

Supporting Information

Solvent-free synthesis of 5-(aryl/alkyl)amino-1,2,4-triazines and α -arylamino-2,2'-bipyridines with a greener prospect

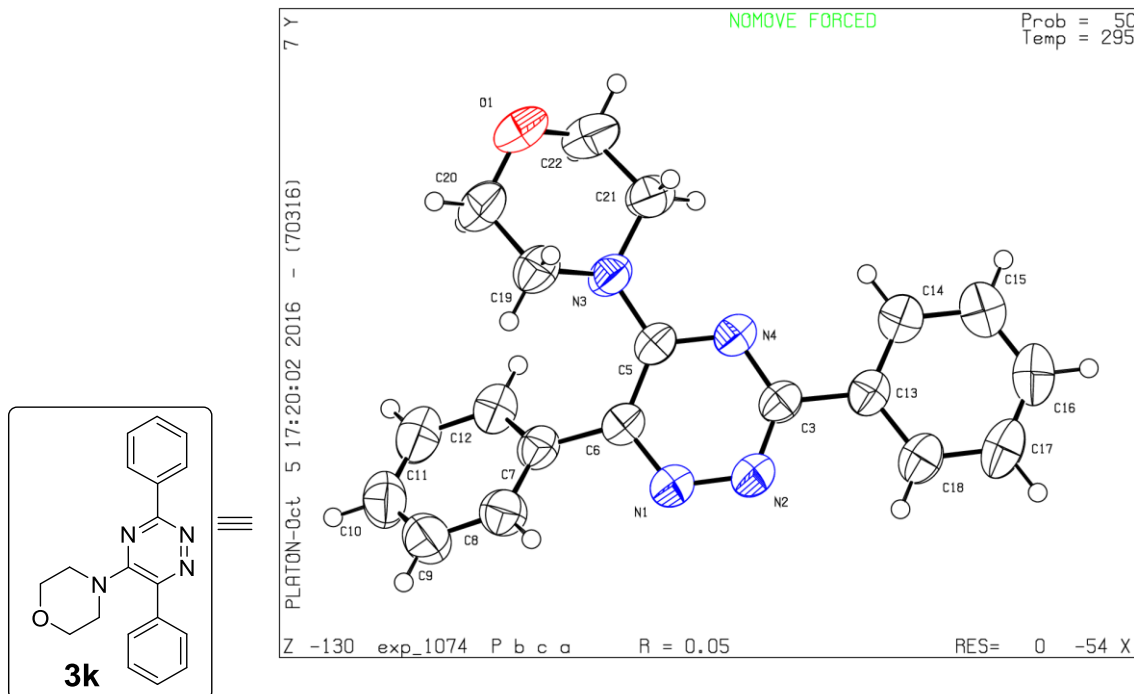
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Table of Contents

Structure Determination (X-ray crystallographic data for 3k)	2
E-factor Calculations	3
¹ H & ¹³ C NMR Spectra of all compounds	4-41

Structure Determination (X-ray crystallographic data for 3k)

The white crystals of **3k** were obtained by crystallization from a solution in ethanol after purification by column chromatography. Chemical formula C₁₉H₁₈N₄O.



ORTEP (with 50% probability) diagram for the structure 5-(Morpholine-4-yl)-3,6-diphenyl-1,2,4-triazine (**3k**).

Wavelength	0.71070 Å	
Formula	C ₁₉ H ₁₈ N ₄ O	
Crystal system	Orthorhombic	
Space group	P b c a	
Unit cell dimensions	a = 11.5790(8) Å	α = 90 °
	b = 9.4791(5) Å	β = 90 °
	c = 30.0335(19) Å	γ = 90 °
Volume	3296.43 Å ³	
Z	8	
R-factor (%)	5.14	

The crystallographic data have been deposited with the Cambridge Crystallographic Data Centres supplementary publication with a CCDC reference number **CCDC 1505314**.

E-factor calculations for compounds 3:

$$\text{For } \mathbf{3a}: E = [0.137 \text{ g (1a)} + 0.047 \text{ g (2a)} - 0.160 \text{ g (product} \times \text{yield)}] / 0.160 \text{ g} = 0.15$$

$$\text{For } \mathbf{3b}: E = [0.130 \text{ g (1b)} + 0.047 \text{ g (2a)} - 0.151 \text{ g (product} \times \text{yield)}] / 0.151 \text{ g} = 0.17$$

$$\text{For } \mathbf{3c}: E = [0.137 \text{ g (1a)} + 0.082 \text{ g (2b)} - 0.190 \text{ g (product} \times \text{yield)}] / 0.190 \text{ g} = 0.15$$

$$\text{For } \mathbf{3d}: E = [0.137 \text{ g (1a)} + 0.069 \text{ g (2c)} - 0.179 \text{ g (product} \times \text{yield)}] / 0.179 \text{ g} = 0.15$$

$$\text{For } \mathbf{3e}: E = [0.137 \text{ g (1a)} + 0.062 \text{ g (2d)} - 0.175 \text{ g (product} \times \text{yield)}] / 0.175 \text{ g} = 0.14$$

$$\text{For } \mathbf{3f}: E = [0.130 \text{ g (1b)} + 0.062 \text{ g (2e)} - 0.169 \text{ g (product} \times \text{yield)}] / 0.169 \text{ g} = 0.14$$

$$\text{For } \mathbf{3g}^a: E = [0.131 \text{ g (1c)} + 0.047 \text{ g (2a)} + 8.65 \text{ g (Toluene)} - 0.150 \text{ g (product} \times \text{yield)}] / 0.150 \text{ g} = 58$$

$$\text{For } \mathbf{3h}^a: E = [0.139 \text{ g (1d)} + 0.047 \text{ g (2a)} + 8.65 \text{ g (Toluene)} - 0.150 \text{ g (product} \times \text{yield)}] / 0.150 \text{ g} = 58$$

$$\text{For } \mathbf{3i}: E = [0.129 \text{ g (1e)} + 0.047 \text{ g (2a)} - 0.160 \text{ g (product} \times \text{yield)}] / 0.160 \text{ g} = 0.10$$

$$\text{For } \mathbf{3j}^a: E = [0.129 \text{ g (1e)} + 0.044 \text{ g (2f)} + 8.65 \text{ g (Toluene)} - 0.146 \text{ g (product} \times \text{yield)}] / 0.146 \text{ g} = 59$$

$$\text{For } \mathbf{3k}^a: E = [0.129 \text{ g (1e)} + 0.051 \text{ g (2g)} + 8.65 \text{ g (Toluene)} - 0.143 \text{ g (product} \times \text{yield)}] / 0.143 \text{ g} = 61$$

$$\text{For } \mathbf{3l}: E = [0.129 \text{ g (1e)} + 0.036 \text{ g (2h)} - 0.137 \text{ g (product} \times \text{yield)}] / 0.137 \text{ g} = 0.20$$

$$\text{For } \mathbf{3m}: E = [0.129 \text{ g (1e)} + 0.050 \text{ g (2i)} - 0.149 \text{ g (product} \times \text{yield)}] / 0.149 \text{ g} = 0.20$$

$$\text{For } \mathbf{3n}^a: E = [0.105 \text{ g (1f)} + 0.047 \text{ g (2a)} + 8.65 \text{ g (Toluene)} - 0.120 \text{ g (product} \times \text{yield)}] / 0.120 \text{ g} = 72$$

$$\text{For } \mathbf{3o}^a: E = [0.137 \text{ g (1a)} + 0.085 \text{ g (2j)} + 8.65 \text{ g (Toluene)} - 0.053 \text{ g (product} \times \text{yield)}] / 0.053 \text{ g} = 166$$

^aFlash chromatography was done using toluene as eluent.

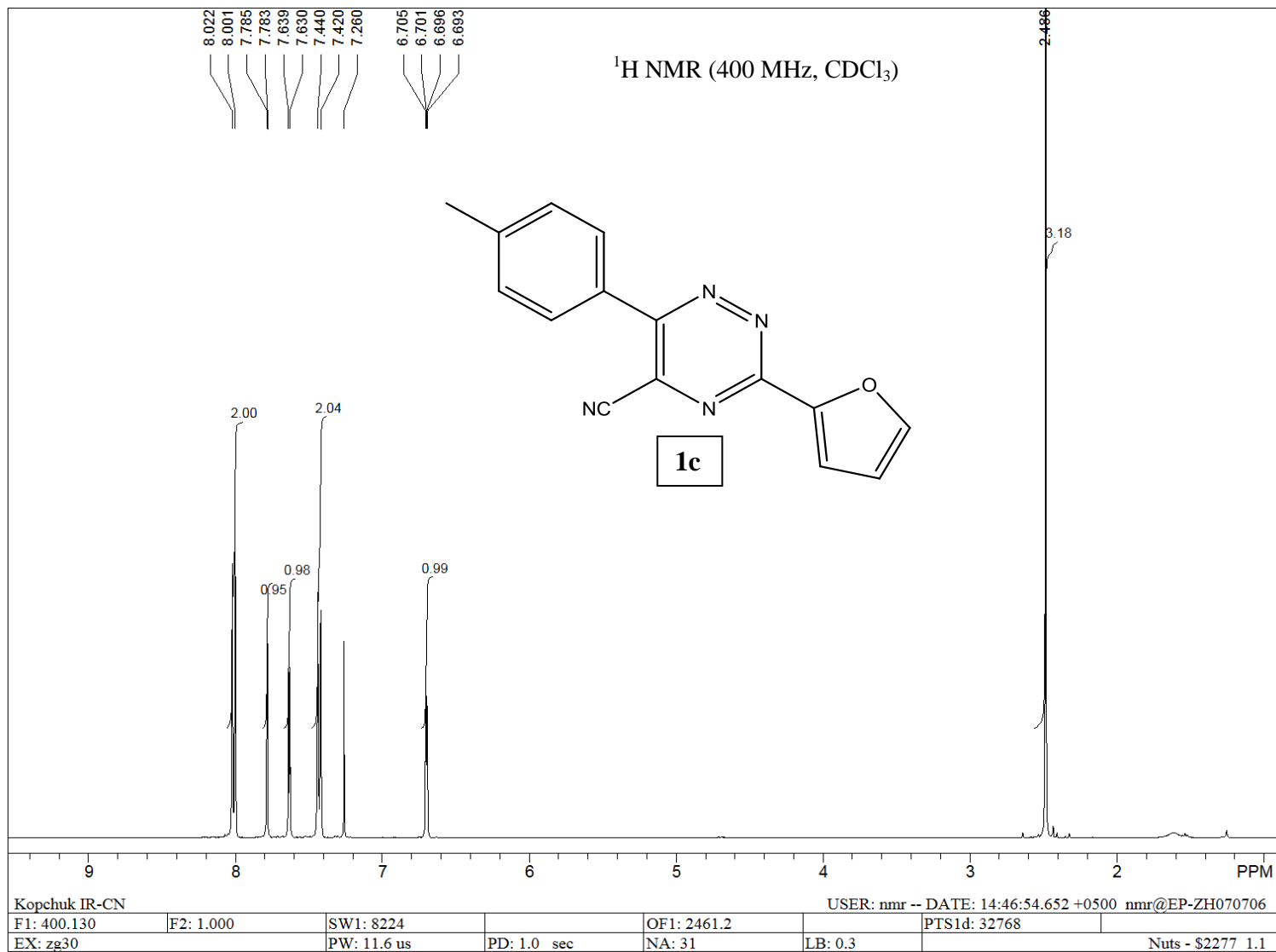
E-factor calculations for compounds 4:^a

