

Electronic Supplementary Information (ESI)
for

**Guest dependent structure and acetone sensing properties of a 2D
Eu³⁺ coordination polymer**

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Table S1. Crystal data and structure refinement parameters for **1**.

1	
chemical formula	C ₃₀ H ₂₂ EuNO ₉
formula weight	692.45
crystal system	Trigonal
space group	P -1
<i>a</i> /Å	9.5960(15)
<i>b</i> /Å	13.277(2)
<i>c</i> /Å	15.277(3)
α (°)	104.260(4)
β (°)	96.594(2)
γ (°)	97.594(3)
<i>V</i> /Å ³	1847.7(5)
<i>Z</i>	2
<i>D</i> /g cm ⁻³	1.245
<i>T</i> /K	113
μ /mm ⁻¹	1.740
<i>F</i> (000)	688
<i>R</i> ^a [<i>I</i> >2σ(<i>I</i>)]	0.0359
<i>wR</i> ^b [all data]	0.0999

^a $R = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|$; ^b $R_w = [\Sigma [w(F_o^2 - F_c^2)^2] / \Sigma w(F_o^2)^2]^{1/2}$.

Table S2. Bond lengths [Å] and angles [deg] for 1.

Eu(1)-O(4)#2	2.400(2)	O(4)#2-Eu(1)-O(5)	73.44(9)
Eu(1)-O(2)#3	2.409(2)	O(2)#3-Eu(1)-O(5)	140.81(9)
Eu(1)-O(8)	2.411(3)	O(8)-Eu(1)-O(5)	75.97(10)
Eu(1)-O(1)	2.424(2)	O(1)-Eu(1)-O(5)	77.53(8)
Eu(1)-O(7)	2.432(3)	O(7)-Eu(1)-O(5)	136.77(9)
Eu(1)-O(5)	2.485(2)	O(4)#2-Eu(1)-O(6)	126.15(8)
Eu(1)-O(6)	2.490(2)	O(2)#3-Eu(1)-O(6)	93.49(8)
Eu(1)-O(3)#2	2.532(3)	O(8)-Eu(1)-O(6)	74.96(10)
Eu(1)-O(1)#3	2.702(2)	O(1)-Eu(1)-O(6)	76.41(8)
O(1)-Eu(1)#3	2.702(2)	O(7)-Eu(1)-O(6)	141.96(9)
O(2)-Eu(1)#3	2.409(2)	O(5)-Eu(1)-O(6)	52.71(8)
O(3)-Eu(1)#2	2.532(3)	O(4)#2-Eu(1)-O(3)#2	52.89(8)
O(4)-Eu(1)#2	2.400(2)	O(2)#3-Eu(1)-O(3)#2	83.21(8)
O(4)#2-Eu(1)-O(2)#3	135.57(9)	O(8)-Eu(1)-O(3)#2	72.01(10)
O(4)#2-Eu(1)-O(8)	93.93(10)	O(1)-Eu(1)-O(3)#2	134.30(9)
O(2)#3-Eu(1)-O(8)	76.24(10)	O(7)-Eu(1)-O(3)#2	70.58(10)
O(4)#2-Eu(1)-O(1)	93.55(9)	O(5)-Eu(1)-O(3)#2	113.33(9)
O(2)#3-Eu(1)-O(1)	117.13(8)	O(6)-Eu(1)-O(3)#2	146.64(9)
O(8)-Eu(1)-O(1)	149.07(10)	O(4)#2-Eu(1)-O(1)#3	149.69(9)
O(4)#2-Eu(1)-O(7)	77.85(9)	O(2)#3-Eu(1)-O(1)#3	50.63(8)
O(2)#3-Eu(1)-O(7)	81.57(9)	O(8)-Eu(1)-O(1)#3	114.71(9)
O(8)-Eu(1)-O(7)	138.24(10)	O(1)-Eu(1)-O(1)#3	67.27(9)
O(1)-Eu(1)-O(7)	72.68(9)	O(7)-Eu(1)-O(1)#3	74.16(8)
O(5)-Eu(1)-O(1)#3	121.32(8)	O(3)#2-Eu(1)-O(1)#3	125.01(8)
O(6)-Eu(1)-O(1)#3	73.78(8)		

