

## Pt/MnO<sub>2</sub> nanosheets: facile synthesis and highly efficient catalyst for ethylene oxidation at low temperature

Min Wang,<sup>a,b</sup> Lingxia Zhang,<sup>\*,a</sup> Weimin Huang,<sup>a</sup> Yajun Zhou,<sup>a,b</sup> Han Zhao,<sup>a,b</sup> Jian Lv,<sup>a,b</sup> Jianjian Tian<sup>a,c</sup>, Xiaotian Kan,<sup>a,b</sup> Jianlin Shi<sup>\*,a</sup>

<sup>a</sup> State Key Laboratory of High Performance Ceramics and Superfine Microstructure, Shanghai Institute of Ceramics, Chinese Academy of Sciences, 1295 Dingxi Road, Shanghai 200050, P. R. China

<sup>b</sup> University of Chinese Academy of Sciences, 19 Yuquan Road, Beijing 100049, P. R. China

<sup>c</sup> School of Materials Science and Engineering, Shanghai University, 99 Shangda Road, Shanghai 200444, P.R. China

Corresponding author. Tel.: +86 21 52412712; Fax: +86 21 52413122.  
E-mail address: jlshi@mail.sic.ac.cn (J.L.Shi).

Corresponding author. Tel.: +86 21 52412706; Fax: +86 21 52413122.  
E-mail address: zhlingxia@mail.sic.ac.cn (L.X.Zhang).

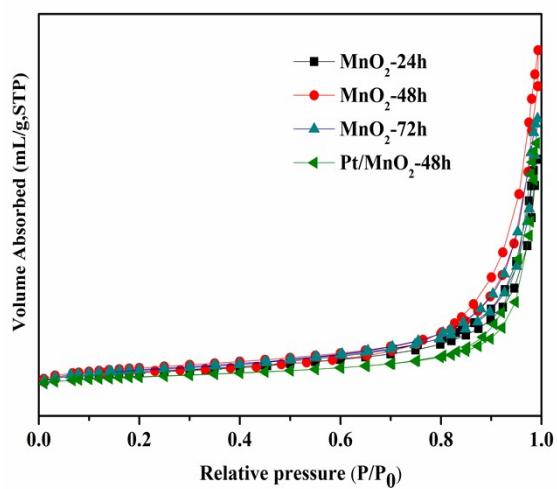


Fig S1. The  $\text{N}_2$  adsorption–desorption isotherms of the as-synthesized  $\text{MnO}_2$  samples and  $\text{Pt}/\text{MnO}_2\text{-}48\text{h}$ .

Table S1. Specific surface areas of the as-prepared MnO<sub>2</sub> samples and Pt/MnO<sub>2</sub>-48h.

Samples	MnO <sub>2</sub> -24h	MnO <sub>2</sub> -48h	MnO <sub>2</sub> -72h	Pt/MnO <sub>2</sub> -48h
S <sub>BET</sub> (m <sup>2</sup> . g <sup>-1</sup> )	157.5	189.1	175.9	117.9

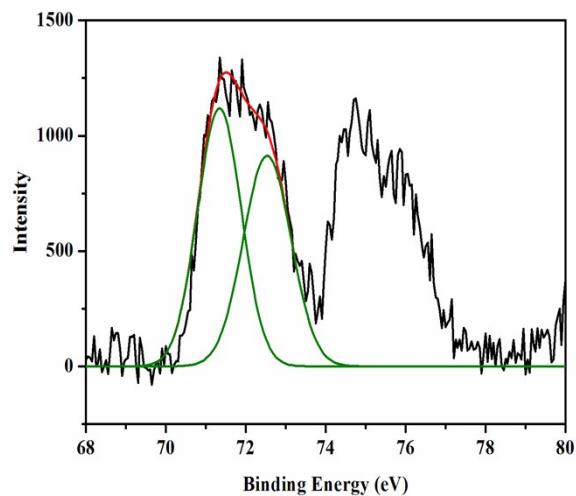


Fig S2. XPS Pt 4f spectrum of Pt/MnO<sub>2</sub>-48h.