

**Supporting Information for**  
**Cobalt phosphide-based electrocatalysts: Synthesis and phase**  
**catalytic activity comparison for hydrogen evolution**

Yuan Pan, Yan Lin, Yinjuan Chen, Yunqi Liu\*, Chenguang Liu\*

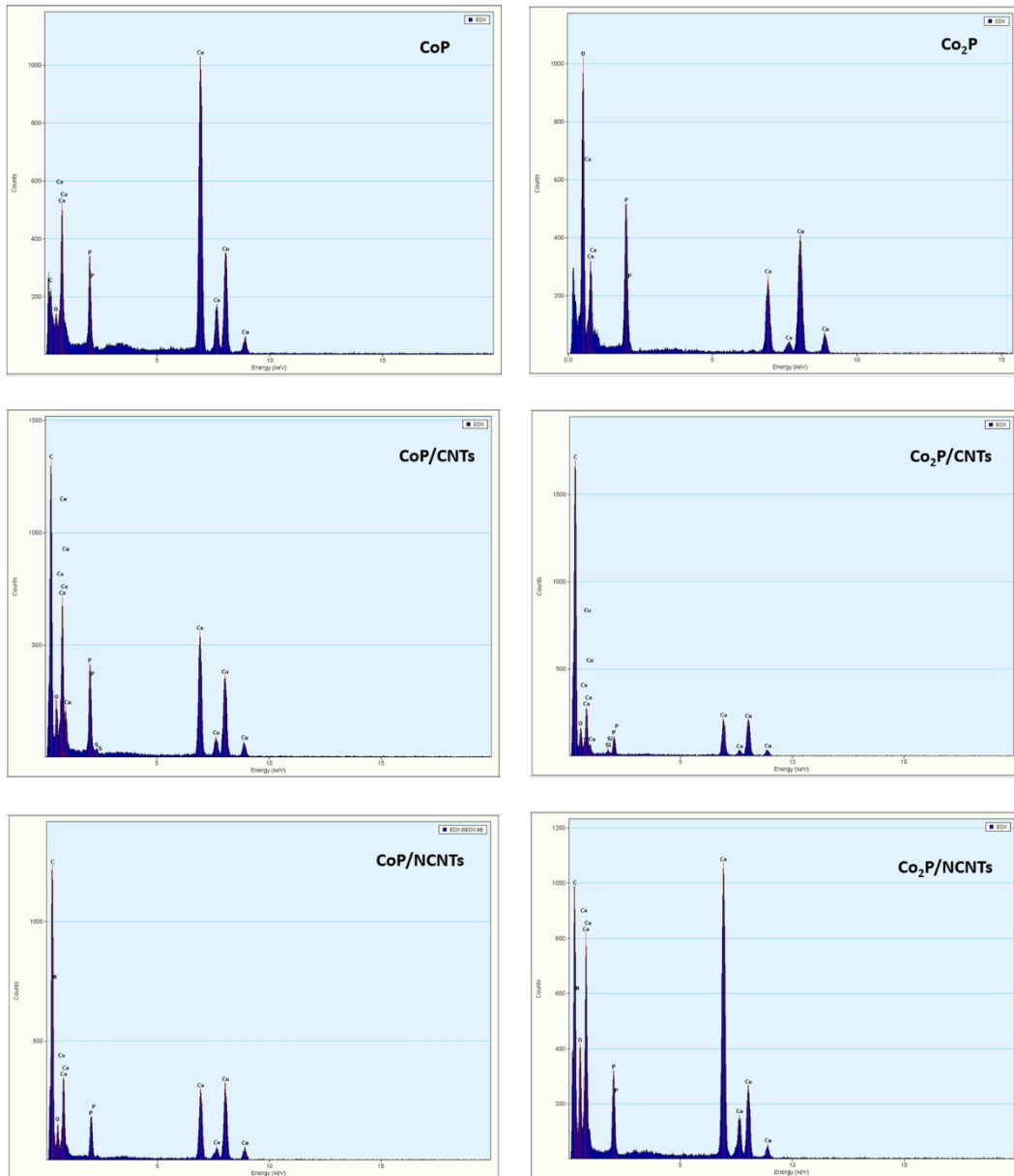
*State Key Laboratory of Heavy Oil Processing, Key Laboratory of Catalysis, China University of*

