Supporting Information

Study of the *E-Z* stilbene isomerisation in perchlorotriphenyl-methane (PTM) derivatives

Francesc Bejarano,^{#[a]} Isaac Alcon, ^{# [b]} Núria Crivillers,^[a] Marta Mas-Torrent,^{*[a]} Stefan T. Bromley,^[b] Jaume Veciana^[a] and Concepció Rovira^{*[a]}



Figure S1. UV-vis spectrum in THF of *E*-2, in red, and of a *Z*-2/*E*-2 mixture (99:1) in blue.

Dr. N. Crivillers, F. Bejarano, Dr. M. Mas-Torrent, Prof. J. Veciana [a] and Prof. C. Rovira Corresponding Authors: M. Mas-Torrent and C. Rovira Institute of Materials Science of Barcelona (ICMAB-CSIC) and Networking Research Center on Bioengineering, Biomaterials and Nanomedicine(CIBER-BBN) Campus de la UAB, 08193, Bellaterra ,Spain E-mail: mmas@icmab.es; cun@icmab.es Prof. S. T. Bromley [b] Departament de Química Física & Institut de Química Teòrica Computacional (IQTCUB), Universitat de Barcelona, 08028, i Barcelona, Spain and Institució Catalana de Recerca i Estudis Avançats (ICREA), 08010 Barcelona, Spain. These authors contributed equally

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7.65 7.60 7.55 7.50 7.45 7.40 7.35 7.30 7.25 7.20 7.15 7.10 7.05 7.00 6.95 6.90 6.85 6.80 6.75 6.70 6.65 6.60 6.55 6.50 6.45 6.40 6.35 fl (pom)

Figure S2. ¹H-NMR (400MHz) of compound *Z*-**2** in CD_2CI_2 after irradiation during 450 min. *Z*-isomer signals are squared in blue, whereas the non-converted *E*-isomer signals are squared in gree





Figure S3. Time evolution of the isomerisation of compound **2** under illumination with a Hg(Xe) lamp. Experimental conditions: 3.2 mM solution of **2** in CHCl₃ was irradiated at 25 °C and aliquots were diluted and injected in HPLC, stationary phase C-18 reversed phase, mobile phase CHCl₃/ACN (20/80), 40 °C, UV detector with a fixed wavelength of 254 nm and a flow of 1ml/min.



Figure S4. HPLC time monitoring of the effect of temperature on *E*-**2**. Experimental conditions: 0.067 mM solution of **2** in toluene was refluxed and aliquots were diluted and injected, stationary phase C-18 reversed phase, mobile phase CHCl₃/ACN (20/80), 40 °C, UV detector with a fixed wavelength of 254 nm and a flow of 1ml/min.



Figure S5. HPLC time monitoring of the isomerization trial catalyzed by I_2 of *Z*-**2**. Experimental conditions: 0.022 mM solution of the mixture obtained after 450 min illumination of *E*-**2** in toluene, with 0.03 equivalents of I_2 , was refluxed and aliquots were washed with sodium sulfite, diluted and injected, stationary phase C-18 reversed phase, mobile phase CHCl₃/ACN (20/80), 40 °C, UV detector with a fixed wavelength of 254 nm and a flow of 1ml/min.





Figure S6. ¹H-NMR (400MHz) of the *Z*-**1** in CD_2CI_2 after irradiation during 145 minutes. *Z*-isomer signals are squared in blue, whereas the non-converted *E*-isomer signals are squared in green.



Figure S7. Manual twisting of the C=C bond (see red arrow on **2** structure) by means of DFT constrained optimizations using the PBE0 hybrid functional and a 6-311-G++ basis set as implemented in the Gaussian09 code. The plotted energies are referenced to the most stable conformation for each compound (i.e. *E*-1 and *E*-2, respectively).



Figure S8. Variation of the dihedral angle of the central vinylene unit (C=C bond, see red arrow on stilbene structure) during 4 ps of a molecular dynamics simulation at 450K for our three studied compounds (**1**, **2** and stilbene) in each conformation (*E* and *Z*). Such calculations were performed using the PBE0 hybrid functional, the Bussi-Donadio-Parrinello thermostat and a light numerical basis set as implemented in the FHI-AIMS code. Red dashed lines point to the hypothetical perpendicular conformations for comparison.