

## **Electronic Supplementary Information(ESI)**

# **Highly Sensitive Ratiometric Fluorescence Probes for Nitric Oxide Based on Dihydropyridine and Potentially Useful in Bioimaging**

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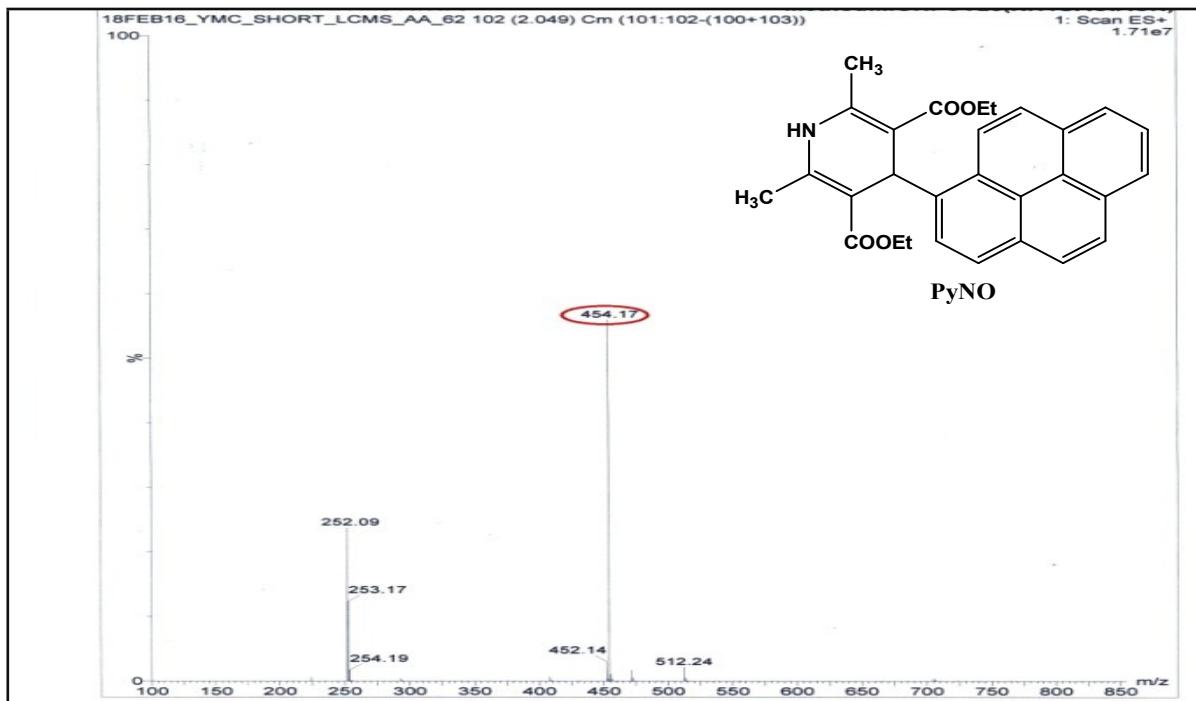
## General Information.

**Finding the Detection Limit.** The detection limit was calculated on the basis of the fluorescence titration .The fluorescence emission spectrum of **PyNO** and **TPANO** were measured, and the standard deviation of blank measurement was achieved. To gain the slope, the ratio of the fluorescence intensity was plotted as a concentration of NO. So the detection limit was calculated with the following equation.

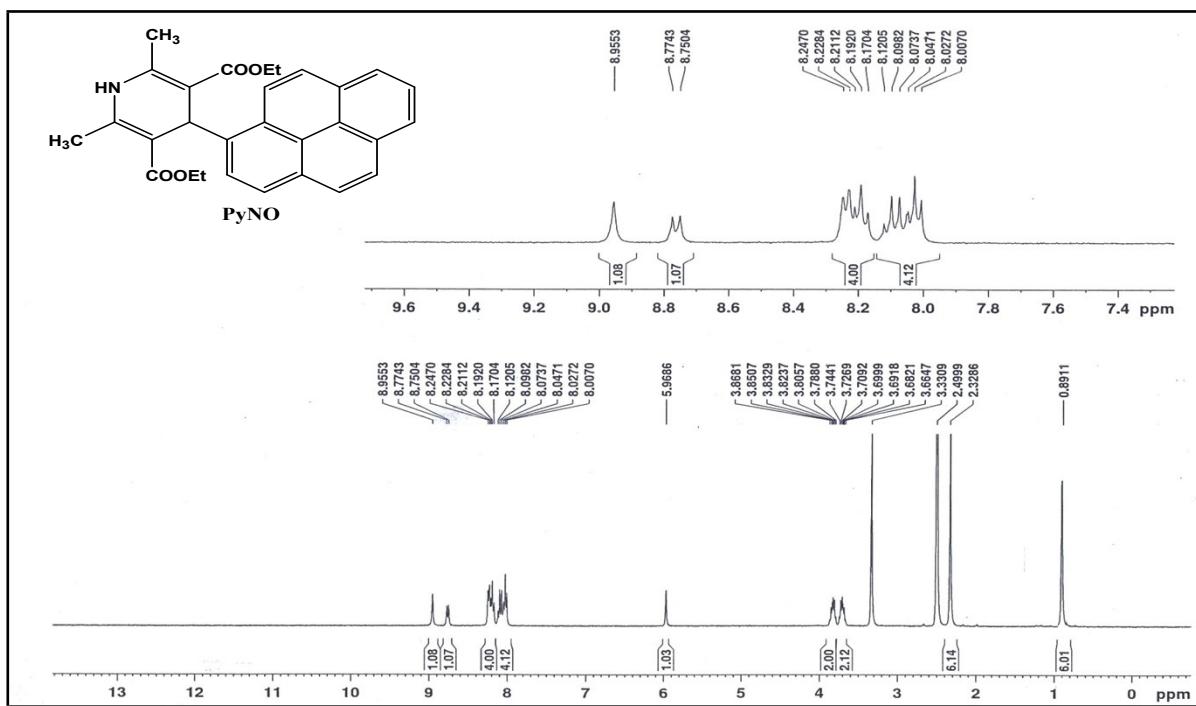
$$\text{Detection limit} = 3Sb1/S \quad (1)$$

where Sbl is the standard deviation of blank measurement and S is the slope of the calibration curve.

**Cytotoxicity Assay.** The cytotoxic effects of **TPANO** and **[TPANO + NO]** were determined by MTT assay following the manufacturer's instruction (MTT 2003, Sigma-Aldrich, MO). Vero cells were seeded onto 96-well plates (approximately 104 cells per well) for 24 h. Next day media was removed and various concentrations of probe **TPANO**, NO (0, 0.625, 1.25, 2.50 and 5.0  $\mu$ M) made in DMEM were added to the cells and incubated for 24 h. Solvent control samples (cells treated with DMSO in DMEM), no cells and cells in DMEM without any treatment were also included in the study. Following incubation, the growth media was removed, and fresh DMEM containing MTT solution was added. Subsequently, the supernatant was removed, the insoluble colored formazan product was solubilized in DMSO, and its absorbance was measured in a microtiter plate reader (Perkin-Elmer) at 570. The cell viability was calculated by the following formula: (mean OD in treated wells / mean OD in control wells) x 100.



