

Supplementary Information

Design and Synthesis Energetic Materials towards High Density and Positive Oxygen Balance by N-dinitromethyl Functionalization of Nitroazoles

X. X. Zhao,^a S. H. Li,^a Y. Wang,^a Y. C. Li,^a F. Q. Zhao^b and S. P. Pang^{*a}

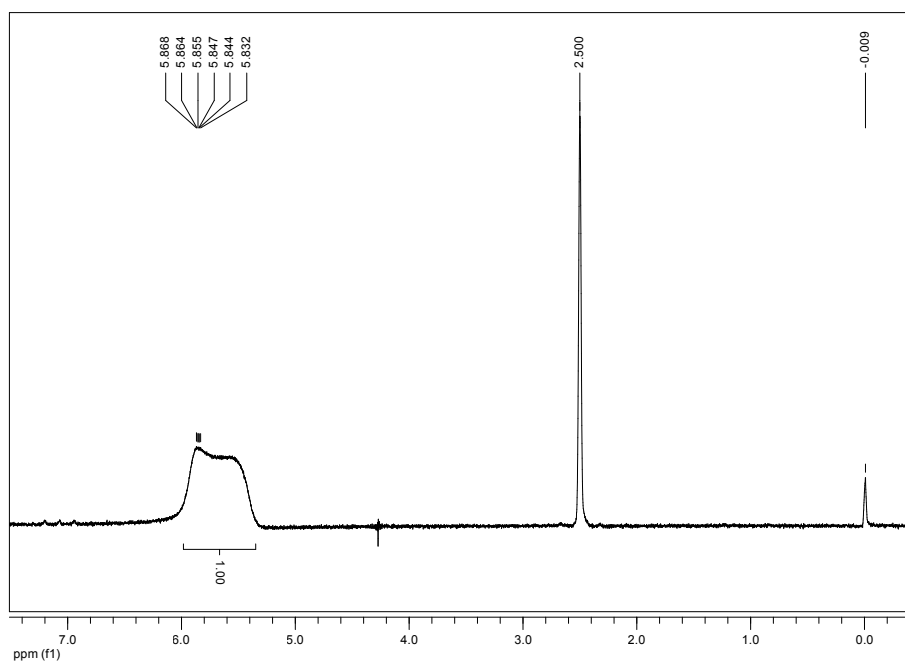
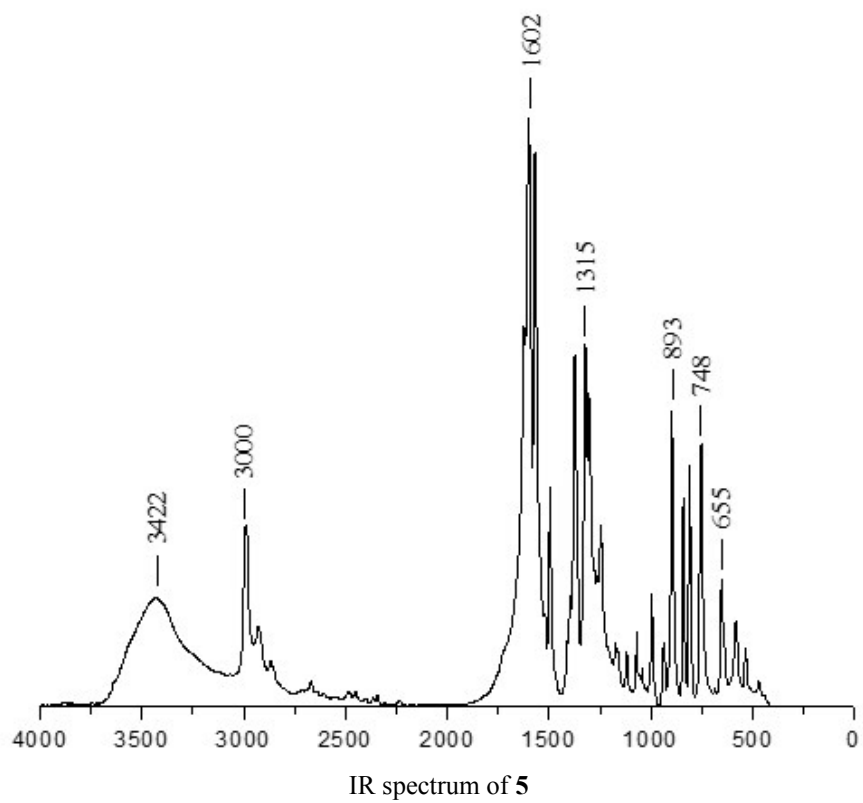
Content

S3-29 Spectral data of compounds **5-8, 5a-5d, 6a-6d, 7a-7d**

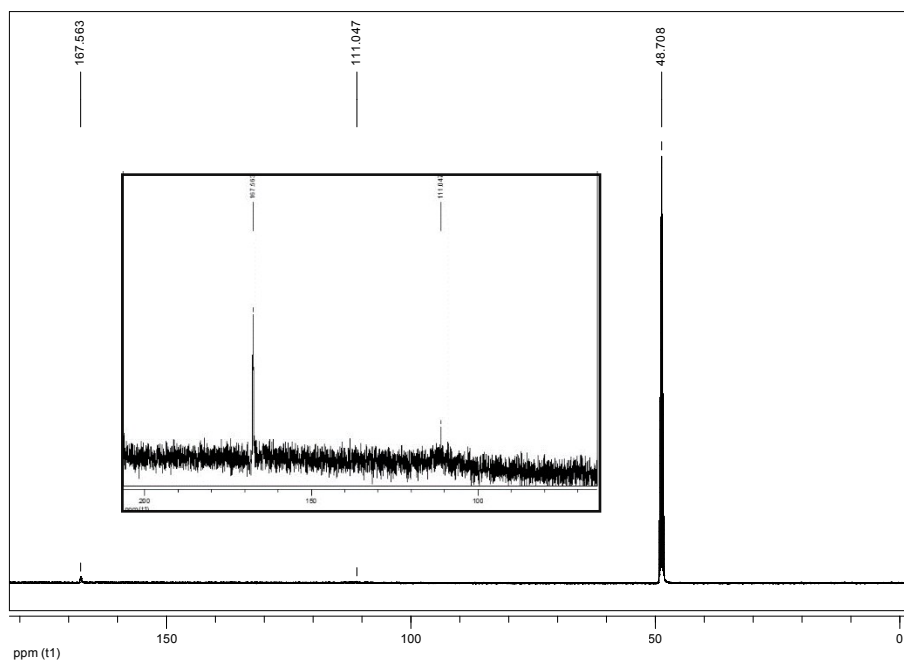
S30-37 DSC data of compounds **5-8, 5a-5d, 6a-6d, 7a-7d**

S38-42 Table of crystal data for **7, 8, 5a, 5b, 5d, 6a-6d, 7a-7d**

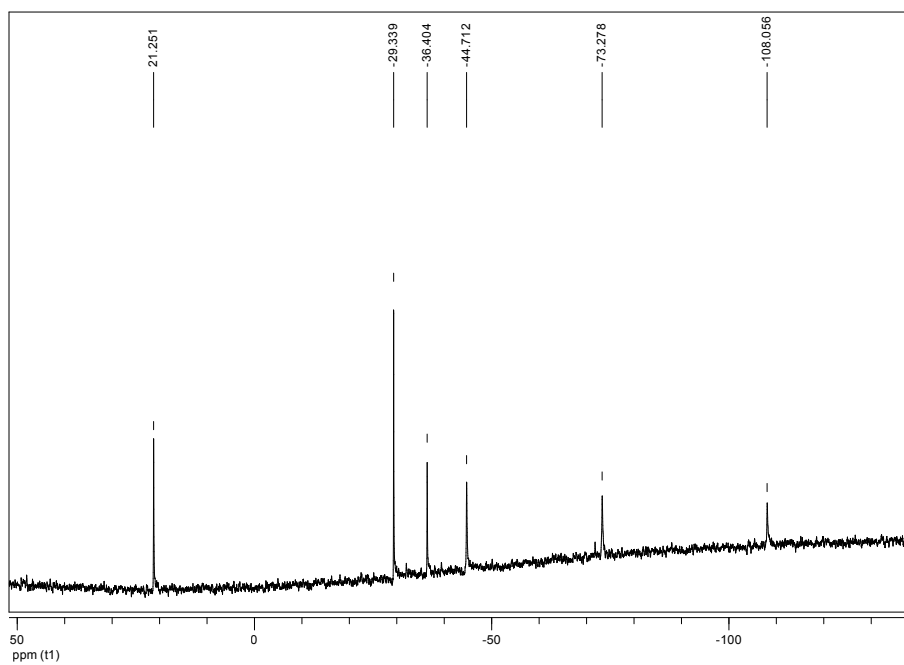
S43-43 Heat of formation calculations of **5-8, 5a-5d, 6a-6d, 7a-7d**



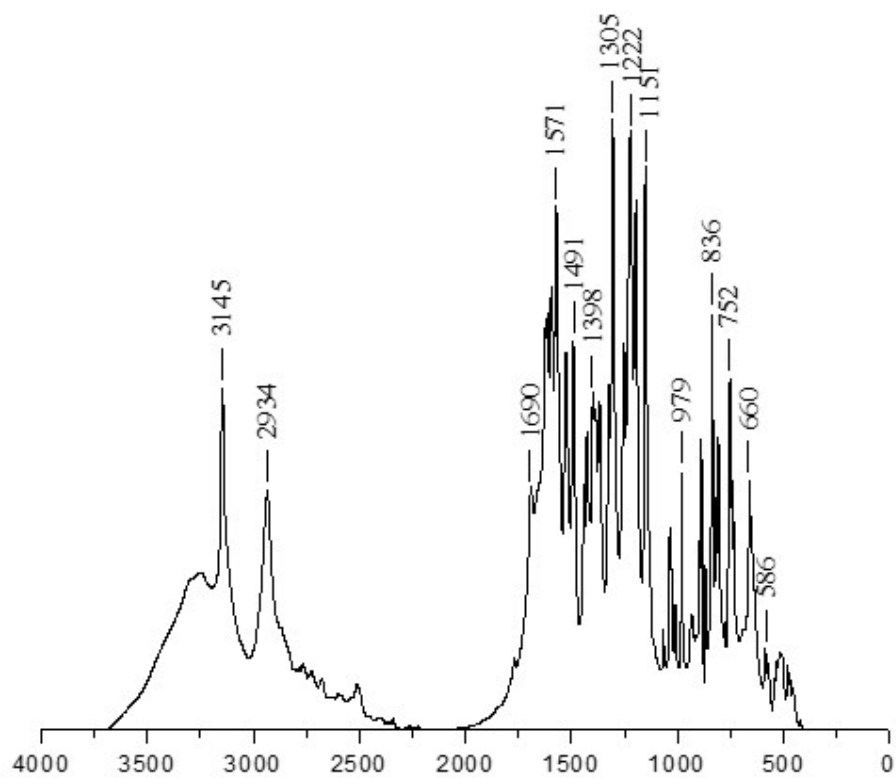
¹H NMR spectrum (400 MHz) of **5** in DMSO-d₆ at 25 °C



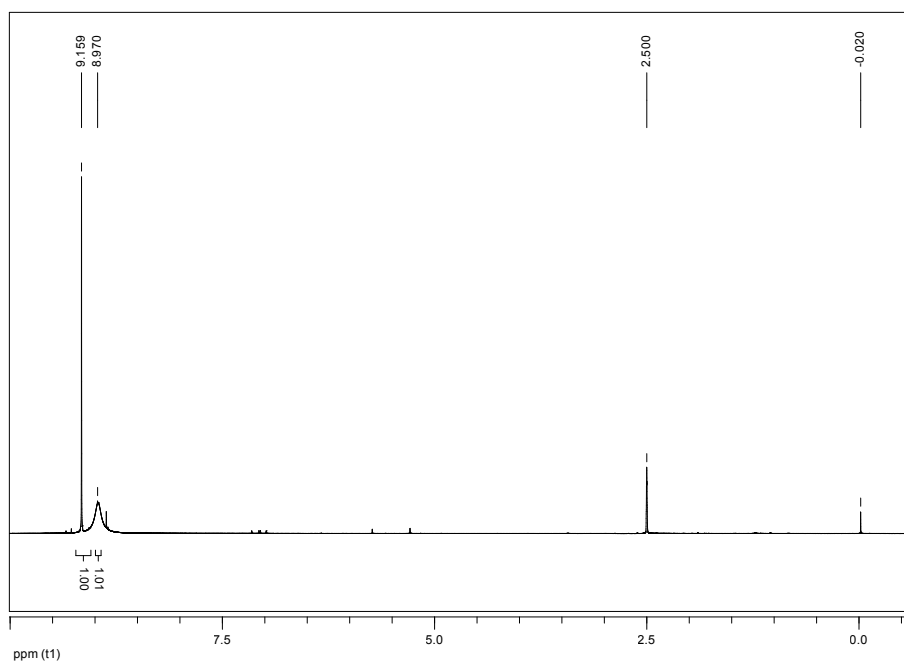
¹³C NMR spectrum (400 MHz) of **5** in methanol-d₄ at 25 °C



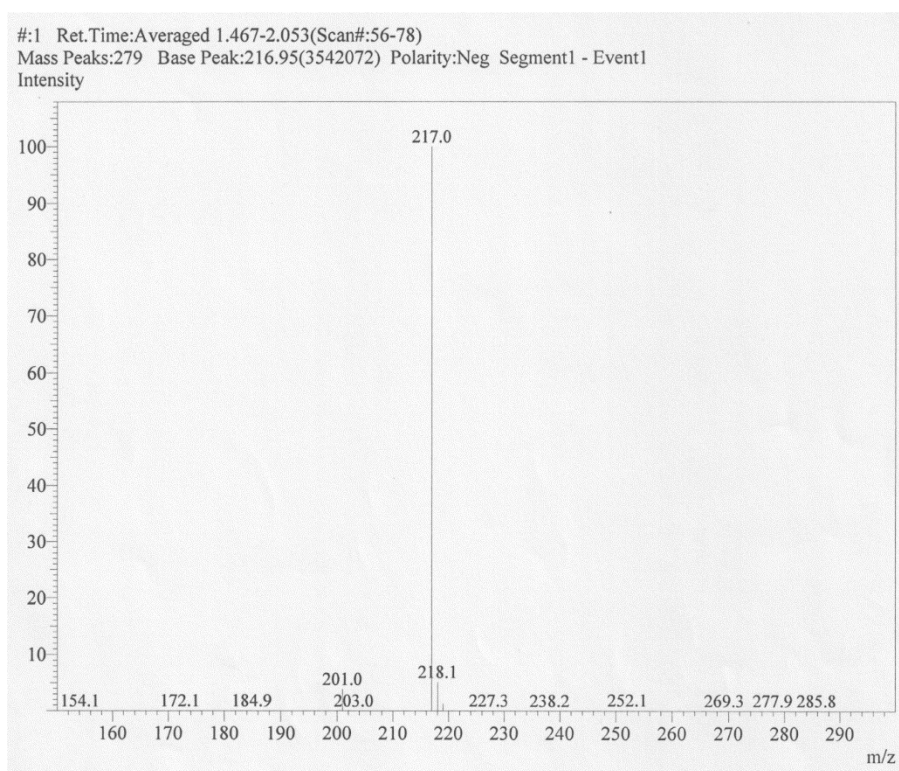
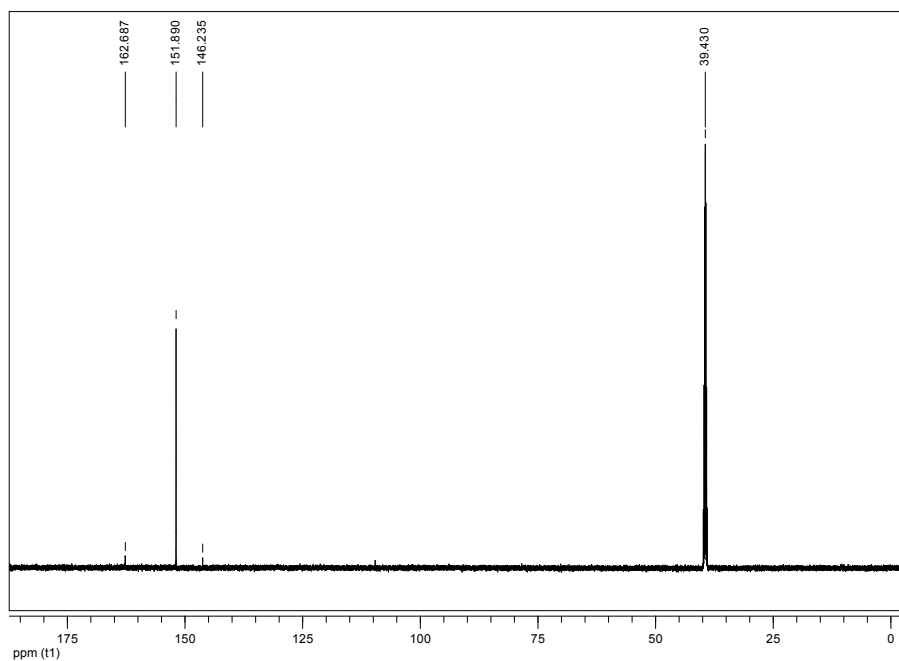
¹⁵N NMR spectrum of **5** in methanol-d₄ at 25 °C



