Supporting Information for

Facile synthesis of $\text{Co}_3\text{V}_2\text{O}_8$ interconnected hollow microsphere anode with superior high-rate capability for Liion battery

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Supplementary Figures

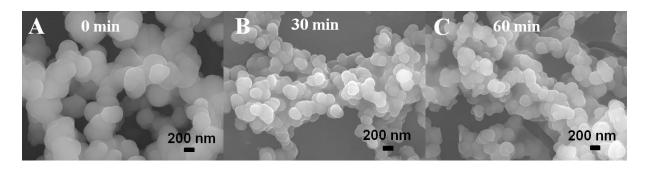


Fig. S1 Typical SEM images of the precursor particles obtained in solutions at step A. (A) 0 min; (B) 30 min; (C) 60 min.

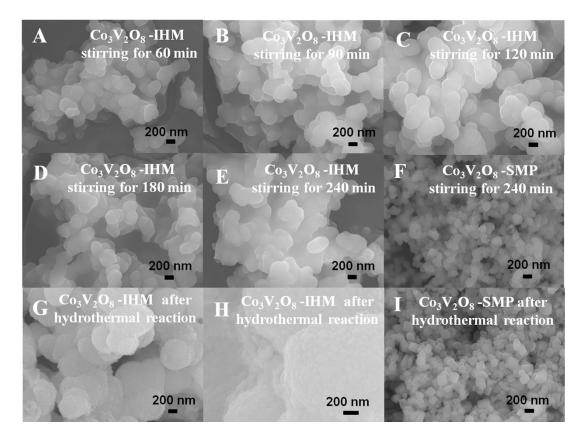


Fig. S2 Typical SEM images of the precursor particles obtained in solutions at step B (A-F). $Co_3V_2O_8$ -IHM precursor particles: (A) 60 min; (B) 90 min; (C) 120 min; (D) 180 min; (E) 240 min. $Co_3V_2O_8$ -SMP precursor particles: (F) 240 min. Typical SEM images of the $Co_3V_2O_8$ -IHM (G, H) and $Co_3V_2O_8$ -SMP (I) precursor particles obtained in solutions after step C.

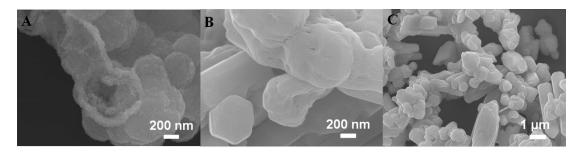


Fig. S3 Typical SEM images of $Co_3V_2O_8$ -IHM (A) and $Co_3V_2O_8$ -SMP (B, C).

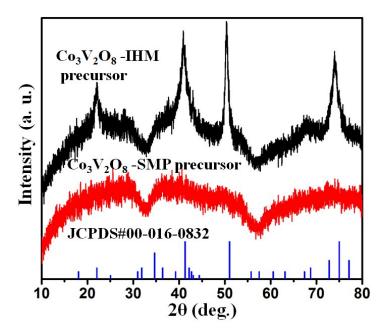


Fig. S4 XRD patterns of as-prepared precursor particles after hydrothermal reaction (step C).

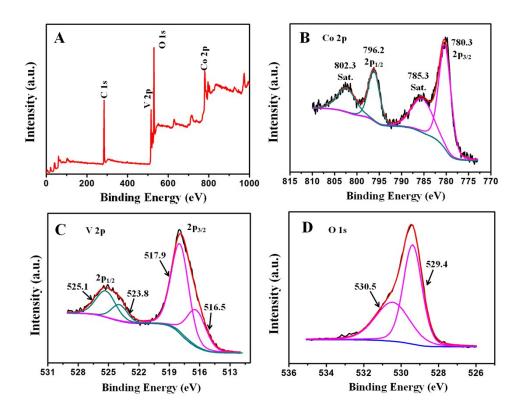


Fig. S5 XPS spectra of the $\text{Co}_3\text{V}_2\text{O}_8\text{-IHM}.$

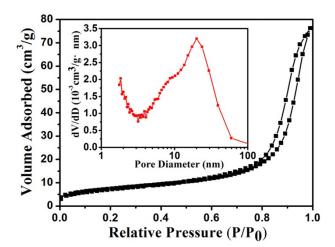


Fig. S6 The N_2 adsorption-desorption isotherms of $\text{Co}_3\text{V}_2\text{O}_8$ -IHM. The insets show the corresponding BJH pore size distribution curves.

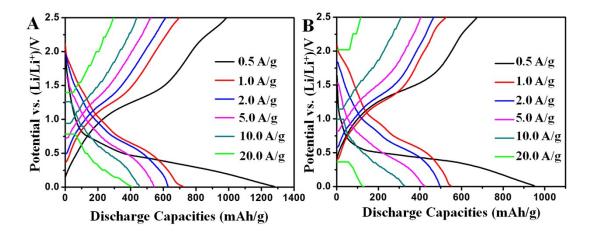


Fig. S7 The corresponding first cycle voltage capacity profiles of Co₃V₂O₈-IHM (A) and Co₃V₂O₈-SMP (B) at different current densities during rate performance test.

Table S1 The data of EIS test of Co₃V₂O₈-IHM after different cycles.

Cycle Numbers	Before Cycle	5 th	10 th	20 th	50 th
$R_{e}\left(\Omega\right)$	4.1	5.9	4.9	8.5	9.5
$R_{f}\left(\Omega\right)$	-	0.1	0.45	1.2	1.4
$R_{ct}\left(\Omega\right)$	30.1	8.1	9.2	8.4	8.6