

Supporting Information (SI)

Phase transformations of novel Cu_xS nanostructures as highly efficient counter electrodes for stable and reproducible quantum dot-sensitized solar cells

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Table S1. Atomic ratios of copper sulfides determined by SEM-EDX quantitative analysis

Thin films	Cu (at%)	S (at%)	Ratio (Cu/S)	Average Ratio
Cu₂S (Thin film1)	66.05	33.5	1.98	
Cu₂S (Thin film2)	67.1	32.9	2.03	2.01±0.02
Cu₂S (Thin film3)	66.9	33.1	2.02	
Cu_{1.75}S (Thin film1)	64.2	35.8	1.79	
Cu_{1.75}S (Thin film2)	63.8	36.2	1.76	1.76±0.03
Cu_{1.75}S (Thin film3)	63.4	36.6	1.73	
Cu_{1.12}S (Thin film1)	53.6	46.4	1.15	
Cu_{1.12}S (Thin film2)	53.2	46.8	1.13	1.13±0.02
Cu_{1.12}S (Thin film3)	52.9	47.1	1.12	
CuS (Thin film1)	51.9	49.1	1.07	
CuS (Thin film2)	51.4	48.6	1.05	1.03±0.04
CuS (Thin film3)	49.3	50.7	0.97	

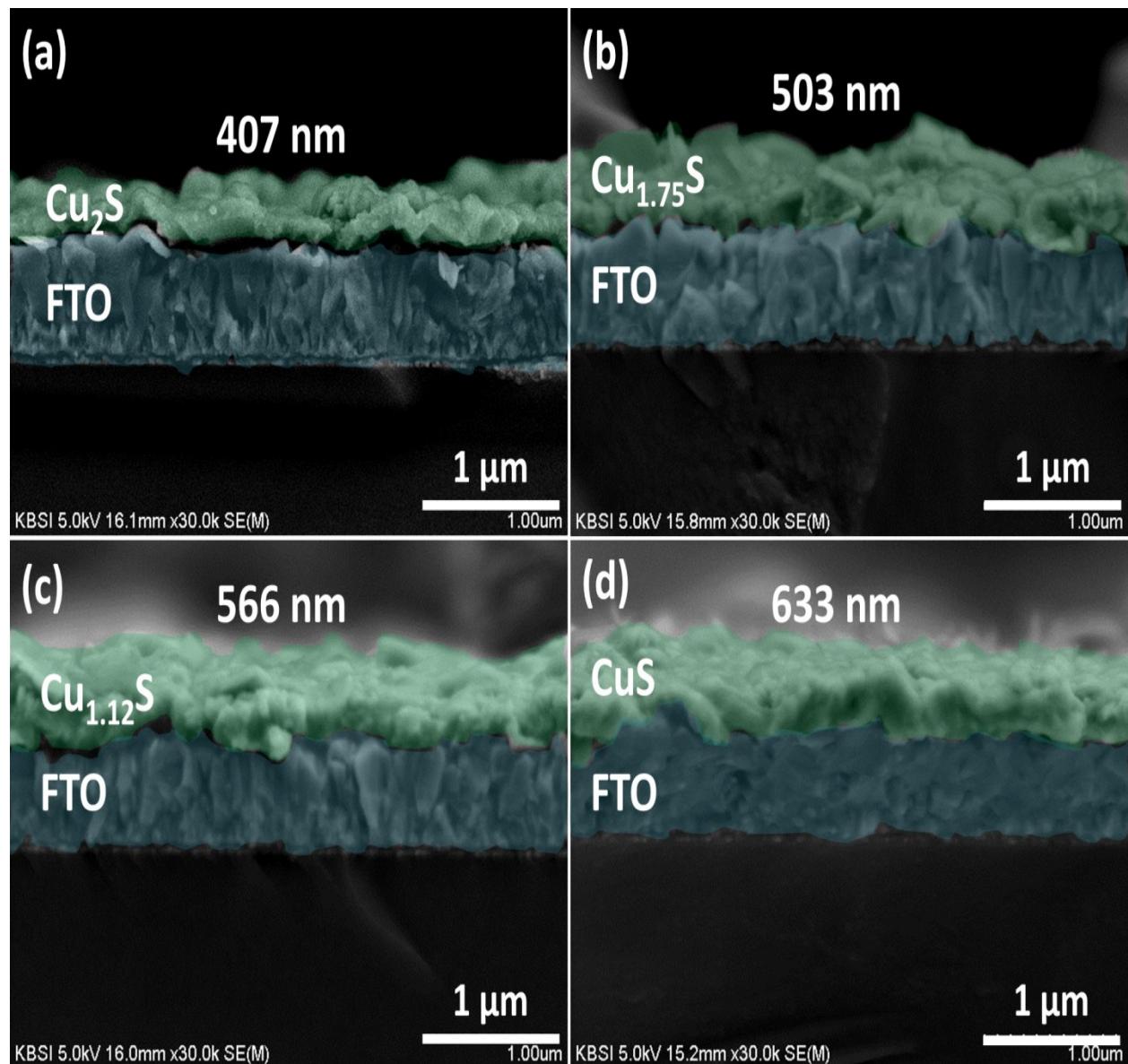


Fig. S1 Cross-sectional SEM images of (a) Cu_2S , (b) $\text{Cu}_{1.75}\text{S}$, (c) $\text{Cu}_{1.12}\text{S}$, and (d) CuS thin film on the surface of FTO.

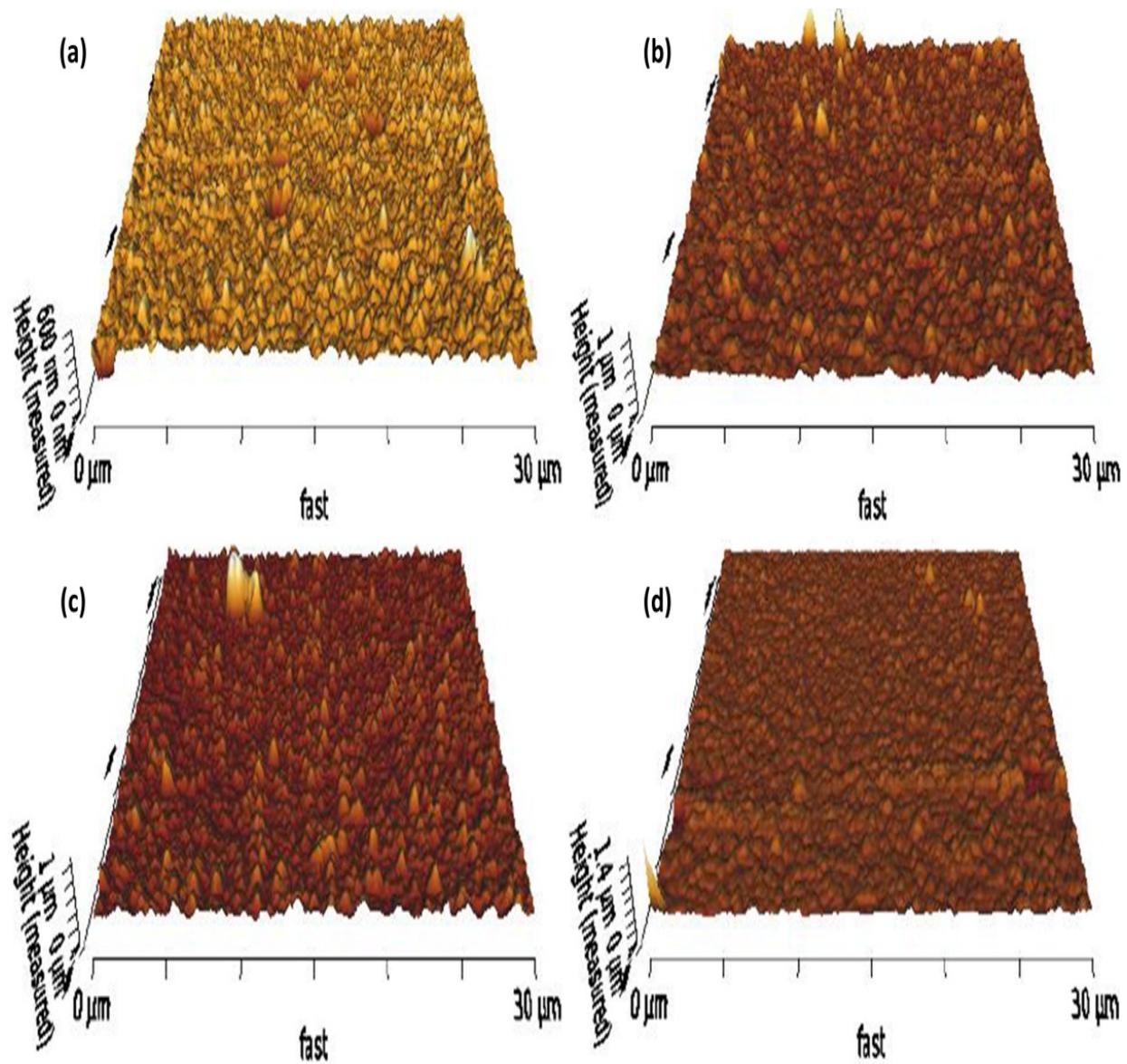


Fig. S2 3-Dimensional (3D) AFM images of (a) 0.005 mM, (b) 0.01 mM, (c) 0.02 mM, and (d) 0.03 mM L-cysteine based Cu_xS thin films on FTO substrate (Cu_2S , $\text{Cu}_{1.75}\text{S}$, $\text{Cu}_{1.12}\text{S}$, and CuS).