Symmetry codes: #1 -x+2,-y+1,-z; #2 -x+1,-y+2,-z; #3 -x+1,-y+1,-z-1

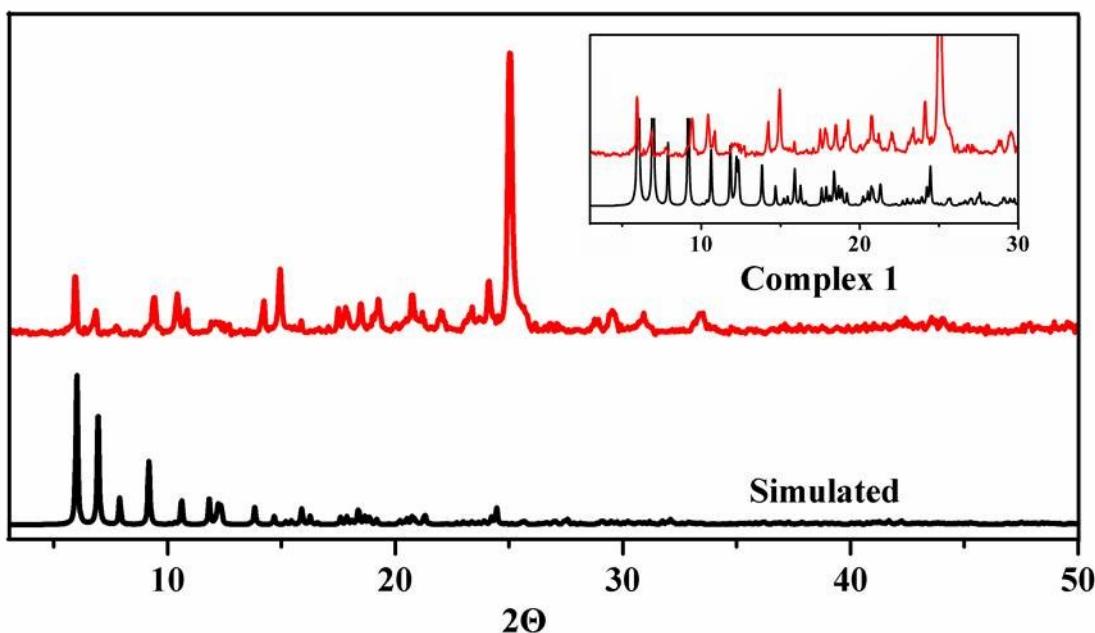


Fig. S1 Simulated and experimental PXRD patterns of complex **1**. The inset shows the detailed view of the patterns in the range of 3-30 degree.

