Bio-inspired enantioselective full transamination with readily available cyclodextrin

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SUPPORTING INFORMATION

Table of Contents

General procedures
Optimization for PLP-catalyzed asymmetric transamination of phenylpyruvic acid in
e aqueous phaseS2
Synthesis of α-keto acids
Representative procedure for asymmetric transamination of α -keto acids
Table S7. Control experiments for PLP-catalyzed asymmetric transamination of
uorophenylpyruvic acidS13
Characterization data
1H NMR of the $\alpha\text{-keto}$ acids and derivatives of amino acidsS13
HPLC for the determination of enantiomeric excessess

1. General procedures

Chemicals: Solvents, inorganic salts, and organic reagents were purchased from commercial resources, and used without further purification unless otherwise mentioned. Column chromatography was performed on silica gel (200-300 mesh).

¹H NMR spectra were recorded on a 400 or 600 MHz NMR spectrometer. Chemical shifts were reported in ppm from tetramethylsilane with the solvent resonance as the internal (DMSO, δ =2.51). Spectra were reported as follows: chemical shift (δ ppm), multiplicity (s=single, d=doublet, t=triplet, q=quartet, m=multiplet), coupling constants (Hz), integration and assignment.

Enantiomeric excesses (ee) were determined by HPLC analysis using the corresponding commercial chiral column as stated in experimental procedures at 30°C with UV detector at 210 nm. CHIRAL PAK AD-H normal-phase analytical columns (4.6 mm Φ ×250mm), CHIRAL PAK OD-H normal-phase analytical columns (4.6 mm Φ ×250mm) and CHIRAL PAK IC normal-phase analytical columns (4.6 mm Φ ×250mm) were used as solid phase.

 α -Keto acids : phenylpyruvic acid, 4-hydroxy phenylpyruvic acid, benzoyl formic acid, pyruvic acid and 4-methyl-2-oxovaleric acid were purchased commercially. α-Keto acids including 2-methoxypyruricacid, 3-methoxypyruric acid. 4-methoxypyruric acid, 4-methylpyruric acid, 3,4-dimethoxypyruric acid, 4fluoropyruric acid, 4-chlorophenyl pyruric acid, 4-bromopyruric acid, and 2-oxo-3-thien-2-ylpropanoic acid were synthesized following a modified literature method¹, from the corresponding aldehydes condensed with hydantoin with ethanolamine as catalystand then hydrolyzed under alkaline conditions. 4-Nitrephenylpyruric acid was prepared by following literature procedure², from4-nitrobenzaldehyde and N-acetyl glycine in the presence of acetic anhydride and sodium acetate, followed by ring opening under reflux, and acetamide hydrolyzed off to give the final product.

2,2-Disubstituted glycines as sacrificial amine sources including methylphenylglycine and diphenylglycine were purchased commercially, and when equimolecular suspension of diphenylglycine and bases were added to the methanol solution, removed the solvent after stirring at room temperature, alkali metal salts of diphenylglycine including lithium, sodium and potassium were prepared.

2. Optimization for PLP-catalyzed asymmetric transamination of phenylpyruvic acidin pure aqueous phase

 Table S1. Additives and reaction time screening for PLP-catalyzed asymmetric

 transamination of phenylpyruvic acid^a

Ρ	Ph Ph Ph $-OH + H_2N Ph -COOH -3a 7a$	PLP β -CD, buffer(-CO	2),50 ℃ Ph Ph NH ₂ 4a	OH + Ph Ph 8
Entry	Additives	Time(h)	Yield(%) ^b	$ee(\%)^{c}$
1	None	24	7	12
2	EDTA	24	4	23
3	EDTA	12	2	21
4	EDTA	36	9	20
5	EDTA	48	9	20
6	EDTA	72	9	20
7	EDTA	96	9	20
8	EDTA	144	10	16

^aAll reactions were carried out with phenylpyruvic acid (**3a**) (0.05 m mol), diphenylglycine (**7a**) (0.05 m mol), EDTA (0.01mmol), β -CD (0.06mmol) and PLP (0.01 m mol) in 300mM Tris buffer (4.0 mL), pH 8.0, at 50 °C for certain time; ^b The yield was determined by chiral HPLC through standard addition method^{3, 4}:

