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Electronic Supplementary Information (ESI)

Cross Metathesis of Methyl Oleate (MO) with Terminal, Internal Olefins by Ruthenium Catalysts: Factors Affecting the Efficient MO Conversion and the Selectivity

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Contents

(i) General procedure and analysis data for their identifications and metathesis experiment

(ii) Calculation: Yield, conversion and selectivity

(iii) GC-FID and GC-MS chromatogram

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(i) General procedure and analysis data for their identifications and metathesis experiment.

1-1. General procedure. All experiments were carried out under a nitrogen atmosphere in a Vacuum Atmospheres drybox unless otherwise specified. All chemicals used were of reagent grade and were purified by the standard purification procedures. Anhydrous grade dichloromethane (DCM) and n-hexane (Kanto Chemical Co., Inc.) were transferred into a bottle containing molecular sieves (mixture of 3A 1/16, 4A 1/8 and 13X 1/16) in the drybox. Methyl oleate (\geq 99%), *cis*-stilbene, β -pinene *cis*-4-octene, cis-1,4-diacetoxy-2-butene, vanillideneacetone, allyl trimethylsilane and allyl glycidal ether were used in the drybox as received (Aldrich Chemical Co. or Tokyo Chemical Industry Co., lLtd.) without purification. Grubbs catalyst 1st generation RuCl₂(PCy₃)₂(CHPh) (1), 2nd generation RuCl₂(PCy₃)(H₂IMes)(CHPh) (2) and Hoveyda-Grubbs catalyst 2nd generation RuCl₂(IMesH₂)(CH-2-O^{*i*}Pr-C₆H₄) (3) were drybox used in the as received $[Cy = cyclohexyl, IMesH_2]$ 1,3-bis(2,4,6-trimethylphenyl)imidazolin-2-ylidene] (Aldrich Chemical Co.). Nonane and dodecane purchased from Tokyo Chemical Industry Co., Ltd were used as the internal standard (IS) for GC analyses.

The gas chromatograph mass spectrometer (GC/MS) analyses were performed on Shimadzu GC-17A gas chromatograph directly coupled to the mass spectrometer system (MS) of Shimadzu GCMS QP5050. Agilent column model DB-1 30m length, 0.25mm in diameter and film thickness and polyethylene glycol stationary phase was used throughout the experiment. Helium was used as the carrier gas at a flow rate of 1.7 mL/min. The oven temperature was programmed from 50°C for 10 min at a rate of 15°C/min with a final hold 200°C for 55 min (the total analysis time was 75 min).

Shimadzu GC-2025 gas chromatograph with FID detector (GC-FID) was employed for quantitative analysis of the starting materials and products using a (0.25 μ m x 0.25 mm x 30m) DB-1 column. Nitrogen gas was used as the carrier gas at a flow rate of 2.0 mL/min. The oven temperature was programmed from 50°C for 10 min at a rate of 15°C/min with a final hold 200°C for 45 min (the total analysis time was 65 min). The quantitative analyses were performed by comparing the peak area of the products with known amount of nonane or dodecane as an internal standard. Calibration coefficient was determined by analyzing the mixtures of MO and the internal standard with different ratios. The amount of MO was calculated by normalization using the internal standard method with the calibration coefficient. The conversion of MO was obtained from the comparison of peak areas before and after reactions.

1-2. The effective carbon number (ECN) of materials.

The effective carbon number (ECN) concept for GC-FID analyses was used to calculate the yields of products.¹ The ECN contributions of some functional groups (CH₂SiMe₃, $C_6H_3(3-OCH_3)(4-OH)$) were calculated from the GC-FID data of the related compounds containing similar moiety (allyltrimethylsilane (ATMS), vanillylidenacetone (VA)). Nonane was used as the internal standard. Contributions to the effective carbon number were summarized in Table S1. ECNs of starting materials and products for both self- and cross-metathesis reactions were shown in Figure S1-S4.

type	atom	ECN contribution ^{<i>a</i>}
Aliphatic	С	1
Aromatic	С	1
Olefinic	С	0.95
Carbonyl	С	0
Ester group	-	-1.49
Ether	О	-1.0
-CH ₂ SiMe ₃	-	4.01^{b}
HO OCH ₃	-	5.2 ^c

Table S1. ECN contributions of atoms or groups.

