

Supporting Information

A pinecone-inspired nanostructure design for long-cycle and high performance Si anode

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Figure S1. XPS of $\text{SiO}_2@\text{GF}$ and encapsulated Si. (a), (b) Si 2p states and O 1s states of $\text{SiO}_2\text{-GF}$. (c) Si 2p states of encapsulated Si, the transformation from (a) to (c) shows clearly the conversion of Si–O–Si bond to Si–Si bond through magnesium thermal reduction. (d) XPS survey spectrum of $\text{SiO}_2\text{-GF}$ and uE-Si, demonstrating the successful reduction of silica to silicon during the magnesium thermal reduction.

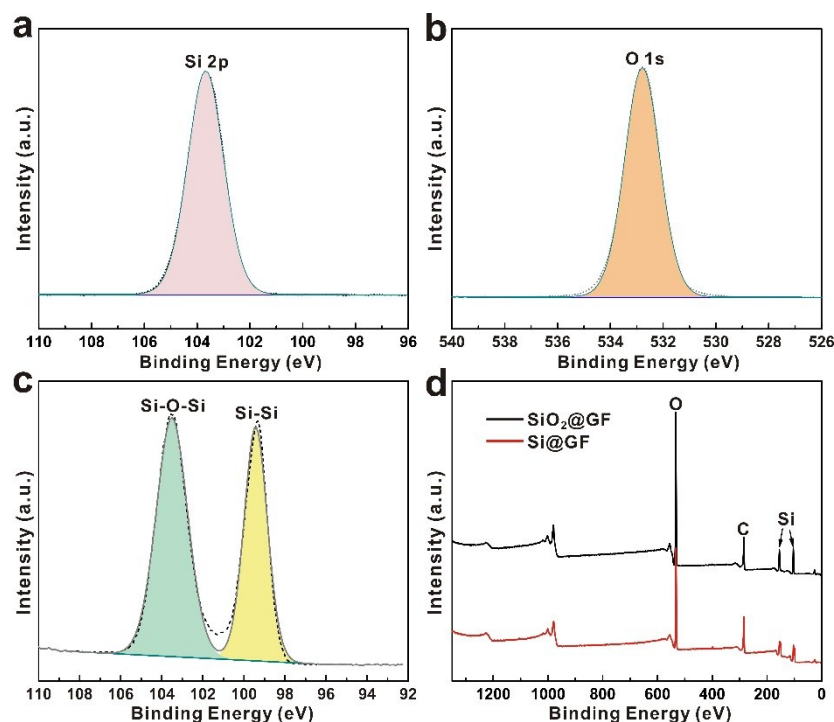


Figure S2. SEM and TEM images of GF, SiO₂-GF and uE-Si (a) SEM image of GF, a smooth surface with some wrinkles. (b) SEM image of SiO₂-GF, close contact between SiO₂ and GF is achieved by SCF method. (c) TEM image of SiO₂-GF, graphene is coated with a thick layer of amorphous SiO₂, as confirmed by the diffraction inset. (d) HRTEM image of SiO₂-GF, confirms the presence of amorphous SiO₂ and graphene. (e) TEM image of uE-Si, as a supporter, GF is loaded with Si nanoparticles. (f) HRTEM image of uE-Si confirms the presence of crystalline Si (111), as well as an amorphous SiO₂ layer on the surface of Si.

