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Supplementary

NMR spectra (in DMSO-d₆) of SCX4, TSCX4, SCX6 hosts, Rh6G and their complexes.



Fig. 1S: ¹H NMR spectrum of Rh6G in DMSO-d₆. (* indicates peaks of solvent)



Fig. 2S: ¹H NMR spectrum of SCX4in DMSO-d₆. (* indicates peak of solvent)



Fig. 3S: ¹H NMR spectrum of TSCX4in DMSO-d₆. (* indicates peak of solvent)

TSCX4



Fig. 4S: ¹H NMR spectrum of SCX6 in DMSO-d₆. (* indicates peak of solvent)

Table 1S: Chemical shifts along with $\Delta\delta$ for TSCX4 (0.01 mM), Rh6G (0.01 mM) and Rh6G \subset TSCX4 in DMSO-d₆.

Proton assignment		Chemical shift	$(\Delta \delta = \delta - \delta_0)$ (ppm)
	δ_0	δ	Δδ
H _f	8.27	8.26	-0.01
H_{g}	7.91	7.93	0.02
H _h	7.84	7.84	0
H _i	7.72	7.48	-0.24
NĤ	7.46	-	-
А	6.94	6.89	-0.05
A'	6.78	6.76	-0.02
CH_2 (e)	3.53	3.56	0.03
CH_2 (b)	2.14	3.45	1.31
$CH_3(c)$	2.08	2.09	0.01
$CH_3(a)(d)$	1.27	1.26	-0.01
Ar-H	7.76	7.8	0.04
Ar-OH	3.38	4.04	0.66

Table 2S: Chemical shifts along with $\Delta\delta$ for SCX6 (0.01 mM) and Rh6G (0.01 mM) and Rh6G \subset SCX6 in DMSO-d₆.

Proton assignment	Chemic	$(\Delta \delta = \delta - \delta_0)$ (ppm)	
	δ_0	δ	Δδ
H _f	8.27	8.23	-0.04
H_{g}	7.91	7.88	0.03
H _h	7.84	7.8	-0.04
H _j	7.72	7.41	-0.31
NH	7.46	-	-
А	6.94	6.86	-0.08
A'	6.78	6.75	-0.03
CH ₂ (e)	3.53	3.53	0
CH_2 (b)	2.14	3.43	1.29
CH ₃ (c)	2.08	2.04	-0.04
CH ₃ (a)(d)	1.27	1.22	-0.05
Ar-H	7.25	7.26	0.01
Ar-OH	4.76	5.39	0.63
Ar-CH ₂ -Ar	3.85	3.84	-0.01



Fig. 5S: ¹H NMR spectra showing aromatic region of Rh6G (0.01 mM) and (Rh6G⊂SCX4) complex with different mole ratio (0.3, 0.4, 0.5, 0.6, and 0.7) in DMSO-d₆ at room temperature.



Fig. 6S: ¹H NMR spectra (500 MHz, 298K) recorded in DMSO-d₆ for a) SCX6, b) 1:4 Rh6G \subset SCX6, c) 1:3 Rh6G \subset SCX6, d) 1:1 Rh6G \subset SCX6, e) 2:1 Rh6G \subset SCX6, f) 3:1 Rh6G \subset SCX6 and g) 4:1 Rh6G \subset SCX6.





Fig. 7S: 2D COSY spectrum of a) Rh6G⊂SCX4, b) Rh6G⊂TSCX4, and c) Rh6G⊂SCX6 1:1 complexes in DMSO-

d₆.



Fig 8S: (a) Partial ¹H NMR spectra (400 MHz, DMSO-d₆, 298 K) of SCX4 at the concentration of 5mM upon addition of Rh6G; (1) 0 mM; (2) 1 mM; (3) 2 mM; (4) 3 mM; (5) 5 mM; (6) 10mM; (7) 20mM; (8) 30 mM (b) Titration curve of SCX4 with Rh6G showing changes in the chemical shifts of –OH against an increasing amount of Rh6G.



Fig 9S: Partial ¹H NMR spectra (400 MHz, DMSO-d₆, 298 K) of TSCX4 at the concentration of 5mM upon addition of Rh6G; (1) 0 mM; (2) 1 mM; (3) 2 mM; (4) 3 mM; (5)4 mM; (6) 5mM; (7) 10mM; (8) 20 mM; (9) 30 mM (b) Titration curve of TSCX4 with Rh6G showing changes in the chemical shifts of –OH against an increasing amount of Rh6G.







Fig. 10S: FT-IR spectra of a) Rh6G, b) SCX4 and Rh6G⊂SCX4 c) TSCX4 and Rh6G⊂TSCX4, and d) SCX6 and Rh6G⊂SCX6 complex in 1:1 mole ratio.



Fig. 11S: ω B97x/6-31G(d,p) optimized high enrgy complexes

	Rh6G		SCX4		Rh6G⊂SCX4		TSCX4		Rh6G⊂TSCX4	
Proton	Theory	Experiment	Theory	Experiment	Theory	Experiment	Theory	Experiment	Theory	Experiment
ArCH2	_	-	3.8	3.9	4.0	3.9	_	-	-	-
Ar-H	_	-	8	7.3	8.1	7.4	8.8	7.8	8.8	7.8
А	7.2	6.9	-	-	6.9	6.9	-	-	6.9	6.9
Α'	6.8	6.8	-	-	6.8	6.8	-	-	6.8	6.8
а	1.3	1.3	-	-	1.3	1.2	-	-	1.3	1.3
b	3.6	2.1	-	-	3.7	3.5	-	-	3.7	3.5
с	2.2	2.1	-	-	2.4	2.1	-	-	2.4	2.1
d	1.4	1.3	-	-	2.0	1.2	-	-	2.0	1.3
е	4	3.5	-	-	4.6	3.6	-	-	4.6	3.6
f	8.8	8.3	-	-	6.3	8.3	-	-	6.2	8.3
g	8.2	7.9	-	-	3.8	7.9	-	-	4.3	7.9
h	8.2	7.8	-	-	6.6	7.8	-	-	6.3	7.8
i=k	4.7	7.5	-	-	4.9	2.5	-	-	6.6	-

Table 3S: A comparison of ¹H NMR chemical shifts in the free Rhodamine-6G, SCX4, TSCX4 and their complexes (Solvent=DMSO)



Fig. 12S: Spectral overlap between excitation and emission spectrum of Rh6G λ_{ex} = 535 nm, λ_{em} = 563 nm.



Fig. 13S: Decay traces of Rh6G (0.001 M) in DMSO λ_{ex} = 535 nm.



Fig. 14S: Antimicrobial activity of sulfonated calixarenes derivatives Rh6G⊂SCX4, Rh6G⊂TSCX4 and Rh6G⊂SCX6 with varying concentrations (25, 50, 100 ppm) against (a, b and c) *B. subtilis* ATCC 6633 and (d,e and f) *E. coli* ATCC 8739 at 37 °C after 24 h.