.30	E5: 0.87(s;9H)
	B6: 0.89 (t ; 3H)	C8: 1.60; 1.50	D7: 0.90	E6: 1.51 (s ; 1H)
	B7: 1.07 (br ; 2H)	C9: 1.30	D8: 7.69 (br)	
		C10: 1.30		
		C11: 1.30		
		C12: 0.89		
		C13: 6.80; 6.30		

Table S2 ¹³C NMR spectroscopic data – indium xanthate/hexylamine model reaction

In xanthate	hexylamine	thiocarbamate	dithiocarbamate	alcohol
(A) [ppm]	(B) [ppm]	isomers (C) [ppm]	(D) [ppm]	(E) [ppm]
A1: 230.9	B1: 42.4	C1: 191.7; 190.9	D1: 205.2	E1: 81.8
A2: 103.2	B2: 34.0	C2: 90.7; 88.1	D2: 51.1	E2: 24.3
A3: 23.5	B3: 26.7	C3: 23.1; 23.3	D3: 27.8	E3: 11.6
A4: 11.1	B4: 31.8	C4: 11.0; 11.2		E4: 35.0
A5: 36.2	B5: 22.7	C5: 35.9; 35.4	other peaks are	E5: 25.8
A6: 26.1	B6: 14.1	C6: 26.2; 26.5	overlapping with	
		C7: 45.3; 43.1	C, B, and E	
		C8: 29.3; 28.7		
		C9: 26.5		
		C10: 31.5		
		C11: 22.6		
		C12: 14.0		



Fig. S5 ¹³C NMR spectra of the pure educts indium xanthate and hexylamine, and spectra of the reaction shortly after mixing and at the end of the reaction (corresponding ¹H NMR spectra can be found in the main text of the article).



Fig. S6 Heteronuclear Multiple Bond Correlation (HMBC) NMR spectrum (¹³C, ¹H) used for the identification of the reaction products by visualization of long-range couplings.



Fig. S7 Heteronuclear Single Quantum Coherence (HSQC) NMR spectrum (${}^{13}C$, ${}^{1}H$) used for the identification of the reaction products by visualization of direct ${}^{1}H{}^{-13}C$ coupling.



Fig. S8 Chemical structures of educts (copper xanthate F, oleylamine G) and reaction products (thiocarbamate H, dithiocarbamate I, alcohol E) and labelling of the carbon/hydrogen atoms used in Tables S3 and S4.

Cu xanthate (F) [ppm]	Cu/In xanthate* (A/F) [ppm]	oleylamine (G) [ppm]	thiocarbamate isomers (H) [ppm]	dithiocarbamate (I) [ppm]
F2: 5.24(t; 1H)	AF2: 5.04(t; 1H)	G1: 2.64 (t ; 2H)	H25: 6.78; 6.28	H1: 3.46
F3: 1.73 (m ; 2H)	AF3: 1.72 (m; 2H)	G2: 1.40 (m, 2H)	H1: 3.27; 3.55	
F4: 0.93 (t ; 3H)	AF4,6: 1.00 – 0.95 (bs : 12 H)	G3,4,5,6,7,12,13,14,15, 16,17: 1.35-1,15 (m, 22H)	H20: ~5.36 (**)	
F6: 0.95(s; 9H)		G8/11: 1.99 (m ; 4H)	all other peaks are	all other peaks are
		G9/10: 5.32 (m ;2H) G18: 0.85 (t ; 3H) G19: 1.03 (bs ; 2H)	overlapping with peaks stemming from oleylamine G, dithiocarbamate I, and alcohol E	overlapping with peaks stemming from oleylamine G, thiocarbamate H and alcohol E

Table S3 ¹H NMR spectroscopic data - Cu/In xanthates - oleylamine

* peaks observed during reaction (reaction time 10 min)

** peak overlaid by oleylamine G9/10 protons

Table S4 ¹³ C NMR s	spectrosco	pic data - Cu	ı/In xanthates – ole	eylamine
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Cu xanthate (F) [ppm]	Cu/In xanthate* (A/F) [ppm]	oleylamine** (G) [ppm]	thiocarbamate isomers (H) [ppm]	dithiocarbamate (I) [ppm]
F1: 230.4 F2: 98.9	AF1: 230 (b) AF2: 98 (b)	G1: 41.9 G2: 33.2 G9/10: 130.0,	C19: 191.8 ; 191.0 C20: 90.5 ; 88.9 C1: 45.3 ; 43.1	C19: 206 (br) C1: 51.5
F3: 23.5	AF3: 23	129.8		
F4: 11.3	AF4: 11	G17: 22.6		
F5: 36.6	AF5: 36.5	G18: 14.0 G3-8, G11-16: 31.9, 29.8, 29.75, 29.72, 29.67, 29.58, 29.46, 29.33, 29.31, 29.25, 27.27,	all other peaks are overlapping with G, I and F	all other peaks are overlapping with G, H and F
F6: 26.2	AF6: 26	26.11		

* peaks observed during the reaction (reaction time 10 min)

** main peaks, additional small peaks by small amounts of isomers and byproducts



Fig. S9 ¹H NMR spectra of oleylamine (G), a mixture of indium and copper xanthate (A + F) and of the reaction mixture comprising of the metal xanthates and oleylamine after different reaction times.



Fig. S10 ¹³C NMR spectra of the reaction mixture (metal xanthates and oleylamine) after different reaction times.