

Electronic Supporting Information for:

Heterobimetallic Pd-K Carbene Complexes via One-Electron Reductions of Palladium Radical Carbenes

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1 X-ray data for compounds 2, 3, 5 and 6

X-Ray crystal structure of $\{PC^{\bullet}(sp^2)P\}^{tBu}PdNH^pTol$ (2). Single crystals were obtained as dark-green blocks from a concentrated *n*-pentane at at -35°C in the glovebox. Crystal and refinement data for **2**: $C_{40}H_{60}NP_2Pd$; $M_r = 723.23$; Triclinic; space group $P\bar{1}$; $a = 9.9475(7)$ Å; $b = 13.7704(10)$ Å; $c = 14.6585(10)$ Å; $\alpha = 98.9950(19)^{\circ}$; $\beta = 100.8508(18)^{\circ}$; $\gamma = 97.3666(18)^{\circ}$; $V = 1921.9(2)$ Å³; $Z = 2$; $T = 120(2)$ K; $\lambda = 0.71073$ Å; $\mu = 0.593$ mm⁻¹; $d_{\text{calc}} = 1.250$ g·cm⁻³; 47002 reflections collected; 6761 unique ($R_{\text{int}} = 0.0350$); giving $R_1 = 0.0253$, $wR_2 = 0.0606$ for 6096 data with [$I > 2\sigma(I)$] and $R_1 = 0.0302$, $wR_2 = 0.0624$ for all 6761 data. Residual electron density (e⁻·Å⁻³) max/min: 0.818/-0.527.

X-Ray crystal structure of $\{PC^{\bullet}(sp^2)P\}^{tBu}PdNPh_2$ (3). Single crystals were obtained as dark-green blocks from a concentrated *n*-pentane at at -35°C in the glovebox. Crystal and refinement data for **3**: $C_{45}H_{62}NP_2Pd$; $M_r = 785.30$; Monoclinic; space group $C2/c$; $a = 20.2994(14)$ Å; $b = 17.5268(14)$ Å; $c = 15.9561(16)$ Å; $\alpha = 90^{\circ}$; $\beta = 123.775(3)^{\circ}$; $\gamma = 90^{\circ}$; $V = 4718.8(7)$ Å³; $Z = 4$; $T = 120(2)$ K; $\lambda = 0.71073$ Å; $\mu = 0.488$ mm⁻¹; $d_{\text{calc}} = 1.105$ g·cm⁻³; 50505 reflections collected; 4154 unique ($R_{\text{int}} = 0.0474$); giving $R_1 = 0.0246$, $wR_2 = 0.0587$ for 3778 data with [$I > 2\sigma(I)$] and $R_1 = 0.0300$, $wR_2 = 0.0604$ for all 4154 data. Residual electron density (e⁻·Å⁻³) max/min: 0.481/-0.418.

X-Ray crystal structure of $\{PC(sp^2)P\}^{tBu}PdNPh_2^- [KOEt_2]^+$ (5). Single crystals were obtained as greenish-brown blocks from a concentrated solution of diethyl ether layered with *n*-pentane at -35°C in the glovebox. Crystal and refinement data for **5**: $C_{49}H_{72}KNOP_2Pd$; $M_r = 898.52$; Monoclinic; space group $P2_1/n$; $a = 10.2409(9)$ Å; $b = 18.1690(15)$ Å; $c = 25.743(2)$ Å; $\alpha = 90^{\circ}$; $\beta = 90.553(3)^{\circ}$; $\gamma = 90^{\circ}$; $V = 4789.7(7)$ Å³; $Z = 4$; $T = 120(2)$ K; $\lambda = 0.71073$ Å; $\mu = 0.575$ mm⁻¹; $d_{\text{calc}} = 1.246$ g·cm⁻³; 75438 reflections collected; 8436 unique ($R_{\text{int}} = 0.0384$); giving $R_1 = 0.0252$, $wR_2 = 0.0563$ for 7432 data with [$I > 2\sigma(I)$] and $R_1 = 0.0318$, $wR_2 = 0.0582$ for all 8436 data. Residual electron density (e⁻·Å⁻³) max/min: 0.420/-0.279.

X-Ray crystal structure of $\{PC(sp^2)P\}^{tBu}PdCH_2Ph^- K^+$ (6). Single crystals were obtained as dark-brown blocks by slow evaporation of diethyl ether solution at room temperature in the glovebox. Crystal and refinement data for **6**: $C_{40}H_{59}KP_2Pd$; $M_r = 747.31$; Monoclinic; space group $C2/c$; $a = 26.989(3)$ Å; $b = 12.0603(10)$ Å; $c = 23.568(2)$ Å; $\alpha = 90^{\circ}$; $\beta = 96.879(4)^{\circ}$; $\gamma = 90^{\circ}$; $V = 7616.1(12)$ Å³; $Z = 8$; $T = 120(2)$ K; $\lambda = 0.71073$ Å; $\mu = 0.707$ mm⁻¹; $d_{\text{calc}} = 1.303$ g·cm⁻³; 90845 reflections collected; 6712 unique ($R_{\text{int}} = 0.0540$); giving $R_1 = 0.0267$, $wR_2 = 0.0661$ for 5799 data with [$I > 2\sigma(I)$] and $R_1 = 0.0349$, $wR_2 = 0.0692$ for all 6712 data. Residual electron density (e⁻·Å⁻³) max/min: 0.917/-0.478.

2 EPR Spectra

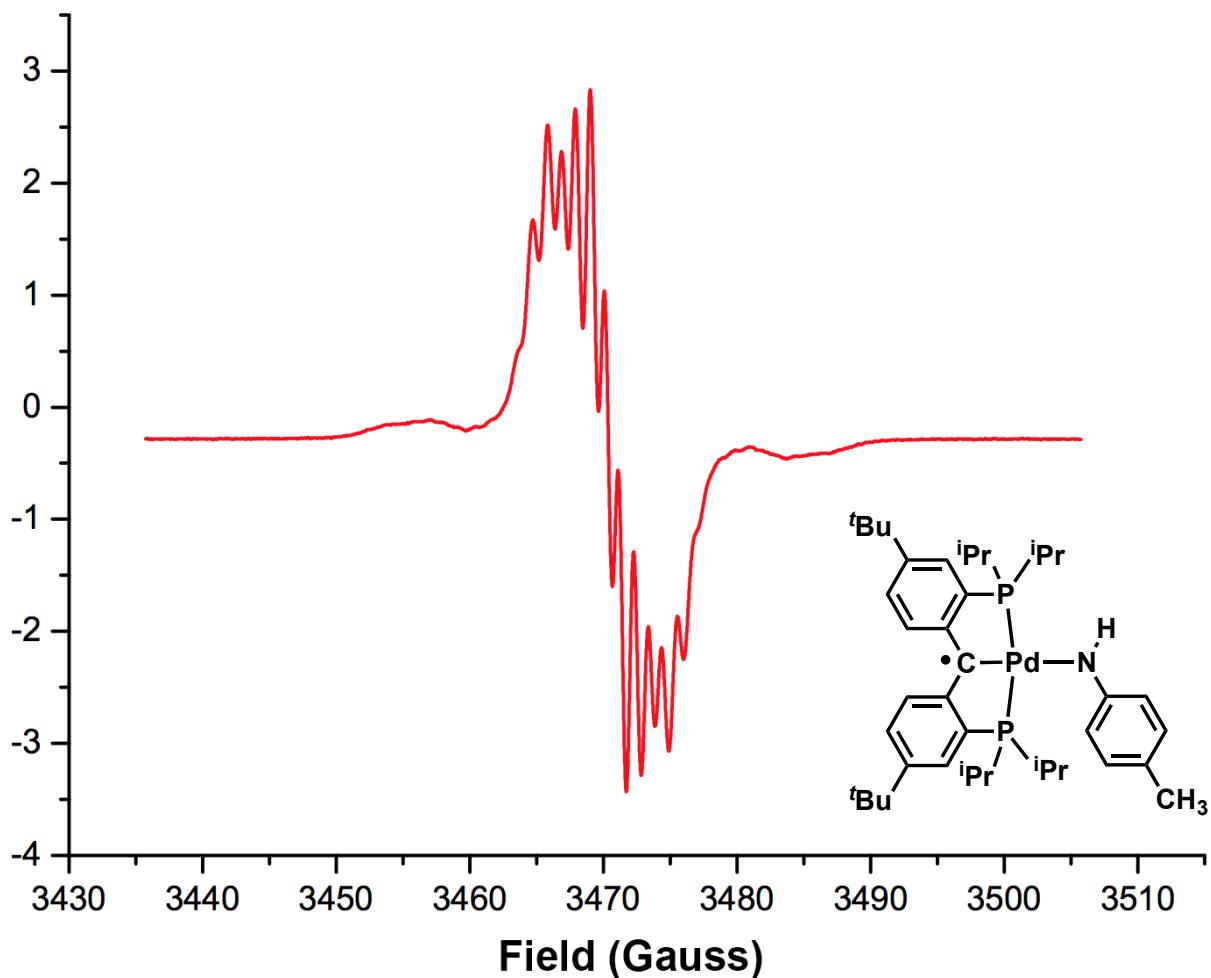


Figure S1. EPR spectrum of $\left[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNH}^{\text{pTol}}\right]$ (**2**) (1 mM solution in toluene, 298 K).

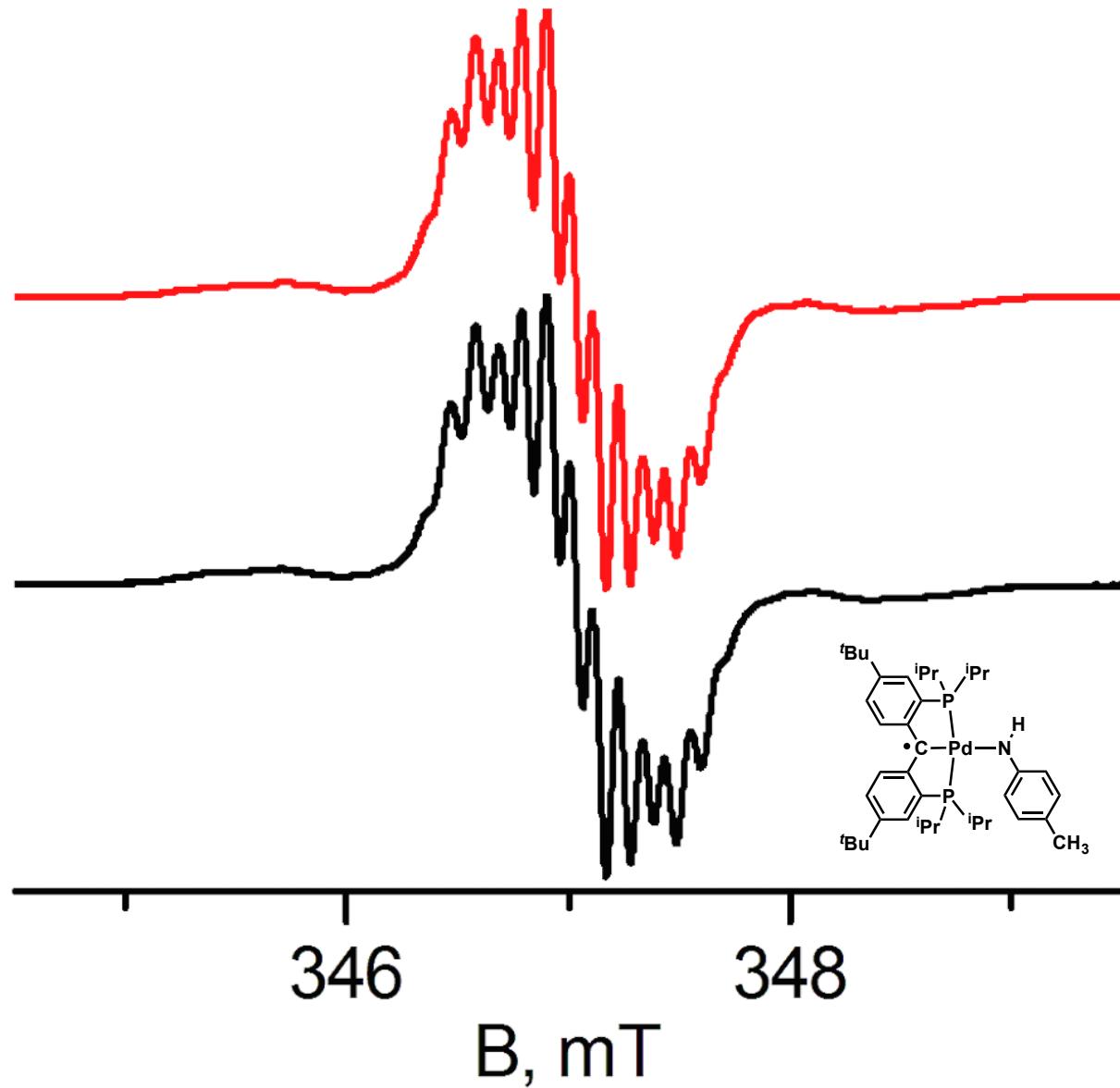


Figure S2. Experimental (black) and simulated (red) X-band EPR spectra of 10^{-4} M toluene solution of $[\{PC^{\bullet}(sp^2)P\}^tBuPdNH^pTol]$ (**2**). Simulation parameters: $a_1(2H) = 0.32$ mT, $a_2(2H) = 0.13$ mT, $a_3(2H) = 0.11$ mT, $a(N) = 0.09$ mT, $a(^{105}Pd) = 0.47$ mT, $g = 2.0088$, 22.33% contribution from the ^{105}Pd species.

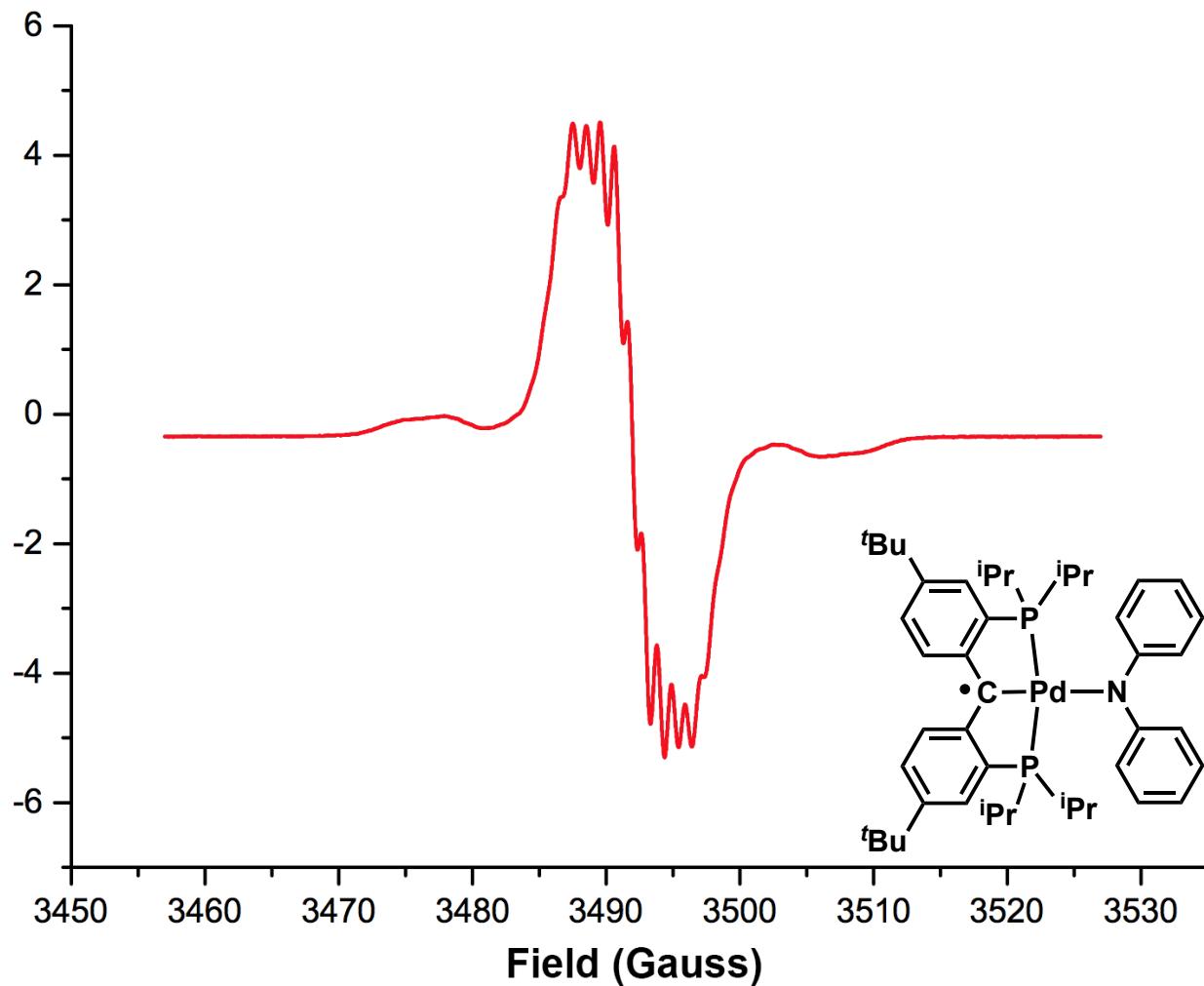


Figure S3. EPR spectrum of $[\{\text{PC}^\bullet(\text{sp}^2)\text{P}\}^t\text{Bu}\text{PdNPh}_2]$ (**3**) (1 mM solution in toluene, 298 K).

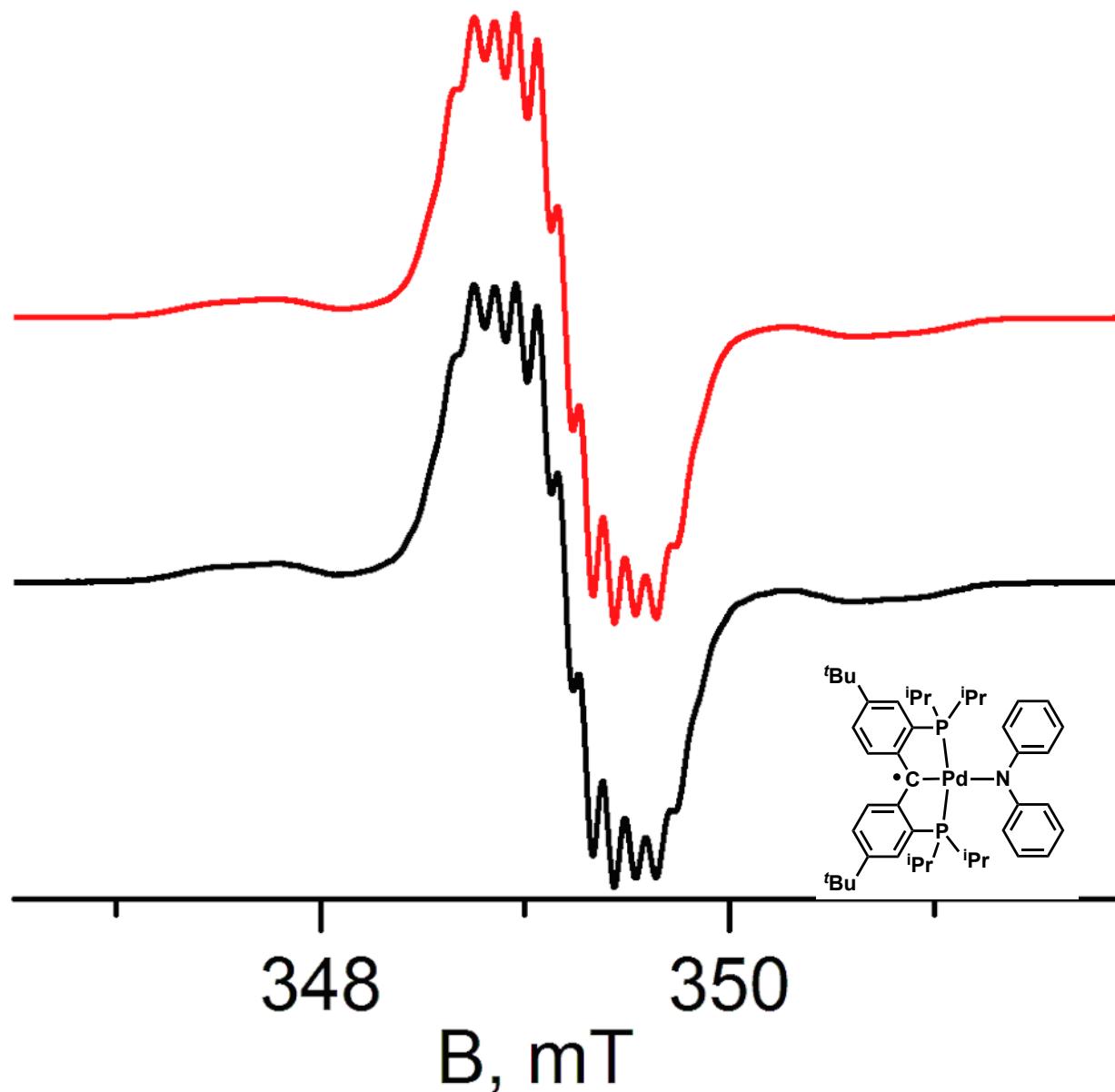


Figure S4. Experimental (black) and simulated (red) X-band EPR spectra of 10^{-4}M toluene solution of $[\{\text{PC}^*(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNPh}_2]$ (**3**). Simulation parameters: $a_1(2\text{H}) = 0.31\text{ mT}$, $a_2(2\text{H}) = 0.13\text{ mT}$, $a_3(2\text{H}) = 0.10\text{ mT}$, $a(\text{N}) = 0.09\text{ mT}$, $a(^{105}\text{Pd}) = 0.51\text{ mT}$, $g = 2.0079$, 22.33% contribution from the ^{105}Pd species.

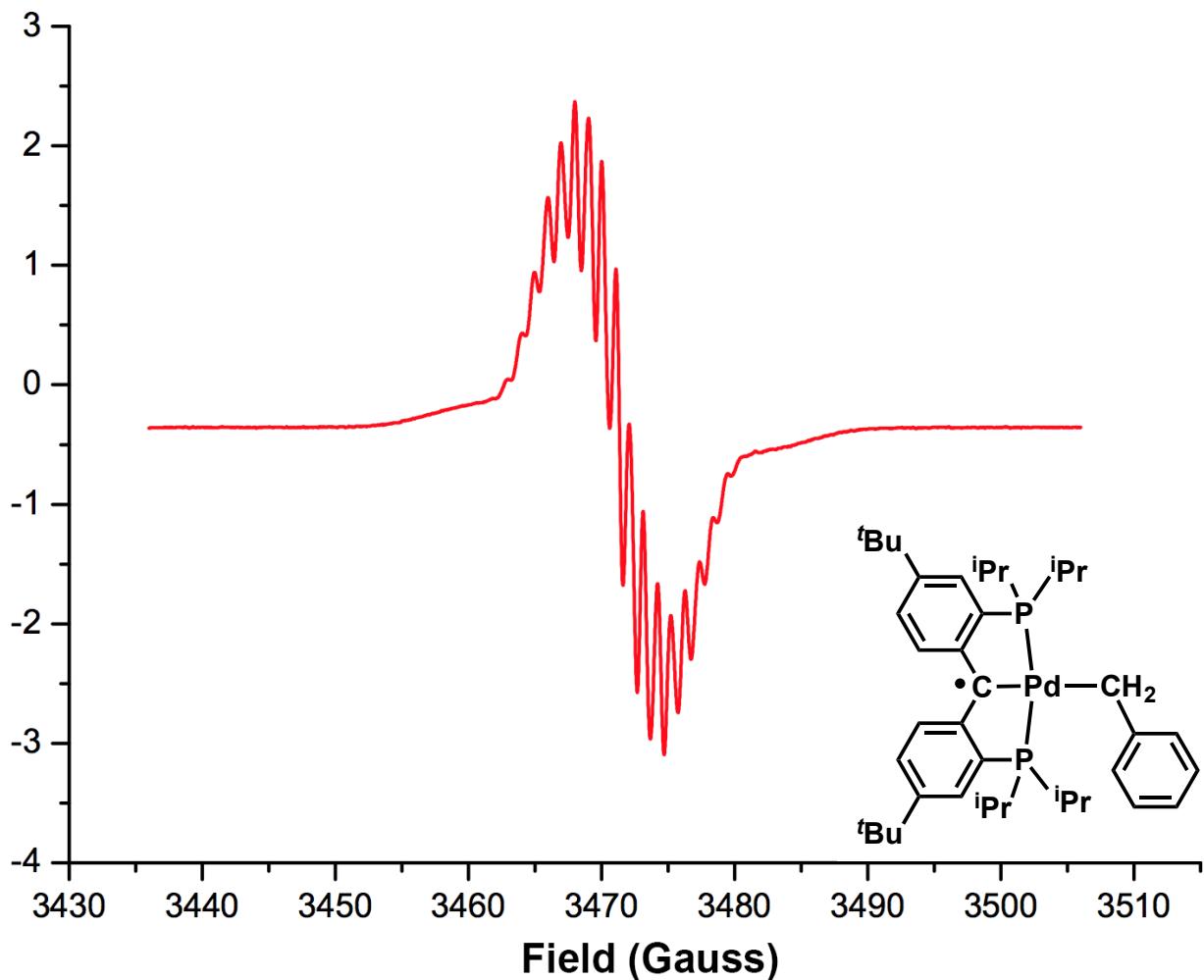


Figure S5. EPR spectrum of $[\{PC^{\bullet}(sp^2)P\}^tBuPdCH_2Ph]$ (**7**) (1 mM solution in toluene, 298 K).

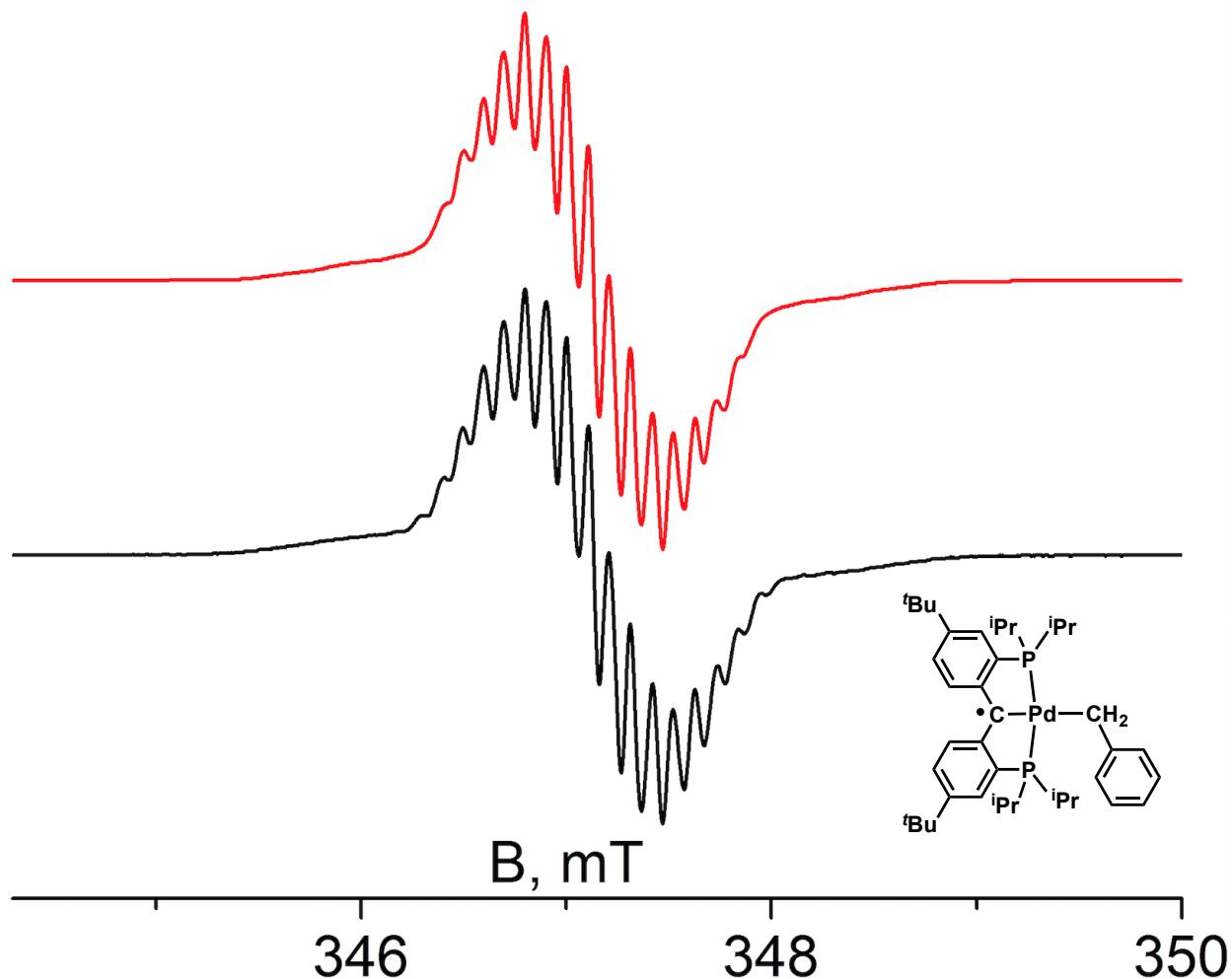


Figure S6. Experimental (black) and simulated (red) X-band EPR spectra of 10^{-4}M toluene solution of $[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}]$ (7). Simulation parameters: $a_1(2\text{H}) = 0.32 \text{ mT}$, $a_2(2\text{H}) = 0.19 \text{ mT}$, $a_3(2\text{H}) = 0.12 \text{ mT}$, $a_4(2\text{H}) = 0.08 \text{ mT}$, $a(^{105}\text{Pd}) = 0.33 \text{ mT}$, $g = 2.0086$, 22.33% contribution from the ^{105}Pd species.

3 DFT Results

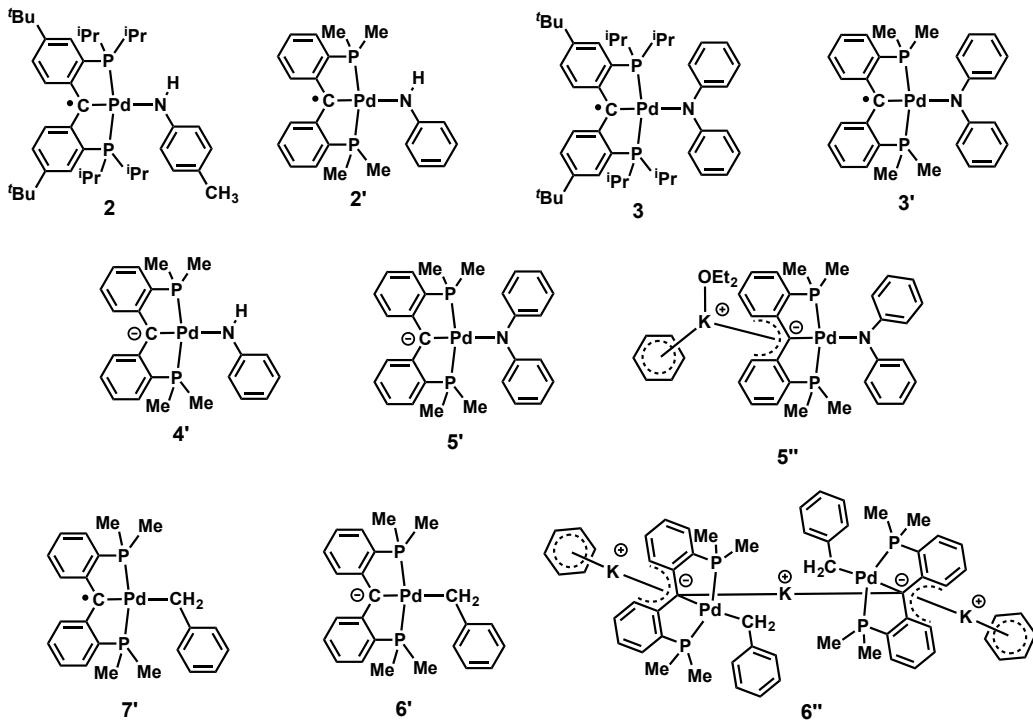


Figure S7. Computed molecules.

3.1 $[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNH^pTol]$ (2)

Table S1. Optimized coordinates for $[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNH^pTol]$ (2).

atom	x	y	z
C	-1.535225	5.049549	0.877464
C	-1.077807	3.810869	0.314745
C	-1.109118	3.720263	-1.111923
C	-1.574046	4.783029	-1.904850
C	-2.027553	5.997024	-1.343648
C	-1.991721	6.098823	0.068320
N	-0.627874	2.765001	1.096325
Pd	-0.152331	0.808087	0.510697
C	0.348286	-1.123105	0.003227
C	1.743297	-1.423738	-0.326165
C	2.785835	-0.539408	0.114803
C	4.130005	-0.773671	-0.196115
C	4.539767	-1.877970	-0.986373

Continued on next page

Table S1. – continued from previous page

atom	x	y	x
C	3.515221	-2.729857	-1.460982
C	2.165868	-2.516989	-1.144322
P	2.163552	0.956563	1.050297
C	2.673042	0.734456	2.892311
C	4.199642	0.665985	3.114801
C	6.035767	-2.084045	-1.311763
C	6.578724	-0.845062	-2.084982
C	-2.512759	7.149255	-2.208016
P	-2.342399	0.103165	-0.152331
C	-3.811275	0.455477	1.023987
C	-4.285940	1.920316	0.893443
C	-2.078444	-1.750860	-0.078395
C	-0.699328	-2.144771	0.004792
C	-0.464342	-3.545098	0.171536
C	-1.512853	-4.474668	0.213802
C	-2.866186	-4.082964	0.089824
C	-3.113197	-2.693006	-0.050932
C	-4.044748	-5.081117	0.126676
C	-3.568480	-6.545329	0.298902
C	-2.975778	0.513615	-1.921610
C	-4.339891	-0.124311	-2.262762
C	3.174733	2.446947	0.404491
C	2.856824	2.687170	-1.088160
C	-4.987409	-4.735238	1.317906
C	-4.846907	-4.985114	-1.205531
C	6.278856	-3.340961	-2.184484
C	6.837997	-2.251282	0.013206
C	-1.901928	0.125468	-2.961314
C	-3.403718	0.122594	2.476253
C	2.882063	3.710118	1.246207
C	1.959798	-0.501637	3.482767
H	-4.142172	-2.347144	-0.117547
H	-1.263149	-5.520845	0.360904
H	0.550998	-3.898299	0.319000
H	-5.393224	-3.719625	1.234241
H	-4.450564	-4.805162	2.272767
H	-5.835525	-5.433063	1.348531
H	-0.611736	2.975361	2.095065
H	-5.245145	-3.975801	-1.367452
H	-5.695841	-5.682332	-1.191327
H	-4.209721	-5.236906	-2.063105
H	-4.436743	-7.216616	0.311547

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Table S1. – continued from previous page

atom	x	y	x
H	-3.024962	-6.687713	1.242008
H	-2.916211	-6.860466	-0.525945
H	4.883027	-0.074113	0.159425
H	3.763698	-3.569485	-2.102739
H	1.418846	-3.175434	-1.575503
H	5.770263	-3.267340	-3.154520
H	5.936346	-4.255934	-1.683828
H	7.353105	-3.451160	-2.381100
H	6.734778	-1.374249	0.664352
H	7.906817	-2.383890	-0.203276
H	6.489014	-3.128699	0.572875
H	6.035980	-0.705281	-3.028643
H	7.644185	-0.978279	-2.317447
H	6.476793	0.076871	-1.499354
H	-2.580655	0.771172	2.800336
H	-3.083946	-0.920845	2.584549
H	-4.257201	0.290174	3.147675
H	-3.076335	1.606819	-1.909615
H	-4.626482	-0.218453	0.721577
H	-2.219681	0.451303	-3.961188
H	-1.754492	-0.961834	-2.989312
H	-0.935102	0.591588	-2.739885
H	-4.662256	2.153222	-0.110260
H	-3.471368	2.616902	1.116909
H	-5.102458	2.104089	1.605457
H	-5.125872	0.156490	-1.551172
H	-4.269260	-1.218696	-2.292918
H	-4.660834	0.213097	-3.258200
H	4.236413	2.176412	0.505130
H	0.873357	-0.441092	3.350414
H	2.176320	-0.579192	4.557144
H	2.306816	-1.423777	2.999336
H	4.719660	1.552373	2.731084
H	4.627449	-0.223756	2.636735
H	4.411311	0.597507	4.190927
H	3.437964	4.562362	0.832603
H	3.183917	3.595505	2.295621
H	1.813089	3.949570	1.211816
H	1.813455	2.997561	-1.214998
H	3.032422	1.790458	-1.694978
H	3.497739	3.491308	-1.475131
H	-0.742127	2.809153	-1.580926

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Table S1. – continued from previous page

atom	x	y	z
H	-1.573893	4.667213	-2.989306
H	-2.330600	7.020210	0.544033
H	-1.528345	5.167742	1.961924
H	-2.572425	6.854120	-3.263342
H	-1.840189	8.018657	-2.148883
H	-3.511830	7.495392	-1.904919
H	2.279189	1.635352	3.384223

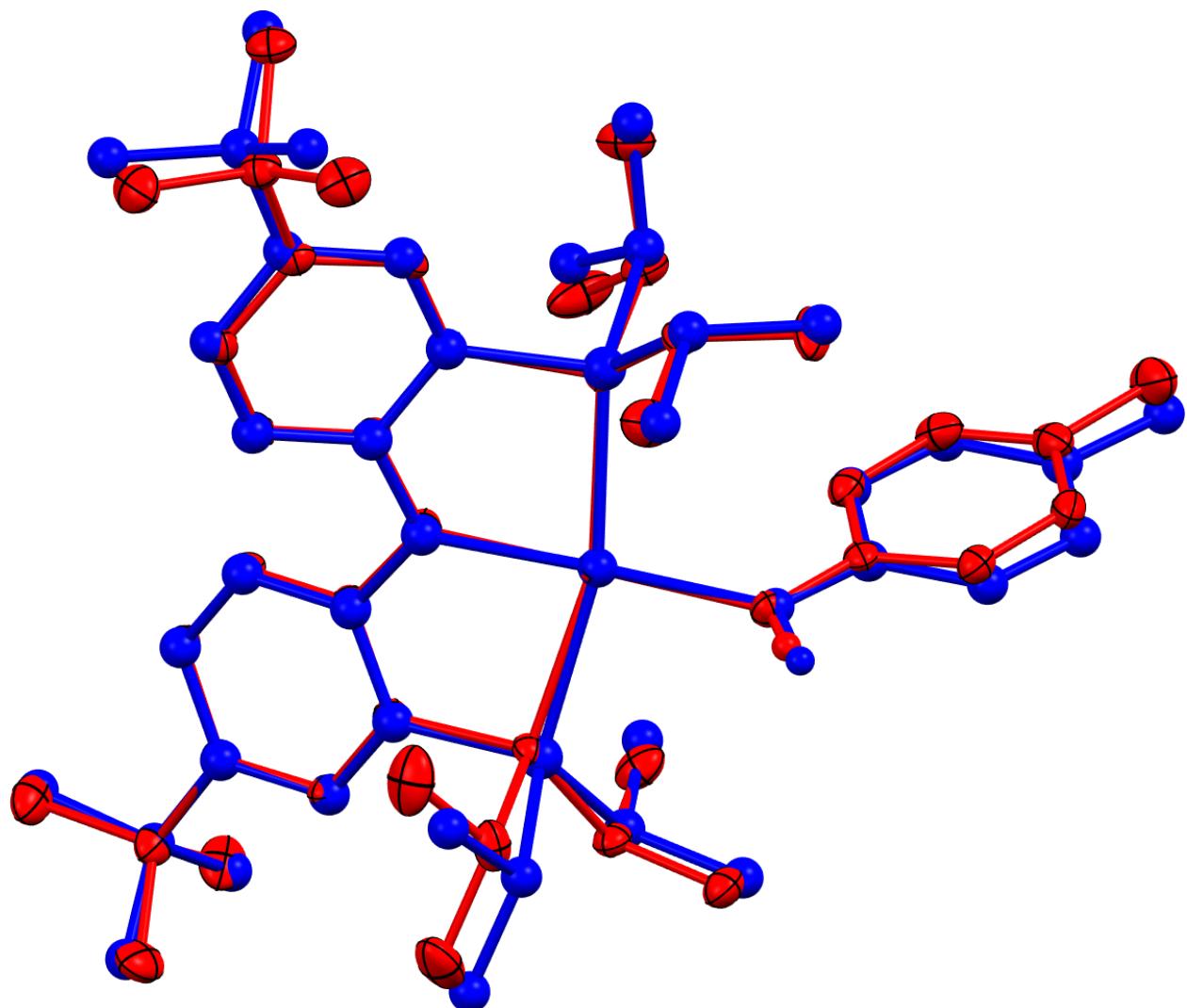


Figure S8. Overlaid structures for $[\{PC\bullet(sp^2)P\}^3BuPdNH^7Tol]$ (**2**) (red: X-ray, blue: optimized).

Table S2. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry and the crystal structure of $[\{\text{PC}^*(\text{sp}^2)\text{P}\}^t\text{Bu}\text{PdNH}^p\text{Tol}]$ (2).

