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Received 00th January 20xx,

Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

Nonsolvent-assisted Fabrication of the StructurizedPloylactideas Superhydrophobic Surfaces

Yafang Chang<sup>a</sup>, Xuying Liu<sup>a</sup>, Huige Yang<sup>a</sup>, Li Zhang<sup>a</sup>, Zhe Cui<sup>a</sup>, Mingjun Niu<sup>a</sup>, Hongzhi Liu <sup>b,\*</sup>, and Jinzhou Chen<sup>a,\*</sup>

**Corresponding Author** 

### Prof. Jinzhou Chen

School of Materials Science and Engineering, Zhengzhou University, Zhengzhou 450001, China

Tel: 0086 0371 67739217.

E-Mail: cjz@zzu.edu.cn

## Prof. Hongzhi Liu

College of Engineering, Zhejiang Agricultural & Forestry University, Lin'an, Hangzhou, Zhejiang, 311300, China.

E-Mail: hzliu@iccas.ac.cn



<sup>&</sup>lt;sup>a,</sup>\*School of Materials Science and Engineering, Zhengzhou University, Zhengzhou

<sup>450001,</sup> China.Email: cjz@zzu.edu.cn

<sup>&</sup>lt;sup>b,\*</sup> College of Engineering, Zhejiang Agricultural & Forestry University, Lin'an,

Hangzhou, Zhejiang, 311300, China. Email:hzliu@iccas.ac.cn

#### 1. Materials

Polylactide (PLA) resin (4032D) was a semicrystalline grade one in the form of pellets and commercially supplied by NatureWorks Co. Ltd (Minnetonka, USA). Chloroform, dichloromethane, absolute ethyl alcohol, n-butyl alcohol, and n-butyl acetate were purchased from Tianjin Chemical Regent Co., Ltd. (China) and used as received.

# 2. Preparation of PLA-i, PLA-ii, PLA-iii, PLA-iv and PLA-v coatings

Table 1 shows the solvent and nonsolvent used to prepare PLA-i, PLA-ii, PLA-iii, PLA-iv, and PLA-v samples. In the case of sample PLA-i, a known amounts of dried PLA (2.34g) pellets were dissolved in a given solvent/nonsolvent mixture at room temperature under magnetic stirring. The weight ratio of chloroform (solvent) to PLA was fixed at 2:1 (w/w) and the weight ratio of PLA to absolute ethyl alcohol (nonsolvent) was varied between 0-5 wt% with a step increment of 0.5 wt%. According to the same procedure, PLA-ii coating was fabricated using chloroform as a solvent and absolute ethyl alcohol/n-butyl alcohol as a binary non-solvent. The ratio of absolute ethyl alcohol to n-butyl alcohol was kept at 1:1. PLAiii coating was fabricated using DCM as solvent and absolute ethyl alcohol as non-solvent; and PLA-iv coating was fabricated by using (DCM) as solvent, n-butyl alcohol and nbutyl acetate as binary non-solvent with the rate of n-butyl alcohol and n-butyl acetate kept as 1:1; likewise, PLA-v coating was fabricated by using DCM as solvent, absolute ethyl alcohol, n-butyl alcohol and n-butyl acetate as ternary nonsolvent with the rate of absolute ethyl alcohol, n-butyl alcohol and n-butyl acetate kept as 1:1:1.

#### 3. Characterization

The morphologies of the film surfaces were observed using scanning electron microscopy (SEM, JEOL, JSM-7500F, Japan) at an accelerating voltage of 5kV and atomic force microscopy (AFM, Shimadzu Corp., SPM-9500J3, Japan) operated in a tapping mode, respectively. Differential scanning calorimetry (DSC) measurements were performed using a PerkinElmer Pyris Diamond DSC under a nitrogen atmosphere at the heating rate of  $10^{\circ}$ C/min from home temperature to  $200^{\circ}$ C. The weight of DSC sample was about 3-5mg. Water contact angleS of the film on the glass slide was measured with a Contact angle analyzer (Powereach, JC2000A, China) based on a sessile drop measuring method at ambient temperature. The volume of DI water droplet was 4µL and five different points for each film were tested. The contact angle hysteresis (CAH) on the sample was measured at the room temperature with CA System (DSA100, Kruss Co., Germany). The adhesion force was assessed by a highly sensitive microelectromechanical balance system (Kruess). A  $10\mu L$  ultrapure Milli-Q water

droplet with an electrical resistivity of  $10^{18}\Omega$ ·cm was suspended on a home-built platinum semispherical cap, which was mounted o the microelectromechanical balance system.

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Sample	Nonsolvent content/mL	CA/°	ϑa/°	ϑr/°	(ϑa-ϑr)/°	<i>T<sub>m</sub>/°</i> C	⊿ <i>H<sub>f</sub></i> /J·g-1	X <sub>c</sub> /%
PLA-i	0.5	82	-	-	-	162.3	18.71	20.10
PLA-i	3	151	151.1	138.9	12.2	162.4	62.17	66.78
PLA- ii	2.5	90	-	-	-	162.4	32.25	34.64
PLA- ii	3.5	150	151	137	14	162.0	55.03	59.11
PLA- iii	0.5	88	-	-	-	162.1	21.65	20.09
PLA- iii	1.5	140.3	-	-		162.4	33.82	36.33
PLA- iv	1	88.3	-	-	-	161.7	48.49	52.08
PLA- iv	3.5	151.3	151.5	135.4	16.1	163.9	57.29	61.54
PLA- v	2	95.3	-	-	-	162.4	33.92	36.43
PLA- v	3	151.8	152.1	139.5	12.6	159.9	51.07	54.83

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Fig.S2 Images of water droplets of 4  $\mu L$  on the PLA-v surface with different tilt angles: (a) 0°, (b) 90°, and (c) 180°.