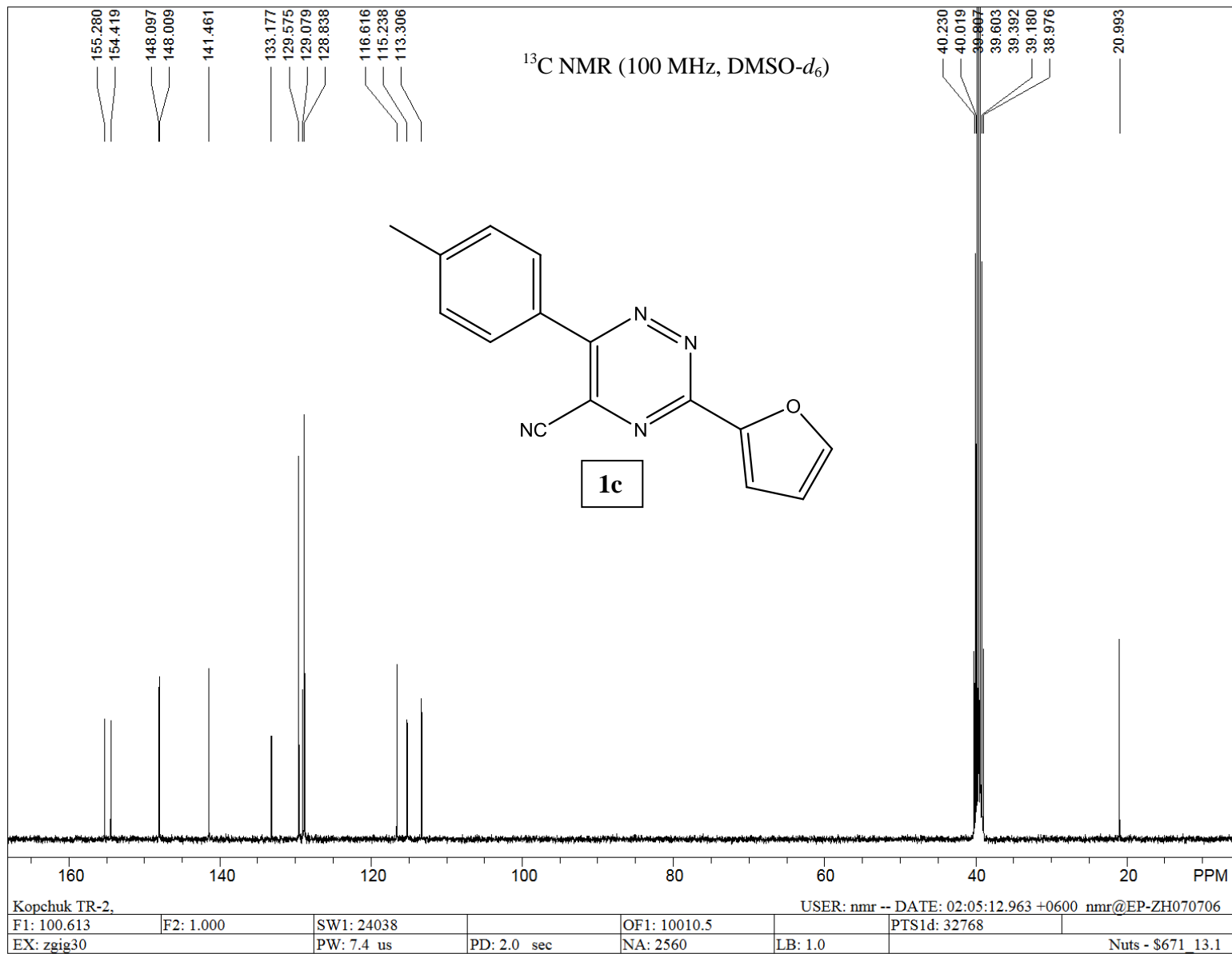
$$\text{For } \mathbf{4a}: E = [0.095 \text{ g (3a)} + 0.064 \text{ g (1-morpholinocyclopentene)} + 3.945 \text{ g (Ethanol)} - 0.100 \text{ g (product} \times \text{yield)}] / 0.100 \text{ g} = 40. \text{ (0.28 mmol reaction was carried out)}$$

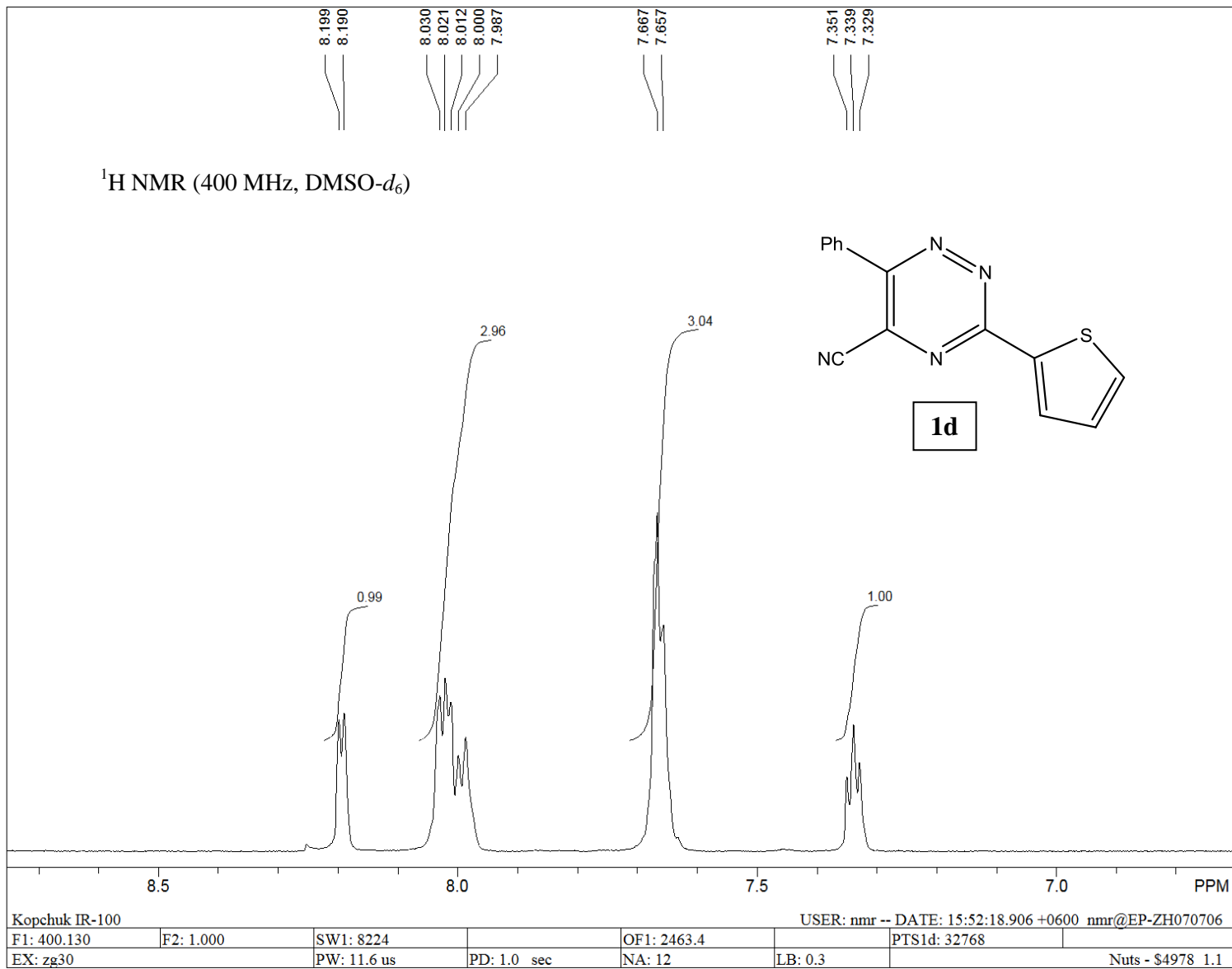
$$\text{For } \mathbf{4b}: E = [0.084 \text{ g (3b)} + 0.060 \text{ g (1-morpholinocyclopentene)} + 3.945 \text{ g (Ethanol)} - 0.096 \text{ g (product} \times \text{yield)}] / 0.096 \text{ g} = 41. \text{ (0.26 mmol reaction was carried out.)}$$

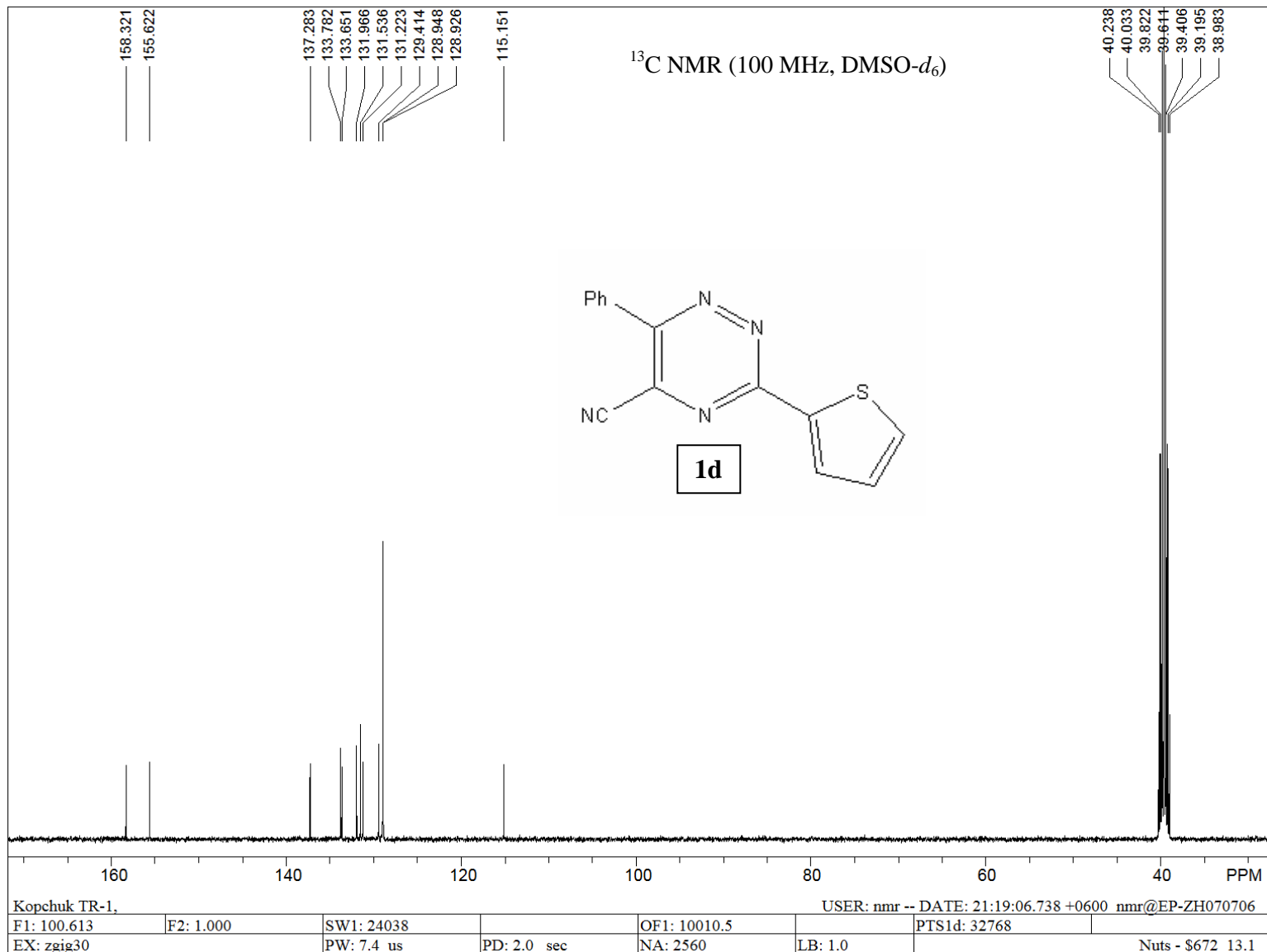
$$\text{For } \mathbf{4c}: E = [0.092 \text{ g (3e)} + 0.057 \text{ g (1-morpholinocyclopentene)} + 3.945 \text{ g (Ethanol)} - 0.100 \text{ g (product} \times \text{yield)}] / 0.100 \text{ g} = 40. \text{ (0.25 mmol reaction was carried out)}$$