The reaction liquid was divided into two parts, with the same subsequent operations except one was added 10% of the target product marked I , another marked II served as blank control ; results of HPLC showed that :A₁ (the peak areas of the target product in I), A_{II} (the peak areas of the target product in I), A_{II} (the peak areas of the target product in I), E_{II} (the ee's of the target product in II);

yield= $A_{II}/(A_1-A_{II}) \times 10\%$ or yield= $E_1/(E_{II}-E_1) \times 10\%$;

^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their N-Bz derivatives⁵, the absolute configuration of **4a** was assigned as *S* by comparison with the standard product.



Fig. S1^{\dagger} ¹H NMR contrast chart of benzophenone (top) and the reaction mixture's CH₂Cl₂ extract fraction (bottom)^a

Reaction was carried out with phenylpyruvic acid (**3a**) (0.5mmol), diphenylglycine (**7a**) (0.5mmol), β -CD (0.6mmol) and PLP (0.1 m mol) in 300mM Tris buffer (40.0 mL), pH 8.0, at 50 °C for 24h. The reaction mixture was extracted with CH₂Cl₂, removed the solvent by vacuum distillation obtaining the off-white solid, compared its ¹H NMR spectrum (bottom) with benzophenone (top) in CDCl₃ which proved benzophenone was produced as a byproduct.

Table S2.	Temperature	screening for P	PLP-catalyzed	asymmetric	transamination

Ph OH + H ₂ h O 3a	Ph PLP, N	EDTA	Ph H ₂ 4a	Ph Ph 8
Entry	Temperature(℃)	Yield(%) ^b	ee(%) ^c	
1	30	Trace		
2	40	2	25	

of phenylpyruvic acid^a

3	50	9	20
4	60	10	19
5	70	12	13
6	80	15	12
7	90	19	7

^aAll reactions were carried out with phenylpyruvic acid (**3a**) (0.05 m mol), diphenylglycine (**7a**) (0.05 m mol), β -CD (0.06mmol),EDTA (0.01mmol) and PLP (0.01 m mol) in Tris buffer (4.0 m L), pH 8.0, for 36 h if not mentioned..^b The yield was determined by chiral HPLC through standard addition method (Table S1[†]).^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their *N*-Bz derivatives, the absolute configuration of **4a** was assigned as *S* by comparison with the standard product.

Table S3. Screening of buffers for PLP-catalyzed asymmetric transaminat	ion of
phenylpyruvic acid ^a	

Ph	OH + H	Ph I ₂ NPh - COOH	PLP,EDTA → β -CD, buffer(-CO ₂), 50 °C	Ph (S	O OH + P	O h Ph
	3a	7a	/		4a	8
Entry	Buffer		Concentration (mM)	pН	Yield(%) ^b	ee(%) ^c
1	Tris buffer		300	8.0	9	20
2	Phosphate buffe	er (PB)	300	8.0	5	9
4	Borate buffer		300	8.0	trace	nd
5	Carbonic acidb	uffer	300	8.0	4	16
6	EDTA buffer		300	8.0	6	12
7	MOPS buffer		300	8.0	4	15
8	HEPES buffer		300	8.0	6	14
9	NaCl aqueous s	solution	300	8.0	trace	nd
10	Tris buffer		150	8.0	6	17
11	Tris buffer		600	8.0	9	21
12	Tris buffer		300	6.0	trace	nd
13	Tris buffer		300	6.5	trace	nd
14	Tris buffer		300	7.0	1	8
15	Tris buffer		300	7.5	3	17
16	Tris buffer		300	8.5	4	23
17	Tris buffer		300	9.0	3	24
18	Tris buffer		300	10.0	2	34

^aAll reactions were carried out with phenylpyruvic acid (**3a**) (0.05 mmol), diphenylglycine (**7a**) (0.05 mmol), β -CD (0.06mmol),EDTA (0.01mmol) and PLP (0.01 mmol) in various buffer (4.0 mL), at 50 °C for 36 h if not mentioned.^b The yield was determined by chiral HPLC through standard addition method(Table S1[†]).^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their *N*-Bz derivatives, the absolute configuration of **4a** was assigned as *S* by comparison with the standard product.

