

Electronic Supplementary Information

**A novel pyrazole based single molecular probe for multi-analyte (Zn^{2+} and Mg^{2+}) detection
in Human gastric adenocarcinoma cells**

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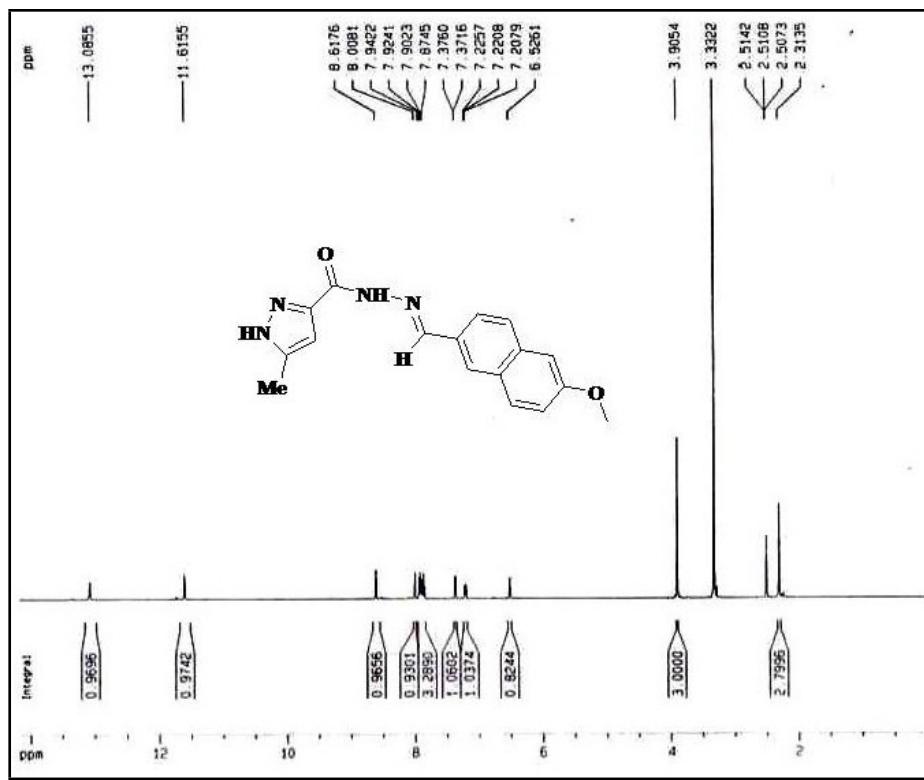


Fig. S1 ¹H NMR (DMSO-*d*₆, 500 MHz) spectrum of the probe **PYN**.

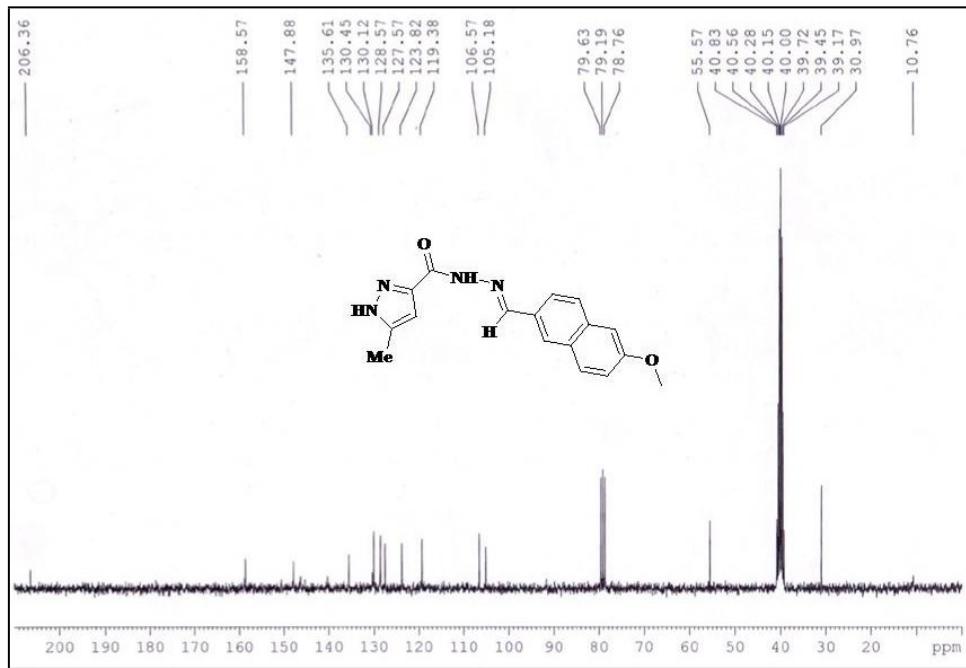


Fig. S2 ¹³C NMR (DMSO-*d*₆, 300 MHz) spectrum of the probe **PYN**.