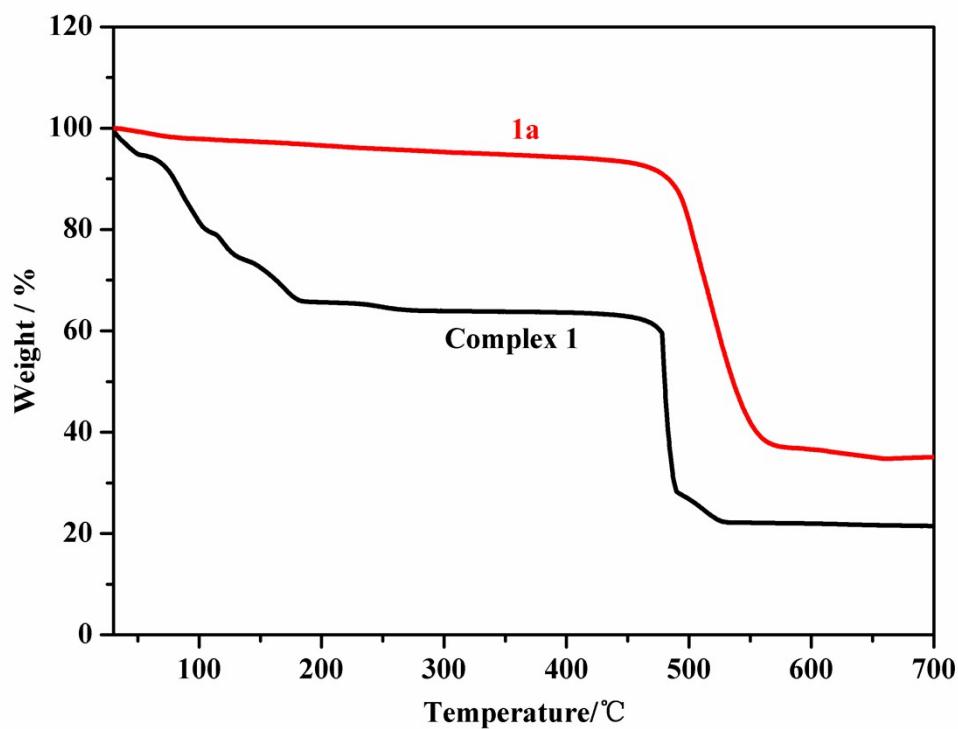


Fig. S2 Thermal gravity analysis (TGA) profiles of complex **1** and **1a**.

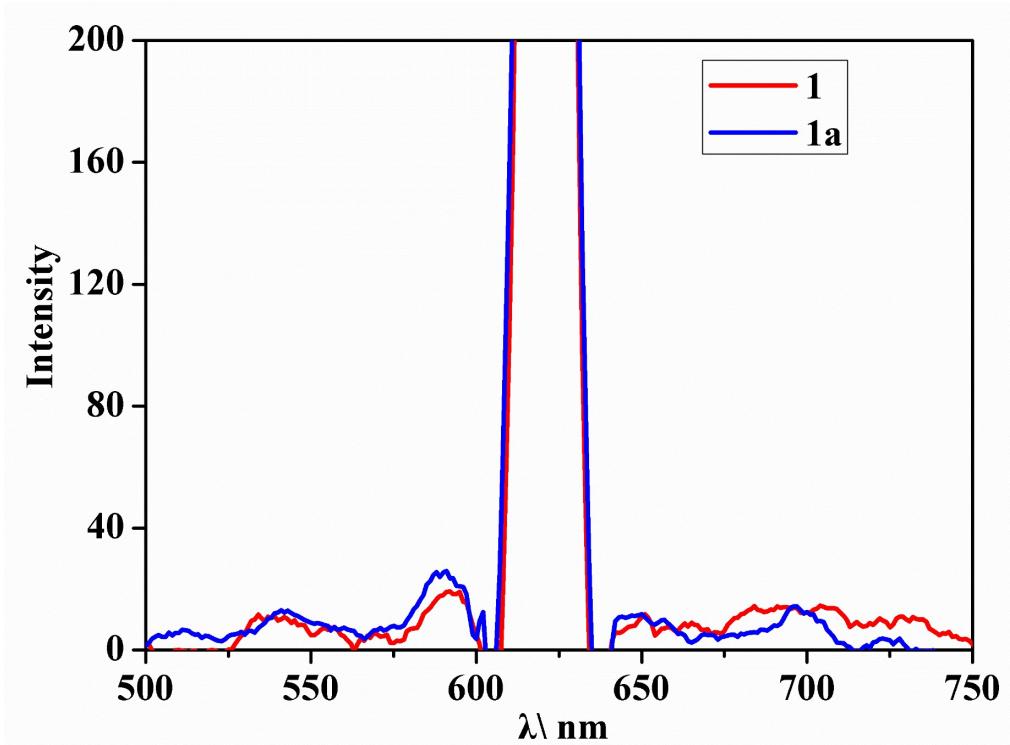


Fig. S3 Solid state fluorescent spectra of **1** and **1a** under excitation at 310 nm. The Y-axis is magnified to present the characteristic emission peak of Eu^{3+} ion with low intensities.