*Petroleum, 66 West Changjiang Road, Qingdao, Shandong 266580, P. R. China*

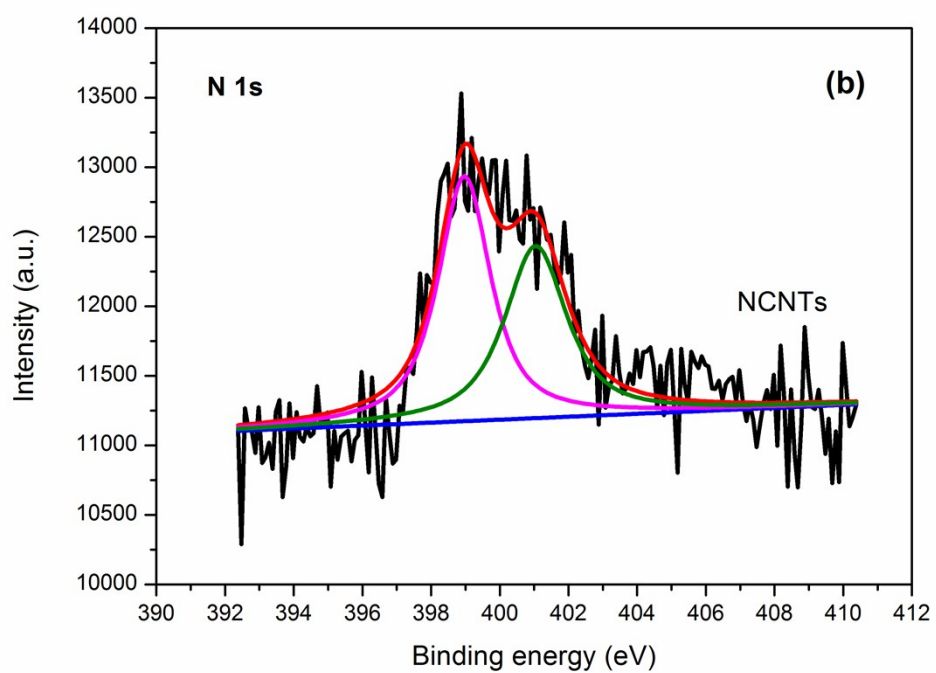
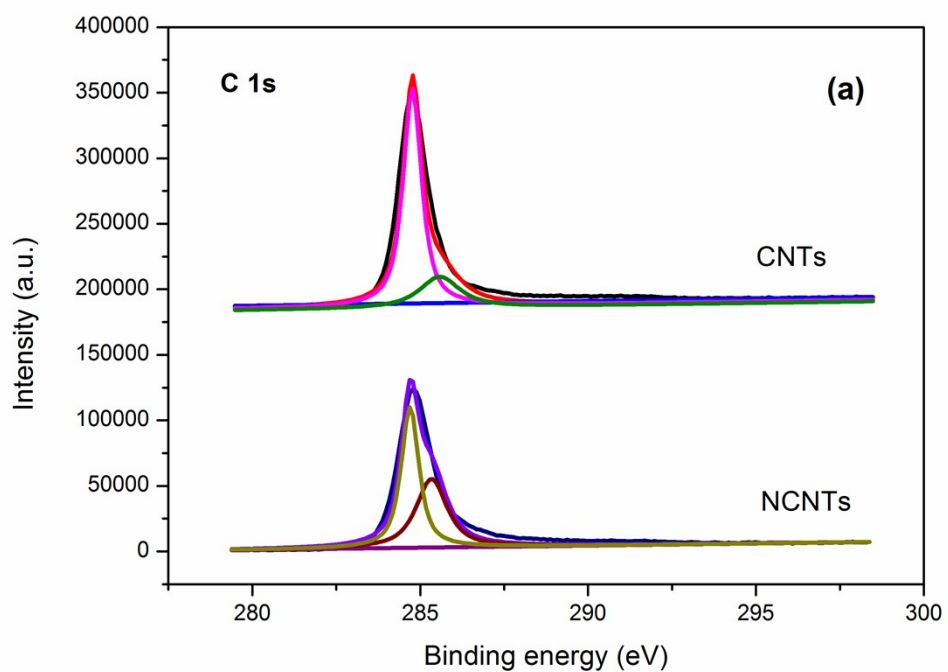
---