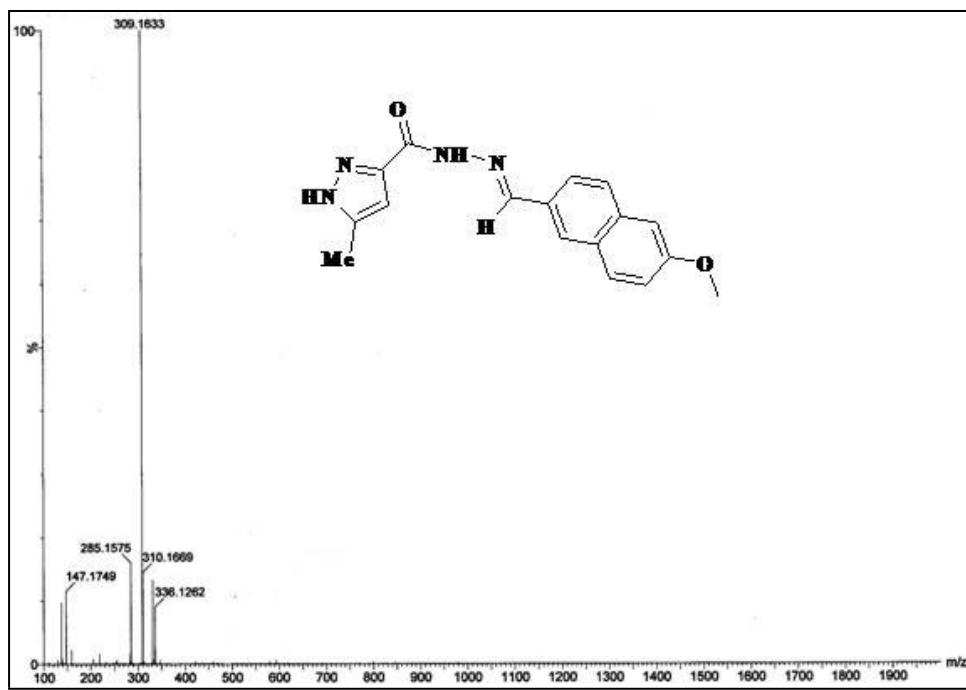


Fig. S3 MS spectra of the probe PYN.

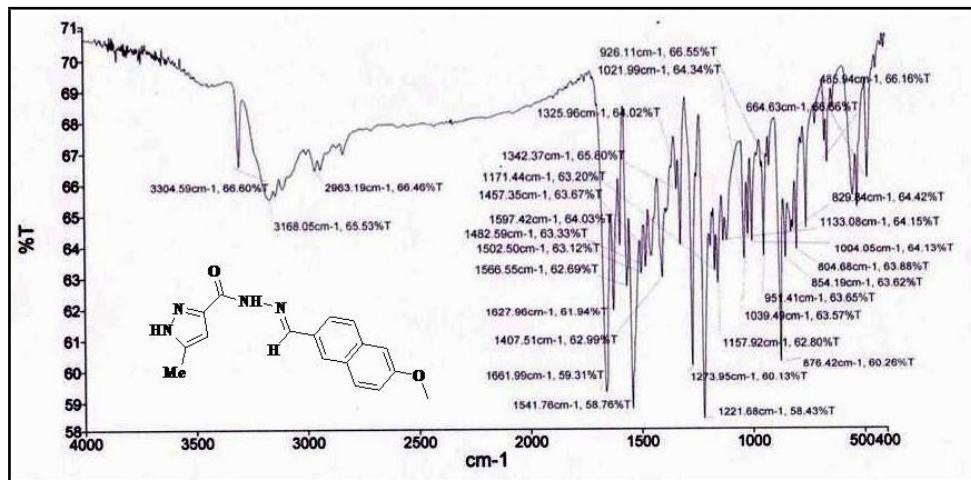


Fig. S4 FT- IR spectra of the probe PYN.

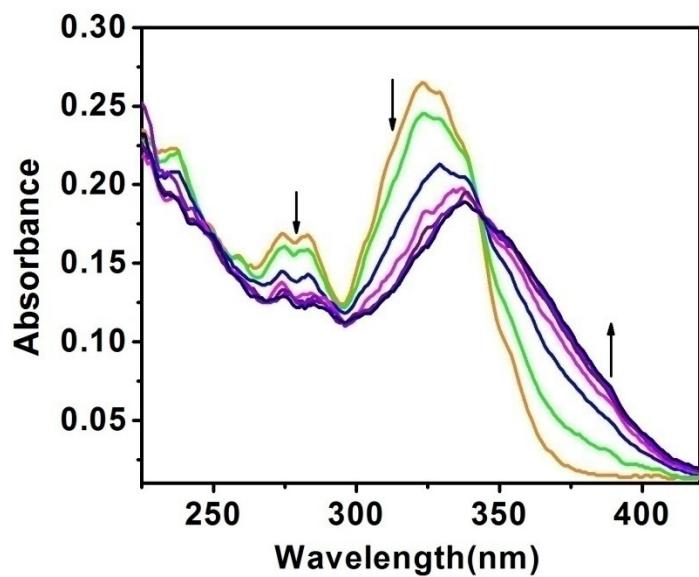


Fig. S5 Uv-vis spectra of the probe **PYN** (5 μM) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 $^{\circ}\text{C}$), in the presence of different amount Zn^{2+} .

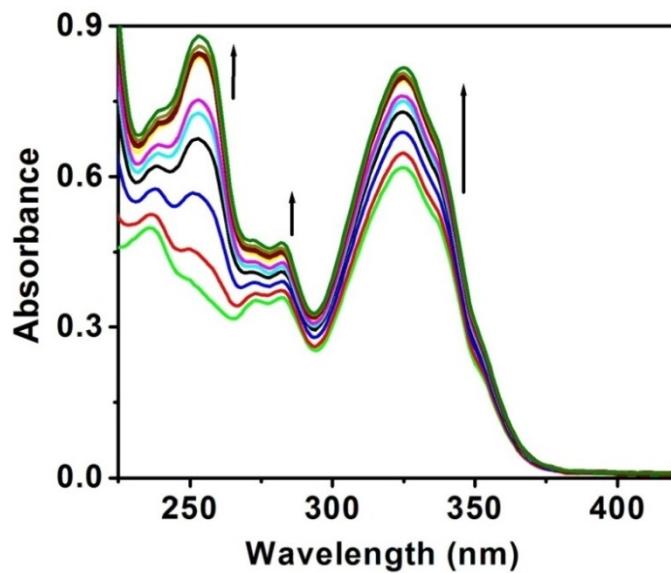


Fig. S6 UV-vis spectra of the probe **PYN** (5 μM) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 $^{\circ}\text{C}$), in the presence of different amount Mg^{2+} .