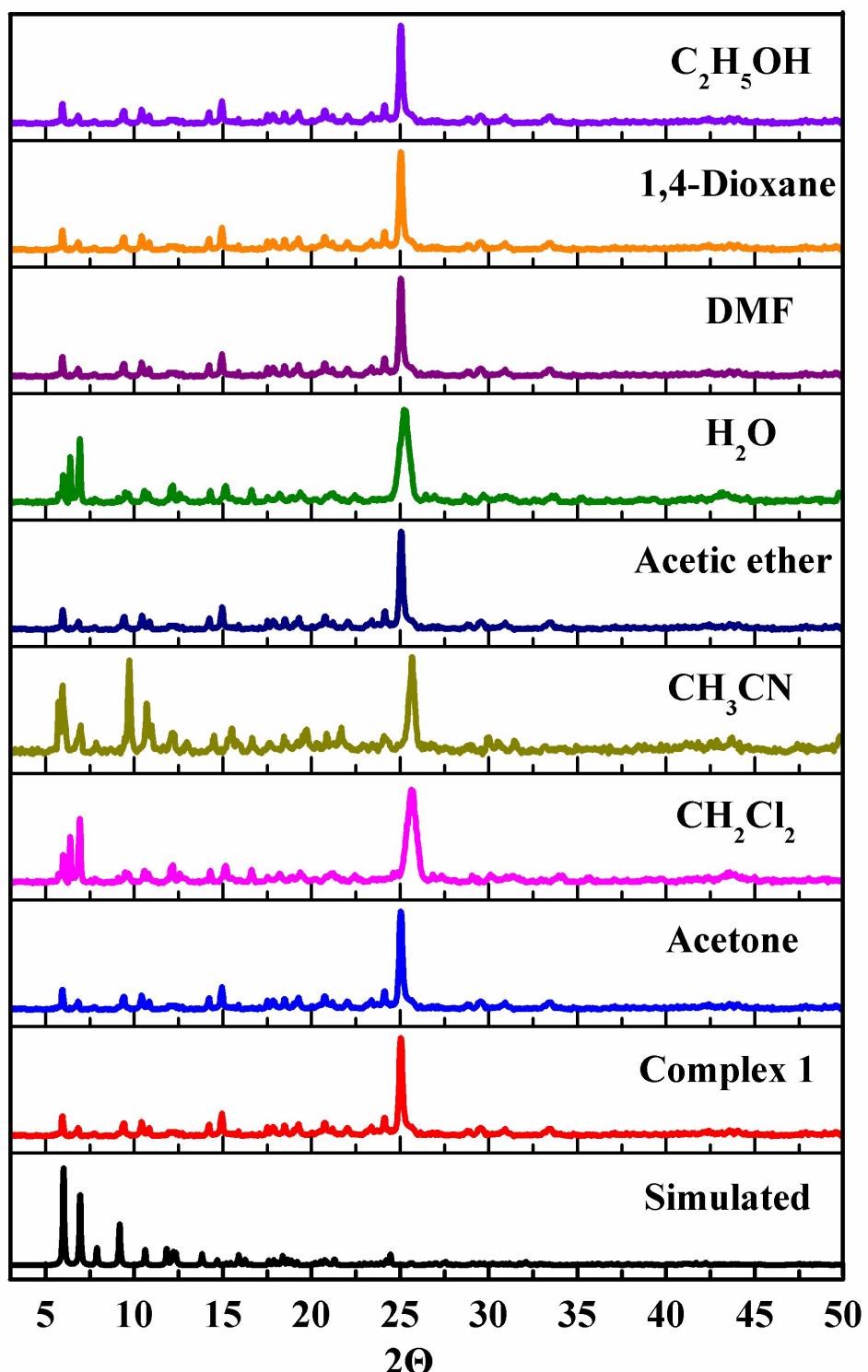


Fig. S4 Simulated and experimental PXRD patterns of **1** soaked in different solvents for 48 hours.