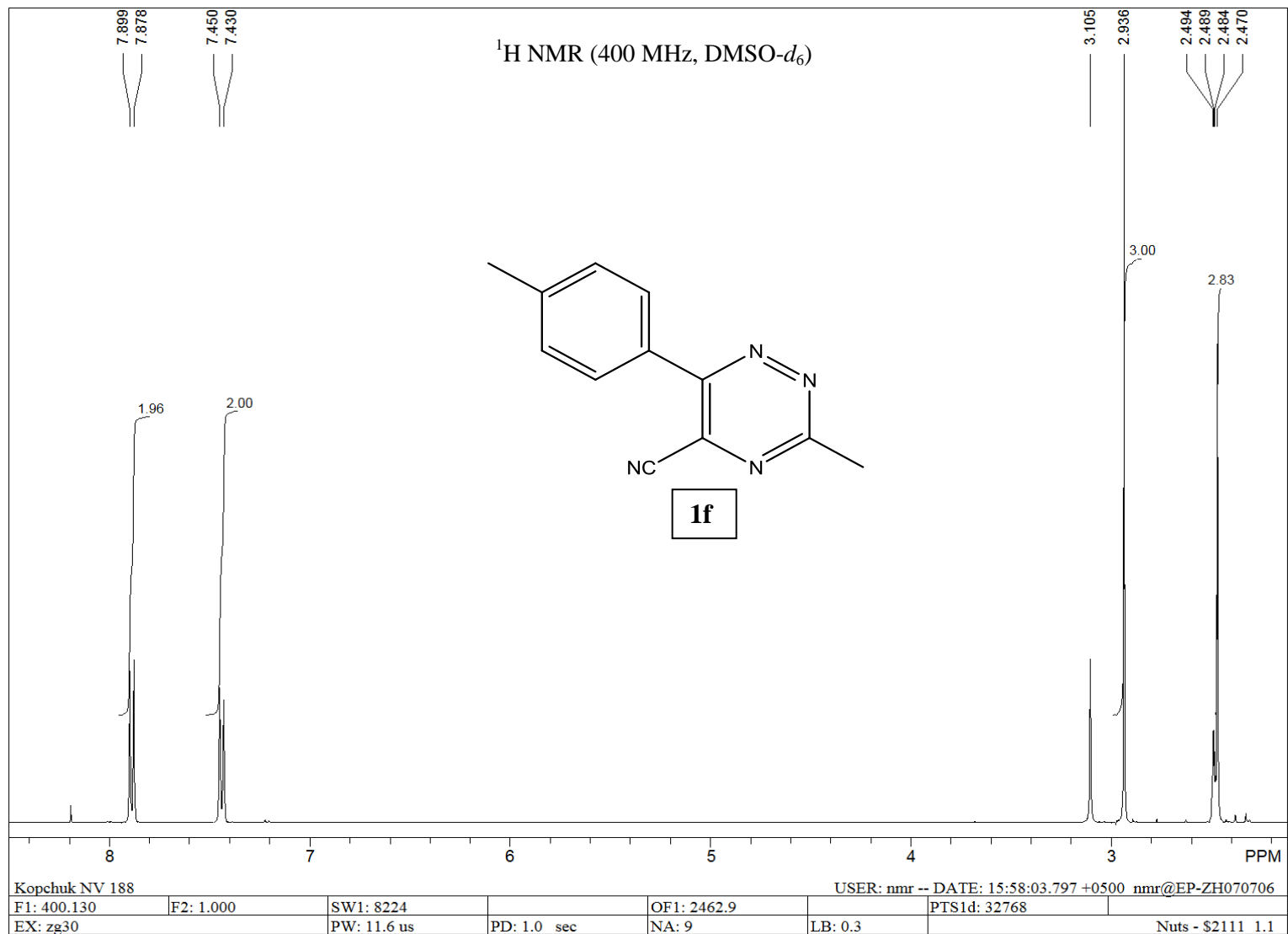
^a 5 mL ethanol was used for recrystallization.