\* Corresponding author. E-mail address: [liuyq@upc.edu.cn](mailto:liuyq@upc.edu.cn); [cgliu1962@sina.com](mailto:cgliu1962@sina.com)

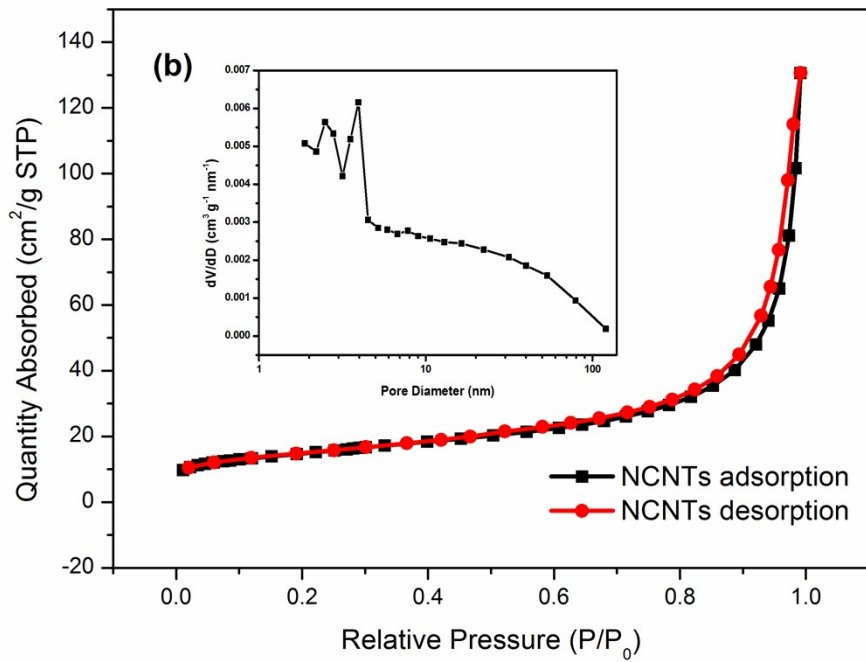
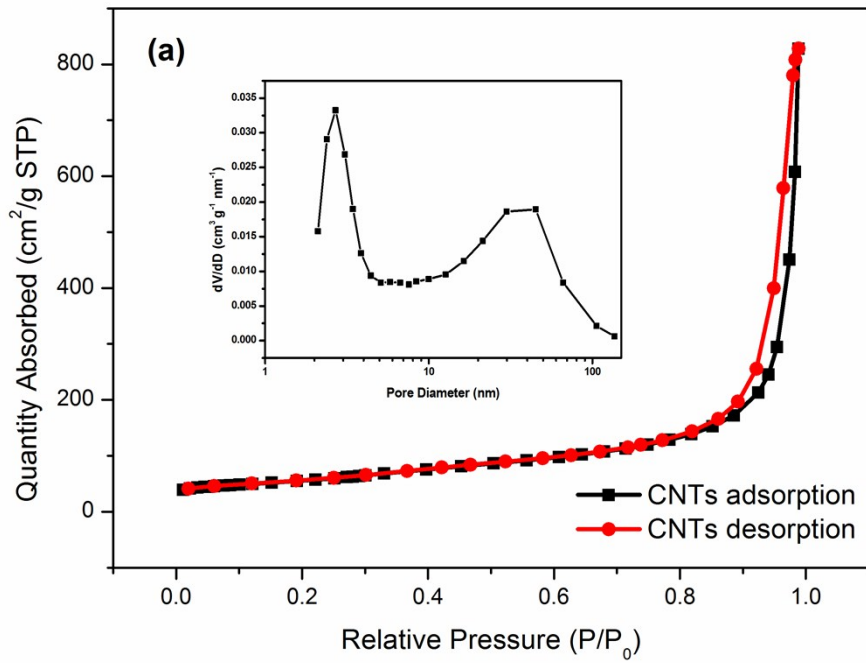
Tel.: +86-532-86981861; +86-532-86981716.