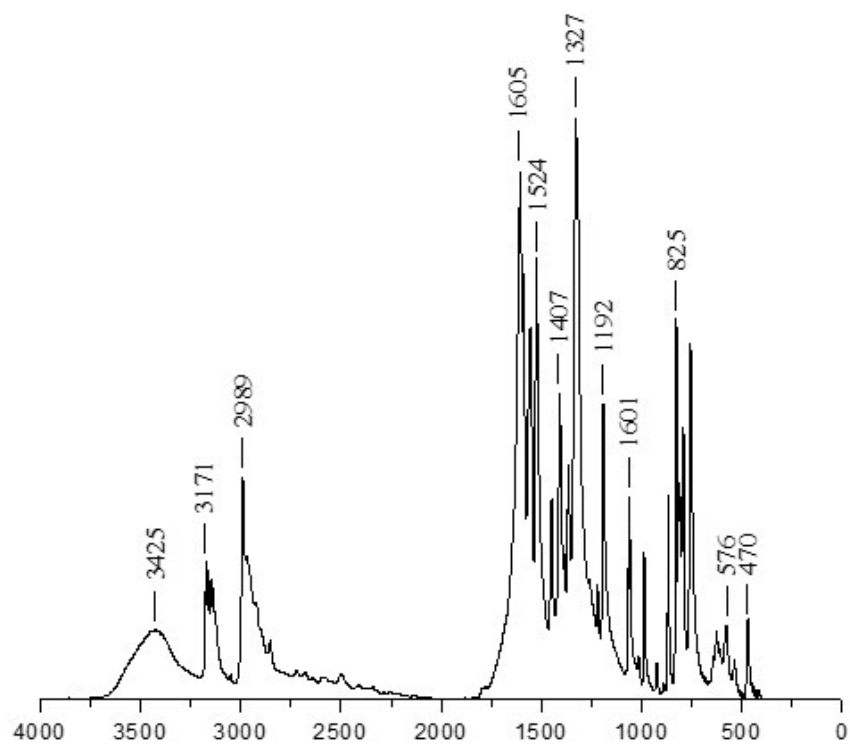
IR spectrum of **6**



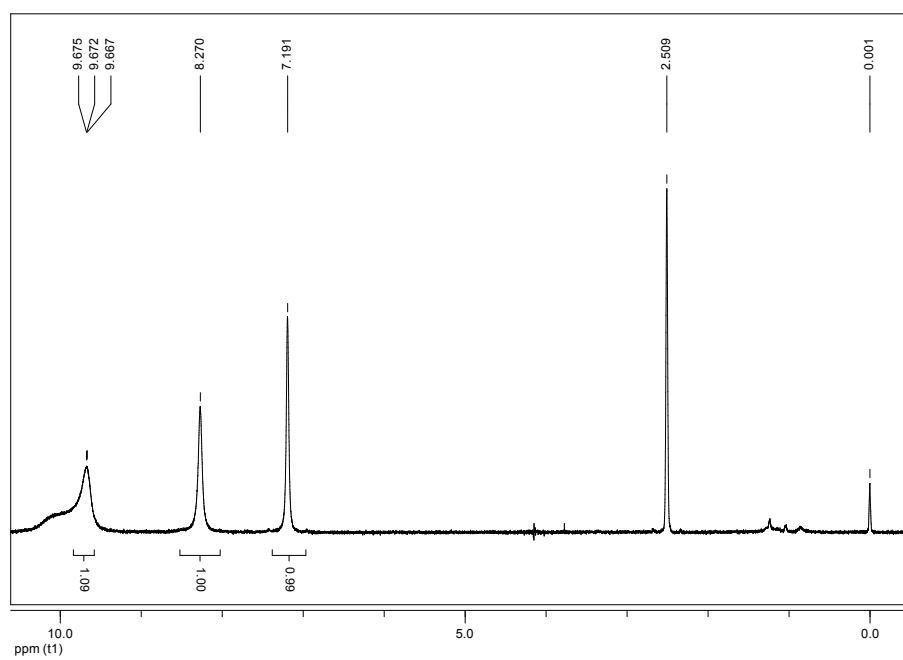
¹H NMR spectrum (400 MHz) of **6** in DMSO-d₆ at 25 °C



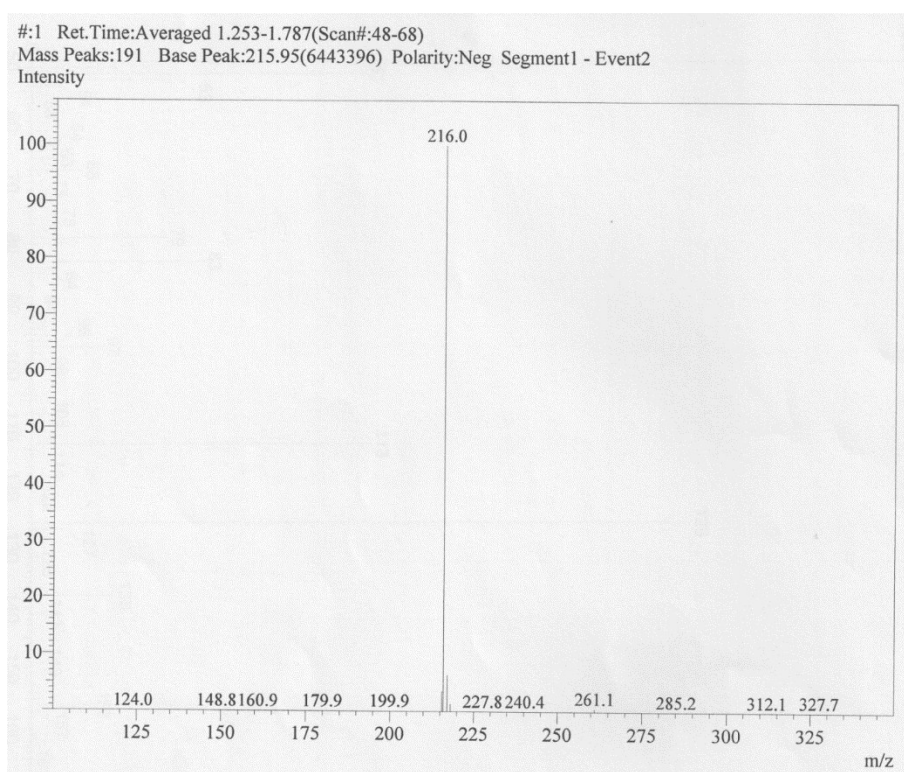
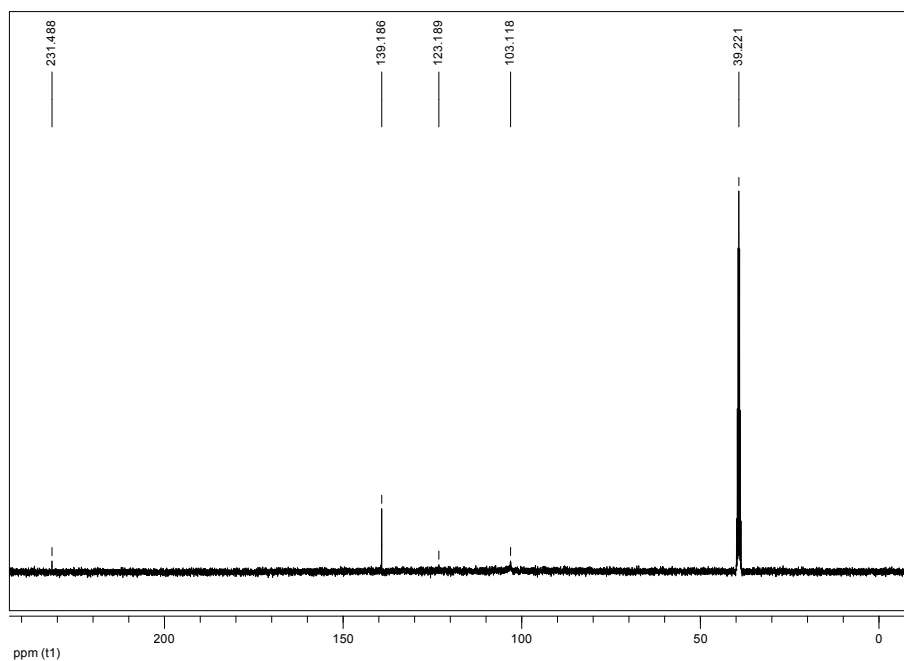
ESI spectrum of **6**

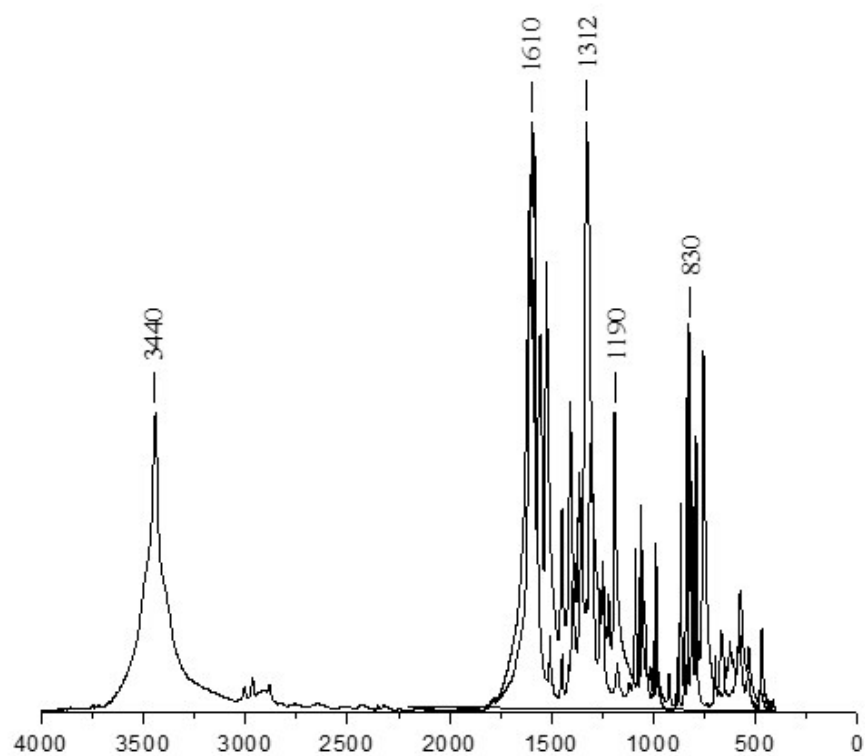


IR spectrum of 7

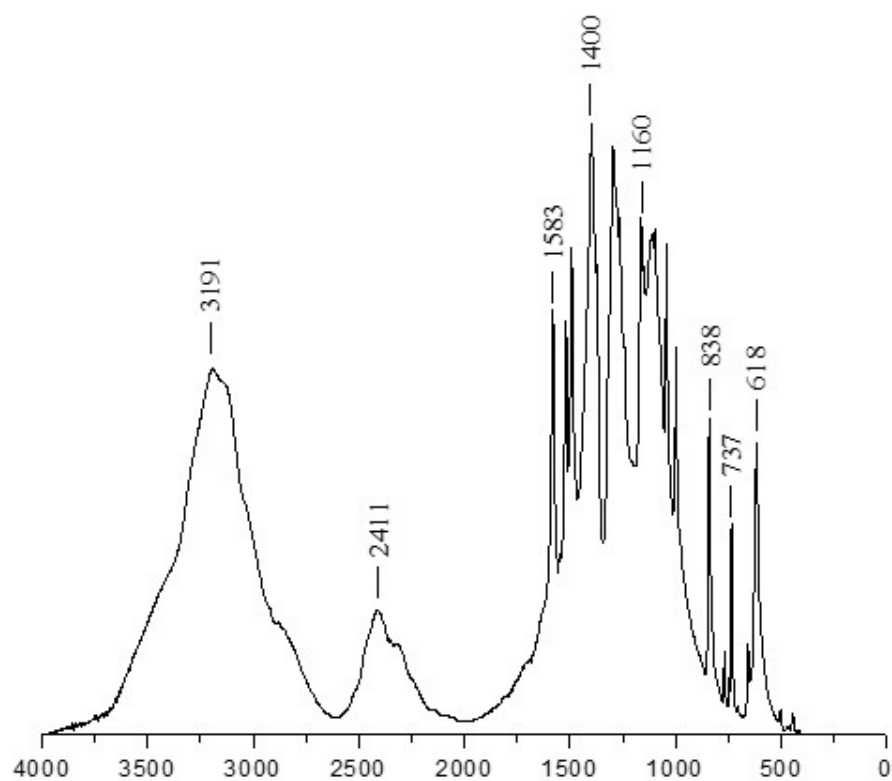


¹H NMR spectrum (400 MHz) of 7 in DMSO-d₆ at 25 °C

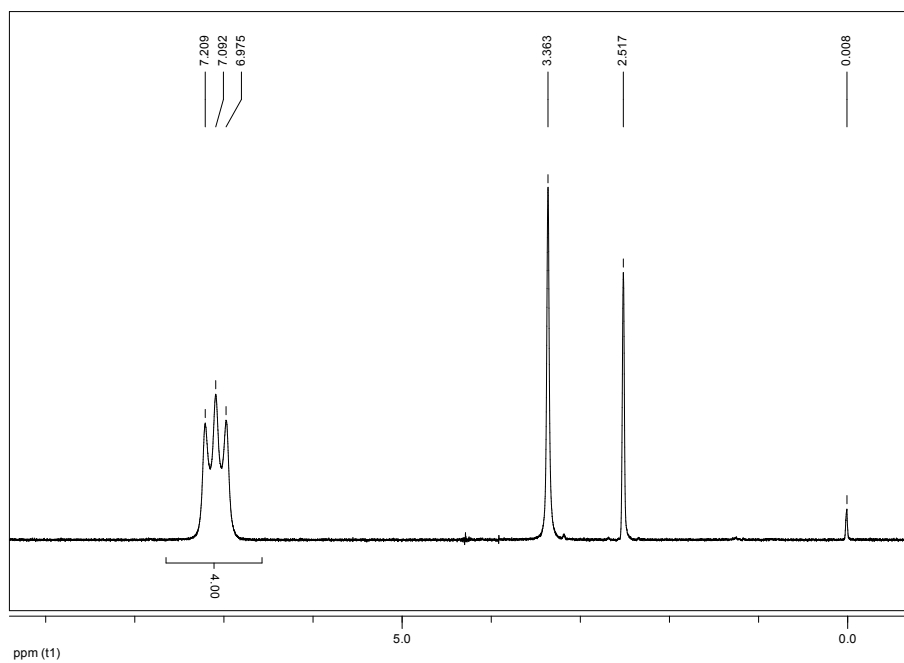




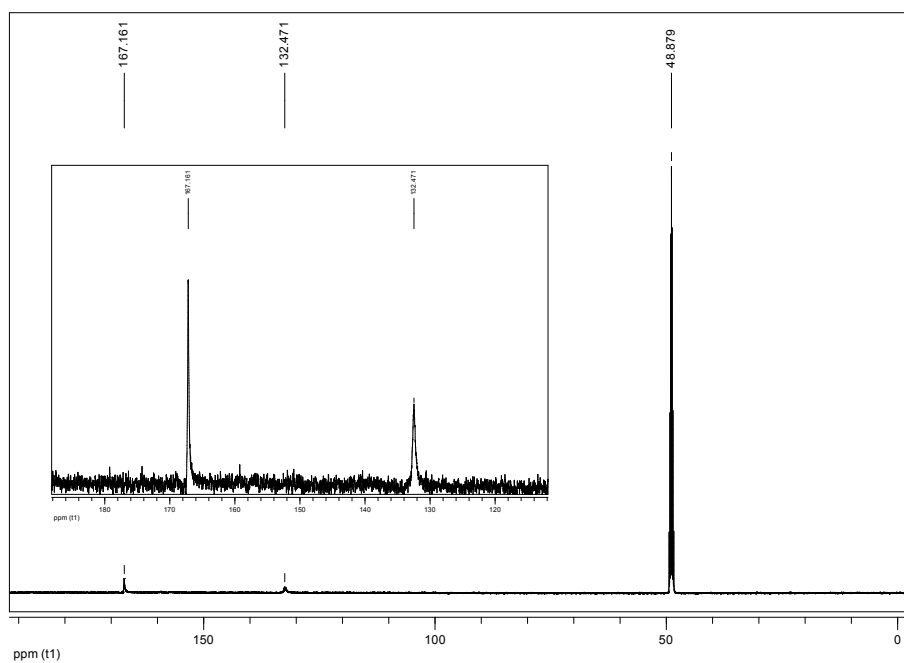
IR spectrum of **8**



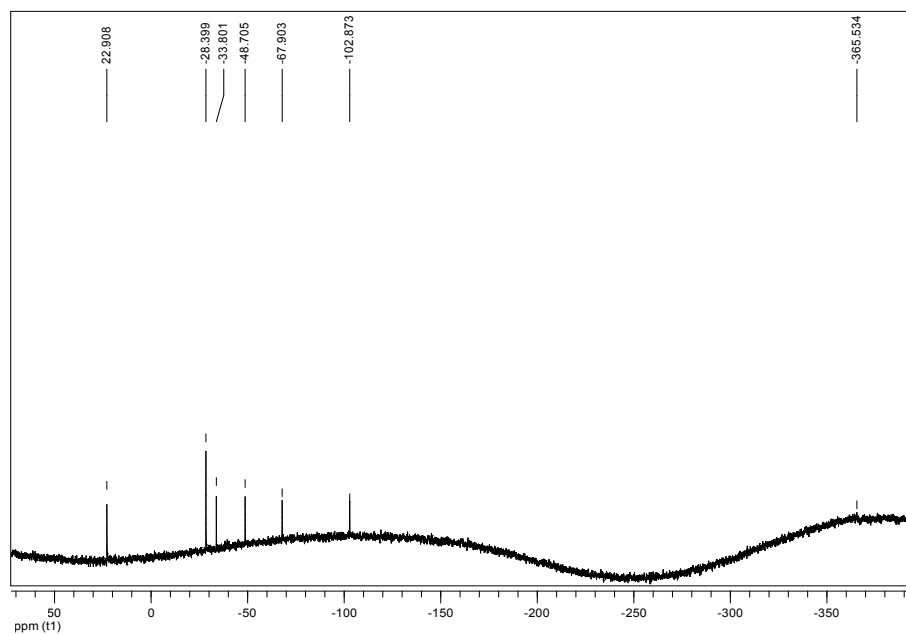
IR spectrum of **5a**



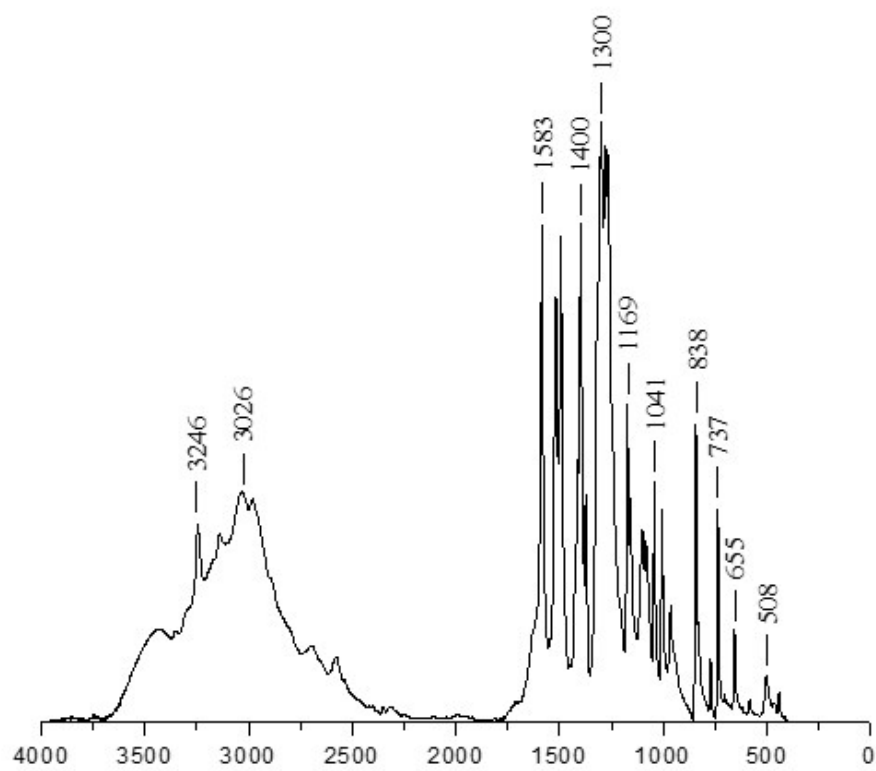
¹H NMR spectrum (400 MHz) of **5a** in DMSO-d₆ at 25 °C