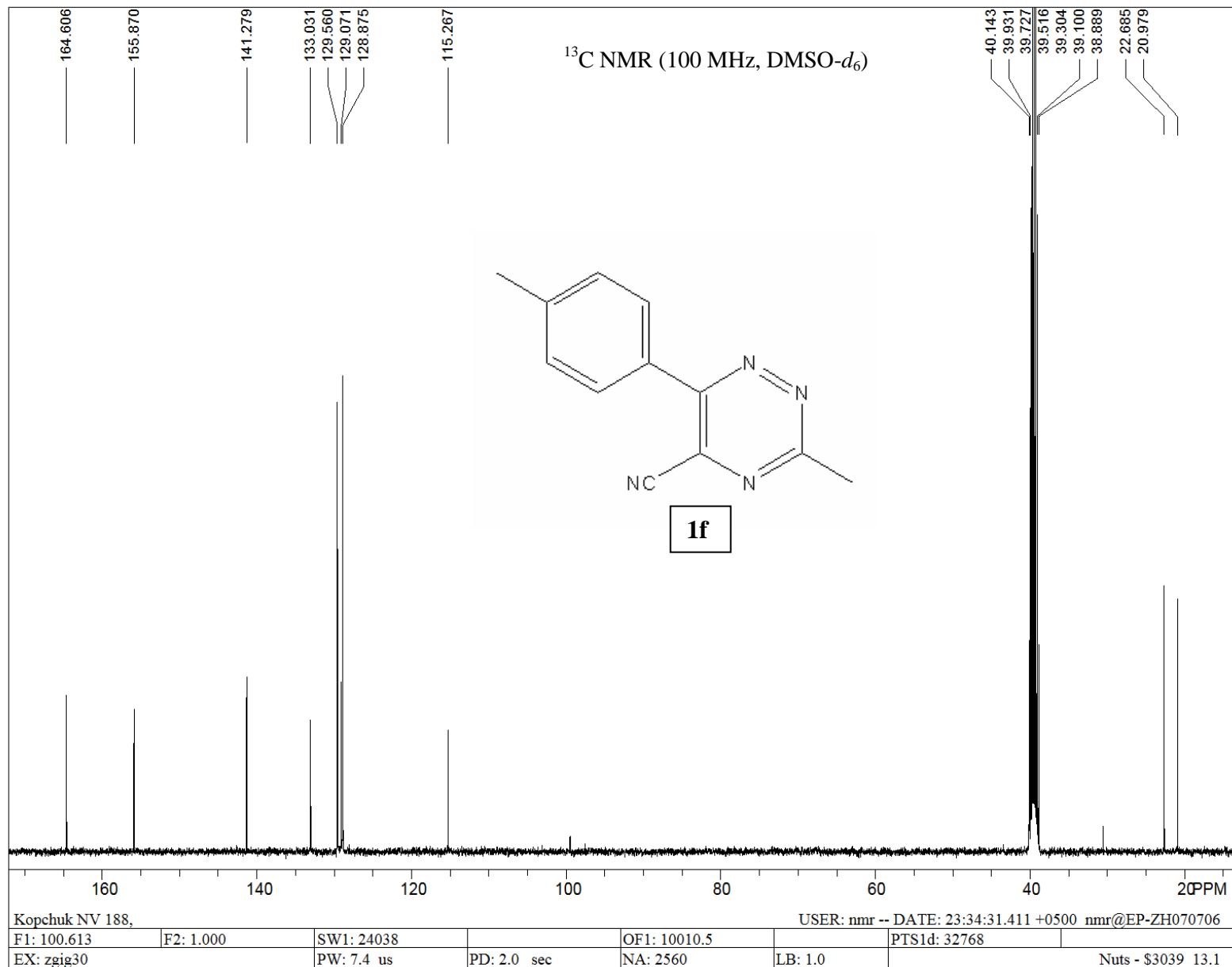


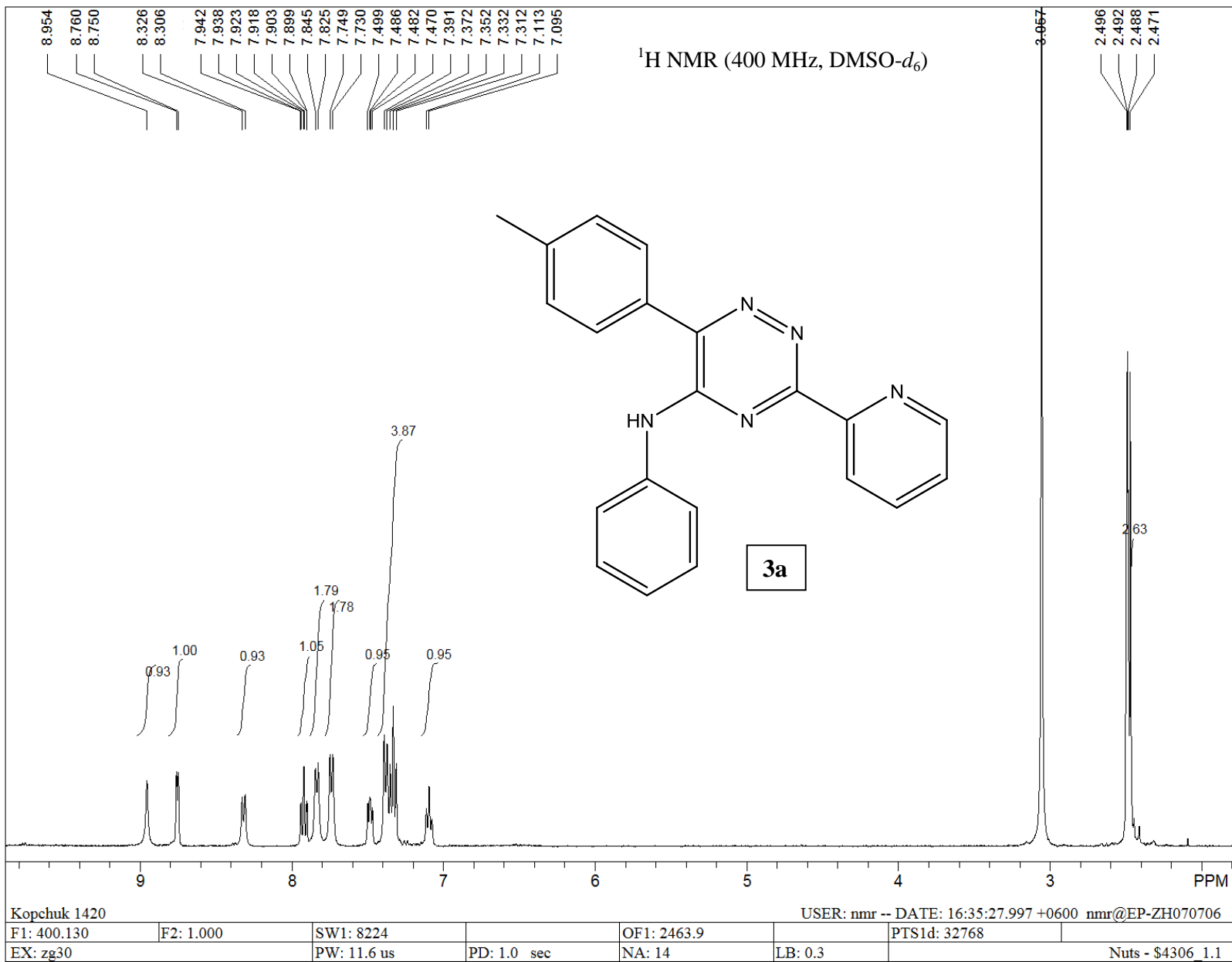


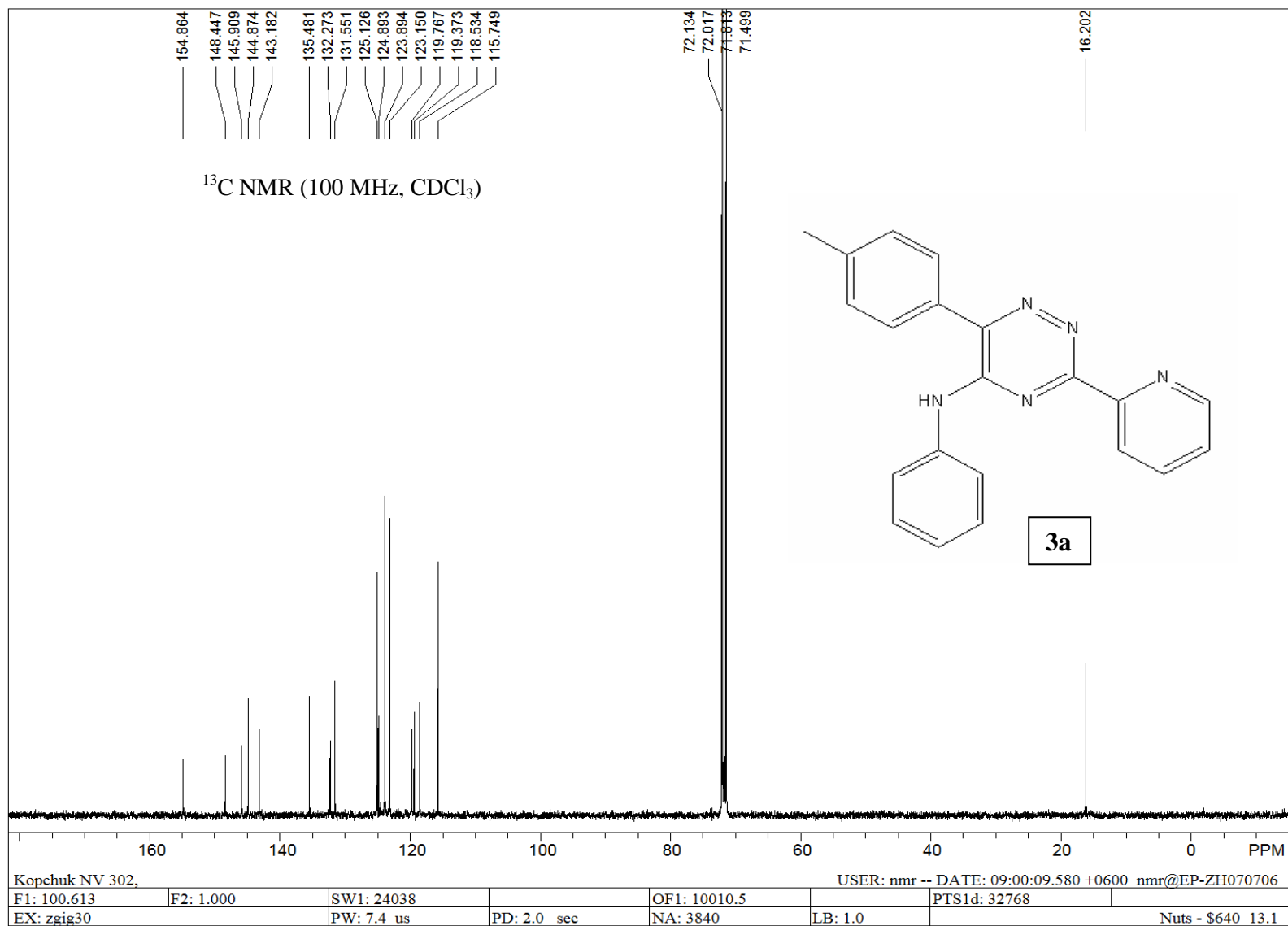


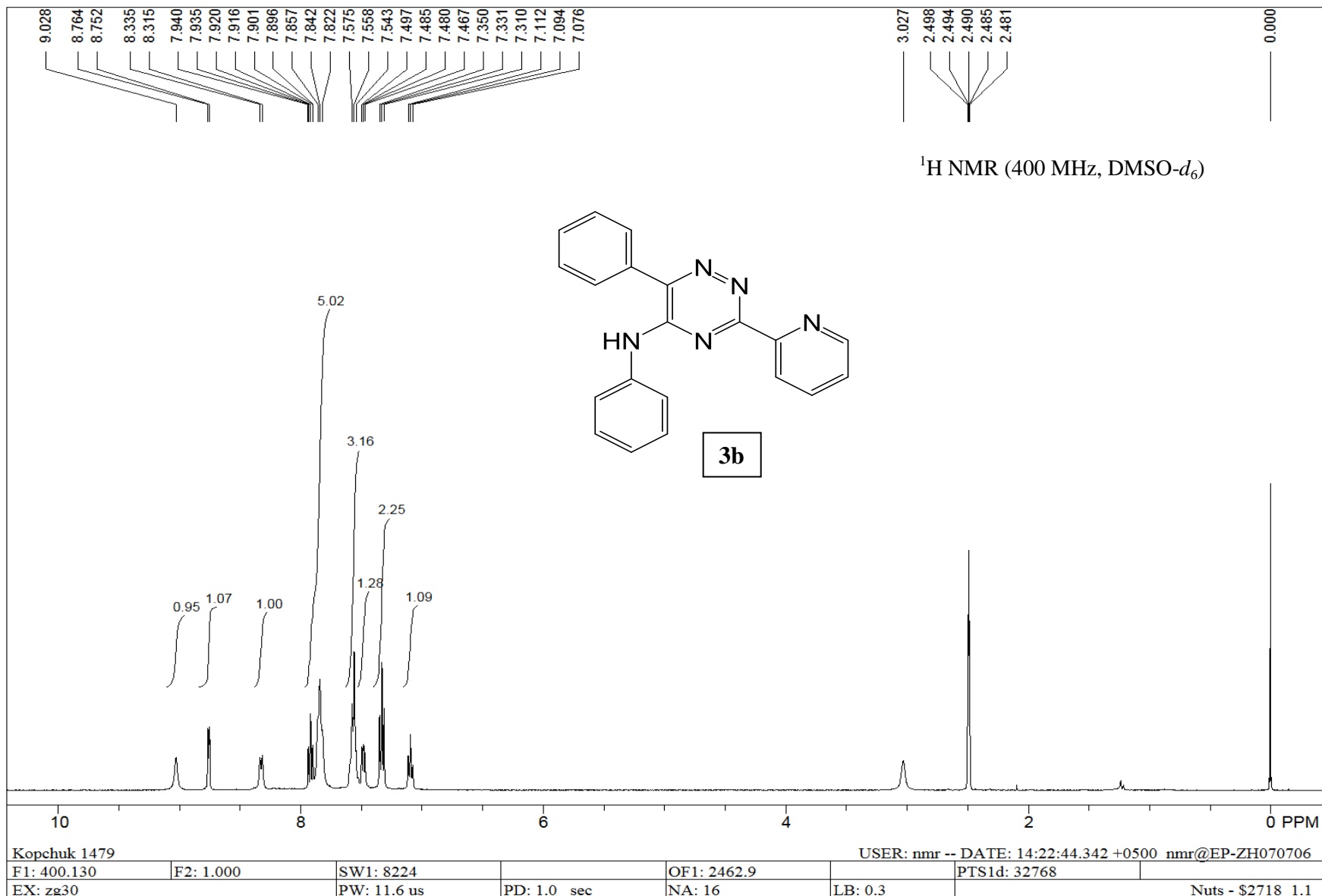


