Supporting information

Single-Crystalline α -Fe₂O₃ Void@Frame

Microframes for Rechargeable Batteries

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Figure S1. The α -Fe₂O₃ precursor used in this work: (a) Low-magnification FESEM image shows dozens of α -Fe₂O₃ microbeads with pore; (b) magnified FESEM image of a typical α -Fe₂O₃ microbead showing the detailed structure and texture; (c) Low-magnification TEM image shows dozens of α -Fe₂O₃ microbeads, and (d) magnified TEM image of a few representative α -Fe₂O₃ microbeads.



Figure S2. Charge-discharge profiles of the 120^{th} , 240^{th} , 360^{th} and 550^{th} cycles under cycling test at 200 mA/g, indicating the same electrochemical reactions involved and high reversibility.



Figure S3. (a) XRD and (b) FESEM image of α -Fe₂O₃ microstructure solid control for LIB cycling test for comparison.



Figure S4. (a) Low magnification and (b) magnified FESEM images for α -Fe₂O₃ microframes anode after 120-cycle rate test in LIB. The outline of the microstructures was maintained after repeated charge-discharge processes.