### Figure S1. LCMS of PyNO



**Figure S2.**  $^1\text{H}$  NMR spectra of **PyNO**

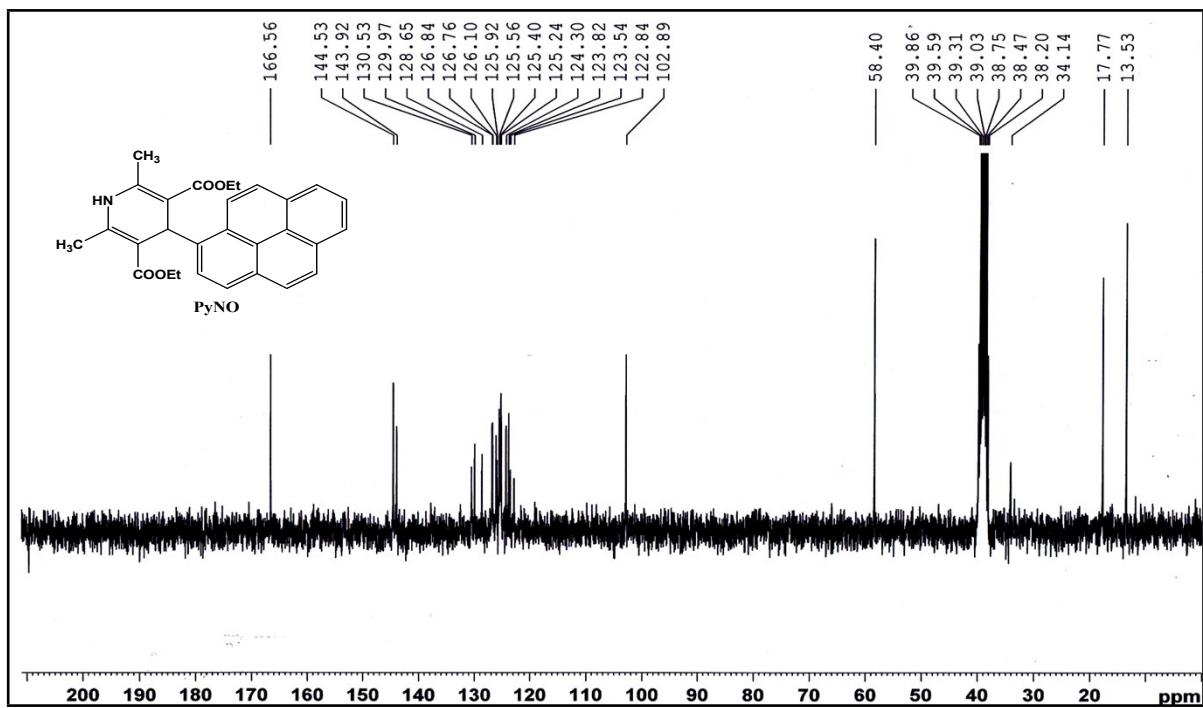


Figure S3.  $^{13}\text{C}$  NMR spectra of PyNO

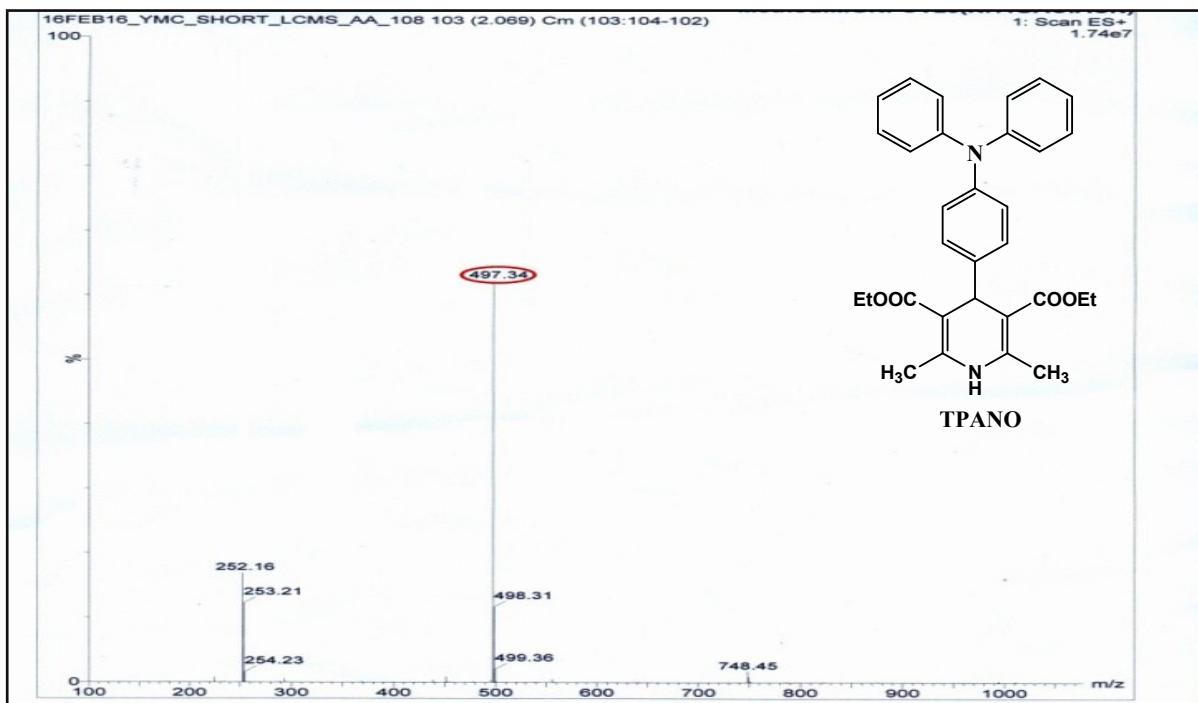


Figure S4. LCMS of TPANO.

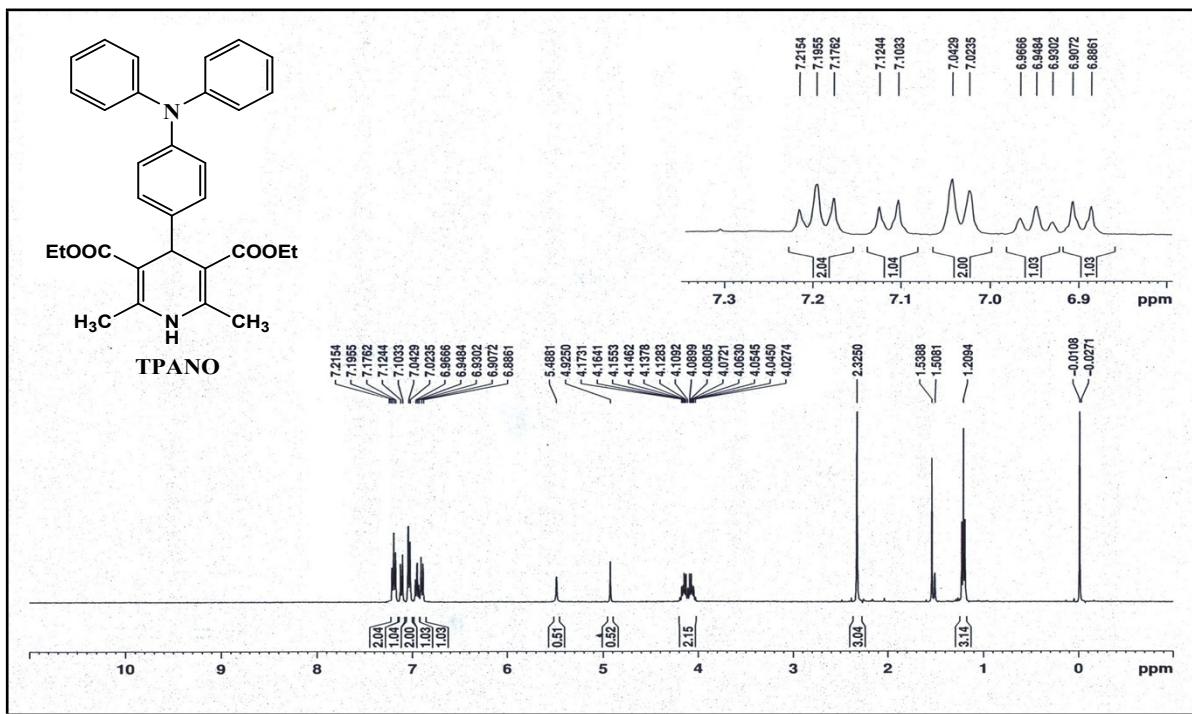


Figure S5.  $^1\text{H}$  NMR spectra of **TPANO**.