Distance	Calcd.	X-Ray	Angle	Calcd.	X-Ray
Pd–N	2.097	2.0787(18)	P(1)–Pd–P(2)	165.97	163.689(19)
Pd–C	2.059	2.019(2)	C–Pd–N	177.88	178.97(9)
Pd–P(1)	2.394	2.2983(5)	C–Pd–P(1)	82.99	82.15(6)
Pd–P(2)	2.383	2.2841(5)	C–Pd–P(2)	82.99	81.80(6)
N–C(51)	1.381	1.354(3)	N–Pd–P(1)	98.32	98.20(5)
C–C(11)	1.463	1.451(3)	N–Pd–P(2)	95.68	97.79(5)
C–C(21)	1.465	1.457(3)	Pd–N–C(51)	128.50	126.94(15)

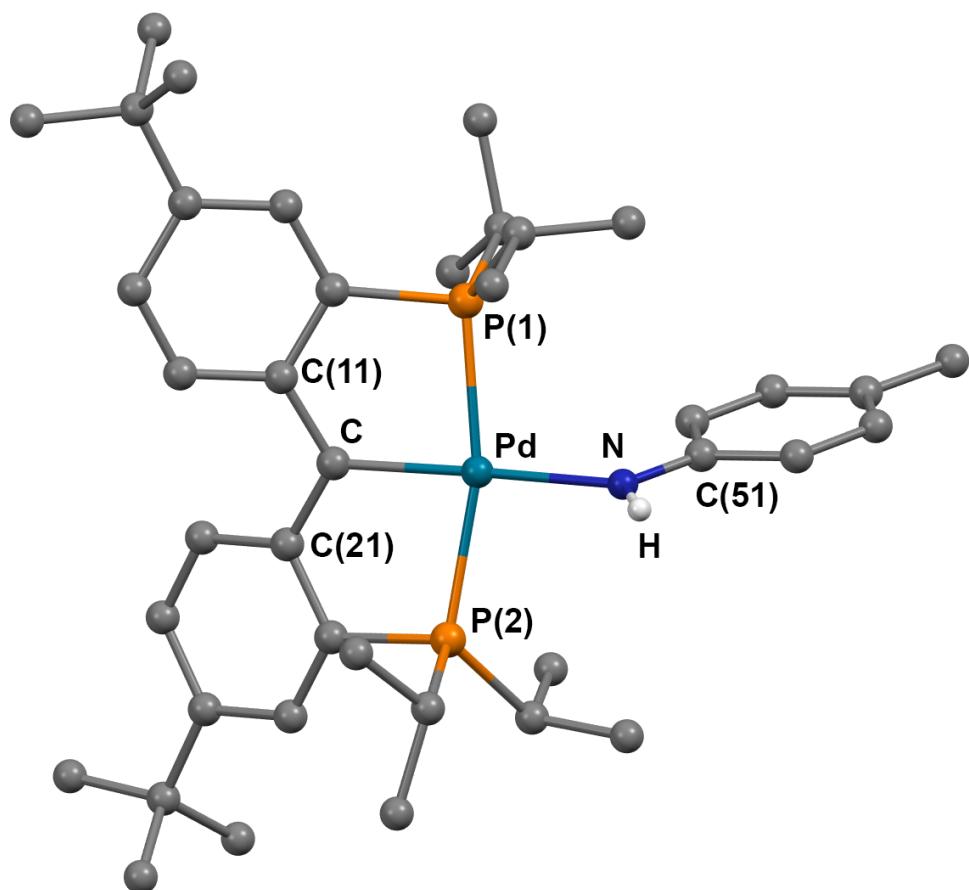


Figure S9. Optimized geometry for $[\{\text{PC}^*(\text{sp}^2)\text{P}\}^t\text{Bu}\text{PdNH}^p\text{Tol}]$ (2).

3.2 $[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNPh_2]$ (3)

Table S3. Optimized coordinates for $[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNPh_2]$ (3).

atom	x	y	z
C	0.006383	-2.885191	2.442283
C	0.249803	-3.557902	1.207922
C	0.828532	-4.863536	1.297817
C	1.121071	-5.448114	2.539367
C	0.860531	-4.769425	3.748525
C	0.304563	-3.476196	3.679741
N	0.001932	-2.887208	0.000046
C	-0.244723	-3.558408	-1.207800
C	-0.821324	-4.864990	-1.297667
C	-1.112616	-5.450201	-2.539210
C	-0.852829	-4.771272	-3.748395
C	-0.298868	-3.477179	-3.679647
C	-0.001956	-2.885526	-2.442194
Pd	0.000398	-0.733781	-0.000051
P	-2.320152	-0.414234	-0.565819
C	-3.726187	-1.365386	0.325857
C	-3.804024	-2.826724	-0.167499
P	2.320480	-0.411129	0.565835
C	3.727755	-1.360493	-0.325781
C	3.807542	-2.821674	0.167739
C	2.503971	1.368132	0.018355
C	1.260229	2.033063	-0.221988
C	1.356737	3.363316	-0.744956
C	2.591267	3.980517	-0.961298
C	3.820621	3.332050	-0.672223
C	3.737308	2.008086	-0.185481
C	-0.000970	1.325591	-0.000045
C	-1.263101	2.031393	0.221894
C	-2.505970	1.364796	-0.018372
C	-3.740150	2.003120	0.185482
C	-3.825196	3.326992	0.672179
C	-2.596693	3.977115	0.961153
C	-1.361354	3.361545	0.744794
C	5.158774	4.063875	-0.915018
C	6.385937	3.194148	-0.543485
C	-5.164302	4.057100	0.914879
C	-6.390321	3.185776	0.543322
C	2.777081	-0.464165	2.436751
C	4.257074	-0.136085	2.731147

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Table S3. – continued from previous page

atom	x	y	z
C	5.201672	5.359192	-0.050715
C	5.275837	4.448582	-2.420342
C	1.834440	0.468283	3.228408
C	3.554087	-1.293221	-1.858030
C	-2.776813	-0.467865	-2.436704
C	-4.257336	-0.142046	-2.730932
C	-5.208801	5.352311	0.050502
C	-5.281946	4.441742	2.420174
C	-1.835705	0.466122	-3.228369
C	-3.552719	-1.297697	1.858105
H	4.652281	1.460682	0.016781
H	2.598448	4.985741	-1.377688
H	0.450248	3.890918	-1.023316
H	4.370492	6.033235	-0.290323
H	6.139003	5.904324	-0.227136
H	5.140590	5.117530	1.018379
H	5.263329	3.552540	-3.054273
H	6.215889	4.986703	-2.603635
H	4.450636	5.096534	-2.739219
H	6.435016	2.278881	-1.147986
H	6.377606	2.909215	0.516922
H	7.307808	3.760842	-0.726761
H	2.566162	-1.501822	2.726860
H	4.655890	-0.836985	-0.053994
H	2.008597	1.520301	2.968237
H	2.018243	0.349128	4.304890
H	0.781090	0.240663	3.032404
H	4.949943	-0.826053	2.235060
H	4.435434	-0.212694	3.812775
H	4.505733	0.888690	2.427929
H	2.664609	-1.849152	-2.172990
H	4.425398	-1.752176	-2.345434
H	3.468283	-0.261348	-2.219357
H	2.882529	-3.361318	-0.057856
H	3.982093	-2.892811	1.248171
H	4.635518	-3.334505	-0.340640
H	-1.064550	-5.407503	-0.389449
H	-1.556643	-6.444438	-2.561265
H	-1.075566	-5.233803	-4.706583
H	-0.088336	-2.925952	-4.595143
H	0.423431	-1.884954	-2.406164
H	-4.654406	1.454502	-0.016735

Continued on next page

Table S3. – continued from previous page

atom	x	y	x
H	-2.605185	4.982368	1.377451
H	-0.455555	3.890374	1.023070
H	-4.378469	6.027404	0.290099
H	-6.146820	5.896277	0.226868
H	-5.147383	5.110667	-1.018578
H	-5.268293	3.545760	3.054165
H	-6.222713	4.978642	2.603383
H	-4.457618	5.090803	2.739054
H	-6.438238	2.270469	1.147853
H	-6.381566	2.900812	-0.517075
H	-7.312933	3.751285	0.726530
H	-2.564331	-1.505180	-2.726899
H	-4.655004	-0.843139	0.053965
H	-2.011521	1.517846	-2.968125
H	-2.019390	0.346736	-4.304845
H	-0.781982	0.240170	-3.032445
H	-4.949084	-0.833138	-2.234845
H	-4.435679	-0.218826	-3.812551
H	-4.507551	0.882312	-2.427591
H	-2.662594	-1.852507	2.173212
H	-4.423509	-1.757648	2.345502
H	-3.468201	-0.265671	2.219296
H	-2.878316	-3.365119	0.058214
H	-3.978420	-2.898203	-1.247934
H	-4.631355	-3.340584	0.340893
H	1.072400	-5.405776	0.389607
H	1.566676	-6.441643	2.561444
H	1.084261	-5.231465	4.706718
H	0.093463	-2.925147	4.595213
H	-0.420476	-1.885248	2.406214

Table S4. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry and the crystal structure of $\{[\text{PC}^{\bullet}(\text{sp}^2)\text{P}]^{\text{tBu}}\text{PdNPh}_2\}$ (**3**).

Distance	Calcd.	X-Ray	Angle	Calcd.	X-Ray
Pd–N	2.153	2.149(2)	P–Pd–P#1	164.69	163.80(2)
Pd–C	2.059	2.024(2)	C–Pd–N	180.00	180.0
Pd–P	2.410	2.3021(5)	C–Pd–P	82.34	81.900(12)
Pd–P#1	2.410	2.3021(5)	C–Pd–P#1	82.34	81.900(12)
N–C(31)	1.404	1.357(2)	N–Pd–P	97.66	98.100(12)
N–C(31)#1	1.404	1.357(2)	N–Pd–P#1	97.65	98.100(12)
C–C(11)	1.463	1.444(2)	Pd–N–C(31)	118.56	117.94(12)
C–C(11)#1	1.463	1.444(2)	Pd–N–C(31)#1	118.55	117.94(12)

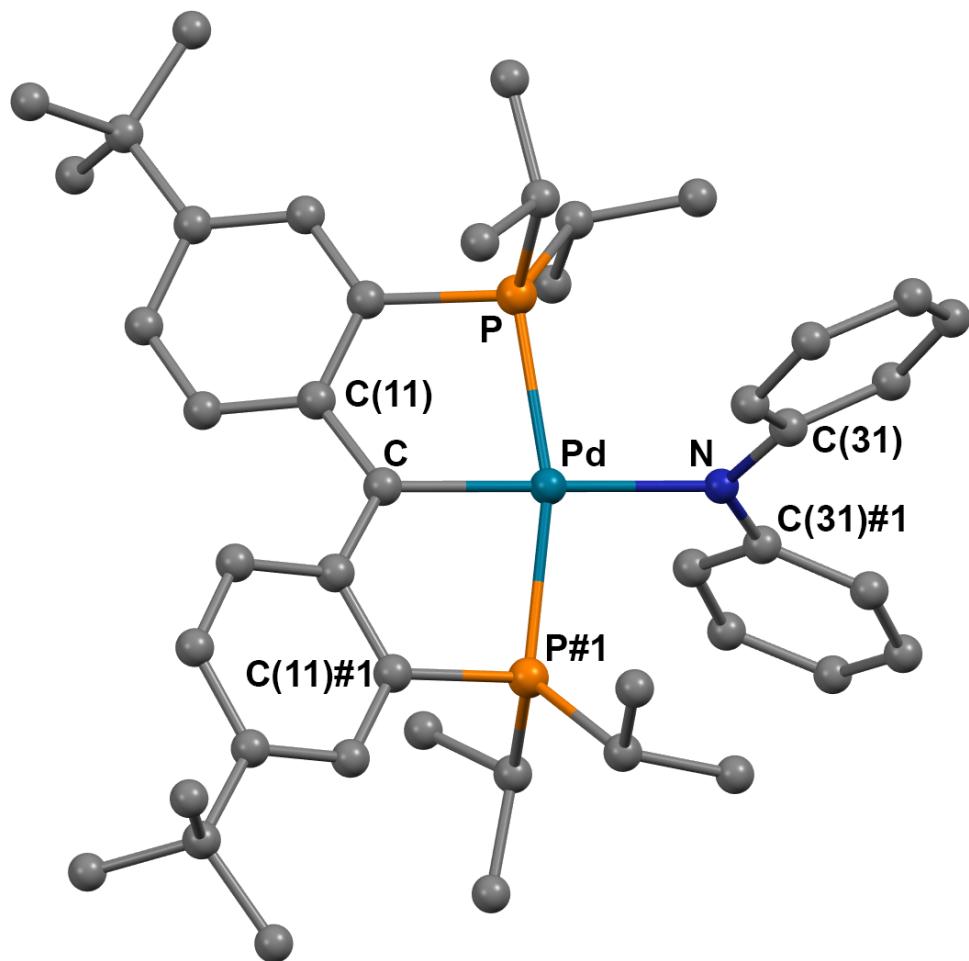


Figure S10. Optimized geometry for $\{[\text{PC}^{\bullet}(\text{sp}^2)\text{P}]^{\text{tBu}}\text{PdNPh}_2\}$ (**3**).

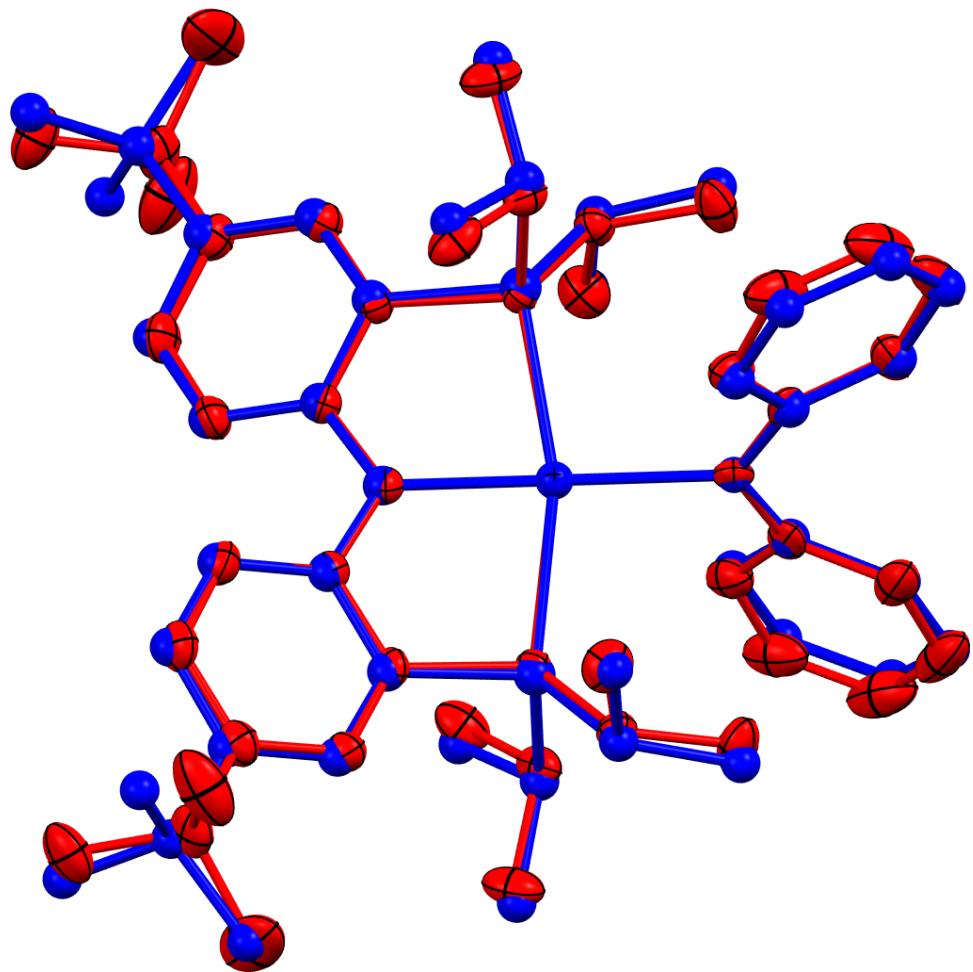


Figure S11. Overlaid structures for $[\{PC\bullet(sp^2)P\}^7BuPdNPh_2]$ (3) (red: X-ray, blue: optimized).

3.3 [{PC[•](sp²)P}^{Me}PdNHPh] (2')

Table S5. Optimized coordinates for [{PC[•](sp²)P}^{Me}PdNHPh] (2').

atom	x	y	z
C	-4.652272	-0.863966	-0.857068
C	-3.372795	-0.799864	-0.210450
C	-3.378490	-0.549231	1.199792
C	-4.578858	-0.366195	1.904055
C	-5.828158	-0.425331	1.249759
C	-5.843522	-0.682758	-0.138846
N	-2.186990	-0.983904	-0.896133
Pd	-0.287925	-0.299570	-0.361140
C	1.618161	0.363918	0.055455
C	2.631644	-0.612466	0.465453
C	2.426267	-2.005643	0.179662
C	3.360790	-2.983358	0.547841
C	4.527453	-2.626618	1.254623
C	4.729635	-1.272820	1.599019
C	3.809841	-0.287900	1.214510
P	0.796411	-2.382630	-0.652267
C	1.213413	-3.066740	-2.359020
H	1.853539	-3.952548	-2.274645
H	5.246395	-3.386164	1.550333
P	-0.866929	1.961545	0.086571
C	-2.008676	2.842756	-1.117560
H	-2.993250	2.367810	-1.067501
C	0.824860	2.733778	-0.112079
C	1.908666	1.790143	-0.114946
C	3.214060	2.325279	-0.368378
C	3.420504	3.698772	-0.555798
C	2.342140	4.607107	-0.495562
C	1.040914	4.108216	-0.279123
H	2.507293	5.672159	-0.636169
C	-1.489441	2.530524	1.771573
H	-1.481979	3.625212	1.828713
C	0.063691	-3.860434	0.246925
H	-0.013890	-3.622355	1.311471
H	-0.837114	2.121890	2.548640
H	-1.612674	2.733310	-2.131149
H	-0.943261	-4.037006	-0.142995
H	1.734696	-2.295989	-2.933616
H	0.200915	4.800658	-0.261561
H	4.423889	4.065209	-0.762724

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Table S5. – continued from previous page

atom	x	y	z
H	4.056394	1.648626	-0.466041
H	-2.319464	-1.202101	-1.885156
H	3.182050	-4.030194	0.307554
H	5.603939	-0.989145	2.181256
H	3.971218	0.736705	1.532685
H	-2.505444	2.153543	1.919589
H	-2.102658	3.906989	-0.873087
H	0.678940	-4.757793	0.114852
H	-2.425267	-0.544449	1.725713
H	-4.540755	-0.190881	2.979058
H	-6.794705	-0.740407	-0.667178
H	-4.689448	-1.058790	-1.929483
H	0.287458	-3.335438	-2.878290
H	-6.754538	-0.288028	1.801521

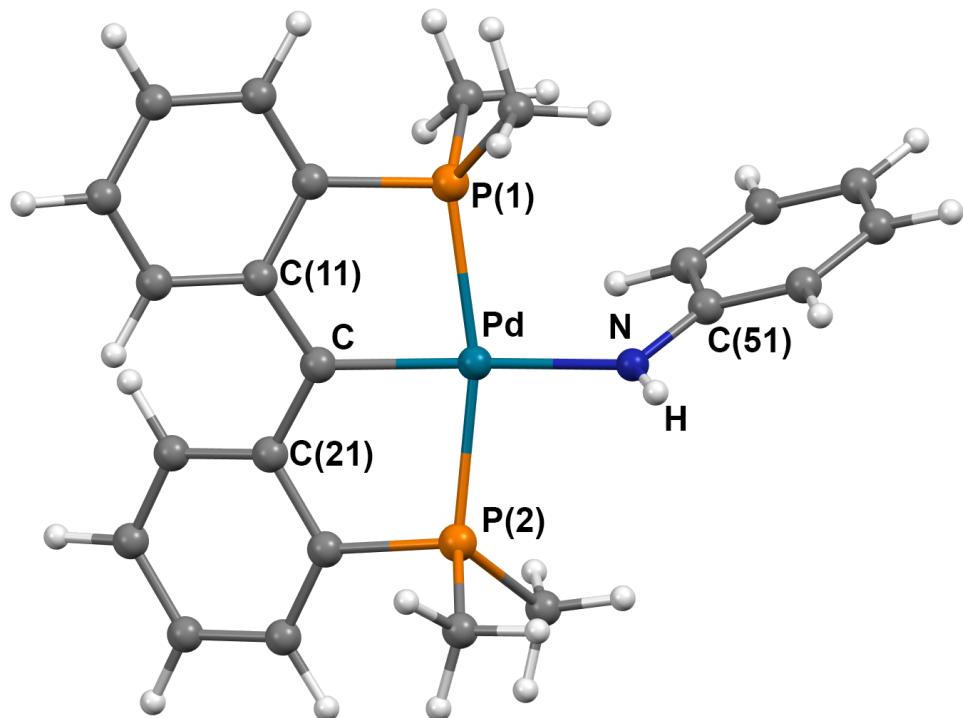


Figure S12. Optimized geometry for $[(PC(sp^2)P)^MePdNHPh]$ (**2'**).

Table S6. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry and the crystal structure of $[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdNHPh}]$ (**2'**).

Distance	Calcd.	X-Ray	Angle	Calcd.	X-Ray
Pd–N	2.088	2.0787(18)	P(1)–Pd–P(2)	166.48	163.689(19)
Pd–C	2.061	2.019(2)	C–Pd–N	176.76	178.97(9)
Pd–P(1)	2.377	2.2983(5)	C–Pd–P(1)	83.16	82.15(6)
Pd–P(2)	2.366	2.2841(5)	C–Pd–P(2)	83.36	81.80(6)
N–C(51)	1.382	1.354(3)	N–Pd–P(1)	97.96	98.20(5)
C–C(11)	1.465	1.451(3)	N–Pd–P(2)	95.55	97.79(5)
C–C(21)	1.466	1.457(3)	Pd–N–C(51)	127.55	126.94(15)

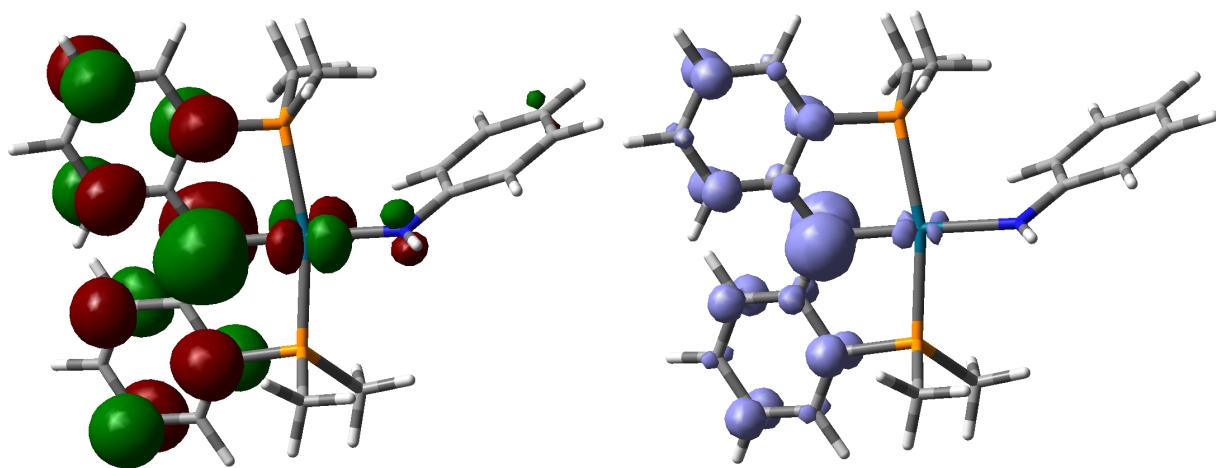


Figure S13. SOMO (left) and spin density (right) for $[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdNHPh}]$ (**2'**).

3.4 [{PC[•](sp²)P}^{Me}PdNPh₂] (3')

Table S7. Optimized coordinates for [{PC[•](sp²)P}^{Me}PdNPh₂] (3').

atom	x	y	z
C	-2.131918	0.814338	-2.304877
C	-2.768275	0.664222	-1.037100
C	-4.043837	1.290399	-0.876910
C	-4.636240	2.010385	-1.925808
C	-3.992094	2.145800	-3.174284
C	-2.730850	1.540407	-3.347082
N	-2.095689	0.000003	0.000006
C	-2.768259	-0.664224	1.037118
C	-4.043816	-1.290416	0.876938
C	-4.636201	-2.010408	1.925841
C	-3.992045	-2.145816	3.174312
C	-2.730806	-1.540408	3.347100
C	-2.131891	-0.814332	2.304890
Pd	0.032860	0.000003	-0.000007
P	0.320678	-2.351202	-0.191606
C	-0.692098	-3.273676	-1.477359
H	-1.746484	-3.220539	-1.189290
P	0.320684	2.351209	0.191589
C	-0.692093	3.273692	1.477334
H	-1.746477	3.220563	1.189259
C	2.096310	2.384220	0.771410
C	2.789220	1.136901	0.605319
C	4.123885	1.077918	1.123468
C	4.729135	2.193475	1.718024
C	4.040272	3.420733	1.821474
C	2.714355	3.502029	1.347954
C	2.090328	0.000000	-0.000002
C	2.789220	-1.136902	-0.605320
C	2.096307	-2.384218	-0.771418
C	2.714351	-3.502028	-1.347961
C	4.040271	-3.420736	-1.821473
C	4.729137	-2.193480	-1.718016
C	4.123888	-1.077923	-1.123462
H	4.515197	4.284721	2.278907
H	4.515196	-4.284725	-2.278904
C	0.240935	3.471109	-1.320434
H	0.516940	4.497438	-1.051656
H	0.934692	3.093345	-2.076912
H	-0.568473	2.787682	2.449070

Continued on next page

Table S7. – continued from previous page

atom	x	y	x
C	0.240918	-3.471104	1.320414
H	0.516921	-4.497434	1.051636
H	0.934672	-3.093344	2.076896
H	-0.568468	-2.787665	-2.449093
H	2.168594	4.438415	1.450538
H	5.738283	2.105794	2.115091
H	4.662807	0.136632	1.098636
H	-0.773589	3.448713	-1.729008
H	-0.384976	4.323233	1.549295
H	-4.554330	-1.222976	-0.079469
H	-5.606072	-2.478472	1.763687
H	-4.458927	-2.701461	3.983657
H	-2.214588	-1.623540	4.302360
H	-1.164354	-0.338386	2.454731
H	2.168587	-4.438412	-1.450549
H	5.738289	-2.105803	-2.115077
H	4.662814	-0.136639	-1.098624
H	-0.773608	-3.448707	1.728983
H	-0.384989	-4.323219	-1.549319
H	-4.554344	1.222954	0.079501
H	-5.606114	2.478438	-1.763646
H	-4.458989	2.701441	-3.983625
H	-2.214640	1.623546	-4.302346
H	-1.164377	0.338403	-2.454726

Table S8. Selected distances (Å) and angles (°) for the optimized geometry and the crystal structure of $\left[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdNPh}_2\right]$ (**3'**).

Distance	Calcd.	X-Ray	Angle	Calcd.	X-Ray
Pd–N	2.129	2.149(2)	P–Pd–P#1	166.09	163.80(2)
Pd–C	2.057	2.024(2)	C–Pd–N	180.00	180.0
Pd–P	2.376	2.3021(5)	C–Pd–P	83.04	81.900(12)
Pd–P#1	2.376	2.3021(5)	C–Pd–P#1	83.04	81.900(12)
N–C(31)	1.403	1.357(2)	N–Pd–P	96.96	98.100(12)
N–C(31)#1	1.403	1.357(2)	N–Pd–P#1	96.96	98.100(12)
C–C(11)	1.465	1.444(2)	Pd–N–C(31)	118.64	117.94(12)
C–C(11)#1	1.465	1.444(2)	Pd–N–C(31)#1	118.64	117.94(12)

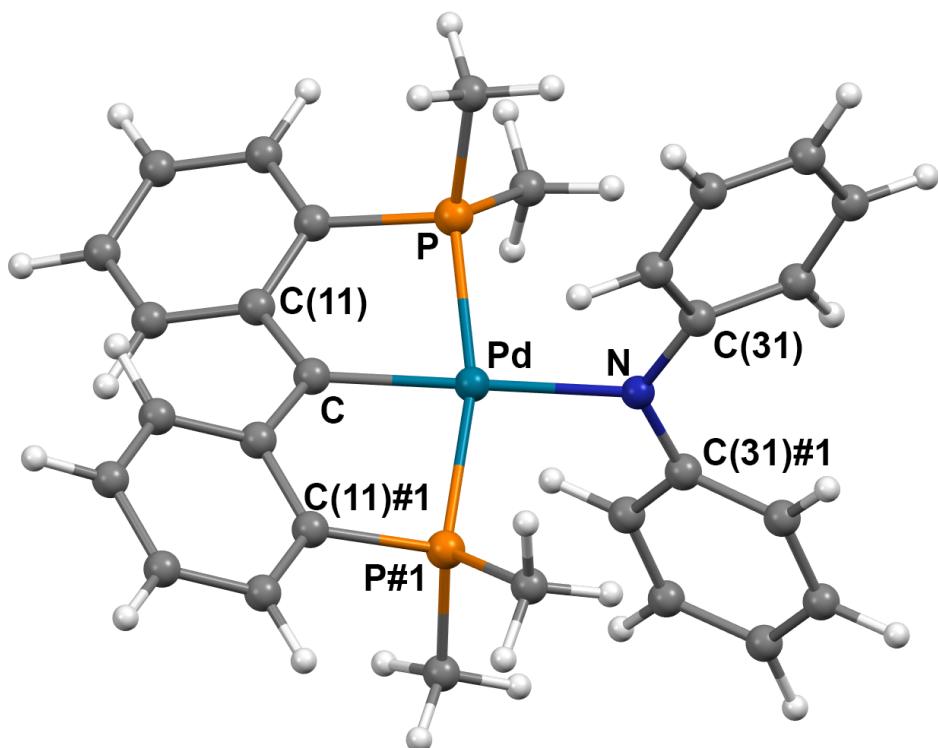


Figure S14. Optimized geometry for $\left[\{PC^{\bullet}(sp^2)P\}^MePdNPh_2\right]$ (**3'**).

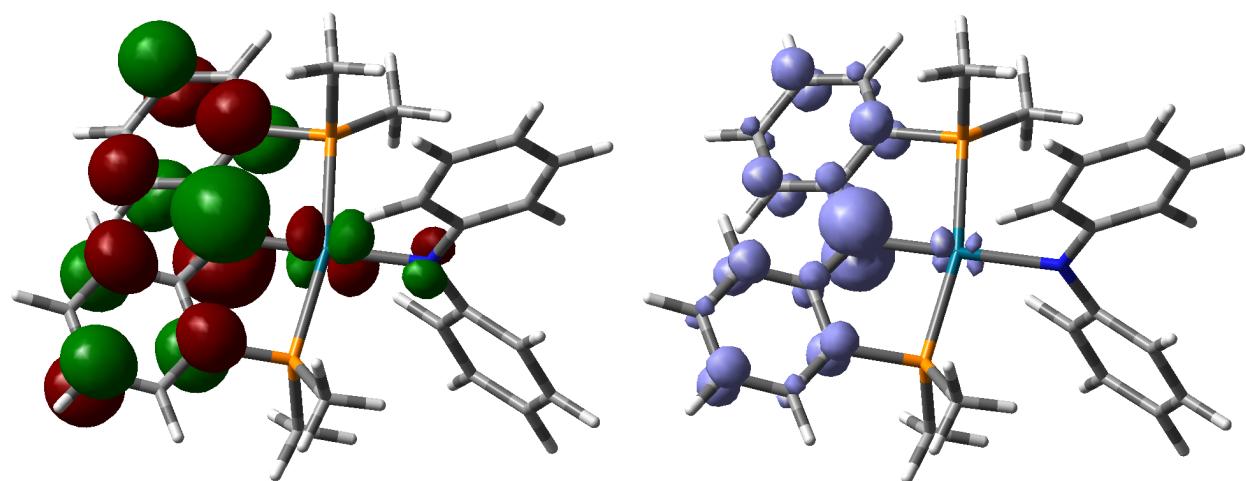


Figure S15. SOMO (left) and spin density (right) for $\left[\{PC^{\bullet}(sp^2)P\}^MePdNPh_2\right]$ (**3'**).

3.5 [{PC(sp²)P}^{Me}PdNPh]⁻ (**4'**)

Table S9. Optimized coordinates for [{PC(sp²)P}^{Me}PdNPh]⁻ (**4'**).

atom	x	y	z
C	-4.836109	-0.170071	-0.817446
C	-3.510597	-0.191309	-0.247653
C	-3.443169	-0.177755	1.190731
C	-4.601098	-0.144457	1.978093
C	-5.892375	-0.122065	1.398701
C	-5.983468	-0.136937	-0.011817
N	-2.371845	-0.225168	-1.006148
Pd	-0.319688	-0.068505	-0.424108
C	1.718517	0.096384	0.044445
C	2.441705	-1.111792	0.373797
C	1.889010	-2.426417	0.094030
C	2.568639	-3.623696	0.364533
C	3.837376	-3.622173	0.976984
C	4.380231	-2.362447	1.342929
C	3.721186	-1.164283	1.062508
P	0.180791	-2.363008	-0.610817
C	0.232058	-3.235646	-2.288051
H	0.612302	-4.258861	-2.179413
H	4.362297	-4.550021	1.192477
P	-0.316437	2.252493	-0.009231
C	-1.114340	3.387705	-1.286262
H	-2.186843	3.170378	-1.327589
C	1.497733	2.614127	-0.006962
C	2.315771	1.413771	-0.003154
C	3.737354	1.688697	-0.127930
C	4.255289	2.984698	-0.173879
C	3.420568	4.130097	-0.101848
C	2.030004	3.911530	-0.033947
H	3.833848	5.135765	-0.129016
C	-1.029299	2.921705	1.607940
H	-0.863471	4.003585	1.684533
C	-0.884518	-3.539661	0.407857
H	-0.836364	-3.226756	1.454927
H	-0.524498	2.418132	2.437776
H	-0.667967	3.179884	-2.263443
H	-1.922278	-3.466660	0.066712
H	0.895373	-2.671872	-2.950809
H	1.354901	4.769421	-0.018462
H	5.333032	3.113547	-0.286894

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Table S9. – continued from previous page

atom	x	y	z
H	4.421536	0.856393	-0.250941
H	-2.553697	-0.231673	-2.012540
H	2.098580	-4.576394	0.112181
H	5.331142	-2.323863	1.876939
H	4.157491	-0.239924	1.424669
H	-2.100607	2.698808	1.651305
H	-0.958721	4.443588	-1.032860
H	-0.529694	-4.574292	0.321098
H	-2.456912	-0.207178	1.650913
H	-4.498098	-0.138769	3.064189
H	-6.964514	-0.121665	-0.489803
H	-4.935182	-0.179948	-1.904552
H	-0.775256	-3.259754	-2.719191
H	-6.785936	-0.096779	2.018921

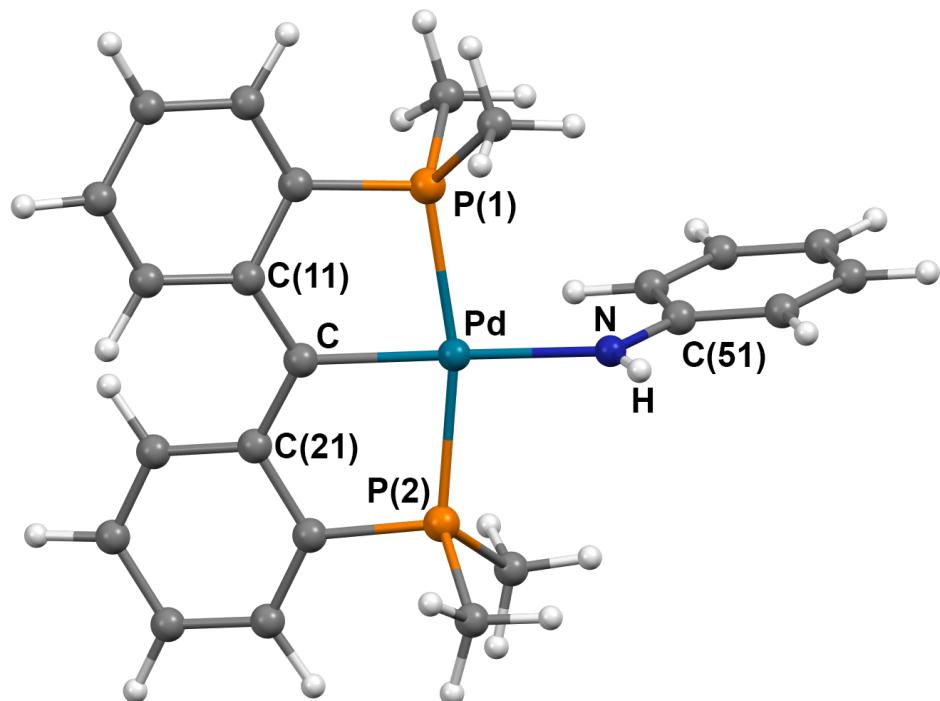


Figure S16. Optimized geometry for $[\{PC(sp^2)P\}^{\text{Me}}\text{PdNHPh}]^-$ (**4'**).

Table S10. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry of $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdNHPh}]^-$ (**4'**).

Distance	Calcd.	Angle	Calcd.
Pd–N	2.139	P(1)–Pd–P(2)	166.51
Pd–C	2.098	C–Pd–N	177.10
Pd–P(1)	2.358	C–Pd–P(1)	83.22
Pd–P(2)	2.357	C–Pd–P(2)	83.56
N–C(51)	1.369	N–Pd–P(1)	96.97
C–C(11)	1.447	N–Pd–P(2)	96.37
C–C(21)	1.446	Pd–N–C(51)	130.22

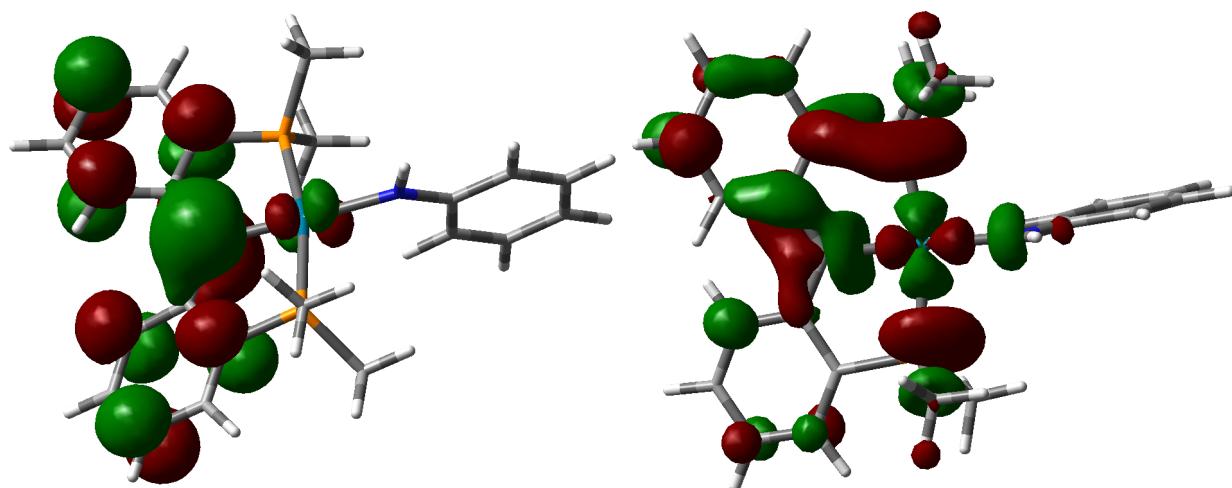


Figure S17. Frontier molecular orbitals for $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdNHPh}]^-$ (**4'**) (left: HOMO, right: LUMO).

3.6 [{PC(sp²)P}^{Me}PdNPh₂]⁻ (**5'**)

Table S11. Optimized coordinates for [{PC(sp²)P}^{Me}PdNPh₂]⁻ (**5'**).

atom	x	y	z
C	-2.145481	0.691039	-2.337995
C	-2.819230	0.580309	-1.080935
C	-4.123319	1.172432	-1.005822
C	-4.705179	1.805790	-2.115326
C	-4.029154	1.886586	-3.351465
C	-2.737049	1.324152	-3.440316
N	-2.150681	-0.000025	0.000009
C	-2.819194	-0.580361	1.080976
C	-4.123283	-1.172493	1.005904
C	-4.705107	-1.805847	2.115428
C	-4.029049	-1.886625	3.351549
C	-2.736949	-1.324173	3.440364
C	-2.145415	-0.691065	2.338020
Pd	0.049024	-0.000004	-0.000017
P	0.339739	-2.341170	-0.196792
C	-0.658549	-3.247601	-1.514298
H	-1.720921	-3.192111	-1.254737
P	0.339704	2.341169	0.196718
C	-0.658642	3.247651	1.514145
H	-1.721001	3.192165	1.254531
C	2.120233	2.430021	0.688958
C	2.816934	1.168705	0.519457
C	4.193725	1.201002	0.980096
C	4.792093	2.358358	1.482411
C	4.088828	3.587538	1.574322
C	2.736711	3.591249	1.177098
C	2.143225	0.000011	0.000013
C	2.816966	-1.168668	-0.519422
C	2.120287	-2.429992	-0.688962
C	2.736799	-3.591207	-1.177089
C	4.088930	-3.587473	-1.574269
C	4.792171	-2.358281	-1.482331
C	4.193770	-1.200939	-0.980026
H	4.565566	4.484919	1.961896
H	4.565694	-4.484844	-1.961832
C	0.135806	3.471549	-1.303256
H	0.398548	4.504297	-1.042110
H	0.806520	3.114743	-2.090522
H	-0.502473	2.750739	2.476051

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Table S11. – continued from previous page

atom	x	y	x
C	0.135780	-3.471609	1.303128
H	0.398538	-4.504346	1.041954
H	0.806456	-3.114831	2.090439
H	-0.502325	-2.750661	-2.476180
H	2.156948	4.511985	1.264725
H	5.825057	2.305084	1.829860
H	4.763166	0.277844	0.988680
H	-0.896173	3.422289	-1.664727
H	-0.350301	4.297457	1.592329
H	-4.660837	-1.156295	0.062393
H	-5.694903	-2.250682	2.009190
H	-4.487402	-2.374341	4.209332
H	-2.184373	-1.377558	4.378100
H	-1.144362	-0.271724	2.414482
H	2.157053	-4.511951	-1.264743
H	5.825145	-2.304989	-1.829749
H	4.763195	-0.277771	-0.988588
H	-0.896216	-3.422368	1.664554
H	-0.350214	-4.297408	-1.592498
H	-4.660848	1.156222	-0.062297
H	-5.694976	2.250613	-2.009058
H	-4.487535	2.374306	-4.209231
H	-2.184497	1.377553	-4.378066
H	-1.144424	0.271712	-2.414485

Table S12. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry of $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdNPh}_2]^-$ (**5'**).