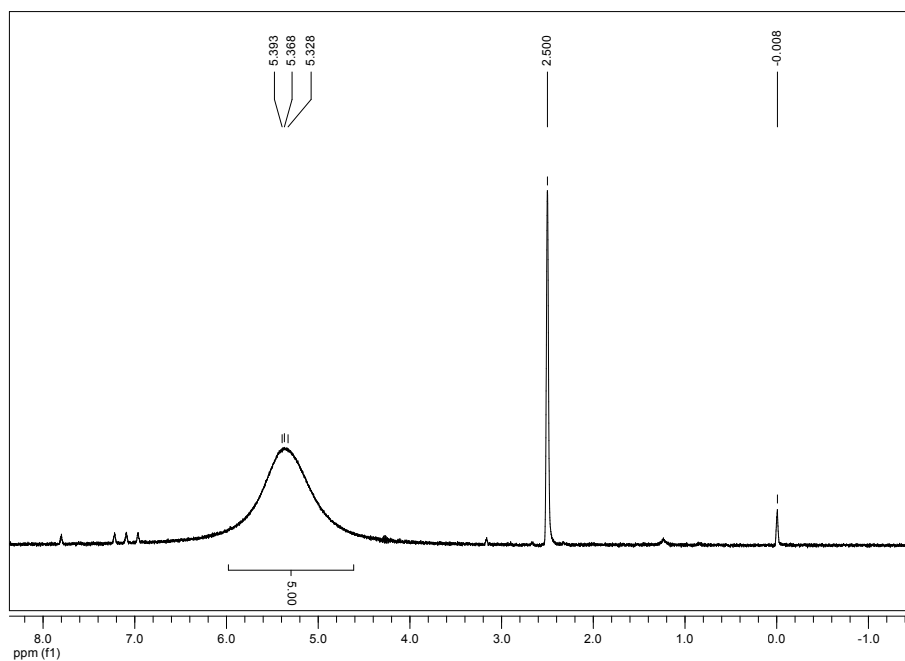
¹³C NMR spectrum (400 MHz) of **5a** in methanol-d₄ at 25 °C



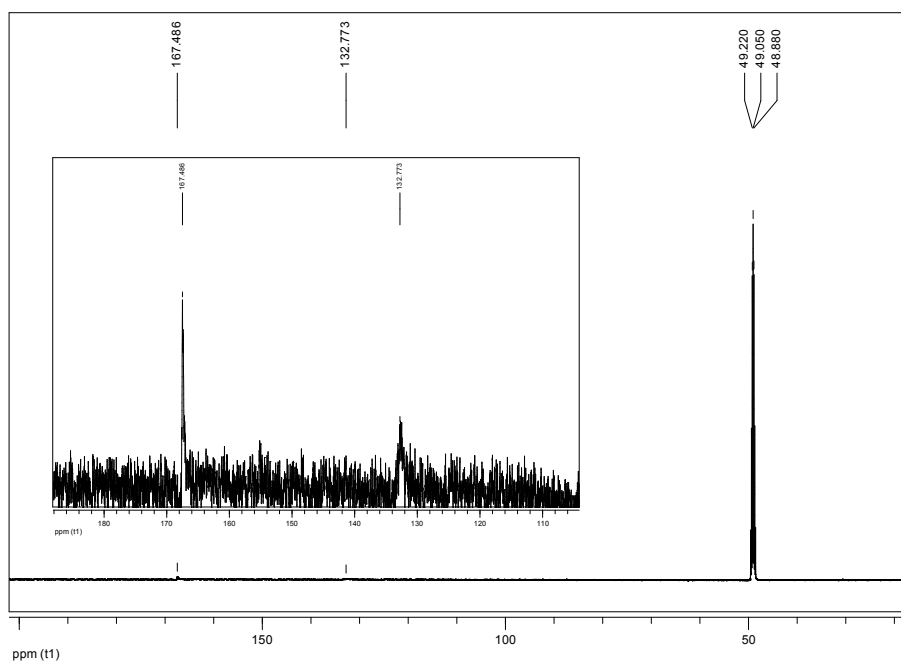
^{15}N NMR spectrum of **5a** in methanol- d_4 at 25 °C



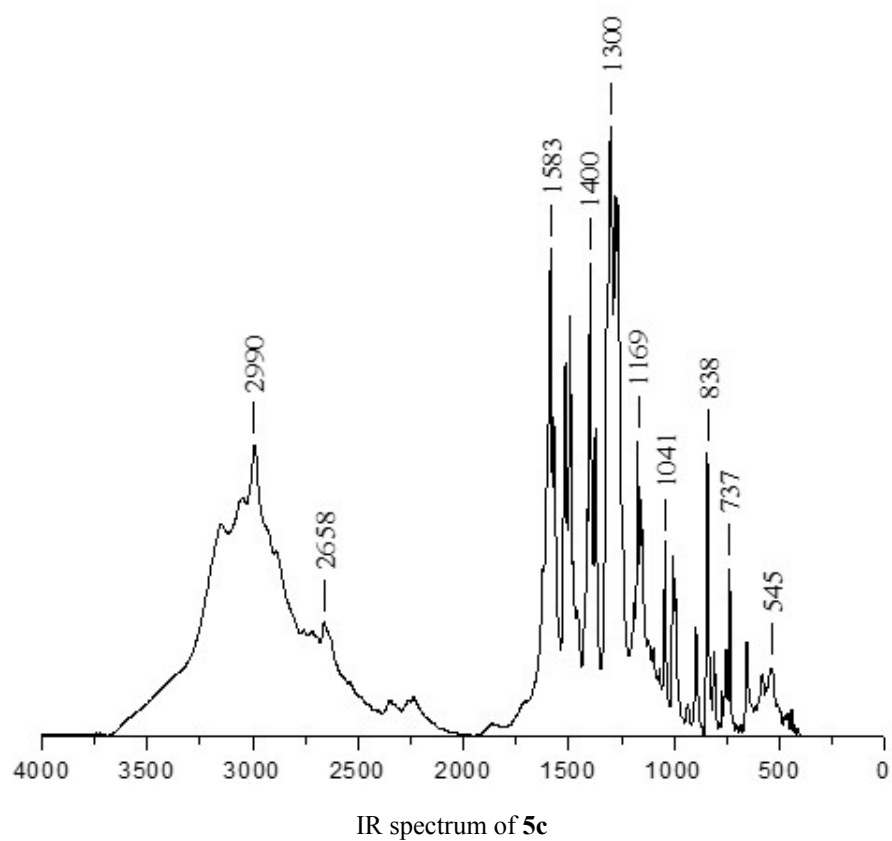
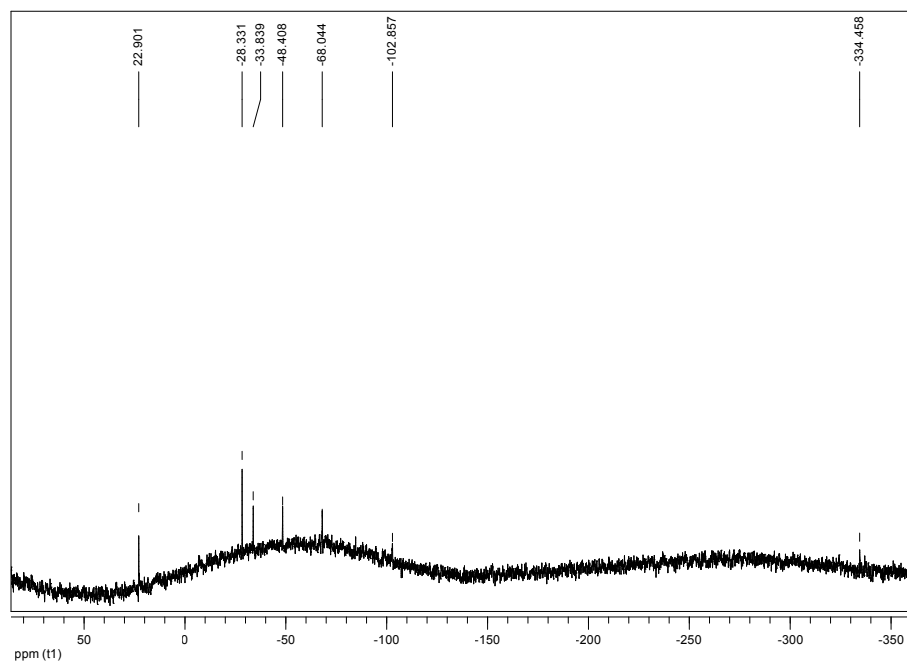
IR spectrum of **5b**

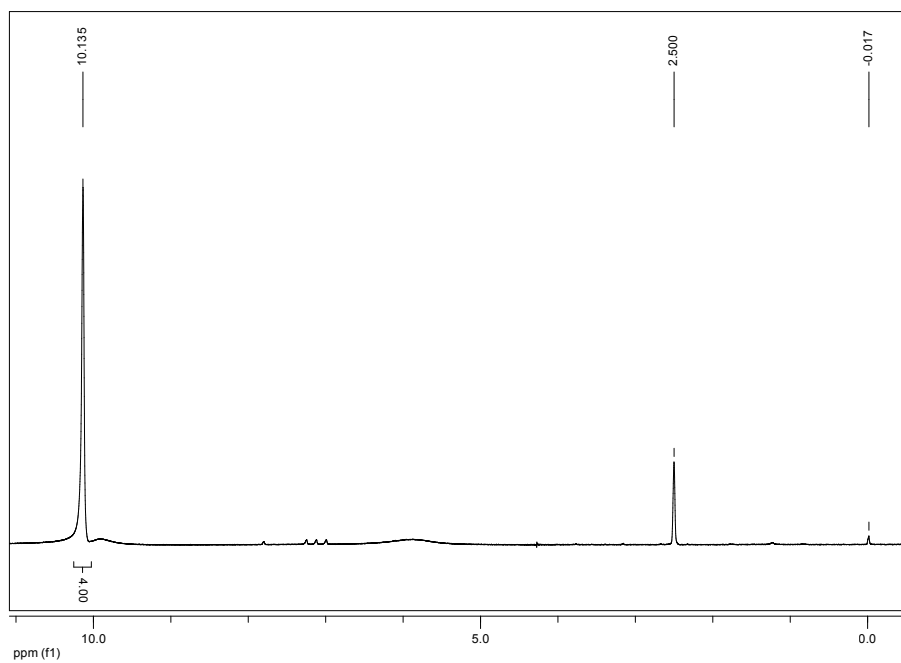


¹H NMR spectrum (400 MHz) of **5b** in DMSO-d₆ at 25 °C

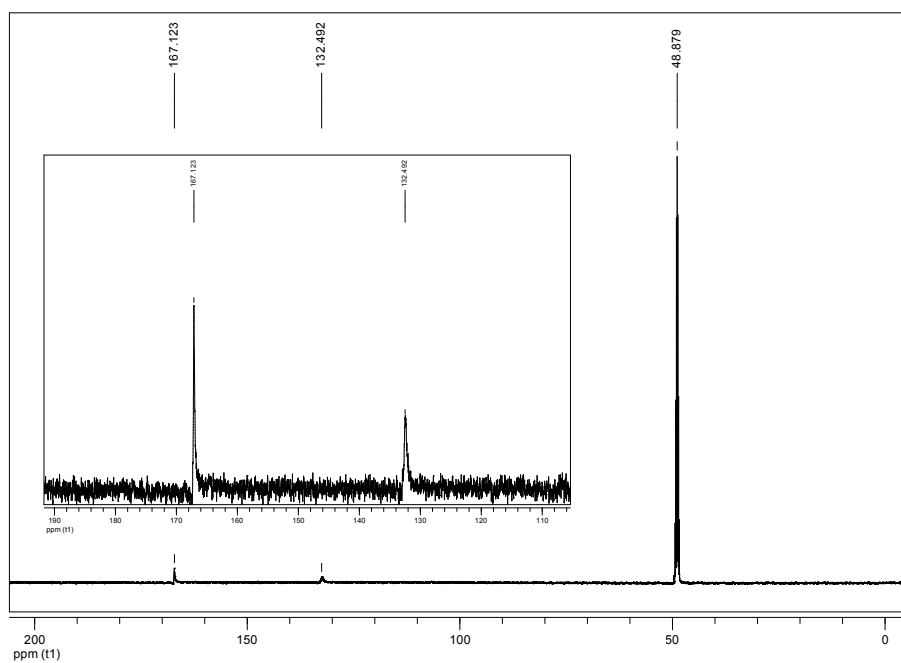


¹³C NMR spectrum (400 MHz) of **5b** in methanol-d₄ at 25 °C

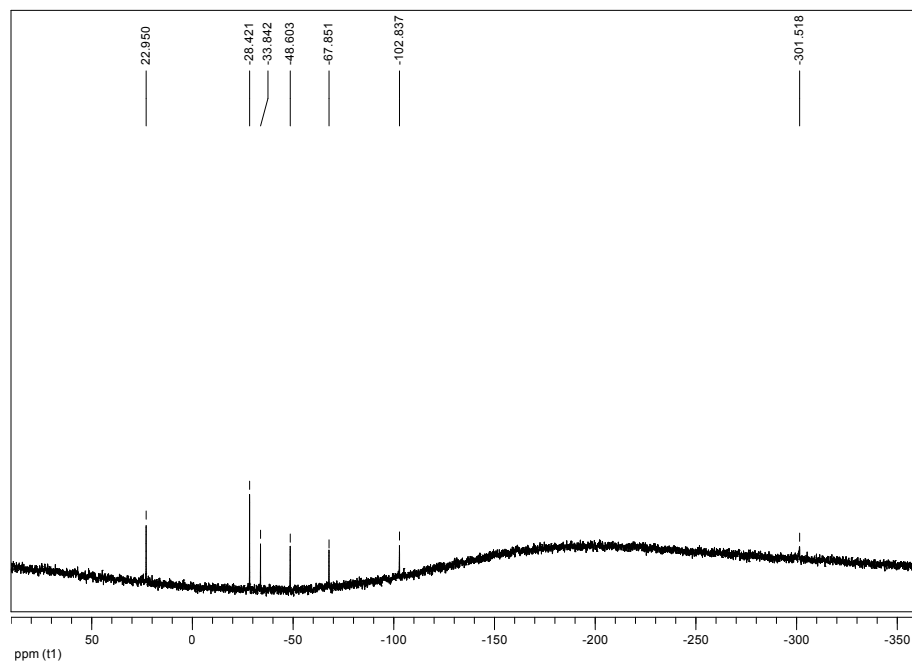




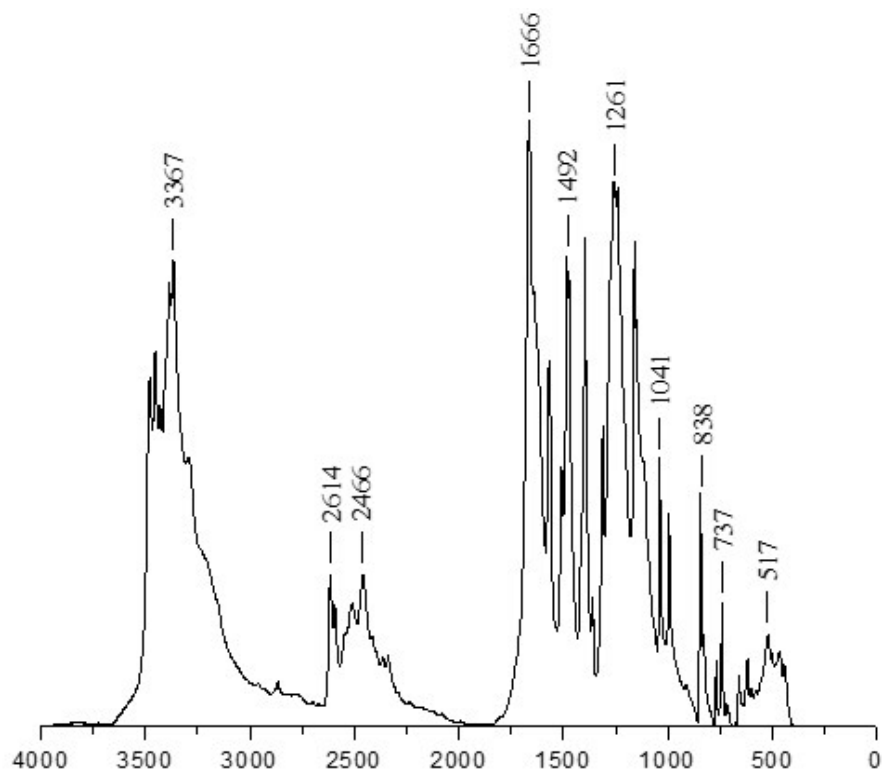
¹H NMR spectrum (400 MHz) of **5c** in DMSO-d₆ at 25 °C



