

Supporting Information

The oxygen reduction reaction mechanism on Sn doped graphene as an electrocatalyst in fuel cells: A DFT study

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Tang,^{*c} and Zhijian Wu^{*a}

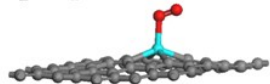
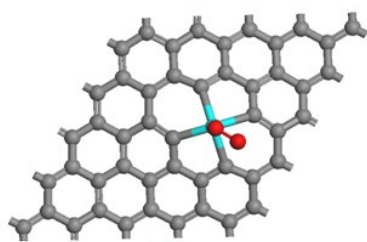
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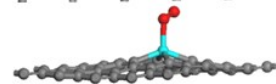
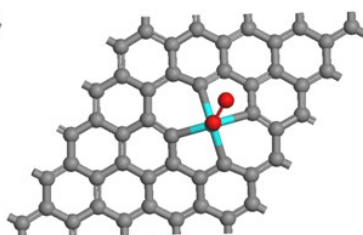
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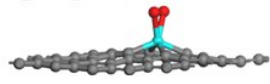
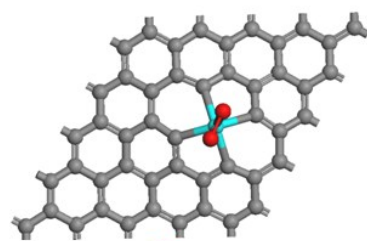
Optimized adsorption structure --- O₂



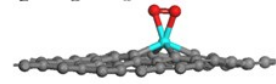
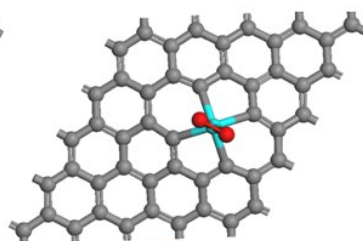
$E_{\text{ads}} = -0.40 \text{ eV}$



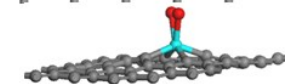
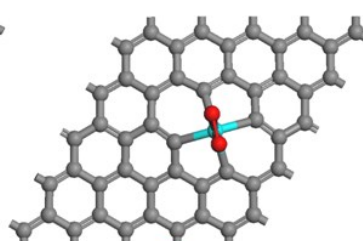
$E_{\text{ads}} = -0.36 \text{ eV}$



$E_{\text{ads}} = 0.15 \text{ eV}$

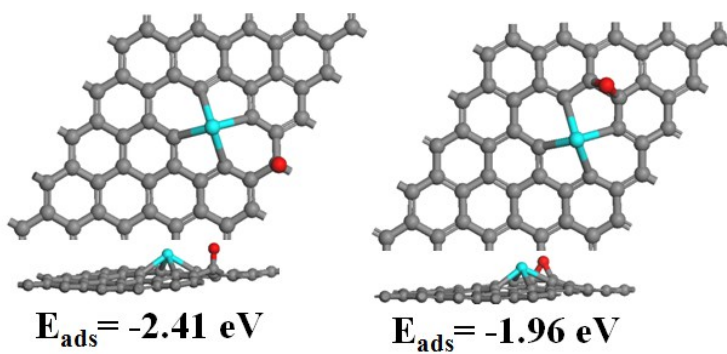
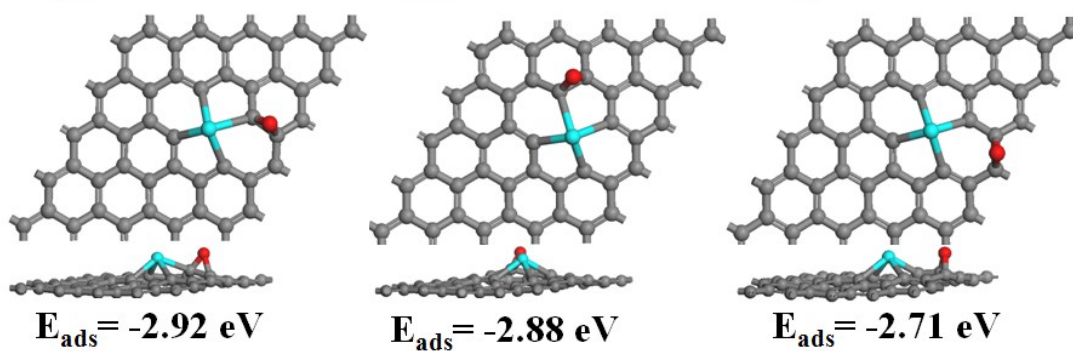
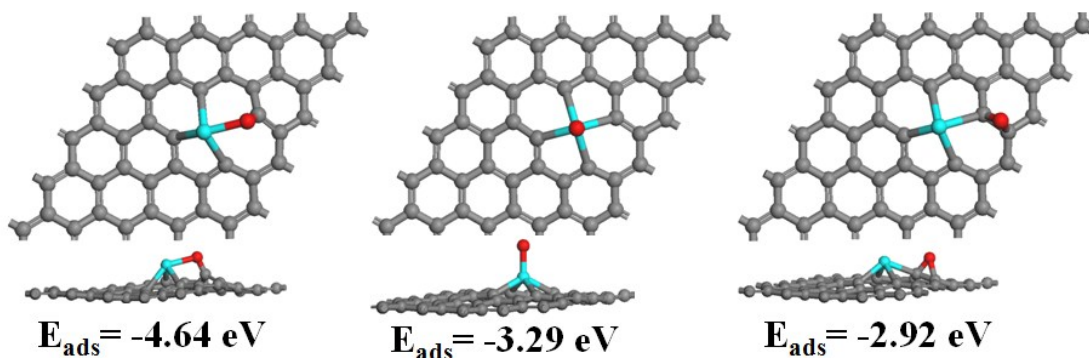