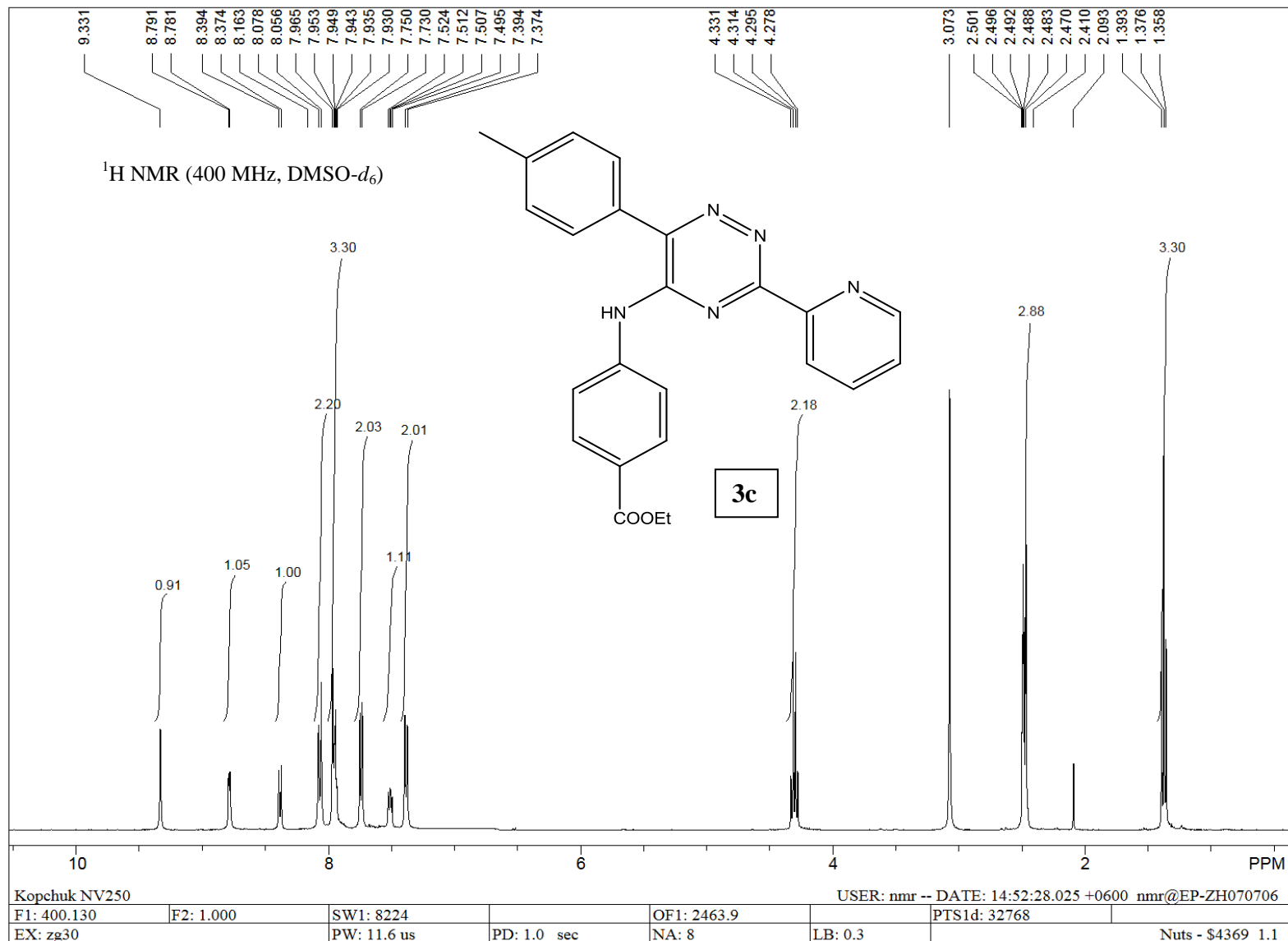


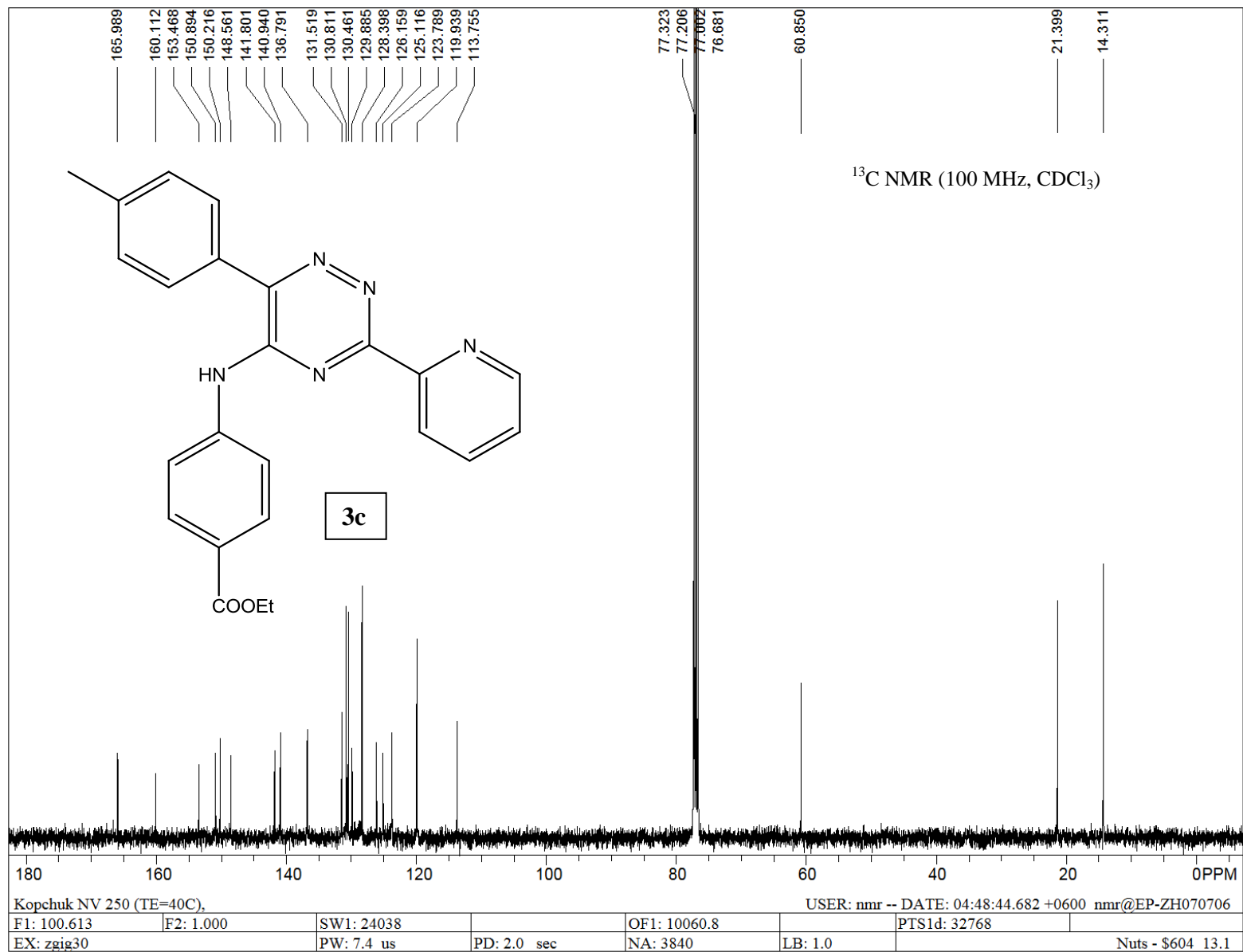


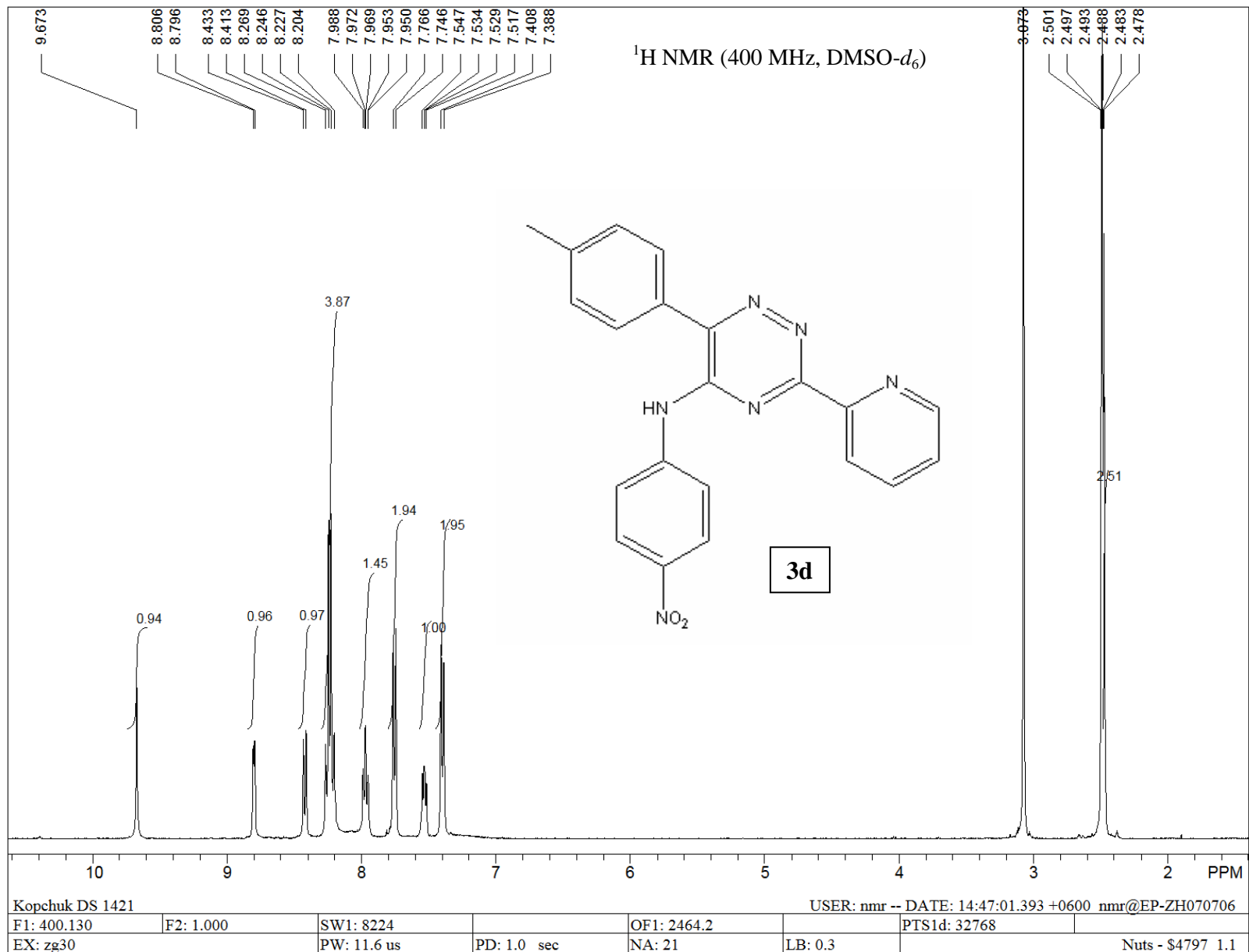


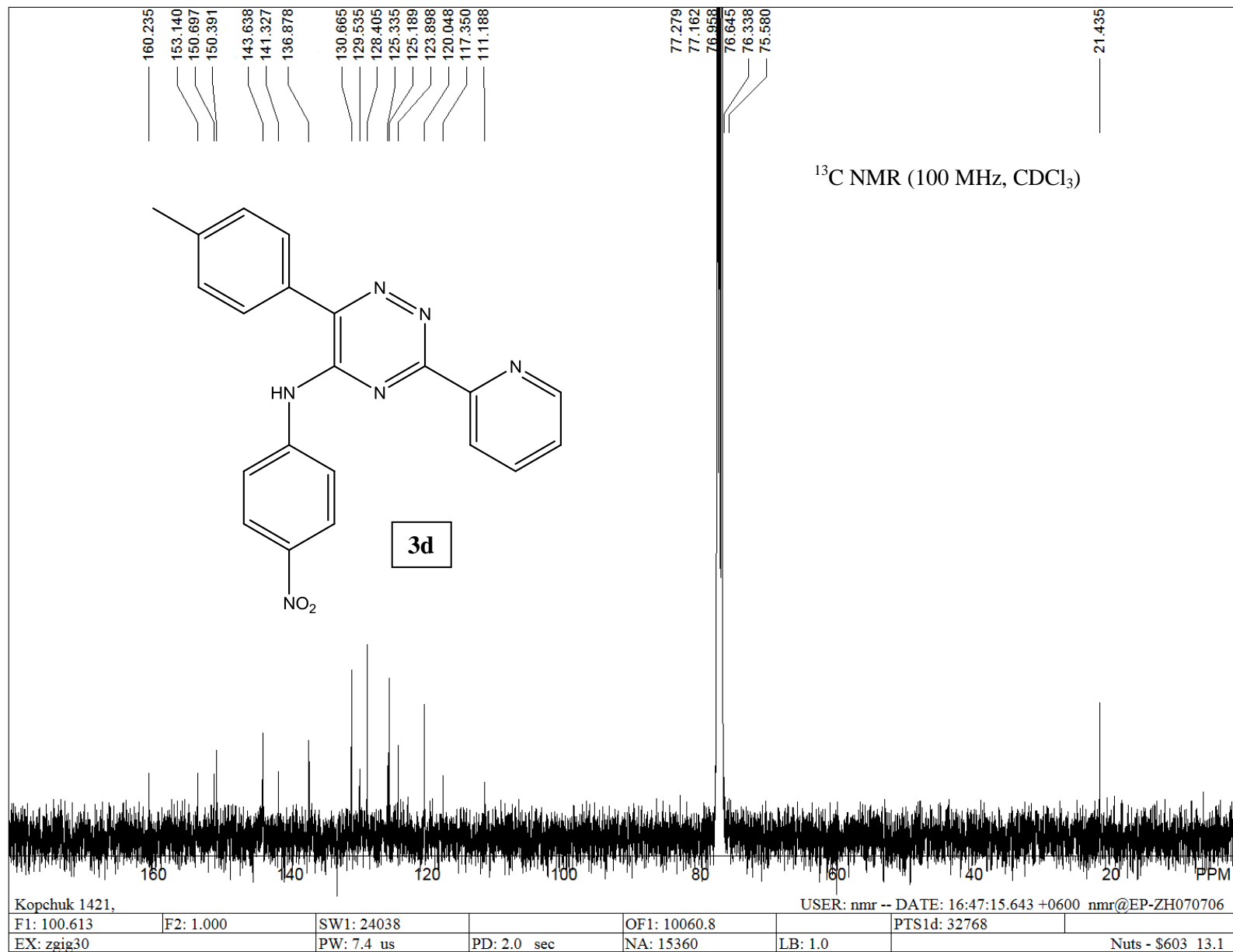