Table S4. Screening of catalysts for asymmetric transamination of phenylpyruvic acid^a



Entry	Catalyst	Dosage (m mol)	Yield(%) ^b	$ee(\%)^{c}$
1	PLP	0.01	9	20
2	PL^d	0.01	10	0
3	PM	0.01	10	0
4	PLP	0.005	5	25
5	PLP	0.015	9	17
6	PLP	0.025	11	15
7	PLP	0.05	10	11

^aAll reactions were carried out with phenylpyruvic acid (**3a**) (0.05 mmol), diphenylglycine (**7a**) (0.05 mmol), β -CD (0.06mmol), EDTA (0.01mmol) and in Tris buffer (4.0 mL), pH 8.0, at 50°C for 36 h if not mentioned..^b The yield was determined by chiral HPLC through standard addition method(Table S1†).^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their *N*-Bz derivatives, the absolute configuration of **4a** was assigned as *S* by comparison with the standard product. ^d PL was an abbreviation for pyridoxal.

Table S5. Screening of CDs for PLP-catalyzed asymmetric transamination	ı of
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Ph O	OH + $H_2N + Ph$ COOH	PLP,EDTA CD, buffer(-CO ₂),50	<mark>→</mark> Ph ́	$ \begin{array}{c} O \\ O \\ O \\ H \\ O \\ H_2 \end{array} O H + P \\ P$
3a	7a			4a 8
Entry	CD	Dosage (m mol)	Yield(%) ^b	$ee(\%)^{c}$
1	β -CD	0.06	9	20
2	a-CD	0.06	2	5
3	γ-CD	0.06	2	0
4	HP - β -CD	0.06	3	0
5	β -CD	0	4	0
6	β -CD	0.015	6	7
7	β -CD	0.03	7	17
8	β -CD	0.12	7	25
9	β -CD	0.24	5	27

phenylpyruvic acid^a

^aAll reactions were carried out with phenylpyruvic acid (**3a**) (0.05 mmol), diphenylglycine (**7a**) (0.05 mmol), EDTA (0.01 mmol) and PLP (0.01 mmol) in Tris buffer (4.0 mL), pH 8.0, at 50 °C for 36 h if not mentioned..^b The yield was determined by chiral HPLC through standard addition method(Table S1[†]).^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their *N*-Bz derivatives, the absolute configuration of **4a** was assigned as *S* by comparison with the standard product.

Table S6. Screening of sacrificial amine sources for PLP-catalyzed asymmetric

transamination of phenylpyruvic acid^a

Ph	$\begin{array}{c} O \\ H \\ O \\ O \\ \mathbf{3a} \\ 7 \end{array} \xrightarrow{Ph} \\ Ph \\ Ph \\ R_2 \\ COOM \\ 7 \end{array}$	PLP, EDTA ³ -CD, buffer(-CO₂),50 ℃		$\begin{array}{c} O \\ H_2 \\ H_2 \\ H_3 \\ H_3 \\ H_3 \\ H_3 \\ H_3 \\ H_3 \\ H_2 \\ H_2 \\ H_3 \\ H_2 \\ H$
Entry	Amine source	Dosage (m mol)	Yield(%) ^b	ee(%) ^c
1	7a(R2:Ph, M:H)	0.05	9	20
2	7b(R2:Me, M:H)	0.05	11	19
3	7c (R2:Ph, M:Li ⁺)	0.05	23	20

4	7d (R2:Ph, M:Na ⁺)	0.05	27	20
5	$7e(R2:Ph, M:K^+)$	0.05	24	20
6	7d	0.1	27	20
7	7d	0.2	28	20
8	7d	0.3	27	20
9	7d	0.4	25	20

^aAll reactions were carried out with phenylpyruvic acid (**3a**) (0.05 mmol), β -CD (0.06mmol),EDTA (0.01mmol) and in Tris buffer (4.0 mL), pH 8.0, at 50 °C for 36 h if not mentioned..^b The yield was determined by chiral HPLC through standard addition method(Table S1[†]).^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their *N*-Bz derivatives, the absolute configuration of **4a** was assigned as *S* by comparison with the standard product.