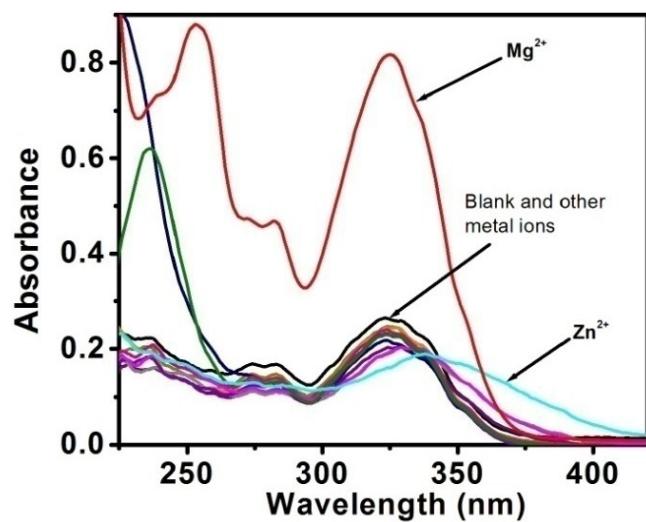


Fig. S7 Changes in the UV-Vis spectra of **PYN** (5 μ M) after addition of 1.2 equivalents of various cations in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C).

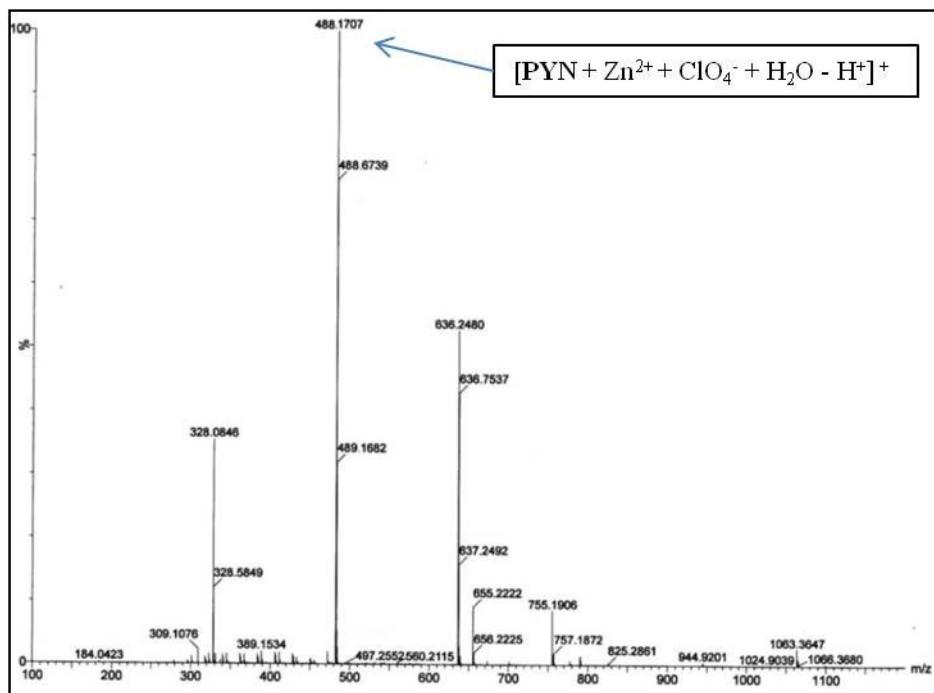


Fig. S8 TOF MS ES⁺ spectra of **PYN-Zn²⁺** complex.

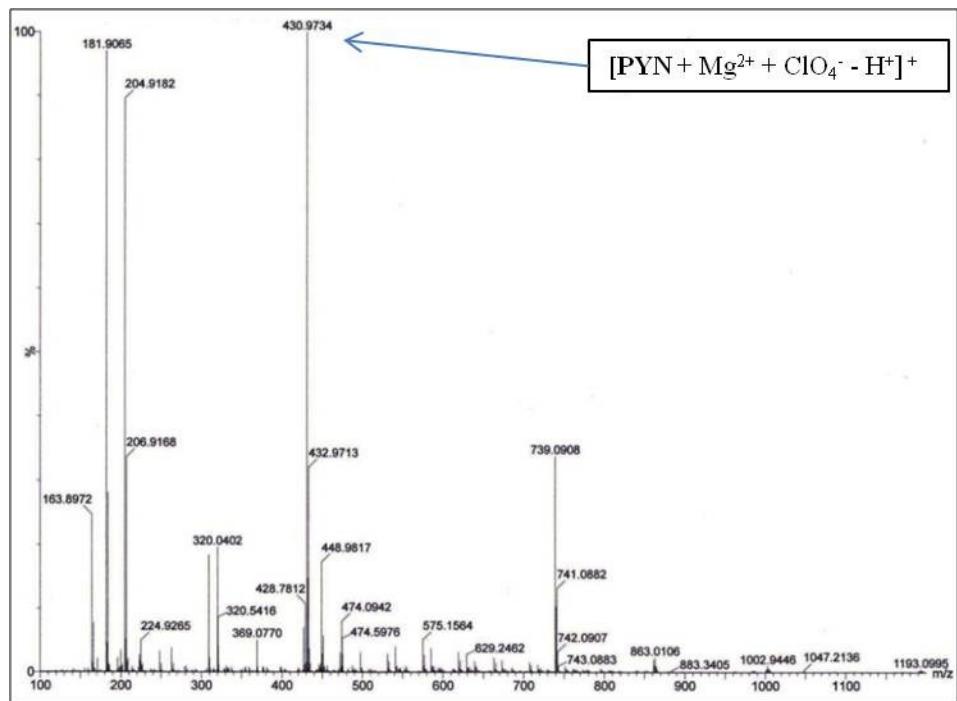


Fig. S9 TOF MS ES⁺ spectra of PYN-Mg²⁺ complex.