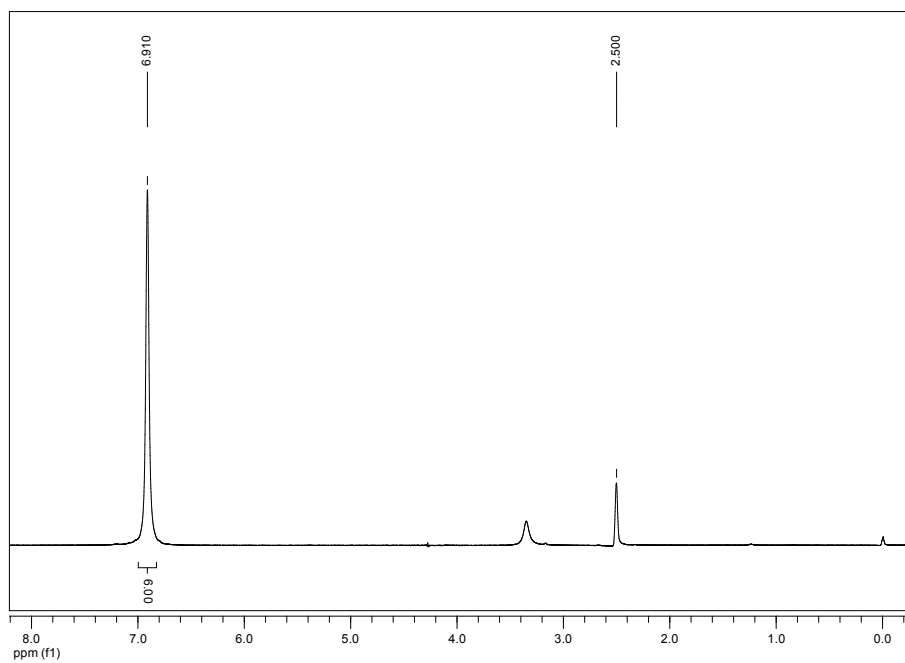
¹³C NMR spectrum (400 MHz) of **5c** in methanol-d₄ at 25 °C



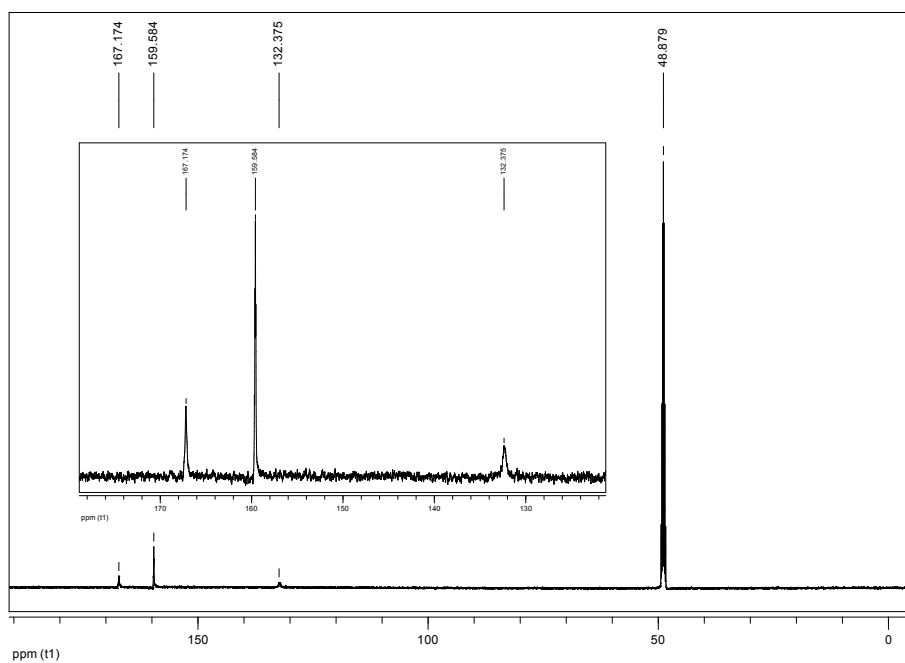
¹⁵N NMR spectrum of **5c** in methanol-d₄ at 25 °C



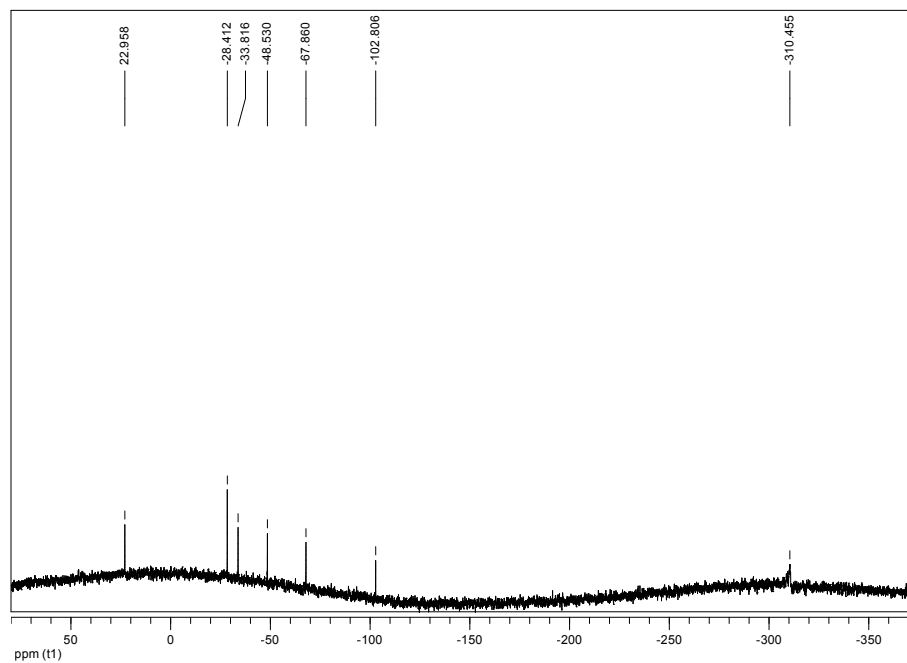
IR spectrum of **5d**



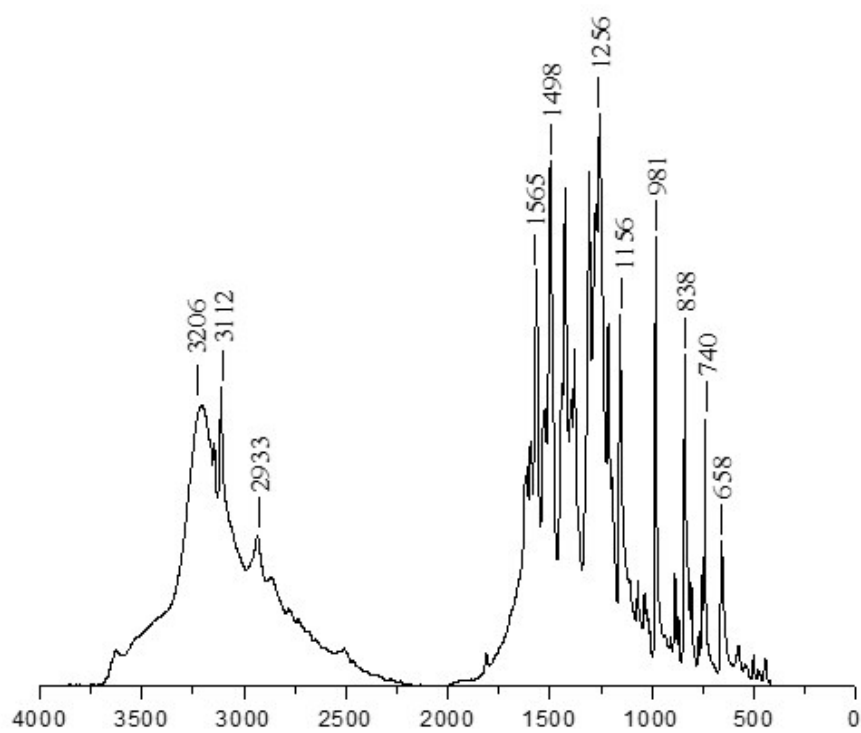
¹H NMR spectrum (400 MHz) of **5d** in DMSO-d₆ at 25 °C



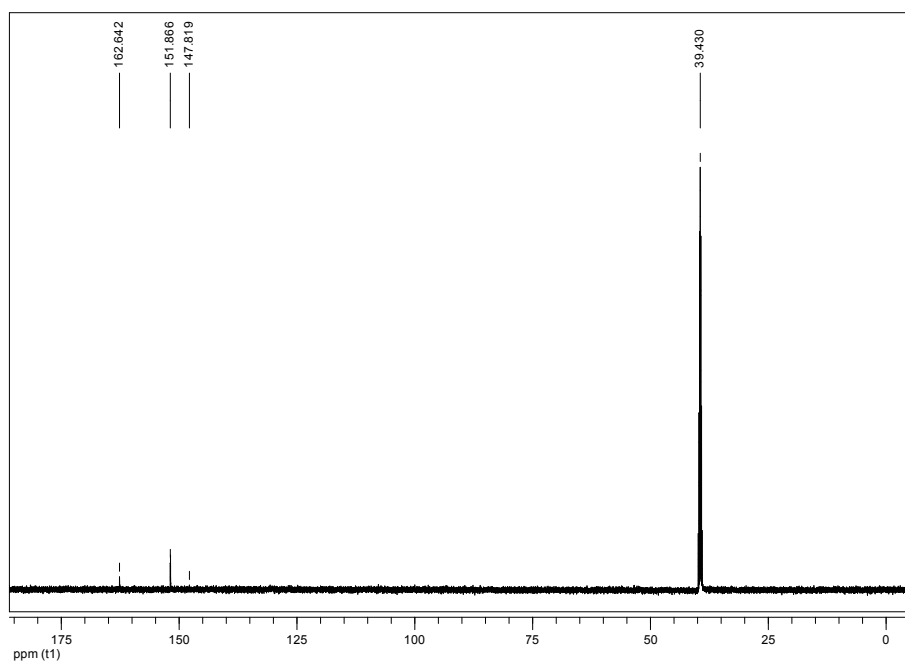
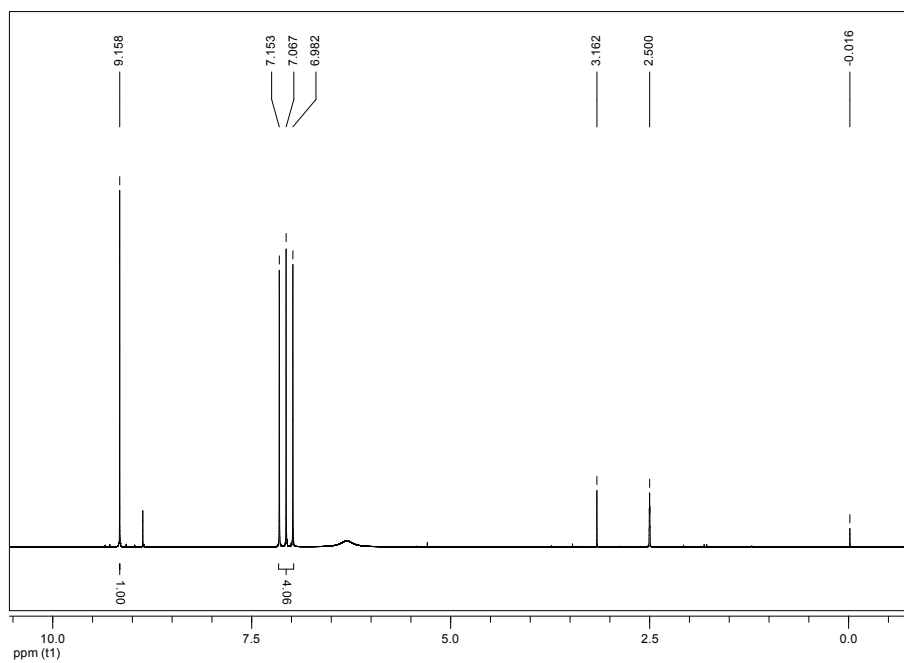
¹³C NMR spectrum (400 MHz) of **5d** in methanol-d₄ at 25 °C