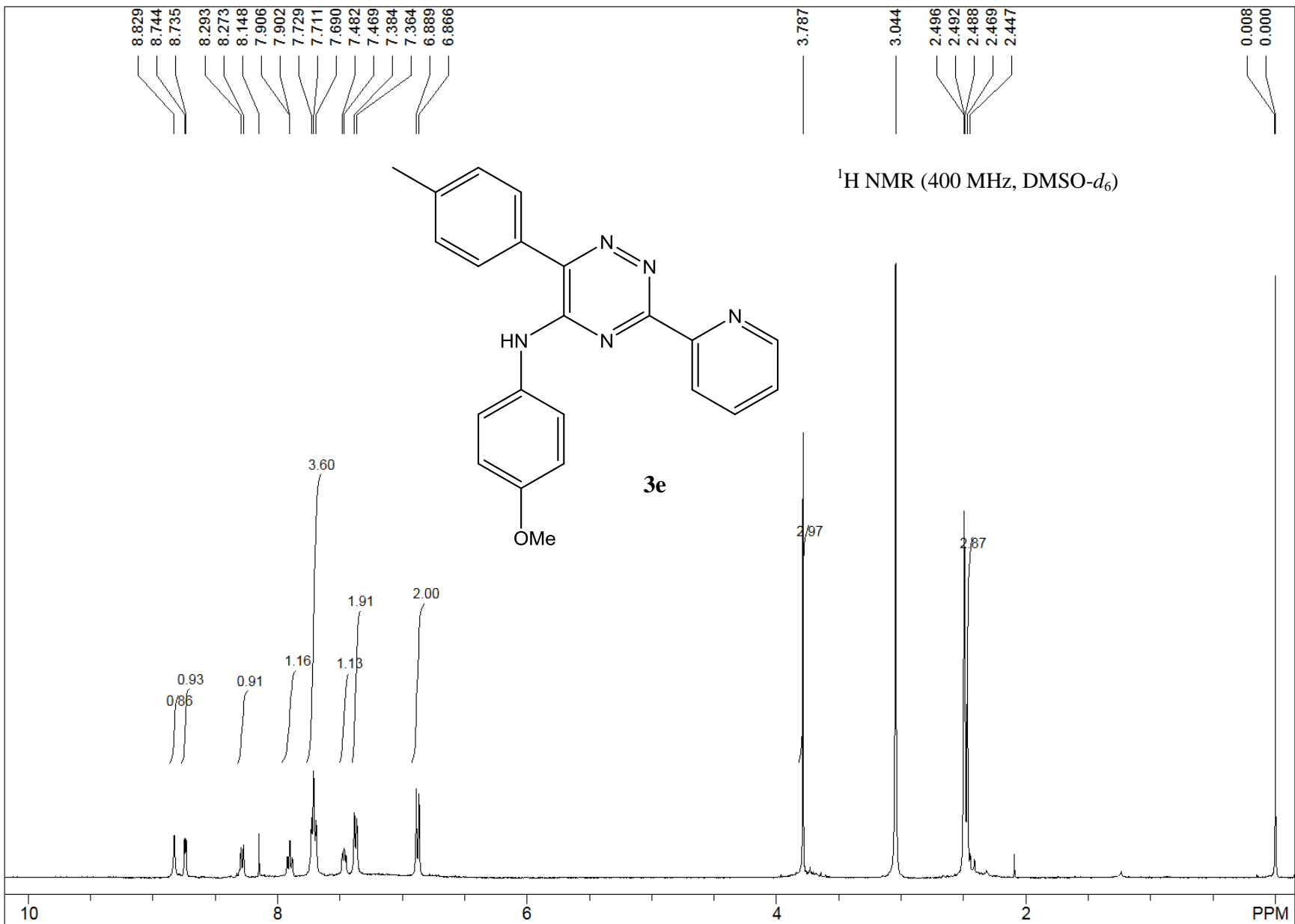




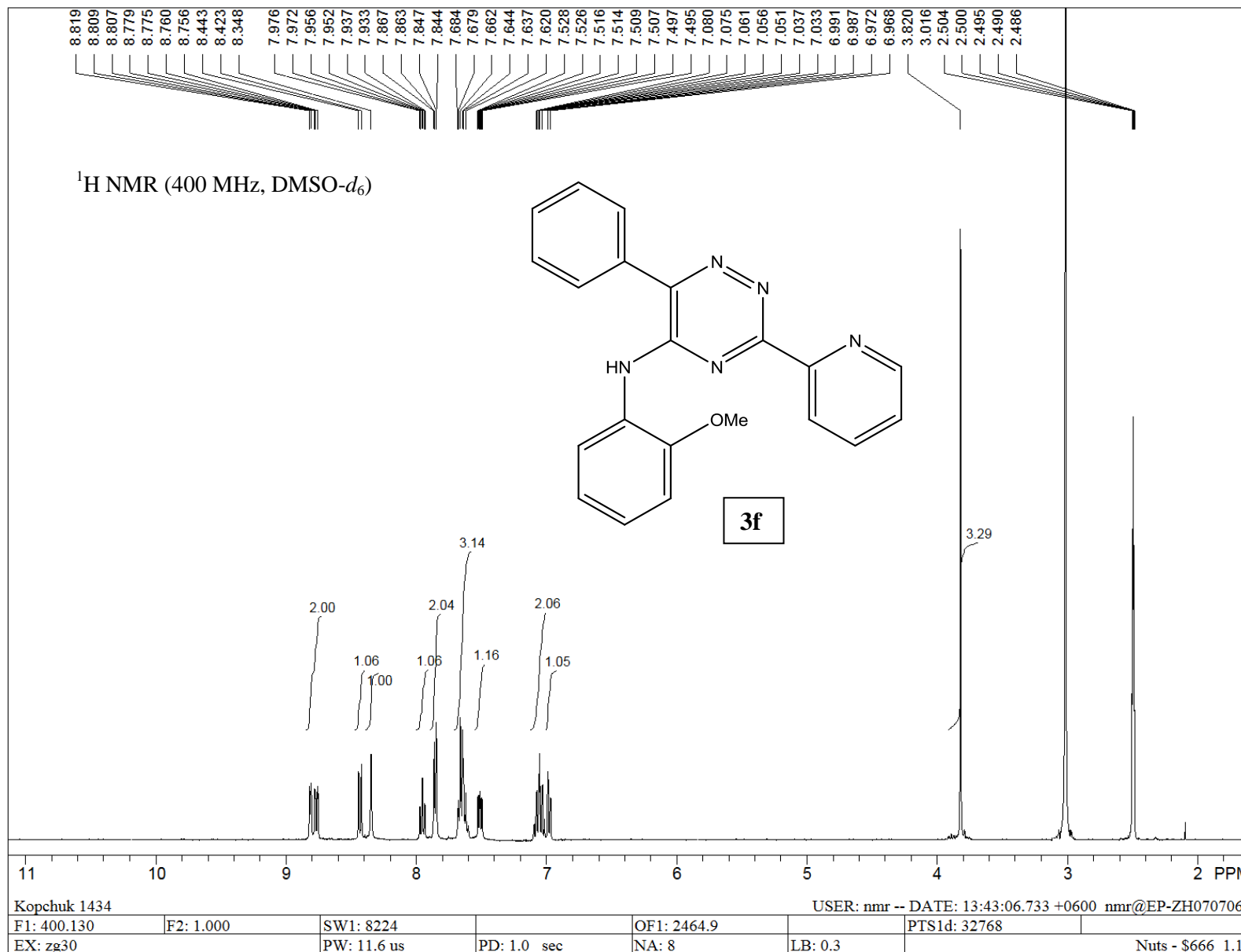


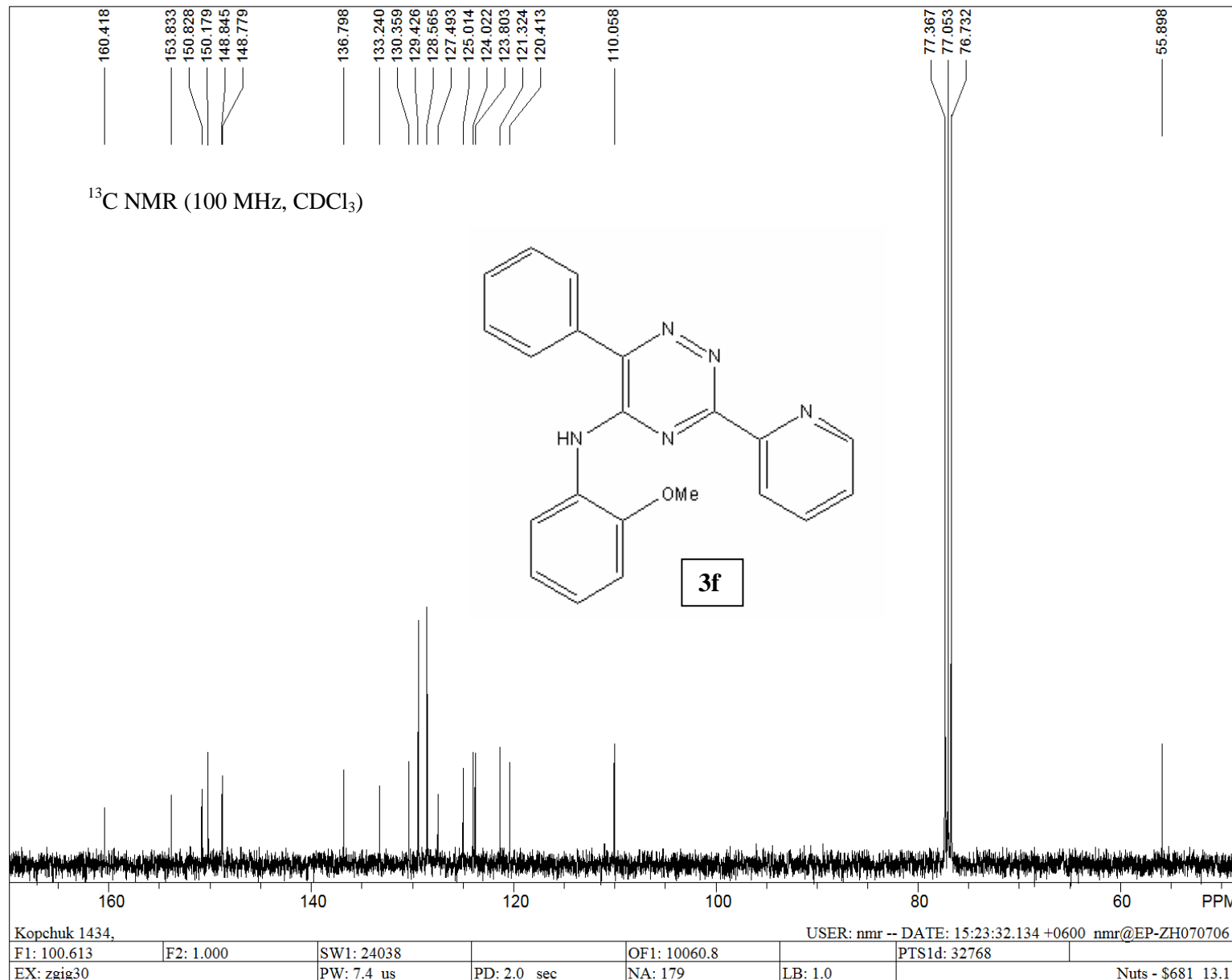


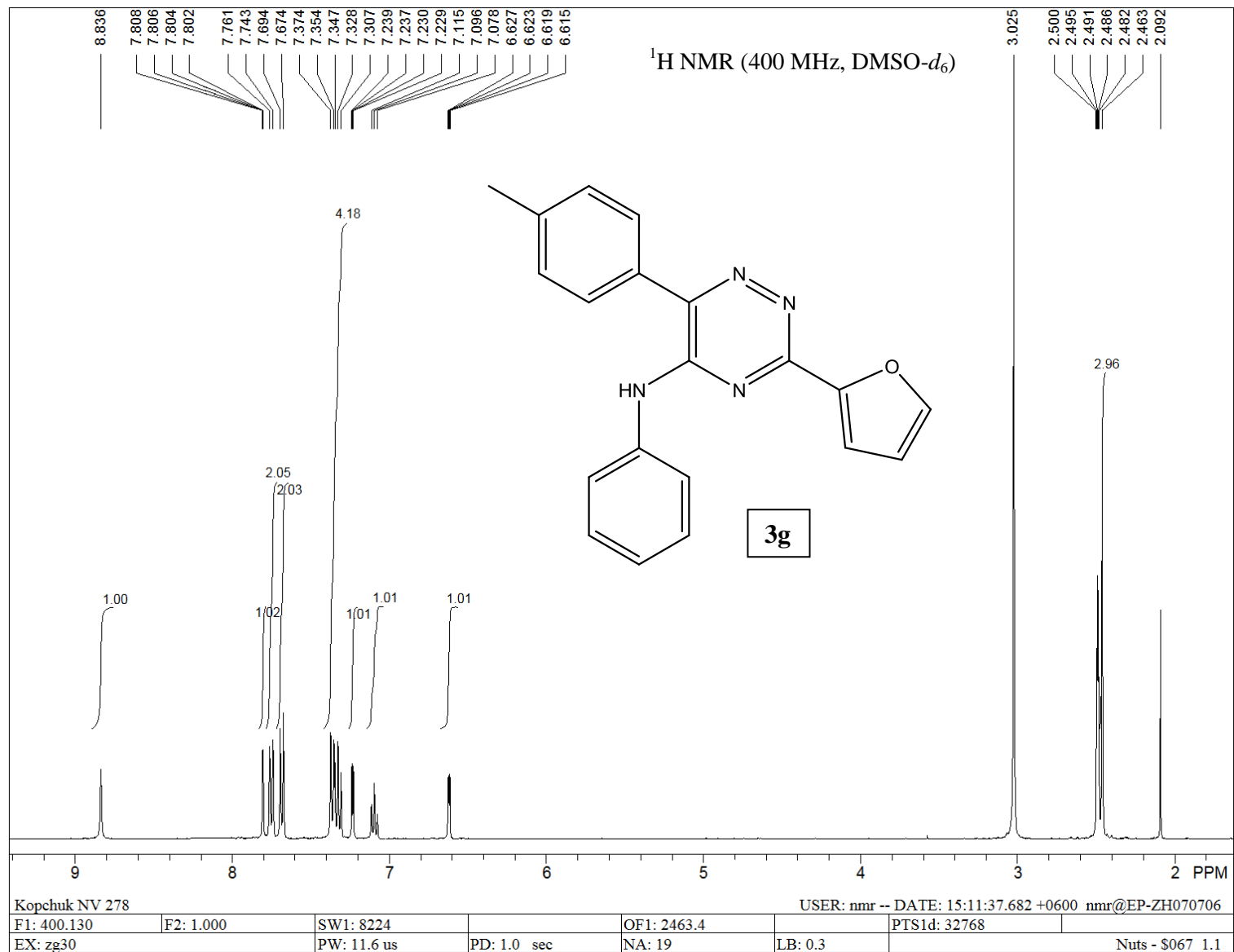


