## **Supporting Information**

## Green synthesis of Si-GQD nanocomposites as cost-effective catalysts for oxygen reduction reaction

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- Figure S1. (A) TEM and (B) HRTEM images of GQDs.















**Figure S5.** (A) CV curves of a Pt/C electrode in (a)  $N_2$ - and (b)  $O_2$ -saturated 0.1 M KOH at a scan rate of 20 mV s<sup>-1</sup>. Current densities were normalized with respect to the geometric area (0.196 cm<sup>2</sup>) of the RDE.

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Figure S6. CV plots of the Si-GQD<sub>1.0</sub> NC electrode (Inset: Pt/C) in 0.1 M KOH in the absence
and presence of 1 M MeOH. Scan rate: 20 mV s<sup>-1</sup>.











84	Elements	Before
85		Pyrolysis
86		(wt%)
87	Si	81.2
88	С	12.2
89	Al	2.3
90	Fe	1.8
91	Na	0.7
92	Ca	0.4
93		
94	К	0.5
95	Mg	0.3
96	Р	0.6
97		

83 Table S1. Elemental composition of rice husk before pyrolysis at 700 °C for 2 h

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Carbon nanomaterials	<sup>a</sup> E <sub>onset</sub>	<sup>b</sup> Peak	Stability	Ref.
	(V)	potential		
		(V)		
N-GQD/graphene	-0.16	-0.27	48 h	1
RN-GQDs-35/graphene	-0.19	-0.31	1000 cycles	2
N-doped colloidal GQDs	-0.1	<sup>c</sup> -0.3	<sup>d</sup> NA	3
N-doped carbon nanodots	-0.15	-0.35	1.9 h	4
N-doped mesoporous	-0.13	-0.25	5.6 h	5
graphitic arrays				
g-C <sub>3</sub> N <sub>4</sub> @CMK	-0.25	-0.3	45 h	6
Si-GQD <sub>1.0</sub> NCs	-0.18	-0.33	8.3 h, 2000 cycles	This work

## 100 Table S2. Comparison of ORR activity of carbon nanomaterials in alkaline media

101 <sup>a,b</sup> E vs. Ag/AgCl electrode, <sup>c</sup> E vs. SCE, <sup>d</sup>NA- not available

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