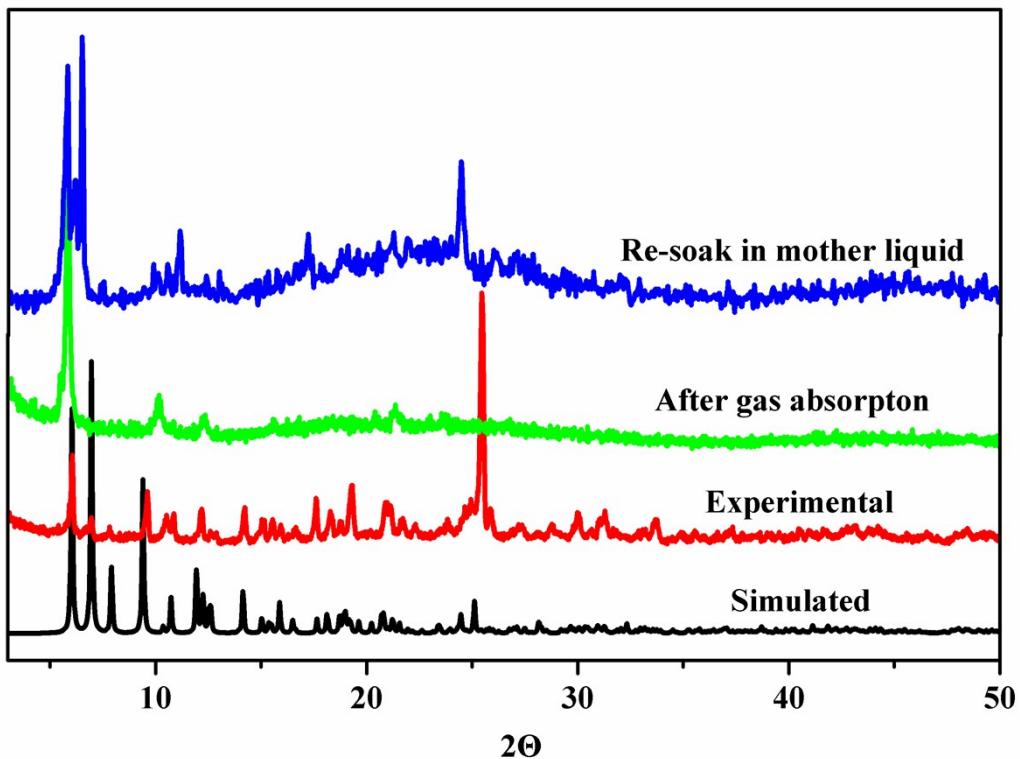


Fig. S5. PXRD patterns of **1** after different treatments.

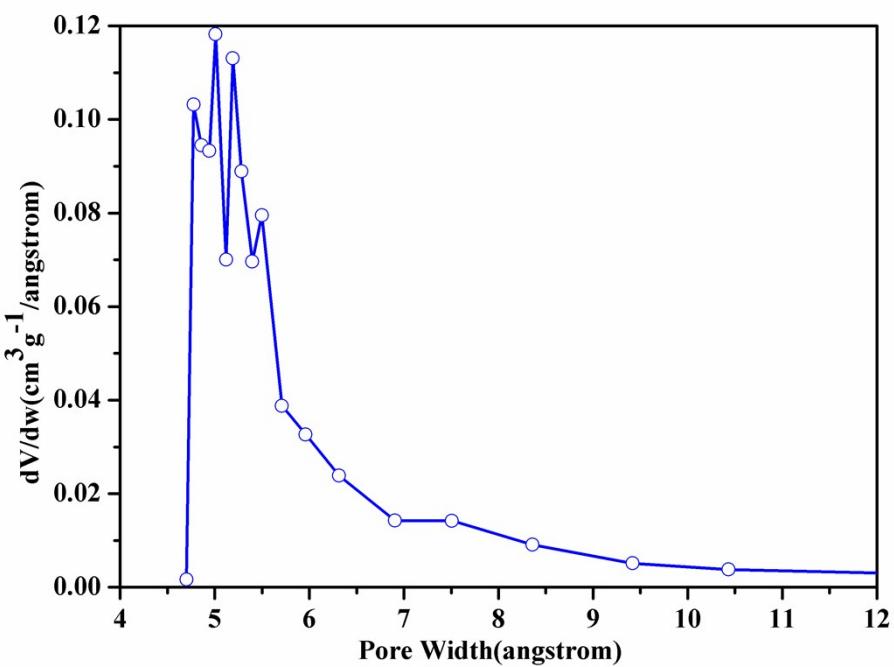


Fig. S6 The pore size distribution of **1a** calculated by Horvath-Kawazoe method.