$E_{\text{ads}} = 0.15 \text{ eV}$

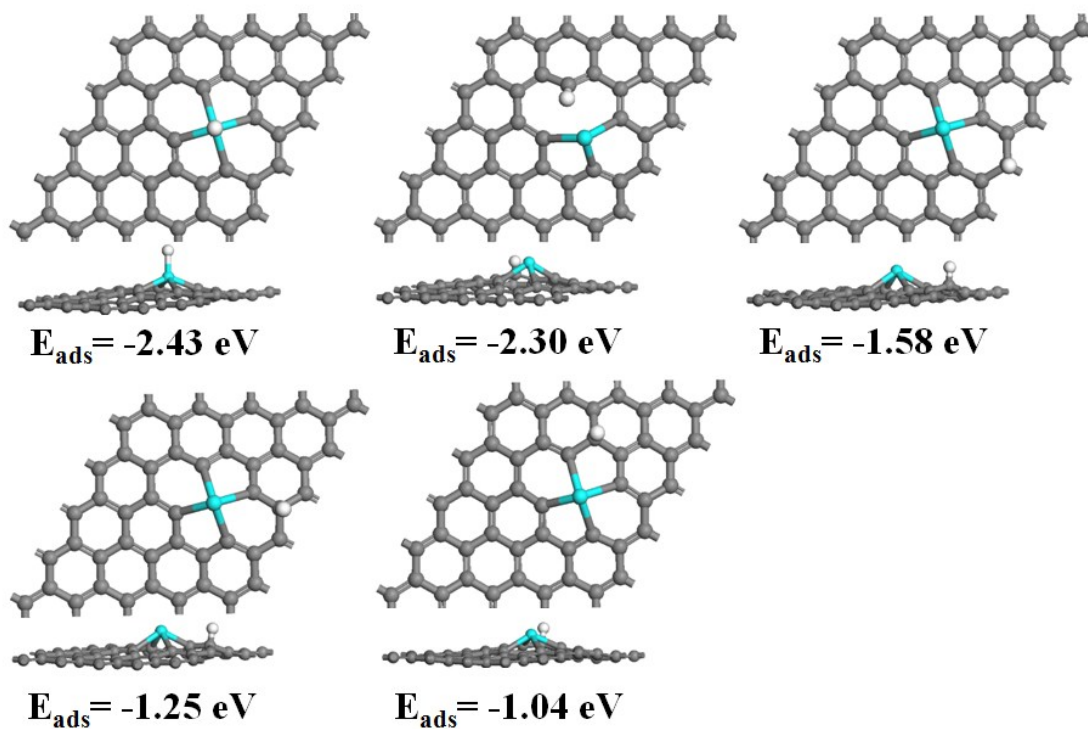


$E_{\text{ads}} = 0.16 \text{ eV}$

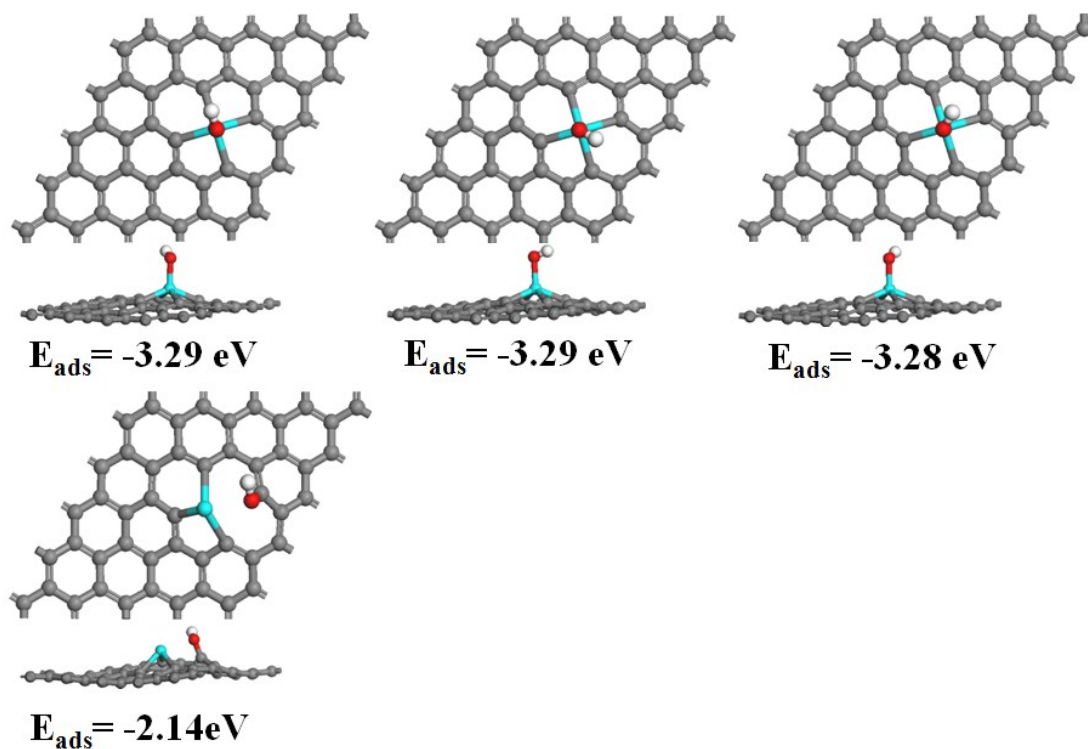
Optimized adsorption structure --- O



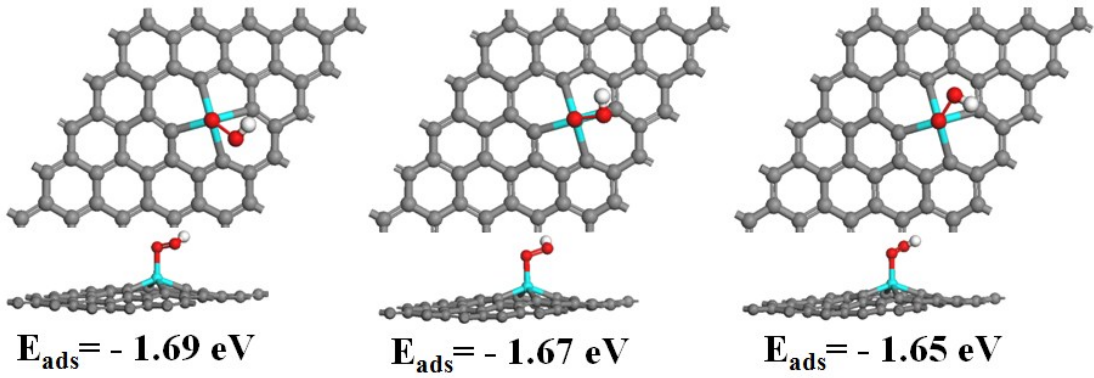
Optimized adsorption structure --- H