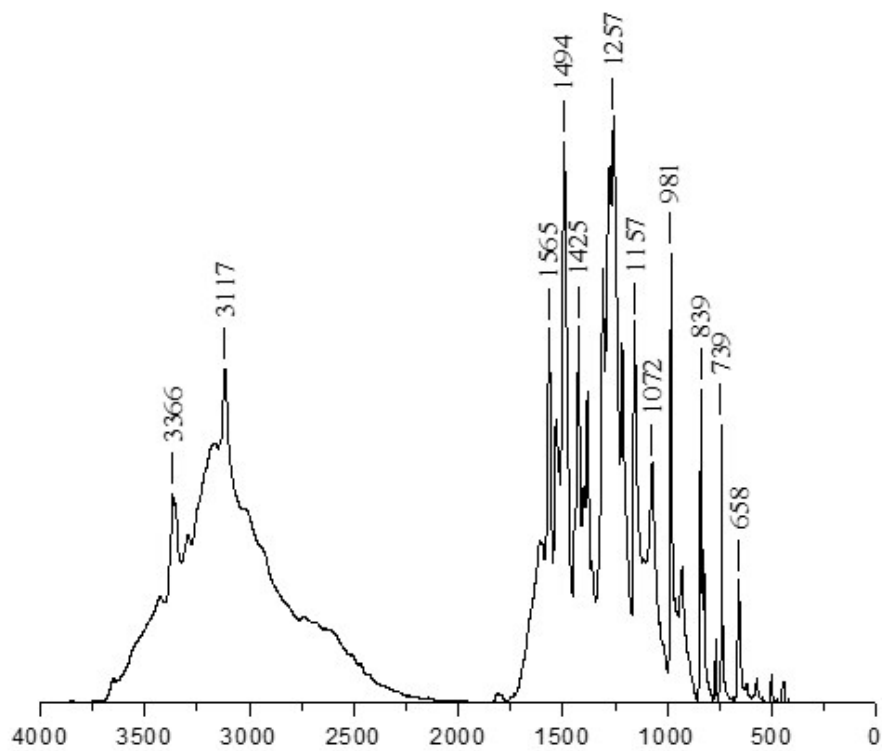


^{15}N NMR spectrum of **5d** in methanol- d_4 at 25 °C

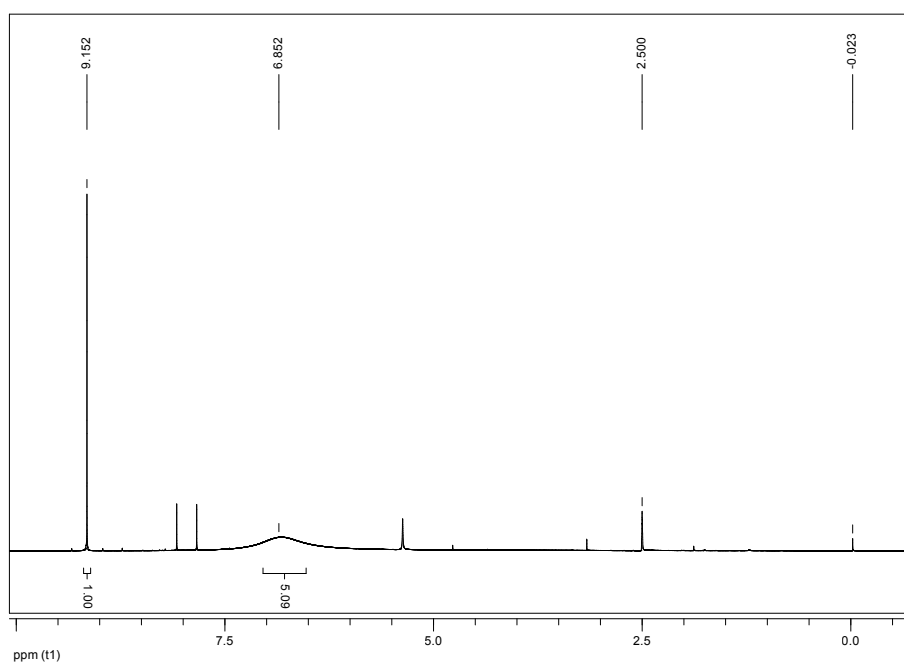


IR spectrum of **6a**

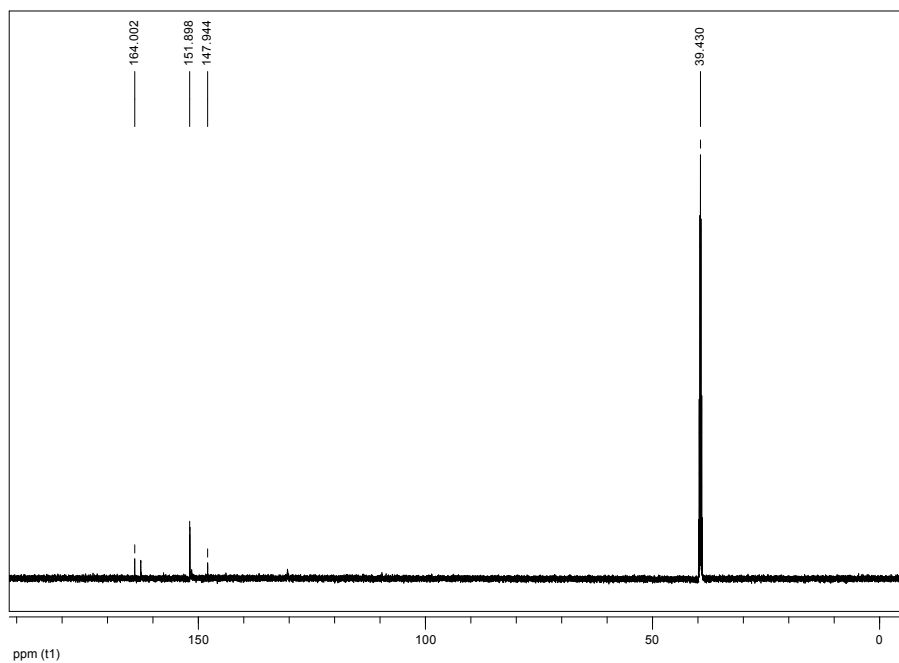




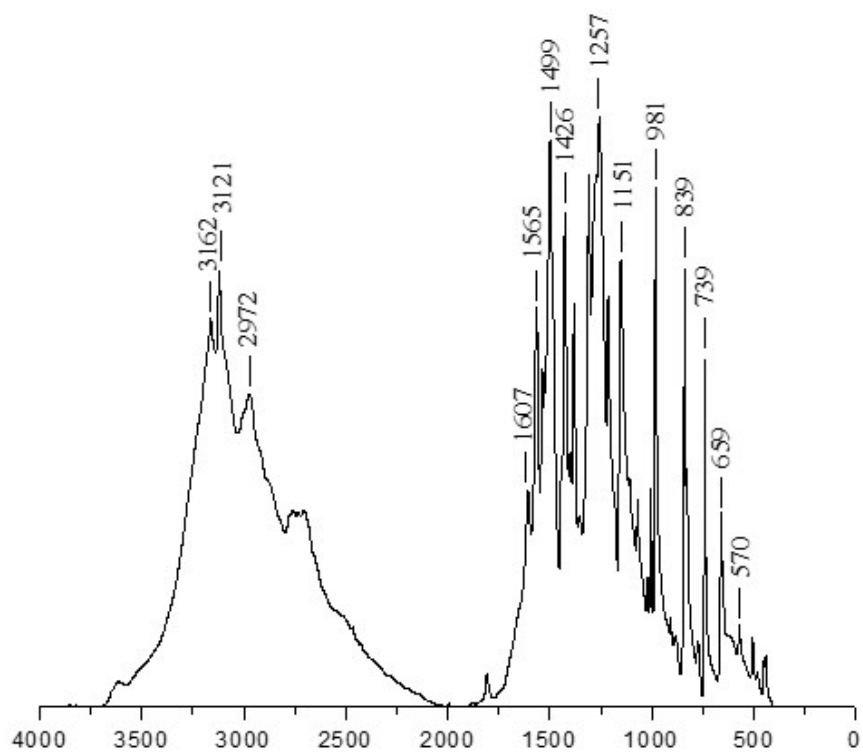
IR spectrum of **6b**



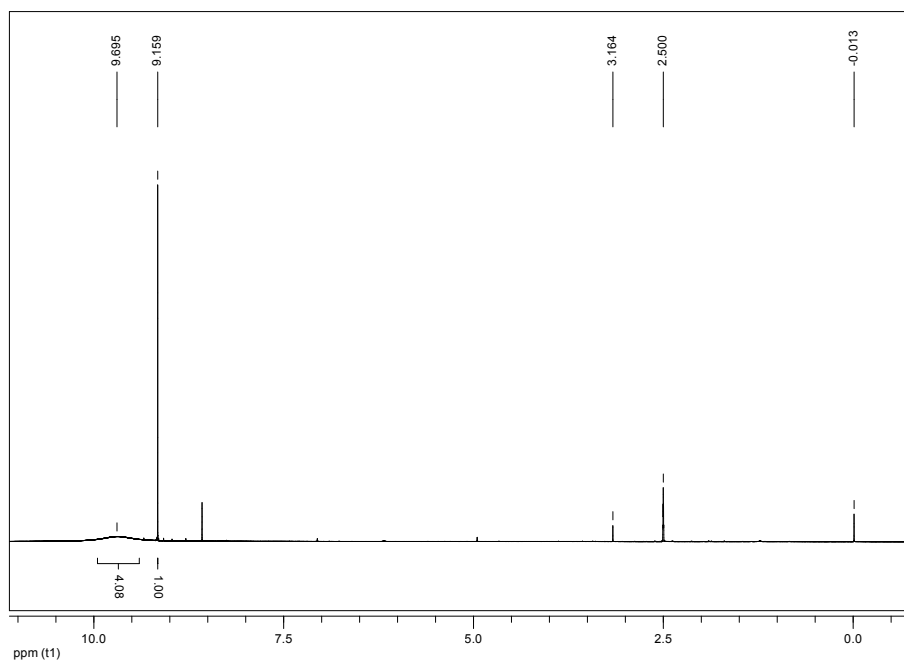
¹H NMR spectrum (400 MHz) of **6b** in DMSO-d₆ at 25 °C



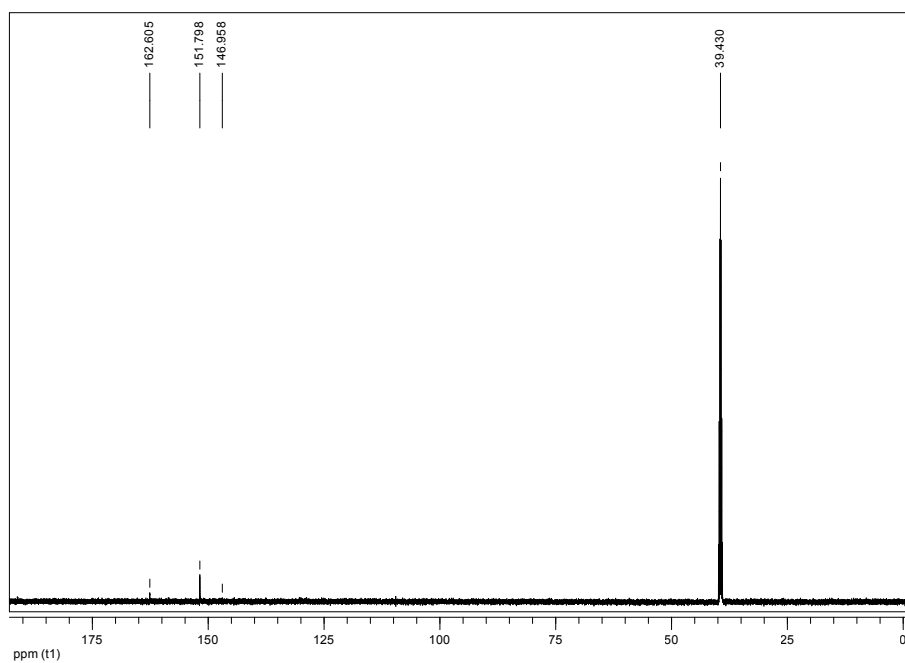
^{13}C NMR spectrum (400 MHz) of **6b** in DMSO- d_6 at 25 °C



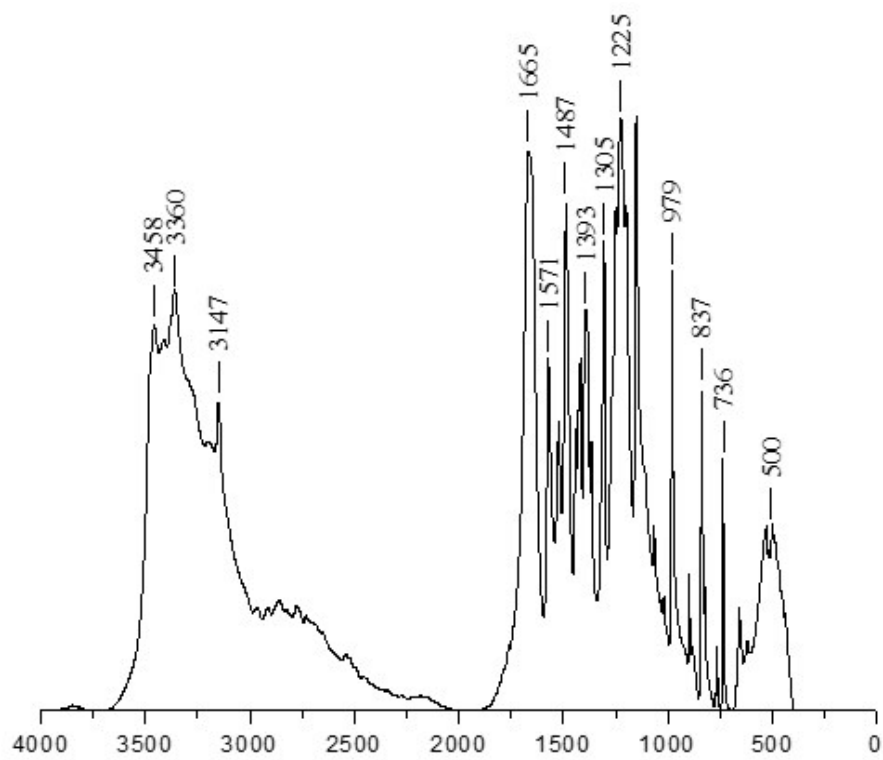
IR spectrum of **6c**



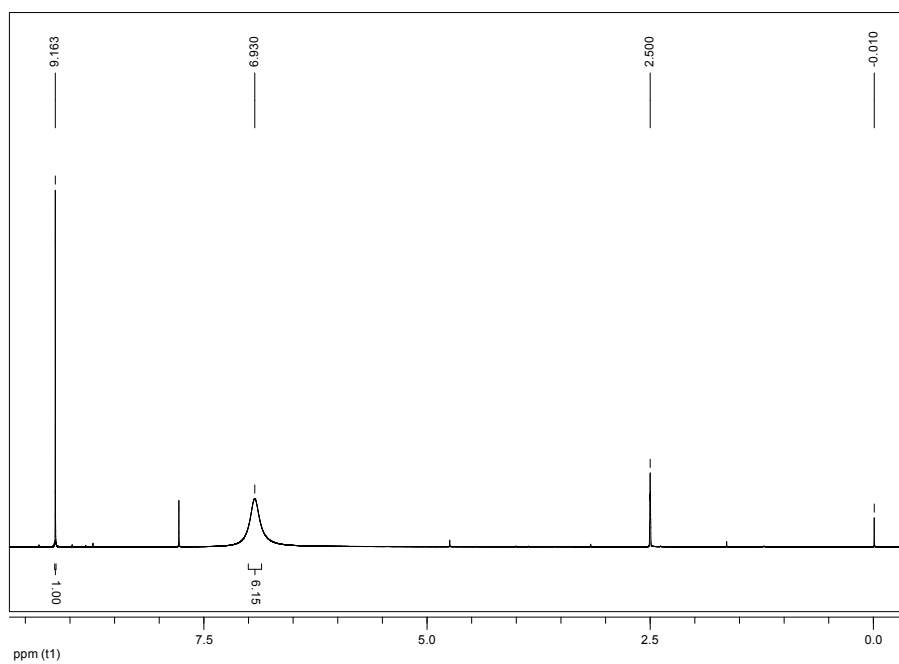
¹H NMR spectrum (400 MHz) of **6c** in DMSO-d₆ at 25 °C



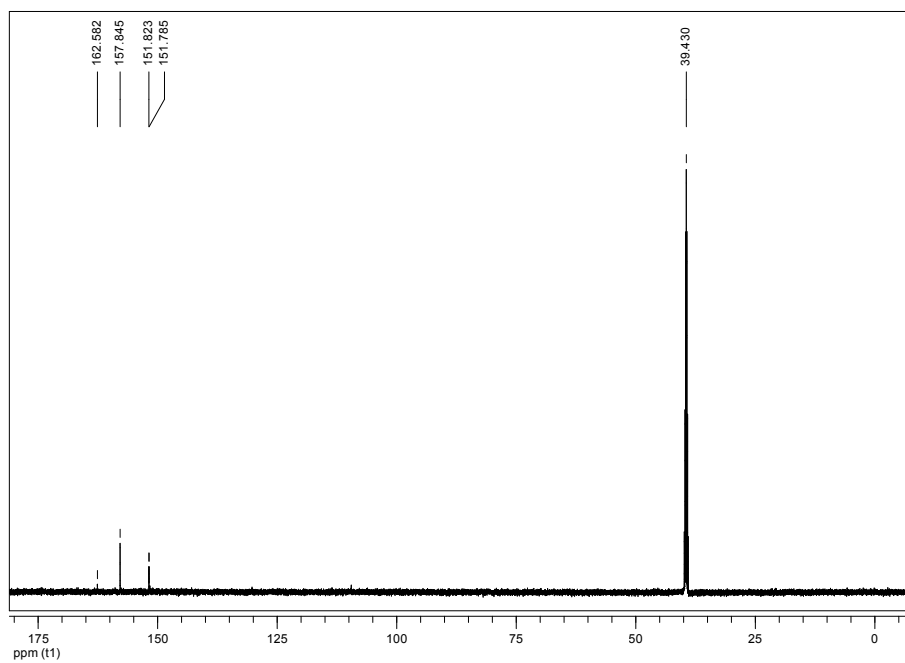
¹³C NMR spectrum (400 MHz) of **6c** in DMSO-d₆ at 25 °C



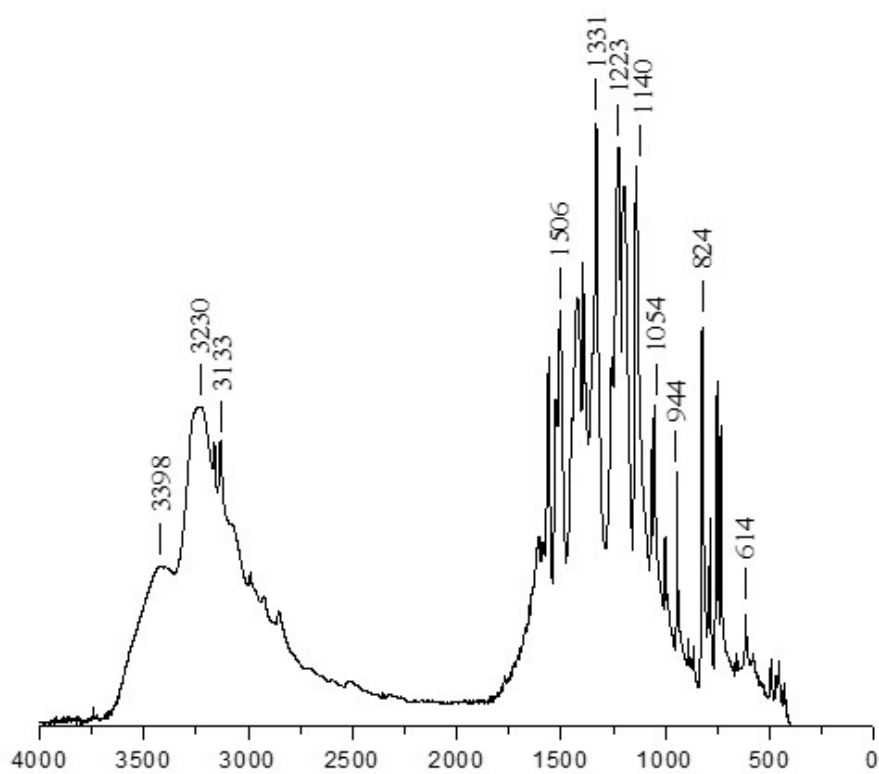
IR spectrum of **6d**



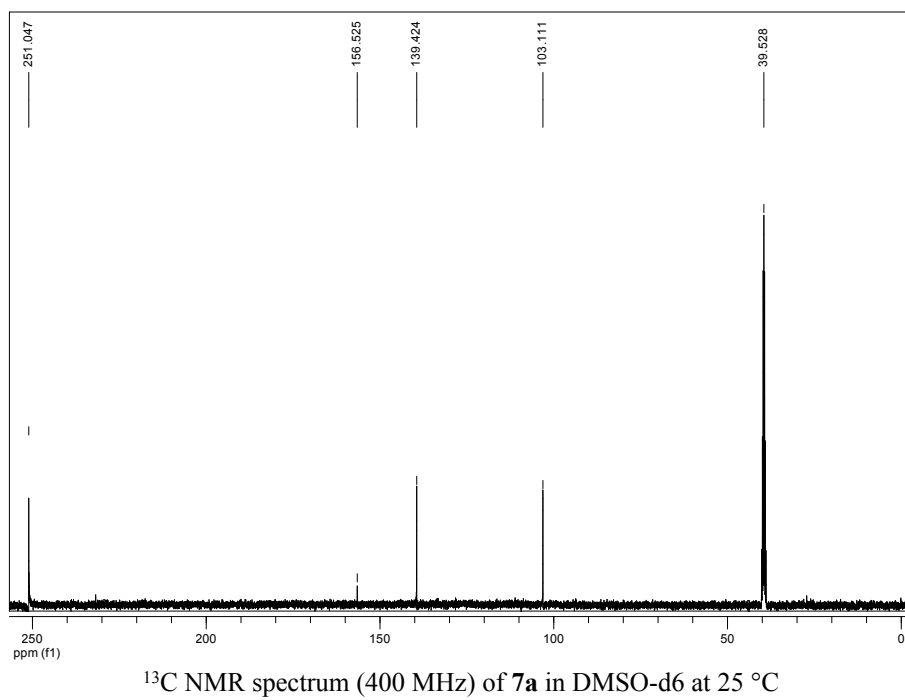
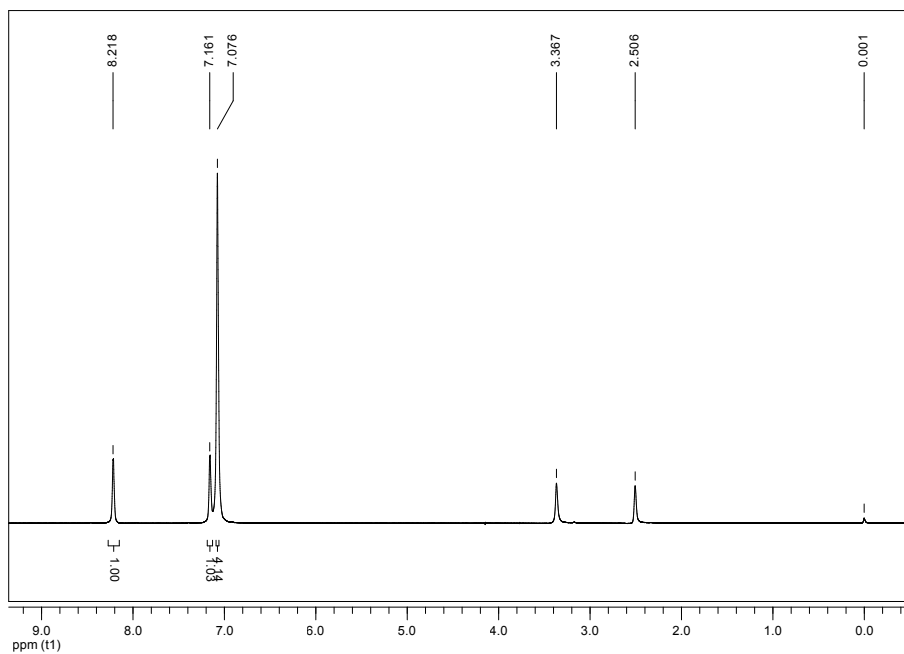
¹H NMR spectrum (400 MHz) of **6d** in DMSO-d₆ at 25 °C