Distance	Calcd.	Angle	Calcd.
Pd–N	2.200	P(1)–Pd–P(2)	165.89
Pd–C	2.094	C–Pd–N	180.00
Pd–P(1)	2.367	C–Pd–P(1)	82.95
Pd–P(2)	2.367	C–Pd–P(2)	82.95
N–C(3)	1.397	N–Pd–P(1)	97.05
N–C(4)	1.397	N–Pd–P(2)	97.05
C–C(1)	1.446	Pd–N–C(3)	118.59
C–C(2)	1.446	Pd–N–C(4)	118.59

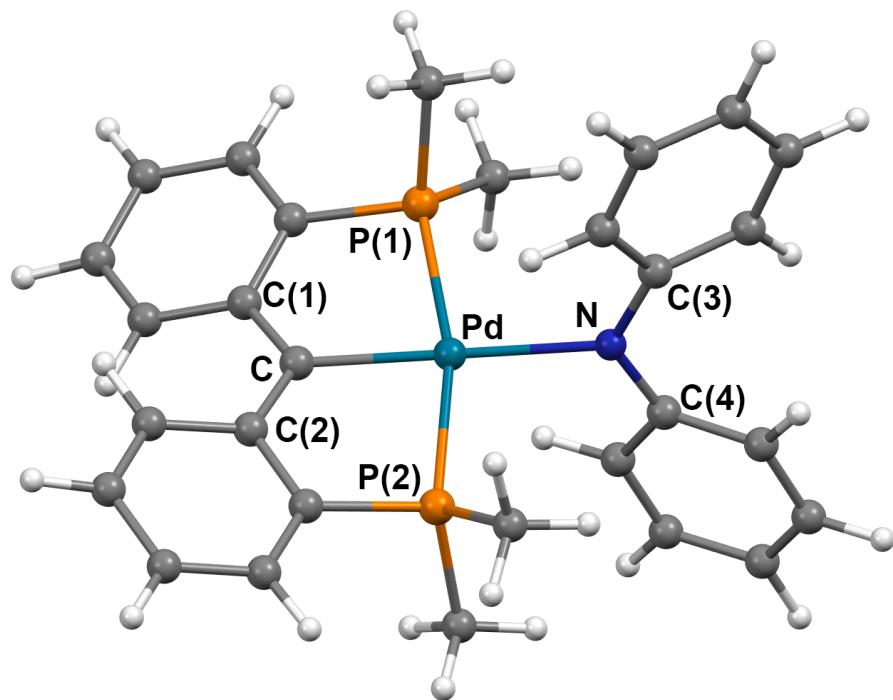


Figure S18. Optimized geometry for $\left[\{PC(sp^2)P\}^{\text{Me}}\text{PdNPh}_2\right]^-$ (**5'**).

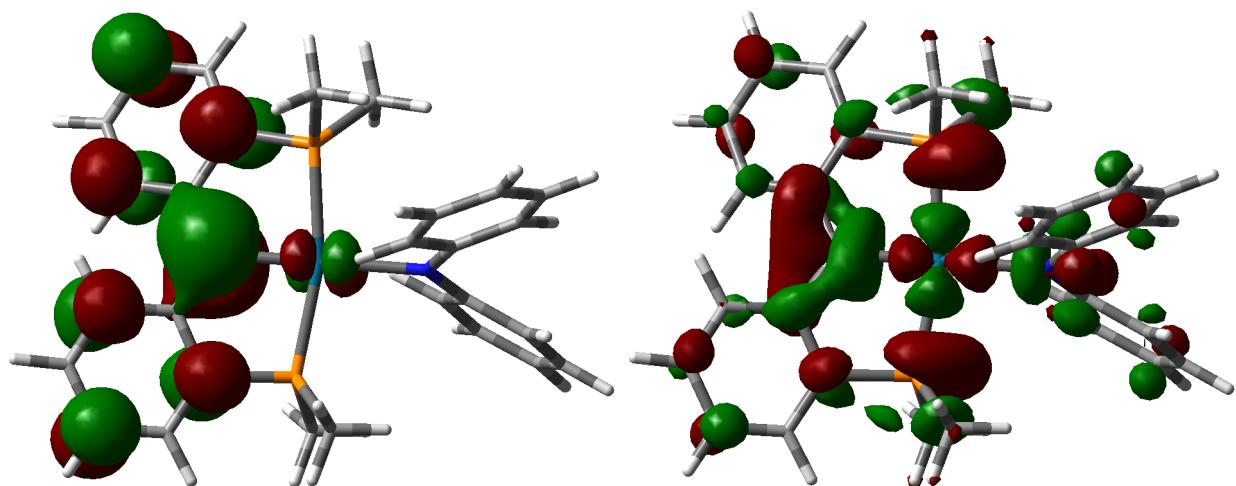


Figure S19. Frontier molecular orbitals for $\left[\{PC(sp^2)P\}^{\text{Me}}\text{PdNPh}_2\right]^-$ (**5'**) (left: HOMO, right: LUMO).

3.7 [{PC(sp²)-K(OEt₂)(C₆H₆)P}^{Me}PdNPh₂] (5’)

Table S13. Optimized coordinates for [{PC(sp²)-K(OEt₂)(C₆H₆)P}^{Me}PdNPh₂] (5’).

atom	x	y	z
Pd	1.219841	-0.385802	0.387587
N	3.045871	0.414415	-0.459697
C	4.211994	-0.364326	-0.380016
C	4.338498	-1.338516	0.653933
C	5.449069	-2.193924	0.727114
H	5.507273	-2.920872	1.536053
H	3.542794	-1.412421	1.391950
C	6.484640	-2.118612	-0.226381
H	7.346412	-2.778744	-0.166513
C	6.372924	-1.168723	-1.263974
H	7.151517	-1.101401	-2.022718
C	5.264720	-0.311109	-1.346422
H	5.195071	0.392681	-2.170993
C	3.035153	1.700170	-1.001603
C	1.811800	2.240617	-1.505782
C	1.728821	3.553360	-1.994836
C	2.857381	4.397746	-2.005957
C	4.074367	3.887802	-1.503675
C	4.169120	2.577296	-1.013231
H	5.116063	2.225985	-0.614413
H	4.959647	4.522196	-1.485896
H	2.794033	5.412974	-2.389812
H	0.778841	3.920783	-2.382655
H	0.938742	1.589356	-1.505585
P	1.244661	0.962530	2.320290
C	1.577359	2.812679	2.241559
C	2.396033	0.393963	3.703271
H	3.435414	0.468011	3.367371
H	2.249850	1.008654	4.599103
H	2.167328	-0.649332	3.938711
H	1.496559	3.268793	3.235407
H	2.583288	2.975098	1.843020
H	0.861864	3.281463	1.560635
P	0.874535	-2.071096	-1.248774
C	0.353035	-1.663540	-3.020971
C	2.269050	-3.302341	-1.538036
H	3.135119	-2.780776	-1.956245
H	2.557288	-3.736833	-0.577971
H	1.944916	-4.098974	-2.218098

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Table S13. – continued from previous page

atom	x	y	x
H	1.159427	-1.110494	-3.513695
H	-0.547618	-1.040605	-3.008752
H	0.144430	-2.582741	-3.581470
C	-0.530410	-1.159108	1.248893
C	-1.163312	-0.406785	2.319552
C	-0.484803	0.718529	2.941983
C	-1.071522	1.524274	3.927067
C	-2.395929	1.306038	4.361301
C	-3.110106	0.243681	3.760621
C	-2.531412	-0.580029	2.786518
C	-0.987782	-2.460246	0.784914
C	-0.512396	-3.022840	-0.465763
C	-0.991411	-4.234876	-0.988349
C	-1.942295	-5.006377	-0.292044
C	-2.349673	-4.545990	0.984422
C	-1.893080	-3.332142	1.506907
H	-0.495540	2.339691	4.365095
H	-2.846112	1.927182	5.130434
H	-4.140918	0.052790	4.060052
H	-3.143281	-1.369701	2.364337
H	-2.175718	-3.072266	2.520817
H	-3.013182	-5.162512	1.590171
H	-2.298392	-5.952403	-0.690309
H	-0.598922	-4.604931	-1.936253
K	-2.432296	0.561117	-0.279970
O	-2.794109	3.158047	-0.947951
C	-2.386857	3.752849	-2.229157
C	-3.262834	4.163464	0.020508
C	-3.729439	3.458464	1.291717
C	-1.926714	2.641904	-3.170449
H	-3.245426	4.296296	-2.655308
H	-1.574608	4.473165	-2.045965
H	-2.437149	4.859646	0.235979
H	-4.088287	4.732096	-0.437457
H	-2.913813	2.921457	1.789815
H	-4.543235	2.754763	1.073288
H	-4.113399	4.197635	2.006000
H	-2.729666	1.911939	-3.333317
H	-1.040234	2.130627	-2.776545
H	-1.652448	3.064992	-4.144648
C	-4.812989	-0.424760	-3.057974
C	-4.046211	-1.548695	-2.683078

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Table S13. – continued from previous page

atom	x	y	z
C	-4.202322	-2.119680	-1.401851
C	-5.132060	-1.562703	-0.497615
C	-5.898782	-0.438156	-0.870313
C	-5.738511	0.132308	-2.150796
H	-6.623429	-0.019709	-0.175522
H	-5.260561	-2.012204	0.483644
H	-3.614979	-2.987246	-1.113452
H	-3.343163	-1.989110	-3.386089
H	-4.703541	0.001063	-4.053058
H	-6.336189	0.992811	-2.442020

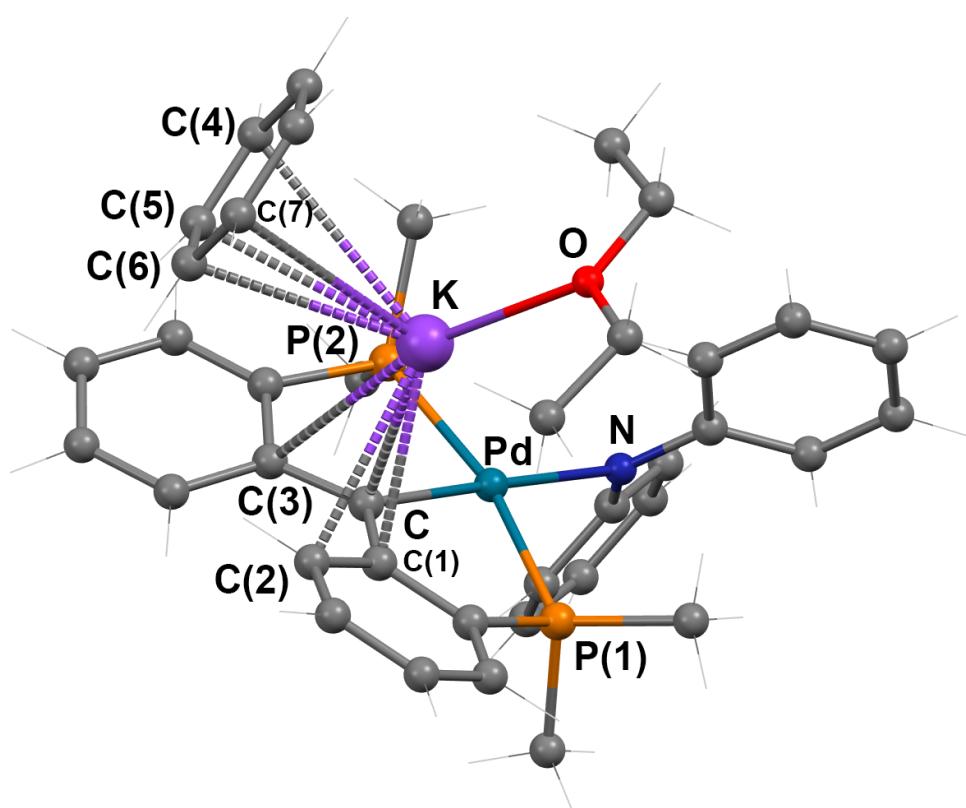


Figure S20. Optimized geometry for $\{PC(sp^2)\text{-K(OEt}_2\text{)}(C_6H_6)\text{P}\}^{Mc}\text{PdNPh}_2$ (**5''**).

Table S14. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry of $[\{\text{PC}(\text{sp}^2)\text{-K(OEt}_2\text{)}(\text{C}_6\text{H}_6)\text{P}\}^{\text{Me}}\text{PdNPh}_2]$ (**5''**).

Distance	Calcd.	Angle	Calcd.
Pd–N	2.166	P(1)–Pd–P(2)	166.62
Pd–C	2.098	C–Pd–N	178.78
Pd–P(1)	2.357	C–Pd–P(1)	83.28
Pd–P(2)	2.374	C–Pd–P(2)	84.26
K–O	2.706	N–Pd–P(1)	95.76
K–C	2.989	N–Pd–P(2)	96.62
K–C(1)	3.050	O–K–C	139.87
K–C(2)	3.273	K–C–Pd	96.25
K–C(3)	3.514		
K–C(4)	3.582		
K–C(5)	3.403		
K–C(6)	3.442		
K–C(7)	3.656		

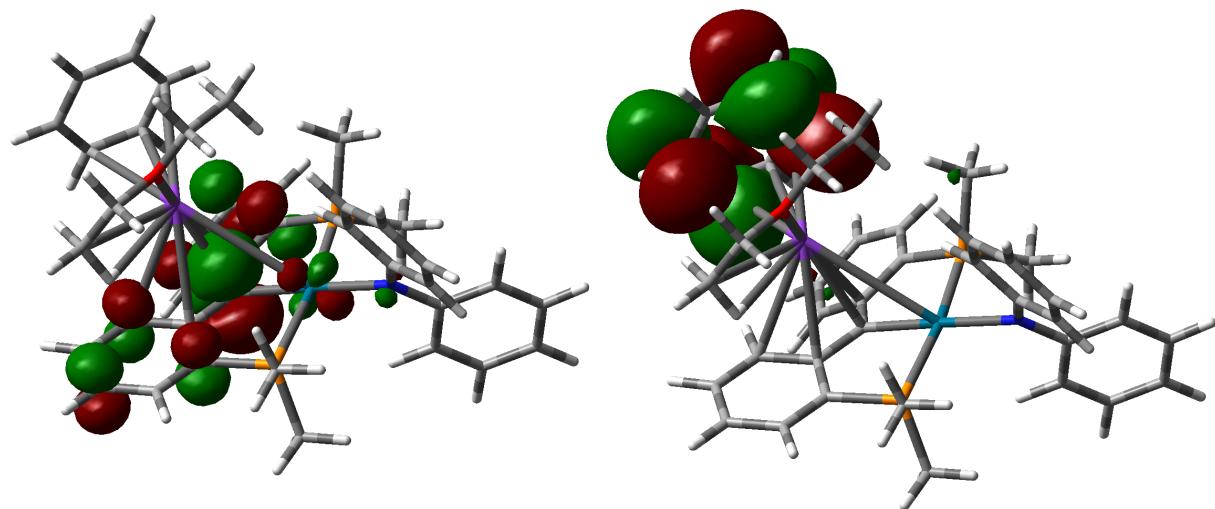


Figure S21. Frontier molecular orbitals for $[\{\text{PC}(\text{sp}^2)\text{-K(OEt}_2\text{)}(\text{C}_6\text{H}_6)\text{P}\}^{\text{Me}}\text{PdNPh}_2]$ (**5''**) (left: HOMO, right: LUMO).

3.8 $[\{PC(sp^2)P\}^{Me}PdCH_2Ph]^-$ (**6'**)

Table S15. Optimized coordinates for $[\{PC(sp^2)P\}^{Me}PdCH_2Ph]^-$ (**6'**).

atom	x	y	z
C	-3.964969	-1.335521	-1.199055
C	-3.283109	-1.458283	0.050811
C	-4.016540	-1.055368	1.209562
C	-5.327010	-0.555925	1.123095
C	-5.975358	-0.437459	-0.125857
C	-5.274696	-0.836998	-1.285622
C	-1.907229	-1.981722	0.136732
Pd	-0.184353	-0.558610	0.024603
C	1.590353	0.641580	-0.019718
C	2.857369	-0.022396	0.165257
C	2.979979	-1.466209	0.037217
C	4.191013	-2.158403	0.185690
C	5.382872	-1.484131	0.518503
C	5.297301	-0.081346	0.728143
C	4.102490	0.620174	0.562931
P	1.367690	-2.299397	-0.311331
C	1.516177	-3.085350	-2.027318
H	2.365828	-3.778604	-2.062907
H	6.321264	-2.020311	0.640755
P	-1.222665	1.542827	0.344760
C	-2.678739	2.097237	-0.719314
H	-3.543752	1.458017	-0.523667
C	0.159158	2.716649	-0.050733
C	1.445341	2.066075	-0.215818
C	2.495845	2.969088	-0.655958
C	2.292172	4.340572	-0.824183
C	1.033336	4.946210	-0.576116
C	-0.029486	4.097912	-0.203063
H	0.884238	6.016516	-0.700631
C	-1.819851	2.003152	2.080063
H	-2.102770	3.062401	2.119089
C	1.298934	-3.827958	0.798070
H	1.376444	-3.499094	1.838853
H	-1.001887	1.819713	2.783309
H	-2.394274	2.000898	-1.771568
H	0.341020	-4.340761	0.655467
H	1.672616	-2.282065	-2.753405
H	-1.021640	4.525544	-0.047128
H	3.123686	4.955324	-1.173477

Continued on next page

Table S15. – continued from previous page

atom	x	y	z
H	3.463282	2.554734	-0.918639
H	-1.687059	-2.682398	-0.681301
H	4.208901	-3.242813	0.057490
H	6.186071	0.466855	1.046046
H	4.089711	1.679118	0.797280
H	-2.679787	1.378929	2.344685
H	-2.934201	3.143209	-0.509399
H	2.122107	-4.517960	0.575444
H	-3.537981	-1.146664	2.184640
H	-5.849798	-0.262747	2.033784
H	-5.754556	-0.759118	-2.261277
H	-3.439795	-1.631224	-2.106768
H	0.590742	-3.618494	-2.274968
H	-6.990601	-0.051285	-0.192738
H	-1.695098	-2.452561	1.107942

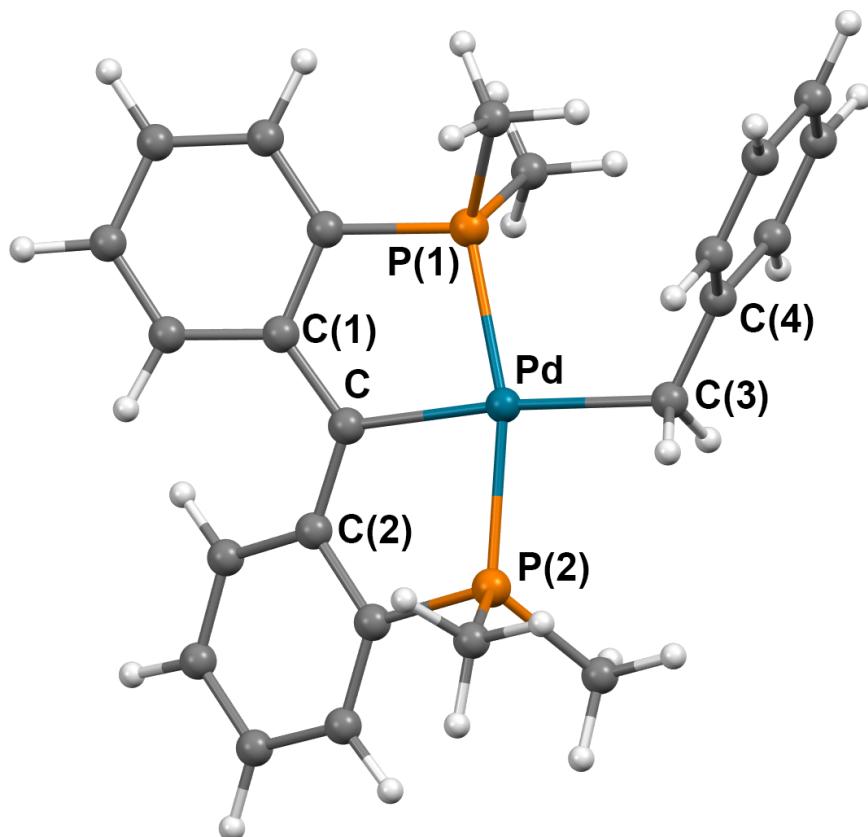


Figure S22. Optimized geometry for $[\{PC(sp^2)P\}^{\text{Me}}\text{PdCH}_2\text{Ph}]^-$ (**6'**).

Table S16. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry of $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdCH}_2\text{Ph}]^-$ (**6'**).

Distance	Calcd.	Angle	Calcd.
Pd–C(3)	2.237	P(1)–Pd–P(2)	164.72
Pd–C	2.143	C–Pd–C(3)	174.26
Pd–P(1)	2.366	C–Pd–P(1)	82.46
Pd–P(2)	2.356	C–Pd–P(2)	82.26
C(3)–C(4)	1.475	C(3)–Pd–P(1)	102.72
C–C(1)	1.445	C(3)–Pd–P(2)	92.55
C–C(2)	1.442	Pd–C(3)–C(4)	119.33

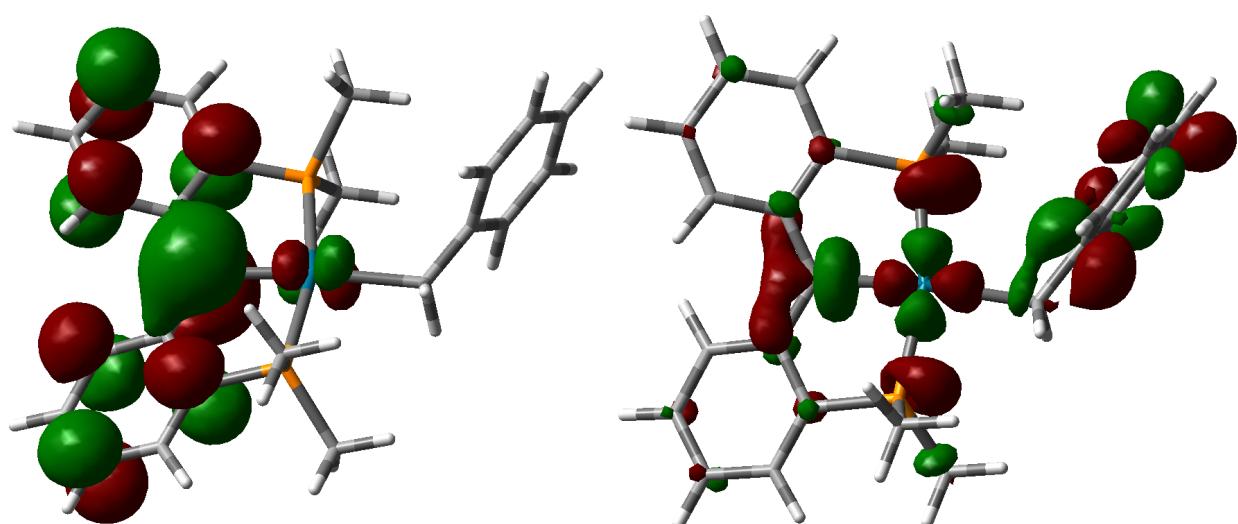


Figure S23. Frontier molecular orbitals for $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdCH}_2\text{Ph}]^-$ (**6'**) (left: HOMO, right: LUMO).

3.9 [{PC(sp²)-K(C₆H₆)P}^{Me}PdCH₂Ph]₂K⁺ (**6'**)

Table S17. Optimized coordinates for [{PC(sp²)-K(C₆H₆)P}^{Me}PdCH₂Ph]₂K⁺ (**6'**).

atom	x	y	z
C	-2.393398	-5.261673	0.406852
C	-3.630152	-4.619908	0.692934
C	-4.777920	-5.097098	0.002042
C	-4.693117	-6.142791	-0.935436
C	-3.453628	-6.753730	-1.214782
C	-2.304899	-6.306795	-0.528543
C	-3.718026	-3.515003	1.689709
Pd	-3.287318	-1.475904	1.023537
P	-3.694889	-1.678123	-1.302469
C	-5.431928	-2.018650	-1.963879
H	-6.147864	-1.314578	-1.527155
C	-2.891118	0.595678	0.542270
C	-2.821696	0.971657	-0.866228
C	-3.239898	0.033071	-1.901596
C	-3.252284	0.362279	-3.266170
C	-2.811632	1.619492	-3.729050
C	-2.294315	2.520452	-2.771431
C	-2.284886	2.213977	-1.403710
K	-5.660247	1.489097	0.680373
C	-2.862890	2.994140	1.566414
C	-2.779907	1.544093	1.654285
C	-2.695247	1.062484	3.026115
C	-2.595235	1.916252	4.135411
C	-2.621551	3.317385	3.990812
C	-2.782509	3.833613	2.687882
P	-2.803955	-0.784719	3.219437
C	-4.114157	-1.101193	4.539723
H	-5.065032	-0.659600	4.226078
H	-2.541074	3.972668	4.852671
H	-2.824082	1.864111	-4.787060
K	0.001563	0.007908	0.004872
C	2.889179	-0.595524	-0.535368
C	2.824029	-0.964868	0.875180
C	3.248102	-0.022371	1.904632
C	3.265233	-0.345388	3.270637
C	2.823953	-1.599614	3.740927
C	2.300895	-2.503789	2.789486
C	2.286702	-2.203549	1.420409
K	5.654430	-1.502218	-0.677034

Continued on next page

Table S17. – continued from previous page

atom	x	y	x
C	2.850693	-2.998534	-1.548382
C	2.771378	-1.548653	-1.642615
C	2.682915	-1.073184	-3.016386
C	2.576134	-1.931819	-4.121264
C	2.598811	-3.332349	-3.970269
C	2.763422	-3.842983	-2.665597
P	2.797246	0.772640	-3.219082
C	4.107953	1.077285	-4.541747
H	5.057738	0.635326	-4.225323
H	2.513013	-3.991383	-4.828746
P	3.703370	1.685278	1.295925
C	5.443822	2.026656	1.947623
H	6.155938	1.316833	1.513998
H	2.840053	-1.839414	4.799991
C	8.507339	-3.228185	0.425452
C	7.462680	-4.169466	0.556900
C	8.512830	-2.065165	1.226701
C	7.470763	-1.841471	2.153454
Pd	3.287104	1.473170	-1.027485
C	3.718274	3.508745	-1.704230
C	3.631670	4.618353	-0.712554
C	4.780479	5.099628	-0.026251
C	4.696765	6.149533	0.906619
C	3.457363	6.760721	1.185755
C	2.307577	6.309763	0.503905
C	2.394986	5.260456	-0.426877
C	1.234299	1.343210	-4.107605
H	1.124482	0.809568	-5.058539
C	2.678695	2.870485	2.345195
H	2.878266	2.704326	3.409821
H	0.350958	1.155925	-3.491606
H	4.247060	2.155962	-4.667671
H	1.613671	2.708803	2.153056
H	5.730815	3.042740	1.663318
C	-8.502469	3.232935	-0.415585
C	-7.453846	4.170442	-0.542408
C	-8.512856	2.073907	-1.222554
C	-7.471726	1.850394	-2.150428
C	-1.239229	-1.354631	4.105239
H	-1.131677	-0.826191	5.059319
C	-2.664211	-2.857233	-2.352771
H	-2.861145	-2.687979	-3.417409

Continued on next page

Table S17. – continued from previous page

atom	x	y	x
H	-0.356571	-1.160172	3.490474
H	-4.250460	-2.181037	4.658405
H	-1.600158	-2.693610	-2.156985
H	-5.717465	-3.037919	-1.689679
H	2.491849	-1.510769	-5.123188
H	2.822169	-4.920080	-2.513223
H	2.988664	-3.469956	-0.582230
H	4.734243	3.407210	-2.115989
H	3.013855	3.676173	-2.531182
H	3.608654	0.395346	3.993228
H	1.822828	-2.922894	0.755254
H	1.883689	-3.453901	3.121532
H	1.303107	2.418928	-4.302129
H	3.809951	0.635899	-5.499372
H	5.752560	4.663516	-0.256775
H	5.598841	6.504250	1.401869
H	3.392448	7.578073	1.899418
H	1.346264	6.784084	0.692711
H	1.498311	4.932101	-0.951817
H	2.928283	3.900221	2.078383
H	5.467335	1.935432	3.039565
H	-2.513430	1.490879	5.135715
H	-2.843825	4.911233	2.540280
H	-2.998727	3.469666	0.601967
H	-4.734400	-3.415994	2.101090
H	-3.014226	-3.686017	2.516474
H	-3.591417	-0.375866	-3.993423
H	-1.825503	2.931451	-0.733433
H	-1.877845	3.472961	-3.097481
H	-1.303900	-2.431756	4.293288
H	-3.817516	-0.665556	5.500377
H	-5.750081	-4.661045	0.232347
H	-5.594447	-6.494515	-1.434174
H	-3.387899	-7.567883	-1.932021
H	-1.343555	-6.780967	-0.717545
H	-1.497512	-4.936439	0.935070
H	-2.912071	-3.888487	-2.090145
H	-5.451161	-1.918013	-3.055072
C	-6.418542	2.783249	-2.272623
C	-6.413540	3.945018	-1.470358
H	-5.614734	4.673594	-1.580647
H	-5.614806	2.610119	-2.983543

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Table S17. – continued from previous page

atom	x	y	z
H	-7.491987	0.971057	-2.789287
H	-9.337898	1.369339	-1.150119
H	-9.317376	3.419108	0.279565
H	-7.460880	5.077541	0.056846
C	6.421407	-3.943862	1.483741
C	6.421476	-2.778110	2.280237
H	9.323053	-3.414365	-0.268745
H	7.473622	-5.079538	-0.037777
H	9.334898	-1.357499	1.150766
H	5.625819	-4.675404	1.597738
H	7.487204	-0.958911	2.787948
H	5.617054	-2.604660	2.990310

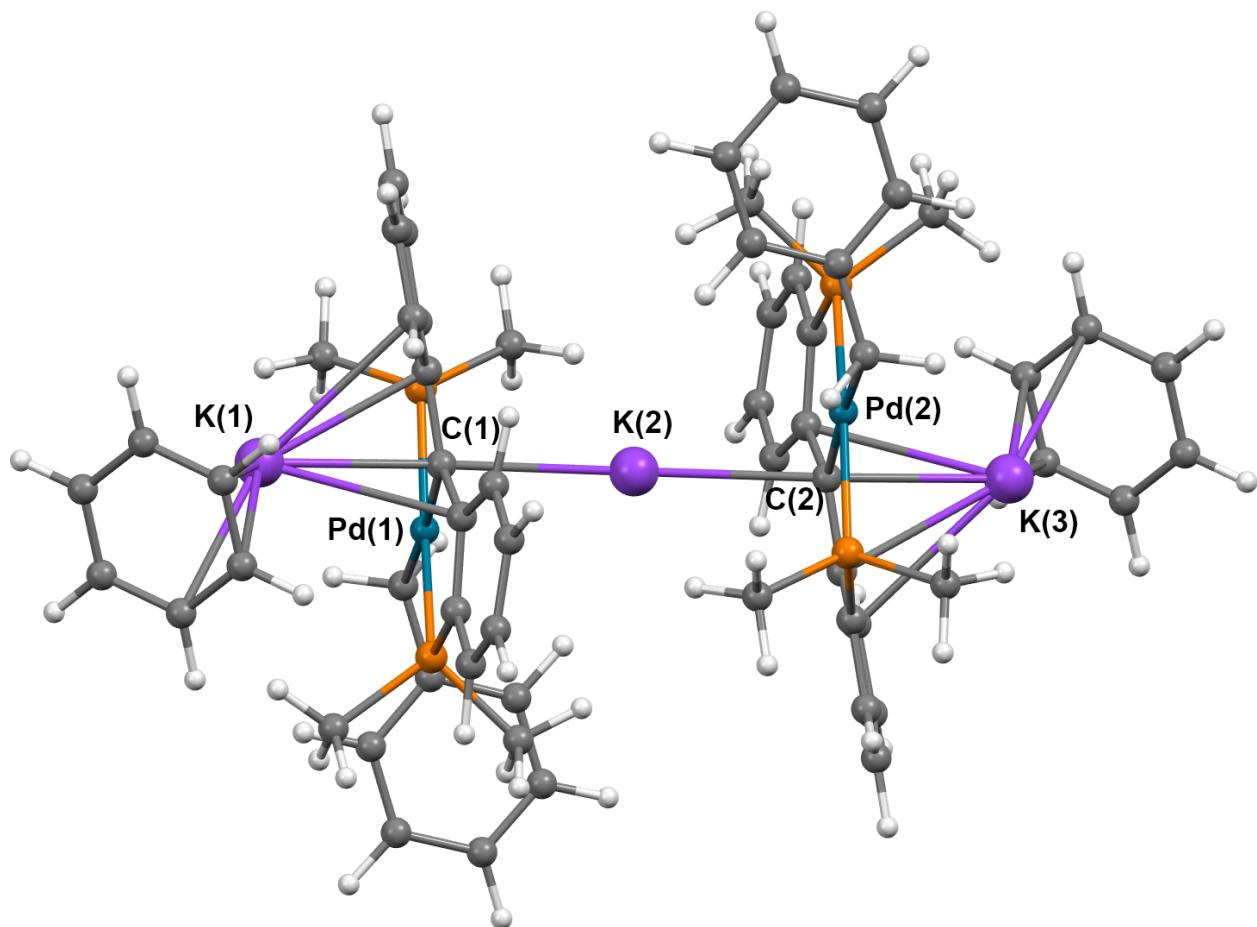


Figure S24. Optimized geometry for $\{[\text{PC}(\text{sp}^2)\text{-K}(\text{C}_6\text{H}_6)\text{P}]^{\text{Me}}\text{PdCH}_2\text{Ph}\}_2\text{K}^+$ (**6''**).

Table S18. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry of $[\{\text{PC}(\text{sp}^2)\text{-K(C}_6\text{H}_6\text{)P}\}^{\text{Me}}\text{PdCH}_2\text{Ph}]_2\text{K}^+$ (**6''**).

Distance	Calcd.	Angle	Calcd.
Pd(1)–C(1)	2.163	K(1)–C(1)–K(2)	170.13
Pd(2)–C(2)	2.163	C(1)–K(2)–C(2)	179.68
K(1)–C(1)	2.914	K(2)–C(2)–K(3)	170.10
K(2)–C(1)	2.999	K(1)–C(1)–Pd(1)	96.43
K(2)–C(2)	3.000	K(3)–C(2)–Pd(2)	96.26
K(3)–C(2)	2.913	Pd(1)–C(1)–K(2)	91.47
		Pd(2)–C(2)–K(2)	91.65

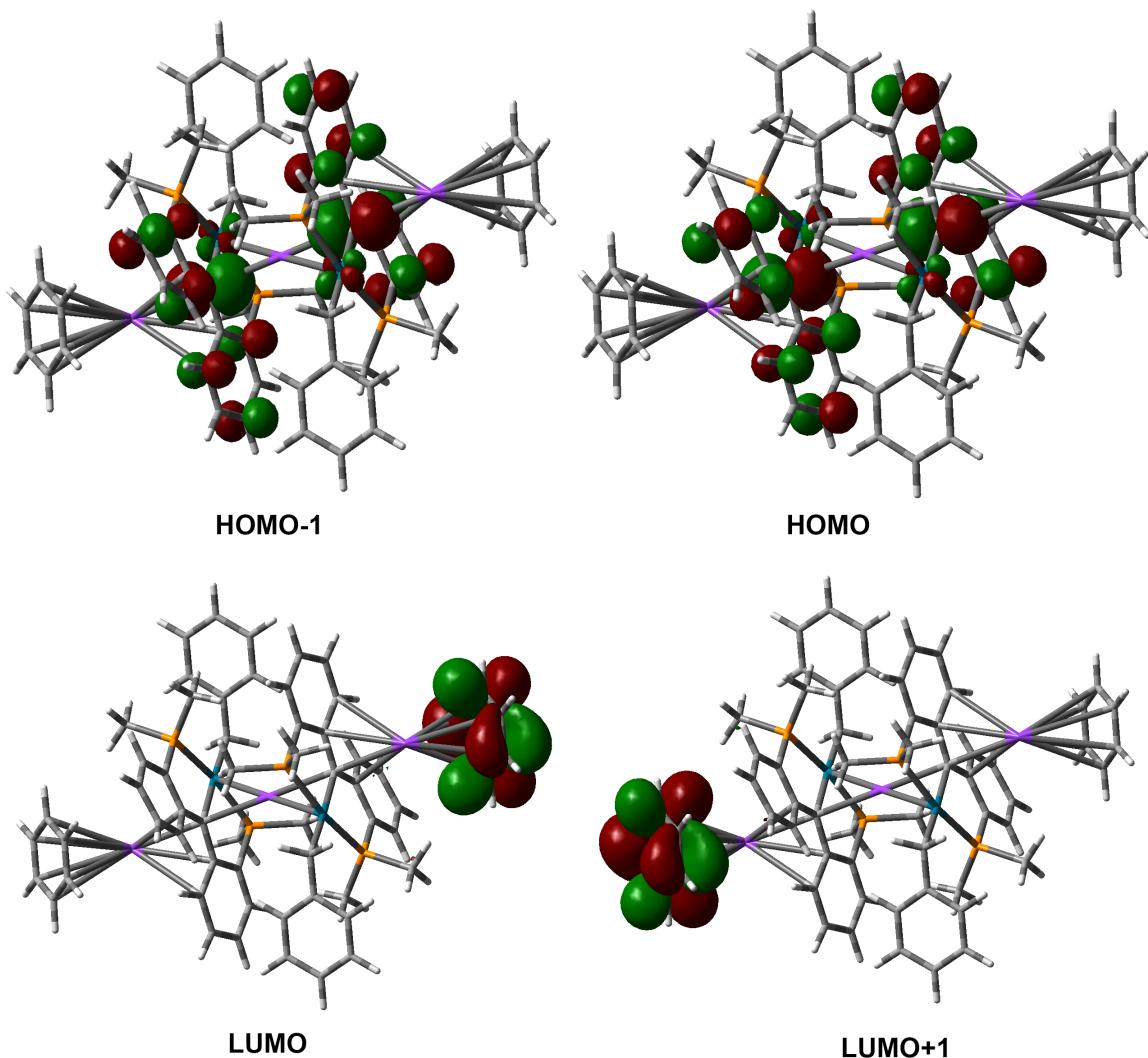


Figure S25. Frontier molecular orbitals for $[\{\text{PC}(\text{sp}^2)\text{-K(C}_6\text{H}_6\text{)P}\}^{\text{Me}}\text{PdCH}_2\text{Ph}]_2\text{K}^+$ (**6''**) (top left: HOMO-1, top right: HOMO, bottom left: LUMO, bottom right: LUMO+1).

3.10 [{PC[•](sp²)P}^{Me}PdCH₂Ph] (7')

Table S19. Optimized coordinates for [{PC[•](sp²)P}^{Me}PdCH₂Ph] (7').

atom	x	y	z
C	-4.347685	-0.536723	-0.580230
C	-3.216833	-1.281455	-0.145991
C	-3.244359	-1.772121	1.189902
C	-4.330930	-1.521964	2.044120
C	-5.437288	-0.768306	1.596635
C	-5.436904	-0.281315	0.274509
C	-2.054142	-1.546242	-1.045002
Pd	-0.257509	-0.445055	-0.501144
C	1.507628	0.570449	0.063499
C	2.638404	-0.220821	0.550299
C	2.710192	-1.618182	0.216965
C	3.765741	-2.431369	0.651993
C	4.780746	-1.905749	1.478198
C	4.707087	-0.552168	1.871754
C	3.667921	0.271995	1.419637
P	1.258663	-2.246565	-0.782120
C	1.935664	-2.595246	-2.509130
H	2.764582	-3.311131	-2.464255
H	5.592314	-2.540222	1.825004
P	-1.234768	1.687879	-0.127769
C	-2.344170	2.502562	-1.413186
H	-3.268783	1.925301	-1.496009
C	0.304269	2.757303	-0.087784
C	1.542754	2.030395	-0.022796
C	2.741301	2.812286	-0.124327
C	2.703022	4.209465	-0.223171
C	1.471777	4.899385	-0.221074
C	0.274142	4.156975	-0.162170
H	1.444444	5.983969	-0.288527
C	-2.138886	2.015437	1.492291
H	-2.374577	3.081245	1.593561
C	0.898014	-3.973305	-0.120713
H	0.667493	-3.903953	0.946406
H	-1.496879	1.703962	2.321080
H	-1.828533	2.502818	-2.377939
H	0.028526	-4.384159	-0.644227
H	2.290337	-1.655492	-2.941602
H	-0.677856	4.684175	-0.195032
H	3.633958	4.765905	-0.312567

Continued on next page

Table S19. – continued from previous page

atom	x	y	z
H	3.700626	2.307316	-0.170337
H	-2.265733	-1.276680	-2.089489
H	3.800472	-3.482864	0.371719
H	5.459840	-0.144135	2.543430
H	3.616064	1.296167	1.774048
H	-3.060132	1.425494	1.514909
H	-2.587263	3.532889	-1.129821
H	1.752515	-4.643786	-0.265242
H	-2.394354	-2.347364	1.554004
H	-4.318646	-1.914353	3.059550
H	-6.291098	0.283180	-0.096038
H	-4.382370	-0.185962	-1.611413
H	1.136126	-2.997026	-3.140649
H	-6.279878	-0.577259	2.257055
H	-1.759006	-2.602898	-1.001993

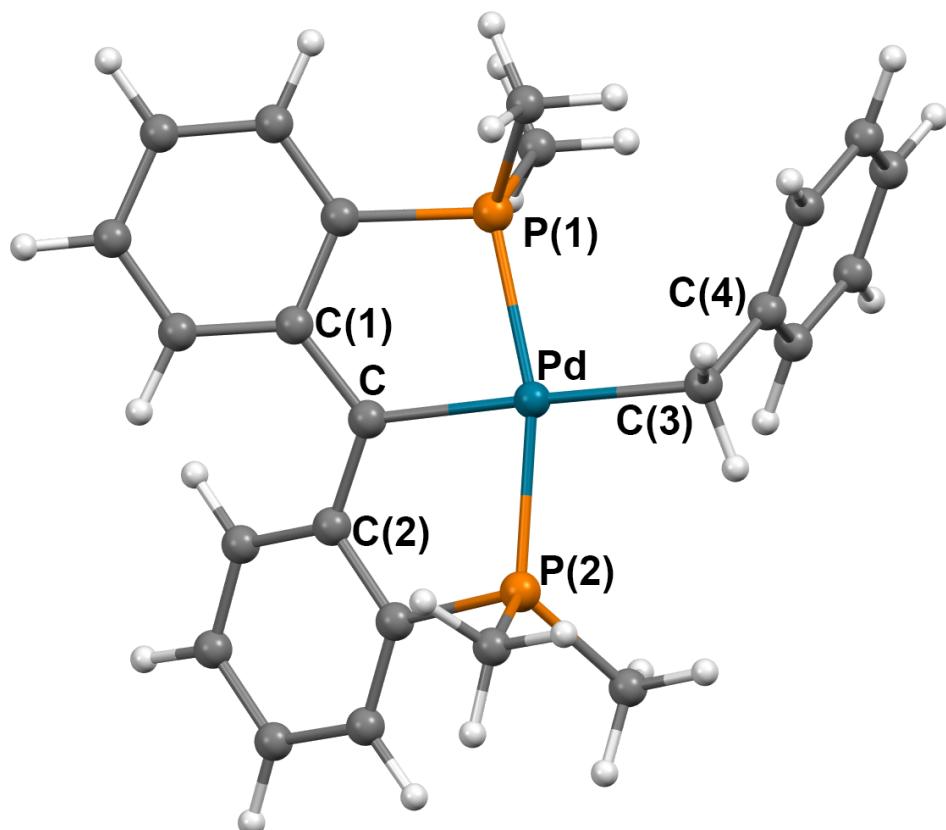


Figure S26. Optimized geometry for $\left[\{PC^{\bullet}(sp^2)P\}^{Me}PdCH_2Ph\right] (7')$.

Table S20. Selected distances (\AA) and angles ($^\circ$) for the optimized geometry of $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdCH}_2\text{Ph}]^-$ (**6'**).