3. Synthesis of α -keto acids



A. A reaction tube was charged with 3 m mol of corresponding aldehydes,3 m mol of hydantoin, and 0.03 m mol of ethanolamine in 7 g of permuted water , under agitation and in an inert atmosphere, heated for 2 to 4 hours with reflux until a large amount of solid was produced turn the clear solution into suspension. Thereafter, 30ml of 6g sodium hydroxyde or 8.5g potassium hydroxide was introduced, continued to reflux for more than another 30 minutes until the muddy liquid turned back to clear. The obtained solution was then cooled to the ambient temperature, brought to acidity with the addition of concentrated hydrochloric acid and extracted with ethyl acetate, the organic phase was washed with iced water, dried with anhydrous Na_2SO_4 , evaporated at reduced pressure to get the crude product which was purified by flash

column or recrystallization.



¹H NMR (400 MHz, DMSO) δ 13.01 (s, 1H), 8.96 (s, 1H), 7.71 (d, *J* = 8.8 Hz, 2H), 6.92 (d, *J* = 8.9 Hz, 2H), 6.38 (s, 1H), 3.76 (s, 3H).



¹H NMR (400 MHz, DMSO) δ 13.18 (s, 1H), 9.26 (s, 1H), 7.37 (s, 1H), 7.32 (d, *J* = 7.8 Hz, 1H), 7.25 (t, *J* = 7.9 Hz, 1H), 6.84 – 6.79 (m, 1H), 6.38 (s, 1H), 3.74 (s, 3H).



¹H NMR (400 MHz, DMSO) δ 13.14 (s, 1H), 9.13 (s, 1H), 8.17 (dd, *J* = 7.8, 1.6 Hz, 1H), 7.25 – 7.19 (m, 1H), 7.03 – 6.93 (m, 2H), 6.80 (s, 1H), 3.81 (s, 3H).



¹H NMR (400 MHz, DMSO) δ 13.11 (s, 1H), 9.09 (s, 1H), 7.64 (d, *J* = 8.1 Hz, 2H), 7.15 (d, *J* = 8.1 Hz, 2H), 6.37 (s, 1H), 2.29 (s, 3H).



¹H NMR (400 MHz, DMSO) δ 13.00 (s, 1H), 8.98 (s, 1H), 7.43 (d, J = 1.7 Hz, 1H), 7.31 (dd, J = 8.4, 1.8 Hz, 1H), 6.94 (d, J = 8.5 Hz, 1H), 6.38 (s, 1H), 3.76 (s, 3H), 3.74 (s, 3H).





¹H NMR (400 MHz, DMSO) δ 13.25 (s, 1H), 9.29 (s, 1H), 7.81 (dd, *J* = 8.5, 5.9 Hz, 2H), 7.17 (t, *J* = 8.9 Hz, 2H), 6.41 (s, 1H).



3j

¹H NMR (400 MHz, DMSO) δ 13.30 (s, 1H), 9.49 (s, 1H), 7.77 (d, *J* = 8.5 Hz, 2H), 7.40 (d, *J* = 8.5 Hz, 2H), 6.39 (s, 1H).



¹H NMR (400 MHz, DMSO) δ 13.29 (s, 1H), 9.49 (s, 1H), 7.71 (d, *J* = 8.5 Hz, 2H), 7.53 (d, *J* = 8.4 Hz, 2H), 6.37 (s, 1H).



¹H NMR (400 MHz, DMSO) δ 13.08 (s, 1H), 9.51 (s, 1H), 7.54 (d, *J* = 5.1 Hz, 1H), 7.26 (d, *J* = 3.4 Hz, 1H), 7.05 (dd, *J* = 5.0, 3.8 Hz, 1H), 6.78 (s, 1H).