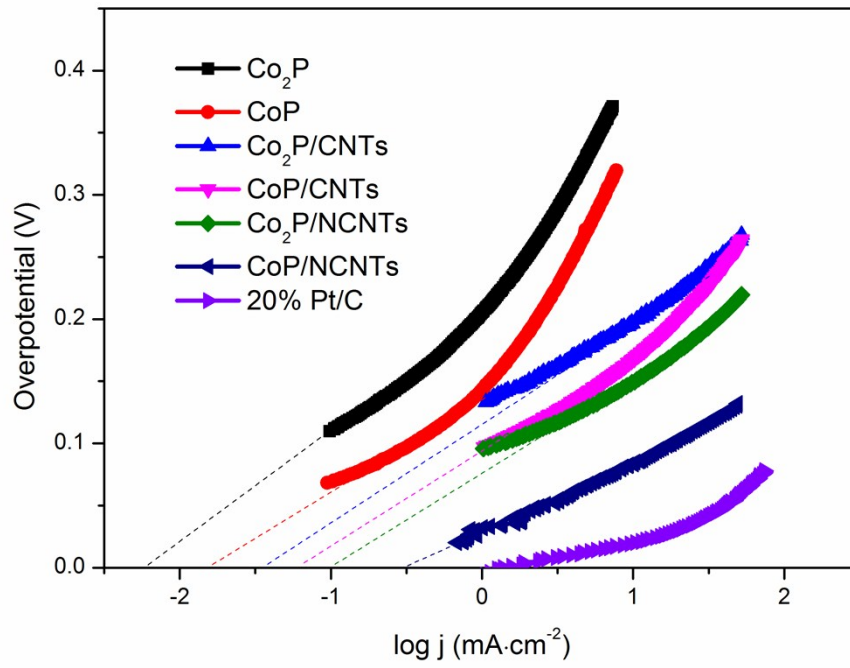
**Fig. S1** EDX spectra of the as-synthesized cobalt phosphide-based catalysts.



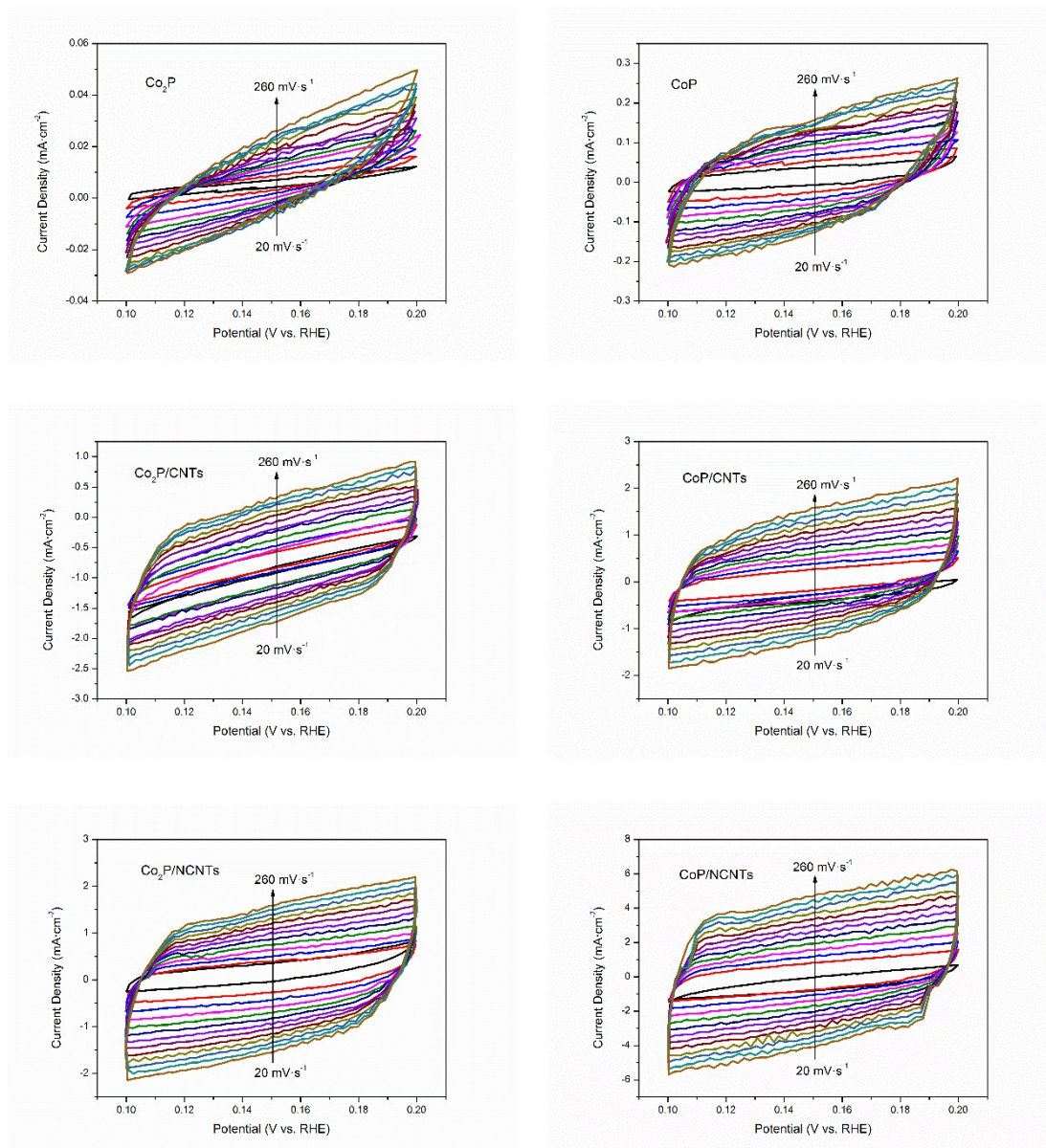
**Fig. S2** XPS spectra in the C (1s) and N (1s) regions for the CNTs and NCNTs.