Distance	Calcd.	Angle	Calcd.
Pd–C(3)	2.176	P(1)–Pd–P(2)	164.52
Pd–C	2.113	C–Pd–C(3)	178.15
Pd–P(1)	2.376	C–Pd–P(1)	82.54
Pd–P(2)	2.371	C–Pd–P(2)	82.11
C(3)–C(4)	1.493	C(3)–Pd–P(1)	98.86
C–C(1)	1.463	C(3)–Pd–P(2)	96.54
C–C(2)	1.463	Pd–C(3)–C(4)	113.74

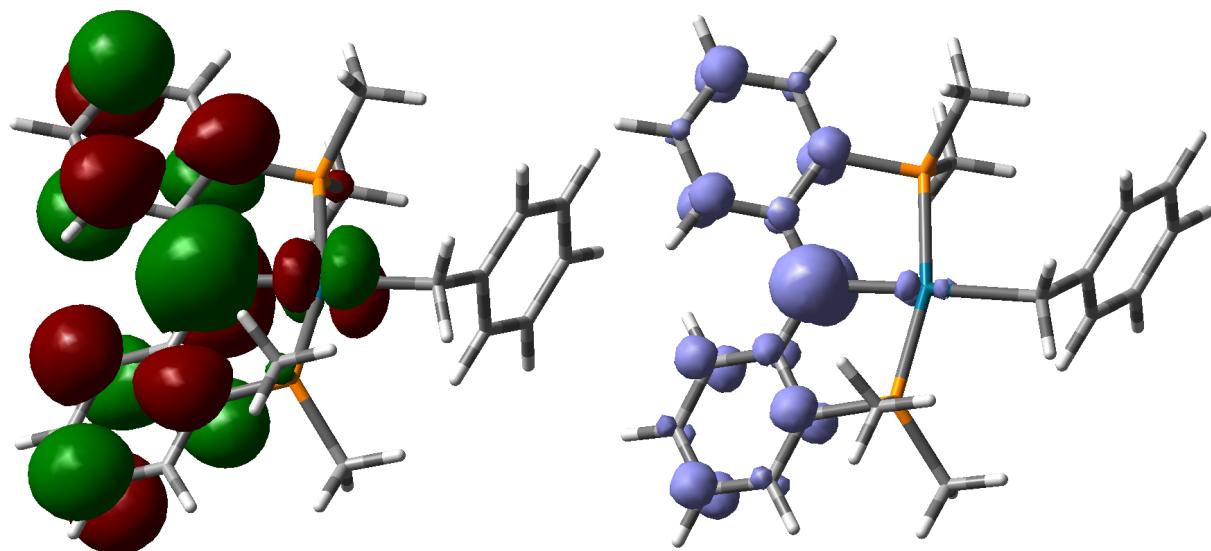


Figure S27. SOMO (left) and spin density (right) for $[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{\text{Me}}\text{PdCH}_2\text{Ph}]$ (**7'**).

4 NMR Spectra

4.1 NMR Spectra for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNH}^p\text{Tol}\right]^- \text{K}^+$ (4)

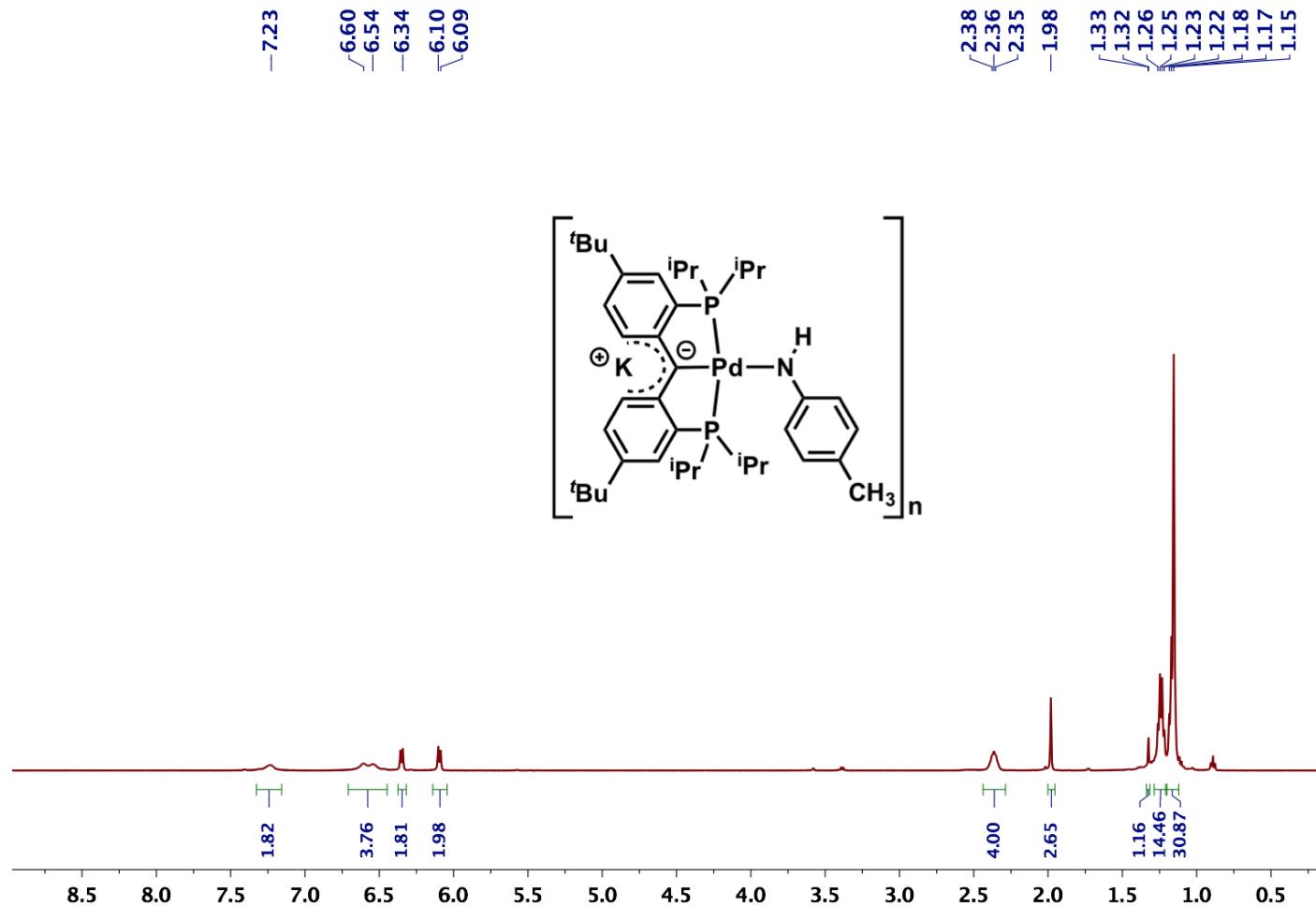


Figure S28. ^1H NMR spectrum for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNH}^p\text{Tol}\right]^- \text{K}^+$ (4).

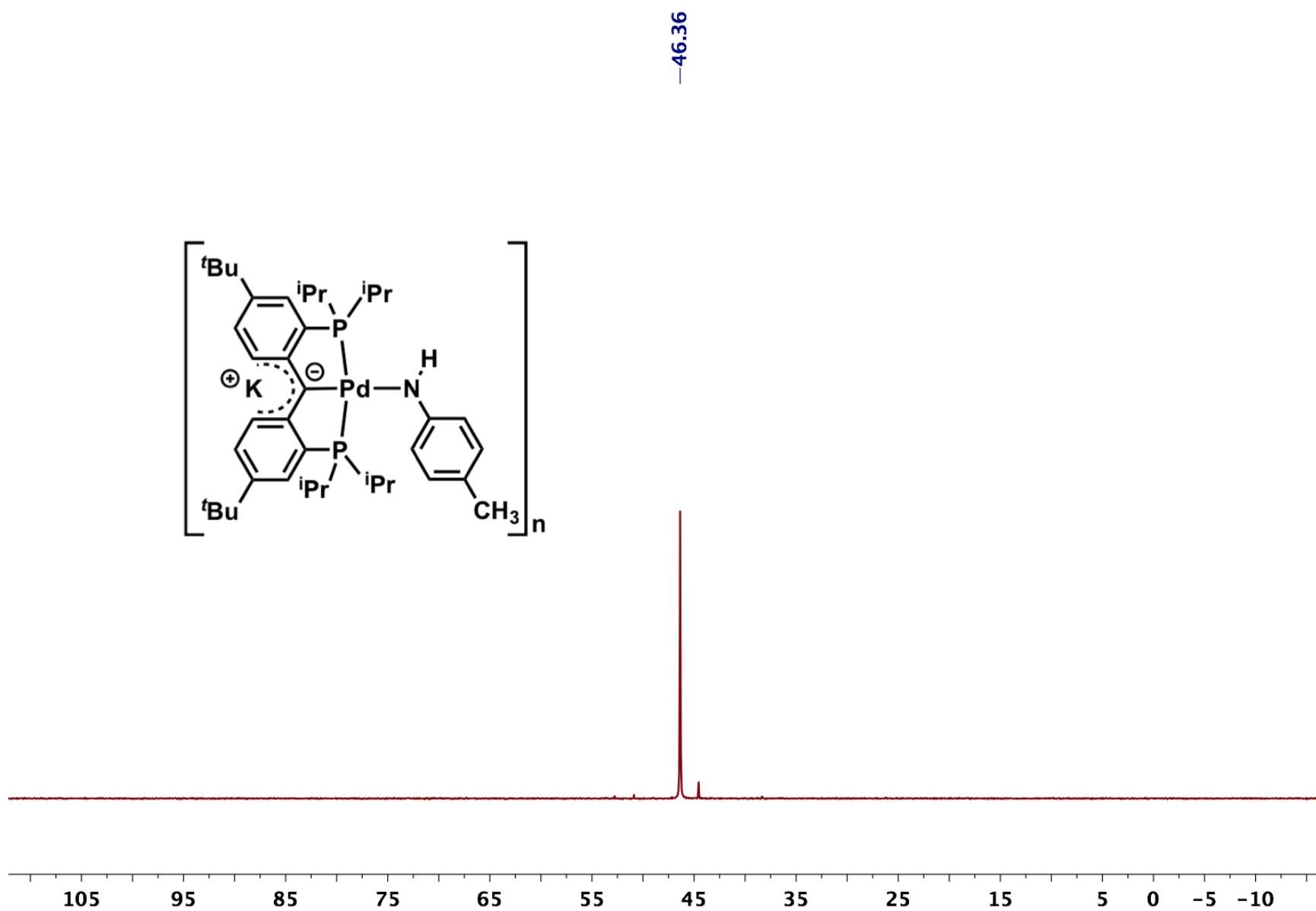


Figure S29. $^{31}\text{P}\{\text{H}\}$ NMR spectrum for $[\{\text{PC}(\text{sp}^2)\text{P}\}^t\text{Bu}\text{PdNH}^p\text{Tol}]^-\text{K}^+$ (**4**).

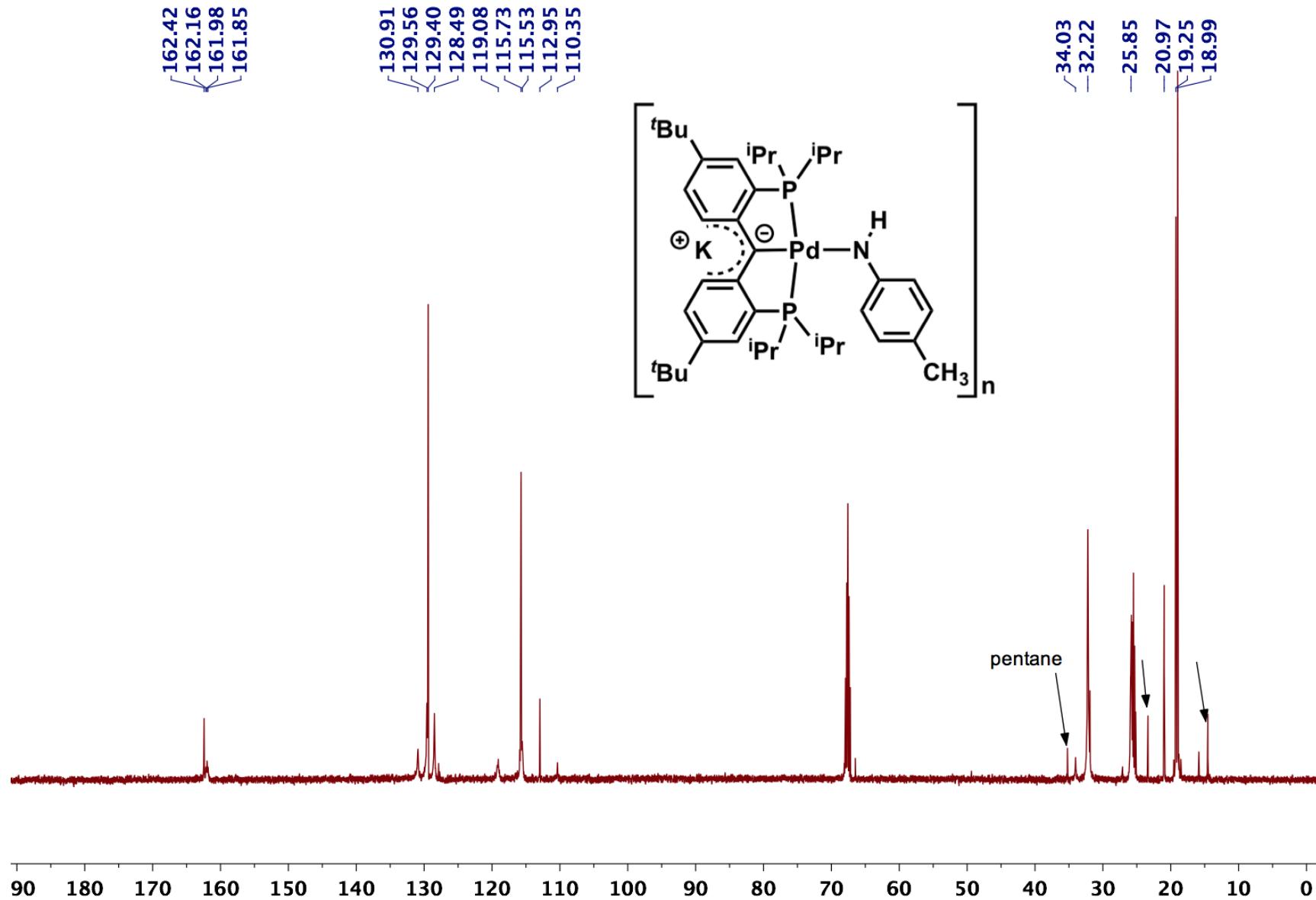


Figure S30. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for $\left[\left\{ \text{PC}(\text{sp}^2)\text{P} \right\}^3\text{BuPdNH}^p\text{Tol} \right]^- \text{K}^+$ (**4**).

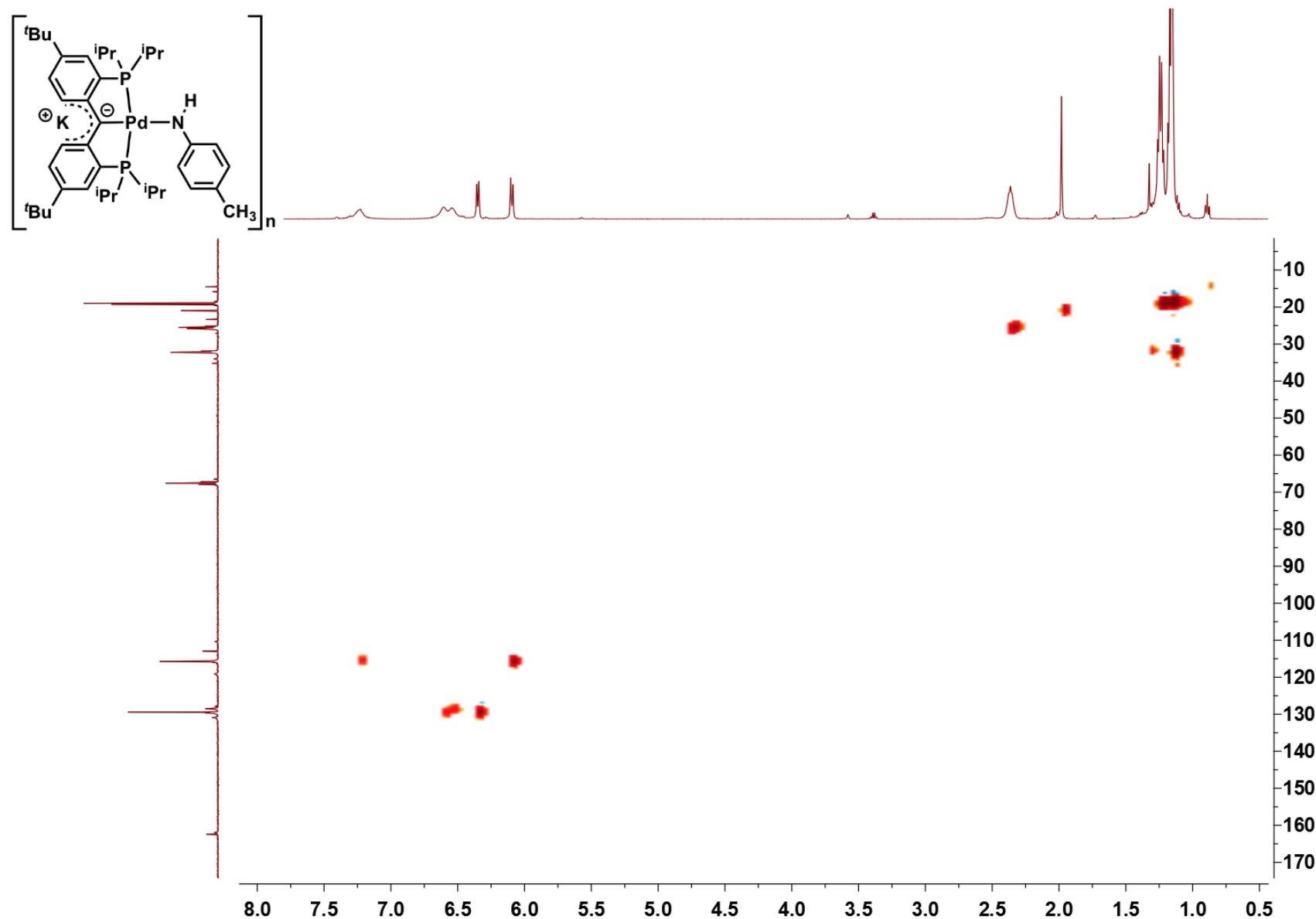


Figure S31. ^1H - ^{13}C HSQC NMR spectrum for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{iBu}}\text{PdNH}^p\text{Tol}\right]^- \text{K}^+$ (**4**).

4.2 NMR Spectra for $\left[\{PC(sp^2)P\}^{tBu}PdNPh_2\right]^- [KOEt_2]^+ (5)$

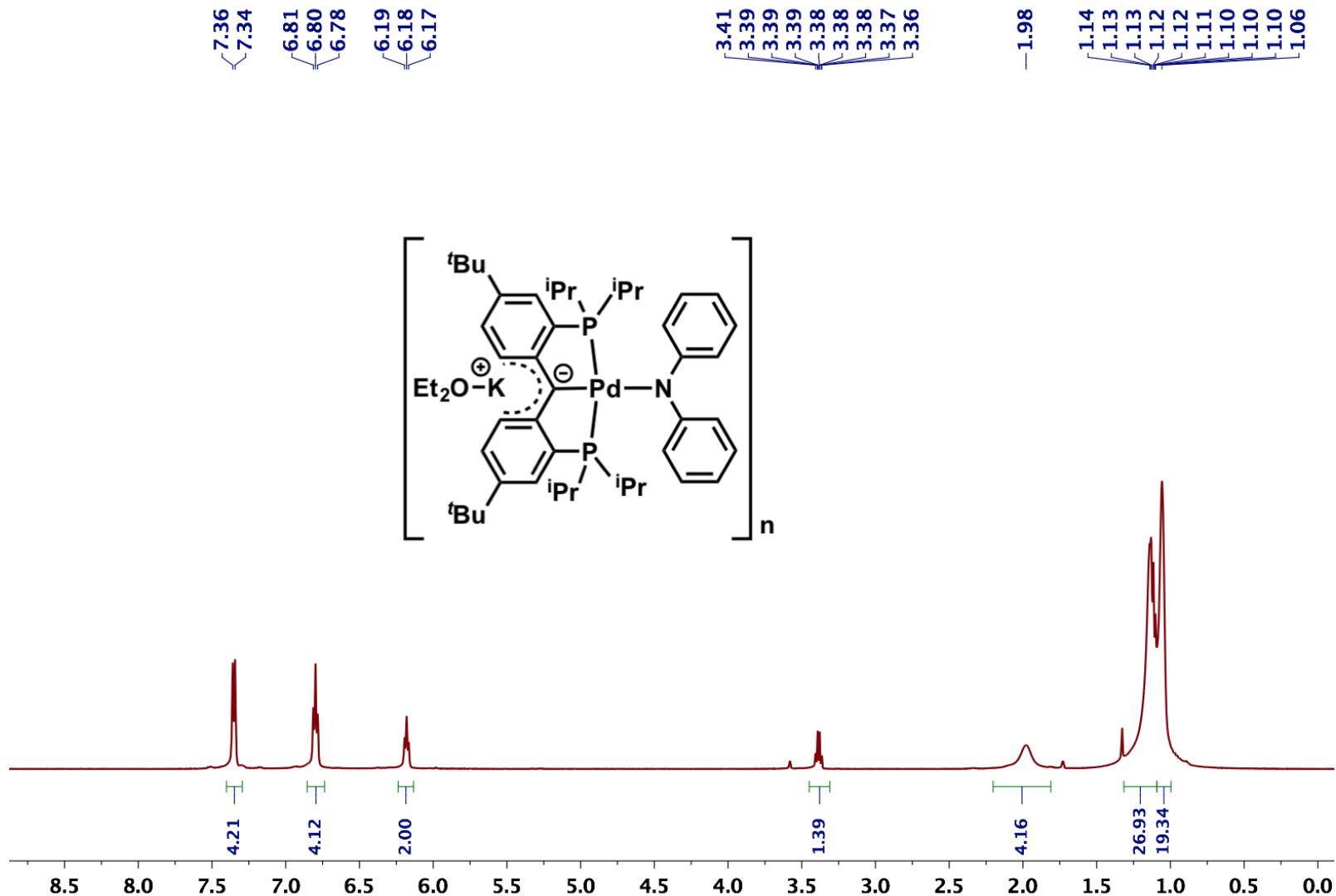


Figure S32. 1H NMR spectrum for $\left[\{PC(sp^2)P\}^{tBu}PdNPh_2\right]^- [KOEt_2]^+ (5)$.

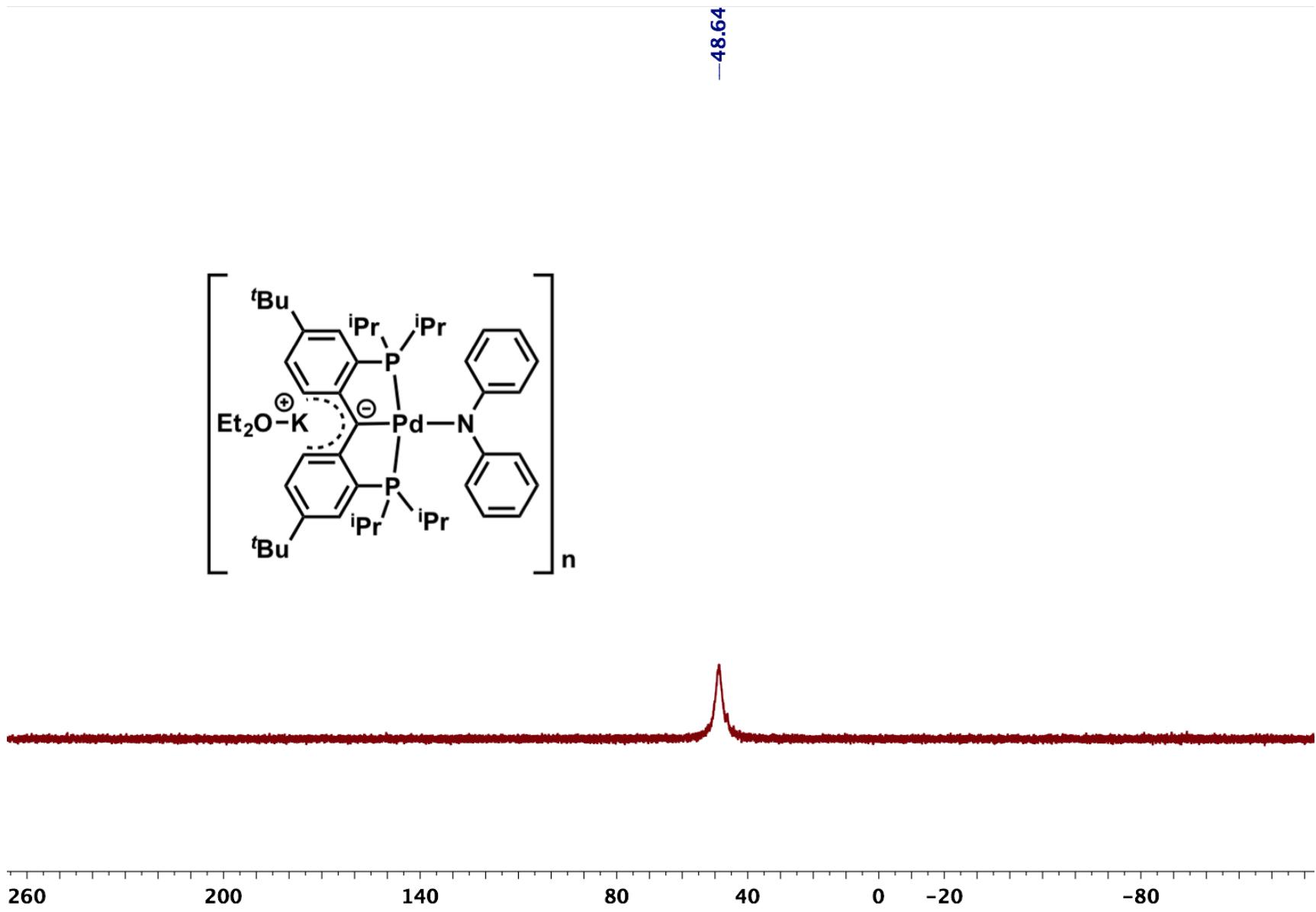


Figure S33. $^{31}\text{P}\{\text{H}\}$ NMR spectrum for $\{[\text{PC}(\text{sp}^2)\text{P}]^{\text{tBu}}\text{PdNPh}_2\}^-[\text{KOEt}_2]^+$ (**5**).

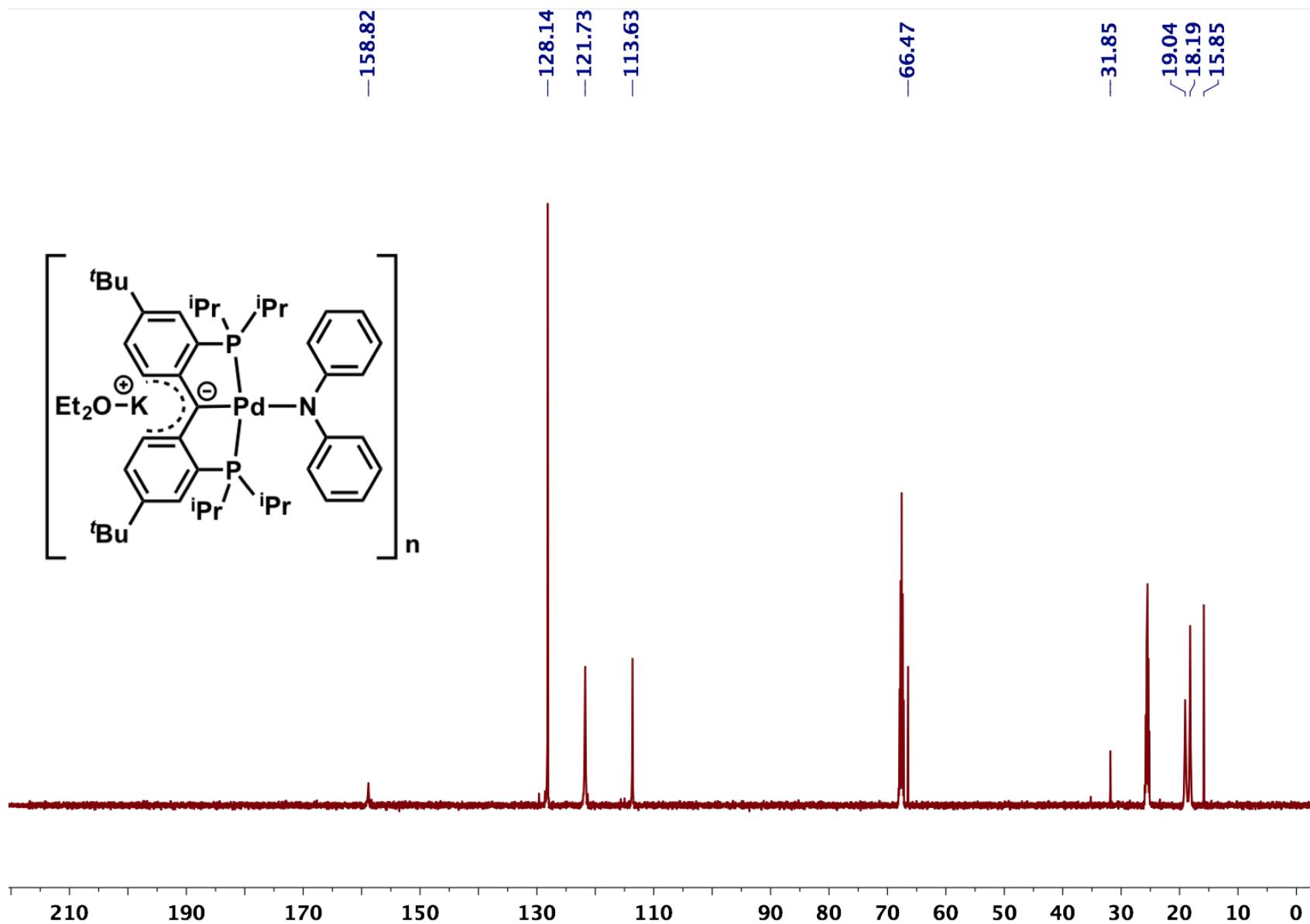


Figure S34. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for $\{[\text{PC}(\text{sp}^2)\text{P}]^{\text{tBu}}\text{PdNPh}_2\}^-[\text{KOEt}_2]^+$ (**5**).

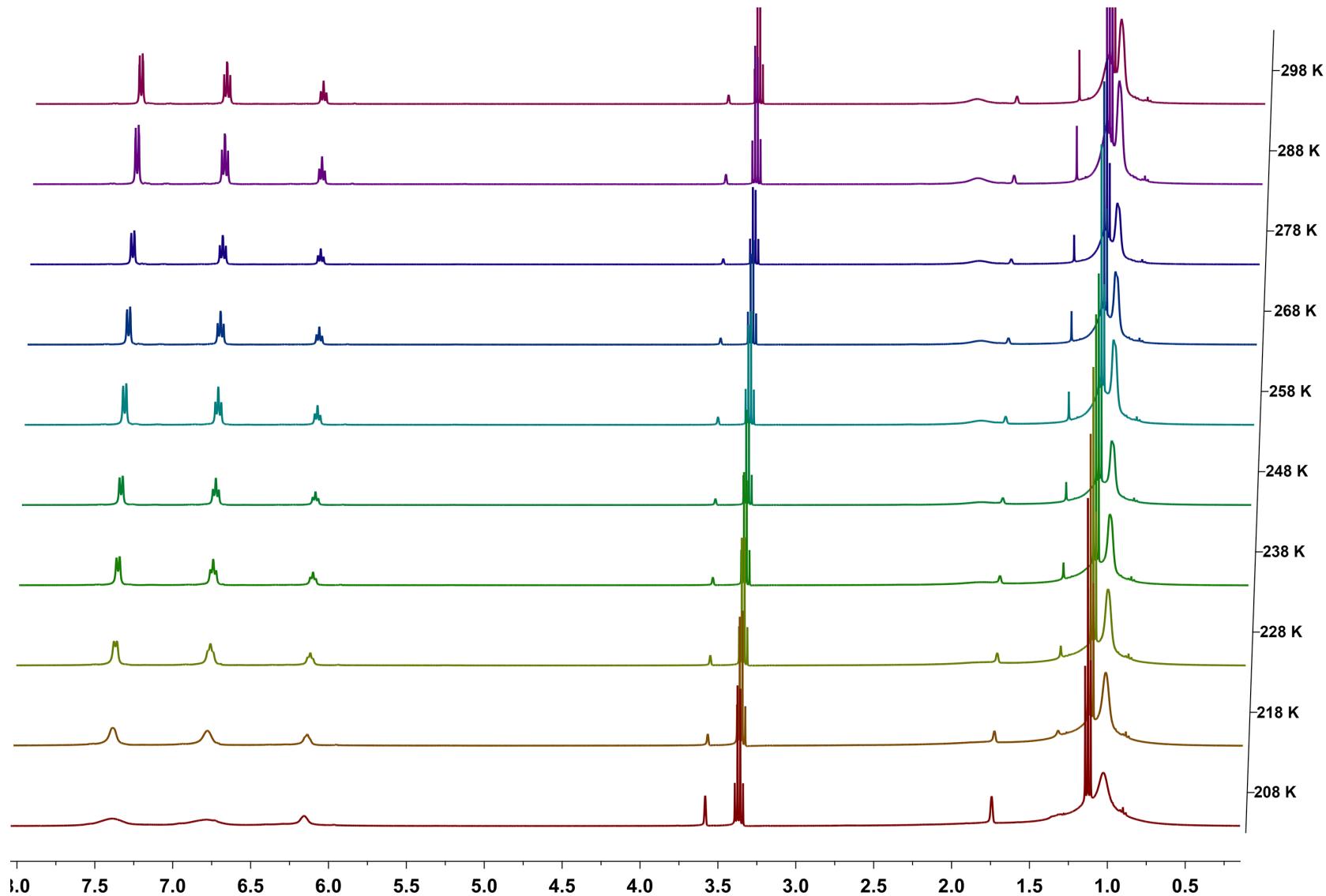


Figure S35. ¹H NMR spectra for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNPh}_2\right]^-[\text{KOEt}_2]^+$ (**5**) at different temperatures.

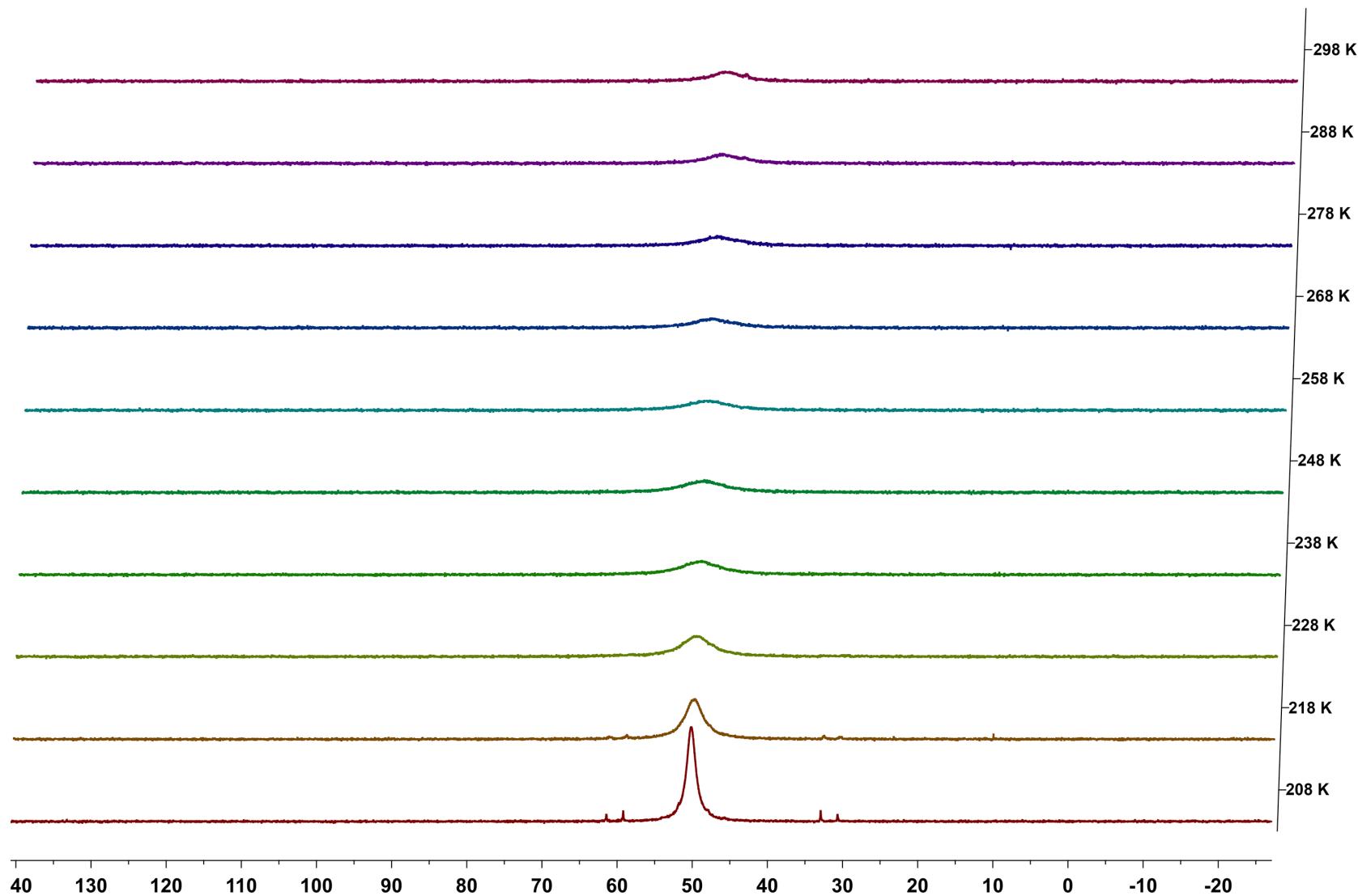


Figure S36. $^{31}\text{P}\{\text{H}\}$ NMR spectra for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNPh}_2\right]^-[\text{KOEt}_2]^+$ (**5**) at different temperatures.

4.3 NMR Spectra for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}\right]^- \text{K}^+$ (6)

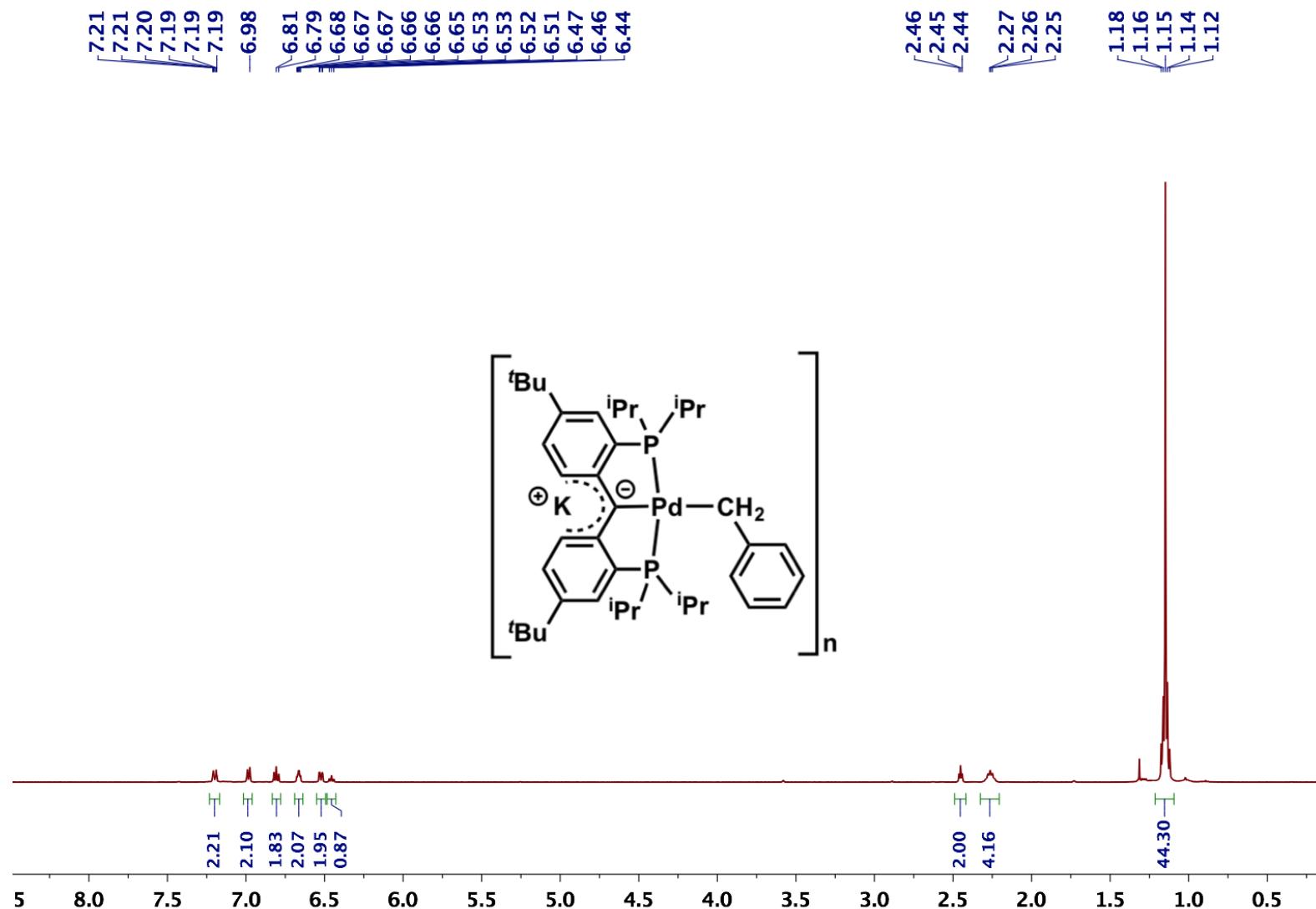


Figure S37. ^1H NMR spectrum for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}\right]^- \text{K}^+$ (6).