^{*a*}Ref X. ^{*b*}Calculated from GC-FID response of allyltrimethylsilane (ATMS) relative to nonane. ^{*c*}Calculated from GC-FID response of vanillylidenacetone (VA) relative to nonane.



Figure S1. ECN of the starting material and products for self metathesis of methyl oleate (MO).



Figure S2. ECN of the starting material and products for cross-metathesis of methyl oleate (MO) with symmetric internal olefins.



Figure S3. ECN of the starting material and products for cross-metathesis of methyl oleate (MO) with terminal olefins.



Figure S4. ECN of the starting material and products for cross-metathesis of methyl oleate (MO) with an asymmetric internal olefin.

(ii) Calculation of yield, conversion and selectivity.

2-1. Self-metathesis reaction of methyl oleate (MO).

Typical procedure (run 2, Table 2, S2) is as follows. Methyl oleate (MO, 0.593 g, 2.00 mmol) and dodecane (0.100 g) as an internal standard for GC analyses were dissolved in 1.0 mL of CH_2Cl_2 (DCM). Grubbs catalyst 2nd generation RuCl₂(PCy₃)(H₂IMes)(CHPh) (2) (0.0085 g, 0.5 mol%, 0.010 mmol) was added into the solution. The mixture was stirred at 50 °C for 1 hour, and then the resulting mixture was filtered through a Celite pad. The obtained filtrate was analyzed using the gas chromatograph with FID detector (GC-FID) and the gas chromatograph mass spectrometer (GC/MS). The results of self-metathesis reactions are summarized in Table 2, S2. GC-FID and GC/MS spectra are shown in Figure S5-S6.

2-2. Cross-metathesis reaction of methyl oleate (MO) with olefins.

Typical procedure (run 11, Table 2, S3) is as follows. Methyl oleate (MO, 0.593 g, 2.00 mmol), *cis*-4-octene (CO, 0.224g, 2.00 mmol) and nonane (0.100 g) as an internal standard for GC analyses were dissolved in 1.0 mL of CH_2Cl_2 (DCM). Grubbs catalyst 2nd generation $RuCl_2(PCy_3)(H_2IMes)(CHPh)$ (2) (0.0085 g, 0.5 mol%, 0.010 mmol) was added into the solution. The mixture was stirred at 50 °C for 1 hour, and then the resulting mixture was filtered through a Celite pad. The obtained filtrate was analyzed using the gas chromatograph with FID detector (GC-FID) and the gas chromatograph mass spectrometer (GC/MS). The results of cross-metathesis reactions are summarized in Table 2-3, S3-S9. GC-FID and GC/MS spectra are shown in Figure S7-S20.

Table S2. Self metathesis of methyl oleate (MO) by RuCl₂(PCy₃)₂(CHPh) (1) or RuCl₂(PCy₃)(H₂IMes)(CHPh) (2).^{*a*}