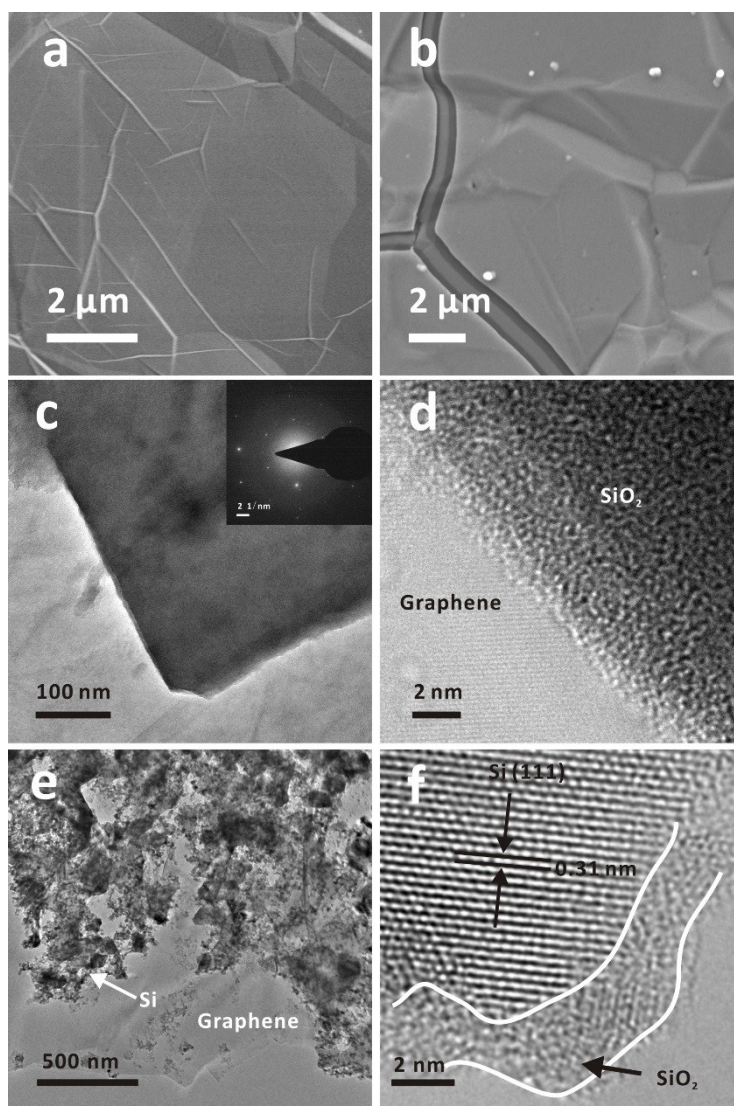


Figure S3. SEM and Characterizations of uE-Si and GE-Si after cycling and statistical analysis of the diameters of GE-Si and uE-Si before and after cycling. (a) SEM image of uE-Si after cycling at low magnification shows severe aggregation. (b) an enlarged image of (a) shows expansion and various particle sizes. (d) SEM image of GE-Si after cycling at low magnification reveals a uniform particle size. (e) enlarged image of (d) shows the porous structure remains, indicating a reversible expansion/contraction process. (c) Particle diameters distribution of GE-Si and uE-Si before cycling. (f) Particle diameters distribution of GE-Si and uE-Si after cycling.

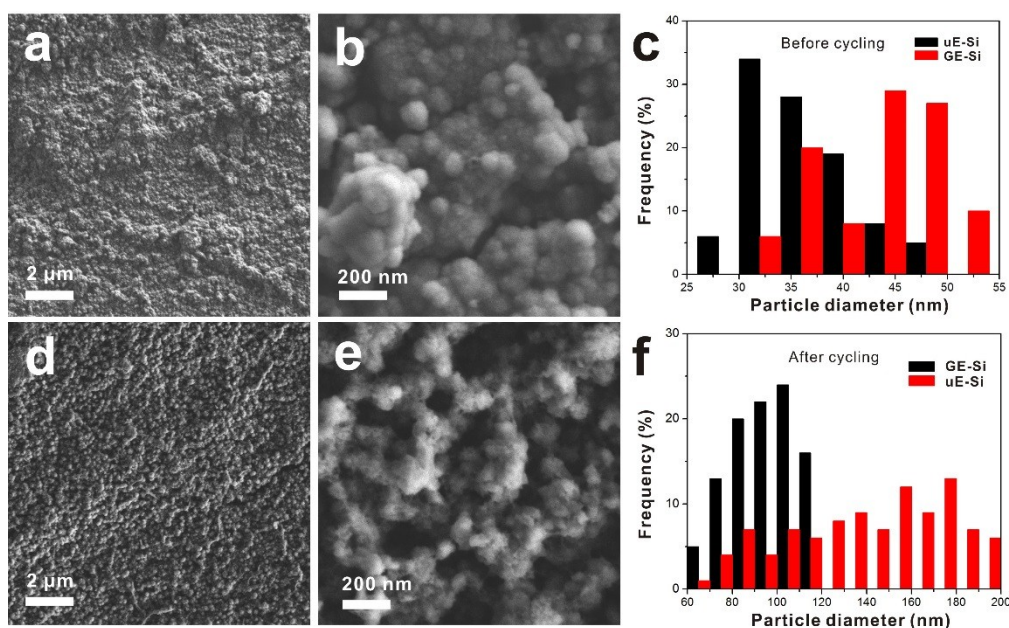


Figure S4. Raman spectra of (a) GF and (b) GE-Si composite. In **Figure S4a**, the Raman intensity of the G band is higher than the 2D band, which is consistent with the multilayer property of the graphene foam. The low intensity of the D bands centred at 1350 cm^{-1} implies a high quality for the GF. In **Figure S4b**, the peak at $\sim 520\text{ cm}^{-1}$ corresponds to the characteristic band of single crystalline silicon. The relatively higher intensity of the D band than that of the G band mainly results from the partially-disordered structure of the overlapped graphene sheets. The absence of GF may be attributed to the covering of a thick Si layer.