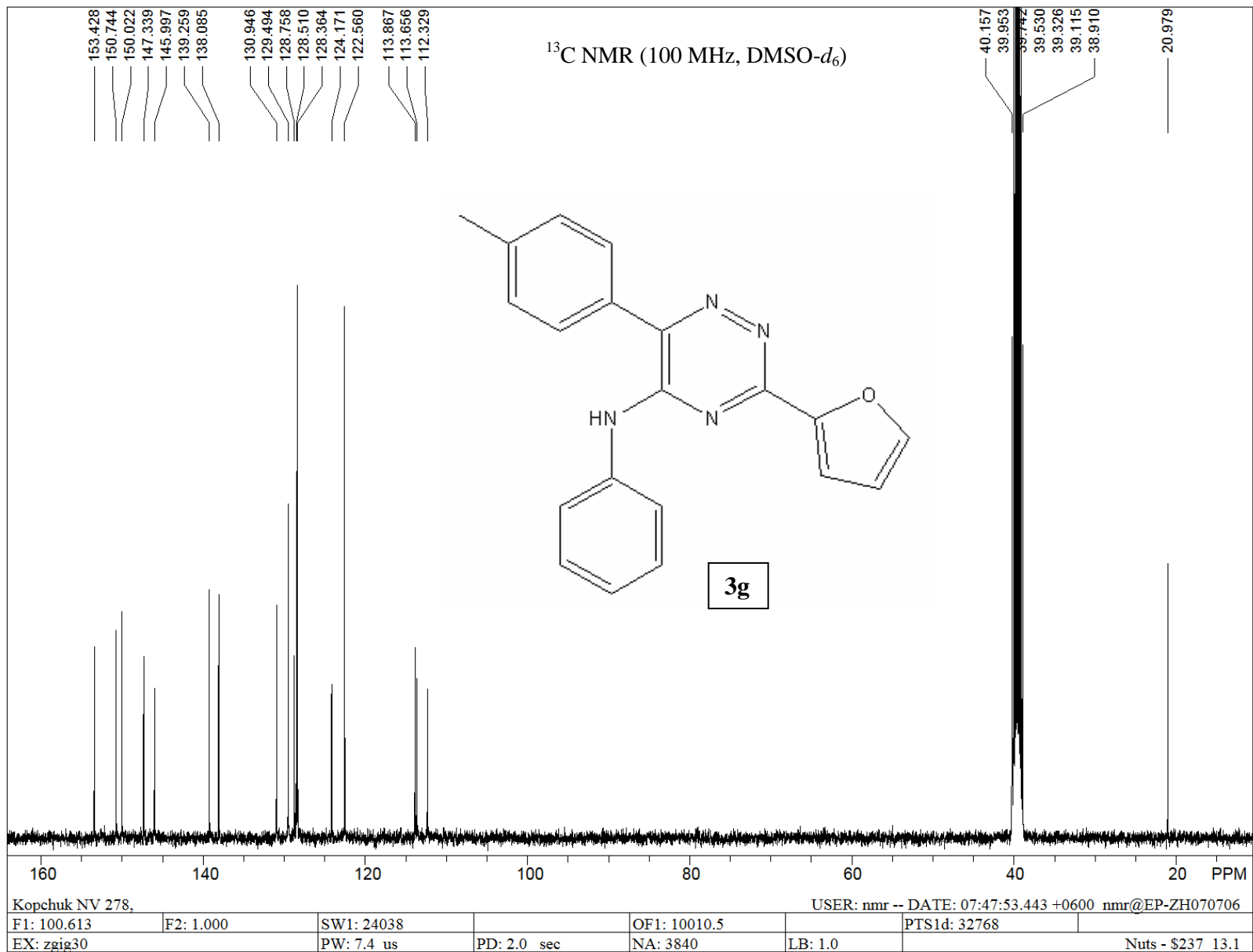


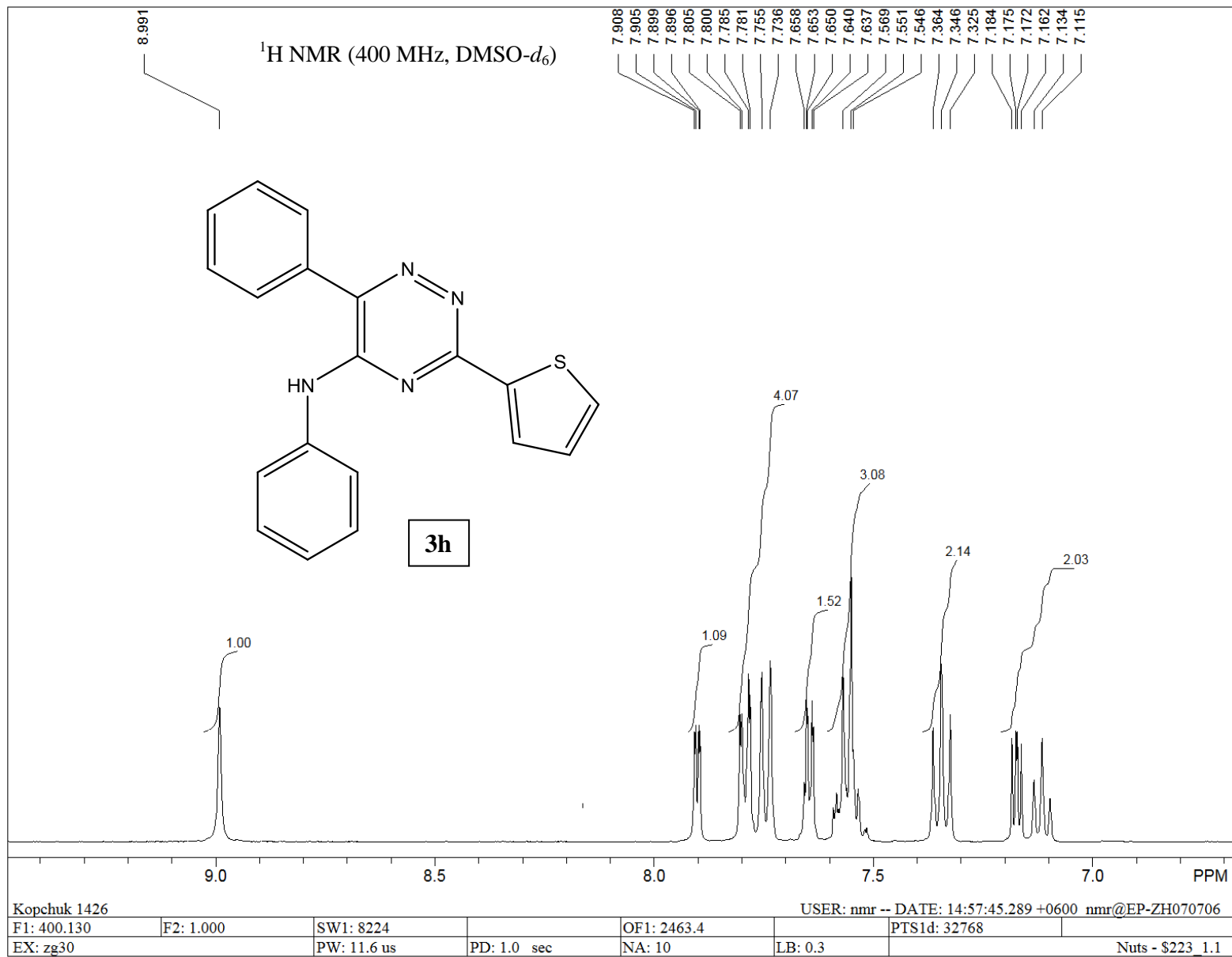
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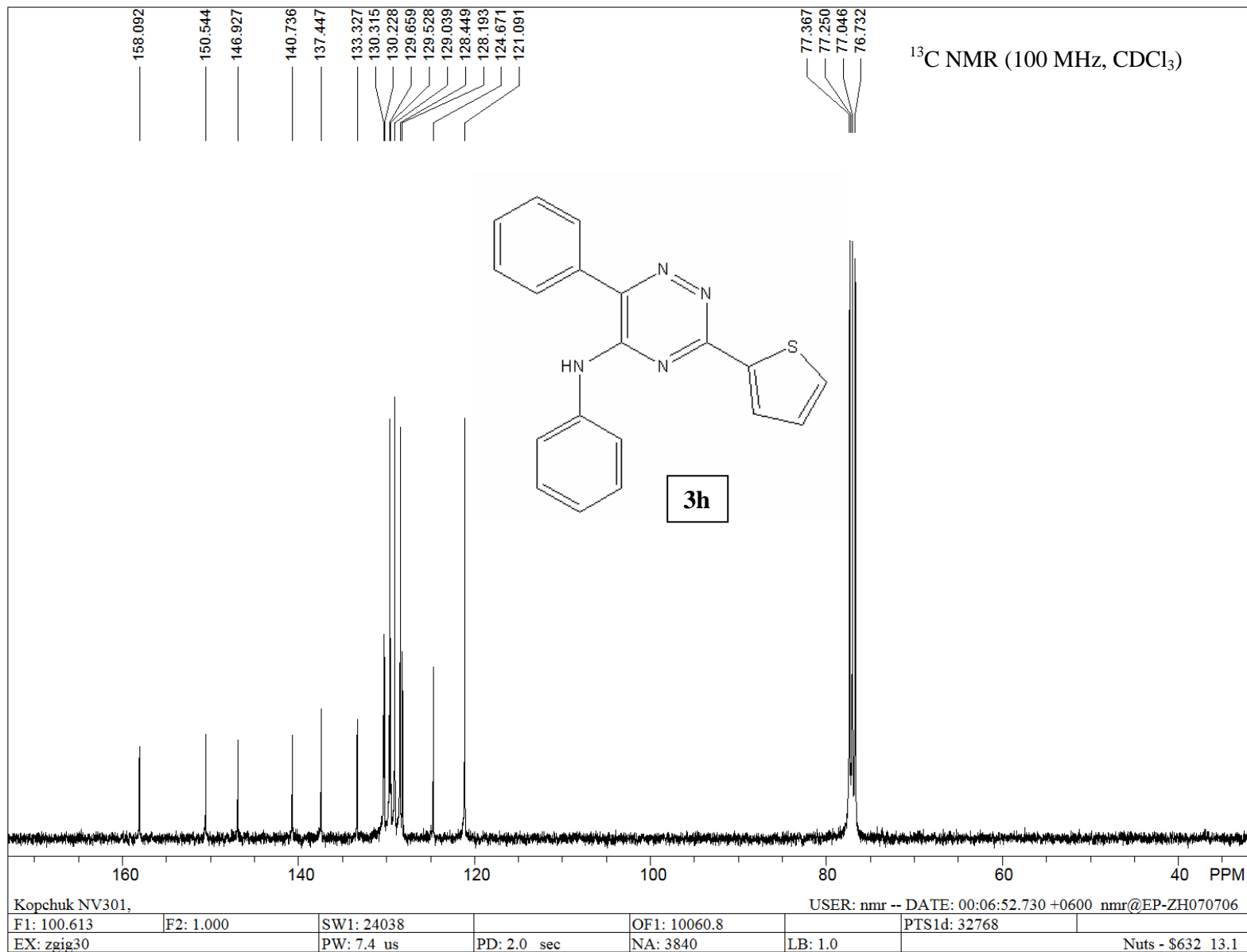