MO

	$\left\langle \right\rangle$	SM1		+	Meo	$\left\langle \right\rangle$	SM2		OMe	+		SM3	OMe
run ^a	catalyst	solvent	temp	Time	Conv.	SM1	SM1	SM2	SM2	SM3	SM3	Select. (1)	Select. (2)
	(mol%)		(°C)	(h)	of MO	trans	cis	trans	cis	trans	cis	p(%)	(%) (%)
					$(\%)^{p}$	$(\%)^c$	(%) ^c	(%) (%)	(%) ^c	$(\%)^c$	$(0/0)^{c}$		
1	2 (0.1)	CH_2Cl_2	50	1	62	23	9	15	4	29	8	LL	56
2	2 (0.5)	CH_2Cl_2	50	1	75	18	5	10	б	19	5	48	60
С	2 (1.0)	CH_2Cl_2	50	1	LL	17	4	8	б	18	5	42	58
4	1 (0.5)	CH_2Cl_2	50	1	99	7	2	6	б	20	15	32	38
5	2 (0.5)	CH_2Cl_2	50	0.33	80	25	7	Г	б	16	5	53	67
9	2 (0.5)	CH_2Cl_2	50	24	65	19	4	14	4	28	8	64	53
L	2 (0.5)	hexane	50	0.33	62	25	٢	L	ю	16	5	53	67
8	2 (0.1)	CH_2Cl_2	23	1	55	24	9	19	5	36	6	98	55
6	2 (0.5)	CH_2Cl_2	23	1	76	17	5	11	4	19	5	49	61
^a Conditi effect ((SM1+S	ions: methyl c of carbon nu iM2)/(SM1+S	oleate 2.00 m umber (ECN M2+SM3).	mol, solve 1) rule. SM1: octa	nt 1.0 mL. ^d Selectivity dec-9-ene, S	^b Conversion of SM1,2 M2: dimethy	of MO e based o	stimated b in the cc 9-enedioat	oy GC usin inversion te, SM3: me	g internation of MO. ethyl octad	onal stand ^e Selectiv ec-9-enoal	ard. ^c GC ity of se te.	yield estimated olf metathesis p	according to the roducts $(\%) =$

~ ~ ~		OMe	Select. (2)	$(\%)^{g}$		80	80	62	67	66	96	80	[Ru(1)], Cy = SH_2 $[Ru(3)].$
			Select. (1)	∫(%)		18	78	61	>99	>99	64	76	(CHPh)(PCy ₃) ₂ O [†] Pr-C ₆ H ₄)(IMe
	SM1	SM2 + SM2 + SM3	SM3	cis	$(\%)^e$	75	3	3	1	7	14	3). ^c RuCl ₂ iCl ₂ (CH-2-
			SM3	trans	$(\%)^e$	8	17	16	9	4	9	16	id on MC lene], Rı
	\langle		SM2	cis	$(\%)^{\ell}$	0	2	-	0	0	0	1	C. ^b Base din-2-ylid
		+ +	SM2	trans	$(0)^e$	1	8	٢	б	1	7	8	ure 50 °C /l)imidazo
•	H ₂ Cl ₂		SM1	cis	$(0)^e$	0	2	1	0	0	0	7	temperat thylpheny
,	ت م	CM2	SM1	trans	$(\%)^e$	0	6	8	б	1	7	10	10 mmol, ,4,6-trime
	8		CM2	$(\%)^{e}$		2	41	31	86	90	50	40	t (2) 0.0] 1,3-bis(2
`	Me +	+	CM1	$(\%)^e$		2	43	32	88	93	44	43	ıL, catalys MesH ₂ =
(\mathbf{H}_4) (3). ^{<i>a</i>}			Conv.	of MO $(\%)^d$		17	81	82	93	94	80	82	I, CH ₂ Cl ₂ 1.0 n [Ph] [Ru(2), I]
-O ⁱ Pr-C ₆	OM		Ru	$\operatorname{cat.}^c$		1	7	3	7	7	7	7	2.00 mmol 1esH ₂)(CH
[2)(CH-2	\langle	>	Time	(h)		17	1	1	1	1	1	8	/l oleate 2 (PCy3)(IN
2(IMesH	$\left\langle \right\rangle$		CO	$(eq.)^b$		1.0	1.0	1.0	5.0	10	10	1.0	ons: methy yl, RuCl ₂
or RuCl			run ^a			10	11	12	13	14	15^{h}	21	^a Conditic cyclohex