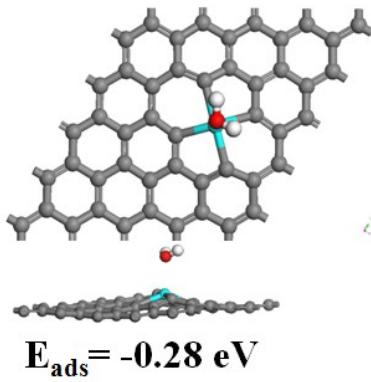
Optimized adsorption structure --- OH



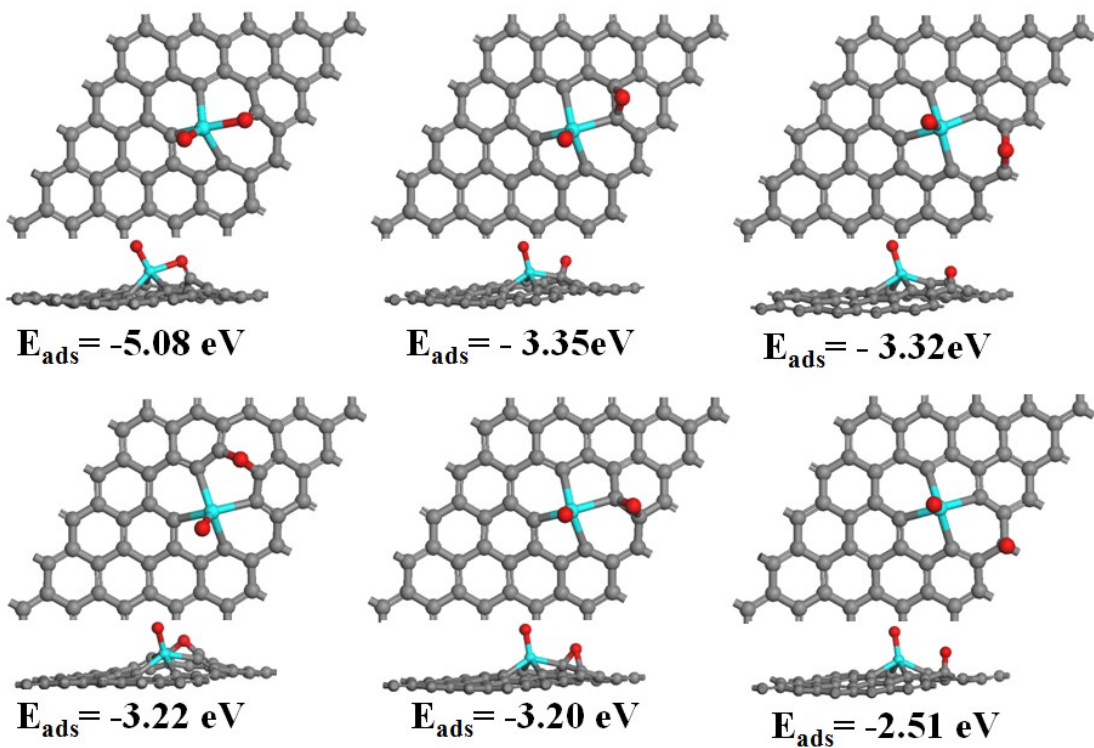
Optimized adsorption structure --- OOH



Optimized adsorption structure --- H₂O



Optimized adsorption structure --- O-O



Optimized adsorption structure --- O-OH

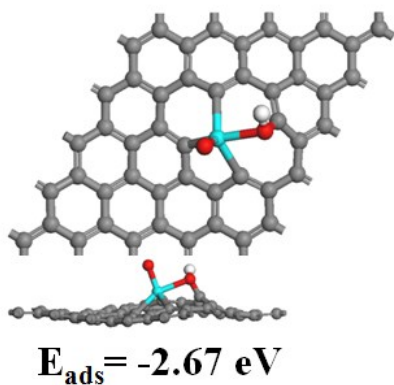


Fig. S1. Possible configurations for each adsorbed species (end-on O₂, side-on O₂, O, H, OH, OOH, H₂O, O-O and O-OH) involved in the ORR on Sn doped divacancy graphene. E_{ads} represents the adsorption energy (eV). The gray and blue balls represent C and Sn atoms, respectively.

