

## Supporting Information On: Poly(Oxymethylene) Dimethyl Ethers - A combined chemical equilibrium investigation towards increasingly efficient and a potentially sustainable synthetic route

M. Ouda,<sup>a,b</sup> G. Yarce,<sup>a</sup> R. J. White,<sup>a,\*</sup> M. Hadrich,<sup>a</sup> D. Himmel,<sup>c</sup> A. Schaadt,<sup>a</sup> H. Klein,<sup>b</sup> E. Jacob<sup>d</sup> and I. Krossing,<sup>c,e</sup>

<sup>a</sup> Fraunhofer Institute for Solar Energy Systems ISE, Heidenhofstraße 2, 79110 Freiburg, Germany; Email: [robin.white@ise.fraunhofer.de](mailto:robin.white@ise.fraunhofer.de), Tel.: +497614588519

<sup>b</sup> Institute of Process and Plant Technology, Technical University Munich, Boltzmannstraße 15, 85748 Garching, Germany

<sup>c</sup> Institute for Inorganic and Analytical Chemistry, Albert-Ludwigs-University Freiburg, Albertstraße 21, 79104 Freiburg, Germany

<sup>d</sup> Motors Emissions Concepts UG, Karwendelstraße 25, 82152 Krailling, Germany

<sup>e</sup> FMF - Freiburger Materialforschungszentrum, Stefan-Meier-Straße 21, D-79104 Freiburg, Germany

### Sources for Thermodynamic Standard Properties in Literature

**Table S1** Sources for standard Gibbs energy of formation  $\Delta g^{\circ}_{if}$

Literature sources	Estimation methods
<ul style="list-style-type: none"><li>DIPPR 801 project, Design institute for physical properties</li><li>NIST, the National Institute for Standards and Technology</li><li>NBS Technical Note 270</li><li>JANAF (Joint Army-Navy-Air Force) tables</li></ul>	<ul style="list-style-type: none"><li>Method of Joback (1984; 1987)</li><li>Method of Constantinou and Gani (1994)</li><li>Method of Benson (1968; 1969)</li><li>CHETAH-software from ASTM (1998)</li></ul>

## Thermodynamic Chemical and Phase Equilibrium Model parameters

**Table S2** Parameters A, B for chemical equilibrium constants (Kj) fitted to experimental data <sup>1,2</sup>

$\ln K(T) = A + \frac{B}{T}$	A	B
$FA + H_2O \rightleftharpoons MG_1$	-2.3250	2579.0
$FA + MG_{n-1} \rightleftharpoons MG_n; n \geq 2$	-2.4334	3039.4
$FA + MeOH \rightleftharpoons HF_1$	-1.9020	3512.0
$FA + HF_{n-1} \rightleftharpoons HF_n; n \geq 2$	-2.2496	3008.8
$HF_1 + MeOH \xrightleftharpoons{H^+} OME_1 + H_2O$	0.8147	340.25
$FA + OME_{n-1} \xrightleftharpoons{H^+} OME_n; n \geq 2$	-2.4154	3029.6
$3FA \xrightleftharpoons{H^+} TRI$	-4.3253	7347.3

**Table S3** UNIFAC group assignment for all components

Group \ Substance	CH <sub>2</sub> O (FA)	H <sub>2</sub> O	C <sub>3</sub> H <sub>8</sub> O <sub>2</sub> (OME1)	OH(CH <sub>2</sub> O)H (MG1)	OH	CH <sub>2</sub>	CH <sub>3</sub> O	CH <sub>2</sub> OH	CH <sub>3</sub> OH	CH <sub>2</sub> O (OME)
Formaldehyde	1									
Water		1								
Methanol									1	
Methylal (OME1)			1							
MG1				1						
HF1							1	1		
MG <sub>n</sub> >2	n-1				2	1				
HF <sub>n</sub> >2	n-1						1	1		
OMEn>2			1							n-1
Trioxane	3									