Table S3. Cross metathesis of methyl oleate (MO) with *cis*-4-octene (CO) by RuCl₂(PCy₃)₂(CHPh) (1), RuCl₂(PCy₃)(H₂IMes)(CHPh) (2)

S8

version of MO estimated by GC using international standard. "GC yield estimated according to the effect of carbon number (ECN) rule. Selectivity of CM1.2 SM1.2 based on the conversion of MO. "Selectivity of cross metablasis products (SM3: methyl octadec-9-enoate. SM2: dimethyl octadec-9-enoate. SM3: methyl octadec-9-enoate. SM3: meth	7 5.0 1 97 88 76 2 0 3 1 2 1 91 96	ditions: methyl oleate 2.00 mmol, CH_2Cl_2 1.0 mL, catalyst (2) 0.010 mmol, temperature 50 °C. ^b Based on MO. ^c Conversion of MO estimated by GC using national standard. ^d GC yield estimated according to the effect of carbon number (ECN) rule. ^e Selectivity of CM1,2 and SM1,2 based on the conversion of MO. ^c Civity of cross metathesis products (%) = (CM1+CM2)/(CM1+CM2+SM1+SM2). CM1: dec-1-en-1-ylbenzene, CM2: methyl 10-phenyldec-9-enoate, SM1:
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S9

	$\left\langle \right\rangle$	$\left\langle \right\rangle$	No Market	$\left\langle \right\rangle$	+ OMe		B	CH2		<	<		<		
		\langle	CM1				CM2		₹ +			SM3 + SM1 + SM3 SM3 SM3 SM3			
run ^a	DAB	Time	Conv.	CM1	CM1	CM2	CM2	SM1	SM1	SM2	SM2	SM3	SM3	Select.	Select.
	$(eq.)^b$	(h)	of MO	trans	cis	trans	cis	trans	cis	trans	cis	trans	cis	(1)	(2)
			(%) (%)	_p (%)	_p (%)	_p (%)	$_{p}(\%)$	_p (%)	p(%)	_p (%)	_p (%)	_p (%)	_p (%)	$(\%)^e$	∫ (%)
18	1.0	1	87	59	6	56	5	5	1	87	65	11	2	86	91
19	5.0	1	80	19	8	22	2	1	1	80	27	13	9	41	88
20	10	1	67	4	2	4	2	0	0	67	9	1	33	10	92
22	1.0	б	06	48	9	44	ю	7	1	5	1	10	7	72	88
¹ Conditic nternatic	ons: methy onal standa:	$\frac{1}{1 \text{ oleate } 2.0}$	00 mmol, C yield estime	CH ₂ Cl ₂ 1.0 ated accord	mL, cataly ing to the e	yst (2) 0.0	10 mmol, t	emperature er (ECN) 1	a 50 °C. Tule. ^e Sel	^b Based on lectivity of	MO. ^c Cc CM1,2 and	d SM1,2 b	of MO esti- ased on the	mated by C conversion	GC usir n of MC

\rangle		MO	$\left\langle \right\rangle$	OMe	+ AT		CH ₂ Cl ₂			<	< <		<	\	
	$\left\langle \right\rangle$	+ CM		+	//				+ WeO			SM2		OMe	
	\rangle	CM3	Silv	Je ³	e ₃ Si	CM4	$\left\langle \right\rangle$	OMe		$\left\langle \right\rangle$		+ +		OMe	
un ^a ATM	S Conv.	CMI	CM2	CM3	CM3	CM4	CM4	SM1	SM1	SM2	SM2	SM3	SM3	Select.	Select
(eq.)	b of MO	_p (%)	_p (%)	trans	cis	trans	cis	trans	cis	trans	cis	trans	cis	(1)	(2)
	(%) (%)			p(%)	p(%)	_p (%)	_p (%)	_p (%)	_p (%)	_p (%)	_p (%)	p(0/0)	$_{p}(0\%)$	$(\%)^{\ell}$	(%) (%)
23 1.0	80	10	10	21	L	19	7	6	2	8	2	17.0	3.2	71.5	78.7
24 5.0	94	36	39	37	13	36	13	7	1	ю	1	4.9	1.3	>99	96.4
25 10	96	46	47	35	13	33	12	1	0	-	0	2.9	1.0	>99	98.4
6 ^g 10	96	27	37	22	8	25	6	1	1	2	0	7	1	71	76