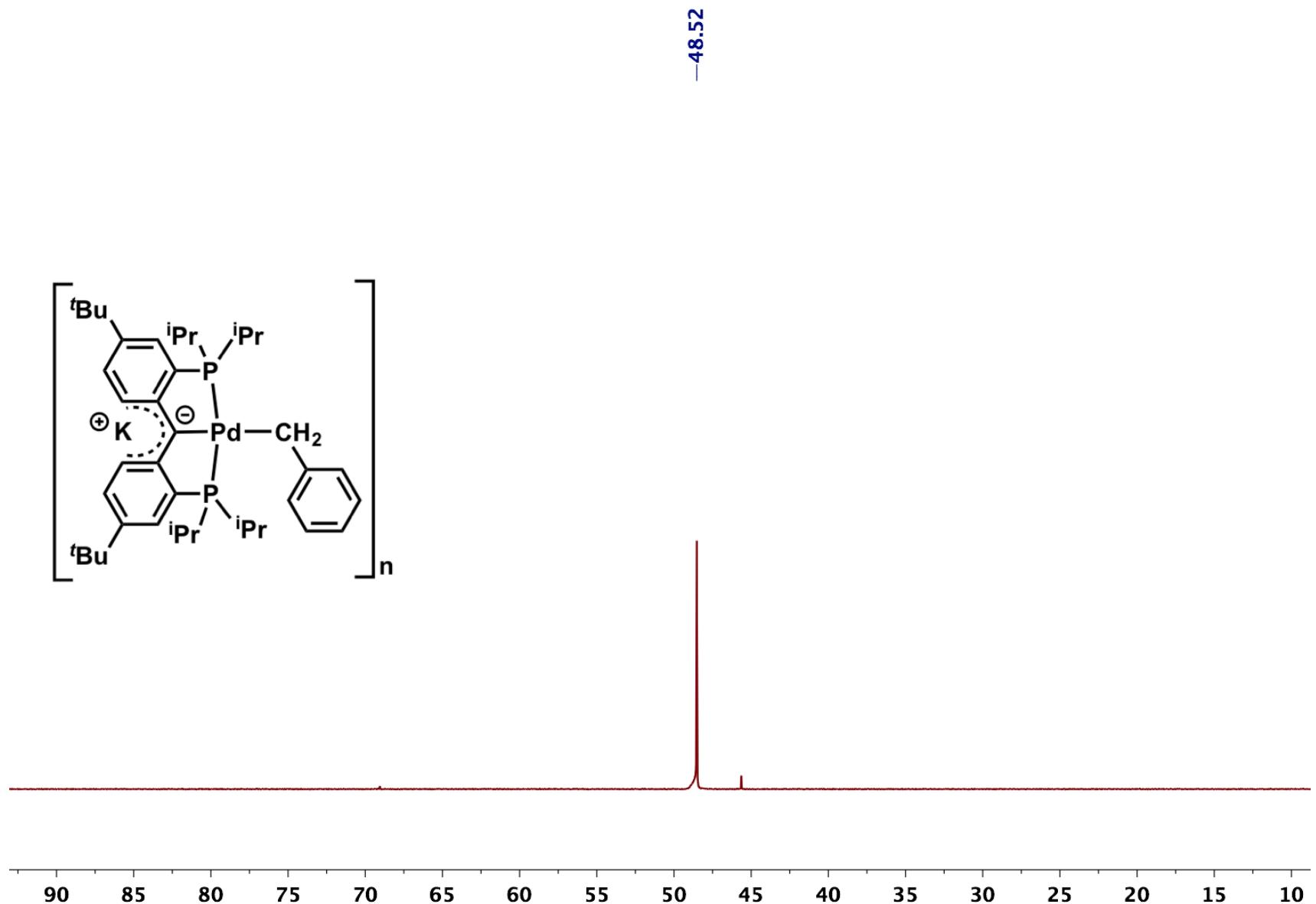


Figure S38. $^{31}\text{P}\{\text{H}\}$ NMR spectrum for $[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdCH}_2\text{Ph}]^-\text{K}^+$ (**6**).

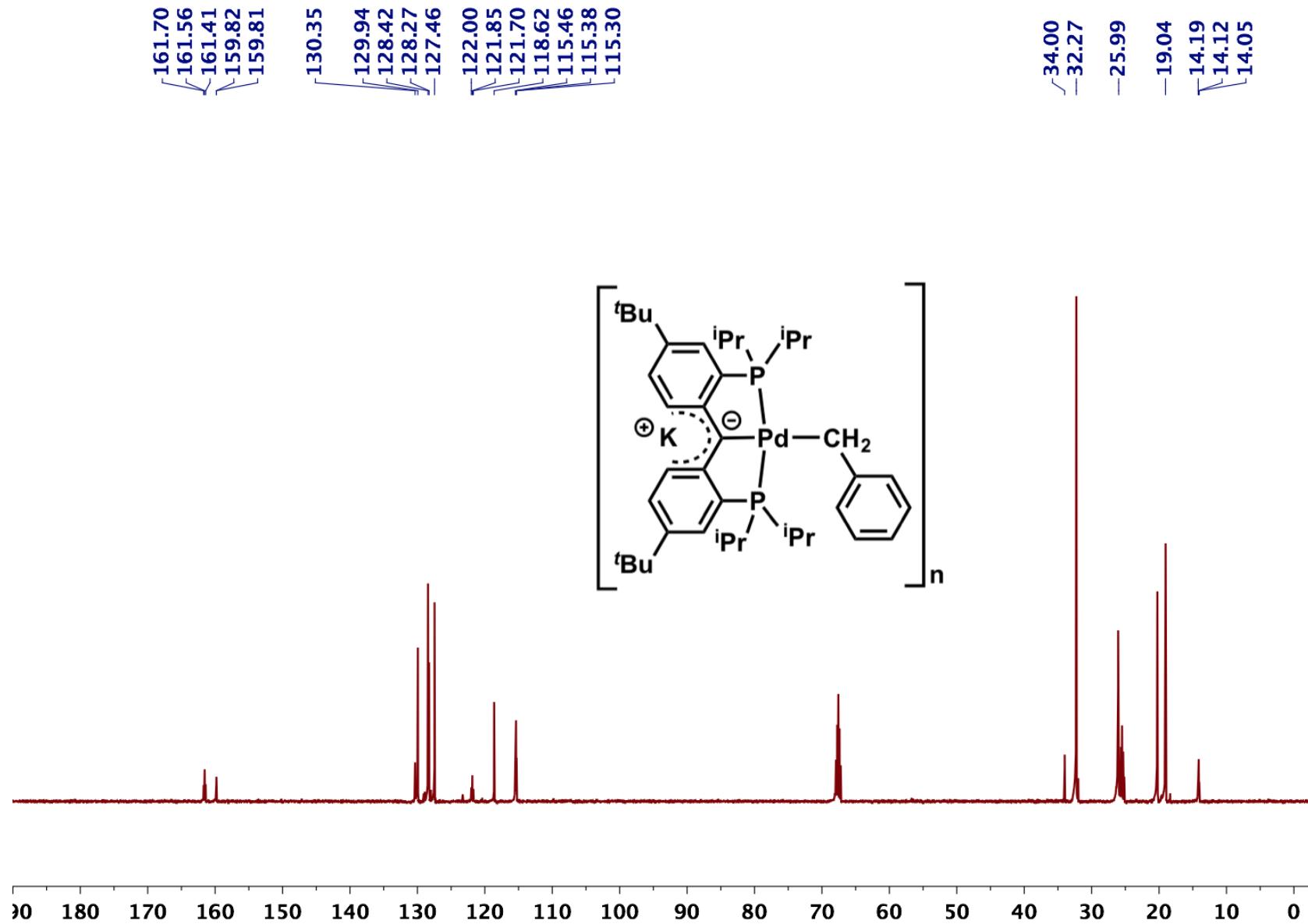


Figure S39. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for $\{[\text{PC}(\text{sp}^2)\text{P}]^t\text{Bu}\text{PdCH}_2\text{Ph}\}^-\text{K}^+$ (**6**).

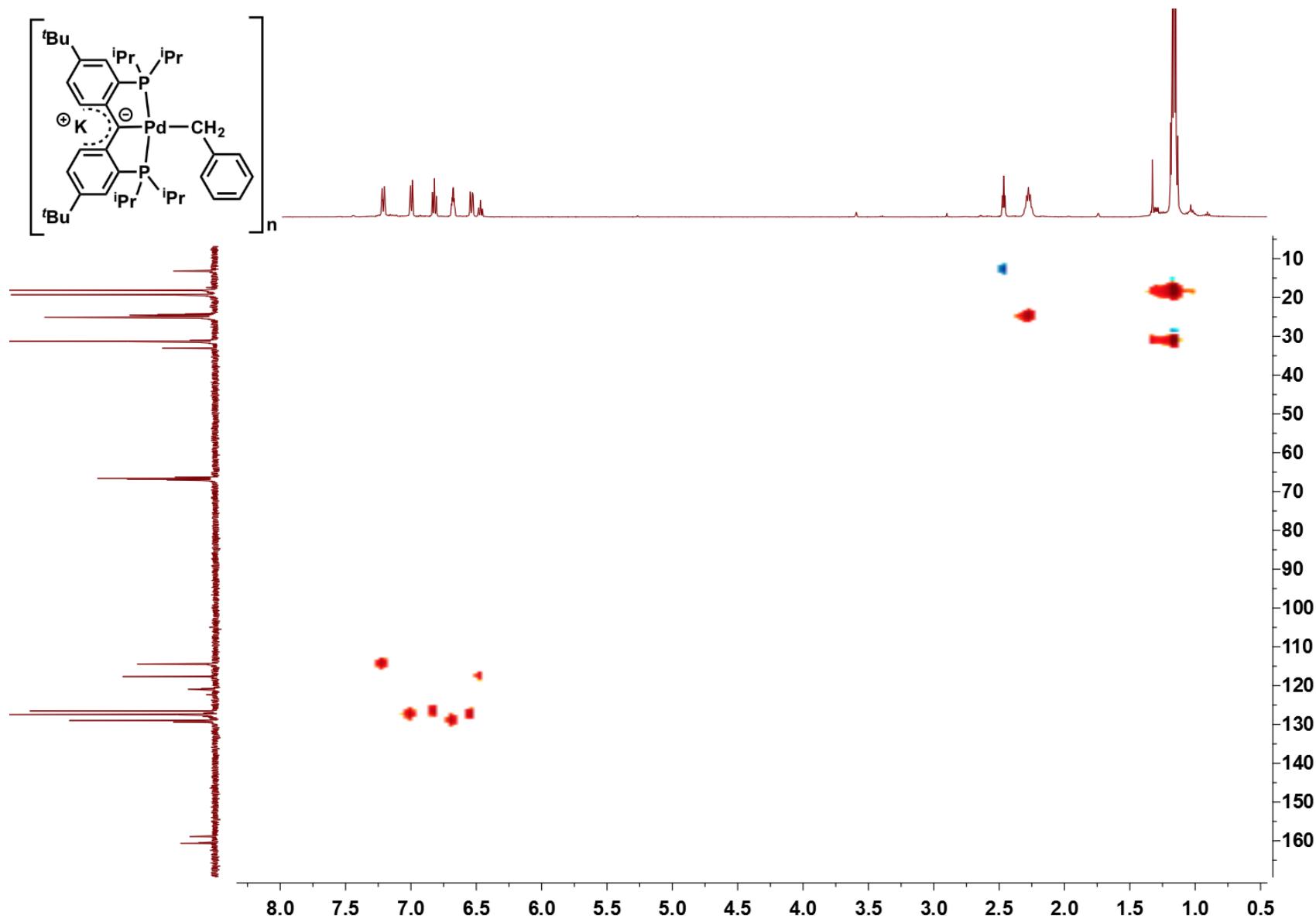


Figure S40. ^1H - ^{13}C HSQC NMR spectrum for $\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdCH}_2\text{Ph}^-\text{K}^+$ (**6**).

4.4 NMR Spectra for $\{\text{PC}(\text{sp}^3)\text{HP}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}$ (8)

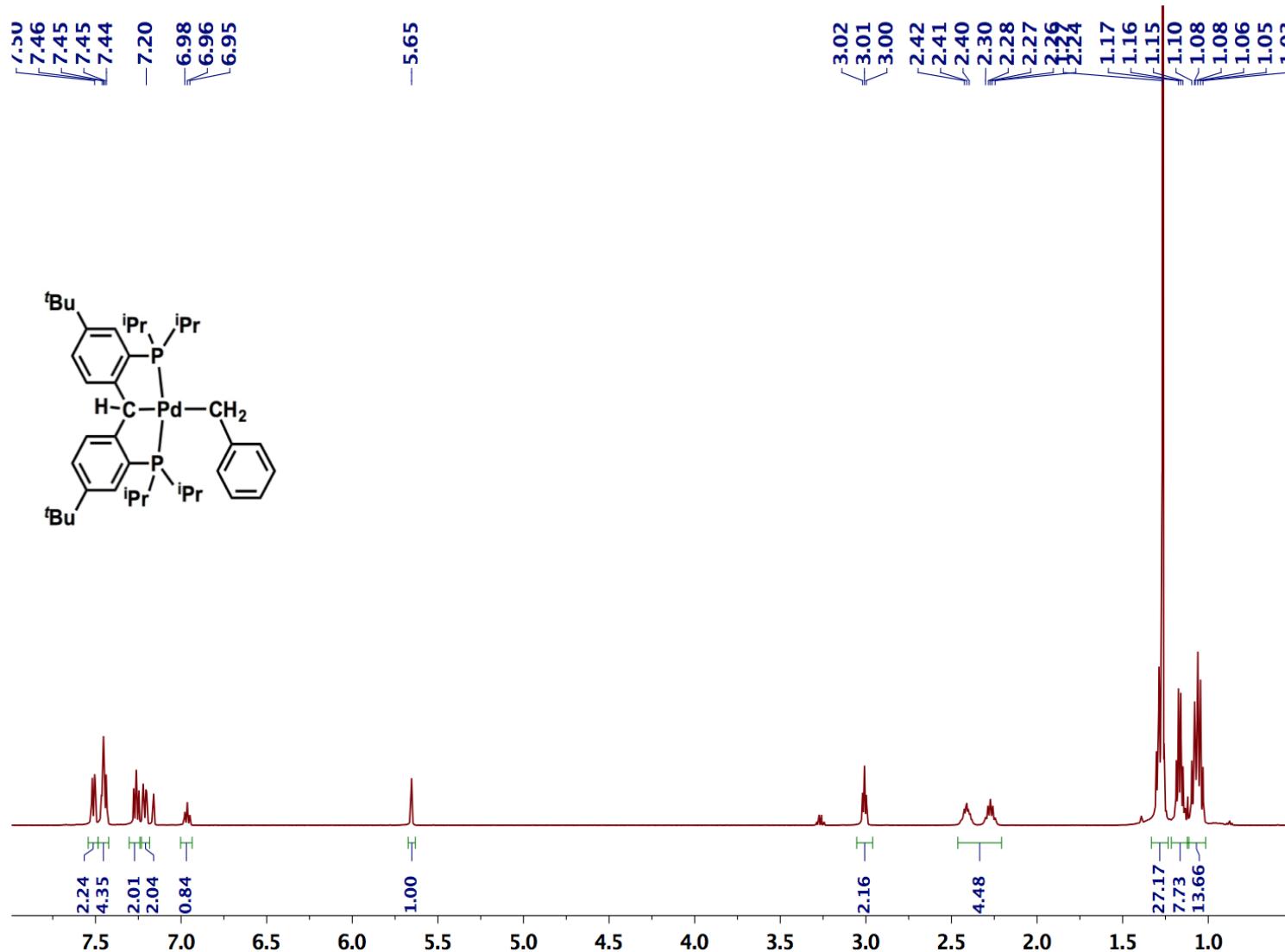


Figure S41. ^1H NMR spectrum for $\{\text{PC}(\text{sp}^3)\text{HP}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}$ (8).

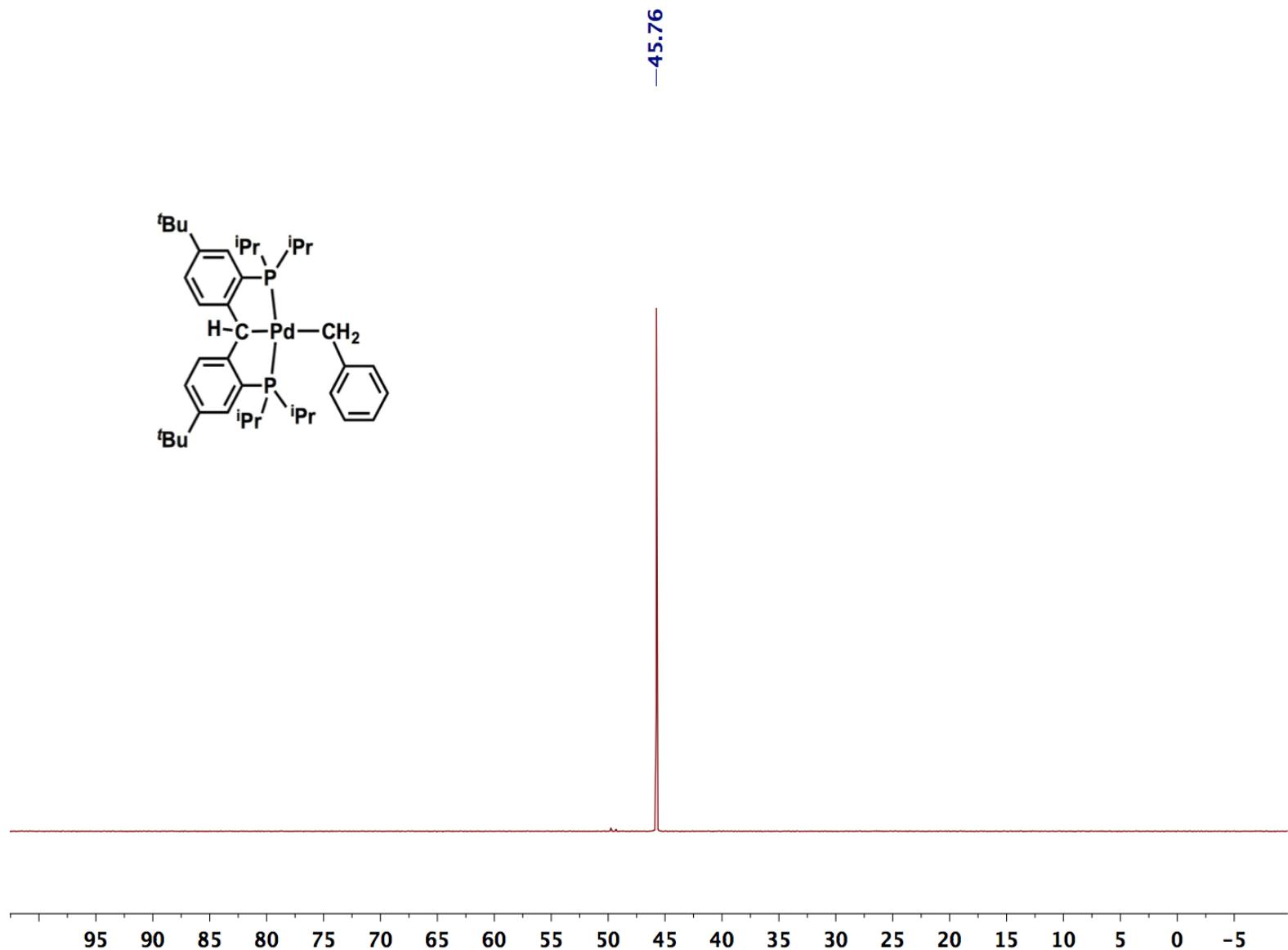


Figure S42. $^{31}\text{P}\{\text{H}\}$ NMR spectrum for $[\{\text{PC}(\text{sp}^3)\text{HP}\}^t\text{Bu}\text{PdCH}_2\text{Ph}]$ (**8**).

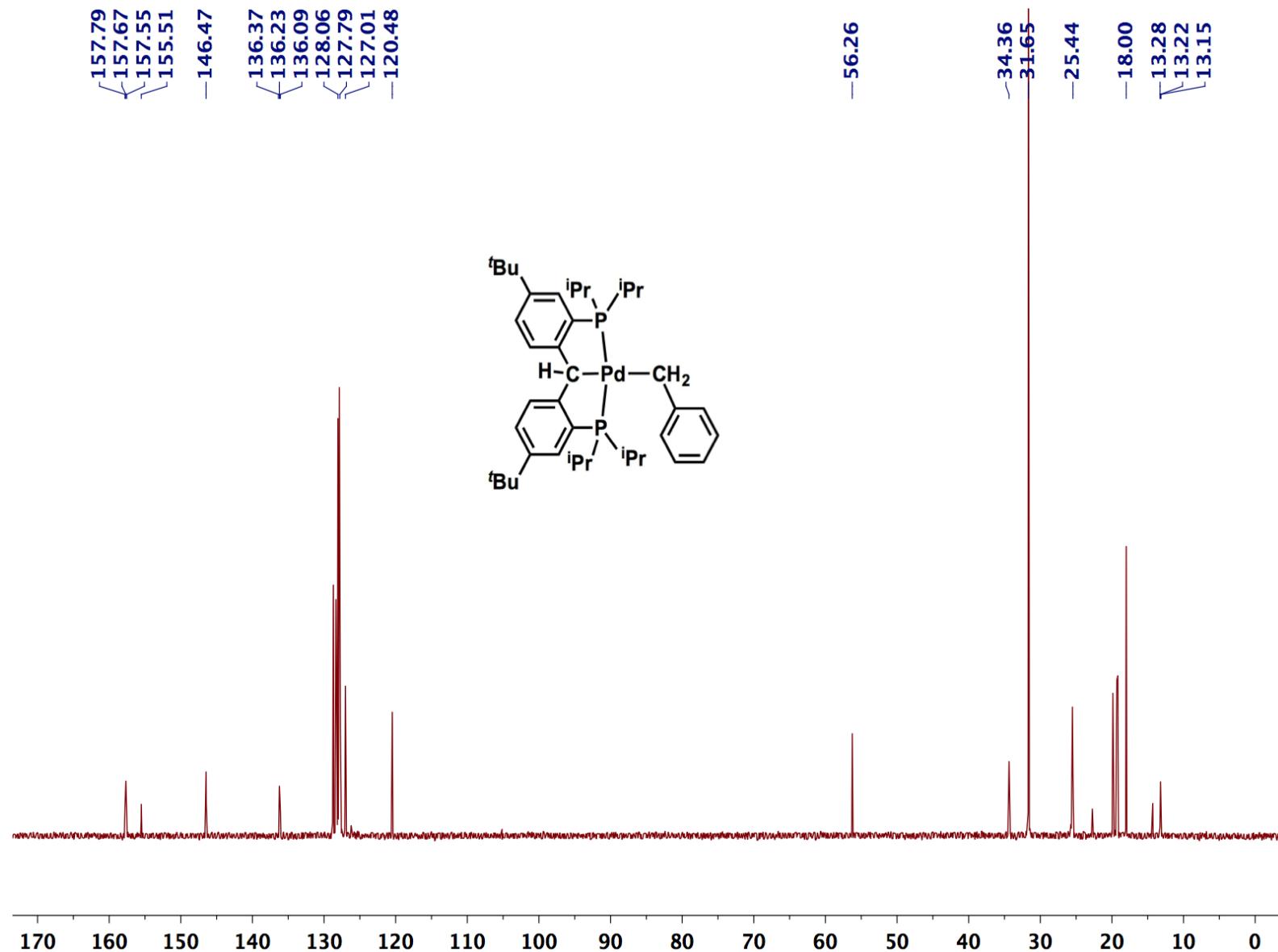


Figure S43. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for $\left[\{\text{PC}(\text{sp}^3)\text{HP}\}^{\text{tBu}}\text{PdCH}_2\text{Ph}\right]$ (**8**).

5 Crystallographic tables

5.1 Crystal data for $\left[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNH}^p\text{Tol}\right]$ (2)

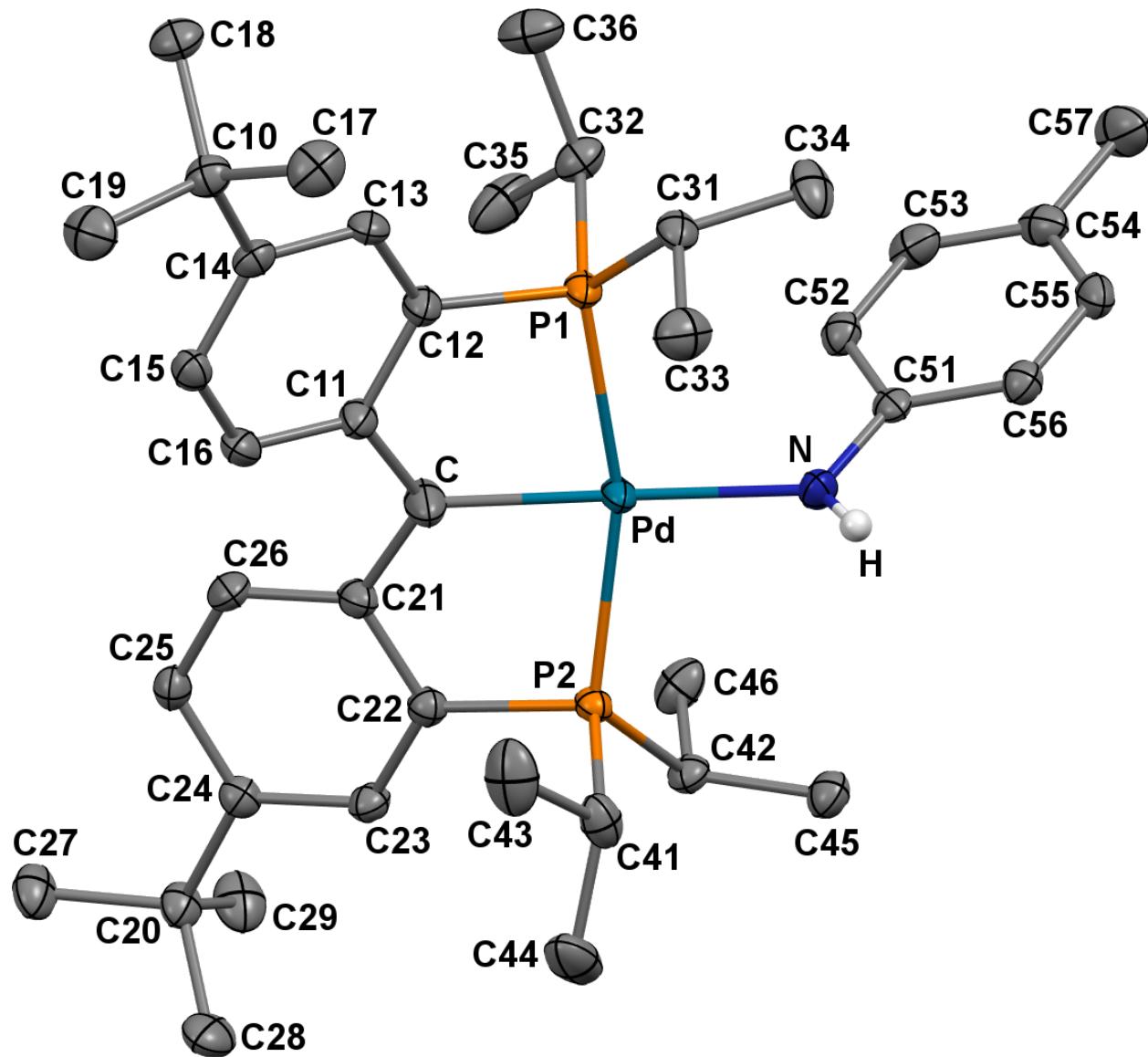


Figure S44. Thermal-ellipsoid representation of $\left[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNH}^p\text{Tol}\right]$ (2) at 50% probability. Most hydrogen atoms were omitted for clarity.

Table S21. Crystal data and structure refinement for $\{[\text{PC}^{\bullet}(\text{sp}^2)\text{P}]^{t\text{Bu}}\text{PdNH}^p\text{Tol}\}$ (**2**).

Identification code:	pc43	
Empirical formula:	$\text{C}_{40}\text{H}_{60}\text{NP}_2\text{Pd}$	
Formula weight:	723.23	
Temperature:	120(2) K	
Wavelength:	0.71073 Å	
Crystal system:	Triclinic	
Space group:	$P\bar{1}$	
Unit cell dimensions:	$a = 9.9475(7)$ Å $b = 13.7704(10)$ Å $c = 14.6585(10)$ Å	$\alpha = 98.9950(19)^\circ$ $\beta = 100.8508(18)^\circ$ $\gamma = 97.3666(18)^\circ$
Volume:	1921.9(2) Å ³	
Z:	2	
Density (calculated):	1.250 g·cm ⁻³	
Absorption coefficient (μ):	0.593 mm ⁻¹	
F(000):	766	
Crystal size:	0.09 × 0.09 × 0.06 mm ³	
θ range for data collection:	1.89 to 25.00°	
Index ranges:	$-11 \leq h \leq 11, -16 \leq k \leq 16, -17 \leq l \leq 17$	
Reflections collected:	47002	
Independent reflections:	6761 [$R_{\text{int}} = 0.0350$]	
Completeness to $\theta = 25.00^\circ$:	100.0 %	
Absorption correction:	Semi-empirical from equivalents	
Max. and min. transmission:	0.7457 and 0.6875	
Refinement method:	Full-matrix least-squares on F^2	
Data / restraints / parameters:	6761 / 0 / 416	
Goodness-of-fit on F^2 :	1.039	
Final R indices [I>2σ(I)]:	$R_1 = 0.0253, wR_2 = 0.0606$	
R indices (all data):	$R_1 = 0.0302, wR_2 = 0.0624$	
Largest diff. peak and hole:	0.818 and -0.527 e ⁻ ·Å ⁻³	

Table S22. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for $[\text{PC}^*(\text{sp}^2)\text{P}]^{t\text{Bu}}\text{PdNH}^p\text{Tol}$ (**2**). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor

atom	x	y	z	U(eq)
Pd	0.80704(2)	0.83723(1)	0.65165(1)	0.018(1)
P(1)	0.94622(5)	0.73444(4)	0.59103(4)	0.019(1)
P(2)	0.72522(5)	0.95588(4)	0.74369(4)	0.016(1)
C(11)	1.1011(2)	0.84841(15)	0.75360(15)	0.022(1)
C(10)	1.3847(2)	0.63478(15)	0.80883(15)	0.022(1)
C(12)	1.0976(2)	0.76051(15)	0.68804(15)	0.021(1)
C(13)	1.1935(2)	0.69736(15)	0.70509(14)	0.020(1)
C(14)	1.2931(2)	0.71367(15)	0.78937(14)	0.019(1)
C(15)	1.2961(2)	0.80055(15)	0.85418(14)	0.020(1)
C(16)	1.2045(2)	0.86679(15)	0.83653(15)	0.022(1)
C(17)	1.2916(2)	0.53418(17)	0.79930(18)	0.032(1)
C(18)	1.4784(2)	0.62452(17)	0.73701(16)	0.027(1)
C(19)	1.4750(2)	0.65969(18)	0.90832(16)	0.032(1)
C(20)	1.0087(2)	1.31771(15)	0.91878(15)	0.023(1)
C(21)	1.0034(2)	1.00972(15)	0.78150(14)	0.021(1)
C(22)	0.8807(2)	1.04791(15)	0.79158(14)	0.018(1)
C(23)	0.8849(2)	1.14537(15)	0.83526(14)	0.020(1)
C(25)	1.1306(2)	1.17412(15)	0.85831(14)	0.021(1)
C(24)	1.0097(2)	1.21086(14)	0.87106(14)	0.019(1)
C	0.9900(2)	0.90616(17)	0.73555(17)	0.033(1)
C(26)	1.1282(2)	1.07709(16)	0.81488(15)	0.023(1)
C(27)	1.1552(2)	1.37241(16)	0.96651(17)	0.031(1)
C(28)	0.9231(2)	1.31690(17)	0.99557(17)	0.033(1)
C(29)	0.9448(3)	1.37390(16)	0.84366(18)	0.034(1)
C(33)	0.8449(3)	0.57378(18)	0.66251(18)	0.034(1)
C(32)	1.0043(2)	0.75898(16)	0.48335(16)	0.025(1)
C(31)	0.8932(2)	0.59885(15)	0.57522(15)	0.021(1)
C(35)	1.0509(3)	0.87130(18)	0.4935(2)	0.041(1)
C(34)	0.7813(2)	0.55835(16)	0.48587(17)	0.032(1)
C(36)	1.1169(3)	0.70061(19)	0.45876(18)	0.035(1)
C(42)	0.5909(2)	1.02208(15)	0.69125(15)	0.020(1)
C(43)	0.7850(3)	0.8700(2)	0.90187(18)	0.042(1)
C(44)	0.6275(3)	0.99892(19)	0.91265(17)	0.039(1)
C(45)	0.4494(2)	0.95419(16)	0.66134(16)	0.025(1)
C(46)	0.6345(2)	1.06190(17)	0.60822(16)	0.029(1)
C(51)	0.58263(19)	0.74656(14)	0.47172(14)	0.017(1)
C(52)	0.6592(2)	0.79960(15)	0.41853(15)	0.021(1)
C(53)	0.6262(2)	0.78118(17)	0.32081(15)	0.026(1)
C(54)	0.5144(2)	0.71004(17)	0.26939(15)	0.026(1)

Continued on next page

Table S22. – continued from previous page

atom	x	y	x	U(eq)
C(55)	0.4375(2)	0.65787(16)	0.32107(15)	0.025(1)
C(56)	0.4698(2)	0.67466(15)	0.41869(15)	0.021(1)
C(57)	0.4784(3)	0.6920(2)	0.16280(16)	0.038(1)
N	0.61756(18)	0.76538(14)	0.56737(12)	0.020(1)
C(41)	0.6706(2)	0.91642(16)	0.84770(15)	0.026(1)
H(13)	1.1915	0.6411	0.6579	0.024
H(15)	1.3628	0.8146	0.9121	0.024
H(16)	1.2121	0.9259	0.8816	0.026
H(17A)	1.2355	0.5146	0.7348	0.047
H(17B)	1.2303	0.5403	0.8443	0.047
H(17C)	1.3495	0.4835	0.8127	0.047
H	0.567(2)	0.7349(17)	0.5891(16)	0.018(6)
H(18A)	1.4210	0.6084	0.6728	0.041
H(18B)	1.5326	0.5712	0.7477	0.041
H(18C)	1.5414	0.6875	0.7446	0.041
H(19A)	1.5321	0.6076	0.9176	0.049
H(19B)	1.4157	0.6635	0.9547	0.049
H(19C)	1.5353	0.7239	0.9166	0.049
H(23)	0.8003	1.1683	0.8410	0.024
H(25)	1.2173	1.2171	0.8802	0.025
H(26)	1.2130	1.0554	0.8074	0.028
H(27A)	1.2116	1.3765	0.9188	0.046
H(27B)	1.1976	1.3359	1.0136	0.046
H(27C)	1.1500	1.4398	0.9976	0.046
H(28A)	0.8267	1.2869	0.9665	0.050
H(28B)	0.9259	1.3854	1.0276	0.050
H(28C)	0.9618	1.2779	1.0416	0.050
H(29A)	1.0012	1.3753	0.7957	0.050
H(29B)	0.9419	1.4422	0.8735	0.050
H(29C)	0.8503	1.3401	0.8138	0.050
H(33A)	0.7621	0.6035	0.6691	0.051
H(33B)	0.9188	0.6004	0.7186	0.051
H(33C)	0.8228	0.5013	0.6560	0.051
H(32)	0.9221	0.7388	0.4295	0.031
H(31)	0.9763	0.5665	0.5691	0.026
H(35A)	1.0668	0.8863	0.4329	0.061
H(35B)	1.1370	0.8923	0.5416	0.061
H(35C)	0.9786	0.9071	0.5126	0.061
H(34A)	0.8154	0.5753	0.4309	0.047
H(34B)	0.6980	0.5880	0.4909	0.047
H(34C)	0.7588	0.4858	0.4785	0.047
H(36A)	1.0838	0.6291	0.4517	0.053

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Table S22. – continued from previous page

atom	x	y	x	U(eq)
H(36B)	1.2003	0.7203	0.5094	0.053
H(36C)	1.1391	0.7149	0.3994	0.053
H(42)	0.5840	1.0799	0.7398	0.024
H(43A)	0.8081	0.8156	0.8590	0.063
H(43B)	0.7527	0.8439	0.9538	0.063
H(43C)	0.8676	0.9209	0.9275	0.063
H(44A)	0.5496	1.0238	0.8772	0.058
H(44B)	0.7059	1.0535	0.9368	0.058
H(44C)	0.5993	0.9725	0.9657	0.058
H(45A)	0.3794	0.9912	0.6336	0.037
H(45B)	0.4236	0.9315	0.7168	0.037
H(45C)	0.4547	0.8964	0.6145	0.037
H(46A)	0.6435	1.0061	0.5605	0.043
H(46B)	0.7239	1.1064	0.6303	0.043
H(46C)	0.5643	1.0988	0.5804	0.043
H(52)	0.7357	0.8495	0.4507	0.025
H(53)	0.6815	0.8182	0.2878	0.031
H(55)	0.3602	0.6090	0.2883	0.030
H(56)	0.4144	0.6367	0.4510	0.025
H(57A)	0.5596	0.6763	0.1381	0.057
H(57B)	0.4505	0.7519	0.1417	0.057
H(57C)	0.4018	0.6360	0.1393	0.057
H(41)	0.5878	0.8629	0.8236	0.031

Table S23. Anisotropic displacement parameters (\AA^2) for $[\{\text{PC}^\bullet(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNH}^p\text{Tol}]$ (2). The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}\text{U}_{11} + \dots + 2hka^{*}b^{*}\text{U}_{12}]$.

atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
Pd	0.0154(1)	0.0164(1)	0.0200(1)	-0.0039(1)	0.0002(1)	0.0049(1)
P(1)	0.0168(3)	0.0149(3)	0.0223(3)	-0.0028(2)	0.0015(2)	0.0047(2)
P(2)	0.0177(3)	0.0139(2)	0.0164(3)	0.0002(2)	0.0025(2)	0.0039(2)
C(11)	0.0170(10)	0.0171(10)	0.0277(11)	-0.0027(9)	0.0032(9)	0.0022(8)
C(10)	0.0221(11)	0.0226(11)	0.0229(11)	0.0066(9)	0.0054(9)	0.0068(9)
C(12)	0.0175(10)	0.0194(10)	0.0238(11)	-0.0019(9)	0.0015(8)	0.0042(8)
C(13)	0.0196(10)	0.0182(10)	0.0210(11)	-0.0030(8)	0.0051(8)	0.0049(8)
C(14)	0.0170(10)	0.0200(10)	0.0212(11)	0.0059(8)	0.0073(8)	0.0038(8)
C(15)	0.0195(10)	0.0235(11)	0.0170(10)	0.0030(8)	0.0027(8)	0.0020(8)
C(16)	0.0214(11)	0.0193(10)	0.0226(11)	-0.0036(9)	0.0063(9)	0.0015(8)
C(17)	0.0335(13)	0.0264(12)	0.0391(14)	0.0141(10)	0.0103(11)	0.0072(10)
C(18)	0.0261(11)	0.0272(12)	0.0322(12)	0.0068(10)	0.0099(10)	0.0117(9)
C(19)	0.0337(13)	0.0362(13)	0.0283(12)	0.0085(10)	0.0013(10)	0.0140(11)
C(20)	0.0212(11)	0.0146(10)	0.0310(12)	0.0017(9)	0.0035(9)	0.0018(8)
C(21)	0.0198(10)	0.0225(11)	0.0191(10)	-0.0016(8)	0.0010(8)	0.0072(8)
C(22)	0.0181(10)	0.0176(10)	0.0181(10)	0.0027(8)	0.0034(8)	0.0033(8)
C(23)	0.0174(10)	0.0182(10)	0.0244(11)	0.0027(8)	0.0061(8)	0.0048(8)
C(25)	0.0169(10)	0.0216(11)	0.0218(11)	0.0031(9)	0.0021(8)	0.0000(8)
C(24)	0.0212(10)	0.0157(10)	0.0198(10)	0.0043(8)	0.0044(8)	0.0031(8)
C	0.0216(11)	0.0292(12)	0.0391(14)	-0.0116(10)	-0.0033(10)	0.0092(10)
C(26)	0.0167(10)	0.0271(12)	0.0260(11)	0.0004(9)	0.0044(9)	0.0075(9)
C(27)	0.0266(12)	0.0173(11)	0.0420(14)	-0.0008(10)	0.0006(10)	0.0010(9)
C(28)	0.0354(13)	0.0216(12)	0.0397(14)	-0.0075(10)	0.0119(11)	0.0039(10)
C(29)	0.0357(13)	0.0173(11)	0.0440(15)	0.0043(10)	-0.0015(11)	0.0071(10)
C(33)	0.0371(13)	0.0284(13)	0.0395(14)	0.0067(11)	0.0145(11)	0.0045(10)
C(32)	0.0225(11)	0.0271(12)	0.0300(12)	0.0086(10)	0.0080(9)	0.0082(9)
C(31)	0.0206(10)	0.0155(10)	0.0255(11)	-0.0018(8)	0.0024(9)	0.0049(8)
C(35)	0.0291(13)	0.0318(14)	0.0707(19)	0.0199(13)	0.0220(13)	0.0081(11)
C(34)	0.0320(13)	0.0169(11)	0.0374(14)	-0.0034(10)	-0.0067(10)	0.0037(9)
C(36)	0.0381(14)	0.0413(14)	0.0345(14)	0.0115(11)	0.0168(11)	0.0195(11)
C(42)	0.0170(10)	0.0194(10)	0.0239(11)	0.0030(8)	0.0045(8)	0.0065(8)
C(43)	0.0436(15)	0.0446(15)	0.0337(14)	0.0215(12)	-0.0058(12)	-0.0021(12)
C(44)	0.0459(15)	0.0431(15)	0.0240(12)	-0.0024(11)	0.0160(11)	-0.0067(12)
C(45)	0.0181(10)	0.0270(12)	0.0295(12)	0.0056(9)	0.0034(9)	0.0058(9)
C(46)	0.0226(11)	0.0343(13)	0.0345(13)	0.0177(10)	0.0060(10)	0.0093(10)
C(51)	0.0154(10)	0.0181(10)	0.0185(10)	0.0019(8)	0.0021(8)	0.0084(8)
C(52)	0.0160(10)	0.0222(11)	0.0246(11)	0.0045(9)	0.0019(8)	0.0048(8)
C(53)	0.0231(11)	0.0344(13)	0.0250(12)	0.0125(10)	0.0092(9)	0.0110(9)
C(54)	0.0251(11)	0.0366(13)	0.0194(11)	0.0041(9)	0.0029(9)	0.0158(10)
C(55)	0.0188(10)	0.0287(12)	0.0237(11)	-0.0021(9)	-0.0022(9)	0.0065(9)

Continued on next page

Table S23. – continued from previous page

atom	U₁₁	U₂₂	U₃₃	U₂₃	U₁₃	U₁₂
C(56)	0.0179(10)	0.0215(11)	0.0246(11)	0.0048(9)	0.0040(9)	0.0042(8)
C(57)	0.0341(13)	0.0604(17)	0.0210(12)	0.0068(12)	0.0027(10)	0.0164(12)
N	0.0172(9)	0.0232(9)	0.0184(9)	0.0029(8)	0.0048(8)	-0.0001(7)
C(41)	0.0293(12)	0.0243(11)	0.0207(11)	0.0056(9)	0.0020(9)	-0.0042(9)

