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Supporting Information

Fabricating SiO₂-coated V₂O₅ nanoflake arrays for highperformance lithium-ion batteries with enhanced cycling capability

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Figure S1 SEM images of NH_4VO_3 nanoflake arrays prepared at pH=2 (a), 3 (b), and 4 (c) with other conditions unchanged.



Figure S2 Top-view (a), and cross-sectional (b) SEM images of the obtained V_2O_5 arrays.



Figure S3 X-ray energy dispersive spectrum (EDS) is to show the presence of elements in V_2O_5 (a SiO₂ nanoflake arrays.



Figure S4 TEM image of V_2O_5 (a) SiO₂ nanoflakes and corresponding X-ray energy dispersive spectrum (EDS) mapping of V, O, and Si.



Figure S5 Photos of (a) NH_4VO_3 nanoflake array, (b) V_2O_5 array, and (c) $V_2O_5@SiO_2$ nanoflake arrays electrodes. This shows the look (color) of real coated electrodes.



Figure S6 Nitrogen sorption isotherm (BET) data to show specific surface areas and pore size distribution (indicated in the inset) for the V_2O_5 @ SiO₂ nanoflakes.



Figure S7 SEM images of V_2O_5 array (a), and V_2O_5 @SiO₂ nanoflake array (b) electrodes after 100 cycles at 0.1 A g⁻¹.



Figure S8 Nyquist plots of the $V_2O_5@SiO_2$ nanoflake array electrode at first cycle and 100^{th} cycle, respectively.