S11

				elect.	(2)	(%)	50	>99	66<	by GC ersion nethyl nethyl
		OMe	OMe	ct. S		в				imated l the conv CM2: r M2: dir
	\langle			Selee	(1)	(0)	16	4	4	MO est sed on t -1-ene, ene, SN
1Fn) (2).				SM3	cis	p(0)	47	36	50	version of SM1,2 ba SM1: dec- octadec-9-e
IMes)(CF			SM3 +	SM3	trans	p(0)	15	7	4	40. ^c Con MI-4 and SM2). C SM1: c
rcy3)(h21				SM2	cis	_p (%)	1	0	0	^b Based on N ectivity of C CM4+SM1+ sc-9-enoate,
y Kuul2(Meo Meo	> \	SM2	trans	$p(0)^{q}$	1	0	0	time 1 h. rule. ^e Sel <i>A</i> 2+CM3+0 thoxy)unde
(AUE) D	t a	+ o Me	OMe	SM1	cis	$_{p}(\%)$	1	0	0	ure 50 °C, 1 ber (ECN) /(CM1+CN) ran-2-ylme
idal ether	CH ₂ Cl ₂	$\left\langle \right\rangle$		SM1	trans	_p (%)	1	0	0	l, temperati arbon numl M3+CM4) yl 11-(oxi
allyl glyc	AGE	+	CM4	CM4	$_{p}(\%)$		1	1	1	0.010 mmo e effect of c M1+CM2+C M4: methy
	• +		0	CM3	p(0)		1	1	۸	catalyst (2) ording to th (%) = (C) oxirane, C
l oleate (l		+	Do	CM2	p(0)		1	1	1	J ₂ 1.0 mL, imated acco products xy)methyl) -enoate.
or memy				CM1	p(0)		1	1	1	mol, CH ₂ C C yield est metathesis 2-en-1-ylo l octadec-9
letathesis (OW	+ CIN1	CM3	Conv.	of MO	(%) (%)	38	55	46	leate 2.00 m mdard. ^d G ^r of cross 2-((undec- SM3: methy
/. Uross n		\langle	\langle	AGE	$(eq.)^b$		1.0	5.0	10	Is: methyl o national sta /Selectivity tte, CM3: enedioate, S
I able S	·			run ^a			27	28	29	^a Conditior using inter of MO. dec-9-enos octadec-9-

Table S7 Cross metathesis of methyl oleate (MO) with allyl olycidal ether (AGE) by RuCl-(PCv,)(H,IMec)(CHPh) (2)^a

$\begin{array}{c ccccc} & & & & & & & & & & & & & & & & &$		eo	0 SM2 + +	SM3	SM2 SM2 SM3 SM3 Select. Select.	rans cis trans cis (1) (2)	$(96)^d$ $(96)^d$ $(96)^d$ $(96)^d$ $(96)^d$ $(96)^e$ $(96)^f$	14 4 17 4 55 9	13 4 15 4 53 17	a 1 h. ^b Based on MO. ^c Conversion of MO estimated by GC b. ^e Selectivity of CM1-4 and SM1,2 based on the conversion CM3+CM4+SM1+SM2). CM1: dec-1-ene, CM2: methyl cyclo[4.1.0]heptan-3-ylidene)nonanoate, SM1: octadec-9-ene,
a moduly occur (MO) MO MO MO MO MO MO MO MO	PN CH2Cl2	CM2 CM2	+	CM4	13 CM4 SM1 SM1	$)^d$ $(^{0/0}_{0})^d$ trans cis	$(0,0)^d$ $(0,0)^d$	1 19 3	1 17 4	st (2) 0.010 mmol, temperature 50 °C to the effect of carbon number (ECN = (CM1+CM2+CM3+CM4)/(CM1+C heptane, CM4: methyl 9-(7,7-dimeth
))))))))	// //////	+	~	CM1 CM2 CN	$(0,0)^d$ $(0,0)^d$ $(0,0)^d$		1 2 (5 2 (D mmol, CH ₂ Cl ₂ 1.0 mL, catal- ^d GC yield estimated according ss metathesis products (%) thyl-3-nonylidenebicyclo[4.1.0



