

Supplementary material

VUV/UV light inducing accelerated phenol degradation with a low electric input

Mengkai Li ^{a,b}, Dong Wen ^a, Zhimin Qiang ^{a,*} and John Kiwi ^{b,*}

^a Key Laboratory of Drinking Water Science and Technology, Research Center for Eco-Environmental Sciences, University of Chinese Academy of Sciences, Chinese Academy of Sciences, 18 Shuang-qing Road, Beijing 100085, China. E-mail: qiangz@rcees.ac.cn

^b Ecole Polytechnique Fédérale de Lausanne, EPFL-SB-ISIC-GPAO, Station 6, CH-1015 Lausanne, Switzerland. E-mail: john.kiwi@pfl.ch

* Corresponding authors.

E-mail: qiangz@rcees.ac.cn, Tel.: +86 10 62849632

E-mail: john.kiwi@epfl.ch, Tel.: +41 21 6936150

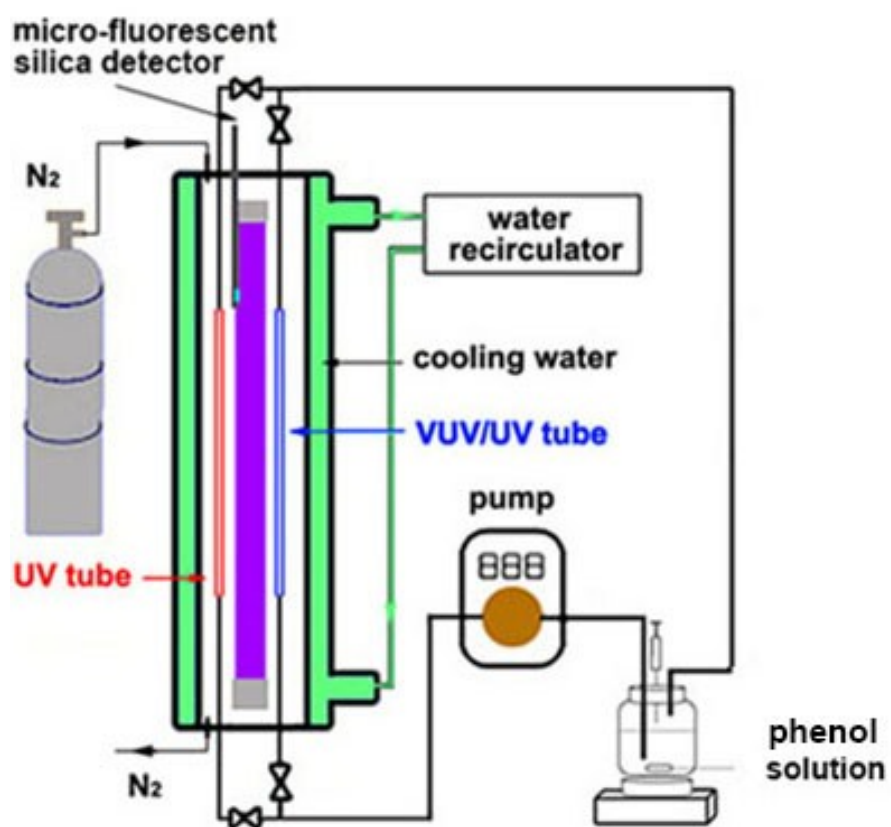


Fig. S1. Schematic diagram of the mini-fluidic VUV/UV photoreaction system (MVPS).¹