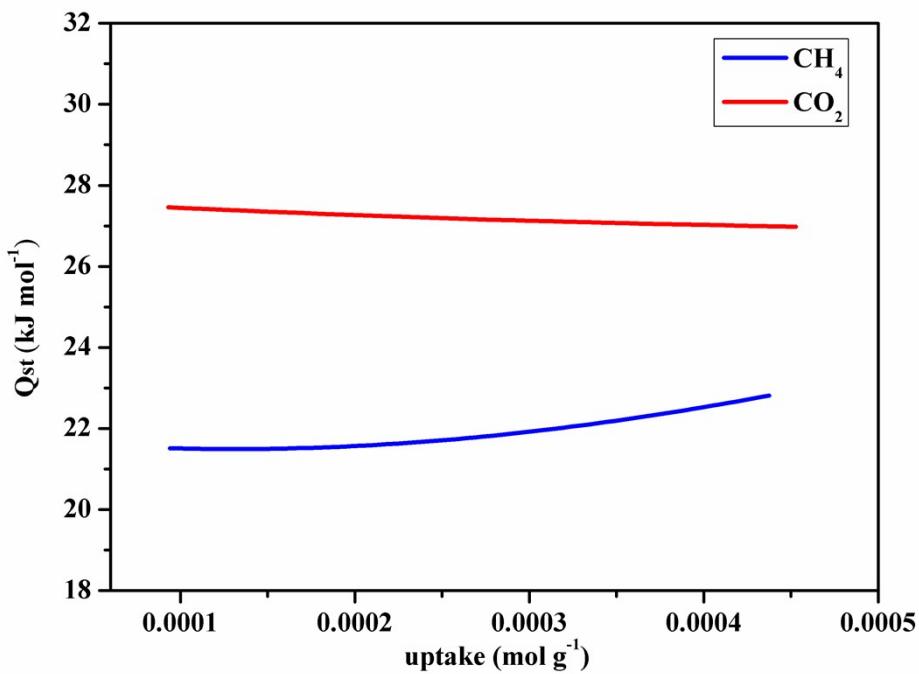


Fig. S7 Isosteric heat of sorption (Q_{st}) of **1a** to CO₂ (red) and CH₄(blue)

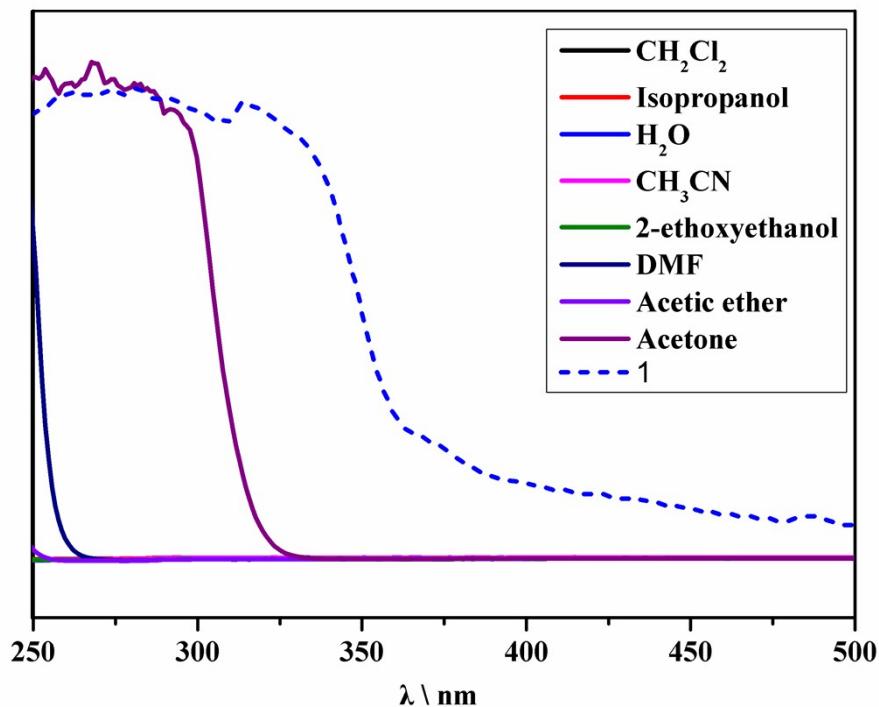


Fig. S8 The solid state UV-Visible spectrum of **1** and the UV-Visible absorption spectra of different solvents.