Table S9. Cross metathesis of methyl oleate (MO) with vanillylidenacetone (VA) by RuCl₂(PCy₃)(H₂IMes)(CHPh) ($\mathbf{2}$).^{*a*}



(iii) GC-FID and GC-MS chromatograms.

3-1. Self-metathesis of methyl oleate (MO).

9-Octadecene (SM1, m/z 252) and 9-octadecene-1,18-dioate (SM2, m/z 341) appears at 22.3 (*cis*-SM1), 22.4 (*trans*-SM1) and 45.6 (*cis*-SM2), 46.2 (*trans*-SM2) min, respectively, while (MO, SM3, m/z 296) which is also the starting material methyl oleate appeared at 28.6 (*cis*), 28.8 (*trans*) min on GC chromatogram (Figure S5). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S6.



Figure S5. GC-FID chromatogram for the self-metathesis reaction of methyl oleate.



Figure S6. GC-MS chromatogram for the self-metathesis reaction of methyl oleate (A), and MS fragmentation patterns of the self-metathesis products (B).

3-2. Cross-metathesis of methyl oleate (MO) with cis-4-octene (CO).

Tridec-4-ene (CM1, m/z 182), methyl tridec-9-enoate (CM2, m/z 226), 9-octadecene (SM1, m/z 252) and 9-octadecene-1,18-dioate (SM2, m/z 341) appears at 17.4 (CM1), 20.4 (CM2), 22.3 (*cis*-SM1), 22.4 (*trans*-SM1), 45.6 (*cis*-SM2), and 46.2 (*trans*-SM2) min, respectively, while (MO, SM3, m/z 296) which is also the starting material methyl oleate appeared at 28.6 (*cis*), 28.8 (*trans*) min (Figure S7). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S8.



Figure S7. GC-FID chromatogram for the cross-metathesis reaction of methyl oleate (MO) with *cis*-4-octene (CO).





Figure S8. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (MO) with *cis*-4-octene (CO) (A), and MS fragmentation patterns of the cross-metathesis products (B).

3-3. Cross metathesis methyl oleate (MO) with *cis*-stilbene (CS).

Two targeted products dec-1-en-1-ylbenzene (CM1, m/z 216) and methyl 10-phenyldec-9-enoate (CM2, m/z 260) were observed at retention time of 21.7 (CM1) and 28.7 (CM2) min, respectively (Figure S9). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S10.



Figure S9. GC-FID chromatograms for the cross-metathesis reaction of methyl oleate (MO) and *cis*-stilbene (CS).



Figure S10. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (MO) and *cis*-stilbene (CS) (A), and MS fragmentation patterns of the cross-metathesis products (B).

3-4. Cross metathesis methyl oleate (MO) with cis-1,4-diacetoxy-2-butene (DAB).

Two targeted products undec-2-en-1-yl acetate (CM1, m/z 212) and methyl 11-acetoxyundec-9-enoate (CM2, m/z 256) were observed at retention time of 19.3 (*cis*-CM1), 19.4 (*trans*-CM1) and 22.5 (*cis*-CM2), 22.6 (*trans*-CM2) min, respectively (Figure S11). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S12.



Figure S11. GC-FID chromatogram for the cross-metathesis reaction of methyl oleate (MO) and *cis*-1,4-diacetoxy-2-butene (DAB).