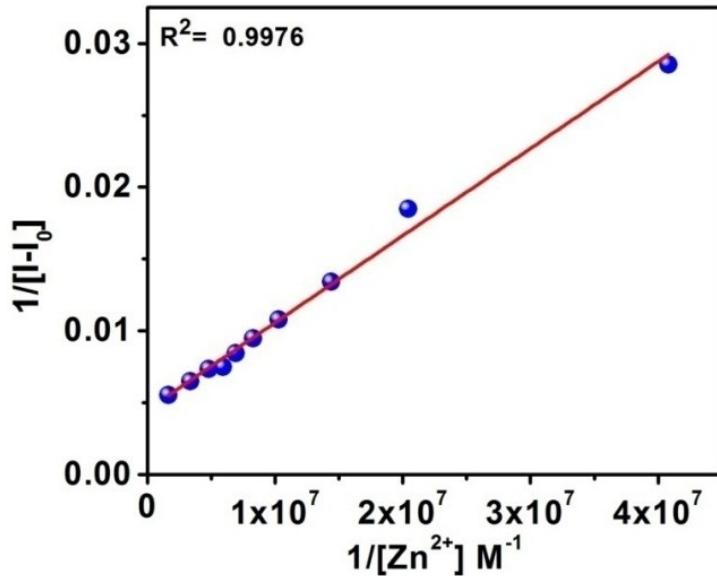


Fig. S10 Benesi-Hildebrand plot of the probe PYN (5 μ M) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C), emission intensity at 456 nm assuming 1:1 stoichiometry between PYN and Zn²⁺.

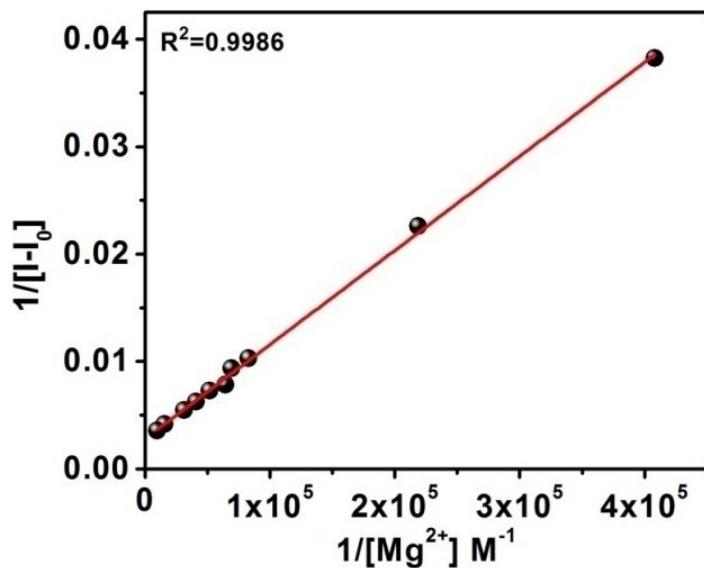


Fig. S11 Benesi-Hildebrand plot of the probe **PYN** (5 μ M) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C), emission intensity at 472 nm assuming 1:1 stoichiometry between **PYN** and Mg^{2+} .

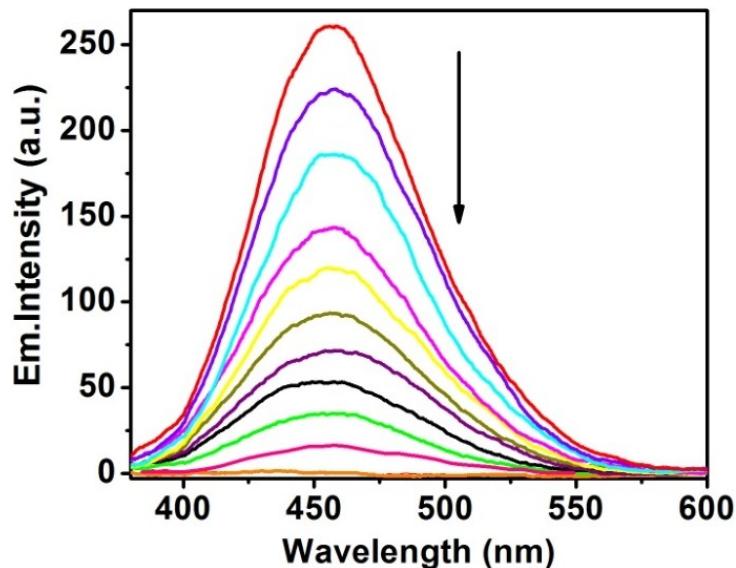


Fig. S12 Fluorescence titration of **PYN-Zn²⁺** complex with EDTA in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C), ($\lambda_{ex} = 370$ nm). Intensity gradually decreases upon gradual addition of EDTA solution.

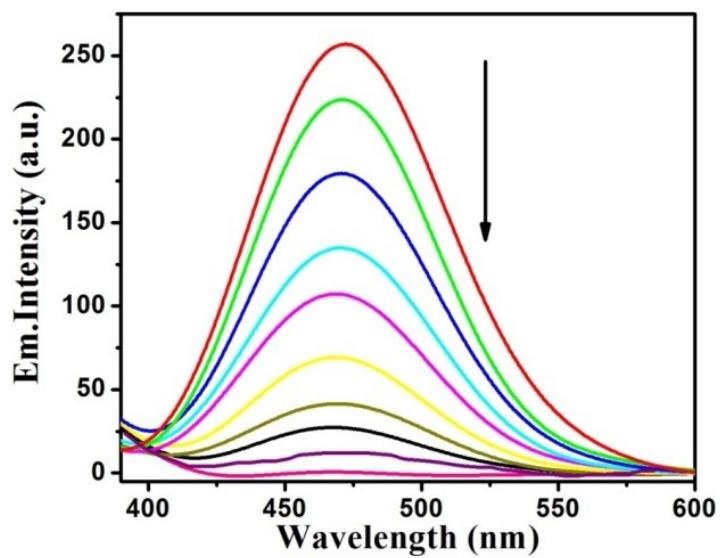


Fig. S13 Fluorescence titration of **PYN**-Mg²⁺ complex with EDTA in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C), ($\lambda_{\text{ex}} = 370 \text{ nm}$). Intensity gradually decreases upon gradual addition of EDTA solution.

