Supplementary Information

Improved performances of LiNi_{0.6}Co_{0.15}Mn_{0.25}O₂ cathode material with full concentration-gradient for lithium ion batteries

Zhonghui Sun ab, Dandan Wang ab Yingying Fan ab, Liansheng Jiao abd, Fenghua Li ab,

Tongshun Wu^{ab}, Dongxue Han^{ab} and Li Niu^{abc*}

a State Key Laboratory of Electroanalytical Chemistry, c/o Engineering Laboratory for

Modern Analytical Techniques, CAS Center for Excellence in Nanoscience,

Changchun, 130022, P. R. China

b University of Chinese Academy of Sciences, Beijing, 100049, P. R. China

c School of Chemistry & Chemical Engineering, Linyi University, Linyi 276005, P. R.

China

d Department of Chemistry, Hebei Normal University for Nationalities, Chengde

067000, P. R. China.

^{*} Corresponding author, email:<u>lniu@ciac.ac.cn</u> (L. Niu), Fax: +86-431-526 2800.



Fig. S1 Digital photograph of as-prepared full concentration gradient precursor $Ni_{0.6}Co_{0.15}Mn_{0.25}(OH)_2$.

	Measured molar ratio			Designed molar ratio		
	Ni	Со	Mn	Ni	Со	Mn
FCG	0.597	0.147	0.256	0.60	0.15	0.25
СС	0.598	0.149	0.253	0.60	0.15	0.25

Table S1 Total chemical composition of FCG and CC precursors by ICP analysis

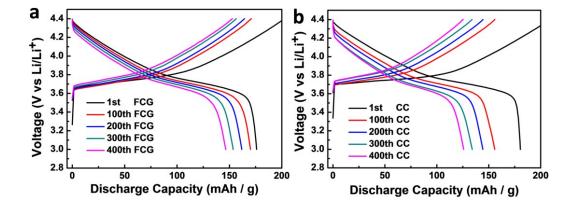


Fig. S2 1st, 100th, 200th, 300th, 400th charge-discharge curves of the FCG and CC $LiNi_{0.6}Co_{0.15}Mn_{0.25}O_2$ electrodes at 0.5 C-rate between 3.0 and 4.4 V.

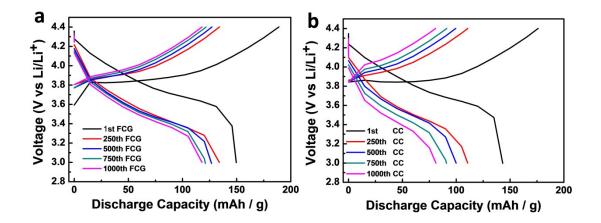


Fig. S3 1st, 250th, 500th, 750th, 1000th charge-discharge curves of the FCG and CC $LiNi_{0.6}Co_{0.15}Mn_{0.25}O_2$ electrodes at 5 C-rate between 3.0 and 4.4 V.