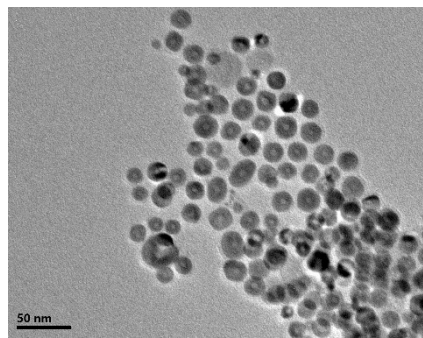
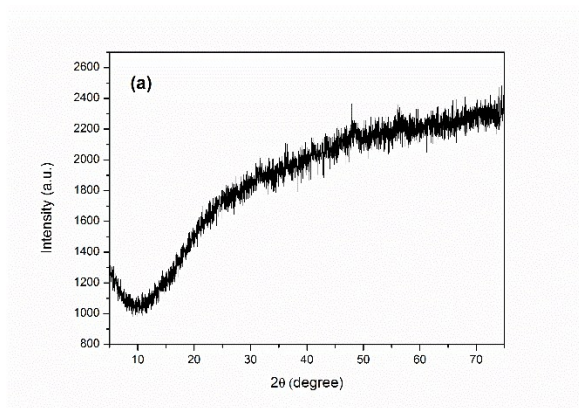
**Fig. S3**  $\text{N}_2$  adsorption-desorption isotherms and Barrett-Joyner-Halenda (BJH) pore-size distribution (inset in Fig. S3) of (a) CNTs and (b) NCNTs



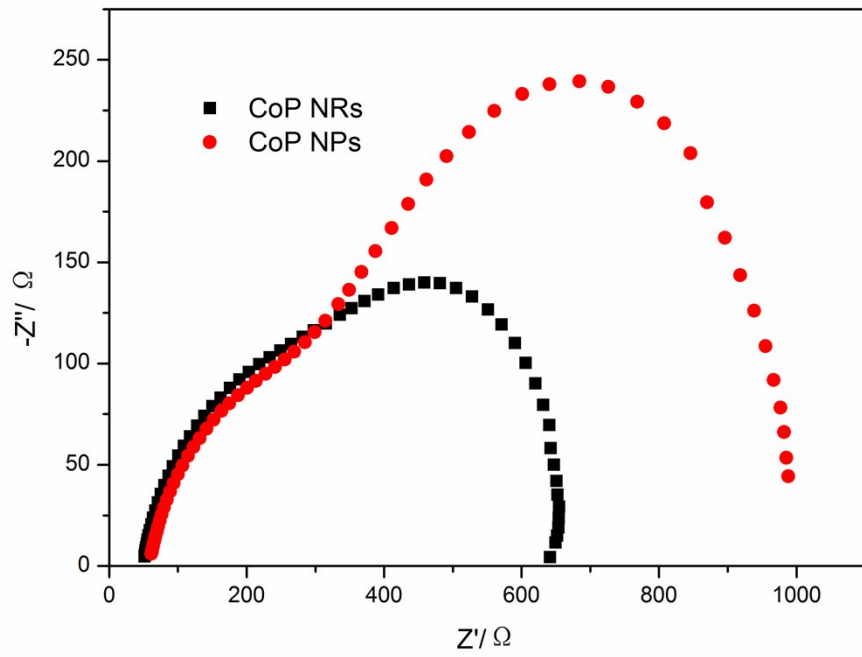
**Fig. S4** Calculated exchange current density of the as-synthesized cobalt phosphide-based catalysts by using extrapolation methods.