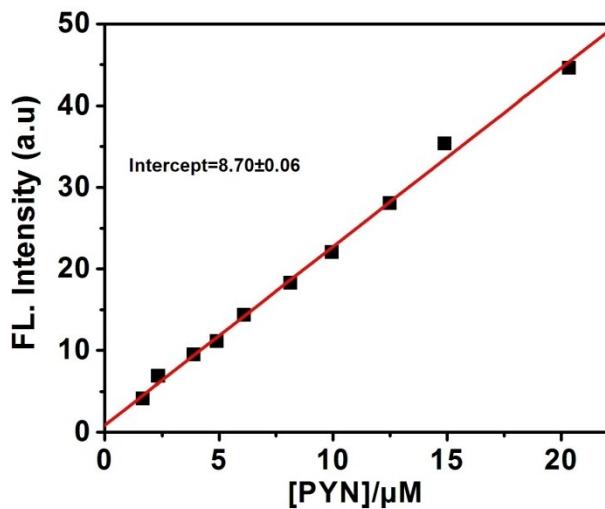


Fig. S14 Determination of Sb of the blank, **PYN** solution.

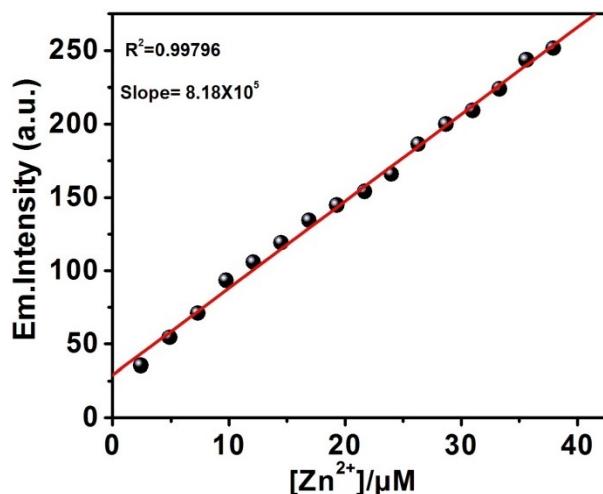


Fig S15 Determination of the detection limit of Zn^{2+} by **PYN** (5 μM) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C), ($\lambda_{ex} = 370$ nm, $\lambda_{em} = 456$ nm). LOD (Zn^{2+}) = $(3 \times 0.06)/8.18 \times 10^5 = 2.2 \times 10^{-7}$

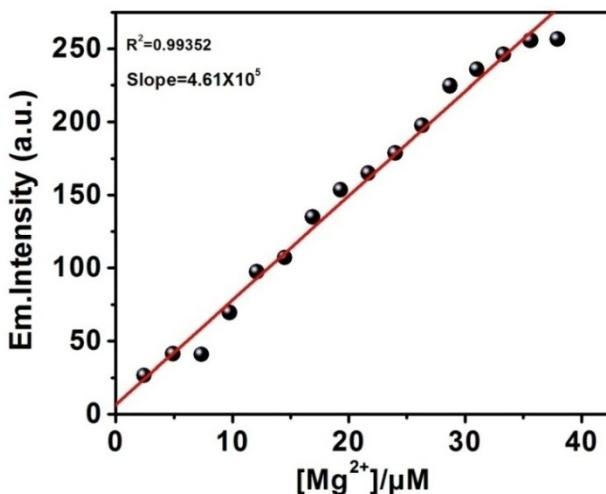


Fig. S16 Determination of the detection limit of Mg^{2+} by **PYN** (5 μM) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C), ($\lambda_{ex} = 370$ nm, $\lambda_{em} = 472$ nm). LOD (Mg^{2+}) = $(3 \times 0.06)/4.61 \times 10^5 = 3.9 \times 10^{-7}$

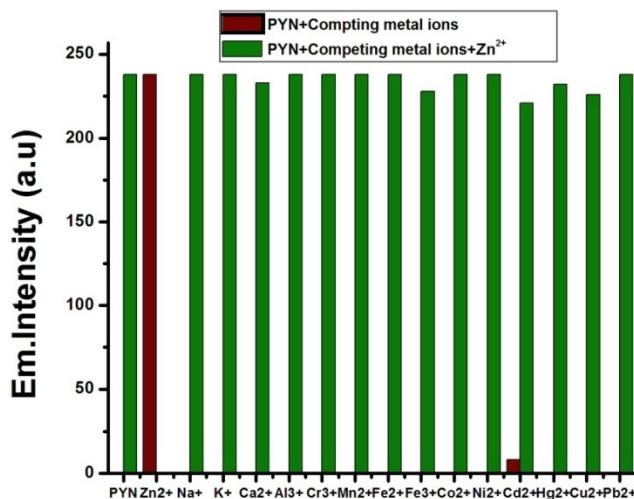


Fig. S17 Fluorescence intensity of PYN (5 μ M) upon the addition of different metal ions (wine bar) and the addition of Zn^{2+} in the presence of background cations (olive bar) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C). The intensities were recorded at 456 nm.