**Table S4** UNIFAC Interaction parameters a<sub>ij</sub> (K) <sup>3,4</sup>

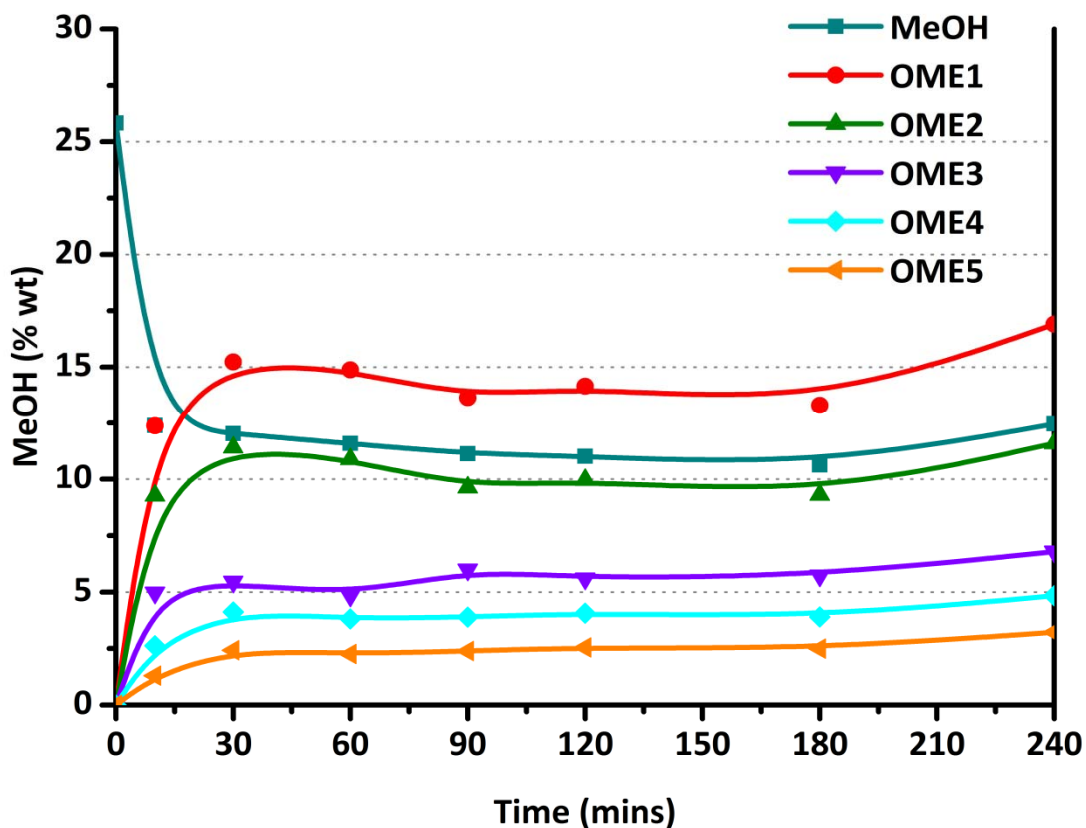
Group j \ Group i	CH <sub>2</sub> O (FA)	H <sub>2</sub> O	OME1	MG1	OH	CH <sub>2</sub>	CH <sub>3</sub> O	CH <sub>2</sub> OH	CH <sub>3</sub> OH	CH <sub>2</sub> O (OME)
CH <sub>2</sub> O FA	-	867.8	0.0	189.2	237.7	83.36	0.0	238.4	238.4	0.0
H <sub>2</sub> O	-254.5	-	a <sub>2,3</sub> (T)	189.5	-229.1	300.0	-219.3	a <sub>2,8</sub> (T)	289.6	a <sub>2,10</sub> (T)
OME1	0.0	a <sub>3,2</sub> (T)	-	a <sub>3,2</sub> (T)	237.7	83.36	0.0	0.0	410.0	26.0
MG1	59.2	-191.8	a <sub>2,3</sub> (T)	-	-229.1	300.0	-142.4	289.6	289.6	59.2
OH	28.06	353.5	28.06	353.5	-	156.4	112.8	-137.1	-137.1	28.06
CH <sub>2</sub>	251.5	1318	251.5	1318	986.5	-	447.8	697.2	697.2	251.5
CH <sub>3</sub> O	0.0	423.8	0.0	774.8	1164.8	273.0	-	238.4	238.4	0.0
CH <sub>2</sub> OH	-128.6	a <sub>8,2</sub> (T)	0.0	-181.0	249.1	16.5	-128.6	-	0.0	-128.6
CH <sub>3</sub> OH	-128.6	-181.0	-71.21	-181.0	249.1	16.5	-128.6	0.0	-	-128.6
CH <sub>2</sub> O (OME)	0.0	670.7	141.5	189.2	237.7	83.36	0.0	238.4	238.4	-

Here,  $a_{2,3}(T) = -225.5 + 0.705(T/K)$ ;  $a_{3,2}(T) = 1031.0 - 1.749(T/K)$ ;  $a_{8,2}(T) = -1018.57 + 329.9(T/K)$ ;  $a_{2,8}(T) = 451.64 - 114100(T/K)$ ;  $a_{2,10}(T) = 168.9 - 0.8776(T/K)$