**Fig. S5** Cyclic voltammety curves of the as-synthesized cobalt phosphide-based catalysts in 0.5 M H<sub>2</sub>SO<sub>4</sub> solution in the region of 0.1-0.2 V vs. RHE with different scan rates from 20 mV·s<sup>-1</sup> to 260 mV·s<sup>-1</sup>.



**Fig. S6** XRD and TEM results of the as-synthesized CoP nanoparticles.



**Fig. S7** Nyquist plots of the as-synthesized CoP NRs and CoP NPs in 0.5 M  $\text{H}_2\text{SO}_4$  with an overpotential of 200 mV.



**Table S1** Comparison of HER performance of some transition metal phosphide catalysts in 0.5 M H<sub>2</sub>SO<sub>4</sub> solution.

Catalyst	Current density (mA·cm <sup>-2</sup> )	Overpotential (mV)	Tafel slope (mV·dec <sup>-1</sup> )	Reference
CoP/NCNTs	10	79	49	This work
CoP/CC	20	100	51	24
CoP Hollow Polyhedrons	10	159	59	25
urchin-like CoP	100	180	46	26
Co <sub>0.59</sub> Fe <sub>0.41</sub> P	10	72	52	27
CoP/RGO	10	156.89	70.2	28
CoP/MPC	10	141.73	69.8	28
CoP/MCV	10	134.34	63.1	28
CoP/OMC	10	112.18	56.7	28
CoP CPHs	10	133	51	29
CoP microspheres	10	226	76	30
Porous FeP	10	240	67	31
IPNTs	10	88	35.5	32
FeP NWs/rGO	10	107	58.5	33
Cu <sub>3</sub> P nanowire/CF	10	143	67	34
MoP-CA2	10	125	54	35
Ni <sub>2</sub> P/CNT	10	124	53	36
WP NAs/CC	10	130	69	37

**Table S2** Calculations of the exchange current density of the as-synthesized cobalt phosphide-based catalysts and 20 % Pt/C catalyst by using extrapolation methods.

Catalyst	$\log( j \text{ (mA}\cdot\text{cm}^{-2}) )$ at $\eta=0$	Exchange current density $j_0$ [mA $\cdot$ cm $^{-2}$ ]
	V	
Co <sub>2</sub> P	-2.2	0.0032
CoP	-1.77	0.017
Co <sub>2</sub> P/CNTs	-1.41	0.039
CoP/CNTs	-1.17	0.068
Co <sub>2</sub> P/NCNTs	-0.99	0.102
CoP/NCNTs	-0.5	0.32
20 % Pt/C	0.17	1.48

**Table S3** ICP element composition and content analysis of the electrolyte after reaction.

	Co (mg/L)	P (mg/L)	S (mg/L)
Electrolyte after reaction	36.84	12.67	5103.68

**Table S4** Values of elements in equivalent circuit resulted from fitting the EIS data of the as-synthesized CoP/NCNTs catalyst.

Potential (mV) vs. RHE	$R_s$ ( $\Omega$ )	Q ( $F \cdot cm^{-2} \cdot S^{n-1}$ )	n	R ( $\Omega$ )
-60	9.85	3.9E-3	0.8	268.3
-80	9.95	4.5E-3	0.8	109.3
-100	9.87	4.9E-3	0.8	50.95
-120	10.16	5.4E-3	0.8	28.22
-140	10.37	5.7E-3	0.8	10.56