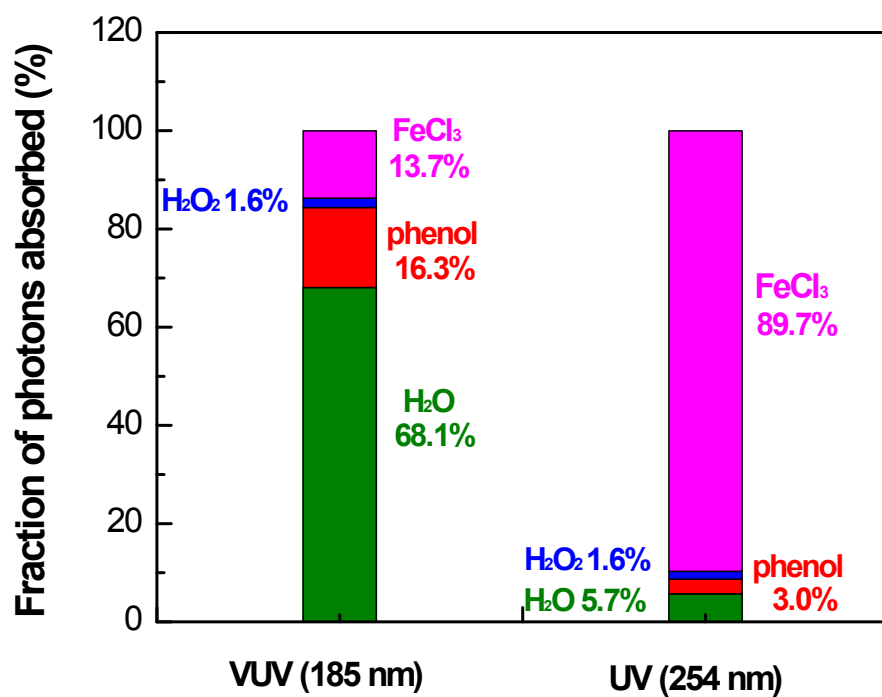


Fig. S2. Fractions of the photons absorbed by each solution component in the VUV/UV photo-Fenton process. Conditions: $[\text{phenol}]_0 = 0.055 \text{ mM}$, $[\text{H}_2\text{O}_2]_0 = 0.735 \text{ mM}$, $[\text{Fe}^{3+}]_0 = 0.25 \text{ mM}$, and $\text{pH}_0 = 3.7$.

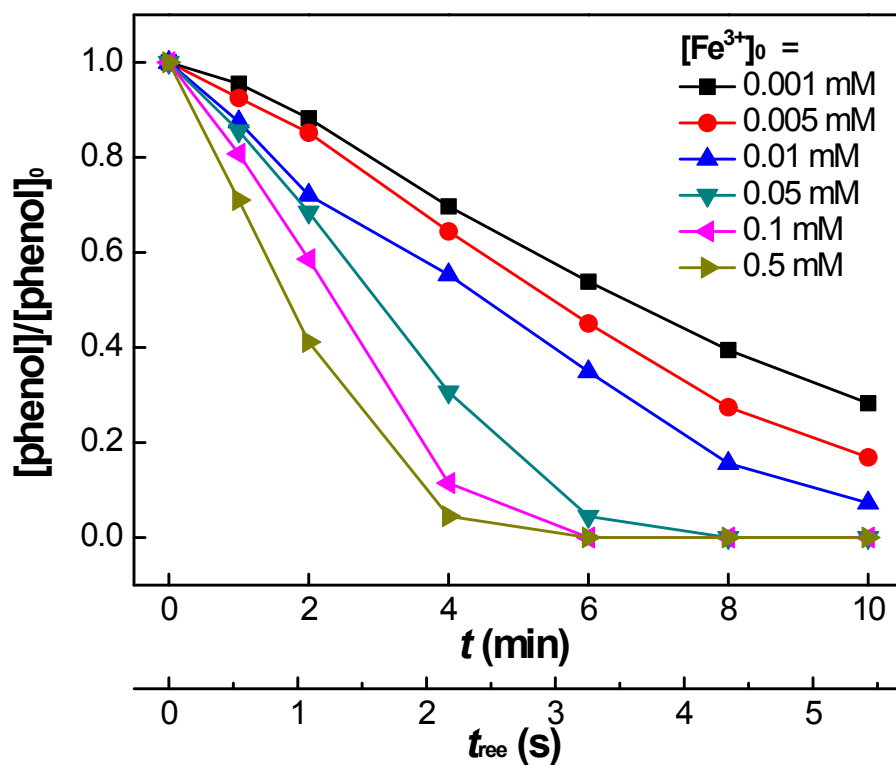


Fig. S3. Phenol degradation by the VUV/UV photo-Fenton process at various initial Fe^{3+} concentrations. Conditions: $[\text{phenol}]_0 = 0.011 \text{ mM}$, $[\text{H}_2\text{O}_2]_0 = 0.147 \text{ mM}$, and $\text{pH}_0 = 3.7$.

Reference

- 1 M. K. Li, Z. M. Qiang, P. Hou, J. R. Bolton, J. H. Qu, P. Li, C. Wang, *Environ. Sci. Technol.* 2016, **50**, 5849–5856.