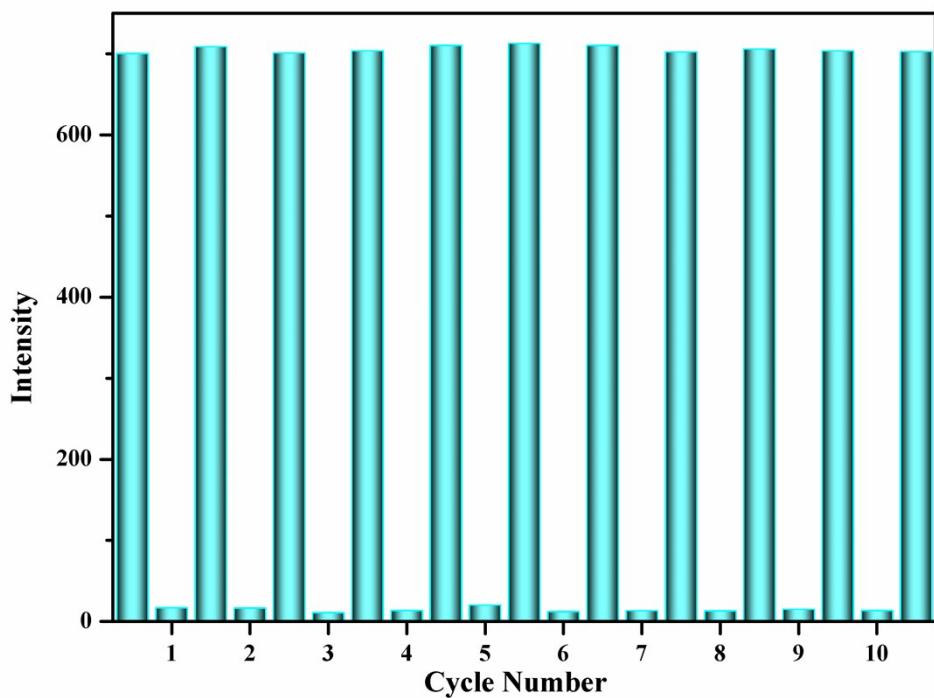


Fig. S9 The sensing recyclability of **1**. The emission at 621 nm was used for comparison.

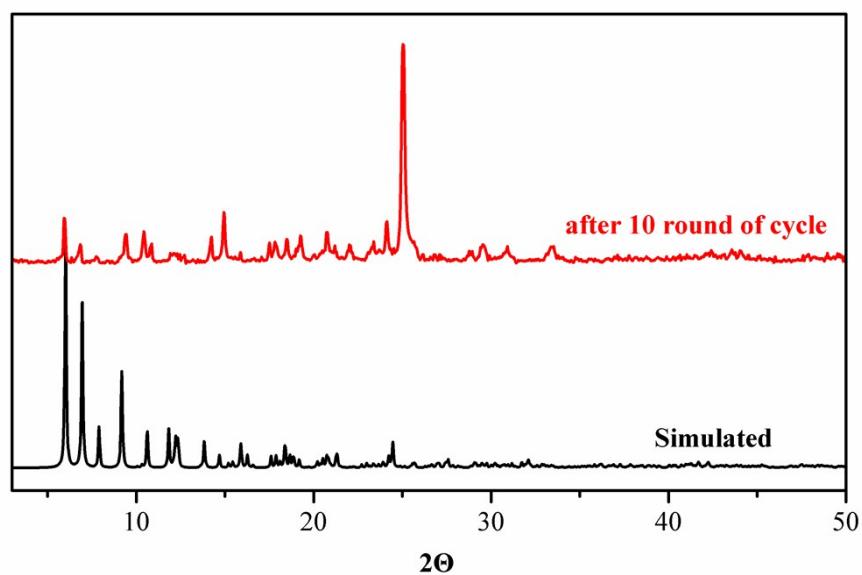


Fig. S10 PXRD pattern of **1** after 10 cycles of acetone sensing.