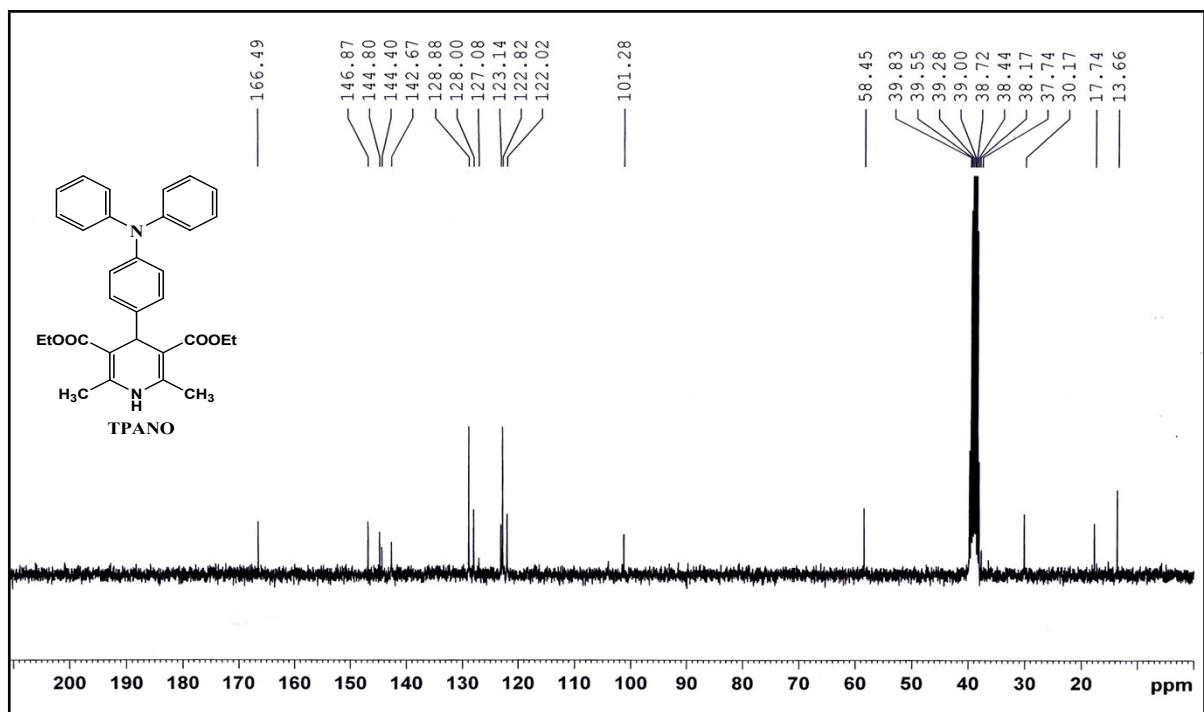
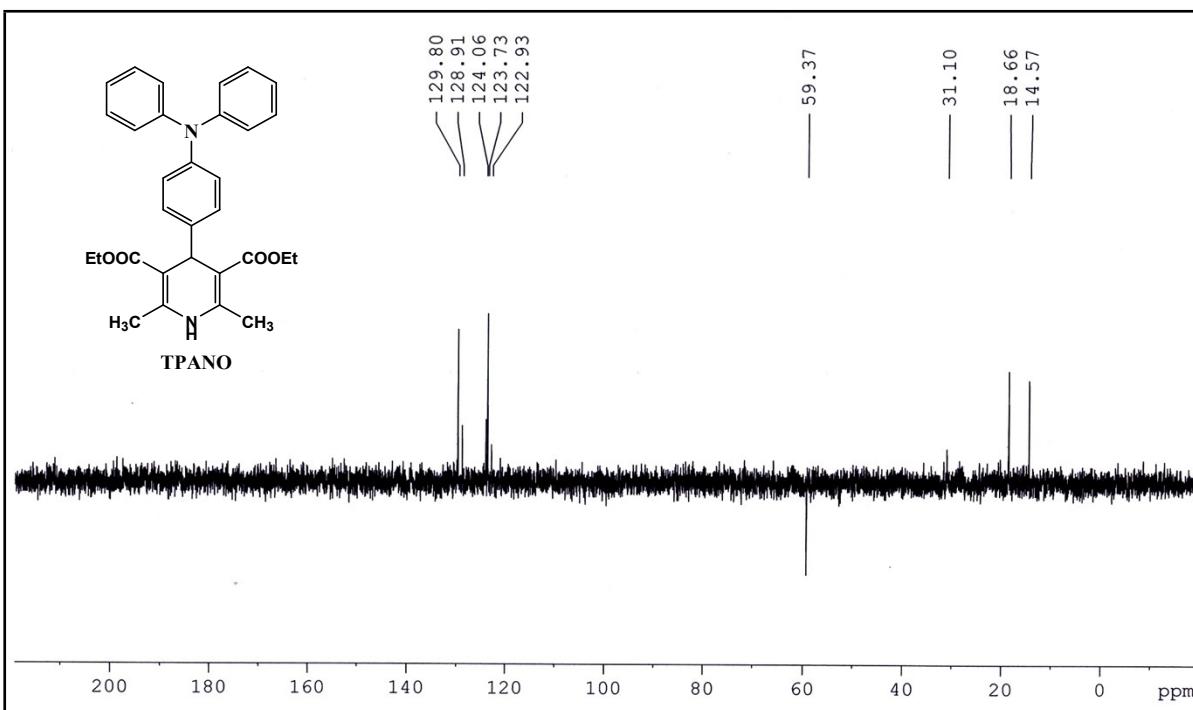
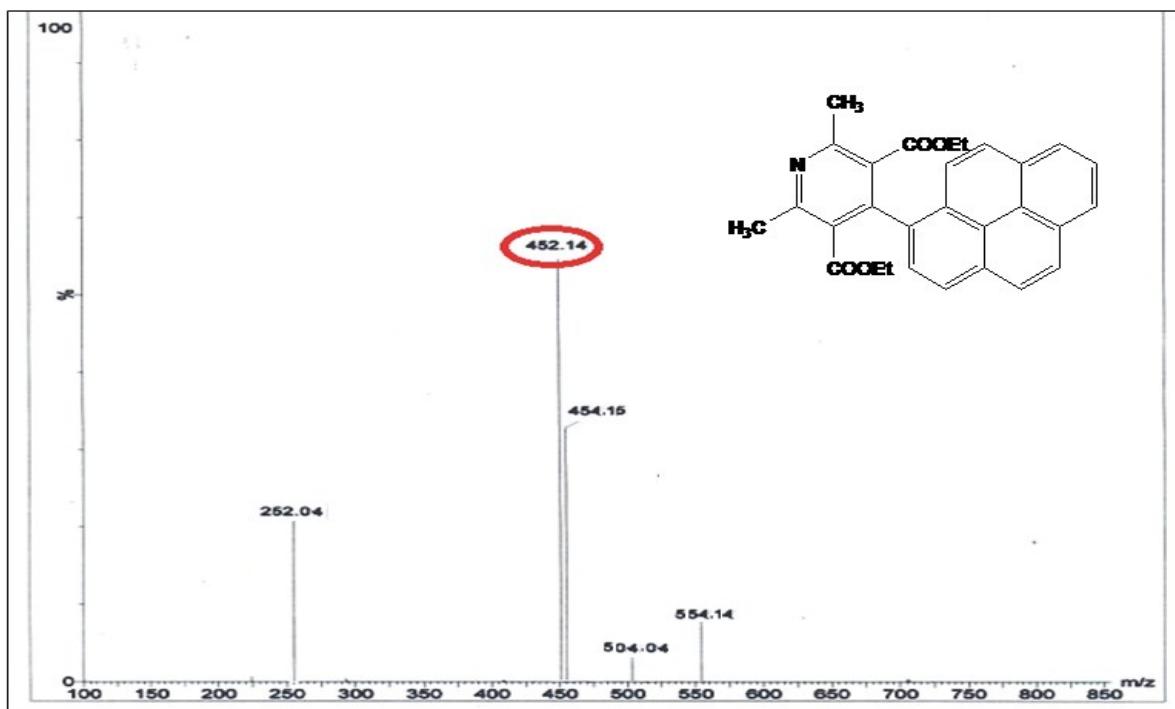