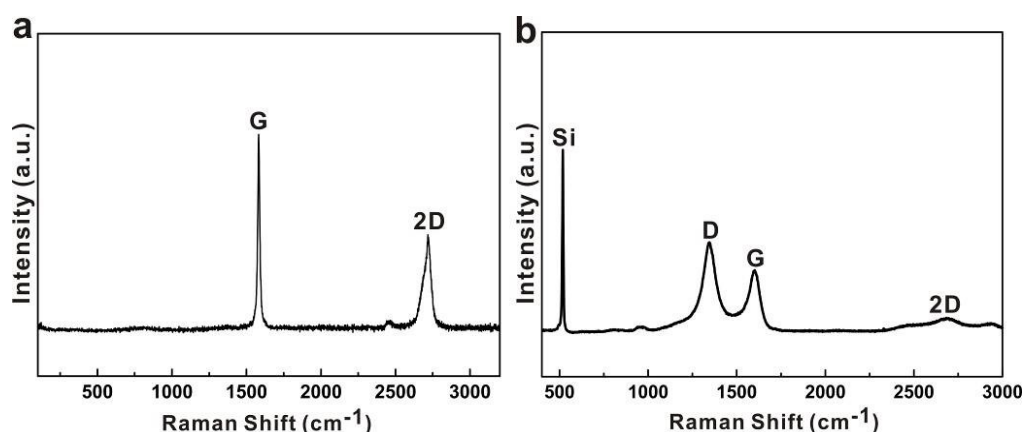


Figure S5. Thermal gravimetric analysis (TGA) of GE-Si composite. The mass loss at about 700 °C is caused by the decomposition of GF and graphene in GE-Si. The mass increased after 900 °C because of the oxidation of Si. The result indicates the GE-Si composite is consisted of 52.9% Si and 47.1% graphene.

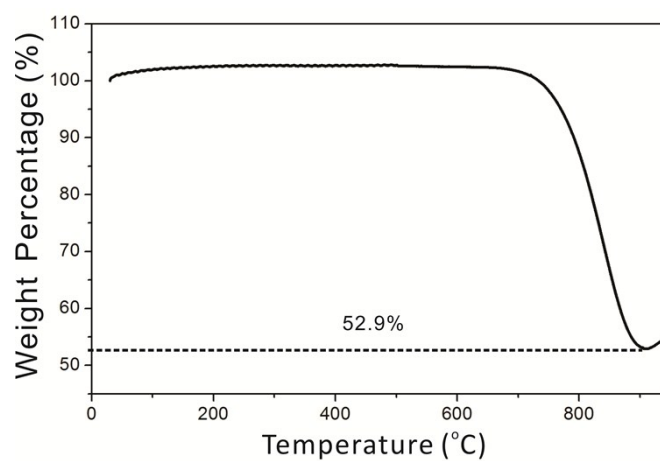
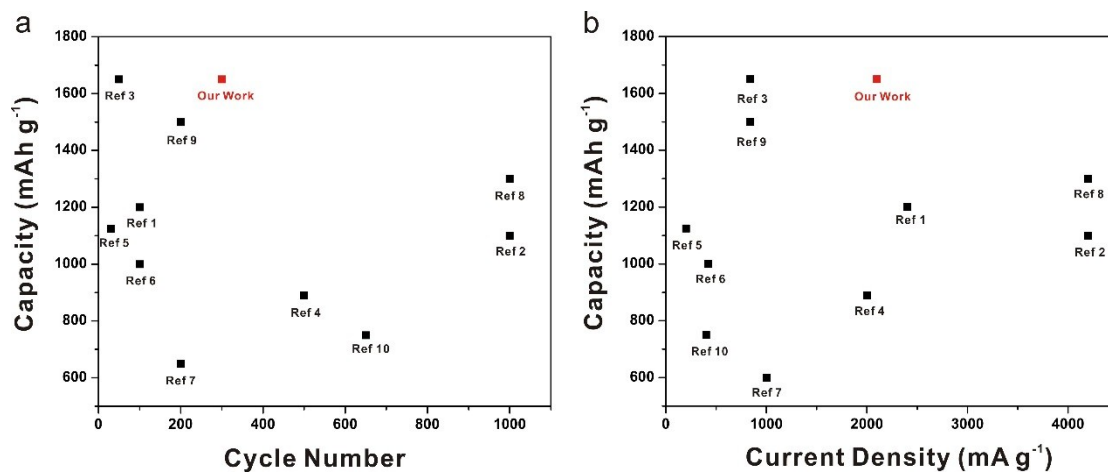


Figure S6. A comparison of electrochemical performance of pinecone-like GE-Si electrode with recent studies on Si-based anodes. (a) Comparison of capacity in different cycle numbers. (b) Comparison of capacity in different current density.^{1–10}



Reference

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