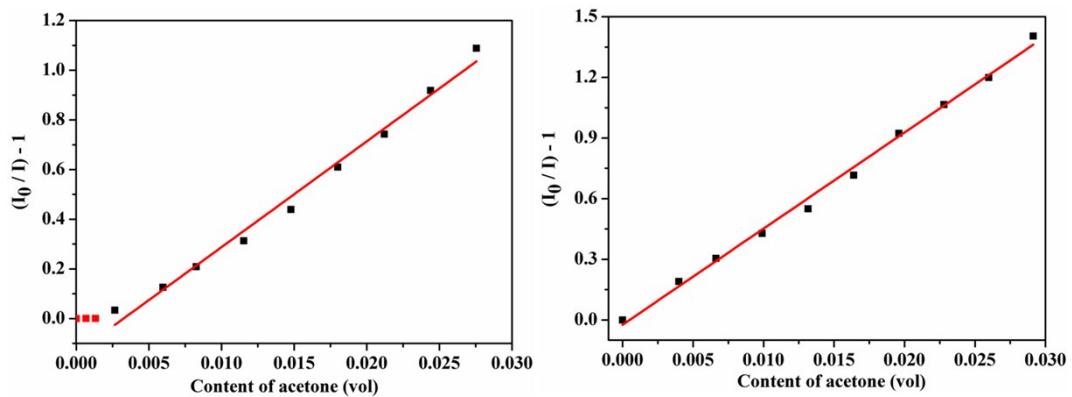


Fig. S11 The relationship between quenching efficiency and acetone content of 1(left) and 1a (right) fitted by the Stern–Volmer (SV) equation.

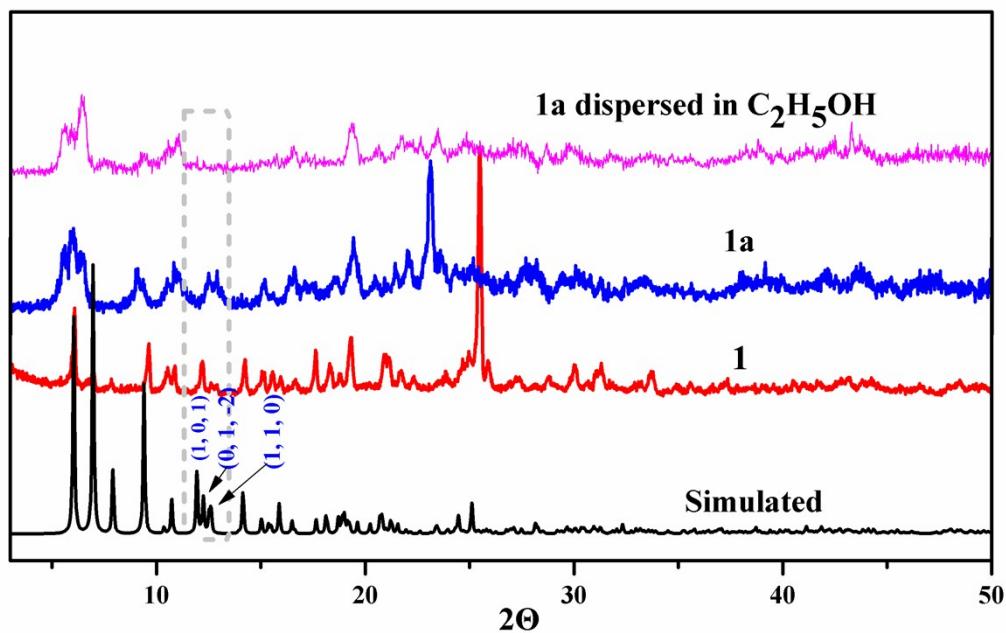


Fig. S12 The PXRD patterns of 1 change after different treatments.