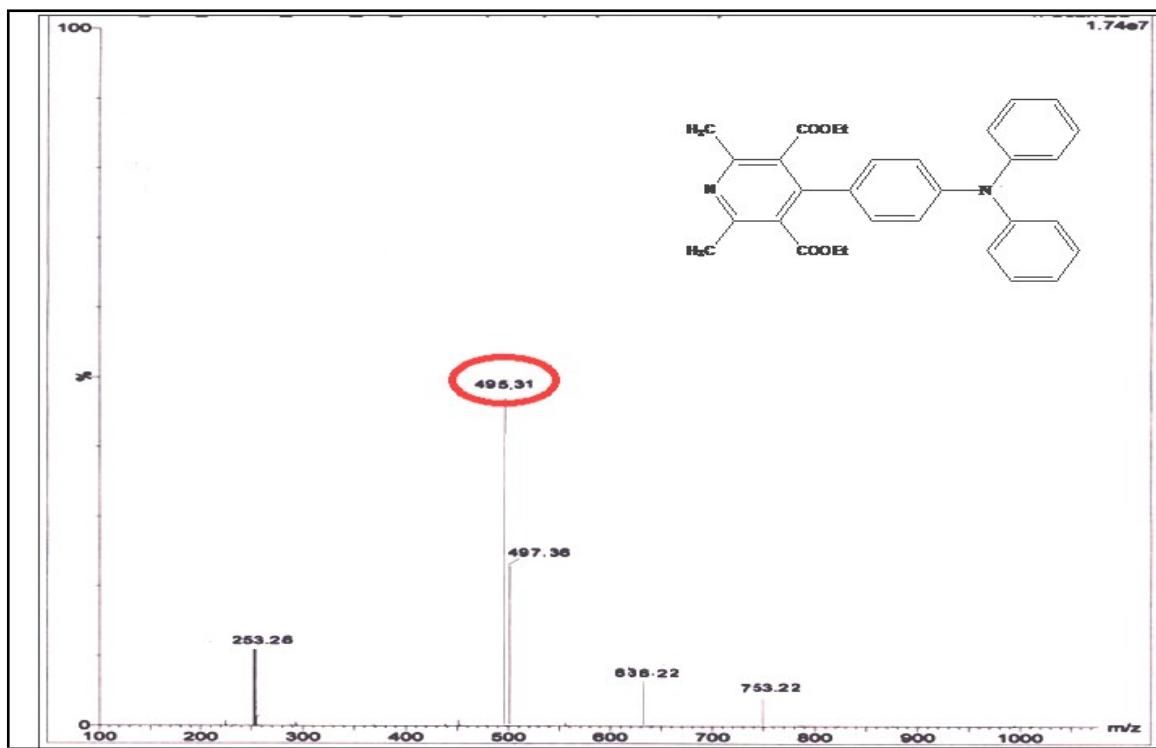
Figure S6.  $^{13}\text{C}$  NMR spectra of **TPANO**.



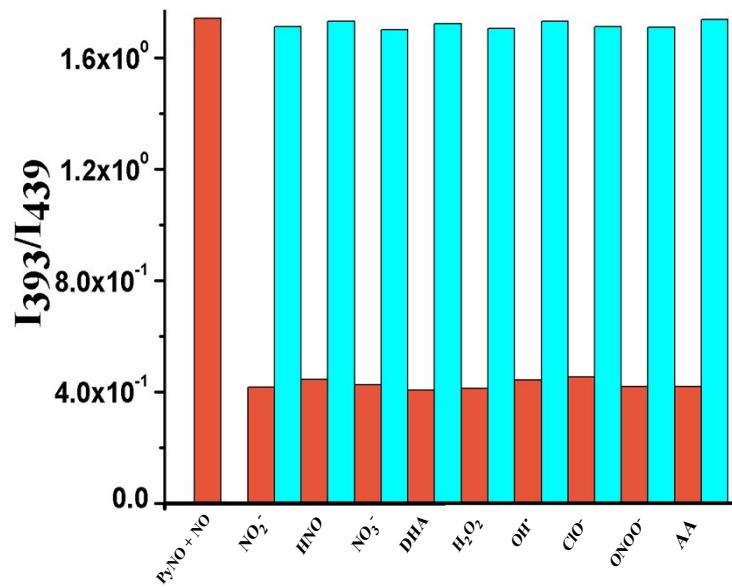
**Figure S7.** DEPT spectra of TPANO.



**Figure S8.** LCMS spectra of PyNO-Ar.



**Figure S9.** LCMS spectra of TPANO-Ar.



**Figure S10.** Competitive fluorescence response of PyNO.

### Quantum yield calculation:

Here, the quantum yield  $\varphi$  was measured by using the following equation:

$$\varphi_x = \varphi_s \left( \frac{F_x}{F_s} \right) \left( \frac{A_s}{A_x} \right) \left( \frac{n_s^2}{n_x^2} \right)$$

Where,

X & S indicate the unknown and standard solution respectively,  $\varphi$  = quantum yield,

F = area under the emission curve, A = absorbance at the excitation wave length,

n = index of refraction of the solvent. Here  $\varphi$  measurements were performed using Fluorescein in 0.1 M NaOH as standard [ $\varphi = 0.79$ ] and Anthracene [ $\varphi = 0.36$ ].