B. A reaction tube was charged with 3 mmol of 4-nitrobenzaldehyde, 15 mmol of anhydrous sodium acetate, and 3.6 mmol of *N*- acetylglycine, the mixture was heated at reflux in 3ml acetic anhydride for 10 hours, this was then cooled to the ambient temperature. After standing overnight in refrigerator, the precipitate was collected on a filter, washed with cold water and ethanol, then dried in vacuo to afford 0.6 g (91%) of brown solid. Following that, 4.5 ml acetone and 3ml H₂O were added to the product obtained from the previous step, heated at reflux for 10 hours, then cooled to the ambient temperature, after removal of the solvent under reduced pressure, 15 ml of 1M HCl solution was added, kept refluxing at 120° C for 10 hours. Whereafter, the solution cooled down and crystallized out brown crystals obtaining by filtering,

and purified by recrystallization yielding 0.1g 4-nitrephenylpyruric acidas the final product (36%).

511

¹H NMR (400 MHz, DMSO) δ 13.58 (s, 1H), 10.17 (s, 1H), 8.20 (d, *J* = 8.7 Hz, 2H), 7.99 (d, *J* = 8.7 Hz, 2H), 6.50 (s, 1H).

4. Representative procedure for asymmetric transamination of α -keto acids

Table 2 Substrate screening for the asymmetric transamination of α -keto acids



The general transamination reaction condition: *a*-keto acid **3** (0.05 mmol), amine source **7d** (0.05 mmol), PLP(0.01 mmol), EDTA (0.01 mmol) and β -CD (0.06 mmol) were added into 4mL 300mM Tris buffer, pH=8. The mixture was stirred at 50°C for

36h or 72h. Followed by *N*-derivatization of the amino acid as the literature procedure shown⁵, and the ee's were determined by chiral HPLC analysis.



¹H NMR (600 MHz, DMSO) δ 12.76 (s, 1H), 8.69 (d, *J* = 8.1 Hz, 1H), 7.83 – 7.76 (m, 2H), 7.52 (t, *J* = 7.4 Hz, 1H), 7.44 (t, *J* = 7.6 Hz, 2H), 7.32 (d, *J* = 7.6 Hz, 2H), 7.26 (t, *J* = 7.6 Hz, 2H), 7.18 (t, *J* = 7.3 Hz, 1H), 4.63 (ddd, *J* = 10.8, 8.3, 4.4 Hz, 1H), 3.20 (dd, *J* = 13.8, 4.3 Hz, 1H), 3.08 (dd, *J* = 13.7, 10.8 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 12.72 (s, 1H), 7.89 (d, *J* = 7.5 Hz, 2H), 7.70 (d, *J* = 8.5 Hz, 1H), 7.65 (t, *J* = 6.9 Hz, 2H), 7.41 (t, *J* = 7.4 Hz, 2H), 7.31 (dt, *J* = 9.9, 7.5 Hz, 2H), 7.19 (d, *J* = 8.5 Hz, 2H), 6.83 (d, *J* = 8.6 Hz, 2H), 4.26 - 4.14 (m, 3H), 4.14 - 4.05 (m, 1H), 3.70 (s, 3H), 3.01 (dd, *J* = 13.8, 4.3 Hz, 1H), 2.80 (dd, *J* = 13.7, 10.6 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 12.78 (s, 1H), 7.88 (d, *J* = 7.5 Hz, 2H), 7.74 (d, *J* = 8.4 Hz, 1H), 7.64 (t, *J* = 6.7 Hz, 2H), 7.40 (dd, *J* = 10.0, 4.5 Hz, 2H), 7.29 (dd, *J* = 7.0, 4.8 Hz, 2H), 7.18 (t, *J* = 7.9 Hz, 1H), 6.93 - 6.81 (m, 2H), 6.81 - 6.71 (m, 1H), 4.28 - 4.07 (m, 4H), 3.71 (s, 3H), 3.06 (dd, *J* = 13.7, 4.1 Hz, 1H), 2.85 (dd, *J* = 13.5, 10.8 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 12.63 (s, 1H), 8.58 (d, *J* = 8.1 Hz, 1H), 7.77 (d, *J* = 7.2 Hz, 2H), 7.51 (t, *J* = 4.8 Hz, 1H), 7.44 (t, *J* = 7.4 Hz, 2H), 7.23 (d, *J* = 7.4 Hz, 1H), 7.17 (dd, *J* = 11.5, 4.0 Hz, 1H), 6.95 (d, *J* = 8.2 Hz, 1H), 6.82 (t, *J* = 7.4 Hz, 1H), 4.65 (ddd, *J* = 10.5, 8.2, 4.5 Hz, 1H), 3.81 (s, 3H), 3.29 – 3.22 (m, 1H), 2.95 (dd, *J* = 13.5, 10.6 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 8.19 (s, 1H), 7.75 (d, *J* = 7.3 Hz, 2H), 7.50 (t, *J* = 7.2 Hz, 1H), 7.43 (t, *J* = 7.4 Hz, 2H), 7.13 (d, *J* = 7.7 Hz, 2H), 7.00 (d, *J* = 7.7 Hz, 2H), 4.45 (s, 1H), 3.18 (dd, *J* = 13.3, 4.1 Hz, 1H), 3.02 (dd, *J* = 13.4, 8.7 Hz, 1H), 2.20 (s, 3H).