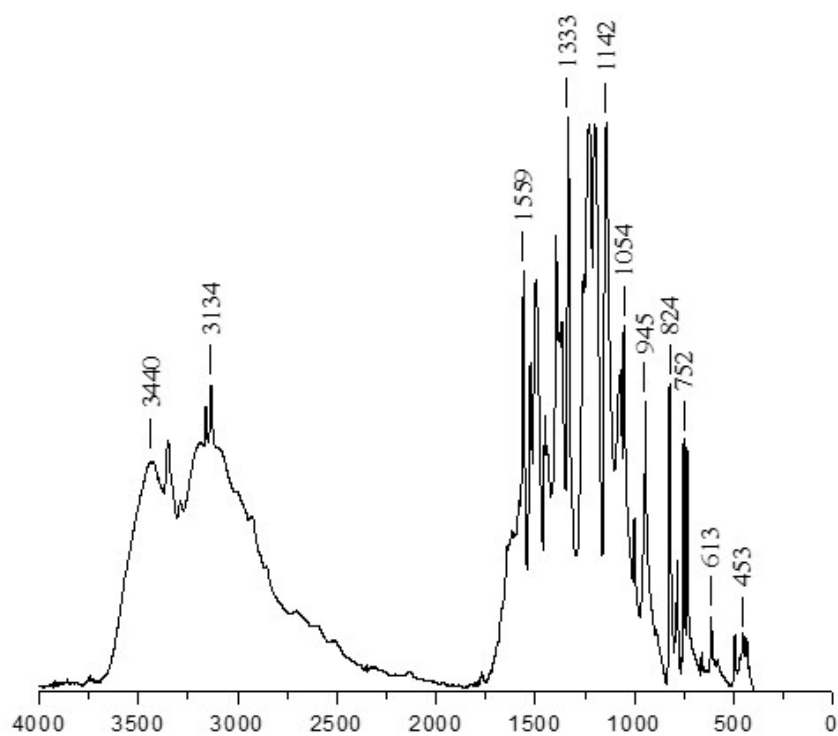


¹³C NMR spectrum (400 MHz) of **6d** in DMSO-d₆ at 25 °C

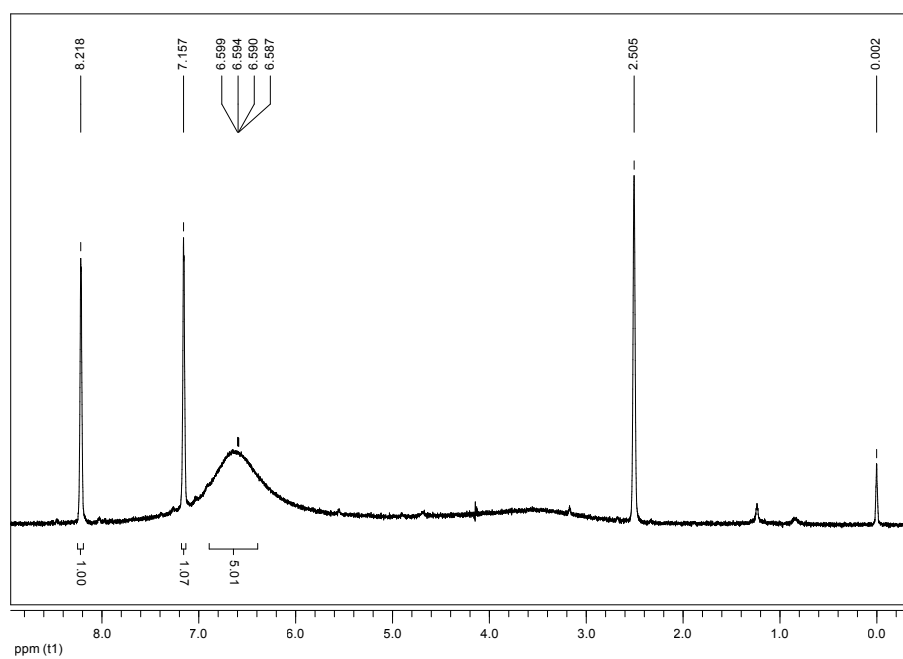


IR spectrum of **7a**

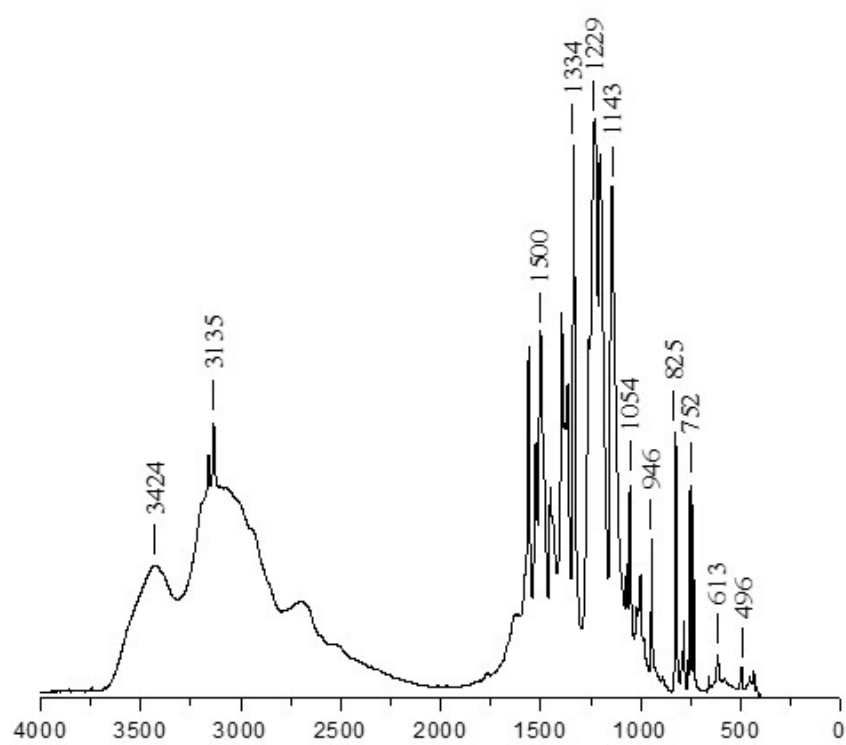
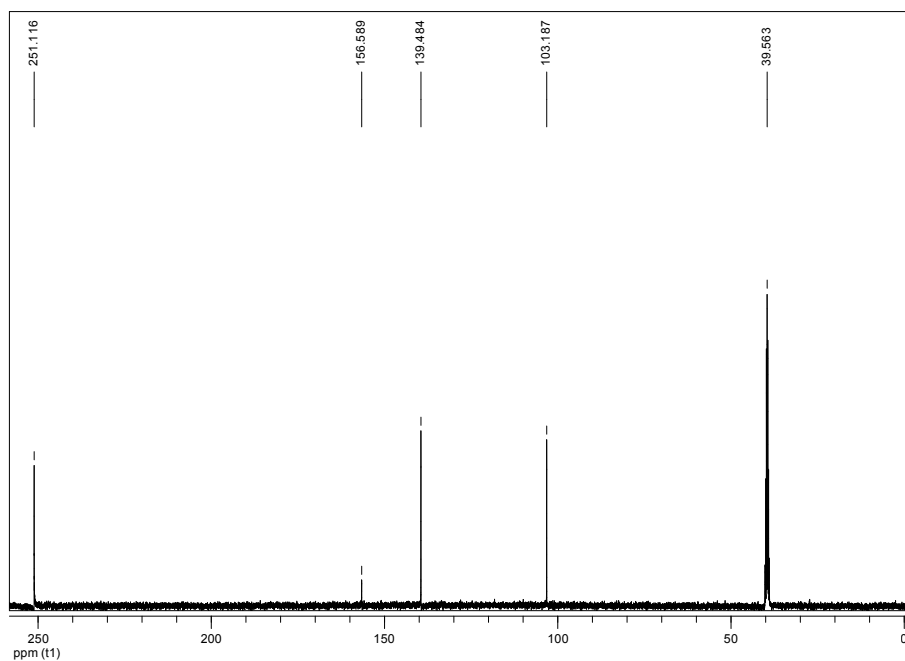


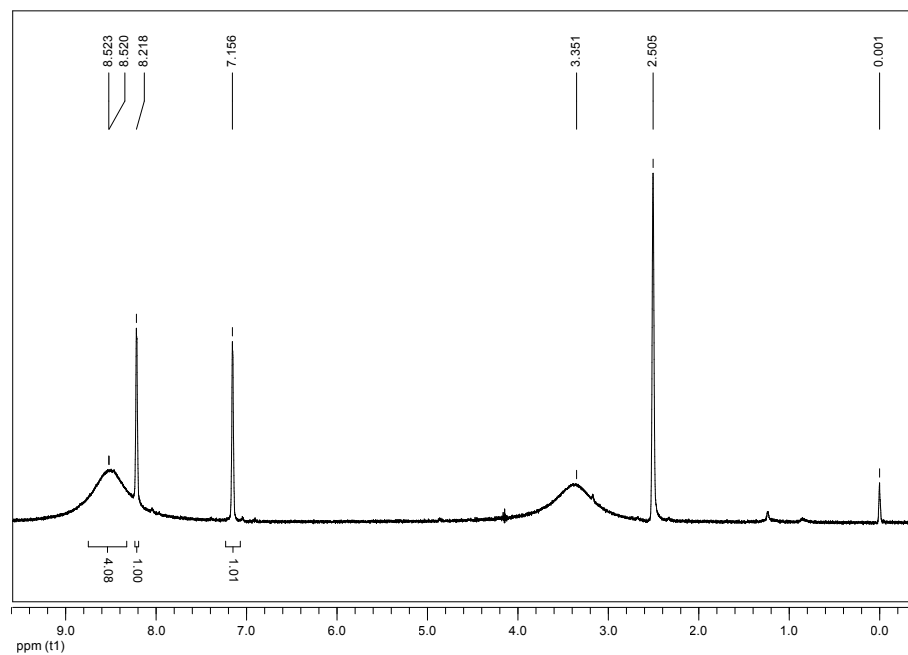


IR spectrum of **7b**

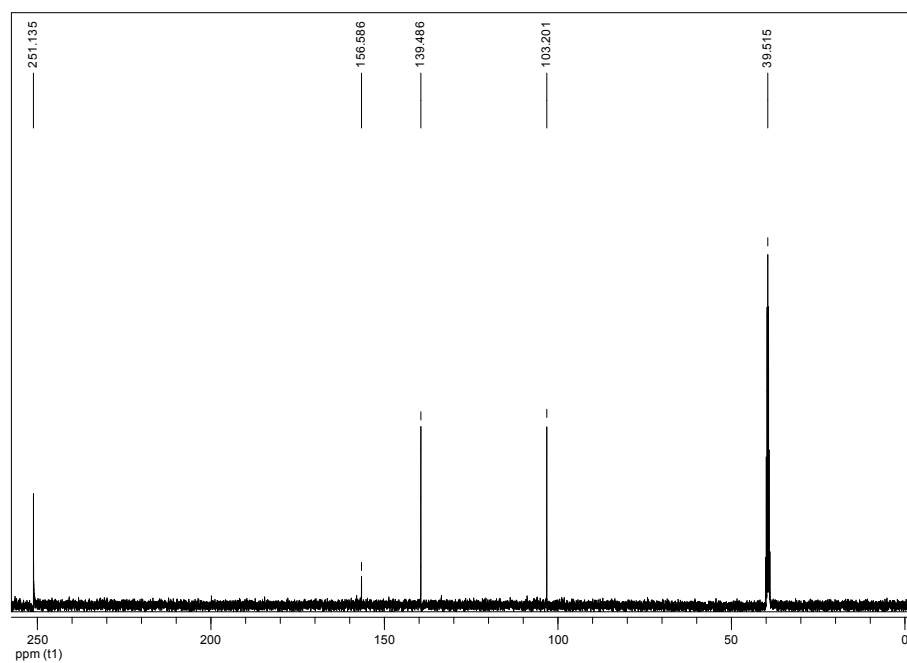


¹H NMR spectrum (400 MHz) of **7b** in DMSO-d₆ at 25 °C

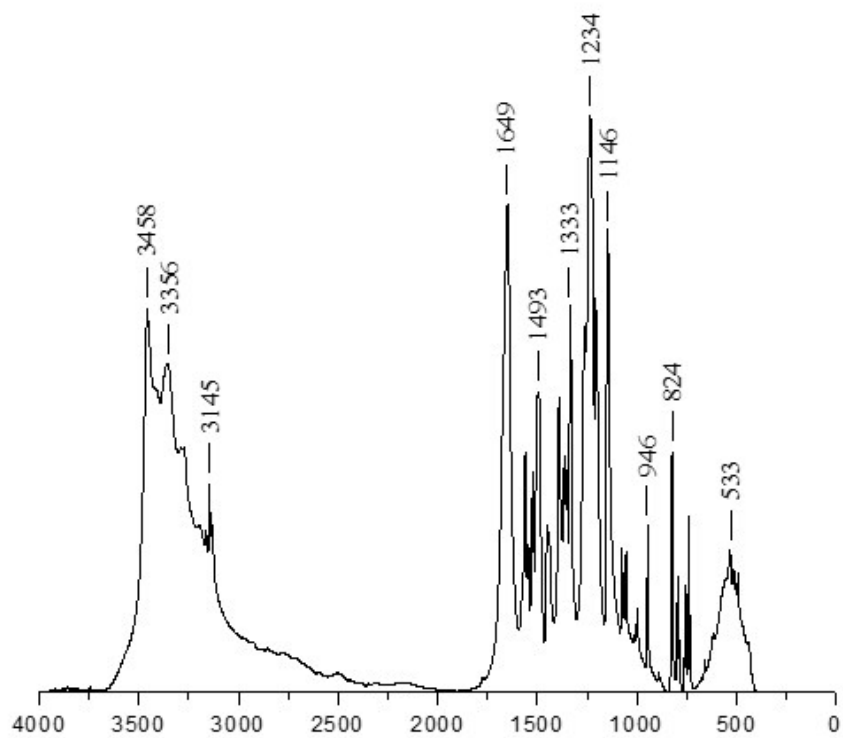




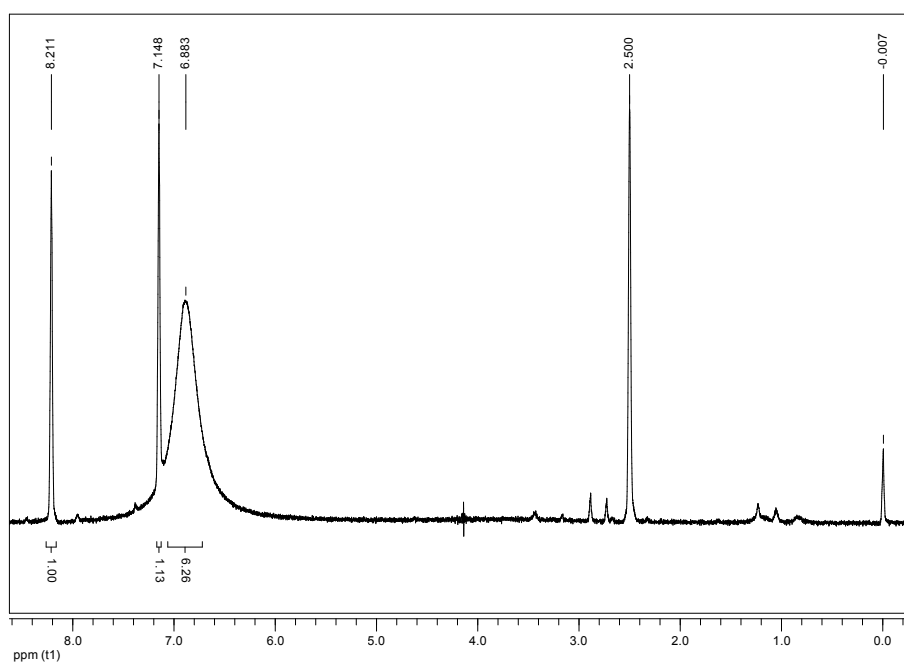
¹H NMR spectrum (400 MHz) of **7c** in DMSO-d₆ at 25 °C



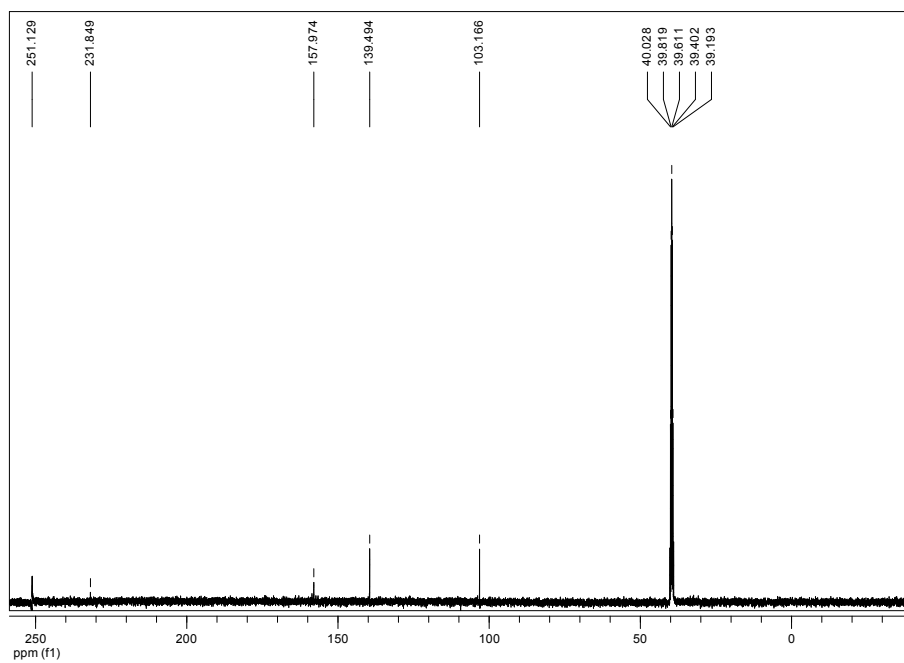
¹³C NMR spectrum (400 MHz) of **7c** in DMSO-d₆ at 25 °C



