## **Tetraphenylethene Modified β-Ketoiminate Boron**

## **Complexes Bearing Aggregation-Induced Emission and**

## Mechanofluorochromism

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solvent	Viscosit	$\lambda_{abs} (nm)$	ε <sup>max</sup>	$\lambda_{em}$	$\Delta v_{st}{}^a$	$\Phi_{\mathrm{f}}{}^{\mathrm{b}}$	$\tau^{c}$	$\mathrm{K_{f}}^{d}/\mathrm{K_{nr}}^{e}$
	y (cP at		(M <sup>-1</sup>	(nm)	(cm <sup>-1</sup> )		(ns)	$(10^9 \text{ s}^{-1})$
	20 °C)		cm <sup>-1</sup> )					
Hexane	0.31	401	46100	480	4104	0.015	1.87	0.0080/0.5267
Cyclohexane	1.00	303, 403	48300	485	4195	0.009	1.66	0.0054/0.5970
Toluene	0.59	303, 405	43600	495	4489	0.007	1.52	0.0046/0.6533
THF	0.55	305, 402	46300	508	5191	0.004	f	—/—
DCM	0.44	302, 402	45100	528	5936	0.005	f	—/—
DMF	0.92	302, 401	44600	532	6141	0.003	f	—/—
DMSO	2.24	304, 403	45000	542	6364	0.005	f	—/—
Ethylene	23.5	305, 405	31800	518	5386	0.105	0.79	0.1329/1.1329
glycol								
Glycerol	1412	406	28500	501	4670	0.165	0.98	0.1684/0.8520

Table S1. Photophysical data of TTPE-H in different solvents.

 ${}^{a}\Delta v_{st} = v_{abs}$ - $v_{em}$ .  ${}^{b}$ The fluorescence quantum yield ( $\Phi_{f}$ ) was measured using quinine sulfate as a standard ( $\Phi$ =0.546 in 0.5 mol L<sup>-1</sup> H<sub>2</sub>SO<sub>4</sub>) as standard.  ${}^{c}$ Fluorescence lifetime.  ${}^{d}$ Radiative rate constant K<sub>f</sub> =  $\Phi_{f}/\tau$ .  ${}^{e}$ Nonradiative rate constant K<sub>nr</sub> = (1 -  $\Phi_{f})/\tau$ .  ${}^{f}$ Too short to be measured (<0.1 ns).

 Table S2. Photophysical data of TTPE-CN in different solvents.

solvent	Viscosit	$\lambda_{abs} (nm)$	€ <sup>max</sup>	$\lambda_{em}$	$\Delta v_{st}{}^a$	$\Phi_{\rm f}{}^{\rm b}$	$\tau^c$	$\mathrm{K_{f}}^{d}/\mathrm{K_{nr}}^{e}$
	y (cP at		(M <sup>-1</sup>	(nm)	(cm <sup>-1</sup> )		(ns)	$(10^9 \text{ s}^{-1})$
	20 °C)		cm <sup>-1</sup> )					
Hexane	0.31	408	22800	498	4429	0.025	1.58	0.0158/0.6171
Cyclohexane	1.00	302, 410	21900	519	5122	0.005	1.32	0.0038/0.7538
Toluene	0.59	301, 409	38500	529	5546	0.004	1.09	0.0037/0.9138
THF	0.55	304, 402	38300	565	7177	0.003	f	—/—
DCM	0.44	304, 405	37900	583	7539	0.001	f	—/—
DMF	0.92	303, 397	35900	607	8714	0.002	f	—/—
DMSO	2.24	397	35100	612	8850	0.002	f	—/—
Ethylene	23.5	309, 414	19300	533	5393	0.172	0.83	0.2072/0.9976
glycol								
Glycerol	1412	418	15600	514	4468	0.204	1.09	0.1872/0.7303

 ${}^{a}\Delta v_{st} = v_{abs}$ - $v_{em}$ .  ${}^{b}$ The fluorescence quantum yield ( $\Phi_{f}$ ) was measured using quinine sulfate as a standard ( $\Phi$ =0.546 in 0.5 mol L<sup>-1</sup> H<sub>2</sub>SO<sub>4</sub>) as standard.  ${}^{c}$ Fluorescence lifetime.  ${}^{d}$ Radiative rate constant K<sub>f</sub> =  $\Phi_{f}/\tau$ .  ${}^{e}$ Nonradiative rate constant K<sub>nr</sub> = (1 -  $\Phi_{f}$ )/ $\tau$ .  ${}^{f}$ Too short to be measured (<0.1 ns).