For standard (s) Fluorescein in 0.1 M NaOH the following values were determined:

$$n_s = 1.3330 \text{ (for 0.1 M NaOH)}; n_x = 1.4793 \text{ (for DMSO)}; \varphi = 0.79.$$

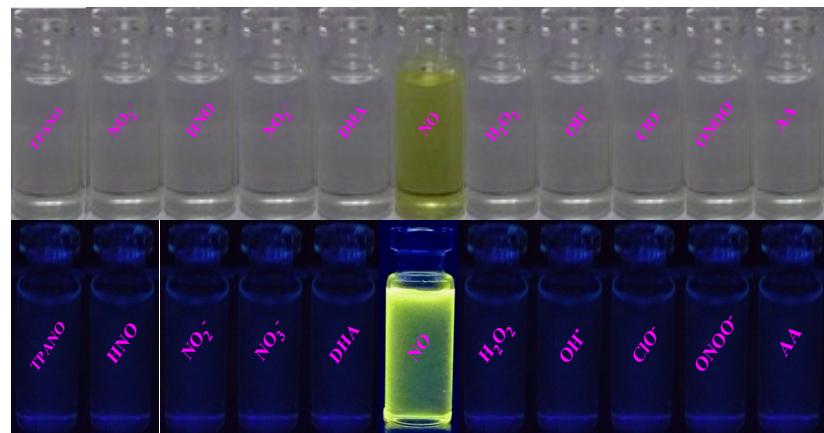
For standard (s) Anthracene the following values were determined:

$$n_s = 1.4793; n_x = 1.4793 \text{ (for DMSO)}; \varphi = 0.36.$$

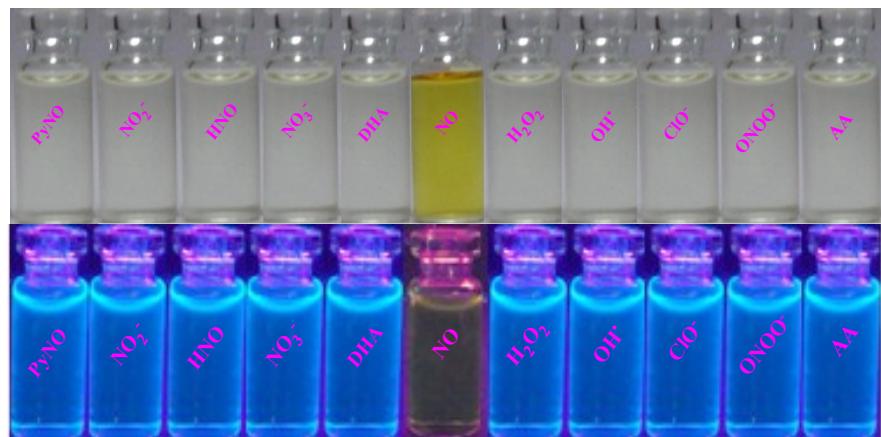
Using the above equation, we calculated quantum yield of probes .

Probes	Quantum yield ( $\varphi$ )
TPANO	0.02
TPANO-Ar	0.51
PyNO	0.15
PyNO-Ar	0.31

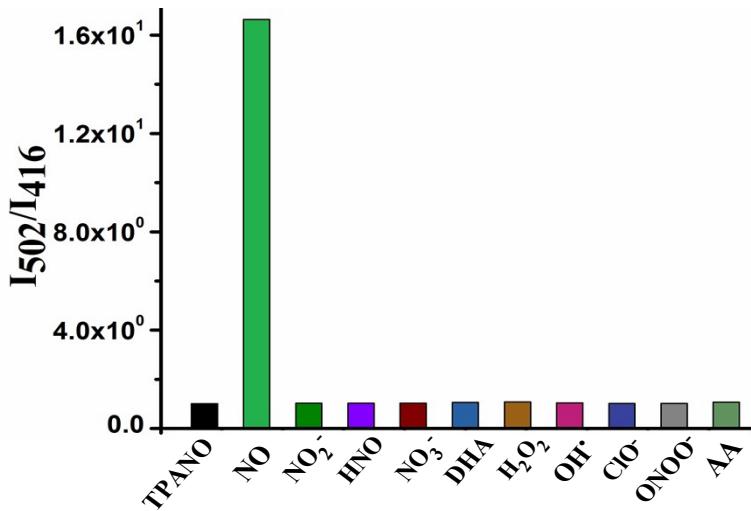
**Table S1.** Quantum yield data



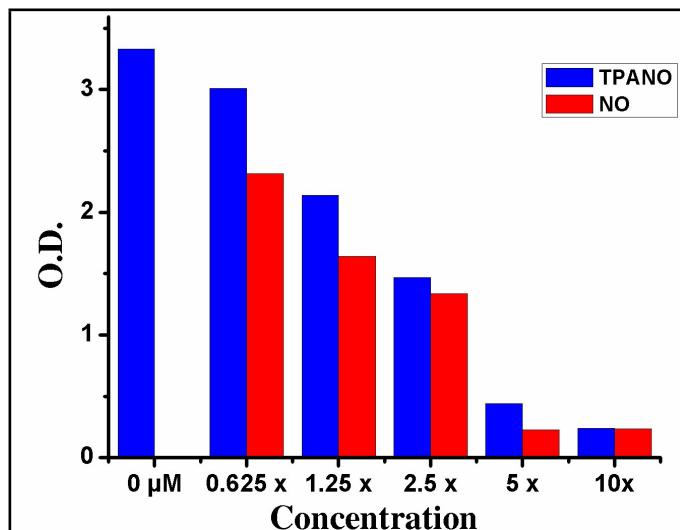
**Figure S11.** The visible color (top) and fluorescence changes (bottom) of receptor **TPANO** in aq. DMSO (DMSO: H<sub>2</sub>O = 7:3 v/v, 20 mM HEPES buffer, pH = 7.4) upon addition of various ions.



**Figure S12.** The visible color (top) and fluorescence changes (bottom) of receptor **PyNO** in aq. DMSO (DMSO: H<sub>2</sub>O = 7:3 v/v, 20 mM HEPES buffer, pH = 7.4) upon addition of various ions.



**Figure S13.** Bar profiles of fluorescence changes of **TPANO** and in presence of other ions.

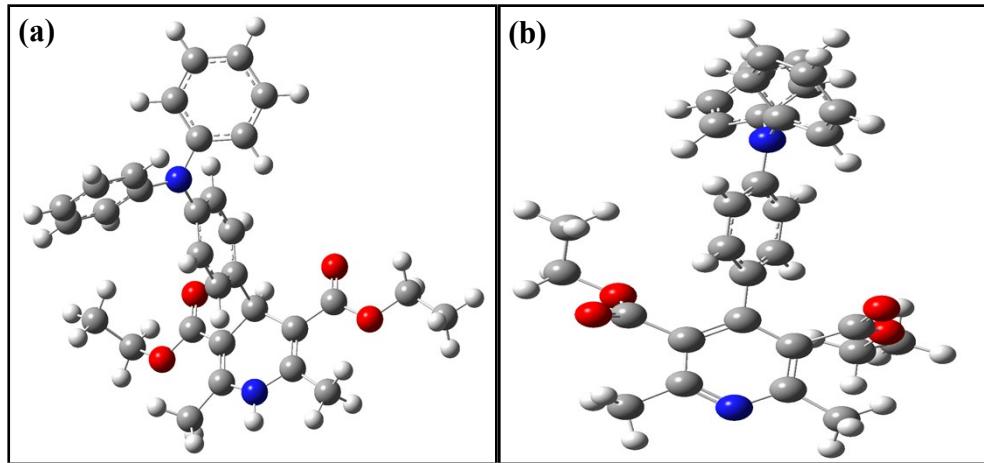


**Figure S14.** MTT assay to determine the cytotoxic effect of **TPANO** on Vero cell .

#### Computational Method:

Geometries have been optimized in B3LYP density functional method with 6-31+G (d, p) basis set for all atom. The geometries are verified as proper minima or not by frequency calculations.