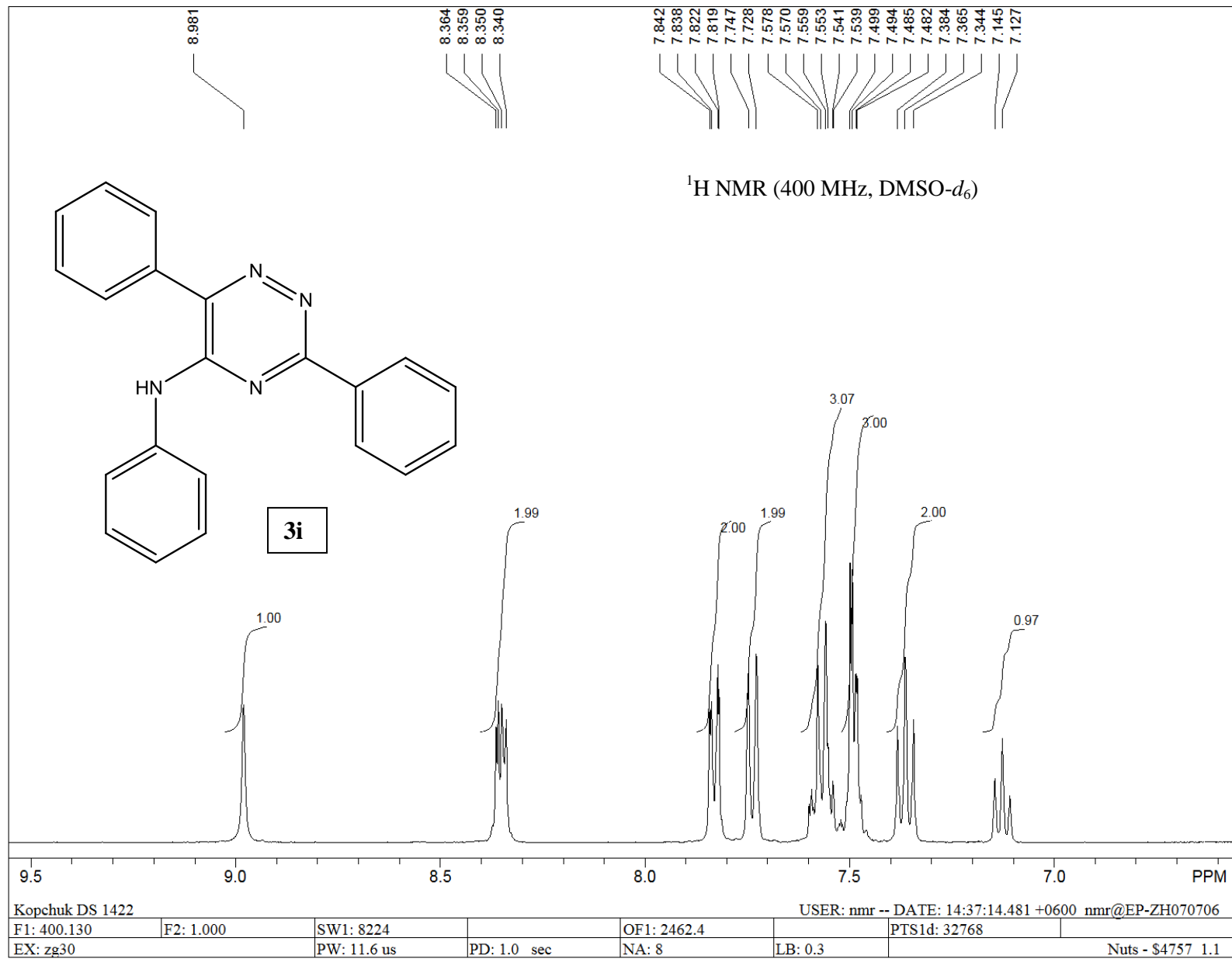


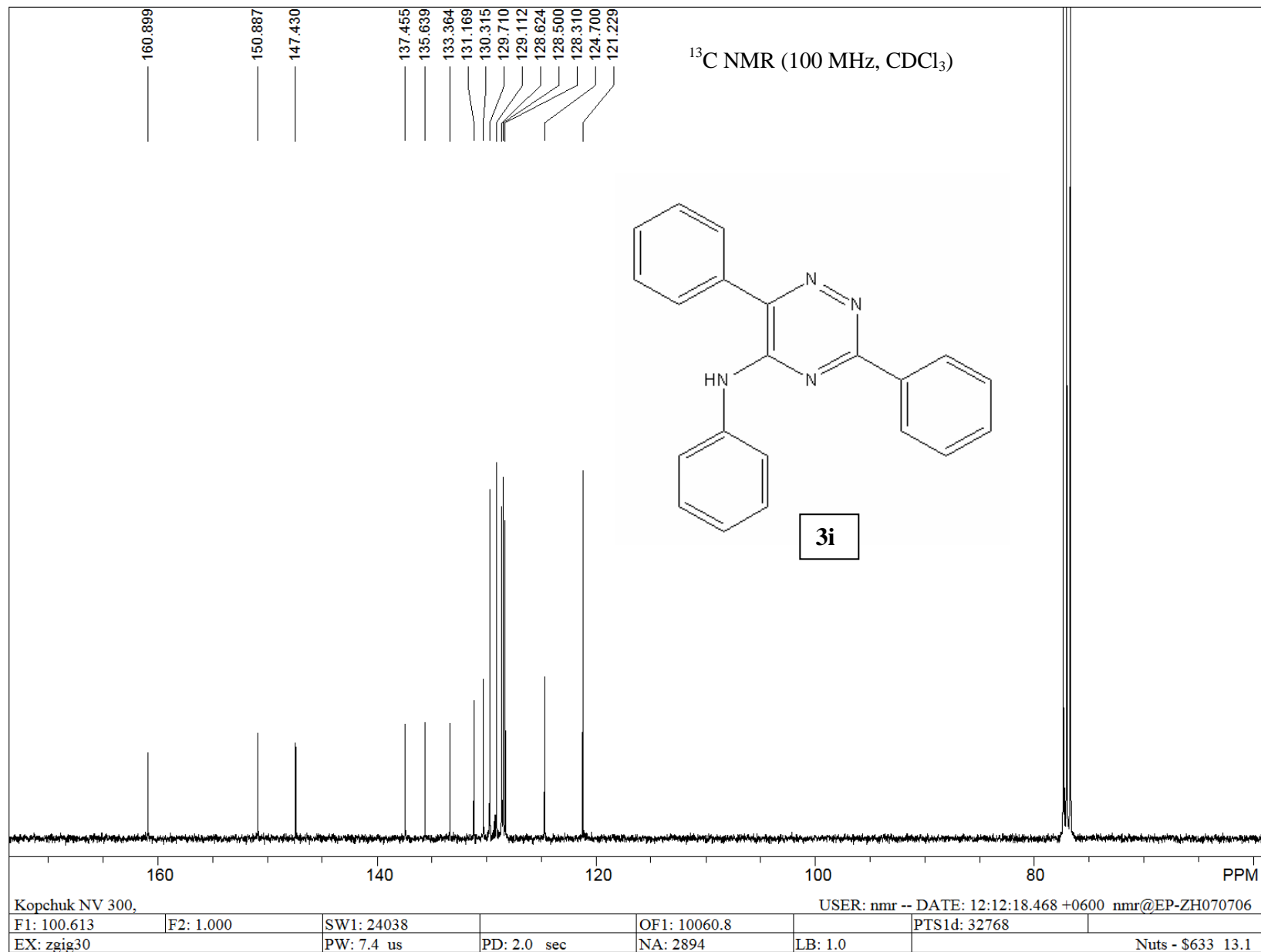


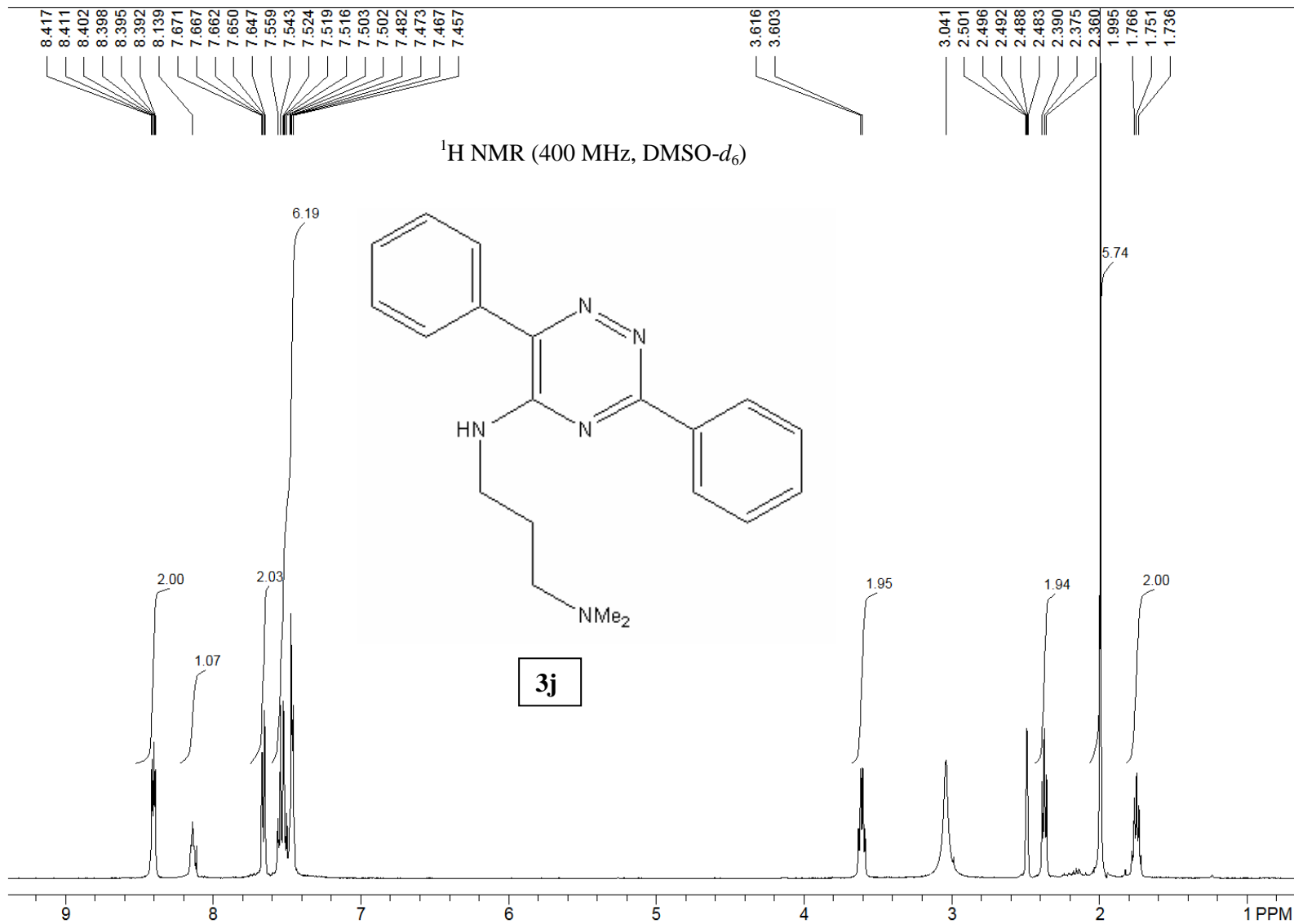












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