Table S24. Distances [Å] for $\left[\{\text{PC}^*(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNH}^{\text{pTol}}\right]$ (**2**).

atom – atom	distance	atom – atom	distance
Pd–C	2.019(2)	Pd–N	2.0787(18)
Pd–P(2)	2.2841(5)	Pd–P(1)	2.2983(5)
P(1)–C(12)	1.821(2)	P(1)–C(31)	1.840(2)
P(1)–C(32)	1.844(2)	P(2)–C(22)	1.817(2)
P(2)–C(42)	1.831(2)	P(2)–C(41)	1.847(2)
C(11)–C(16)	1.402(3)	C(11)–C(12)	1.415(3)
C(11)–C	1.451(3)	C(10)–C(19)	1.524(3)
C(10)–C(14)	1.532(3)	C(10)–C(18)	1.534(3)
C(10)–C(17)	1.536(3)	C(12)–C(13)	1.386(3)
C(13)–C(14)	1.398(3)	C(13)–H(13)	0.9500
C(14)–C(15)	1.400(3)	C(15)–C(16)	1.388(3)
C(15)–H(15)	0.9500	C(16)–H(16)	0.9500
C(17)–H(17A)	0.9800	C(17)–H(17B)	0.9800
C(17)–H(17C)	0.9800	C(18)–H(18A)	0.9800
C(18)–H(18B)	0.9800	C(18)–H(18C)	0.9800
C(19)–H(19A)	0.9800	C(19)–H(19B)	0.9800
C(19)–H(19C)	0.9800	C(20)–C(24)	1.529(3)
C(20)–C(29)	1.531(3)	C(20)–C(27)	1.533(3)
C(20)–C(28)	1.535(3)	C(21)–C(26)	1.405(3)
C(21)–C(22)	1.413(3)	C(21)–C	1.457(3)
C(22)–C(23)	1.385(3)	C(23)–C(24)	1.396(3)
C(23)–H(23)	0.9500	C(25)–C(26)	1.382(3)
C(25)–C(24)	1.396(3)	C(25)–H(25)	0.9500
C(26)–H(26)	0.9500	C(27)–H(27A)	0.9800
C(27)–H(27B)	0.9800	C(27)–H(27C)	0.9800
C(28)–H(28A)	0.9800	C(28)–H(28B)	0.9800
C(28)–H(28C)	0.9800	C(29)–H(29A)	0.9800
C(29)–H(29B)	0.9800	C(29)–H(29C)	0.9800
C(33)–C(31)	1.521(3)	C(33)–H(33A)	0.9800
C(33)–H(33B)	0.9800	C(33)–H(33C)	0.9800
C(32)–C(36)	1.524(3)	C(32)–C(35)	1.532(3)
C(32)–H(32)	1.0000	C(31)–C(34)	1.526(3)
C(31)–H(31)	1.0000	C(35)–H(35A)	0.9800
C(35)–H(35B)	0.9800	C(35)–H(35C)	0.9800
C(34)–H(34A)	0.9800	C(34)–H(34B)	0.9800
C(34)–H(34C)	0.9800	C(36)–H(36A)	0.9800
C(36)–H(36B)	0.9800	C(36)–H(36C)	0.9800
C(42)–C(46)	1.522(3)	C(42)–C(45)	1.533(3)
C(42)–H(42)	1.0000	C(43)–C(41)	1.529(3)
C(43)–H(43A)	0.9800	C(43)–H(43B)	0.9800
C(43)–H(43C)	0.9800	C(44)–C(41)	1.524(3)

Continued on next page

Table S24. – continued from previous page

atom – atom	distance	atom – atom	distance
C(44)–H(44A)	0.9800	C(44)–H(44B)	0.9800
C(44)–H(44C)	0.9800	C(45)–H(45A)	0.9800
C(45)–H(45B)	0.9800	C(45)–H(45C)	0.9800
C(46)–H(46A)	0.9800	C(46)–H(46B)	0.9800
C(46)–H(46C)	0.9800	C(51)–N	1.354(3)
C(51)–C(52)	1.407(3)	C(51)–C(56)	1.411(3)
C(52)–C(53)	1.383(3)	C(52)–H(52)	0.9500
C(53)–C(54)	1.392(3)	C(53)–H(53)	0.9500
C(54)–C(55)	1.386(3)	C(54)–C(57)	1.509(3)
C(55)–C(56)	1.382(3)	C(55)–H(55)	0.9500
C(56)–H(56)	0.9500	C(57)–H(57A)	0.9800
C(57)–H(57B)	0.9800	C(57)–H(57C)	0.9800
N–H	0.76(2)	C(41)–H(41)	1.0000

Table S25. Angles [°] for $[\{PC\bullet(sp^2)P\}^{tBu}PdNH^pTol]$ (**2**).

atom – atom – atom	angle	atom – atom – atom	angle
C–Pd–N	178.97(9)	C–Pd–P(2)	81.80(6)
N–Pd–P(2)	97.79(5)	C–Pd–P(1)	82.15(6)
N–Pd–P(1)	98.20(5)	P(2)–Pd–P(1)	163.689(19)
C(12)–P(1)–C(31)	103.37(9)	C(12)–P(1)–C(32)	108.03(10)
C(31)–P(1)–C(32)	106.95(10)	C(12)–P(1)–Pd	100.58(7)
C(31)–P(1)–Pd	118.47(7)	C(32)–P(1)–Pd	117.77(7)
C(22)–P(2)–C(42)	106.81(9)	C(22)–P(2)–C(41)	105.16(10)
C(42)–P(2)–C(41)	106.28(10)	C(22)–P(2)–Pd	101.61(7)
C(42)–P(2)–Pd	120.79(7)	C(41)–P(2)–Pd	114.74(7)
C(16)–C(11)–C(12)	116.58(18)	C(16)–C(11)–C	124.86(19)
C(12)–C(11)–C	118.18(18)	C(19)–C(10)–C(14)	112.17(17)
C(19)–C(10)–C(18)	108.99(18)	C(14)–C(10)–C(18)	110.09(17)
C(19)–C(10)–C(17)	107.65(18)	C(14)–C(10)–C(17)	108.99(17)
C(18)–C(10)–C(17)	108.87(18)	C(13)–C(12)–C(11)	121.13(19)
C(13)–C(12)–P(1)	125.28(15)	C(11)–C(12)–P(1)	113.18(15)
C(12)–C(13)–C(14)	122.21(18)	C(12)–C(13)–H(13)	118.9
C(14)–C(13)–H(13)	118.9	C(13)–C(14)–C(15)	116.42(18)
C(13)–C(14)–C(10)	119.61(18)	C(15)–C(14)–C(10)	123.85(18)
C(16)–C(15)–C(14)	122.09(19)	C(16)–C(15)–H(15)	119.0
C(14)–C(15)–H(15)	119.0	C(15)–C(16)–C(11)	121.46(18)
C(15)–C(16)–H(16)	119.3	C(11)–C(16)–H(16)	119.3
C(10)–C(17)–H(17A)	109.5	C(10)–C(17)–H(17B)	109.5
H(17A)–C(17)–H(17B)	109.5	C(10)–C(17)–H(17C)	109.5
H(17A)–C(17)–H(17C)	109.5	H(17B)–C(17)–H(17C)	109.5
C(10)–C(18)–H(18A)	109.5	C(10)–C(18)–H(18B)	109.5
H(18A)–C(18)–H(18B)	109.5	C(10)–C(18)–H(18C)	109.5
H(18A)–C(18)–H(18C)	109.5	H(18B)–C(18)–H(18C)	109.5
C(10)–C(19)–H(19A)	109.5	C(10)–C(19)–H(19B)	109.5
H(19A)–C(19)–H(19B)	109.5	C(10)–C(19)–H(19C)	109.5
H(19A)–C(19)–H(19C)	109.5	H(19B)–C(19)–H(19C)	109.5
C(24)–C(20)–C(29)	108.57(17)	C(24)–C(20)–C(27)	111.92(17)
C(29)–C(20)–C(27)	109.06(18)	C(24)–C(20)–C(28)	109.84(17)
C(29)–C(20)–C(28)	109.99(19)	C(27)–C(20)–C(28)	107.45(18)
C(26)–C(21)–C(22)	116.22(18)	C(26)–C(21)–C	125.83(19)
C(22)–C(21)–C	117.93(19)	C(23)–C(22)–C(21)	121.35(18)
C(23)–C(22)–P(2)	125.91(15)	C(21)–C(22)–P(2)	112.74(15)
C(22)–C(23)–C(24)	122.05(18)	C(22)–C(23)–H(23)	119.0
C(24)–C(23)–H(23)	119.0	C(26)–C(25)–C(24)	122.06(19)
C(26)–C(25)–H(25)	119.0	C(24)–C(25)–H(25)	119.0
C(23)–C(24)–C(25)	116.60(18)	C(23)–C(24)–C(20)	119.99(17)
C(25)–C(24)–C(20)	123.39(18)	C(11)–C–C(21)	121.98(19)

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Table S25. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(11) – C – Pd	118.57(15)	C(21) – C – Pd	119.30(15)
C(25) – C(26) – C(21)	121.67(19)	C(25) – C(26) – H(26)	119.2
C(21) – C(26) – H(26)	119.2	C(20) – C(27) – H(27A)	109.5
C(20) – C(27) – H(27B)	109.5	H(27A) – C(27) – H(27B)	109.5
C(20) – C(27) – H(27C)	109.5	H(27A) – C(27) – H(27C)	109.5
H(27B) – C(27) – H(27C)	109.5	C(20) – C(28) – H(28A)	109.5
C(20) – C(28) – H(28B)	109.5	H(28A) – C(28) – H(28B)	109.5
C(20) – C(28) – H(28C)	109.5	H(28A) – C(28) – H(28C)	109.5
H(28B) – C(28) – H(28C)	109.5	C(20) – C(29) – H(29A)	109.5
C(20) – C(29) – H(29B)	109.5	H(29A) – C(29) – H(29B)	109.5
C(20) – C(29) – H(29C)	109.5	H(29A) – C(29) – H(29C)	109.5
H(29B) – C(29) – H(29C)	109.5	C(31) – C(33) – H(33A)	109.5
C(31) – C(33) – H(33B)	109.5	H(33A) – C(33) – H(33B)	109.5
C(31) – C(33) – H(33C)	109.5	H(33A) – C(33) – H(33C)	109.5
H(33B) – C(33) – H(33C)	109.5	C(36) – C(32) – C(35)	111.24(19)
C(36) – C(32) – P(1)	114.12(16)	C(35) – C(32) – P(1)	108.91(16)
C(36) – C(32) – H(32)	107.4	C(35) – C(32) – H(32)	107.4
P(1) – C(32) – H(32)	107.4	C(33) – C(31) – C(34)	111.27(19)
C(33) – C(31) – P(1)	109.42(14)	C(34) – C(31) – P(1)	111.89(15)
C(33) – C(31) – H(31)	108.0	C(34) – C(31) – H(31)	108.0
P(1) – C(31) – H(31)	108.0	C(32) – C(35) – H(35A)	109.5
C(32) – C(35) – H(35B)	109.5	H(35A) – C(35) – H(35B)	109.5
C(32) – C(35) – H(35C)	109.5	H(35A) – C(35) – H(35C)	109.5
H(35B) – C(35) – H(35C)	109.5	C(31) – C(34) – H(34A)	109.5
C(31) – C(34) – H(34B)	109.5	H(34A) – C(34) – H(34B)	109.5
C(31) – C(34) – H(34C)	109.5	H(34A) – C(34) – H(34C)	109.5
H(34B) – C(34) – H(34C)	109.5	C(32) – C(36) – H(36A)	109.5
C(32) – C(36) – H(36B)	109.5	H(36A) – C(36) – H(36B)	109.5
C(32) – C(36) – H(36C)	109.5	H(36A) – C(36) – H(36C)	109.5
H(36B) – C(36) – H(36C)	109.5	C(46) – C(42) – C(45)	111.79(18)
C(46) – C(42) – P(2)	108.71(14)	C(45) – C(42) – P(2)	111.01(14)
C(46) – C(42) – H(42)	108.4	C(45) – C(42) – H(42)	108.4
P(2) – C(42) – H(42)	108.4	C(41) – C(43) – H(43A)	109.5
C(41) – C(43) – H(43B)	109.5	H(43A) – C(43) – H(43B)	109.5
C(41) – C(43) – H(43C)	109.5	H(43A) – C(43) – H(43C)	109.5
H(43B) – C(43) – H(43C)	109.5	C(41) – C(44) – H(44A)	109.5
C(41) – C(44) – H(44B)	109.5	H(44A) – C(44) – H(44B)	109.5
C(41) – C(44) – H(44C)	109.5	H(44A) – C(44) – H(44C)	109.5
H(44B) – C(44) – H(44C)	109.5	C(42) – C(45) – H(45A)	109.5
C(42) – C(45) – H(45B)	109.5	H(45A) – C(45) – H(45B)	109.5
C(42) – C(45) – H(45C)	109.5	H(45A) – C(45) – H(45C)	109.5
H(45B) – C(45) – H(45C)	109.5	C(42) – C(46) – H(46A)	109.5

Continued on next page

Table S25. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(42)–C(46)–H(46B)	109.5	H(46A)–C(46)–H(46B)	109.5
C(42)–C(46)–H(46C)	109.5	H(46A)–C(46)–H(46C)	109.5
H(46B)–C(46)–H(46C)	109.5	N–C(51)–C(52)	120.79(19)
N–C(51)–C(56)	123.67(19)	C(52)–C(51)–C(56)	115.54(18)
C(53)–C(52)–C(51)	121.8(2)	C(53)–C(52)–H(52)	119.1
C(51)–C(52)–H(52)	119.1	C(52)–C(53)–C(54)	122.0(2)
C(52)–C(53)–H(53)	119.0	C(54)–C(53)–H(53)	119.0
C(55)–C(54)–C(53)	116.66(19)	C(55)–C(54)–C(57)	122.0(2)
C(53)–C(54)–C(57)	121.3(2)	C(56)–C(55)–C(54)	122.1(2)
C(56)–C(55)–H(55)	118.9	C(54)–C(55)–H(55)	118.9
C(55)–C(56)–C(51)	121.8(2)	C(55)–C(56)–H(56)	119.1
C(51)–C(56)–H(56)	119.1	C(54)–C(57)–H(57A)	109.5
C(54)–C(57)–H(57B)	109.5	H(57A)–C(57)–H(57B)	109.5
C(54)–C(57)–H(57C)	109.5	H(57A)–C(57)–H(57C)	109.5
H(57B)–C(57)–H(57C)	109.5	C(51)–N–Pd	126.94(15)
C(51)–N–H	112.3(18)	Pd–N–H	119.0(17)
C(44)–C(41)–C(43)	111.3(2)	C(44)–C(41)–P(2)	114.06(16)
C(43)–C(41)–P(2)	109.61(16)	C(44)–C(41)–H(41)	107.2
C(43)–C(41)–H(41)	107.2	P(2)–C(41)–H(41)	107.2

5.2 Crystal data for $\left[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNPh_2\right]$ (3)

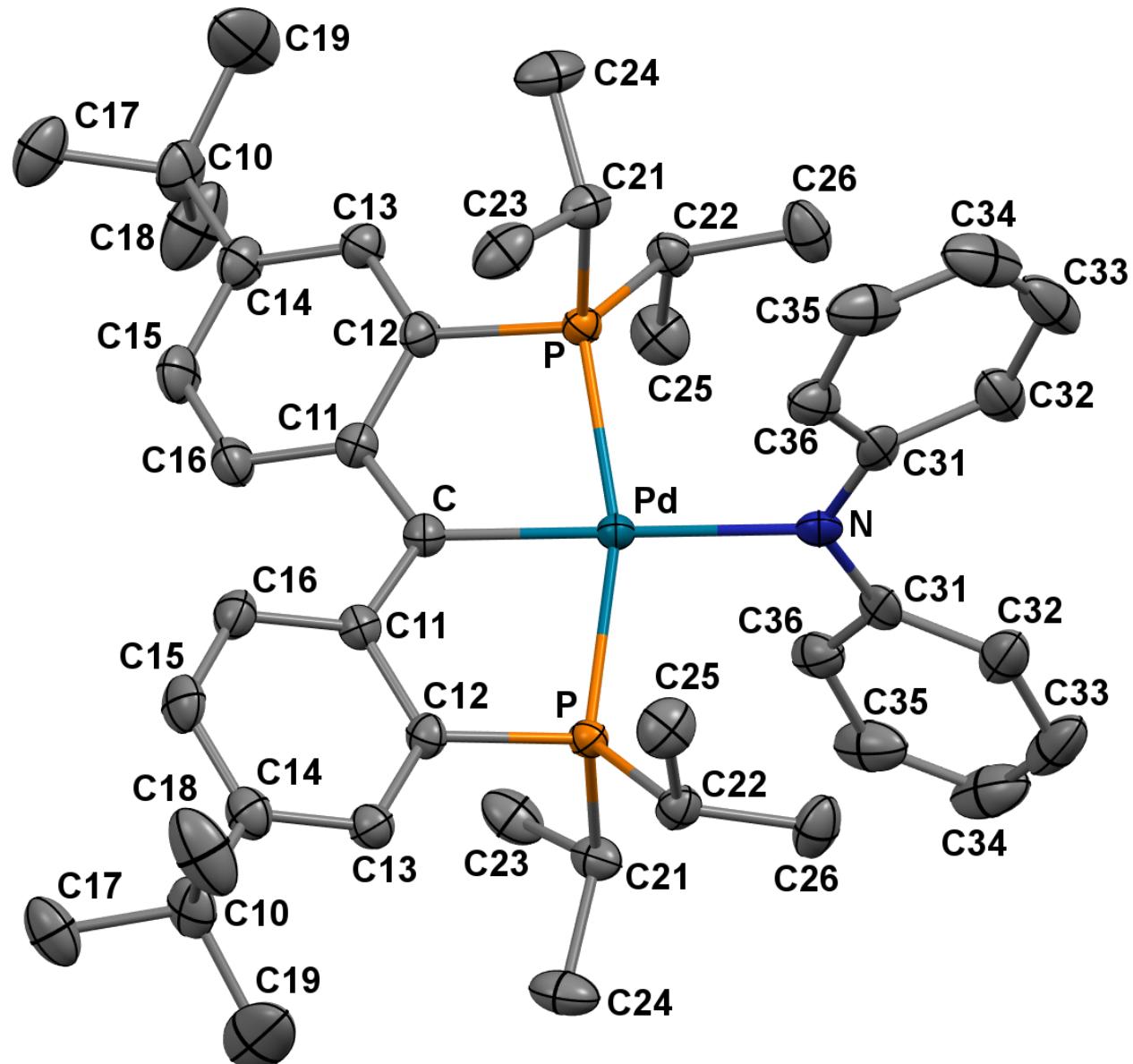


Figure S45. Thermal-ellipsoid representation of $\left[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNPh_2\right]$ (3) at 50% probability. Hydrogen atoms were omitted for clarity.

Table S26. Crystal data and structure refinement for $\left[\{\text{PC}^{\bullet}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNPh}_2\right]$ (**3**).

Identification code:	pc40b	
Empirical formula:	C ₄₅ H ₆₂ N Pd ₂	
Formula weight:	785.30	
Temperature:	120(2) K	
Wavelength:	0.71073 Å	
Crystal system:	Monoclinic	
Space group:	C2/c	
Unit cell dimensions:	$a = 20.2994(14)$ Å	$\alpha = 90^\circ$
	$b = 17.5268(14)$ Å	$\beta = 123.775(3)^\circ$
	$c = 15.9561(16)$ Å	$\gamma = 90^\circ$
Volume:	4718.8(7) Å ³	
Z:	4	
Density (calculated):	1.105 g·cm ⁻³	
Absorption coefficient (μ):	0.488 mm ⁻¹	
F(000):	1660	
Crystal size:	0.12 × 0.10 × 0.08 mm ³	
θ range for data collection:	1.68 to 25.00°	
Index ranges:	-24 ≤ h ≤ 24, -20 ≤ k ≤ 20, -18 ≤ l ≤ 18	
Reflections collected:	50505	
Independent reflections:	4154 [R _{int} = 0.0474]	
Completeness to θ = 25.00°:	100.0 %	
Absorption correction:	Semi-empirical from equivalents	
Max. and min. transmission:	0.7454 and 0.6915	
Refinement method:	Full-matrix least-squares on F ²	
Data / restraints / parameters:	4154 / 0 / 230	
Goodness-of-fit on F ² :	1.059	
Final R indices [I>2σ(I)]:	R ₁ = 0.0246, wR ₂ = 0.0587	
R indices (all data):	R ₁ = 0.0300, wR ₂ = 0.0604	
Largest diff. peak and hole:	0.481 and -0.418 e ⁻ ·Å ⁻³	

Table S27. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for $[\{\text{PC}^*(\text{sp}^2)\text{P}\}^t\text{Bu}\text{PdNPh}_2]$ (**3**). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor

atom	x	y	z	U(eq)
P	0.63131(3)	0.45974(2)	0.37636(3)	0.019(1)
N	0.5000	0.31863(12)	0.2500	0.025(1)
C	0.5000	0.55672(14)	0.2500	0.022(1)
Pd	0.5000	0.44123(1)	0.2500	0.018(1)
C(11)	0.57372(10)	0.59663(10)	0.28824(13)	0.022(1)
C(10)	0.80519(12)	0.70037(11)	0.39744(16)	0.035(1)
C(12)	0.64563(11)	0.55754(10)	0.35315(13)	0.022(1)
C(13)	0.71865(11)	0.59130(11)	0.39028(14)	0.025(1)
C(14)	0.72503(11)	0.66519(11)	0.36329(14)	0.029(1)
C(15)	0.65382(12)	0.70351(11)	0.29699(15)	0.032(1)
C(16)	0.58089(12)	0.67163(11)	0.26115(14)	0.028(1)
C(17)	0.80785(14)	0.78470(12)	0.42291(18)	0.044(1)
C(18)	0.81714(16)	0.69292(15)	0.3115(2)	0.060(1)
C(19)	0.87322(14)	0.66097(17)	0.4919(2)	0.073(1)
C(21)	0.65540(11)	0.45620(11)	0.50644(13)	0.026(1)
C(22)	0.71085(11)	0.40597(10)	0.38065(14)	0.024(1)
C(23)	0.59758(13)	0.50631(13)	0.51392(16)	0.037(1)
C(24)	0.74111(12)	0.47677(14)	0.58793(15)	0.040(1)
C(25)	0.70521(13)	0.41267(13)	0.28225(16)	0.037(1)
C(26)	0.71066(12)	0.32295(11)	0.40863(17)	0.036(1)
C(31)	0.48248(10)	0.28236(11)	0.16480(15)	0.028(1)
C(32)	0.44830(11)	0.20830(12)	0.13526(17)	0.039(1)
C(33)	0.42575(13)	0.17757(15)	0.0427(2)	0.054(1)
C(34)	0.43374(14)	0.21792(17)	-0.0255(2)	0.057(1)
C(35)	0.46666(13)	0.28965(15)	0.00073(16)	0.046(1)
C(36)	0.49112(11)	0.32041(12)	0.09395(14)	0.032(1)
H(13)	0.7656	0.5633	0.4353	0.030
H(15)	0.6560	0.7536	0.2759	0.039
H(16)	0.5344	0.7005	0.2173	0.034
H(17A)	0.7658	0.8121	0.3634	0.066
H(17B)	0.8594	0.8061	0.4437	0.066
H(17C)	0.8001	0.7901	0.4780	0.066
H(18A)	0.8198	0.6388	0.2981	0.090
H(18B)	0.8667	0.7182	0.3303	0.090
H(18C)	0.7726	0.7170	0.2507	0.090
H(19A)	0.8779	0.6082	0.4757	0.109
H(19B)	0.8625	0.6613	0.5447	0.109
H(19C)	0.9228	0.6882	0.5159	0.109
H(21)	0.6467	0.4025	0.5193	0.032

Continued on next page

Table S27. – continued from previous page

atom	x	y	x	U(eq)
H(22)	0.7625	0.4289	0.4348	0.029
H(23A)	0.6074	0.5600	0.5073	0.056
H(23B)	0.6050	0.4983	0.5794	0.056
H(23C)	0.5432	0.4929	0.4599	0.056
H(24A)	0.7770	0.4413	0.5847	0.060
H(24B)	0.7492	0.4732	0.6544	0.060
H(24C)	0.7521	0.5290	0.5770	0.060
H(25A)	0.6559	0.3889	0.2281	0.055
H(25B)	0.7506	0.3868	0.2885	0.055
H(25C)	0.7055	0.4667	0.2665	0.055
H(26A)	0.6621	0.2981	0.3547	0.055
H(26B)	0.7130	0.3206	0.4716	0.055
H(26C)	0.7568	0.2967	0.4176	0.055
H(32)	0.4408	0.1794	0.1796	0.047
H(33)	0.4043	0.1275	0.0258	0.065
H(34)	0.4169	0.1966	-0.0892	0.069
H(35)	0.4727	0.3183	-0.0451	0.055
H(36)	0.5149	0.3696	0.1106	0.039

Table S28. Anisotropic displacement parameters (\AA^2) for $[\{\text{PC}^\bullet(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNPh}_2]$ (**3**). The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^*{}^2\textbf{U}_{11} + \dots + 2hka^*b^*\textbf{U}_{12}]$.

atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
P	0.0158(2)	0.0192(2)	0.0195(2)	0.0015(2)	0.0086(2)	-0.0017(2)
N	0.0133(10)	0.0252(12)	0.0225(11)	0.000	0.0051(9)	0.000
C	0.0230(13)	0.0201(13)	0.0215(12)	0.000	0.0112(11)	0.000
Pd	0.0154(1)	0.0165(1)	0.0184(1)	0.000	0.0078(1)	0.000
C(11)	0.0229(10)	0.0206(9)	0.0229(9)	-0.0009(7)	0.0126(8)	-0.0009(7)
C(10)	0.0325(11)	0.0284(11)	0.0495(13)	0.0020(9)	0.0266(10)	-0.0090(9)
C(12)	0.0242(10)	0.0199(9)	0.0228(9)	0.0019(7)	0.0138(8)	-0.0023(8)
C(13)	0.0230(10)	0.0254(10)	0.0264(9)	0.0019(8)	0.0130(8)	-0.0018(8)
C(14)	0.0305(11)	0.0266(10)	0.0323(10)	-0.0003(8)	0.0196(9)	-0.0055(8)
C(15)	0.0376(12)	0.0226(10)	0.0390(11)	0.0063(8)	0.0224(10)	-0.0046(9)
C(16)	0.0306(11)	0.0222(10)	0.0305(10)	0.0052(8)	0.0161(9)	0.0031(8)
C(17)	0.0465(14)	0.0378(13)	0.0588(14)	-0.0127(11)	0.0357(12)	-0.0189(11)
C(18)	0.0634(17)	0.0557(16)	0.093(2)	-0.0278(15)	0.0634(17)	-0.0282(14)
C(19)	0.0291(13)	0.0628(18)	0.096(2)	0.0280(16)	0.0159(14)	-0.0176(12)
C(21)	0.0255(10)	0.0317(11)	0.0210(9)	0.0003(8)	0.0124(8)	-0.0068(8)
C(22)	0.0189(9)	0.0240(10)	0.0285(10)	0.0040(8)	0.0123(8)	0.0010(8)
C(23)	0.0391(12)	0.0459(13)	0.0336(11)	-0.0131(10)	0.0245(10)	-0.0111(10)
C(24)	0.0312(12)	0.0587(14)	0.0223(10)	-0.0023(10)	0.0105(9)	-0.0106(11)
C(25)	0.0375(12)	0.0406(12)	0.0403(12)	0.0022(10)	0.0269(10)	0.0051(10)
C(26)	0.0302(11)	0.0285(11)	0.0555(13)	0.0106(10)	0.0267(11)	0.0081(9)
C(31)	0.0149(9)	0.0222(10)	0.0386(11)	-0.0020(8)	0.0100(8)	0.0091(7)
C(32)	0.0222(10)	0.0309(11)	0.0532(13)	-0.0117(10)	0.0145(10)	0.0016(9)
C(33)	0.0248(12)	0.0448(14)	0.0731(18)	-0.0340(13)	0.0154(12)	-0.0022(10)
C(34)	0.0331(13)	0.0748(19)	0.0468(14)	-0.0314(14)	0.0118(12)	0.0114(13)
C(35)	0.0331(12)	0.0638(16)	0.0341(12)	-0.0087(11)	0.0144(10)	0.0180(11)
C(36)	0.0244(10)	0.0346(11)	0.0301(10)	-0.0048(9)	0.0106(9)	0.0096(8)

Table S29. Distances [Å] for $[\{PC^*(sp^2)P\}^{tBu}PdNPh_2]$ (**3**).

atom – atom	distance	atom – atom	distance
P–C(12)	1.8110(17)	P–C(22)	1.8376(18)
P–C(21)	1.8500(18)	P–Pd	2.3021(5)
N–C(31)	1.357(2)	N–C(31)#1	1.357(2)
N–Pd	2.149(2)	C–C(11)	1.444(2)
C–C(11)#1	1.444(2)	C–Pd	2.024(2)
Pd–P#1	2.3021(5)	C(11)–C(12)	1.412(3)
C(11)–C(16)	1.416(3)	C(10)–C(17)	1.526(3)
C(10)–C(18)	1.526(3)	C(10)–C(14)	1.529(3)
C(10)–C(19)	1.530(3)	C(12)–C(13)	1.386(3)
C(13)–C(14)	1.393(3)	C(13)–H(13)	0.9500
C(14)–C(15)	1.400(3)	C(15)–C(16)	1.375(3)
C(15)–H(15)	0.9500	C(16)–H(16)	0.9500
C(17)–H(17A)	0.9800	C(17)–H(17B)	0.9800
C(17)–H(17C)	0.9800	C(18)–H(18A)	0.9800
C(18)–H(18B)	0.9800	C(18)–H(18C)	0.9800
C(19)–H(19A)	0.9800	C(19)–H(19B)	0.9800
C(19)–H(19C)	0.9800	C(21)–C(23)	1.523(3)
C(21)–C(24)	1.527(3)	C(21)–H(21)	1.0000
C(22)–C(25)	1.514(3)	C(22)–C(26)	1.523(3)
C(22)–H(22)	1.0000	C(23)–H(23A)	0.9800
C(23)–H(23B)	0.9800	C(23)–H(23C)	0.9800
C(24)–H(24A)	0.9800	C(24)–H(24B)	0.9800
C(24)–H(24C)	0.9800	C(25)–H(25A)	0.9800
C(25)–H(25B)	0.9800	C(25)–H(25C)	0.9800
C(26)–H(26A)	0.9800	C(26)–H(26B)	0.9800
C(26)–H(26C)	0.9800	C(31)–C(36)	1.405(3)
C(31)–C(32)	1.423(3)	C(32)–C(33)	1.389(3)
C(32)–H(32)	0.9500	C(33)–C(34)	1.381(4)
C(33)–H(33)	0.9500	C(34)–C(35)	1.375(4)
C(34)–H(34)	0.9500	C(35)–C(36)	1.389(3)
C(35)–H(35)	0.9500	C(36)–H(36)	0.9500

Symmetry transformations used to generate equivalent atoms: #1 $-x + 1, y, -z + \frac{1}{2}$

Table S30. Angles [°] for $[\{PC^{\bullet}(sp^2)P\}^{tBu}PdNPh_2]$ (**3**).

atom – atom – atom	angle	atom – atom – atom	angle
C(12) – P – C(22)	104.80(8)	C(12) – P – C(21)	106.78(8)
C(22) – P – C(21)	105.20(9)	C(12) – P – Pd	101.17(6)
C(22) – P – Pd	121.73(6)	C(21) – P – Pd	115.66(6)
C(31) – N – C(31)#1	124.1(2)	C(31) – N – Pd	117.94(12)
C(31)#1 – N – Pd	117.94(12)	C(11) – C – C(11)#1	122.1(2)
C(11) – C – Pd	118.97(11)	C(11)#1 – C – Pd	118.97(11)
C – Pd – N	180.0	C – Pd – P	81.900(12)
N – Pd – P	98.100(12)	C – Pd – P#1	81.900(12)
N – Pd – P#1	98.100(12)	P – Pd – P#1	163.80(2)
C(12) – C(11) – C(16)	115.76(16)	C(12) – C(11) – C	118.75(16)
C(16) – C(11) – C	125.36(17)	C(17) – C(10) – C(18)	108.82(18)
C(17) – C(10) – C(14)	111.02(17)	C(18) – C(10) – C(14)	108.33(18)
C(17) – C(10) – C(19)	107.2(2)	C(18) – C(10) – C(19)	109.9(2)
C(14) – C(10) – C(19)	111.57(17)	C(13) – C(12) – C(11)	122.03(16)
C(13) – C(12) – P	124.87(14)	C(11) – C(12) – P	113.04(13)
C(12) – C(13) – C(14)	121.72(17)	C(12) – C(13) – H(13)	119.1
C(14) – C(13) – H(13)	119.1	C(13) – C(14) – C(15)	116.43(17)
C(13) – C(14) – C(10)	121.87(17)	C(15) – C(14) – C(10)	121.53(17)
C(16) – C(15) – C(14)	122.70(18)	C(16) – C(15) – H(15)	118.7
C(14) – C(15) – H(15)	118.7	C(15) – C(16) – C(11)	121.33(18)
C(15) – C(16) – H(16)	119.3	C(11) – C(16) – H(16)	119.3
C(10) – C(17) – H(17A)	109.5	C(10) – C(17) – H(17B)	109.5
H(17A) – C(17) – H(17B)	109.5	C(10) – C(17) – H(17C)	109.5
H(17A) – C(17) – H(17C)	109.5	H(17B) – C(17) – H(17C)	109.5
C(10) – C(18) – H(18A)	109.5	C(10) – C(18) – H(18B)	109.5
H(18A) – C(18) – H(18B)	109.5	C(10) – C(18) – H(18C)	109.5
H(18A) – C(18) – H(18C)	109.5	H(18B) – C(18) – H(18C)	109.5
C(10) – C(19) – H(19A)	109.5	C(10) – C(19) – H(19B)	109.5
H(19A) – C(19) – H(19B)	109.5	C(10) – C(19) – H(19C)	109.5
H(19A) – C(19) – H(19C)	109.5	H(19B) – C(19) – H(19C)	109.5
C(23) – C(21) – C(24)	111.10(16)	C(23) – C(21) – P	109.52(13)
C(24) – C(21) – P	114.38(13)	C(23) – C(21) – H(21)	107.2
C(24) – C(21) – H(21)	107.2	P – C(21) – H(21)	107.2
C(25) – C(22) – C(26)	111.53(17)	C(25) – C(22) – P	111.30(13)
C(26) – C(22) – P	110.84(13)	C(25) – C(22) – H(22)	107.7
C(26) – C(22) – H(22)	107.7	P – C(22) – H(22)	107.7
C(21) – C(23) – H(23A)	109.5	C(21) – C(23) – H(23B)	109.5
H(23A) – C(23) – H(23B)	109.5	C(21) – C(23) – H(23C)	109.5
H(23A) – C(23) – H(23C)	109.5	H(23B) – C(23) – H(23C)	109.5
C(21) – C(24) – H(24A)	109.5	C(21) – C(24) – H(24B)	109.5

Symmetry transformations used to generate equivalent atoms: #1 $-x + 1, y, -z + \frac{1}{2}$

Continued on next page

Table S30. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
H(24A)–C(24)–H(24B)	109.5	C(21)–C(24)–H(24C)	109.5
H(24A)–C(24)–H(24C)	109.5	H(24B)–C(24)–H(24C)	109.5
C(22)–C(25)–H(25A)	109.5	C(22)–C(25)–H(25B)	109.5
H(25A)–C(25)–H(25B)	109.5	C(22)–C(25)–H(25C)	109.5
H(25A)–C(25)–H(25C)	109.5	H(25B)–C(25)–H(25C)	109.5
C(22)–C(26)–H(26A)	109.5	C(22)–C(26)–H(26B)	109.5
H(26A)–C(26)–H(26B)	109.5	C(22)–C(26)–H(26C)	109.5
H(26A)–C(26)–H(26C)	109.5	H(26B)–C(26)–H(26C)	109.5
N–C(31)–C(36)	120.22(18)	N–C(31)–C(32)	124.5(2)
C(36)–C(31)–C(32)	115.00(19)	C(33)–C(32)–C(31)	121.2(2)
C(33)–C(32)–H(32)	119.4	C(31)–C(32)–H(32)	119.4
C(34)–C(33)–C(32)	121.7(2)	C(34)–C(33)–H(33)	119.1
C(32)–C(33)–H(33)	119.1	C(35)–C(34)–C(33)	118.6(2)
C(35)–C(34)–H(34)	120.7	C(33)–C(34)–H(34)	120.7
C(34)–C(35)–C(36)	120.3(3)	C(34)–C(35)–H(35)	119.8
C(36)–C(35)–H(35)	119.8	C(35)–C(36)–C(31)	123.2(2)
C(35)–C(36)–H(36)	118.4	C(31)–C(36)–H(36)	118.4

Symmetry transformations used to generate equivalent atoms: #1 $-x + 1, y, -z + \frac{1}{2}$

5.3 Crystal data for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNPh}_2\right]^-[\text{KOEt}_2]^+$ (5)

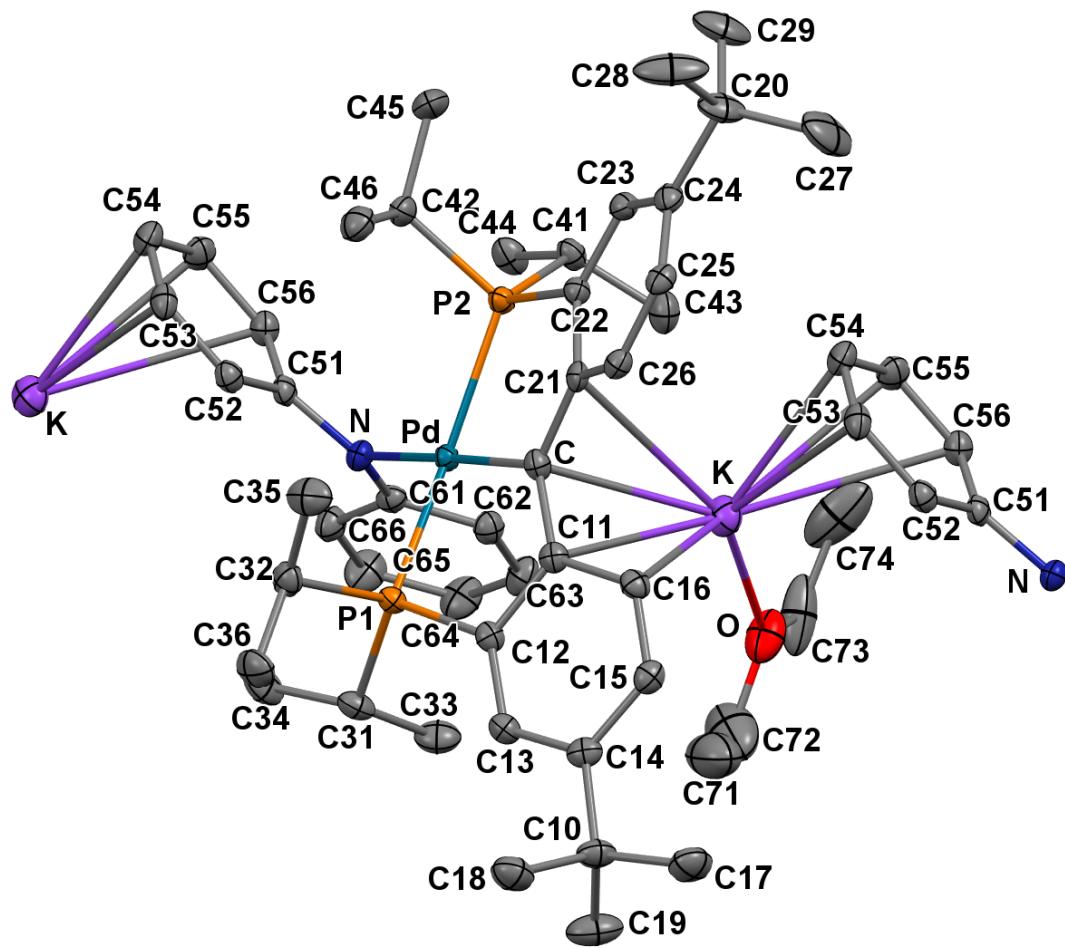


Figure S46. Thermal-ellipsoid representation of $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNPh}_2\right]^-[\text{KOEt}_2]^+$ (5) at 50% probability. Hydrogen atoms were omitted for clarity.

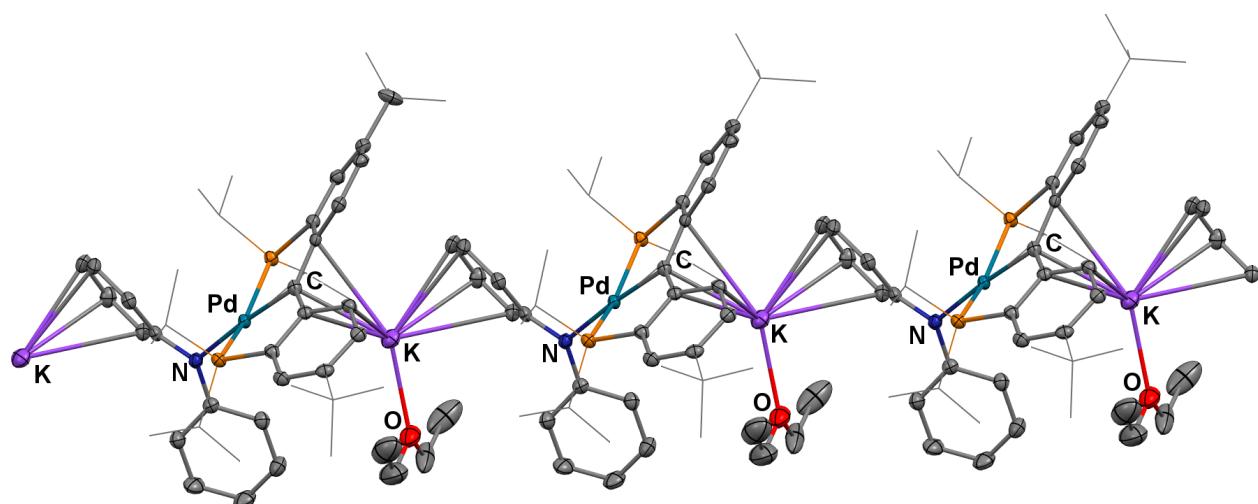


Figure S47. Polymeric structure of $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdNPh}_2\right]^-[\text{KOEt}_2]^+$ (5).