Figure S12. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (**MO**) with *cis*-1,4-diacetoxy-2-butene (**DAB**) (A), and MS fragmentation patterns of the cross-metathesis products (B).

3-5. Cross metathesis of methyl oleate (MO) with Allyltrimethylsilane (ATMS).

Four targeted products dec-1-ene (CM1, m/z 140), methyl dec-9-enoate (CM2, m/z 184), trimethyl(undec-2-en-1-yl)silane (CM3, m/z 226), and methyl 11-(trimethylsilyl)undec-9-enoate (CM4, m/z 270) were observed at retention time of 12.3 (CM1), 17.8 (CM2), 18.9 (*cis*-CM3), 18.8 (*trans*-CM3), 21.8 (*cis*-CM4), and 21.9 (*trans*-CM4) min, respectively (Figure S13). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S14.



Figure S13. GC-FID chromatogram for the cross-metathesis reaction of methyl oleate (MO) and allyltrimethylsilane (ATMS).









Figure S14. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (MO) and allyltrimethylsilane (ATMS) (A), and MS fragmentation patterns of the cross-metathesis products (B).

3-6. Cross metathesis of methyl oleate (MO) with allyl glycidal ether (AGE).

Four targeted products dec-1-ene (CM1, m/z 140), methyl dec-9-enoate (CM2, m/z 184), 2-((undec-2-en-1-yloxy)methyl)oxirane (CM3, m/z 226), and methyl 11-(oxiran-2-ylmethoxy)undec-9-enoate (CM4, m/z 270) were observed at retention time of 12.7 (CM1), 17.7 (CM2), 21.6 (CM3), and 27.1 (CM4) min, respectively (Figure S15). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S16.



Figure S15. GC-FID chromatogram for the cross-metathesis reaction of methyl oleate (MO) and allyl glycidal ether (AGE).





Figure S16. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (**MO**) and allyl glycidal ether (**AGE**) (A), and MS fragmentation patterns of the cross-metathesis products (B).

(A)

3-7. Cross metathesis of methyl oleate (MO) with β-pinene (PN).

Four targeted products dec-1-ene (CM1, m/z 140), methyl dec-9-enoate (CM2, m/z 184), 7,7-dimethyl-3-nonylidenebicyclo[4.1.0]heptane (CM3, m/z 248), and methyl 9-(7,7-dimethylbicyclo[4.1.0]heptan-3-ylidene)nonanoate (CM4, m/z 292) were observed at retention time of 12.7 (CM1), 17.7 (CM2), 22.3 (CM3), and 30.3 (CM4) min, respectively (Figure S17). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S18.



Figure S17. GC-FID chromatogram for the cross-metathesis reaction of methyl oleate (MO) and β -pinene (PN).



Figure S18. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (MO) and β -pinene (PN) (A), and MS fragmentation patterns of the cross-metathesis products (B).

3-8. Cross metathesis of methyl oleate (MO) with vanillidenacetone (VA).

Four targeted products dodec-3-en-2-one (CM1, m/z 182), methyl 11-oxododec-9-enoate (CM2, m/z 226), 4-(dec-1-en-1-yl)-2-methoxyphenol (CM3 m/z 262), and methyl 10-(4-hydroxy-3-methoxyphenyl)dec-9-enoate (CM4, m/z 306) were observed at retention time of 18.9 (CM1), 22.1 (CM2), 32.1 (CM3), and 56.0 (CM4) min, respectively (Figure S19). GC–MS chromatogram and the mass fragmentation patterns of each peak are shown in Figure S20.



Figure S19. GC-FID chromatogram for the cross-metathesis reaction of methyl oleate (MO) and vanillidenacetone (VA).





(B)











Figure S20. GC-MS chromatogram for the cross-metathesis reaction of methyl oleate (**MO**) and vanillideneacetone (**VA**) (A), and MS fragmentation patterns of the cross-metathesis products (B).

References

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