Time-dependent density functional theory calculation has also been performed at the same level of theory.



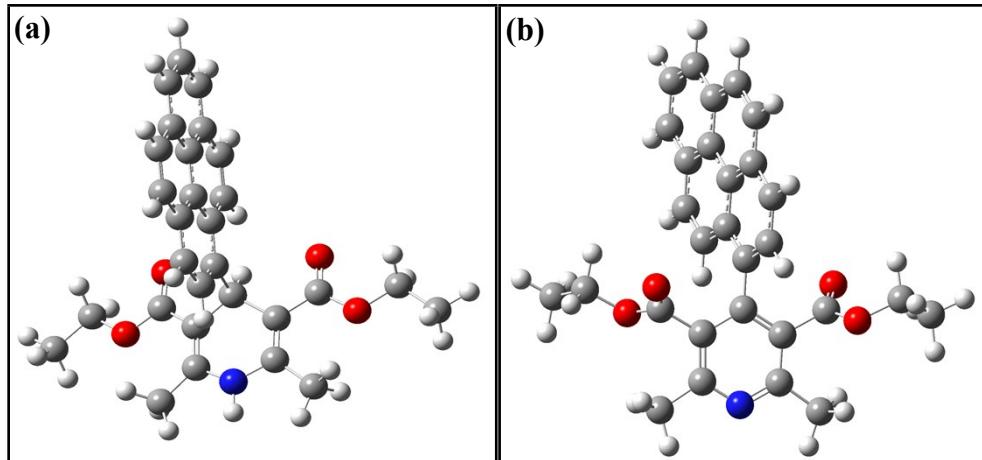
**Figure S15.** Energy minimization structure of (a) TPANO and (b) TPANO-Ar.

Species	E(HOMO)	E(LUMO)	E(LUMO+2)	E(LUMO+3)	E(LUMO+4)
TPANO					
	(-3.45 eV)	(-1.12 eV)	(0.25 eV)	(+0.65 eV)	(+1.02 eV)
TPANO-Ar					
	(-6.91 eV)	(5.10 eV)	(-1.70 eV)	(+1.82 eV)	(+2.70 eV)

**Table S2.** HOMO-LUMO energy calculated data of TPANO and TPANO-Ar.

$\lambda_{\max}(\text{nm})$	Osc. Strength ( $f$ )	Key Excitations
307.63	0.3139	(96.55%)HOMO $\rightarrow$ LUMO+2
300.37	0.3043	(96.71%)HOMO $\rightarrow$ LUMO+3

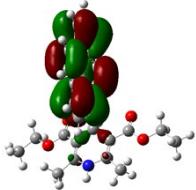
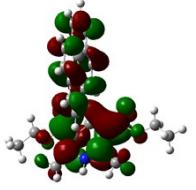
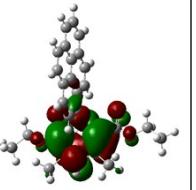
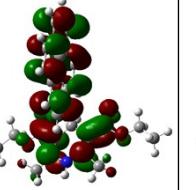
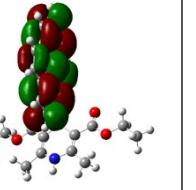
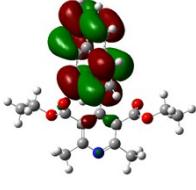
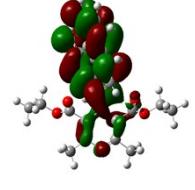
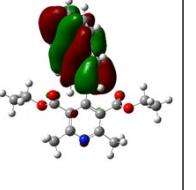
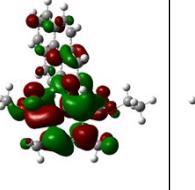
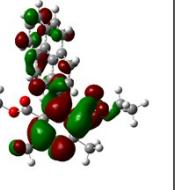
**Table S3.** Selected vertical electronic transitions of **TPANO** calculated by TDDFT method.



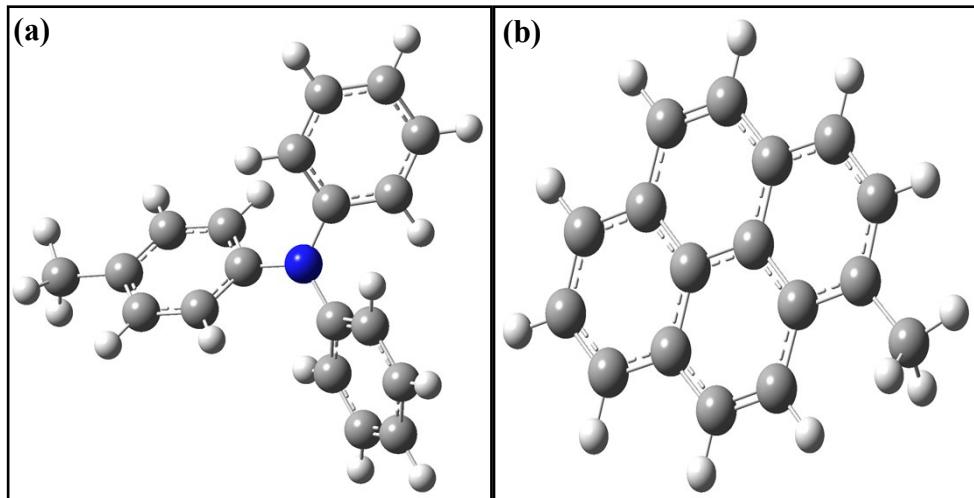
**Figure S16.** Energy minimization structure of (a) **PyNO** and (b) **PyNO-Ar**.

$\lambda_{\max}(\text{nm})$	Osc. Strength ( $f$ )	Key Excitations
323.34	0.1138	(82.07%)HOMO $\rightarrow$ LUMO+1
320.73	0.1776	(98.36%)HOMO-1 $\rightarrow$ LUMO+1

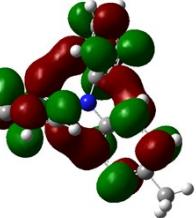
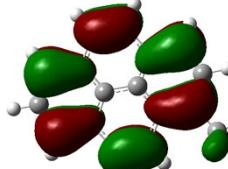
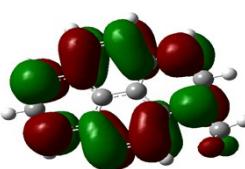
**Table S4.** Selected vertical electronic transitions of **PyNO** calculated by TDDFT method.