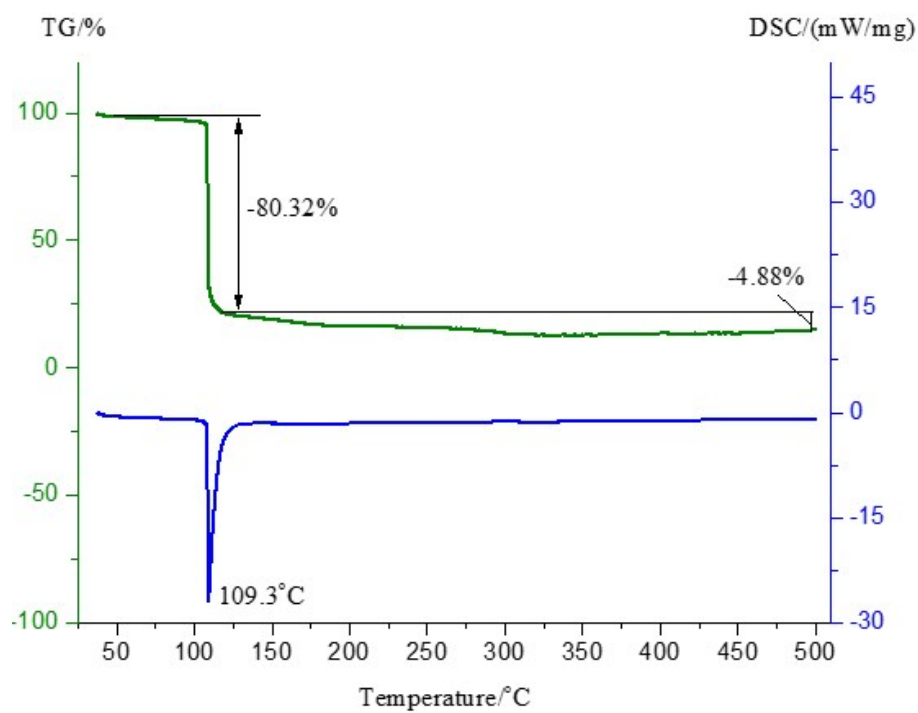
IR spectrum of **7d**



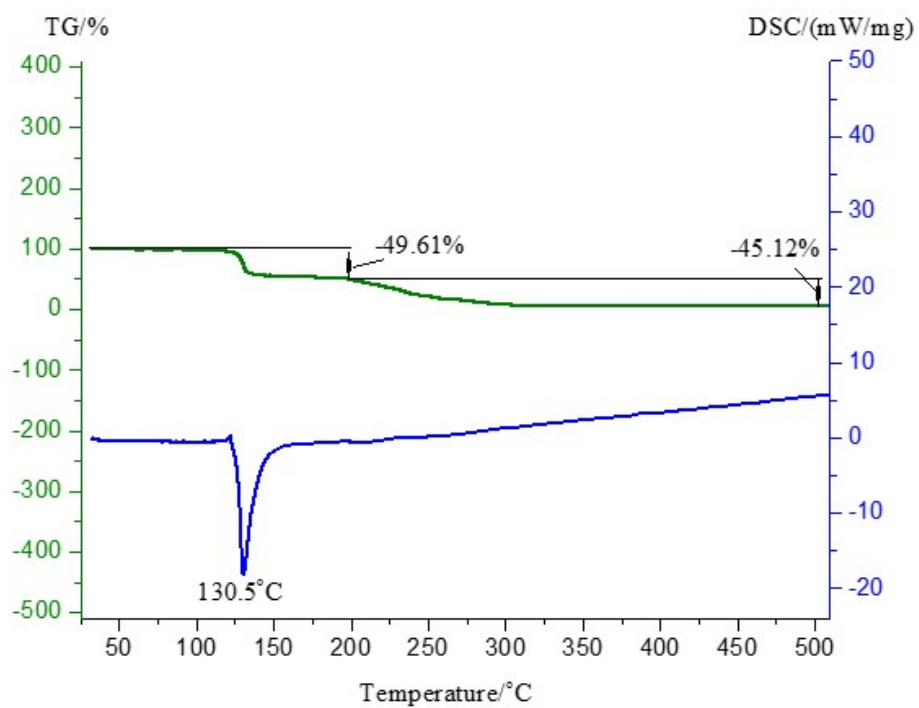
¹H NMR spectrum (400 MHz) of **7d** in DMSO-d₆ at 25 °C



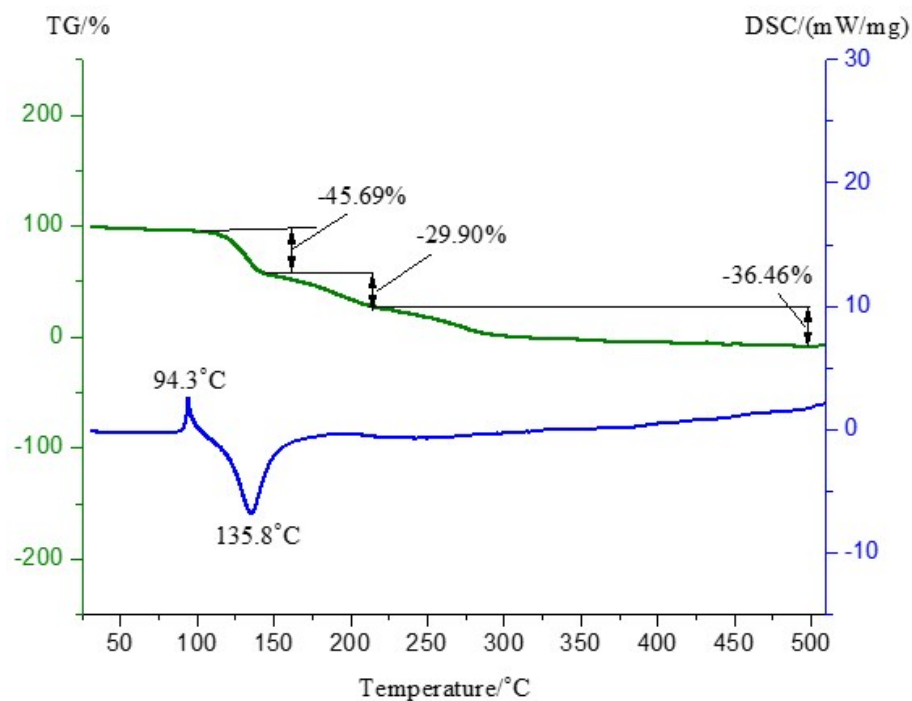
^{13}C NMR spectrum (400 MHz) of **7d** in DMSO-d_6 at $25\text{ }^\circ\text{C}$