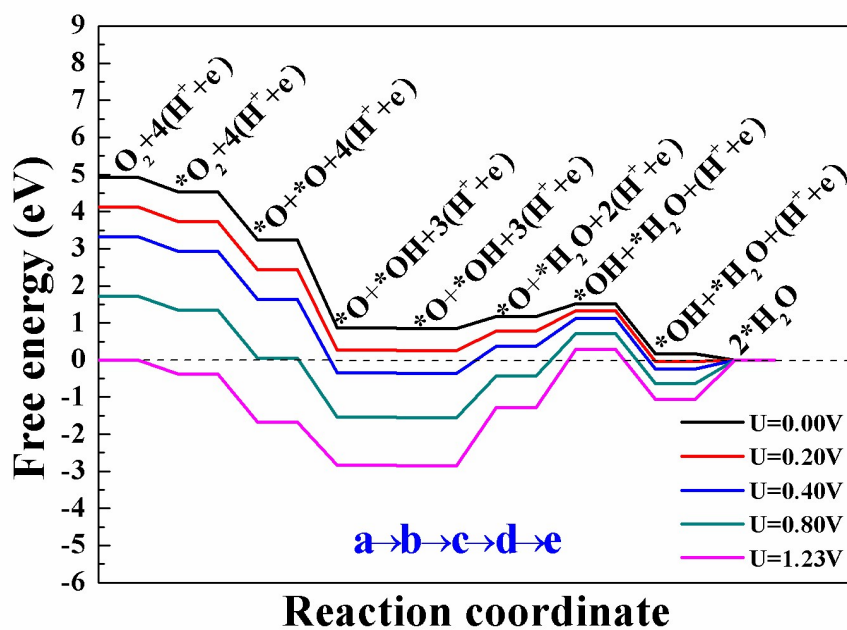


Fig. S2. The free energy diagram for the reaction pathway I ($a \rightarrow b \rightarrow c \rightarrow d \rightarrow e$) on Sn doped divacancy graphene at different electrode potentials.

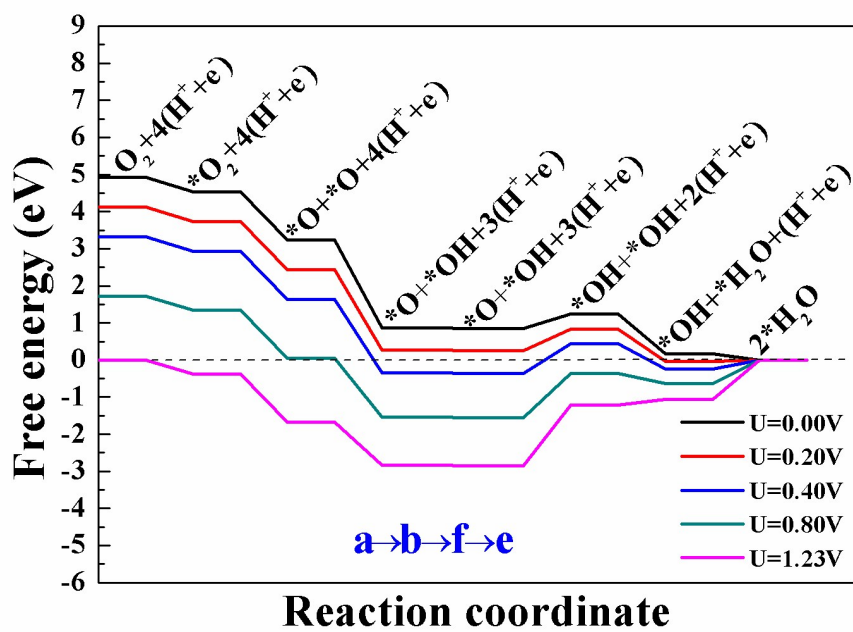


Fig. S3. The free energy diagram for the reaction pathway II ($a \rightarrow b \rightarrow f \rightarrow e$) on Sn doped divacancy graphene at different electrode potentials.