Compound	$f_w$	$\lambda_{abs} (nm)$	$\lambda_{em} (nm)$	$\Phi_{\mathrm{f}}{}^{a}$	$T^{b}(ns)$	$K_{\rm f}^{c}/K_{\rm nr}^{d}(10^9 {\rm s}^{-1})$				
	0%	402	508	0.0042	e	—/—				
	20%	403	509	0.0043	e	—/—				
	40%	403	507	0.0047	e	—/—				
ТТРЕ-Н	60%	403	510	0.0053	e	—/—				
	70%	403	513	0.0056	e	—/—				
	80%	411	508	0.0375	0.78	0.0481/1.2340				
	85%	409	508	0.0983	0.96	0.1024/0.9393				
	90%	408	513	0.2355	1.32	0.1784/0.5792				
	0%	402	565	0.0034	e	—/—				
	20%	403	579	0.0018	e	—/—				
	40%	403	580	0.0019	e	—/—				
TTPE-CN	60%	404	583	0.0024	e	—/—				
	70%	403	583	0.0026	e	—/—				
	80%	408	556	0.0224	0.77	0.0291/1.2696				
	85%	408	553	0.0699	0.99	0.0706/0.9395				
	90%	410	553	0.2082	1.39	0.1498/0.5696				

**Table S3**. Photophysical data of **TTPE-H** and **TTPE-CN** in water/THF mixtures with different water fractions ( $f_w$ , 1.0 × 10<sup>-5</sup> mol L<sup>-1</sup>).

<sup>*a*</sup>The fluorescence quantum yield ( $\Phi_f$ ) was measured using fluorescein ( $\Phi$ =0.79 in 0.1 M sodium hydroxide water solution) as standard. <sup>*b*</sup>Fluorescence lifetime, the fluorescence decays can be fitted with a single exponential function. <sup>*c*</sup>Radiative rate constant  $K_f = \Phi_f/\tau$ . <sup>*d*</sup>Nonradiative rate constant  $K_{nr} = (1 - \Phi_f)/\tau$ . <sup>*e*</sup>Too short to be measured (<0.1 ns).



**Fig. S1** DLS data of **TTPE-H**  $(1.0 \times 10^{-5} \text{ mol } \text{L}^{-1})$  in ethylene glycol–THF mixture with ethylene glycol fraction of 90%.

Sample ID Operator ID Elapsed Time Mean Diam. Rel. Var. Skew RmsError	90 Unkn 00:01 187.3 0.000 0.008 3.952	own Op 1:33 5 (nm) 0 3 27e-03	erator			100 50 0 50.0		Diame	500.0 ter (nm)
d	G(d)	C(d)	d	G(d)	C(d)	d	G(d)	C(d)	
174.12	0	0	187.48	100	66	201.87	0	100	
175.30	0	0	188.75	67	89	203.23	0	100	
176.48	0	0	190.02	34	100	204.60	0	100	
177.67	0	0	191.30	0	100	205.98	0	100	
178.87	0	0	192.59	0	100	207.37	0	100	
180.07	0	0	193.89	0	100	208.77	0	100	Print Window
181.29	0	0	195.20	0	100	210.18	0	100	
182.51	0	0	196.51	0	100	211.59	0	100	Copy For Spreadsheet
183.74	0	0	197.84	0	100	213.02	0	100	
184.98	33	11	199.17	0	100	214.46	0	100	<u>C</u> opy to Clipboard
186.23	66	33	200.52	0	100	215.90	0	100	Close

**Fig. S2** DLS data of **TTPE-CN** ( $1.0 \times 10^{-5}$  mol L<sup>-1</sup>) in ethylene glycol–THF mixture with ethylene glycol fraction of 90%.



**Fig. S3** DLS data of **TTPE-H**  $(1.0 \times 10^{-5} \text{ mol } \text{L}^{-1})$  in ethylene glycol–THF mixture with ethylene glycol fraction of 85%.

Sample ID Operator ID Elapsed Time Mean Diam. Rel. Var. Skew RmsError	85 Unkn 00:02 199.5 0.001 0.013 8.857	own Op 2:04 5 (nm) 1 3 73e-03	erator			100 50 0 50.0		Diame	500.0 ter (nm)
d	G(d)	C(d)	d	G(d)	C(d)	d	G(d)	C(d)	
153.89	0	0	210.66	37	100	288.38	0	100	
158.34	0	0	216.76	0	100	296.73	0	100	
162.93	0	0	223.04	0	100	305.32	0	100	
167.65	0	0	229.50	0	100	314.16	0	100	
172.50	0	0	236.14	0	100	323.26	0	100	
177.50	0	0	242.98	0	100	332.62	0	100	Print Window
182.64	0	0	250.02	0	100	342.25	0	100	
187.93	29	10	257.26	0	100	352.17	0	100	Copy For Spreadsheet
193.37	65	31	264.71	0	100	362.36	0	100	
198.97	100	64	272.37	0	100	372.86	0	100	<u>Copy to Clipboard</u>
204.73	72	88	280.26	0	100	383.65	0	100	Close

**Fig. S4** DLS data of **TTPE-CN** ( $1.0 \times 10^{-5}$  mol L<sup>-1</sup>) in ethylene glycol–THF mixture with ethylene glycol fraction of 85%.