Table S31. Crystal data and structure refinement for $[\text{PC}(\text{sp}^2)\text{P}]^{t\text{Bu}}\text{PdNPh}_2^-[\text{KOEt}_2]^+$ (**5**).

Identification code:	pc47	
Empirical formula:	C ₄₉ H ₇₂ KNOP ₂ Pd	
Formula weight:	898.52	
Temperature:	120(2) K	
Wavelength:	0.71073 Å	
Crystal system:	Monoclinic	
Space group:	P2 ₁ /n	
Unit cell dimensions:	$a = 10.2409(9)$ Å	$\alpha = 90^\circ$
	$b = 18.1690(15)$ Å	$\beta = 90.553(3)^\circ$
	$c = 25.743(2)$ Å	$\gamma = 90^\circ$
Volume:	4789.7(7) Å ³	
Z:	4	
Density (calculated):	1.246 g·cm ⁻³	
Absorption coefficient (μ):	0.575 mm ⁻¹	
F(000):	1904	
Crystal size:	0.12 × 0.11 × 0.07 mm ³	
θ range for data collection:	1.94 to 25.00°	
Index ranges:	-12 ≤ h ≤ 12, -15 ≤ k ≤ 21, -30 ≤ l ≤ 30	
Reflections collected:	75438	
Independent reflections:	8436 [R _{int} = 0.0384]	
Completeness to θ = 25.00°:	99.9 %	
Absorption correction:	Semi-empirical from equivalents	
Max. and min. transmission:	0.7457 and 0.6996	
Refinement method:	Full-matrix least-squares on F ²	
Data / restraints / parameters:	8436 / 0 / 512	
Goodness-of-fit on F ² :	1.049	
Final R indices [I>2σ(I)]:	R ₁ = 0.0252, wR ₂ = 0.0563	
R indices (all data):	R ₁ = 0.0318, wR ₂ = 0.0582	
Largest diff. peak and hole:	0.420 and -0.279 e ⁻ ·Å ⁻³	

Table S32. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for $\{\text{PC(sp}^2\text{)P}\}^{\text{tBu}}\text{PdNPh}_2\}^-\text{[KOEt}_2]^+$ (**5**). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor

atom	x	y	z	U(eq)
Pd	0.49711(1)	0.09320(1)	0.20563(1)	0.013(1)
C(22)	0.66834(18)	0.09791(11)	0.31091(7)	0.016(1)
P(1)	0.48476(5)	0.17596(3)	0.13905(2)	0.015(1)
C(23)	0.72261(19)	0.09162(11)	0.36063(7)	0.018(1)
P(2)	0.53023(5)	0.04634(3)	0.28798(2)	0.015(1)
C(24)	0.81603(19)	0.14046(12)	0.37919(8)	0.019(1)
C(26)	0.79972(18)	0.20395(11)	0.29617(8)	0.017(1)
C(21)	0.71084(18)	0.15156(11)	0.27465(8)	0.015(1)
C	0.66299(19)	0.15148(11)	0.22225(8)	0.017(1)
C(20)	0.8791(2)	0.13562(14)	0.43302(8)	0.030(1)
C(28)	0.8456(3)	0.20500(15)	0.46354(10)	0.054(1)
C(27)	1.0275(3)	0.12999(19)	0.42759(11)	0.053(1)
C(29)	0.8303(3)	0.06954(15)	0.46354(9)	0.041(1)
C(12)	0.64850(19)	0.21001(11)	0.13508(8)	0.017(1)
C(11)	0.72145(19)	0.19388(11)	0.18149(8)	0.016(1)
C(10)	0.8855(2)	0.31327(12)	0.04687(8)	0.023(1)
C(15)	0.9054(2)	0.25212(11)	0.13664(8)	0.019(1)
C(14)	0.8319(2)	0.26995(11)	0.09265(8)	0.019(1)
C(13)	0.70181(19)	0.24704(11)	0.09343(8)	0.018(1)
C(17)	1.0306(2)	0.32939(14)	0.05312(9)	0.032(1)
C(18)	0.8139(2)	0.38720(13)	0.04304(9)	0.031(1)
C(19)	0.8650(3)	0.27129(15)	-0.00369(9)	0.039(1)
N	0.33184(15)	0.02650(9)	0.18497(6)	0.016(1)
C(31)	0.4302(2)	0.15321(12)	0.07285(8)	0.022(1)
C(33)	0.5107(2)	0.09055(13)	0.05058(8)	0.031(1)
C(34)	0.2846(2)	0.13484(14)	0.07147(9)	0.032(1)
C(42)	0.39437(19)	0.06894(12)	0.33238(8)	0.020(1)
C(46)	0.3827(2)	0.15188(12)	0.33794(9)	0.026(1)
C(51)	0.22261(18)	0.03394(11)	0.21513(8)	0.017(1)
C(52)	0.18579(19)	0.10431(11)	0.23196(8)	0.019(1)
C(53)	0.09146(19)	0.11463(12)	0.26896(8)	0.022(1)
C(54)	0.0262(2)	0.05511(12)	0.29029(8)	0.024(1)
C(55)	0.0575(2)	-0.01465(12)	0.27271(8)	0.023(1)
C(56)	0.15123(19)	-0.02551(12)	0.23540(8)	0.019(1)
C(61)	0.34516(19)	-0.03273(11)	0.15206(8)	0.017(1)
C(62)	0.4703(2)	-0.05907(11)	0.14013(8)	0.020(1)
C(63)	0.4879(2)	-0.11539(12)	0.10518(9)	0.026(1)
C(64)	0.3834(2)	-0.14855(13)	0.08055(9)	0.033(1)
C(65)	0.2596(2)	-0.12290(13)	0.09134(9)	0.032(1)

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Table S32. – continued from previous page

atom	x	y	x	U(eq)
C(66)	0.2407(2)	-0.06625(12)	0.12570(8)	0.023(1)
C(71)	0.8918(4)	0.0677(2)	0.06195(12)	0.068(1)
C(72)	0.8650(4)	-0.0136(2)	0.05975(13)	0.073(1)
C(73)	0.8817(3)	-0.12236(17)	0.10856(15)	0.062(1)
C(74)	0.9095(4)	-0.15203(17)	0.16145(15)	0.071(1)
C(45)	0.4011(2)	0.03314(14)	0.38637(8)	0.030(1)
C(41)	0.56477(19)	-0.05181(11)	0.29988(8)	0.019(1)
C(43)	0.6881(2)	-0.07526(12)	0.27211(9)	0.027(1)
C(44)	0.4482(2)	-0.09905(12)	0.28349(9)	0.025(1)
C(32)	0.3831(2)	0.25641(11)	0.15708(8)	0.022(1)
C(35)	0.4233(3)	0.28431(13)	0.21019(9)	0.032(1)
C(36)	0.3853(2)	0.31911(12)	0.11770(9)	0.031(1)
C(16)	0.85414(19)	0.21697(11)	0.17915(8)	0.018(1)
O	0.88447(16)	-0.04438(10)	0.10966(7)	0.041(1)
C(25)	0.84866(19)	0.19826(11)	0.34570(8)	0.018(1)
K	0.87015(5)	0.04914(3)	0.19096(2)	0.029(1)
H(23)	0.6947	0.0526	0.3825	0.021
H(26)	0.8260	0.2444	0.2753	0.020
H(28A)	0.8777	0.2483	0.4449	0.080
H(28B)	0.8870	0.2027	0.4980	0.080
H(28C)	0.7507	0.2086	0.4673	0.080
H(27A)	1.0495	0.0844	0.4093	0.080
H(27B)	1.0684	0.1296	0.4622	0.080
H(27C)	1.0597	0.1723	0.4079	0.080
H(29A)	0.7357	0.0735	0.4681	0.062
H(29B)	0.8735	0.0683	0.4977	0.062
H(29C)	0.8503	0.0243	0.4445	0.062
H(15)	0.9953	0.2649	0.1373	0.023
H(13)	0.6477	0.2573	0.0641	0.022
H(17A)	1.0791	0.2829	0.0549	0.049
H(17B)	1.0455	0.3575	0.0851	0.049
H(17C)	1.0607	0.3581	0.0233	0.049
H(18A)	0.8468	0.4150	0.0133	0.047
H(18B)	0.8291	0.4155	0.0750	0.047
H(18C)	0.7200	0.3784	0.0385	0.047
H(19A)	0.8962	0.3011	-0.0328	0.058
H(19B)	0.7719	0.2608	-0.0086	0.058
H(19C)	0.9139	0.2249	-0.0023	0.058
H(31)	0.4441	0.1976	0.0506	0.026
H(33A)	0.4818	0.0805	0.0149	0.046
H(33B)	0.4988	0.0463	0.0718	0.046
H(33C)	0.6031	0.1044	0.0507	0.046

Continued on next page

Table S32. – continued from previous page

atom	x	y	x	U(eq)
H(34A)	0.2581	0.1224	0.0359	0.047
H(34B)	0.2345	0.1775	0.0833	0.047
H(34C)	0.2678	0.0928	0.0943	0.047
H(42)	0.3120	0.0513	0.3153	0.024
H(46A)	0.3781	0.1745	0.3034	0.040
H(46B)	0.4591	0.1709	0.3568	0.040
H(46C)	0.3033	0.1638	0.3572	0.040
H(52)	0.2273	0.1461	0.2174	0.023
H(53)	0.0708	0.1631	0.2800	0.026
H(54)	-0.0380	0.0620	0.3162	0.029
H(55)	0.0133	-0.0560	0.2867	0.028
H(56)	0.1679	-0.0739	0.2232	0.023
H(62)	0.5444	-0.0375	0.1566	0.024
H(63)	0.5739	-0.1317	0.0979	0.032
H(64)	0.3961	-0.1879	0.0568	0.039
H(65)	0.1862	-0.1449	0.0747	0.038
H(66)	0.1544	-0.0493	0.1318	0.027
H(71A)	0.8811	0.0889	0.0272	0.102
H(71B)	0.9813	0.0760	0.0744	0.102
H(71C)	0.8303	0.0912	0.0857	0.102
H(72A)	0.7739	-0.0223	0.0480	0.087
H(72B)	0.9243	-0.0373	0.0346	0.087
H(73A)	0.9479	-0.1408	0.0840	0.074
H(73B)	0.7948	-0.1394	0.0965	0.074
H(74A)	0.9061	-0.2059	0.1605	0.106
H(74B)	0.8441	-0.1335	0.1857	0.106
H(74C)	0.9966	-0.1361	0.1729	0.106
H(45A)	0.4033	-0.0205	0.3826	0.045
H(45B)	0.3241	0.0474	0.4063	0.045
H(45C)	0.4802	0.0497	0.4046	0.045
H(41)	0.5794	-0.0587	0.3380	0.023
H(43A)	0.6752	-0.0701	0.2345	0.040
H(43B)	0.7077	-0.1267	0.2805	0.040
H(43C)	0.7611	-0.0441	0.2834	0.040
H(44A)	0.4287	-0.0908	0.2466	0.037
H(44B)	0.3720	-0.0855	0.3042	0.037
H(44C)	0.4689	-0.1511	0.2892	0.037
H(32)	0.2908	0.2389	0.1595	0.027
H(35A)	0.5136	0.3020	0.2091	0.048
H(35B)	0.4167	0.2443	0.2355	0.048
H(35C)	0.3656	0.3248	0.2203	0.048
H(36A)	0.3502	0.3017	0.0844	0.046

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Table S32. – continued from previous page

atom	x	y	x	U(eq)
H(36B)	0.4754	0.3359	0.1131	0.046
H(36C)	0.3319	0.3600	0.1303	0.046
H(16)	0.9097	0.2077	0.2082	0.022
H(25)	0.9072	0.2351	0.3579	0.021

Table S33. Anisotropic displacement parameters (\AA^2) for $\{[\text{PC}(\text{sp}^2)\text{P}]^{t\text{Bu}}\text{PdNPh}_2\}^-[\text{KOEt}_2]^+$ (**5**). The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}\text{U}_{11} + \dots + 2hka^*b^*\text{U}_{12}]$.

atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
Pd	0.0121(1)	0.0128(1)	0.0138(1)	0.0005(1)	-0.0009(1)	-0.0008(1)
C(22)	0.0144(9)	0.0161(10)	0.0161(10)	-0.0011(8)	0.0004(8)	0.0005(8)
P(1)	0.0155(3)	0.0147(3)	0.0147(3)	0.0010(2)	-0.0020(2)	-0.0007(2)
C(23)	0.0182(10)	0.0190(10)	0.0156(10)	0.0003(9)	0.0024(8)	0.0011(9)
P(2)	0.0144(2)	0.0160(3)	0.0154(3)	0.0021(2)	-0.0008(2)	-0.0010(2)
C(24)	0.0171(10)	0.0239(11)	0.0162(10)	-0.0038(9)	0.0003(8)	0.0021(9)
C(26)	0.0152(10)	0.0151(10)	0.0196(10)	-0.0007(8)	0.0032(8)	0.0004(8)
C(21)	0.0115(9)	0.0155(10)	0.0183(10)	-0.0012(8)	0.0011(8)	0.0034(8)
C	0.0145(10)	0.0166(10)	0.0198(10)	0.0004(8)	-0.0021(8)	0.0004(8)
C(20)	0.0390(14)	0.0356(14)	0.0155(11)	0.0007(10)	-0.0080(10)	-0.0128(11)
C(28)	0.100(3)	0.0421(17)	0.0187(13)	-0.0078(12)	0.0069(14)	-0.0323(17)
C(27)	0.0405(16)	0.078(2)	0.0412(16)	0.0167(16)	-0.0246(13)	-0.0161(15)
C(29)	0.0603(18)	0.0410(15)	0.0224(12)	0.0075(11)	-0.0183(12)	-0.0176(13)
C(12)	0.0167(10)	0.0152(10)	0.0177(10)	-0.0014(8)	-0.0021(8)	-0.0011(8)
C(11)	0.0172(10)	0.0132(10)	0.0176(10)	-0.0023(8)	-0.0006(8)	-0.0004(8)
C(10)	0.0270(12)	0.0240(12)	0.0187(11)	0.0026(9)	0.0025(9)	-0.0069(9)
C(15)	0.0176(10)	0.0184(11)	0.0215(11)	-0.0031(9)	0.0010(8)	-0.0034(8)
C(14)	0.0242(11)	0.0144(10)	0.0173(10)	-0.0023(8)	0.0019(8)	-0.0025(9)
C(13)	0.0202(10)	0.0165(10)	0.0173(10)	0.0005(8)	-0.0026(8)	-0.0003(8)
C(17)	0.0325(13)	0.0351(14)	0.0301(13)	0.0084(11)	0.0056(10)	-0.0077(11)
C(18)	0.0360(13)	0.0299(13)	0.0272(12)	0.0075(10)	-0.0005(10)	-0.0034(11)
C(19)	0.0556(17)	0.0399(15)	0.0205(12)	-0.0030(11)	0.0066(11)	-0.0163(13)
N	0.0138(8)	0.0169(9)	0.0189(9)	-0.0014(7)	0.0010(7)	-0.0004(7)
C(31)	0.0261(11)	0.0217(11)	0.0168(10)	0.0033(9)	-0.0055(9)	-0.0048(9)
C(33)	0.0418(14)	0.0296(13)	0.0200(11)	-0.0046(10)	0.0006(10)	-0.0037(11)
C(34)	0.0290(13)	0.0350(14)	0.0306(13)	0.0034(11)	-0.0129(10)	-0.0071(11)
C(42)	0.0146(10)	0.0257(11)	0.0195(11)	0.0025(9)	0.0015(8)	-0.0007(9)
C(46)	0.0257(12)	0.0286(13)	0.0250(12)	-0.0034(10)	0.0050(9)	0.0027(10)
C(51)	0.0118(9)	0.0199(11)	0.0180(10)	-0.0008(8)	-0.0032(8)	0.0005(8)
C(52)	0.0159(10)	0.0196(11)	0.0214(11)	0.0013(9)	-0.0026(8)	-0.0008(8)
C(53)	0.0165(10)	0.0229(11)	0.0249(11)	-0.0057(9)	-0.0019(9)	0.0047(9)
C(54)	0.0173(11)	0.0292(12)	0.0250(12)	-0.0013(10)	0.0047(9)	0.0041(9)
C(55)	0.0187(11)	0.0236(12)	0.0276(12)	0.0051(10)	0.0037(9)	-0.0021(9)
C(56)	0.0165(10)	0.0175(11)	0.0241(11)	-0.0005(9)	-0.0005(8)	0.0006(8)
C(61)	0.0185(10)	0.0151(10)	0.0167(10)	0.0015(8)	0.0011(8)	-0.0007(8)
C(62)	0.0172(10)	0.0187(11)	0.0232(11)	0.0009(9)	0.0000(8)	-0.0023(8)
C(63)	0.0233(11)	0.0231(12)	0.0329(13)	-0.0035(10)	0.0059(10)	0.0033(9)
C(64)	0.0362(14)	0.0274(13)	0.0343(14)	-0.0140(11)	0.0051(11)	-0.0007(11)
C(65)	0.0288(13)	0.0342(13)	0.0318(13)	-0.0138(11)	-0.0018(10)	-0.0088(11)
C(66)	0.0172(10)	0.0273(12)	0.0241(11)	-0.0029(10)	0.0005(9)	-0.0021(9)

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Table S33. – continued from previous page

atom	U₁₁	U₂₂	U₃₃	U₂₃	U₁₃	U₁₂
C(71)	0.083(3)	0.078(3)	0.0435(18)	0.0083(17)	0.0097(17)	0.029(2)
C(72)	0.083(3)	0.083(3)	0.051(2)	-0.0286(19)	-0.0119(18)	0.036(2)
C(73)	0.0251(14)	0.0452(18)	0.115(3)	-0.045(2)	-0.0109(16)	0.0021(13)
C(74)	0.078(2)	0.0309(17)	0.104(3)	-0.0127(18)	0.050(2)	-0.0190(16)
C(45)	0.0251(12)	0.0430(15)	0.0223(12)	0.0073(11)	0.0073(9)	0.0025(11)
C(41)	0.0186(10)	0.0176(11)	0.0216(11)	0.0048(9)	-0.0028(8)	-0.0010(9)
C(43)	0.0207(11)	0.0226(12)	0.0374(13)	0.0002(10)	-0.0009(10)	0.0030(9)
C(44)	0.0242(11)	0.0191(11)	0.0311(12)	0.0056(10)	-0.0057(9)	-0.0044(9)
C(32)	0.0217(11)	0.0187(11)	0.0260(11)	0.0025(9)	0.0013(9)	0.0040(9)
C(35)	0.0506(15)	0.0187(12)	0.0278(13)	-0.0027(10)	0.0049(11)	0.0076(11)
C(36)	0.0391(14)	0.0229(12)	0.0303(13)	0.0064(10)	0.0001(11)	0.0105(10)
C(16)	0.0186(10)	0.0192(11)	0.0163(10)	-0.0018(9)	-0.0029(8)	-0.0009(8)
O	0.0311(9)	0.0398(11)	0.0517(12)	-0.0208(9)	0.0016(8)	-0.0006(8)
C(25)	0.0164(10)	0.0184(11)	0.0183(10)	-0.0055(9)	0.0018(8)	-0.0019(8)
K	0.0322(3)	0.0233(3)	0.0317(3)	-0.0040(2)	-0.0002(2)	0.0052(2)

Table S34. Distances [Å] for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdNPh}_2\right]^-[\text{KOEt}_2]^+$ (**5**).

atom – atom	distance	atom – atom	distance
Pd–C	2.043(2)	Pd–N	2.1445(16)
Pd–P(1)	2.2827(5)	Pd–P(2)	2.3064(5)
C(22)–C(23)	1.395(3)	C(22)–C(21)	1.421(3)
C(22)–P(2)	1.792(2)	P(1)–C(12)	1.791(2)
P(1)–C(31)	1.835(2)	P(1)–C(32)	1.856(2)
C(23)–C(24)	1.386(3)	C(23)–H(23)	0.9500
P(2)–C(41)	1.843(2)	P(2)–C(42)	1.855(2)
C(24)–C(25)	1.401(3)	C(24)–C(20)	1.526(3)
C(26)–C(25)	1.369(3)	C(26)–C(21)	1.425(3)
C(26)–H(26)	0.9500	C(21)–C	1.431(3)
C(21)–K	3.292(2)	C–C(11)	1.437(3)
C–K	2.939(2)	C(20)–C(29)	1.522(3)
C(20)–C(28)	1.526(4)	C(20)–C(27)	1.532(4)
C(28)–H(28A)	0.9800	C(28)–H(28B)	0.9800
C(28)–H(28C)	0.9800	C(27)–H(27A)	0.9800
C(27)–H(27B)	0.9800	C(27)–H(27C)	0.9800
C(29)–H(29A)	0.9800	C(29)–H(29B)	0.9800
C(29)–H(29C)	0.9800	C(12)–C(13)	1.382(3)
C(12)–C(11)	1.433(3)	C(11)–C(16)	1.424(3)
C(11)–K	3.047(2)	C(10)–C(19)	1.521(3)
C(10)–C(17)	1.522(3)	C(10)–C(14)	1.524(3)
C(10)–C(18)	1.533(3)	C(15)–C(16)	1.375(3)
C(15)–C(14)	1.392(3)	C(15)–H(15)	0.9500
C(14)–C(13)	1.396(3)	C(13)–H(13)	0.9500
C(17)–H(17A)	0.9800	C(17)–H(17B)	0.9800
C(17)–H(17C)	0.9800	C(18)–H(18A)	0.9800
C(18)–H(18B)	0.9800	C(18)–H(18C)	0.9800
C(19)–H(19A)	0.9800	C(19)–H(19B)	0.9800
C(19)–H(19C)	0.9800	N–C(51)	1.375(3)
N–C(61)	1.377(3)	C(31)–C(33)	1.521(3)
C(31)–C(34)	1.528(3)	C(31)–H(31)	1.0000
C(33)–H(33A)	0.9800	C(33)–H(33B)	0.9800
C(33)–H(33C)	0.9800	C(34)–H(34A)	0.9800
C(34)–H(34B)	0.9800	C(34)–H(34C)	0.9800
C(42)–C(46)	1.519(3)	C(42)–C(45)	1.536(3)
C(42)–H(42)	1.0000	C(46)–H(46A)	0.9800
C(46)–H(46B)	0.9800	C(46)–H(46C)	0.9800
C(51)–C(52)	1.403(3)	C(51)–C(56)	1.407(3)
C(52)–C(53)	1.376(3)	C(52)–H(52)	0.9500
C(53)–C(54)	1.387(3)	C(53)–H(53)	0.9500

Symmetry transformations used to generate equivalent atoms: #1 $x + 1, y, z$

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Table S34. – continued from previous page

atom – atom	distance	atom – atom	distance
C(54)–C(55)	1.384(3)	C(54)–H(54)	0.9500
C(55)–C(56)	1.379(3)	C(55)–H(55)	0.9500
C(56)–H(56)	0.9500	C(61)–C(66)	1.401(3)
C(61)–C(62)	1.405(3)	C(62)–C(63)	1.376(3)
C(62)–H(62)	0.9500	C(63)–C(64)	1.377(3)
C(63)–H(63)	0.9500	C(64)–C(65)	1.382(3)
C(64)–H(64)	0.9500	C(65)–C(66)	1.372(3)
C(65)–H(65)	0.9500	C(66)–H(66)	0.9500
C(71)–C(72)	1.504(5)	C(71)–H(71A)	0.9800
C(71)–H(71B)	0.9800	C(71)–H(71C)	0.9800
C(72)–O	1.414(4)	C(72)–H(72A)	0.9900
C(72)–H(72B)	0.9900	C(73)–O	1.417(3)
C(73)–C(74)	1.489(5)	C(73)–H(73A)	0.9900
C(73)–H(73B)	0.9900	C(74)–H(74A)	0.9800
C(74)–H(74B)	0.9800	C(74)–H(74C)	0.9800
C(45)–H(45A)	0.9800	C(45)–H(45B)	0.9800
C(45)–H(45C)	0.9800	C(41)–C(43)	1.519(3)
C(41)–C(44)	1.527(3)	C(41)–H(41)	1.0000
C(43)–H(43A)	0.9800	C(43)–H(43B)	0.9800
C(43)–H(43C)	0.9800	C(44)–H(44A)	0.9800
C(44)–H(44B)	0.9800	C(44)–H(44C)	0.9800
C(32)–C(35)	1.512(3)	C(32)–C(36)	1.525(3)
C(32)–H(32)	1.0000	C(35)–H(35A)	0.9800
C(35)–H(35B)	0.9800	C(35)–H(35C)	0.9800
C(36)–H(36A)	0.9800	C(36)–H(36B)	0.9800
C(36)–H(36C)	0.9800	C(16)–K	3.069(2)
C(16)–H(16)	0.9500	O–K	2.7010(17)
C(25)–H(25)	0.9500	K–C(54)#1	3.004(2)
K–C(55)#1	3.062(2)	K–C(53)#1	3.240(2)
K–C(56)#1	3.372(2)	K–C(52)#1	3.536(2)
K–C(51)#1	3.667(2)		

Symmetry transformations used to generate equivalent atoms: #1 $x + 1, y, z$

Table S35. Angles [°] for $[\{PC(sp^2)P\}^{tBu}PdNPh_2]^- [KOEt_2]^+ (\mathbf{5})$.

atom – atom – atom	angle	atom – atom – atom	angle
C–Pd–N	175.74(7)	C–Pd–P(1)	81.71(6)
N–Pd–P(1)	98.52(5)	C–Pd–P(2)	83.36(6)
N–Pd–P(2)	97.36(5)	P(1)–Pd–P(2)	160.036(19)
C(23)–C(22)–C(21)	122.49(18)	C(23)–C(22)–P(2)	124.44(15)
C(21)–C(22)–P(2)	112.84(14)	C(12)–P(1)–C(31)	107.54(9)
C(12)–P(1)–C(32)	105.66(10)	C(31)–P(1)–C(32)	104.07(10)
C(12)–P(1)–Pd	103.02(7)	C(31)–P(1)–Pd	124.27(7)
C(32)–P(1)–Pd	110.99(7)	C(24)–C(23)–C(22)	122.08(19)
C(24)–C(23)–H(23)	119.0	C(22)–C(23)–H(23)	119.0
C(22)–P(2)–C(41)	107.55(9)	C(22)–P(2)–C(42)	105.99(9)
C(41)–P(2)–C(42)	104.81(9)	C(22)–P(2)–Pd	102.61(7)
C(41)–P(2)–Pd	122.44(7)	C(42)–P(2)–Pd	112.34(7)
C(23)–C(24)–C(25)	115.82(18)	C(23)–C(24)–C(20)	124.11(19)
C(25)–C(24)–C(20)	120.04(18)	C(25)–C(26)–C(21)	122.55(19)
C(25)–C(26)–H(26)	118.7	C(21)–C(26)–H(26)	118.7
C(22)–C(21)–C(26)	113.66(17)	C(22)–C(21)–C	120.96(18)
C(26)–C(21)–C	125.36(18)	C(22)–C(21)–K	101.48(12)
C(26)–C(21)–K	108.25(12)	C–C(21)–K	63.23(10)
C(21)–C–C(11)	123.14(18)	C(21)–C–Pd	118.31(14)
C(11)–C–Pd	118.45(14)	C(21)–C–K	91.00(11)
C(11)–C–K	80.34(11)	Pd–C–K	102.45(8)
C(29)–C(20)–C(24)	112.10(18)	C(29)–C(20)–C(28)	108.0(2)
C(24)–C(20)–C(28)	109.0(2)	C(29)–C(20)–C(27)	109.0(2)
C(24)–C(20)–C(27)	109.44(19)	C(28)–C(20)–C(27)	109.3(2)
C(13)–C(12)–C(11)	122.64(18)	C(13)–C(12)–P(1)	126.11(15)
C(11)–C(12)–P(1)	111.25(14)	C(16)–C(11)–C(12)	113.23(18)
C(16)–C(11)–C	126.38(18)	C(12)–C(11)–C	120.06(17)
C(16)–C(11)–K	77.37(11)	C(12)–C(11)–K	119.95(13)
C–C(11)–K	71.96(11)	C(19)–C(10)–C(17)	108.25(19)
C(19)–C(10)–C(14)	110.81(18)	C(17)–C(10)–C(14)	112.09(18)
C(19)–C(10)–C(18)	108.81(19)	C(17)–C(10)–C(18)	107.72(18)
C(14)–C(10)–C(18)	109.06(18)	C(16)–C(15)–C(14)	123.25(19)
C(16)–C(15)–H(15)	118.4	C(14)–C(15)–H(15)	118.4
C(15)–C(14)–C(13)	115.29(18)	C(15)–C(14)–C(10)	123.63(18)
C(13)–C(14)–C(10)	121.06(18)	C(12)–C(13)–C(14)	122.70(19)
C(12)–C(13)–H(13)	118.6	C(14)–C(13)–H(13)	118.6
C(51)–N–C(61)	120.68(17)	C(51)–N–Pd	116.68(13)
C(61)–N–Pd	120.77(13)	C(33)–C(31)–C(34)	111.08(19)
C(33)–C(31)–P(1)	110.96(15)	C(34)–C(31)–P(1)	111.03(15)
C(33)–C(31)–H(31)	107.9	C(34)–C(31)–H(31)	107.9

Symmetry transformations used to generate equivalent atoms: #1 $x + 1, y, z$

Continued on next page

Table S35. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
P(1)–C(31)–H(31)	107.9	C(46)–C(42)–C(45)	109.75(18)
C(46)–C(42)–P(2)	109.76(14)	C(45)–C(42)–P(2)	115.89(15)
C(46)–C(42)–H(42)	107.0	C(45)–C(42)–H(42)	107.0
P(2)–C(42)–H(42)	107.0	N–C(51)–C(52)	119.13(18)
N–C(51)–C(56)	124.23(18)	C(52)–C(51)–C(56)	116.28(18)
C(53)–C(52)–C(51)	122.1(2)	C(53)–C(52)–H(52)	119.0
C(51)–C(52)–H(52)	119.0	C(52)–C(53)–C(54)	120.8(2)
C(52)–C(53)–H(53)	119.6	C(54)–C(53)–H(53)	119.6
C(55)–C(54)–C(53)	118.09(19)	C(55)–C(54)–H(54)	121.0
C(53)–C(54)–H(54)	121.0	C(56)–C(55)–C(54)	121.6(2)
C(56)–C(55)–H(55)	119.2	C(54)–C(55)–H(55)	119.2
C(55)–C(56)–C(51)	121.1(2)	C(55)–C(56)–H(56)	119.5
C(51)–C(56)–H(56)	119.5	N–C(61)–C(66)	123.91(18)
N–C(61)–C(62)	119.78(18)	C(66)–C(61)–C(62)	116.12(19)
C(63)–C(62)–C(61)	121.45(19)	C(63)–C(62)–H(62)	119.3
C(61)–C(62)–H(62)	119.3	C(62)–C(63)–C(64)	121.4(2)
C(62)–C(63)–H(63)	119.3	C(64)–C(63)–H(63)	119.3
C(63)–C(64)–C(65)	118.0(2)	C(63)–C(64)–H(64)	121.0
C(65)–C(64)–H(64)	121.0	C(66)–C(65)–C(64)	121.2(2)
C(66)–C(65)–H(65)	119.4	C(64)–C(65)–H(65)	119.4
C(65)–C(66)–C(61)	121.8(2)	C(65)–C(66)–H(66)	119.1
C(61)–C(66)–H(66)	119.1	O–C(72)–C(71)	109.3(3)
O–C(72)–H(72A)	109.8	C(71)–C(72)–H(72A)	109.8
O–C(72)–H(72B)	109.8	C(71)–C(72)–H(72B)	109.8
H(72A)–C(72)–H(72B)	108.3	O–C(73)–C(74)	109.9(2)
O–C(73)–H(73A)	109.7	C(74)–C(73)–H(73A)	109.7
O–C(73)–H(73B)	109.7	C(74)–C(73)–H(73B)	109.7
H(73A)–C(73)–H(73B)	108.2	C(43)–C(41)–C(44)	111.36(18)
C(43)–C(41)–P(2)	110.63(14)	C(44)–C(41)–P(2)	110.46(14)
C(43)–C(41)–H(41)	108.1	C(44)–C(41)–H(41)	108.1
P(2)–C(41)–H(41)	108.1	C(35)–C(32)–C(36)	110.17(18)
C(35)–C(32)–P(1)	109.98(15)	C(36)–C(32)–P(1)	114.20(15)
C(35)–C(32)–H(32)	107.4	C(36)–C(32)–H(32)	107.4
P(1)–C(32)–H(32)	107.4	C(15)–C(16)–C(11)	122.81(19)
C(15)–C(16)–K	121.31(14)	C(11)–C(16)–K	75.71(11)
C(15)–C(16)–H(16)	118.6	C(11)–C(16)–H(16)	118.6
K–C(16)–H(16)	73.4	C(72)–O–C(73)	112.0(2)
C(72)–O–K	116.56(17)	C(73)–O–K	130.0(2)
C(26)–C(25)–C(24)	122.83(19)	C(26)–C(25)–H(25)	118.6
C(24)–C(25)–H(25)	118.6	O–K–C	130.92(6)

Symmetry transformations used to generate equivalent atoms: #1 $x + 1, y, z$

Continued on next page

Table S35. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
O – K – C(54)#1	130.53(6)	C – K – C(54)#1	97.09(6)
O – K – C(11)	120.78(6)	C – K – C(11)	27.70(5)
C(54)#1 – K – C(11)	107.36(6)	O – K – C(55)#1	104.84(6)
C – K – C(55)#1	120.02(6)	C(54)#1 – K – C(55)#1	26.36(6)
C(11) – K – C(55)#1	133.71(6)	O – K – C(16)	123.48(6)
C – K – C(16)	50.25(5)	C(54)#1 – K – C(16)	94.35(6)
C(11) – K – C(16)	26.93(5)	C(55)#1 – K – C(16)	118.48(6)
O – K – C(53)#1	131.95(5)	C – K – C(53)#1	95.75(5)
C(54)#1 – K – C(53)#1	25.31(6)	C(11) – K – C(53)#1	94.51(5)
C(55)#1 – K – C(53)#1	44.20(6)	C(16) – K – C(53)#1	74.50(5)
O – K – C(21)	153.36(5)	C – K – C(21)	25.76(5)
C(54)#1 – K – C(21)	71.67(5)	C(11) – K – C(21)	46.69(5)
C(55)#1 – K – C(21)	94.32(5)	C(16) – K – C(21)	58.45(5)
C(53)#1 – K – C(21)	74.63(5)	O – K – C(56)#1	87.55(5)
C – K – C(56)#1	141.03(6)	C(54)#1 – K – C(56)#1	43.97(5)
C(11) – K – C(56)#1	142.99(5)	C(55)#1 – K – C(56)#1	24.13(5)
C(16) – K – C(56)#1	118.55(5)	C(53)#1 – K – C(56)#1	49.28(5)
C(21) – K – C(56)#1	115.61(5)	O – K – C(52)#1	110.69(5)
C – K – C(52)#1	113.52(5)	C(54)#1 – K – C(52)#1	42.12(5)
C(11) – K – C(52)#1	103.53(5)	C(55)#1 – K – C(52)#1	48.49(5)
C(16) – K – C(52)#1	78.21(5)	C(53)#1 – K – C(52)#1	22.90(5)
C(21) – K – C(52)#1	95.84(5)	C(56)#1 – K – C(52)#1	40.34(5)
O – K – C(51)#1	91.34(5)	C – K – C(51)#1	135.48(5)
C(54)#1 – K – C(51)#1	48.97(5)	C(11) – K – C(51)#1	124.67(5)
C(55)#1 – K – C(51)#1	41.04(5)	C(16) – K – C(51)#1	98.23(5)
C(53)#1 – K – C(51)#1	40.62(5)	C(21) – K – C(51)#1	115.14(5)
C(56)#1 – K – C(51)#1	22.56(5)	C(52)#1 – K – C(51)#1	22.36(5)

Symmetry transformations used to generate equivalent atoms: #1 $x + 1, y, z$

5.4 Crystal data for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdCH}_2\text{Ph}\right]^- \text{K}^+$ (6)

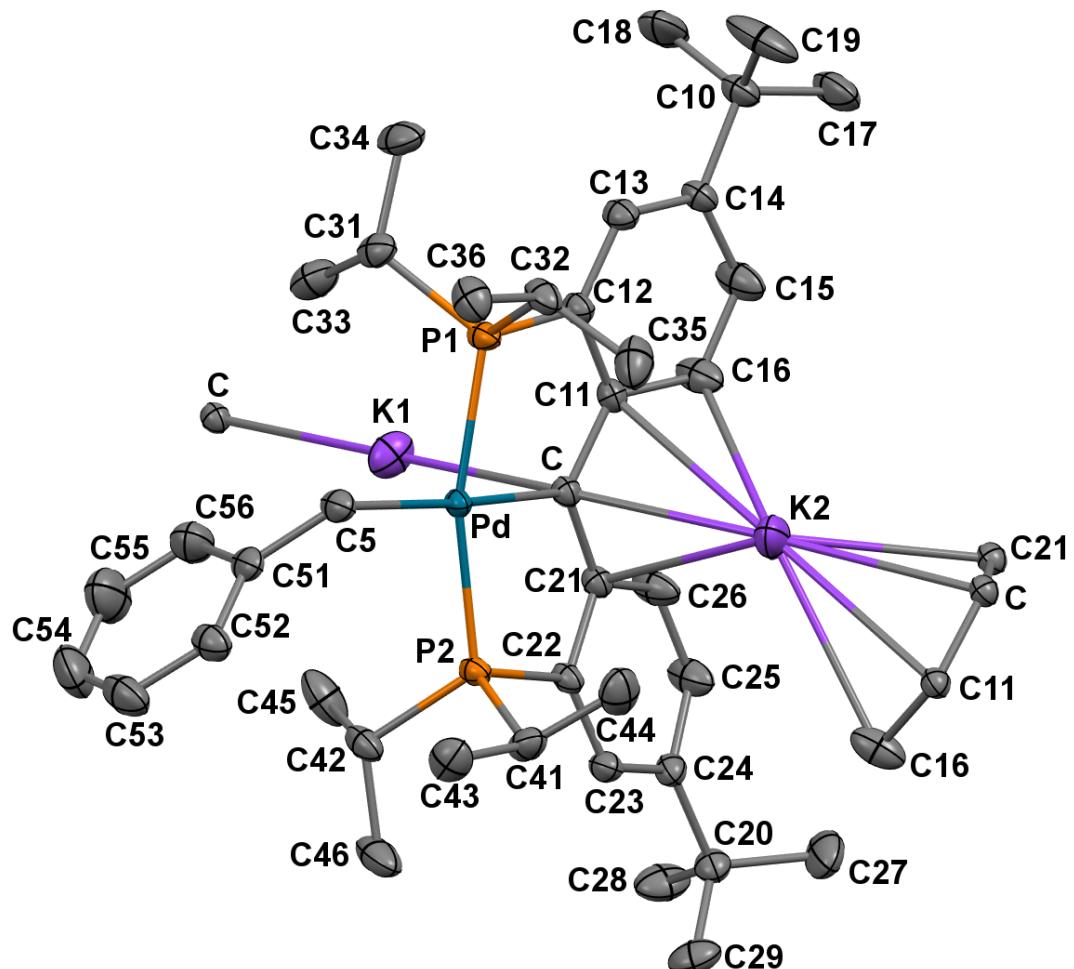


Figure S48. Thermal-ellipsoid representation of $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdCH}_2\text{Ph}\right]^- \text{K}^+$ (6) at 50% probability. Hydrogen atoms were omitted for clarity.

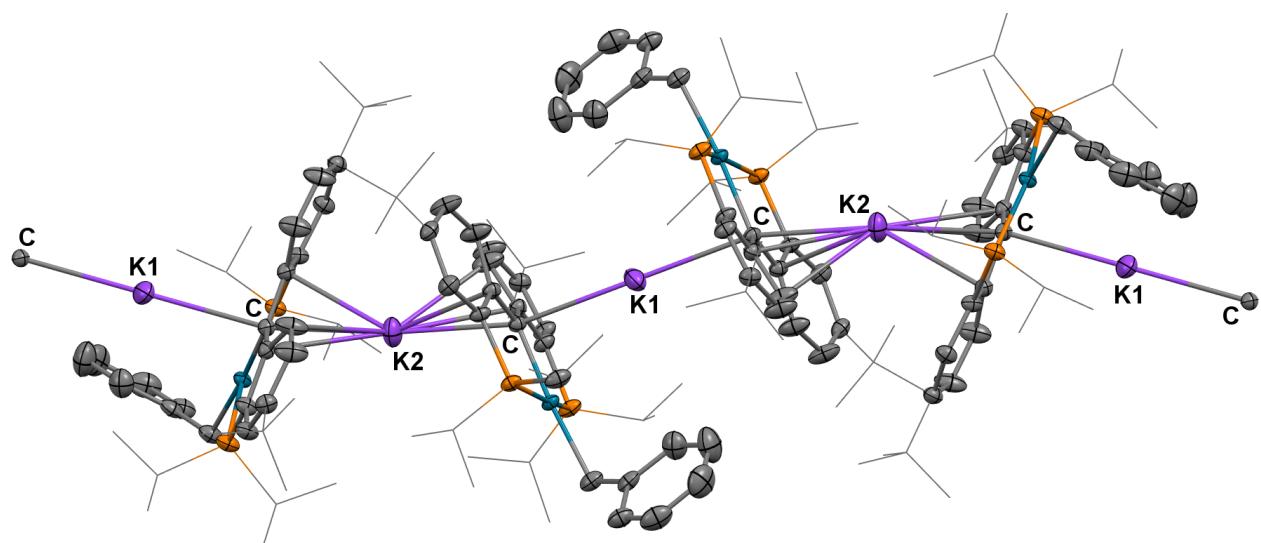


Figure S49. Polymeric structure of $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{\text{tBu}}\text{PdCH}_2\text{Ph}\right]^- \text{K}^+$ (6).

Table S36. Crystal data and structure refinement for $[\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}]^-\text{K}^+$ (**6**).