Species	E(HOMO)	E(LUMO)	E(HOMO-1)	E(LUMO+1)	E(LUMO+2)
PyNO	 (-3.75 eV)	 (-1.16 eV)	 (-5.13 eV)	 (-0.85 eV)	 (-0.43 eV)
PyNO-Ar	 (-5.32 eV)	 (-1.60 eV)	 (-6.22 eV)	 (-1.04 eV)	 (-0.78 eV)

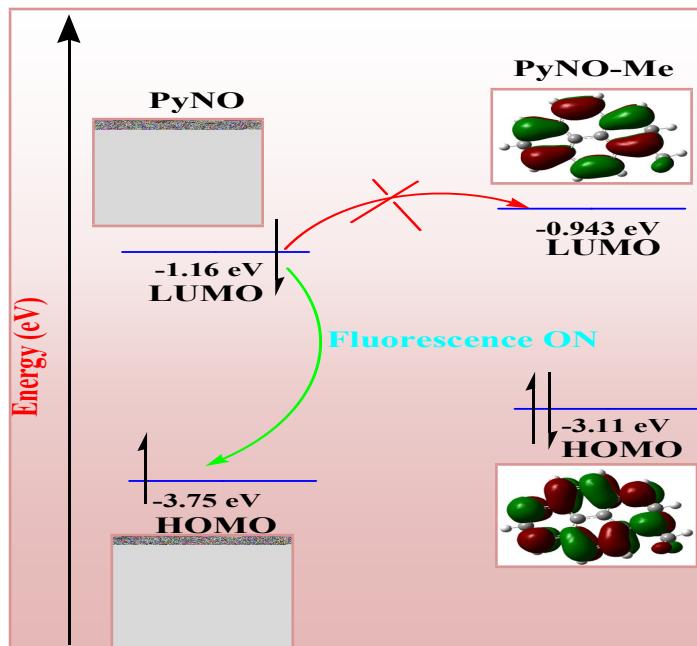
**Table S5.** HOMO-LUMO energy calculated data of PyNO and PyNO-Ar.



**Figure S17.** Energy minimization structure of (a) TPANO-Me and (b) PyNO-Me.

Species	E(HOMO)	E(LUMO)
TPANO-Me	 (-4.88 eV)	 (-2.92 eV)
PyNO-Me	 (-3.11 eV)	 (-0.94 eV)

**Table S6.** HOMO-LUMO energy calculated data of **TPANO-Me** and **PyNO-Me**.



**Figure S18.** Calculated energy level diagram of PyNO and PyNO-Me.

**Table S7.** Crystal Data and Details of the Structure Determination for **TPANO** and **PyNO**

Compound	TPANO	PyNO
CCDC Entry	1499388	1499389
Formula	$C_{31} H_{32} N_2 O_4$	$C_{29} H_{22} N O_4$
Formula Weight	496.58	448.13
Crystal Color	green	yellow
Crystal System	Monoclinic	Monoclinic
Space group	P21/n	P21
a, b, c (Å)	11.9094(13), 7.3443(8), 30.494(4)	8.690(5), 9.930(5), 13.591(5)
$\alpha(^{\circ})$	90.00	90.000(5)
$\beta(^{\circ})$	91.547(4)	90.853(5)
$\gamma(^{\circ})$	90.00	90.000(5)
$V(\text{\AA}^3)$	2666.2(5)	1172.7(10)
Z	4	2
D(calculated) [g/cm <sup>3</sup> ]	1.237	1.270
$\mu(\text{MoKa}) [\text{mm}^{-1}]$	0.082	0.085
F(000)	1056	470
Crystal Size [mm]	$0.30 \times 0.22 \times 0.16$	$0.20 \times 0.12 \times 0.10$
Absorption Correction	Multi-scan	Multi-scan
Temperature (K)	295(2)	293(2)

Radiation [Å]	0.71073	0.71073
θ(Min-Max) [°]	2.60– 28.28	2.3–28.3
Dataset (h; k; l)	-15 to 15; - 9 to 9; -40 to 40	-11 to 11; - 13 to 13; -18 to 18
Total, Unique Data, R(int)	37027, 5365,0.0877	22479, 4735,0.1134
Observed data [I > 2σ(I)]	5365	4735
Nref, Npar	6656, 351	5847, 316
R, wR, S	0.1066, 0.2118, 1.028	0.0867, 0.2181, 1.047
Δq(max)and Δq(min) (e/Å³)	0.44 and -0.33	0.365 and -0.258
Goodness of Fits	1.028	1.047

$$R = \sum |(|F_o| - |F_c|)| / \sum |F_o|$$

$$wR = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$$

$$w = 1/[\sigma^2(F_o^2) + (0.0721P)^2 + 2.3397 P] \text{ for TPANO}$$

$$w = 1/[\sigma^2(F_o^2) + (0.1015P)^2 + 0.3581P] \text{ for PyNO}$$

$$\text{Where } P = (F_c^2 + 2 F_c^2)/3.$$

$GOF (S) = \{\sum [w(F_o^2 - F_c^2)^2] / (n - p)\}^{1/2}$ , where  $n$  = number of measured data and  $p$  = number of parameters.

**Table S8.** Selected bond distances (Å) and bond angles (°) of **TPANO** and **PyNO**

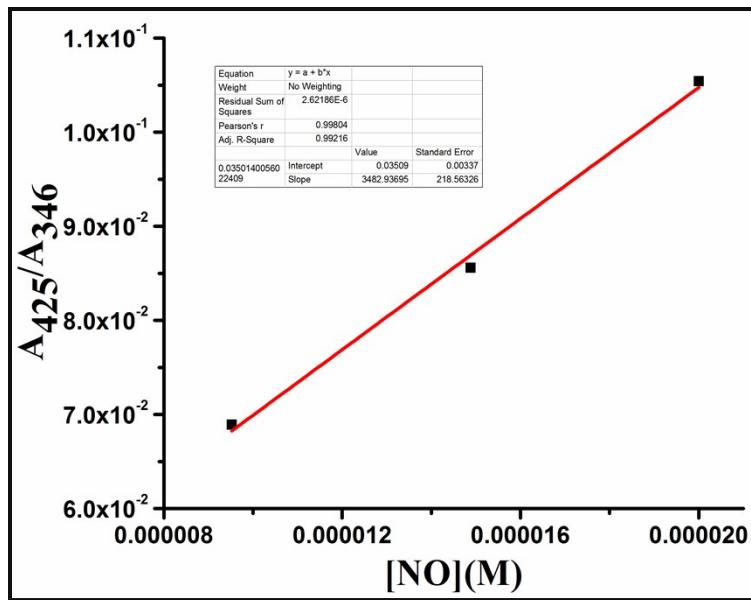
TPANO			PyNO		
Bonds(Å)	X-ray	DFT data	Bonds(Å)	X-ray	DFT data
O1 – C25	1.202(3)	1.2202	C1 – C17	1.529(6)	1.5399
O2 – C25	1.335(3)	1.3642	C17 – C18	1.531(6)	1.5270
O3 – C29	1.195(3)	1.2198	C17 – C21	1.515(7)	1.5271
O4 – C29	1.380(9)	1.3632	N1 – C19	1.381(6)	1.3895
N2 – H3A	0.8600	1.0079	N1 – C20	1.366(7)	1.3895
N2 – C21	1.370(3)	1.3899	O1 – C22	1.192(9)	1.2199
N2 – C22	1.384(3)	1.3895	O3 – C25	1.206(7)	1.2199
C19 – H19	0.980	1.0915	O2 – C22	1.331(7)	1.3619
C16 – C19	1.522(3)	1.5326	O4 – C25	1.333(6)	1.3619
<b>Angles (°)</b>			O2 – C23	1.432(7)	1.4445
C16 – C19 – C20	111.6(2)	111.76	O4 – C26	1.444(7)	1.4445
C16 – C19 – C23	109.0(2)	111.77	C26 – C27	1.474(13)	1.5168
C20 – C19 – C23	110.2(2)	111.12	<b>Angles (°)</b>		
C21 – N2 – C22	123.1(2)	124.28	C18 – C17 – C21	110.1(2)	111.13
			C17 – C21 – C20	120.7(3)	120.767
			C17 – C18 – C19	120.8(2)	120.767
			C18 – C19 – N1	119.2(3)	118.514

			C19 – N1 – C20	123.1(3)	124.27
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**Table S9.** Analysis of the Hydrogen Bonding Network