¹H NMR (400 MHz, DMSO) δ 9.19 (s, 1H), 7.88 (d, *J* = 7.4 Hz, 2H), 7.75 – 7.57 (m, 2H), 7.41 (td, *J* = 7.1, 3.2 Hz, 2H), 7.31 (dd, *J* = 16.2, 7.9 Hz, 2H), 7.01 (d, *J* = 8.1 Hz, 2H), 6.63 (d, *J* = 8.1 Hz, 2H), 4.25 (dd, *J* = 13.2, 8.0 Hz, 1H), 4.22 – 4.08 (m, 2H), 4.00 (dd, *J* = 18.4, 6.2 Hz, 1H), 2.98 (dd, *J* = 13.8, 3.3 Hz, 1H), 2.75 (dd, *J* = 13.6, 9.5 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 12.70 (s, 1H), 7.88 (d, *J* = 7.5 Hz, 2H), 7.67 (dd, *J* = 21.3, 7.8 Hz, 3H), 7.41 (t, *J* = 7.2 Hz, 2H), 7.36 - 7.17 (m, 2H), 6.92 (s, 1H), 6.80 (dd, *J* = 21.1, 8.1 Hz, 2H), 4.34 - 4.00 (m, 4H), 3.70 (d, *J* = 8.9 Hz, 6H), 3.01 (dd, *J* = 13.6, 4.1 Hz, 1H), 2.79 (dd, *J* = 13.7, 10.8 Hz, 1H).



¹H NMR (600 MHz, DMSO) δ 8.13 (d, J = 8.3 Hz, 2H), 7.86 (d, J = 7.4 Hz, 2H), 7.76 (d, J = 8.5 Hz, 1H), 7.67 – 7.57 (m, 2H), 7.54 (d, J = 8.3 Hz, 2H), 7.39 (t, J = 7.4 Hz, 2H), 7.28 (dd, J = 16.2, 7.8 Hz, 2H), 4.30 – 4.24 (m, 1H), 4.21 (dd, J = 12.2, 9.2 Hz, 2H), 4.15 (t, J = 6.7 Hz, 1H), 3.25 – 3.22 (m, 1H), 3.05 – 2.96 (m, 1H).