Identification code:	pc44	
Empirical formula:	C ₄₀ H ₅₉ KP ₂ Pd	
Formula weight:	747.31	
Temperature:	120(2) K	
Wavelength:	0.71073 Å	
Crystal system:	Monoclinic	
Space group:	C2/c	
Unit cell dimensions:	$a = 26.989(3)$ Å	$\alpha = 90^\circ$
	$b = 12.0603(10)$ Å	$\beta = 96.879(4)^\circ$
	$c = 23.568(2)$ Å	$\gamma = 90^\circ$
Volume:	7616.1(12) Å ³	
Z:	8	
Density (calculated):	1.303 g·cm ⁻³	
Absorption coefficient (μ):	0.707 mm ⁻¹	
F(000):	3152	
Crystal size:	0.13 × 0.12 × 0.10 mm ³	
θ range for data collection:	1.74 to 25.00°	
Index ranges:	$-32 \leq h \leq 32, -14 \leq k \leq 14, -28 \leq l \leq 28$	
Reflections collected:	90845	
Independent reflections:	6712 [$R_{\text{int}} = 0.0540$]	
Completeness to $\theta = 25.00^\circ$:	100.0 %	
Absorption correction:	Semi-empirical from equivalents	
Max. and min. transmission:	0.7457 and 0.6757	
Refinement method:	Full-matrix least-squares on F ²	
Data / restraints / parameters:	6712 / 0 / 458	
Goodness-of-fit on F ² :	1.066	
Final R indices [I>2σ(I)]:	$R_1 = 0.0267, wR_2 = 0.0661$	
R indices (all data):	$R_1 = 0.0349, wR_2 = 0.0692$	
Largest diff. peak and hole:	0.917 and $-0.478 \text{ e}^- \cdot \text{\AA}^{-3}$	

Table S37. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for $[\text{PC(sp}^2\text{P)}^{\text{tBu}}\text{PdCH}_2\text{Ph}]^-\text{K}^+$ (**6**). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor

atom	x	y	z	U(eq)
Pd	0.04254(1)	0.78119(1)	0.09393(1)	0.016(1)
P(1)	-0.03530(2)	0.70860(5)	0.07545(3)	0.022(1)
K(2)	0.0000	0.90423(8)	0.2500	0.038(1)
P(2)	0.11202(2)	0.87644(5)	0.13175(2)	0.019(1)
C	0.00435(8)	0.91767(18)	0.12240(9)	0.018(1)
C(10)	-0.21394(8)	0.8679(2)	0.11379(11)	0.026(1)
C(11)	-0.04919(8)	0.91110(18)	0.12137(9)	0.019(1)
C(12)	-0.07573(8)	0.81553(19)	0.09731(10)	0.022(1)
C(13)	-0.12711(9)	0.80309(19)	0.09665(10)	0.024(1)
C(15)	-0.13161(9)	0.9687(2)	0.14729(13)	0.038(1)
C(14)	-0.15714(8)	0.88093(19)	0.11867(10)	0.022(1)
C(16)	-0.08056(10)	0.9837(2)	0.14885(13)	0.036(1)
C(17)	-0.23774(10)	0.9537(3)	0.14899(14)	0.047(1)
C(5)	0.07820(9)	0.6421(2)	0.05736(11)	0.025(1)
C(18)	-0.23563(10)	0.8813(3)	0.05164(13)	0.056(1)
C(19)	-0.22805(11)	0.7531(3)	0.13385(19)	0.060(1)
C(20)	0.12494(9)	1.27972(19)	0.23522(10)	0.025(1)
C(21)	0.03330(8)	1.01061(18)	0.14598(9)	0.019(1)
C(22)	0.08662(8)	1.00268(18)	0.15874(9)	0.019(1)
C(24)	0.09491(9)	1.18521(19)	0.20441(10)	0.022(1)
C(23)	0.11499(8)	1.08563(19)	0.18808(10)	0.023(1)
C(27)	0.09845(12)	1.3215(3)	0.28479(13)	0.049(1)
C(26)	0.01486(9)	1.1181(2)	0.15825(12)	0.035(1)
C(25)	0.04403(9)	1.1987(2)	0.18638(13)	0.036(1)
C(28)	0.12824(11)	1.3756(2)	0.19344(12)	0.040(1)
C(29)	0.17765(10)	1.2444(2)	0.25852(12)	0.038(1)
C(31)	-0.05742(9)	0.6628(2)	0.00203(10)	0.027(1)
C(32)	-0.04818(9)	0.5887(2)	0.12049(11)	0.028(1)
C(33)	-0.04488(12)	0.7479(3)	-0.04066(13)	0.038(1)
C(34)	-0.11257(9)	0.6305(2)	-0.01002(13)	0.039(1)
C(36)	-0.02327(10)	0.4819(2)	0.10375(12)	0.033(1)
C(35)	-0.03142(12)	0.6165(2)	0.18297(12)	0.043(1)
C(43)	0.18009(10)	0.7135(2)	0.17619(12)	0.035(1)
C(44)	0.11497(12)	0.7734(3)	0.23639(12)	0.038(1)
C(45)	0.13226(13)	0.9873(3)	0.03519(16)	0.045(1)
C(46)	0.20533(10)	0.9760(3)	0.11179(14)	0.047(1)
C(51)	0.11661(9)	0.6687(2)	0.01925(10)	0.026(1)
C(52)	0.16562(9)	0.6305(2)	0.02957(11)	0.029(1)
C(53)	0.20072(10)	0.6551(2)	-0.00706(13)	0.042(1)

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Table S37. – continued from previous page

atom	x	y	x	U(eq)
C(54)	0.18797(12)	0.7191(3)	-0.05510(14)	0.052(1)
C(55)	0.14040(13)	0.7586(3)	-0.06582(13)	0.056(1)
C(56)	0.10535(10)	0.7327(3)	-0.02987(12)	0.040(1)
C(42)	0.15828(9)	0.9180(2)	0.08351(11)	0.027(1)
C(41)	0.14946(9)	0.8110(2)	0.19381(10)	0.027(1)
K(1)	0.0000	1.0000	0.0000	0.036(1)
H(13)	-0.1423	0.7377	0.0802	0.029
H(15)	-0.1500	1.0207	0.1666	0.045
H(16)	-0.0649(10)	1.044(2)	0.1715(12)	0.041(8)
H(17A)	-0.2299	1.0282	0.1360	0.071
H(17B)	-0.2247	0.9453	0.1894	0.071
H(17C)	-0.2740	0.9432	0.1444	0.071
H(5B)	0.0909(9)	0.593(2)	0.0890(11)	0.029(7)
H(5A)	0.0512(10)	0.604(2)	0.0338(11)	0.031(7)
H(18A)	-0.2282	0.9559	0.0385	0.085
H(18B)	-0.2719	0.8708	0.0481	0.085
H(18C)	-0.2208	0.8260	0.0283	0.085
H(19A)	-0.2145	0.6964	0.1102	0.090
H(19B)	-0.2645	0.7465	0.1304	0.090
H(19C)	-0.2142	0.7428	0.1739	0.090
H(23)	0.1498	1.0736	0.1974	0.027
H(27A)	0.0650	1.3476	0.2702	0.074
H(27B)	0.1176	1.3826	0.3040	0.074
H(27C)	0.0958	1.2610	0.3120	0.074
H(26)	-0.0192	1.1344	0.1463	0.041
H(25)	0.0289	1.2673	0.1940	0.043
H(28A)	0.1451	1.3504	0.1612	0.060
H(28B)	0.1472	1.4367	0.2130	0.060
H(28C)	0.0945	1.4012	0.1794	0.060
H(29A)	0.1957	1.2214	0.2268	0.057
H(29B)	0.1760	1.1821	0.2850	0.057
H(29C)	0.1951	1.3067	0.2787	0.057
H(31)	-0.0378	0.5948	-0.0050	0.033
H(32)	-0.0851	0.5762	0.1161	0.034
H(33C)	-0.0545(12)	0.725(3)	-0.0776(15)	0.051(9)
H(33B)	-0.0635(11)	0.816(3)	-0.0355(12)	0.040(8)
H(33A)	-0.0112(13)	0.765(2)	-0.0361(13)	0.045(9)
H(34A)	-0.1186	0.5943	-0.0474	0.059
H(34B)	-0.1333	0.6972	-0.0100	0.059
H(34C)	-0.1211	0.5794	0.0197	0.059
H(36A)	-0.0361	0.4621	0.0644	0.050
H(36B)	-0.0306	0.4220	0.1296	0.050

Continued on next page

Table S37. – continued from previous page

atom	x	y	x	U(eq)
H(36C)	0.0129	0.4931	0.1065	0.050
H(35A)	0.0050	0.6247	0.1888	0.065
H(35B)	-0.0414	0.5566	0.2073	0.065
H(35C)	-0.0471	0.6860	0.1929	0.065
H(43A)	0.1576	0.6558	0.1588	0.052
H(43B)	0.2004	0.6834	0.2099	0.052
H(43C)	0.2020	0.7387	0.1484	0.052
H(44C)	0.0881(12)	0.724(2)	0.2165(13)	0.045(9)
H(44B)	0.1332(12)	0.732(3)	0.2656(14)	0.046(9)
H(44A)	0.0999(11)	0.836(3)	0.2517(13)	0.046(9)
H(45A)	0.1230(13)	1.055(3)	0.0480(14)	0.062(11)
H(45B)	0.1031(13)	0.952(3)	0.0171(13)	0.053(10)
H(45C)	0.1525(13)	1.003(3)	0.0087(15)	0.058(10)
H(46A)	0.2295	0.9829	0.0841	0.070
H(46B)	0.2200	0.9323	0.1446	0.070
H(46C)	0.1966	1.0499	0.1247	0.070
H(52)	0.1752	0.5865	0.0625	0.035
H(53)	0.2338	0.6276	0.0010	0.050
H(54)	0.2119	0.7355	-0.0803	0.062
H(55)	0.1314	0.8042	-0.0983	0.067
H(56)	0.0723	0.7596	-0.0389	0.047
H(42)	0.1695	0.8483	0.0659	0.033
H(41)	0.1729	0.8678	0.2127	0.033

Table S38. Anisotropic displacement parameters (\AA^2) for $[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}]^-\text{K}^+$ (**6**). The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}\mathbf{U}_{11} + \dots + 2hka^{*}b^{*}\mathbf{U}_{12}]$.

atom	\mathbf{U}_{11}	\mathbf{U}_{22}	\mathbf{U}_{33}	\mathbf{U}_{23}	\mathbf{U}_{13}	\mathbf{U}_{12}
Pd	0.0137(1)	0.0157(1)	0.0190(1)	-0.0036(1)	0.0036(1)	0.0000(1)
P(1)	0.0155(3)	0.0194(3)	0.0310(3)	-0.0090(3)	0.0045(2)	-0.0016(2)
K(2)	0.0336(4)	0.0565(6)	0.0238(4)	0.000	0.0061(3)	0.000
P(2)	0.0139(3)	0.0179(3)	0.0254(3)	-0.0047(2)	0.0037(2)	0.0002(2)
C	0.0173(11)	0.0174(11)	0.0189(11)	-0.0006(9)	0.0044(9)	0.0006(9)
C(10)	0.0169(12)	0.0277(13)	0.0357(14)	-0.0009(11)	0.0065(10)	-0.0007(10)
C(11)	0.0185(11)	0.0188(11)	0.0191(11)	0.0005(9)	0.0050(9)	0.0014(9)
C(12)	0.0174(11)	0.0211(12)	0.0272(12)	-0.0060(10)	0.0050(9)	0.0002(9)
C(13)	0.0204(12)	0.0210(12)	0.0312(13)	-0.0064(10)	0.0044(10)	-0.0029(9)
C(15)	0.0236(14)	0.0258(14)	0.0655(19)	-0.0106(13)	0.0131(13)	0.0022(11)
C(14)	0.0157(11)	0.0228(12)	0.0282(12)	0.0008(10)	0.0052(9)	0.0009(9)
C(16)	0.0237(13)	0.0232(13)	0.0620(19)	-0.0139(13)	0.0098(12)	-0.0023(11)
C(17)	0.0216(14)	0.0537(19)	0.070(2)	-0.0204(17)	0.0172(14)	-0.0020(13)
C(5)	0.0210(13)	0.0230(13)	0.0319(14)	-0.0093(11)	0.0049(10)	-0.0012(10)
C(18)	0.0182(14)	0.105(3)	0.0455(18)	0.0010(18)	0.0001(12)	-0.0047(16)
C(19)	0.0230(15)	0.0386(18)	0.123(3)	0.0149(19)	0.0243(18)	-0.0034(13)
C(20)	0.0267(13)	0.0201(12)	0.0286(13)	-0.0046(10)	0.0059(10)	-0.0054(10)
C(21)	0.0183(11)	0.0182(11)	0.0208(11)	-0.0003(9)	0.0051(9)	0.0009(9)
C(22)	0.0158(11)	0.0162(11)	0.0262(12)	-0.0029(9)	0.0065(9)	0.0003(9)
C(24)	0.0233(12)	0.0202(12)	0.0234(12)	-0.0046(10)	0.0070(9)	-0.0039(10)
C(23)	0.0175(11)	0.0240(13)	0.0267(12)	-0.0045(10)	0.0036(9)	-0.0018(10)
C(27)	0.056(2)	0.0487(18)	0.0465(18)	-0.0275(15)	0.0192(15)	-0.0160(16)
C(26)	0.0187(13)	0.0254(14)	0.0600(18)	-0.0109(13)	0.0067(12)	0.0021(11)
C(25)	0.0235(13)	0.0235(13)	0.0607(18)	-0.0157(12)	0.0105(12)	0.0003(10)
C(28)	0.0414(16)	0.0306(15)	0.0465(17)	0.0058(13)	-0.0042(13)	-0.0160(13)
C(29)	0.0374(16)	0.0282(14)	0.0444(16)	-0.0070(12)	-0.0081(13)	-0.0064(12)
C(31)	0.0238(13)	0.0236(13)	0.0339(14)	-0.0118(11)	-0.0003(10)	0.0025(10)
C(32)	0.0238(13)	0.0228(13)	0.0397(14)	-0.0065(11)	0.0126(11)	-0.0054(10)
C(33)	0.0349(17)	0.0372(17)	0.0378(17)	-0.0046(13)	-0.0090(13)	0.0026(13)
C(34)	0.0265(14)	0.0388(16)	0.0508(17)	-0.0228(13)	-0.0029(12)	-0.0031(12)
C(36)	0.0369(15)	0.0241(13)	0.0404(15)	-0.0004(11)	0.0125(12)	0.0012(11)
C(35)	0.061(2)	0.0353(16)	0.0359(16)	-0.0049(13)	0.0200(14)	-0.0103(14)
C(43)	0.0328(15)	0.0339(15)	0.0356(14)	-0.0017(12)	-0.0016(11)	0.0129(12)
C(44)	0.0385(17)	0.0485(18)	0.0257(14)	0.0031(14)	0.0007(12)	0.0101(15)
C(45)	0.0422(19)	0.0424(19)	0.057(2)	0.0168(16)	0.0276(17)	0.0066(15)
C(46)	0.0322(16)	0.0421(17)	0.070(2)	-0.0212(15)	0.0255(14)	-0.0156(13)
C(51)	0.0244(13)	0.0240(13)	0.0291(13)	-0.0123(11)	0.0057(10)	-0.0018(10)
C(52)	0.0235(13)	0.0259(13)	0.0400(15)	-0.0074(11)	0.0079(11)	0.0007(10)
C(53)	0.0264(14)	0.0397(17)	0.063(2)	-0.0113(15)	0.0160(13)	0.0000(12)
C(54)	0.0479(19)	0.064(2)	0.0481(19)	-0.0024(17)	0.0271(15)	-0.0057(16)

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Table S38. – continued from previous page

atom	U₁₁	U₂₂	U₃₃	U₂₃	U₁₃	U₁₂
C(55)	0.053(2)	0.079(2)	0.0356(17)	0.0111(16)	0.0106(15)	-0.0027(18)
C(56)	0.0292(15)	0.0523(18)	0.0366(15)	0.0005(14)	0.0026(12)	0.0014(13)
C(42)	0.0207(12)	0.0222(13)	0.0418(15)	-0.0058(11)	0.0140(11)	-0.0011(10)
C(41)	0.0217(12)	0.0299(13)	0.0289(13)	-0.0076(11)	-0.0037(10)	0.0035(10)
K(1)	0.0468(5)	0.0304(4)	0.0299(4)	0.0062(3)	-0.0023(4)	-0.0007(4)

Table S39. Distances [Å] for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}\right]^-\text{K}^+$ (**6**).

atom–atom	distance	atom–atom	distance
Pd–C	2.095(2)	Pd–C(5)	2.163(2)
Pd–P(1)	2.2695(6)	Pd–P(2)	2.2868(6)
Pd–K(1)	3.5460(3)	P(1)–C(12)	1.804(2)
P(1)–C(31)	1.846(2)	P(1)–C(32)	1.852(3)
K(2)–C(21)#1	2.998(2)	K(2)–C(21)	2.998(2)
K(2)–C#1	3.028(2)	K(2)–C	3.028(2)
K(2)–C(11)#1	3.162(2)	K(2)–C(11)	3.162(2)
K(2)–C(16)#1	3.176(3)	K(2)–C(16)	3.176(3)
K(2)–C(26)#1	3.420(3)	K(2)–C(26)	3.420(3)
K(2)–C(44)#1	3.530(3)	K(2)–C(44)	3.530(3)
K(2)–H(16)	2.93(3)	K(2)–H(44A)	2.82(3)
P(2)–C(22)	1.816(2)	P(2)–C(41)	1.852(2)
P(2)–C(42)	1.855(2)	C–C(21)	1.439(3)
C–C(11)	1.444(3)	C–K(1)	3.040(2)
C(10)–C(17)	1.517(4)	C(10)–C(18)	1.520(4)
C(10)–C(19)	1.525(4)	C(10)–C(14)	1.532(3)
C(11)–C(16)	1.426(3)	C(11)–C(12)	1.437(3)
C(11)–K(1)	3.465(2)	C(12)–C(13)	1.393(3)
C(13)–C(14)	1.381(3)	C(13)–H(13)	0.9500
C(15)–C(16)	1.386(4)	C(15)–C(14)	1.393(3)
C(15)–H(15)	0.9500	C(16)–H(16)	0.97(3)
C(17)–H(17A)	0.9800	C(17)–H(17B)	0.9800
C(17)–H(17C)	0.9800	C(5)–C(51)	1.486(3)
C(5)–H(5B)	0.98(3)	C(5)–H(5A)	0.98(3)
C(18)–H(18A)	0.9800	C(18)–H(18B)	0.9800
C(18)–H(18C)	0.9800	C(19)–H(19A)	0.9800
C(19)–H(19B)	0.9800	C(19)–H(19C)	0.9800
C(20)–C(29)	1.523(4)	C(20)–C(27)	1.526(4)
C(20)–C(28)	1.528(4)	C(20)–C(24)	1.531(3)
C(21)–C(26)	1.430(3)	C(21)–C(22)	1.438(3)
C(21)–K(1)	3.453(2)	C(22)–C(23)	1.392(3)
C(24)–C(23)	1.391(3)	C(24)–C(25)	1.397(3)
C(23)–H(23)	0.9500	C(27)–H(27A)	0.9800
C(27)–H(27B)	0.9800	C(27)–H(27C)	0.9800
C(26)–C(25)	1.371(4)	C(26)–H(26)	0.9500
C(25)–H(25)	0.9500	C(28)–H(28A)	0.9800
C(28)–H(28B)	0.9800	C(28)–H(28C)	0.9800
C(29)–H(29A)	0.9800	C(29)–H(29B)	0.9800
C(29)–H(29C)	0.9800	C(31)–C(33)	1.504(4)
C(31)–C(34)	1.532(3)	C(31)–H(31)	1.0000

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

Continued on next page

Table S39. – continued from previous page

atom – atom	distance	atom – atom	distance
C(32)–C(35)	1.525(4)	C(32)–C(36)	1.526(3)
C(32)–H(32)	1.0000	C(33)–K(1)	3.369(3)
C(33)–H(33C)	0.92(3)	C(33)–H(33B)	0.98(3)
C(33)–H(33A)	0.93(3)	C(34)–H(34A)	0.9800
C(34)–H(34B)	0.9800	C(34)–H(34C)	0.9800
C(36)–H(36A)	0.9800	C(36)–H(36B)	0.9800
C(36)–H(36C)	0.9800	C(35)–H(35A)	0.9800
C(35)–H(35B)	0.9800	C(35)–H(35C)	0.9800
C(43)–C(41)	1.523(3)	C(43)–H(43A)	0.9800
C(43)–H(43B)	0.9800	C(43)–H(43C)	0.9800
C(44)–C(41)	1.518(4)	C(44)–H(44C)	1.01(3)
C(44)–H(44B)	0.94(3)	C(44)–H(44A)	0.95(3)
C(45)–C(42)	1.516(4)	C(45)–H(45A)	0.91(4)
C(45)–H(45B)	0.95(3)	C(45)–H(45C)	0.90(4)
C(46)–C(42)	1.531(4)	C(46)–H(46A)	0.9800
C(46)–H(46B)	0.9800	C(46)–H(46C)	0.9800
C(51)–C(52)	1.394(3)	C(51)–C(56)	1.395(4)
C(52)–C(53)	1.388(4)	C(52)–H(52)	0.9500
C(53)–C(54)	1.380(4)	C(53)–H(53)	0.9500
C(54)–C(55)	1.364(5)	C(54)–H(54)	0.9500
C(55)–C(56)	1.379(4)	C(55)–H(55)	0.9500
C(56)–H(56)	0.9500	C(42)–H(42)	1.0000
C(41)–H(41)	1.0000	K(1)–C#2	3.040(2)
K(1)–C(33)#2	3.369(3)	K(1)–C(21)#2	3.453(2)
K(1)–C(11)#2	3.465(2)	K(1)–Pd#2	3.5460(3)
K(1)–H(33B)	2.87(3)	K(1)–H(33A)	2.96(3)
K(1)–H(45B)	2.82(3)		

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

Table S40. Angles [°] for $\left[\{\text{PC}(\text{sp}^2)\text{P}\}^{t\text{Bu}}\text{PdCH}_2\text{Ph}\right]^-\text{K}^+$ (**6**).

atom–atom–atom	angle	atom–atom–atom	angle
C–Pd–C(5)	175.02(9)	C–Pd–P(1)	83.38(6)
C(5)–Pd–P(1)	94.03(7)	C–Pd–P(2)	83.83(6)
C(5)–Pd–P(2)	99.30(7)	P(1)–Pd–P(2)	164.91(2)
C–Pd–K(1)	58.68(6)	C(5)–Pd–K(1)	117.03(7)
P(1)–Pd–K(1)	86.801(17)	P(2)–Pd–K(1)	93.357(17)
C(12)–P(1)–C(31)	110.05(11)	C(12)–P(1)–C(32)	103.49(11)
C(31)–P(1)–C(32)	103.99(11)	C(12)–P(1)–Pd	104.31(8)
C(31)–P(1)–Pd	118.90(8)	C(32)–P(1)–Pd	115.11(8)
C(21)#1–K(2)–C(21)	129.34(9)	C(21)#1–K(2)–C#1	27.63(6)
C(21)–K(2)–C#1	147.16(7)	C(21)#1–K(2)–C	147.16(7)
C(21)–K(2)–C	27.63(6)	C#1–K(2)–C	173.86(9)
C(21)#1–K(2)–C(11)#1	48.50(6)	C(21)–K(2)–C(11)#1	129.81(6)
C#1–K(2)–C(11)#1	26.88(6)	C–K(2)–C(11)#1	152.77(6)
C(21)#1–K(2)–C(11)	129.81(6)	C(21)–K(2)–C(11)	48.50(6)
C#1–K(2)–C(11)	152.77(6)	C–K(2)–C(11)	26.88(6)
C(11)#1–K(2)–C(11)	177.00(9)	C(21)#1–K(2)–C(16)#1	60.16(6)
C(21)–K(2)–C(16)#1	103.84(7)	C#1–K(2)–C(16)#1	48.92(6)
C–K(2)–C(16)#1	128.67(7)	C(11)#1–K(2)–C(16)#1	26.01(6)
C(11)–K(2)–C(16)#1	151.99(7)	C(21)#1–K(2)–C(16)	103.84(7)
C(21)–K(2)–C(16)	60.16(6)	C#1–K(2)–C(16)	128.67(7)
C–K(2)–C(16)	48.92(6)	C(11)#1–K(2)–C(16)	151.99(7)
C(11)–K(2)–C(16)	26.01(6)	C(16)#1–K(2)–C(16)	144.87(10)
C(21)#1–K(2)–C(26)#1	24.64(6)	C(21)–K(2)–C(26)#1	105.28(7)
C#1–K(2)–C(26)#1	46.39(6)	C–K(2)–C(26)#1	127.52(7)
C(11)#1–K(2)–C(26)#1	56.05(6)	C(11)–K(2)–C(26)#1	121.26(6)
C(16)#1–K(2)–C(26)#1	54.51(7)	C(16)–K(2)–C(26)#1	97.20(7)
C(21)#1–K(2)–C(26)	105.28(7)	C(21)–K(2)–C(26)	24.64(6)
C#1–K(2)–C(26)	127.52(7)	C–K(2)–C(26)	46.39(6)
C(11)#1–K(2)–C(26)	121.26(6)	C(11)–K(2)–C(26)	56.05(6)
C(16)#1–K(2)–C(26)	97.20(7)	C(16)–K(2)–C(26)	54.51(7)
C(26)#1–K(2)–C(26)	82.10(9)	C(21)#1–K(2)–C(44)#1	76.36(6)
C(21)–K(2)–C(44)#1	128.20(6)	C#1–K(2)–C(44)#1	78.15(6)
C–K(2)–C(44)#1	104.67(6)	C(11)#1–K(2)–C(44)#1	101.38(6)
C(11)–K(2)–C(44)#1	79.98(6)	C(16)#1–K(2)–C(44)#1	126.65(7)
C(16)–K(2)–C(44)#1	70.91(7)	C(26)#1–K(2)–C(44)#1	96.24(7)
C(26)–K(2)–C(44)#1	124.44(7)	C(21)#1–K(2)–C(44)	128.20(6)
C(21)–K(2)–C(44)	76.36(6)	C#1–K(2)–C(44)	104.67(6)
C–K(2)–C(44)	78.15(6)	C(11)#1–K(2)–C(44)	79.98(6)
C(11)–K(2)–C(44)	101.38(6)	C(16)#1–K(2)–C(44)	70.91(7)
C(16)–K(2)–C(44)	126.65(7)	C(26)#1–K(2)–C(44)	124.44(7)

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

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Table S40. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(26)–K(2)–C(44)	96.24(7)	C(44)#1–K(2)–C(44)	126.90(12)
C(21)#1–K(2)–H(16)	93.1(6)	C(21)–K(2)–H(16)	56.8(5)
C#1–K(2)–H(16)	120.3(6)	C–K(2)–H(16)	55.5(6)
C(11)#1–K(2)–H(16)	137.8(6)	C(11)–K(2)–H(16)	39.5(6)
C(16)#1–K(2)–H(16)	127.2(6)	C(16)–K(2)–H(16)	17.7(6)
C(26)#1–K(2)–H(16)	81.8(6)	C(26)–K(2)–H(16)	43.3(6)
C(44)#1–K(2)–H(16)	81.3(6)	C(44)–K(2)–H(16)	131.8(6)
C(21)#1–K(2)–H(44A)	119.6(6)	C(21)–K(2)–H(44A)	75.9(6)
C#1–K(2)–H(44A)	98.7(6)	C–K(2)–H(44A)	83.1(6)
C(11)#1–K(2)–H(44A)	72.7(6)	C(11)–K(2)–H(44A)	108.3(6)
C(16)#1–K(2)–H(44A)	60.6(6)	C(16)–K(2)–H(44A)	131.9(6)
C(26)#1–K(2)–H(44A)	113.3(6)	C(26)–K(2)–H(44A)	92.6(6)
C(44)#1–K(2)–H(44A)	136.0(6)	C(44)–K(2)–H(44A)	11.4(6)
H(16)–K(2)–H(44A)	132.7(8)	C(22)–P(2)–C(41)	105.89(11)
C(22)–P(2)–C(42)	107.37(11)	C(41)–P(2)–C(42)	104.79(11)
C(22)–P(2)–Pd	103.38(7)	C(41)–P(2)–Pd	115.99(8)
C(42)–P(2)–Pd	118.51(8)	C(21)–C–C(11)	123.02(19)
C(21)–C–Pd	118.06(15)	C(11)–C–Pd	118.76(15)
C(21)–C–K(2)	75.05(12)	C(11)–C–K(2)	81.73(12)
Pd–C–K(2)	110.80(8)	C(21)–C–K(1)	93.99(13)
C(11)–C–K(1)	94.42(13)	Pd–C–K(1)	85.25(7)
K(2)–C–K(1)	163.40(8)	C(17)–C(10)–C(18)	108.3(2)
C(17)–C(10)–C(19)	108.2(2)	C(18)–C(10)–C(19)	108.3(3)
C(17)–C(10)–C(14)	112.0(2)	C(18)–C(10)–C(14)	109.1(2)
C(19)–C(10)–C(14)	110.8(2)	C(16)–C(11)–C(12)	112.1(2)
C(16)–C(11)–C	127.3(2)	C(12)–C(11)–C	120.06(19)
C(16)–C(11)–K(2)	77.56(15)	C(12)–C(11)–K(2)	119.31(15)
C–C(11)–K(2)	71.39(12)	C(16)–C(11)–K(1)	119.46(16)
C(12)–C(11)–K(1)	98.31(14)	C–C(11)–K(1)	61.02(11)
K(2)–C(11)–K(1)	129.89(7)	C(13)–C(12)–C(11)	122.6(2)
C(13)–C(12)–P(1)	123.72(18)	C(11)–C(12)–P(1)	113.33(16)
C(14)–C(13)–C(12)	123.4(2)	C(14)–C(13)–H(13)	118.3
C(12)–C(13)–H(13)	118.3	C(16)–C(15)–C(14)	123.0(2)
C(16)–C(15)–H(15)	118.5	C(14)–C(15)–H(15)	118.5
C(13)–C(14)–C(15)	114.8(2)	C(13)–C(14)–C(10)	121.9(2)
C(15)–C(14)–C(10)	123.2(2)	C(15)–C(16)–C(11)	123.4(2)
C(15)–C(16)–K(2)	124.7(2)	C(11)–C(16)–K(2)	76.43(14)
C(15)–C(16)–H(16)	118.5(17)	C(11)–C(16)–H(16)	118.0(17)
K(2)–C(16)–H(16)	66.5(17)	C(10)–C(17)–H(17A)	109.5
C(10)–C(17)–H(17B)	109.5	H(17A)–C(17)–H(17B)	109.5

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

Continued on next page

Table S40. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(10)–C(17)–H(17C)	109.5	H(17A)–C(17)–H(17C)	109.5
H(17B)–C(17)–H(17C)	109.5	C(51)–C(5)–Pd	116.71(17)
C(51)–C(5)–H(5B)	112.6(15)	Pd–C(5)–H(5B)	107.3(15)
C(51)–C(5)–H(5A)	106.6(15)	Pd–C(5)–H(5A)	104.8(15)
H(5B)–C(5)–H(5A)	108(2)	C(10)–C(18)–H(18A)	109.5
C(10)–C(18)–H(18B)	109.5	H(18A)–C(18)–H(18B)	109.5
C(10)–C(18)–H(18C)	109.5	H(18A)–C(18)–H(18C)	109.5
H(18B)–C(18)–H(18C)	109.5	C(10)–C(19)–H(19A)	109.5
C(10)–C(19)–H(19B)	109.5	H(19A)–C(19)–H(19B)	109.5
C(10)–C(19)–H(19C)	109.5	H(19A)–C(19)–H(19C)	109.5
H(19B)–C(19)–H(19C)	109.5	C(29)–C(20)–C(27)	108.6(2)
C(29)–C(20)–C(28)	108.7(2)	C(27)–C(20)–C(28)	108.2(2)
C(29)–C(20)–C(24)	112.4(2)	C(27)–C(20)–C(24)	109.8(2)
C(28)–C(20)–C(24)	109.0(2)	C(26)–C(21)–C(22)	112.4(2)
C(26)–C(21)–C	126.7(2)	C(22)–C(21)–C	120.83(19)
C(26)–C(21)–K(2)	94.42(15)	C(22)–C(21)–K(2)	101.02(14)
C–C(21)–K(2)	77.32(12)	C(26)–C(21)–K(1)	100.35(15)
C(22)–C(21)–K(1)	109.91(13)	C–C(21)–K(1)	61.44(11)
K(2)–C(21)–K(1)	136.98(7)	C(23)–C(22)–C(21)	122.2(2)
C(23)–C(22)–P(2)	124.63(17)	C(21)–C(22)–P(2)	113.13(16)
C(23)–C(24)–C(25)	114.7(2)	C(23)–C(24)–C(20)	124.9(2)
C(25)–C(24)–C(20)	120.2(2)	C(24)–C(23)–C(22)	123.3(2)
C(24)–C(23)–H(23)	118.4	C(22)–C(23)–H(23)	118.4
C(20)–C(27)–H(27A)	109.5	C(20)–C(27)–H(27B)	109.5
H(27A)–C(27)–H(27B)	109.5	C(20)–C(27)–H(27C)	109.5
H(27A)–C(27)–H(27C)	109.5	H(27B)–C(27)–H(27C)	109.5
C(25)–C(26)–C(21)	123.2(2)	C(25)–C(26)–K(2)	109.5(2)
C(21)–C(26)–K(2)	60.94(13)	C(25)–C(26)–H(26)	118.4
C(21)–C(26)–H(26)	118.4	K(2)–C(26)–H(26)	99.2
C(26)–C(25)–C(24)	123.4(2)	C(26)–C(25)–H(25)	118.3
C(24)–C(25)–H(25)	118.3	C(20)–C(28)–H(28A)	109.5
C(20)–C(28)–H(28B)	109.5	H(28A)–C(28)–H(28B)	109.5
C(20)–C(28)–H(28C)	109.5	H(28A)–C(28)–H(28C)	109.5
H(28B)–C(28)–H(28C)	109.5	C(20)–C(29)–H(29A)	109.5
C(20)–C(29)–H(29B)	109.5	H(29A)–C(29)–H(29B)	109.5
C(20)–C(29)–H(29C)	109.5	H(29A)–C(29)–H(29C)	109.5
H(29B)–C(29)–H(29C)	109.5	C(33)–C(31)–C(34)	109.9(2)
C(33)–C(31)–P(1)	110.45(18)	C(34)–C(31)–P(1)	116.59(18)
C(33)–C(31)–H(31)	106.4	C(34)–C(31)–H(31)	106.4
P(1)–C(31)–H(31)	106.4	C(35)–C(32)–C(36)	110.3(2)

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

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Table S40. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(35)–C(32)–P(1)	109.10(17)	C(36)–C(32)–P(1)	113.08(17)
C(35)–C(32)–H(32)	108.1	C(36)–C(32)–H(32)	108.1
P(1)–C(32)–H(32)	108.1	C(31)–C(33)–K(1)	121.85(18)
C(31)–C(33)–H(33C)	111(2)	K(1)–C(33)–H(33C)	126(2)
C(31)–C(33)–H(33B)	109.2(17)	K(1)–C(33)–H(33B)	51.6(17)
H(33C)–C(33)–H(33B)	106(3)	C(31)–C(33)–H(33A)	111.8(19)
K(1)–C(33)–H(33A)	56.5(19)	H(33C)–C(33)–H(33A)	110(3)
H(33B)–C(33)–H(33A)	108(3)	C(31)–C(34)–H(34A)	109.5
C(31)–C(34)–H(34B)	109.5	H(34A)–C(34)–H(34B)	109.5
C(31)–C(34)–H(34C)	109.5	H(34A)–C(34)–H(34C)	109.5
H(34B)–C(34)–H(34C)	109.5	C(32)–C(36)–H(36A)	109.5
C(32)–C(36)–H(36B)	109.5	H(36A)–C(36)–H(36B)	109.5
C(32)–C(36)–H(36C)	109.5	H(36A)–C(36)–H(36C)	109.5
H(36B)–C(36)–H(36C)	109.5	C(32)–C(35)–H(35A)	109.5
C(32)–C(35)–H(35B)	109.5	H(35A)–C(35)–H(35B)	109.5
C(32)–C(35)–H(35C)	109.5	H(35A)–C(35)–H(35C)	109.5
H(35B)–C(35)–H(35C)	109.5	C(41)–C(43)–H(43A)	109.5
C(41)–C(43)–H(43B)	109.5	H(43A)–C(43)–H(43B)	109.5
C(41)–C(43)–H(43C)	109.5	H(43A)–C(43)–H(43C)	109.5
H(43B)–C(43)–H(43C)	109.5	C(41)–C(44)–K(2)	122.83(18)
C(41)–C(44)–H(44C)	109.4(17)	K(2)–C(44)–H(44C)	73.6(17)
C(41)–C(44)–H(44B)	109.6(19)	K(2)–C(44)–H(44B)	124.0(19)
H(44C)–C(44)–H(44B)	108(2)	C(41)–C(44)–H(44A)	109.8(19)
K(2)–C(44)–H(44A)	35.8(18)	H(44C)–C(44)–H(44A)	109(2)
H(44B)–C(44)–H(44A)	111(3)	C(42)–C(45)–H(45A)	111(2)
C(42)–C(45)–H(45B)	112(2)	H(45A)–C(45)–H(45B)	108(3)
C(42)–C(45)–H(45C)	112(2)	H(45A)–C(45)–H(45C)	104(3)
H(45B)–C(45)–H(45C)	109(3)	C(42)–C(46)–H(46A)	109.5
C(42)–C(46)–H(46B)	109.5	H(46A)–C(46)–H(46B)	109.5
C(42)–C(46)–H(46C)	109.5	H(46A)–C(46)–H(46C)	109.5
H(46B)–C(46)–H(46C)	109.5	C(52)–C(51)–C(56)	115.8(2)
C(52)–C(51)–C(5)	122.6(2)	C(56)–C(51)–C(5)	121.5(2)
C(53)–C(52)–C(51)	121.7(3)	C(53)–C(52)–H(52)	119.2
C(51)–C(52)–H(52)	119.2	C(54)–C(53)–C(52)	120.5(3)
C(54)–C(53)–H(53)	119.7	C(52)–C(53)–H(53)	119.7
C(55)–C(54)–C(53)	118.9(3)	C(55)–C(54)–H(54)	120.5
C(53)–C(54)–H(54)	120.5	C(54)–C(55)–C(56)	120.5(3)
C(54)–C(55)–H(55)	119.8	C(56)–C(55)–H(55)	119.8
C(55)–C(56)–C(51)	122.5(3)	C(55)–C(56)–H(56)	118.7
C(51)–C(56)–H(56)	118.7	C(45)–C(42)–C(46)	111.1(2)

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

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Table S40. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(45)–C(42)–P(2)	108.90(18)	C(46)–C(42)–P(2)	116.14(19)
C(45)–C(42)–H(42)	106.7	C(46)–C(42)–H(42)	106.7
P(2)–C(42)–H(42)	106.7	C(44)–C(41)–C(43)	110.2(2)
C(44)–C(41)–P(2)	109.25(18)	C(43)–C(41)–P(2)	112.09(17)
C(44)–C(41)–H(41)	108.4	C(43)–C(41)–H(41)	108.4
P(2)–C(41)–H(41)	108.4	C#2–K(1)–C	180.0
C#2–K(1)–C(33)#2	86.95(7)	C–K(1)–C(33)#2	93.05(7)
C#2–K(1)–C(33)	93.05(7)	C–K(1)–C(33)	86.95(7)
C(33)#2–K(1)–C(33)	180.0	C#2–K(1)–C(21)#2	24.58(5)
C–K(1)–C(21)#2	155.42(5)	C(33)#2–K(1)–C(21)#2	111.02(6)
C(33)–K(1)–C(21)#2	68.98(6)	C#2–K(1)–C(21)	155.42(5)
C–K(1)–C(21)	24.58(5)	C(33)#2–K(1)–C(21)	68.98(6)
C(33)–K(1)–C(21)	111.02(6)	C(21)#2–K(1)–C(21)	180.00(7)
C#2–K(1)–C(11)	155.44(5)	C–K(1)–C(11)	24.56(5)
C(33)#2–K(1)–C(11)	101.81(7)	C(33)–K(1)–C(11)	78.19(7)
C(21)#2–K(1)–C(11)	137.01(5)	C(21)–K(1)–C(11)	42.99(5)
C#2–K(1)–C(11)#2	24.56(5)	C–K(1)–C(11)#2	155.44(5)
C(33)#2–K(1)–C(11)#2	78.19(7)	C(33)–K(1)–C(11)#2	101.81(7)
C(21)#2–K(1)–C(11)#2	42.99(5)	C(21)–K(1)–C(11)#2	137.01(5)
C(11)–K(1)–C(11)#2	180.00(7)	C#2–K(1)–Pd	143.93(4)
C–K(1)–Pd	36.07(4)	C(33)#2–K(1)–Pd	114.55(5)
C(33)–K(1)–Pd	65.45(5)	C(21)#2–K(1)–Pd	128.37(4)
C(21)–K(1)–Pd	51.63(4)	C(11)–K(1)–Pd	51.81(4)
C(11)#2–K(1)–Pd	128.19(4)	C#2–K(1)–Pd#2	36.07(4)
C–K(1)–Pd#2	143.93(4)	C(33)#2–K(1)–Pd#2	65.45(5)
C(33)–K(1)–Pd#2	114.55(5)	C(21)#2–K(1)–Pd#2	51.63(4)
C(21)–K(1)–Pd#2	128.37(4)	C(11)–K(1)–Pd#2	128.19(4)
C(11)#2–K(1)–Pd#2	51.81(4)	Pd–K(1)–Pd#2	180.0
C#2–K(1)–H(33B)	91.3(6)	C–K(1)–H(33B)	88.7(6)
C(33)#2–K(1)–H(33B)	164.5(6)	C(33)–K(1)–H(33B)	15.5(6)
C(21)#2–K(1)–H(33B)	66.8(6)	C(21)–K(1)–H(33B)	113.2(6)
C(11)–K(1)–H(33B)	74.1(6)	C(11)#2–K(1)–H(33B)	105.9(6)
Pd–K(1)–H(33B)	75.1(6)	Pd#2–K(1)–H(33B)	104.9(6)
C#2–K(1)–H(33A)	92.8(6)	C–K(1)–H(33A)	87.2(6)
C(33)#2–K(1)–H(33A)	164.9(6)	C(33)–K(1)–H(33A)	15.1(6)
C(21)#2–K(1)–H(33A)	70.9(6)	C(21)–K(1)–H(33A)	109.1(6)
C(11)–K(1)–H(33A)	84.5(6)	C(11)#2–K(1)–H(33A)	95.5(6)
Pd–K(1)–H(33A)	58.8(6)	Pd#2–K(1)–H(33A)	121.2(6)
H(33B)–K(1)–H(33A)	30.5(8)	C#2–K(1)–H(45B)	97.3(6)
C–K(1)–H(45B)	82.7(6)	C(33)#2–K(1)–H(45B)	80.2(7)

Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$

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Table S40. – continued from previous page

atom – atom – atom	angle	atom – atom – atom	angle
C(33)–K(1)–H(45B)	99.8(7)	C(21)#2–K(1)–H(45B)	105.6(6)
C(21)–K(1)–H(45B)	74.4(6)	C(11)–K(1)–H(45B)	106.7(6)
C(11)#2–K(1)–H(45B)	73.3(6)	Pd–K(1)–H(45B)	61.2(7)
Pd#2–K(1)–H(45B)	118.8(7)	H(33B)–K(1)–H(45B)	115.3(9)
H(33A)–K(1)–H(45B)	84.8(10)	Symmetry transformations used to generate equivalent atoms: #1 $-x, y, -z + \frac{1}{2}$; #2 $-x, -y + 2, -z$	