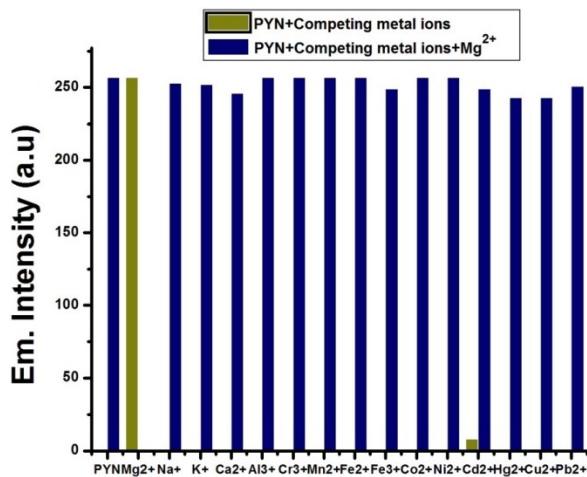


Fig. S18 Fluorescence intensity of PYN (5 μ M) upon the addition of different metal ions (dark yellow bar) and the addition of Mg^{2+} in the presence of background cations (navy bar) in 7:3 (v/v) MeCN-water solution (10 mM Tris-HCl, pH 7.2 at 25 °C). The intensities were recorded at 472 nm.

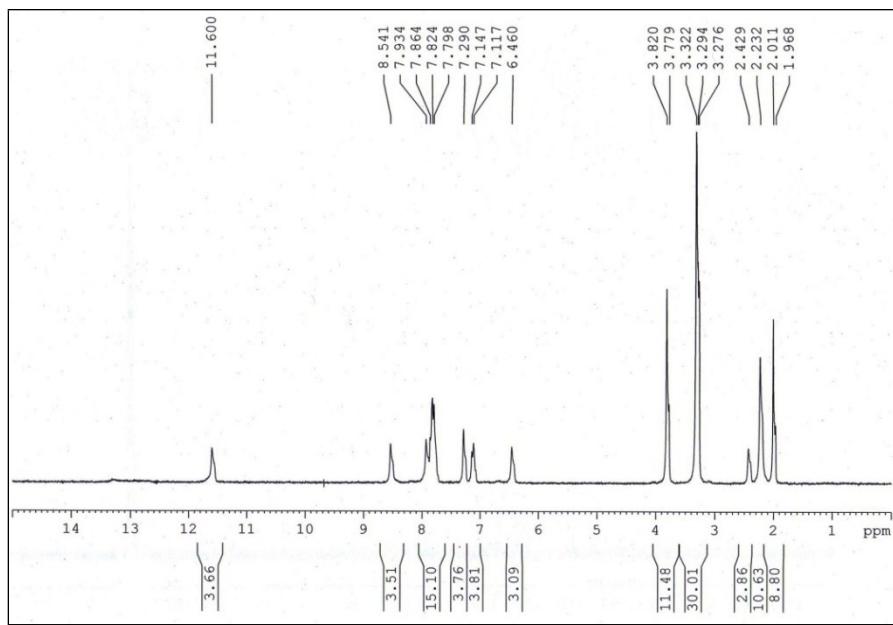


Fig. S19 ^1H NMR spectrum of **PYN-Zn** $^{2+}$ (1:1), in DMSO- d_6 in Bruker 300 MHz instrument.

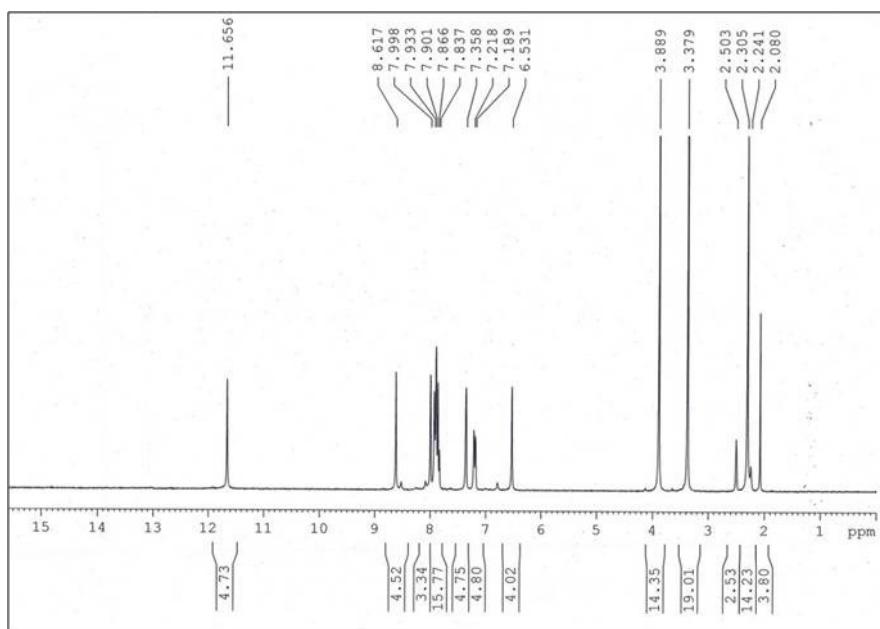


Fig. S20 ^1H NMR spectrum of **PYN-Mg** $^{2+}$ (1:1), in DMSO- d_6 in Bruker 300 MHz instrument

