Supporting Information

Lateral actuation of an organic droplet on conjugated polymer electrodes via

imbalanced interfacial tensions

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1. Schematic of experimental setup.



Fig. S1 Schematic of the experimental setup. The PPy(DBS) samples were placed in a quartz cell filled with 0.1 M NaNO₃. A copper tape was used as the counter electrode, and a saturated calomel electrode (SCE) was used as the reference electrode. The voltage was applied by using a potentiostat. The organic droplet was monitored through the camera of the goniometer.

2. Calculation of the interfacial tension of a flattened sessile droplet.



Fig. S2 Schematic of the cross-section view of a flattened droplet on a substrate. a and b indicate the outside and inside of the vertex of a droplet. d and e indicate the inside and outside of the point where the radius of droplet is maximal. c is in the droplet and at the same height as d. h is the distance between points b and c. w is the width of the droplet.

In Fig. S2, the following equations can be formulated for hydrostatic pressures,

- $P_{\rm a} + \Delta P_{\rm ab} = P_{\rm b} \tag{S1}$
- $P_{\rm b} + \Delta P_{\rm bc} = P_{\rm c} \tag{S2}$
- $P_{\rm c} = P_{\rm d} \tag{S3}$
- $P_{\rm d} \Delta P_{\rm de} = P_{\rm e} \tag{S4}$
- $P_{\rm e} \Delta P_{\rm ae} = P_{\rm a} , \qquad (S5)$

where *P* represents the pressure at the indicated positions. ΔP_{ab} and ΔP_{de} are the Laplace pressures due to the surface curvature. Adding Equation S1 to Equation S5 results in

$$\Delta P_{\rm ab} + \Delta P_{\rm bc} = \Delta P_{\rm de} + \Delta P_{\rm ae}$$
 (S6)

or

$$\Delta P_{\rm bc} - \Delta P_{\rm ae} = \Delta P_{\rm de} - \Delta P_{\rm ab} , \qquad (S7)$$

in which

$$\Delta P_{\rm bc} - \Delta P_{\rm ae} = \Delta \rho g h \tag{S8}$$

$$\Delta P_{\rm de} = \gamma_{\rm L_O} \left(\frac{1}{R_{\rm d1}} + \frac{1}{R_{\rm d2}} \right)$$
 (S9)

$$\Delta P_{\rm ab} = \gamma_{\rm L_O} \left(\frac{1}{R_{\rm b1}} + \frac{1}{R_{\rm b2}} \right), \qquad (S10)$$

where $\Delta \rho$ is the density difference between a droplet (e.g., DCM) and surrounding medium (i.e., 0.1 M NaNO₃), *g* is the constant acceleration due to gravity, *h* is the height difference between position *b* and *c*, γ_{L_0} is the interfacial tension of the droplet and surrounding medium. R_{d1} and R_{d2} are the principal radii of curvature at position *d*, and R_{b1} and R_{b2} are the principal radii of curvature at position *b*. Substituting Equation S8 to S10 into Equation S7 results in

$$\Delta \rho g h = \gamma_{\rm L_O} \left(\frac{1}{R_{\rm d1}} + \frac{1}{R_{\rm d2}} - \frac{1}{R_{\rm b1}} - \frac{1}{R_{\rm b2}} \right). \quad (S11)$$

Thus, the interfacial tension between a droplet and the surrounding medium can be computed according to the droplet shape as

$$\gamma_{\rm L_O} = \frac{\Delta \rho g h}{\left(\frac{1}{R_{\rm d1}} + \frac{1}{R_{\rm d2}} - \frac{1}{R_{\rm b1}} - \frac{1}{R_{\rm b2}}\right)}.$$
 (S12)

According to Equation S12, the interfacial tension of the flattened droplet shown in Fig. 2b-ii is calculated as $0.27 \text{ mN} \cdot \text{m}^{-1}$.