¹H NMR (400 MHz, DMSO) δ 12.79 (s, 1H), 8.72 (d, J = 8.2 Hz, 1H), 7.79 (d, J = 7.2 Hz, 2H), 7.52 (t, J = 7.3 Hz, 1H), 7.45 (t, J = 7.4 Hz, 2H), 7.35 (dd, J = 8.4, 5.7 Hz, 2H), 7.09 (t, J = 8.8 Hz, 2H), 4.60 (ddd, J = 10.9, 8.3, 4.4 Hz, 1H), 3.18 (dd, J = 13.8, 4.3 Hz, 1H), 3.05 (dd, J = 13.6, 11.0 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 12.87 (s, 1H), 7.88 (d, *J* = 7.5 Hz, 2H), 7.75 (d, *J* = 8.5 Hz, 1H), 7.69 – 7.56 (m, 2H), 7.41 (t, *J* = 7.3 Hz, 2H), 7.35 – 7.25 (m, 6H), 4.17 (dt, *J* = 12.5, 4.9 Hz, 4H), 3.08 (dd, *J* = 13.7, 4.1 Hz, 1H), 2.86 (dd, *J* = 13.4, 11.1 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 8.74 (d, *J* = 8.1 Hz, 1H), 7.79 (d, *J* = 7.6 Hz, 2H), 7.52 (t, *J* = 7.3 Hz, 1H), 7.45 (dd, *J* = 7.7, 6.1 Hz, 4H), 7.28 (d, *J* = 8.2 Hz, 2H), 4.66 – 4.55 (m, 1H), 3.17 (dd, *J* = 13.8, 4.3 Hz, 1H), 3.08 – 2.96 (dd, *J* = 13.8, 8.9 Hz, 1H).



¹H NMR (400 MHz, DMSO) δ 12.85 (s, 1H), 8.76 (d, J = 16.2 Hz, 1H), 7.85 (d, J = 7.3 Hz, 2H), 7.51 (dt, J = 27.5, 7.2 Hz, 3H), 7.32 (d, J = 4.8 Hz, 1H), 6.93 (dd, J = 11.8, 7.0 Hz, 2H), 4.69 – 4.47 (m, 1H), 3.48 – 3.29 (m, 2H).



4m

¹H NMR (400 MHz, DMSO) δ 12.87 (s, 1H), 7.88 (d, *J* = 7.5 Hz, 2H), 7.75 (d, *J* = 8.5 Hz, 1H), 7.69 – 7.57 (m, 2H), 7.41 (t, *J* = 7.3 Hz, 2H), 7.33 – 7.26 (m, 5H), 4.17 (dt, *J* = 12.5, 4.9 Hz, 4H), 3.08 (dd, *J* = 13.7, 4.1 Hz, 1H), 2.86 (dd, *J* = 13.4, 11.1 Hz, 1H).



4n

¹H NMR (400 MHz, DMSO) δ 12.52 (s, 1H), 8.65 (d, *J* = 7.1 Hz, 1H), 7.88 (d, *J* = 7.2 Hz, 2H), 7.54 (t, *J* = 7.3 Hz, 1H), 7.47 (t, *J* = 7.4 Hz, 2H), 4.42 (p, *J* = 7.4 Hz, 1H), 1.39 (d, *J* = 7.3 Hz, 3H).



¹H NMR (400 MHz, DMSO) δ 12.53 (s, 1H), 8.57 (d, *J* = 7.9 Hz, 1H), 7.89 (d, *J* = 7.3 Hz, 2H), 7.54 (t, *J* = 7.2 Hz, 1H), 7.47 (t, *J* = 7.3 Hz, 2H), 4.59 – 4.33 (m, 1H), 1.91 – 1.63 (m, 2H), 1.59 (ddd, *J* = 12.7, 8.8, 4.2 Hz, 1H), 0.90 (dd, *J* = 17.1, 6.3 Hz, 6H).

F	$ \begin{array}{c} $	PLP,EDTA β -CD, buffer(-CO ₂),50 °C	F 4i	• Ph Ph 8
entry	condition	Yield(%) ^b	ee(%) ^c	
1	one-pot method ^a	37	35	
2	$3i + \beta - CD^d$	39	44	
3	$7d+\beta$ -CD ^e	36	35	
4	$3i+\beta$ -CD(2.4eq) ^f	38	50	