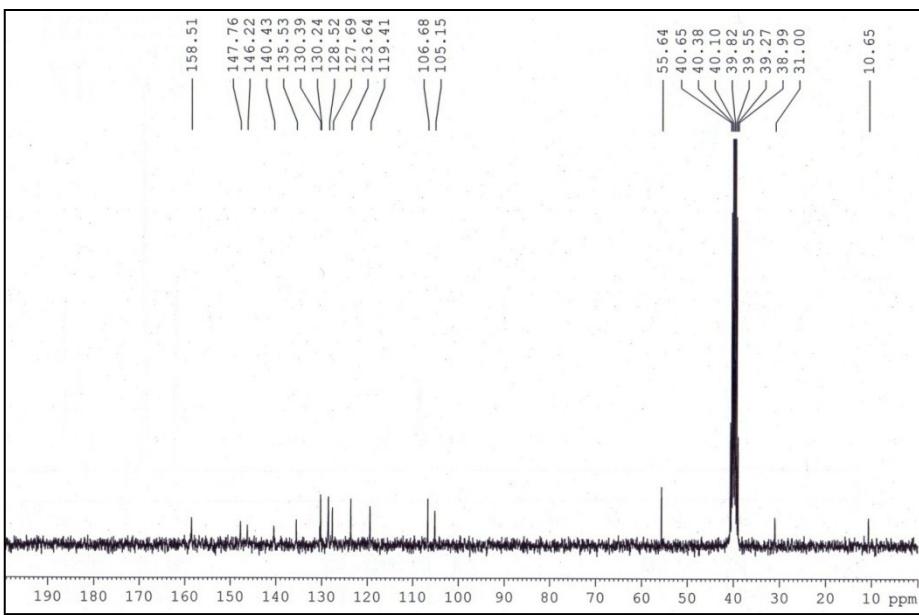


Fig. S21 ¹³C NMR spectrum of PYN-Zn²⁺ (1:1), in DMSO-d₆ in Bruker 300 MHz instrument.

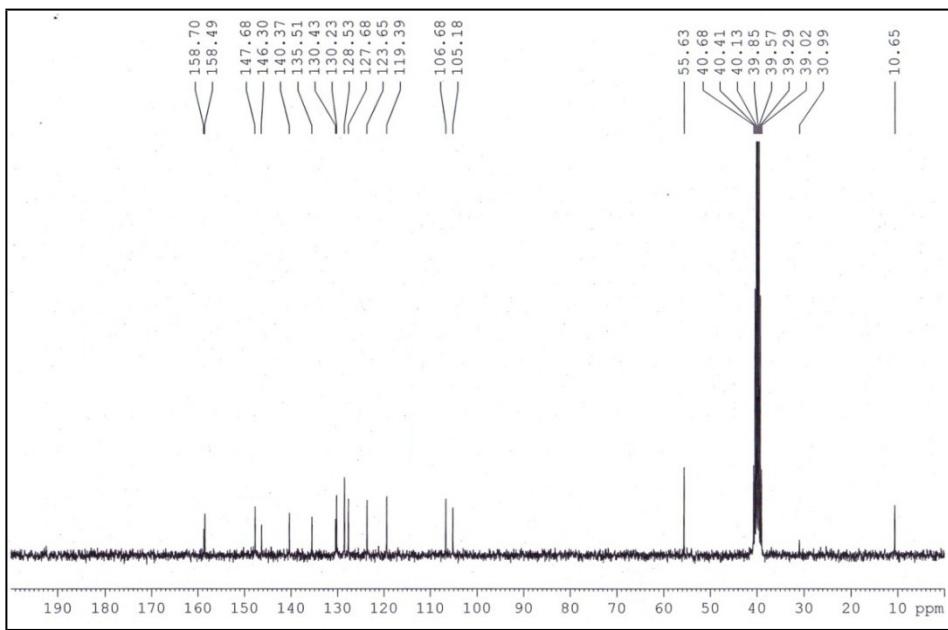


Fig. S22 ¹³C NMR spectrum of PYN-Mg²⁺ (1:1), in DMSO-d₆ in Bruker 300 MHz instrument.

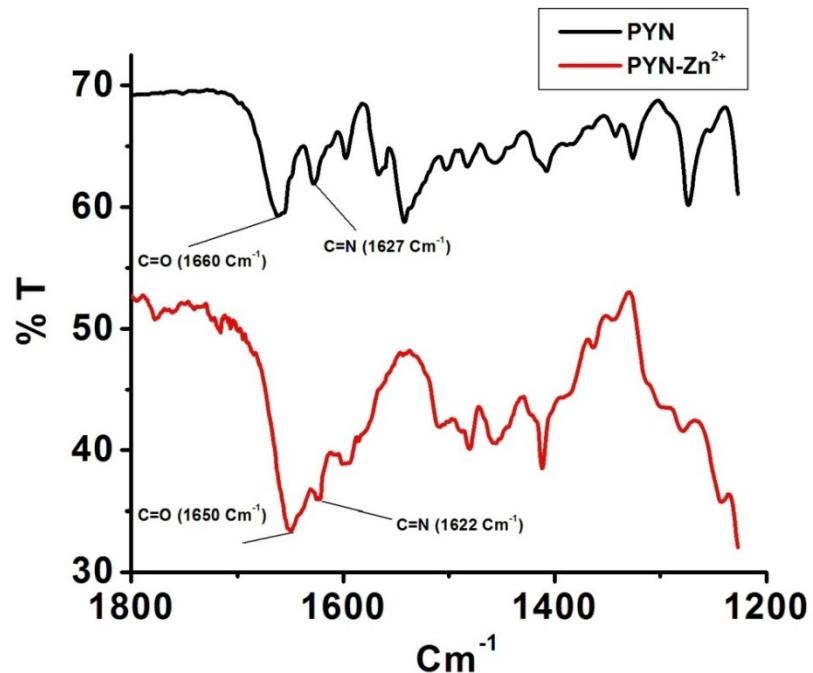


Fig. S23 FT-IR spectrum of **PYN- Zn^{2+}** in KBr pellet.