**Fig. S5** DLS data of **TTPE-H** ( $1.0 \times 10^{-5}$  mol L<sup>-1</sup>) in ethylene glycol–THF mixture with ethylene glycol fraction of 80%.

Sample ID Operator ID Elapsed Time Mean Diam. Rel. Var. Skew RmsError	80 Unkn 00:03 669.7 0.235 -0.18 1.442	own Ope 3:58 7 (nm) 5 14 26e-04	erator			100 50 0 50.0		Diame	5000.0 eter (nm)
d	G(d)	C(d)	d	G(d)	C(d)	d	G(d)	C(d)	
226.73	0	0	421.84	0	44	784.85	0	44	
239.90	0	0	446.34	0	44	830.42	30	49	
253.83	0	0	472.25	0	44	878.64	55	58	
268.57	15	2	499.68	0	44	929.66	100	74	
284.16	51	11	528.69	0	44	983.64	85	88	
300.66	86	25	559.39	0	44	1040.75	57	98	Print Window
318.12	75	37	591.87	0	44	1101.19	14	100	
336.59	38	44	626.23	0	44	1165.13	0	100	Copy For Spreadsheet
356.13	0	44	662.60	0	44	1232.78	0	100	
376.81	0	44	701.07	0	44	1304.36	0	100	<u>C</u> opy to Clipboard
398.69	0	44	741.78	0	44	1380.10	0	100	Close

**Fig. S6** DLS data of **TTPE-CN**  $(1.0 \times 10^{-5} \text{ mol } L^{-1})$  in ethylene glycol–THF mixture with ethylene glycol fraction of 80%.



**Fig. S7** PL spectra of **TTPE-H** in ethylene glycol–THF mixtures with different ethylene glycol fractions (from 0% to 100%,  $1.0 \times 10^{-5}$  mol L<sup>-1</sup>, excited at 390 nm). Photographs are **TTPE-H** in THF and ethylene glycol under UV light (365 nm), respectively.



Fig. S8 spectra of TTPE-CN in ethylene glycol–THF mixtures with different ethylene glycol fractions (from 0% to 100%,  $1.0 \times 10^{-5}$  mol L<sup>-1</sup>, excited at 410 nm). Photographs are TTPE-H in THF and ethylene glycol under UV light (365 nm), respectively.



Fig. S9 PL spectra of TTPE-H in THF at different temperature (excited at 390 nm,  $1.0 \times 10^{-5} \text{ mol } \text{L}^{-1}$ ).



Fig. S10 PL spectra of TTPE-CN in THF at different temperature (excited at 410 nm,  $1.0 \times 10^{-5} \text{ mol } \text{L}^{-1}$ ).



**Fig. S11** Fluorescence microscopy image of **TTPE-H** powders. The excitation wavelength for measurements was 330–385 nm.



**Fig. S12** Fluorescence microscopy image of **TTPE-CN** powders. The excitation wavelength for measurements was 330–385 nm.



**Fig. S13** Normalized UV-vis absorption spectra of **TTPE-H** in different solid-states: as-prepared and grinding.



**Fig. S14** Normalized UV-vis absorption spectra of **TTPE-CN** in different solid-states: as-prepared and grinding.



**Fig. S15** Maximum fluorescent emission of **TTPE-H** (a) and **TTPE-CN** (b) upon repeating treated by grinding and fuming with DCM.







Fig. S17 <sup>13</sup>C NMR (100 MHz) spectrum of compound **3**.



Fig. S18 TOF LC/MS spectrum of compound 3.



Fig. S19 <sup>1</sup>H NMR (400 MHz) spectrum of compound TTPE-H.







Fig. S21 MALDI/TOF MS spectrum of compound TTPE-H.



Fig. S22 <sup>1</sup>H NMR (400 MHz) spectrum of compound TTPE-CN.



**Fig. S23** <sup>13</sup>C NMR (100 MHz) spectrum of compound **TTPE-CN**.



Fig. S24 MALDI/TOF MS spectrum of compound TTPE-CN.