Table S2. Solar cell parameters of QDSSC (multiple cells) with various Cu_xS and Pt CEs

Counter Electrode	V _{oc} (V)	J _{SC} (mA cm ⁻²)	FF	η (%)
Cu₂S (Cell 1)	0.642	17.31	0.528	5.88
Cu₂S (Cell 2)	0.645	17.32	0.527	5.89
Cu₂S (Cell 3)	0.640	17.33	0.531	5.90
Cu_{1.75}S (Cell 1)	0.640	15.86	0.523	5.32
Cu_{1.75}S (Cell 2)	0.642	15.85	0.522	5.33
Cu_{1.75}S (Cell 3)	0.644	15.84	0.521	5.32
Cu_{1.12}S (Cell 1)	0.640	15.29	0.521	5.10
Cu_{1.12}S (Cell 2)	0.646	15.26	0.518	5.11
Cu_{1.12}S (Cell 3)	0.637	15.23	0.524	5.09
CuS (Cell 1)	0.643	14.46	0.525	4.89
CuS (Cell 2)	0.642	14.44	0.527	4.90
CuS (Cell 3)	0.642	14.45	0.527	4.90
Pt (Cell 1)	0.514	9.71	0.265	1.32
Pt (Cell 2)	0.517	9.98	0.263	1.36
Pt (Cell 3)	0.511	10.03	0.264	1.35

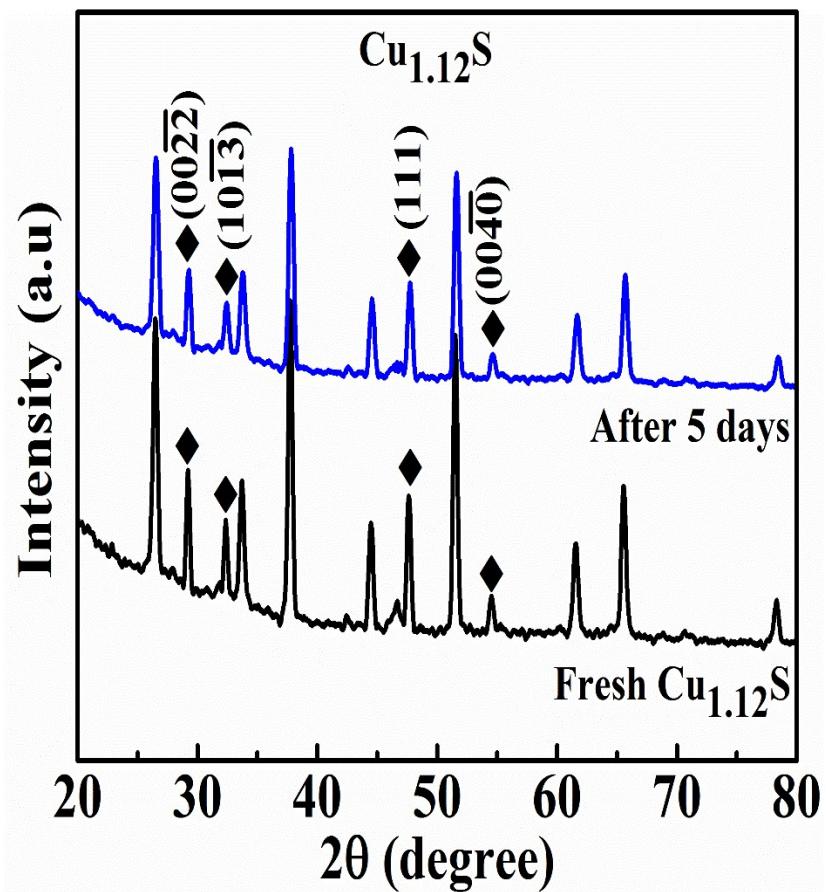


Fig. S3 XRD spectra of the as-prepared $\text{Cu}_{1.12}\text{S}$ CE, the $\text{Cu}_{1.12}\text{S}$ CE after 5 days of immersion in the polysulfide electrolyte.