5. Table S7. Control experiments for PLP-catalyzed asymmetric transamination of 4-fluorophenylpyruvic acid^a

^aAll reactions were carried out with 4-fluorophenylpyruric acid (**3i**) (0.05 m mol), β -CD (0.06 m mol), EDTA (0.01 m mol) and in Tris buffer (4.0 m L), pH 8.0, at 50 °C for 48 h if not mentioned. ^b The yield was determined by chiral HPLC through standard addition method(Table S1†).^c The ee's were determined by chiral HPLC (Chiralpak AD-H column) after the amines were converted into their *N*-Bz derivatives, the absolute configuration of **4i** was assigned as *S* by comparison with the standard product. ^d **3i**, β -CD, EDTA and Tris buffer solution were stirred overnight at 50 °C to get the resulting mixtures, then added **7d** and PLP keeping on stirring for another 48h. ^e**7d**, β -CD, EDTA and Tris buffer solution were stirred overnight at 50 °C to get the resulting mixtures, then added **3i** and PLP keeping on stirring for another 48h. ^f The same as ^d except dosage of β -CD was 2.4 eq.(0.12mmol).

6. Characterization data

¹H NMR of the α -keto acids and derivatives of amino acids

¹H NMR of the α-keto acids











S17



¹H NMR of the α -amino acid derivatives









S21







S24



HPLC for the determination of enantiomeric excessess

Table 3, compound 4a



HPLC Conditions:Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(14/1) containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	19.470	47234600	49.025
2	22.490	49113239	50.975



Peak#	RT(min)	Area	Aera%
1	19.180	34667956	60.861
2	22.655	22294653	39.139

Table 2, compound 4b



4b

HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(10:1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	24.052	32654064	49.814
2	25.940	32898368	50.186



Peak#	RT(min)	Area	Aera%
1	23.875	6560276	30.889
2	26.035	14677691	69.111

Table 2, compound 4c



HPLC Conditions: Column: Chiralcel OD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA (14/1) containing 0.3% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	46.767	63690881	49.722
2	63.597	64403855	50.278



Peak#	RT(min)	Area	Aera%
1	47.378	27909618	35.787
2	63.822	50078348	64.213

Table 2, compound 4d



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA (8/1) containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	19.020	3750251	48.362
2	24.248	3859756	49.775

Table 2, compound 4e



4e

HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(10:1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



The Reaction Result



Peak#	RT(min)	Area	Aera%
1	20.228	12598767	72.052
2	22.650	4886826	27.948

Table 2, compound 4f



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(14/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	60.205	95597538	50.084
2	68.530	95276723	49.916

Table 2, compound 4g



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(14/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm





Peak#	RT(min)	Area	Aera%
1	53.343	50882363	49.896
2	65.230	51094863	50.104

Table 2, compound 4h



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(7/3)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic





Peak#	RT(min)	Area	Aera%
1	6.500	2304690	42.022
2	7.722	3179808	57.978

Table 2, compound 4i



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(5.4/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection:

UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	8.048	21053054	49.609
2	9.430	21385147	50.391



Peak#	RT(min)	Area	Aera%
1	8.538	6493229	71.988
2	10.082	2526666	28.012



4j

HPLC Conditions: Column: Chiralcel IC, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(14/1)containing 0.1% TFA;Flow rate: 0.8 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	19.743	30332380	50.175
2	24.612	30120286	49.825



Peak#	RT(min)	Area	Aera%
1	19.677	12624137	65.249
2	24.122	6723489	34.751



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(5.4/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



Peak#	RT(min)	Area	Aera%
1	8.823	17498240	50.098
2	10.940	17429685	49.902



Peak#	RT(min)	Area	Aera%
1	8.827	7777919	73.019
2	11.072	2874055	26.981



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(10/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic





Peak#	RT(min)	Area	Aera%
1	15.003	1564163	56.912
2	17.357	1184238	43.088

Table 2, compound 4m



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(14/1) containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic





Peak#	RT(min)	Area	Aera%
1	4318863	3827439	52.460
2	3468480	3468480	47.540

Table 2, compound 4n



HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(25/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection:

UV 210 nm

Racemic







HPLC Conditions: Column: Chiralcel AD-H, Daicel Chemical Industries, Ltd., Eluent: Hexanes / IPA(10/1)containing 0.1% TFA; Flow rate: 1.0 mL/min; Detection: UV 210 nm

Racemic



The Reaction Result



References

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