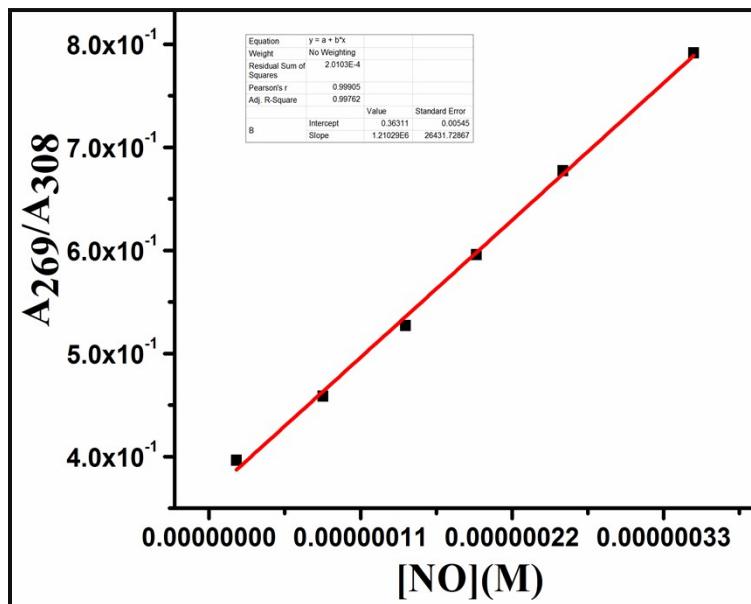
<b>TPANO</b>					
Donor(D) --- H ...Acceptor (A)	$d_{N-H}(\text{\AA})$	$d_{H-O}(\text{\AA})$	$d_{N-O}(\text{\AA})$	$\angle(D-H...A)^{\circ}$	symmetry
N2—H3A.....O3	0.8600	2.1400	2.996(2)	175.00	x, 1+y, z
<b>PyNO</b>					
N1—H1.....O3	0.86(4)	2.16(4)	2.997(4)	165(4)	1-x,-1/2+y,2-z

Crystallographic data for the structures reported in this paper in the form of CIF files have been deposited with the Cambridge Crystallographic Data Center as supplementary publication Nos. 1499388-1499389 for compound **TPANO** and **PyNO** respectively. Copies of the data can be obtained free of charge via <http://www.ccdc.cam.ac.uk/conts/retrieving.html> or from CCDC, 12 Union Road, Cambridge CB2 IEZ, UK (Fax: + 44 1223 336 033; E-mail: [deposit@ccdc.cam.ac.uk](mailto:deposit@ccdc.cam.ac.uk)).



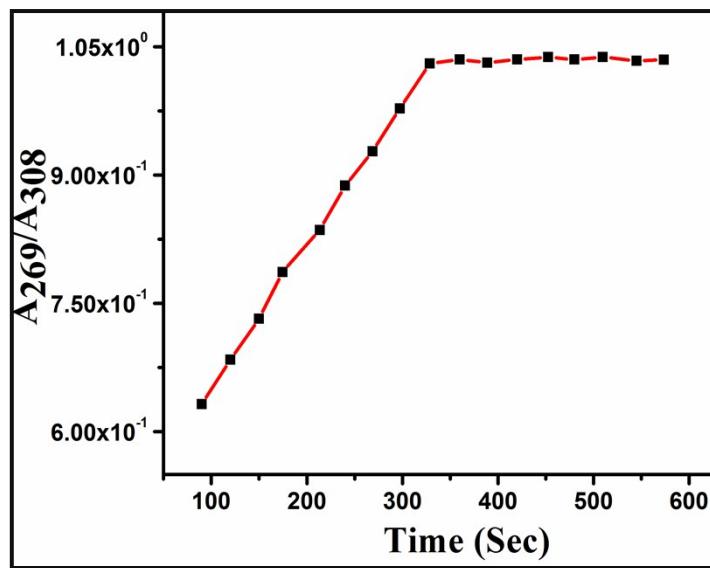
**Figure S19.** Calibration curve for **PyNO** with **NO** by UV-Vis titration.

Thus using the formula we get the detection limit= 2.9  $\mu$ M.

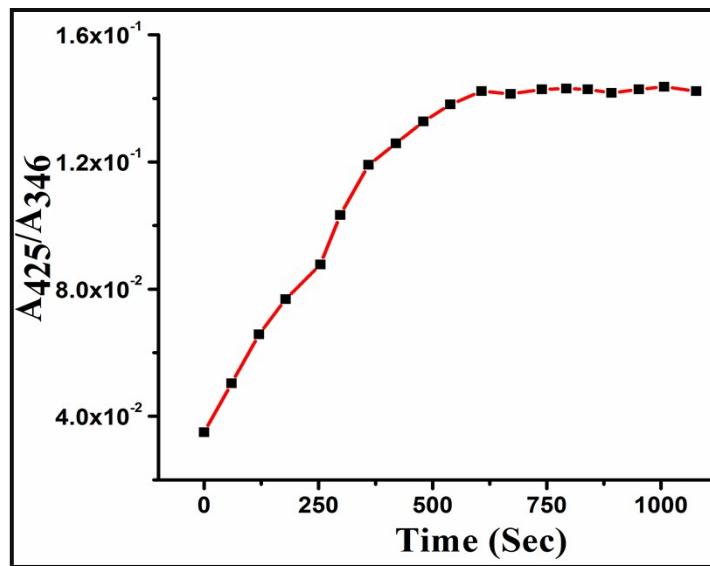


**Figure S20.** Calibration curve for **TPANO** with **NO** by UV-Vis titration.

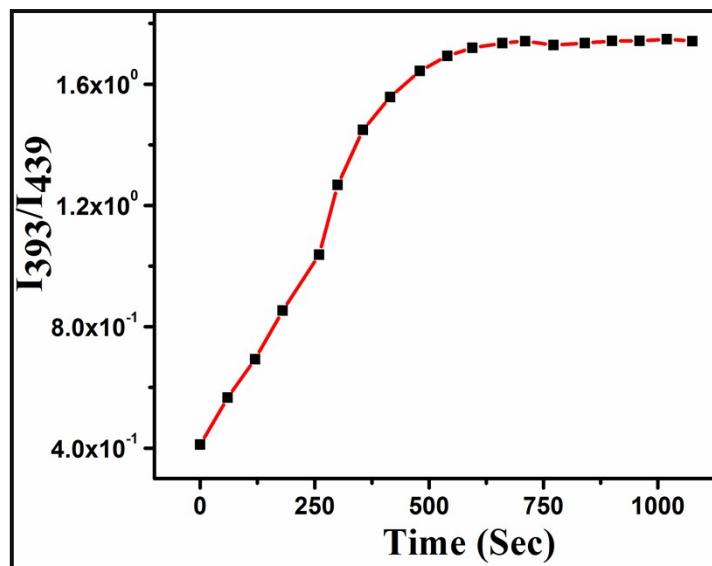
Thus using the formula we get the detection limit= 0.013  $\mu$ M.



**Figure S21.** Increments of absorption intensity of **TPANO** in pH 7.4, 10 mM HEPES buffer against time (0 - 6 min) after the addition of NO at different concentration.



**Figure S22.** Increments of absorption intensity of **PyNO** in pH 7.4, 10 mM HEPES buffer against time (0 - 11 min) after the addition of NO at different concentration.



**Figure S23.** Increments of fluorescence intensity of **PyNO** in pH 7.4, 10 mM HEPES buffer against time (0 - 11min) after the addition of NO at different concentration.