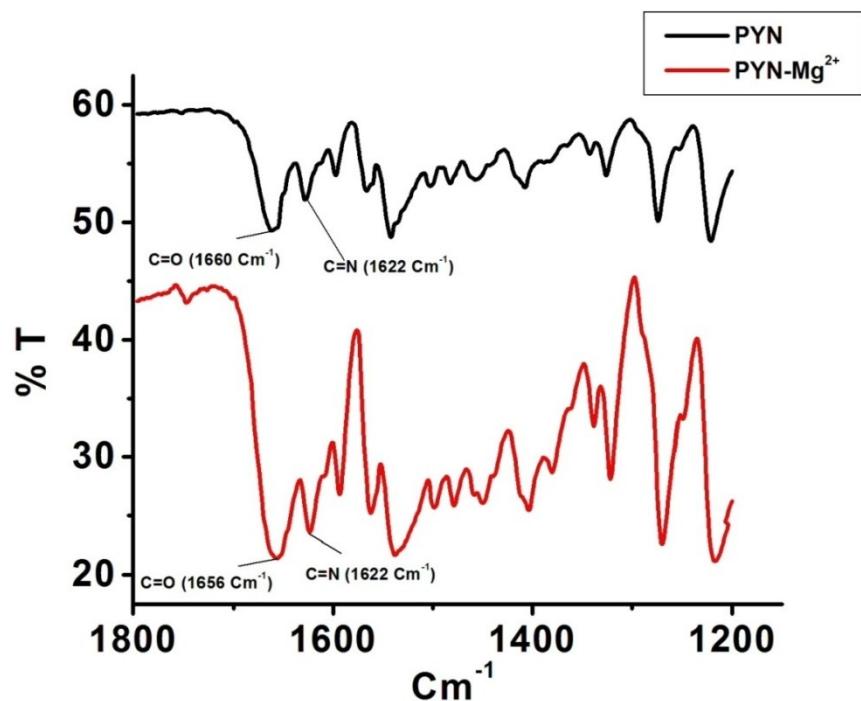


Fig. S24 FT-IR spectrum of **PYN- Mg^{2+}** in KBr pellet.

Table 1Fluorescence lifetimes of **PYN** and **PYN-Zn²⁺/Mg²⁺** complex in ACN/H₂O solvent

ACN/ H₂O	τ_{1(ns)}	τ_{2(ns)}	τ_{3(ns)}	α₁	α₂	α₃	χ²	τ_{av}	Φ
PYN	1.07	5.17	0.06	0.06	0.02	0.93	0.96	0.71	0.003
PYN+Zn²⁺	1.73	7.48	0.17	0.20	0.06	0.74	1.09	4.42	0.15
PYN+Mg²⁺	1.18	7.48	0.17	0.18	0.03	0.79	0.91	3.46	0.12

Table 2

Calculated selected bond angles ($^{\circ}$) and bond lengths (Å) data of **PYN** and complex **PYN-Zn²⁺/Mg²⁺** (B3LYP/ LANL2DZ level of theory).

bond angles ($^{\circ}$)	PYN-Zn²⁺	bond angles ($^{\circ}$)	PYN-Mg²⁺	bond angles ($^{\circ}$)	PYN
O ₂₂ -Zn ₃₉ -N ₂₆	79.32	O ₂₂ -Mg ₃₉ -N ₂₆	79.26	O ₂₂ -C ₂₁ -N ₂₀	123.86
Zn ₃₉ -N ₂₆ -N ₂₇	120.73	Mg ₃₉ -N ₂₆ -N ₂₇	145.30	N ₂₀ -N ₁₉ -C ₁₇	117.64
N ₂₆ -C ₂₃ -C ₂₁	117.90	N ₂₆ -C ₂₃ -C ₂₁	116.78	O ₂₂ -C ₂₁ -C ₂₄	124.24
Zn ₃₉ -O ₂₂ -C ₂₁	18.07	Mg ₃₉ -O ₂₂ -C ₂₁	84.97	C ₂₁ -N ₂₀ -N ₁₉	121.13
Zn ₃₉ -N ₂₆ -C ₂₃	108.12	Mg ₃₉ -N ₂₀ -C ₂₃	44.07	C ₂₁ -C ₂₄ -N ₂₇	119.24
O ₂₂ -C ₂₁ -C ₂₃	116.58	O ₂₂ -C ₂₁ -C ₂₃	15.10	C ₂₄ -N ₂₇ -N ₂₈	104.28
O ₂₂ -C ₂₁ -N ₂₀	129.36	O ₂₂ -C ₂₁ -N ₂₀	130.41		
C ₂₁ -N ₂₀ -N ₁₉	114.31	C ₂₁ -N ₂₀ -N ₁₉	114.12		
Bond lengths (Å)	PYN-Zn²⁺	Bond lengths (Å)	PYN		
Zn ₃₉ -O ₂₂	2.181	C ₂₁ -O ₂₂	1.206		
Zn ₃₉ -N ₂₆	1.991	C ₂₁ -N ₂₀	1.394		
O ₂₂ -C ₂₁	1.296	N ₁₉ -N ₂₀	1.352		
C ₂₁ -C ₂₃	1.486	N ₁₉ -C ₁₇	1.280		
N ₂₆ -C ₂₃	1.338	C ₂₁ -C ₂₄	1.494		
N ₂₆ -N ₂₇	1.349	C ₂₄ -N ₂₇	1.330		
C ₂₁ -N ₂₀	1.304	N ₂₇ -N ₂₈	1.391		
N ₂₀ -N ₁₉	1.379	C ₂₄ -C ₂₅	1.421		
Bond lengths (Å)	PYN-Mg²⁺				
Mg ₃₉ -O ₂₂	2.091				
Mg ₃₉ -N ₂₆	2.372				
O ₂₂ -C ₂₁	1.295				
C ₂₁ -C ₂₃	1.487				
N ₂₆ -C ₂₃	1.341				
N ₂₆ -N ₂₇	1.350				
C ₂₁ -N ₂₀	1.302				
N ₂₀ -N ₁₉	1.380				