## Experimental data T1 and T2

**Table S5** Experimental data of OME-synthesis on T1

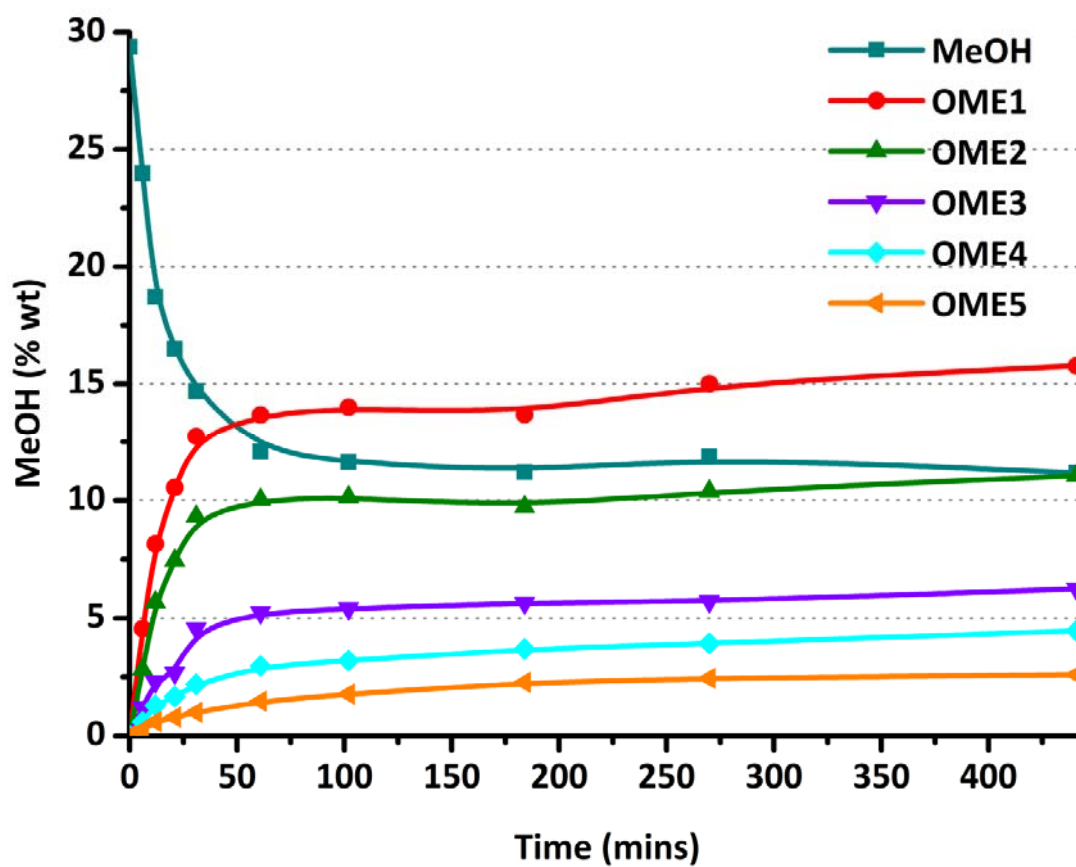
Test 13 – Autoclave – D50WX2 – Experimental data						
sample Time (min)	p-FA = 119.94 g MeOH = 80.20 g		Catalyst Mass = 1.0048 g MeOH for cat. loading = 7.41 g			
	MeOH (%wt.)	OME1 (%wt.)	OME2 (%wt.)	OME3 (%wt.)	OME4 (%wt.)	OME5 (%wt.)
0	25.83	0.00	0.000	0.000	0.000	0.000
10	12.38	12.37	9.27	4.97	2.63	1.30
30	12.02	15.23	11.41	5.45	4.12	2.42
60	11.58	14.88	10.89	4.84	3.81	2.26
90	11.13	13.64	9.63	5.99	3.89	2.39
120	11.01	14.15	9.97	5.59	4.07	2.55
180	10.63	13.30	9.31	5.73	3.90	2.49
240	12.45	16.91	11.59	6.79	4.84	3.22
<b>AV</b>	<b>11.48</b>	<b>14.69</b>	<b>10.47</b>	<b>5.74</b>	<b>4.11</b>	<b>2.56</b>



**Figure S1** Graphic progress of OME-synthesis on T1

**Table S6** Experimental data of OME-synthesis on T2

Test 14 – Autoclave – A36 – Experimental data							
p-FA = 120.05 g MeOH = 80.25 g			Catalyst Mass = 1.0038 g MeOH for cat. loading = 6.41 g				
sample Time (min)	MeOH (%wt.)	FA (%wt.)	OME1 (%wt.)	OME2 (%wt.)	OME3 (%wt.)	OME4 (%wt.)	OME5 (%wt.)
0	29.37	57.21	0.00	0.00	0.000	0.00	0.00
6	23.98	--	4.55	2.81	1.20	0.60	0.25
12	18.71	43.81	8.15	5.67	2.30	1.29	0.60
21	16.49	--	10.55	7.45	2.68	1.67	0.79
31	14.68	36.35	12.73	9.32	4.56	2.19	1.00
61	12.07	33.47	13.67	10.04	5.23	2.97	1.47
102	11.63	32.33	14.00	10.14	5.40	3.20	1.77
184	11.20	32.04	13.69	9.72	5.65	3.69	2.27
270	11.86	31.96	15.01	10.39	5.71	3.93	2.44
441	11.17	31.74	15.77	11.05	6.24	4.46	2.60
AV	11.41	31.91	14.83	10.39	5.87	4.03	2.44



**Figure S2** Graphic progress of OME-synthesis on T2

## References

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2. N. Schmitz, F. Homberg, J. Berje, J. Burger and H. Hasse, *Ind. Eng. Chem. Res.*, 2015, **54**(25), 6409.
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