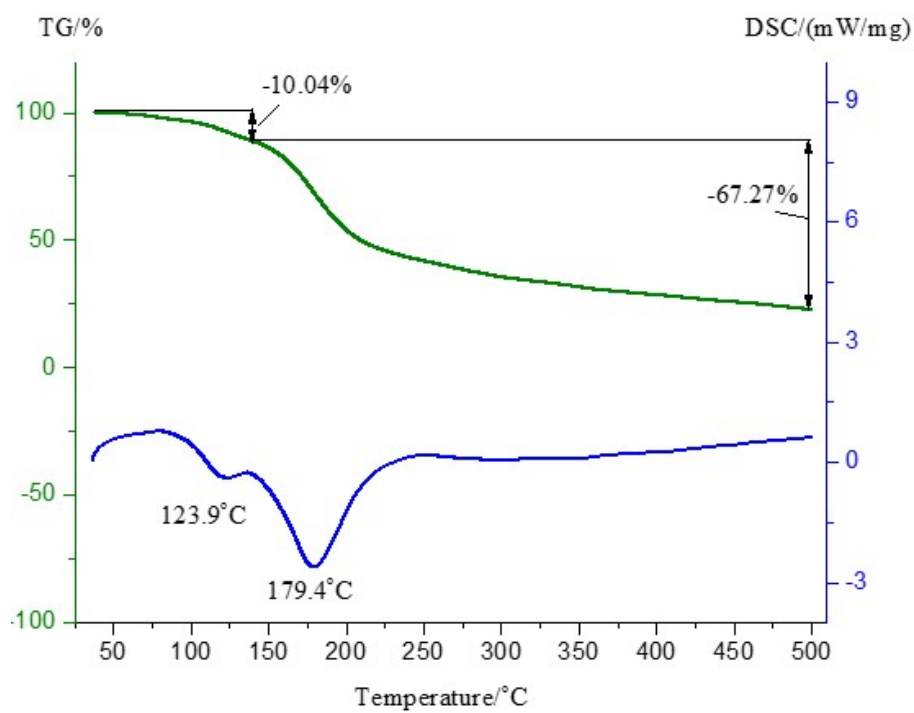
DSC spectrum of 5



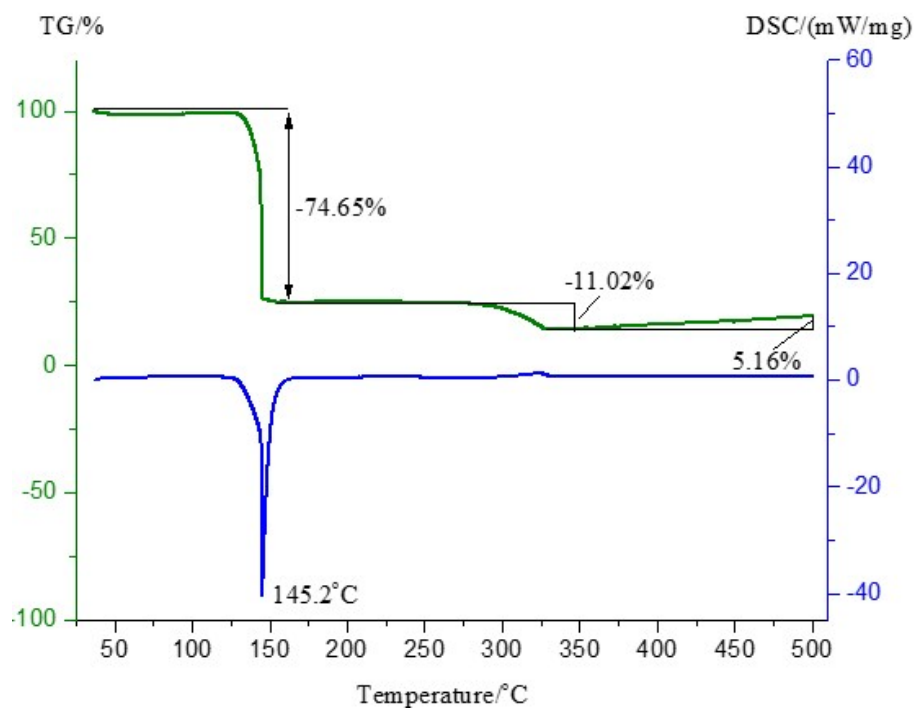
DSC spectrum of 6



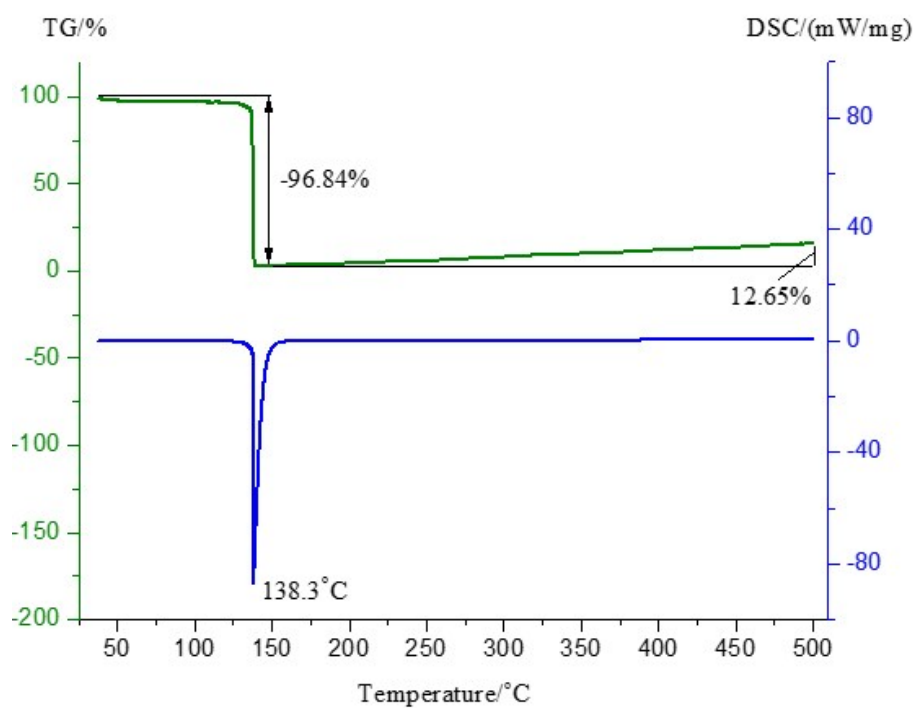
DSC spectrum of 7



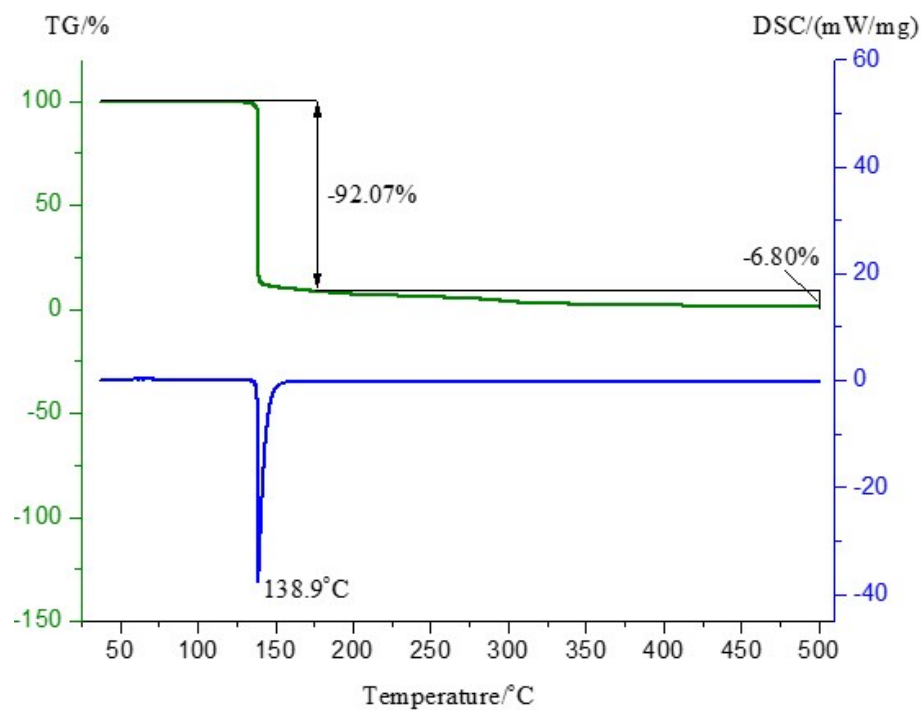
DSC spectrum of 8



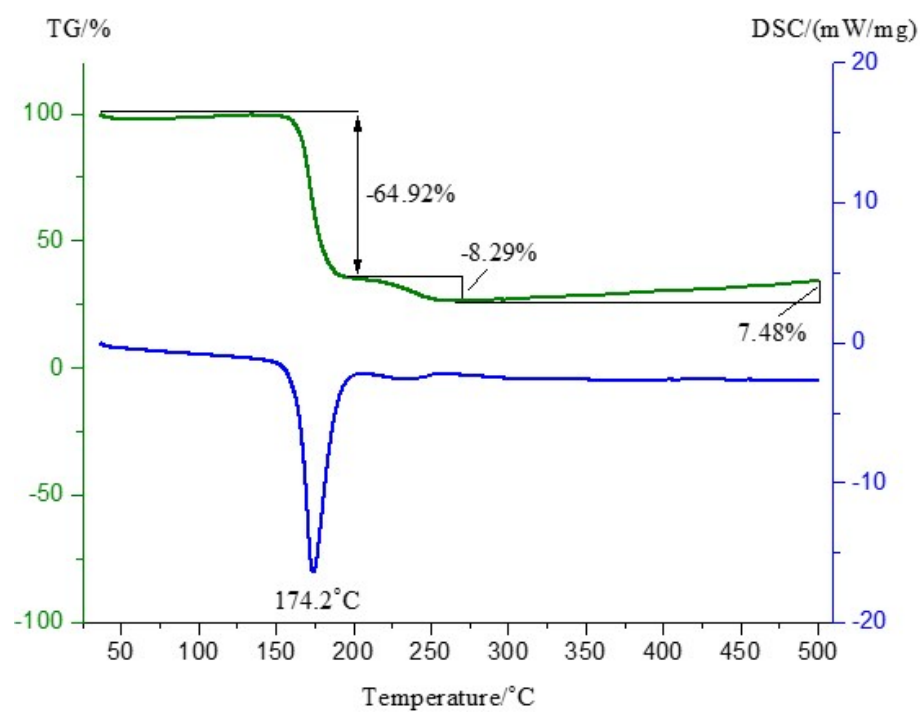
DSC spectrum of **5a**



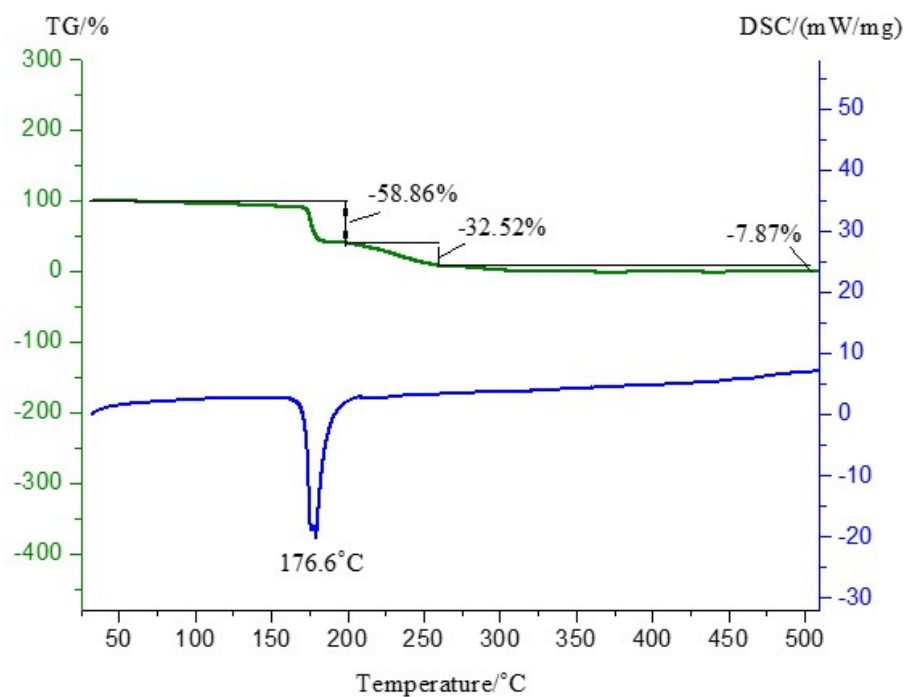
DSC spectrum of **5b**



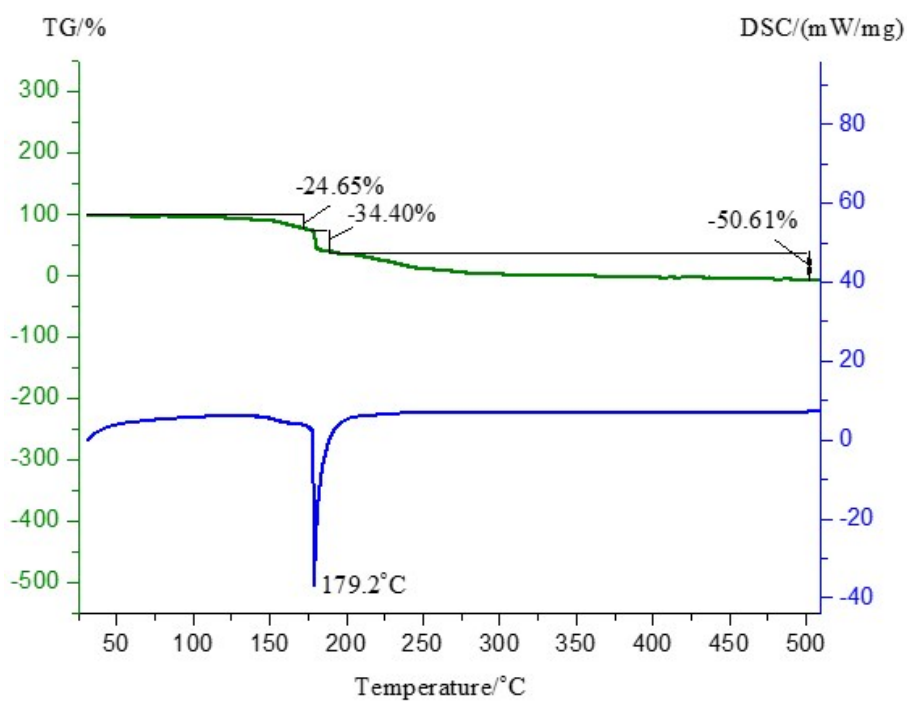
DSC spectrum of **5c**



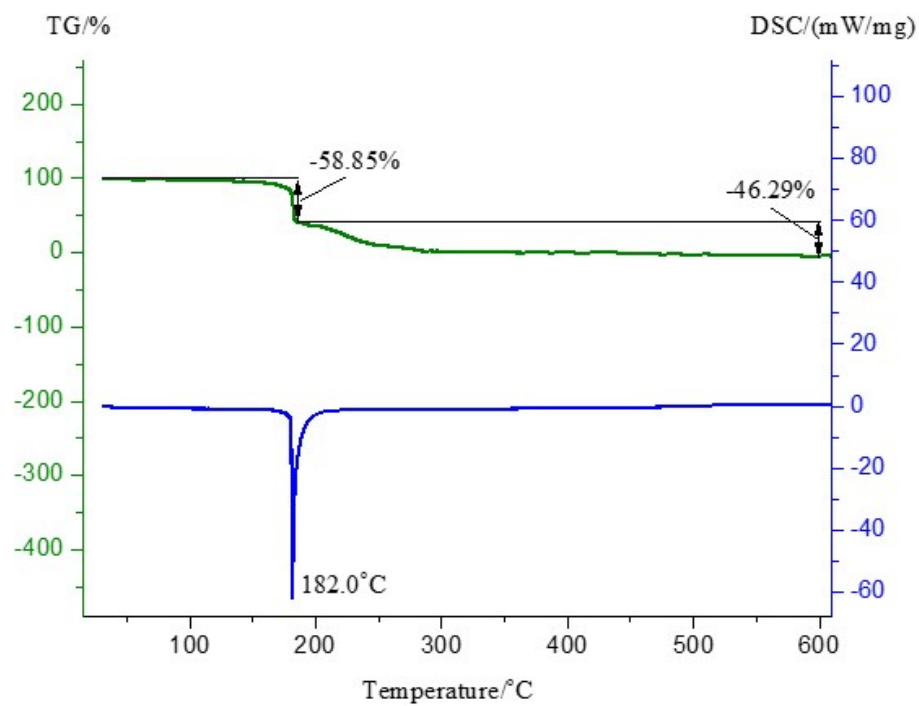
DSC spectrum of **5d**



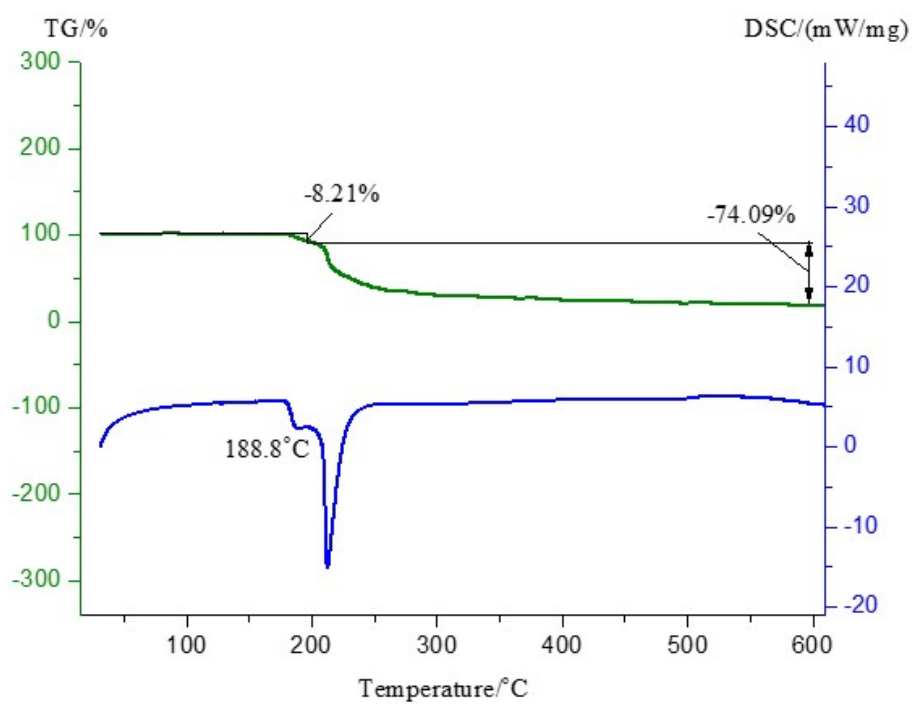
DSC spectrum of **6a**



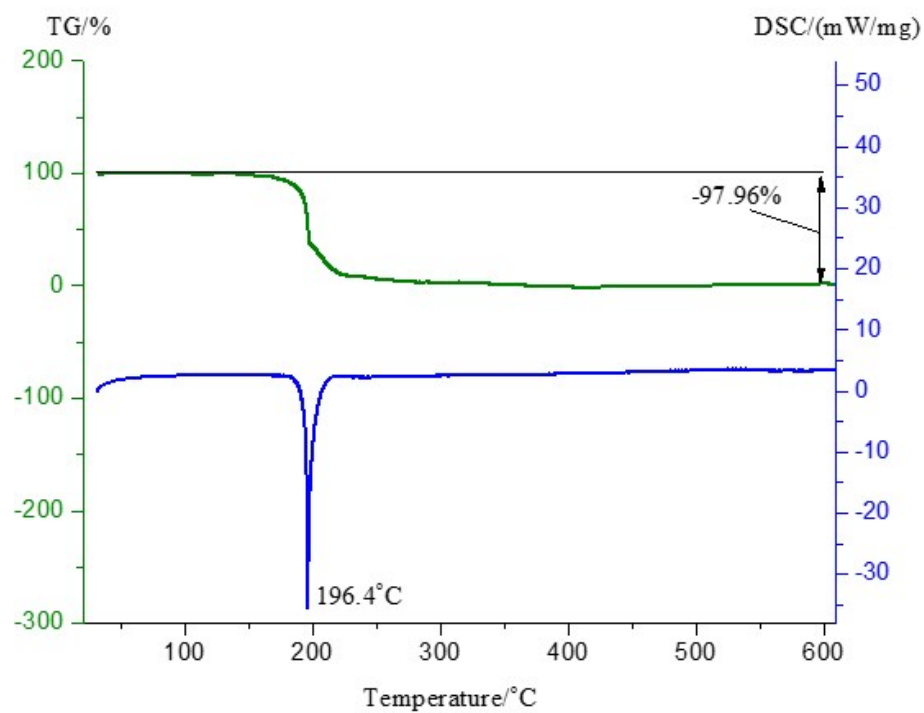
DSC spectrum of **6b**



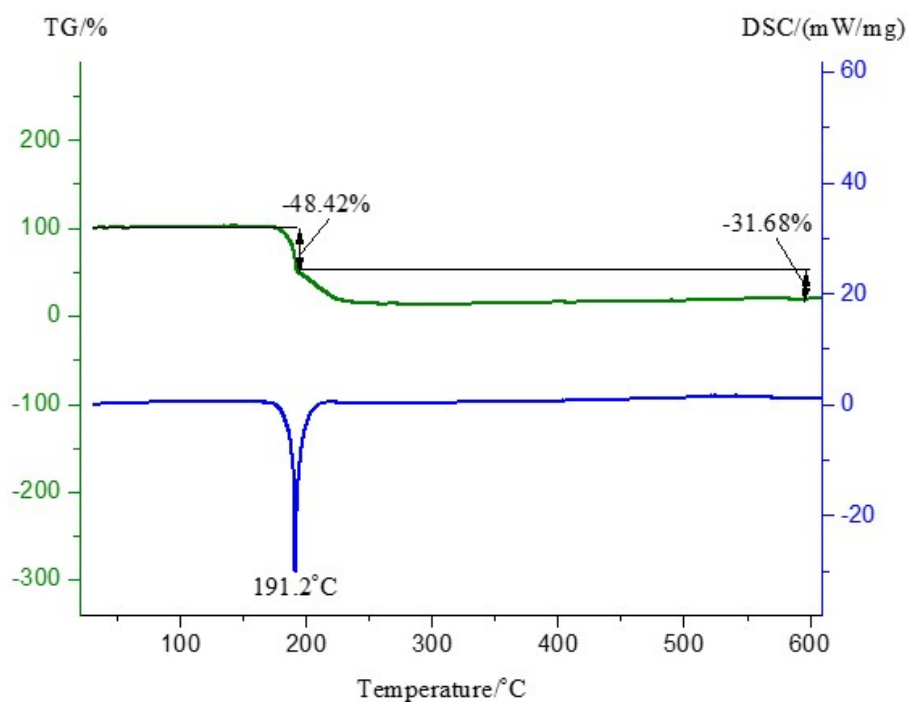
DSC spectrum of **6c**



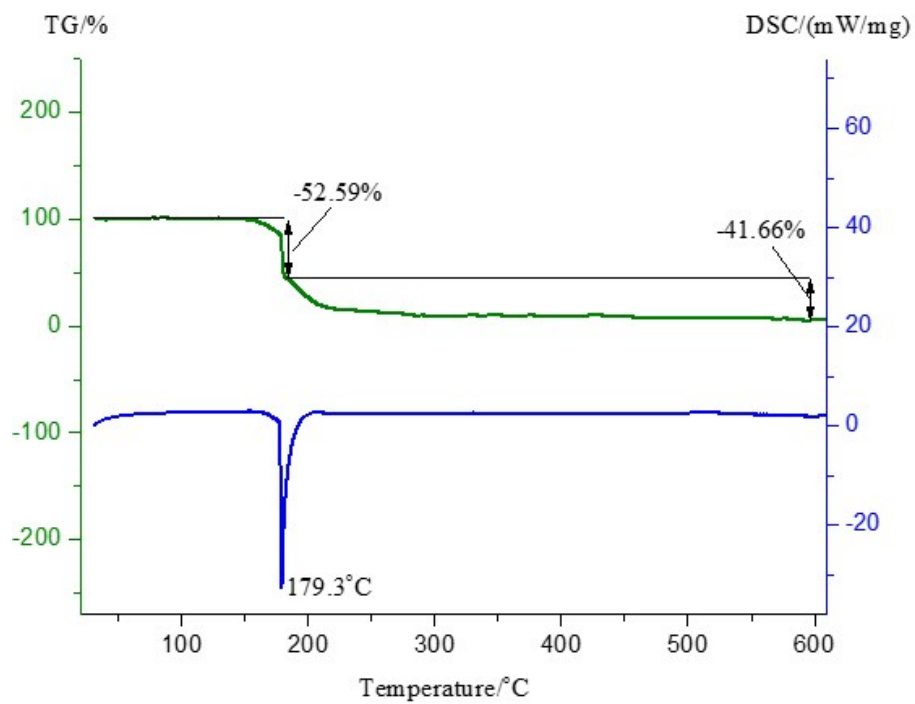
DSC spectrum of **6d**



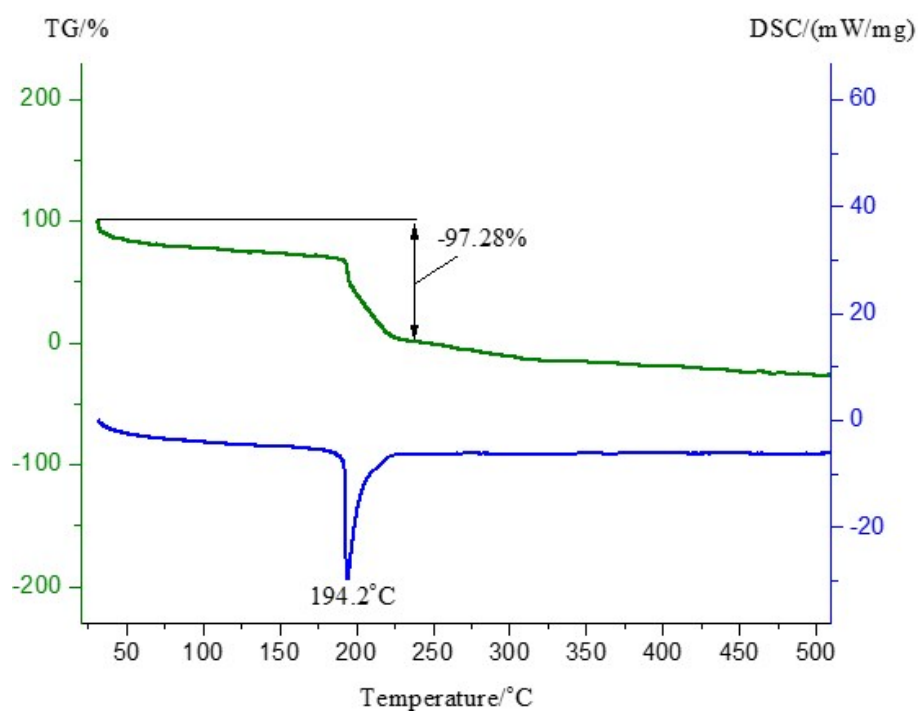
DSC spectrum of **7a**



DSC spectrum of **7b**



DSC spectrum of 7c



DSC spectrum of 7d

	5a	5b	5d
CCDC	1442351	1442350	1442346
Empirical formula	C ₂ H ₄ N ₈ O ₆	C ₂ H ₅ N ₉ O ₆	C ₃ H ₆ N ₁₀ O ₆
Formula mass	236.13	251.15	278.18
Temperature (K)	98.7	99.4	98.7
Crystal system	Monoclinic	orthorhombic	Monoclinic
Space group	P ₁ 2 ₁ /c ₁	Pbca	P ₁ 2 ₁ /c ₁
a / Å, b / Å, c / Å	8.0847(3)	6.43139(19)	9.1227(5)
	7.5557(3)	16.4647(4)	7.0640(4)
	14.1232(6)	16.8638(5)	16.2597(11)
$\alpha / ^\circ, \beta / ^\circ, \gamma / ^\circ$	90.00, 99.661(4), 90.00	90.00, 90.00, 90.00	90.00, 103.139(6), 90.00
Z	4	8	4
Volume (Å ³)	850.49(6)	1785.73(9)	1020.39(10)
ρ_{calc} (g·cm ⁻³)	1.844	1.868	1.811
μ (mm ⁻¹)	0.177	0.178	0.168
F(000)	480	1024	568
Crystal size / mm ³	0.15 × 0.14 × 0.12	0.15 × 0.10 × 0.08	0.50 × 0.45 × 0.40
2 θ range for data collection	5.86 to 52°	6.92 to 51.98°	6.06 to 52°
Index ranges	-9 ≤ h ≤ 9	-7 ≤ h ≤ 7	-11 ≤ h ≤ 11
	-8 ≤ k ≤ 9	-12 ≤ k ≤ 20	-4 ≤ k ≤ 8
	-16 ≤ l ≤ 17	-20 ≤ l ≤ 14	-19 ≤ l ≤ 20
Reflections collected	3503	4817	4294
Independent reflections	1669[R(int) = 0.0192 (inf-0.9Å)]	1743[R(int)=0.0244 (inf-0.9Å)]	1999[R(int) = 0.0235 (inf-0.9Å)]
Goodness-of-fit on F ²	1.070	1.034	1.062
Final R indexes [I>2 σ (I) i.e. Fo>4 σ (Fo)]	R ₁ = 0.0319 wR ₂ = 0.0744	R ₁ = 0.0333 wR ₂ = 0.0772	R ₁ = 0.0368 wR ₂ = 0.0840
Final R indexes [all data]	R ₁ = 0.0393 wR ₂ = 0.0786	R ₁ = 0.0400 wR ₂ = 0.0808	R ₁ = 0.0448 wR ₂ = 0.0893

	6a	6b	6c
CCDC	1442343	1442345	1442398
Empirical formula	C ₃ H ₅ N ₇ O ₆ ·H ₂ O	C ₃ H ₆ N ₈ O ₆	C ₃ H ₅ N ₇ O ₇
Formula mass	253.04	250.16	251.14
Temperature (K)	173	173	173
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /c	P2 ₁ /c
a / Å, b / Å, c / Å	6.3136(5) 15.3604(13) 9.3411(8)	6.4094(9) 16.106(2) 9.1462(11)	6.4054(9) 15.829(2) 8.9085(10)
α / °, β / °, γ / °	90.00, 98.223(2), 90.00	90.00, 100.123(4), 90.00	90.00, 99.085(4), 90.00
Z	4	4	4
Volume (Å ³)	896.58(13)	929.5(2)	891.9(2)
ρ _{calc} (g·cm ⁻³)	1.759	1.788	1.870
μ (mm ⁻¹)	0.167	0.168	0.180
F(000)	458	512	512
Crystal size / mm ³	0.25 × 0.18 × 0.15	0.19 × 0.17 × 0.17	0.22 × 0.15 × 0.12
2θ range for data collection	6.90 to 50.50°	6.78 to 50.78°	6.44 to 50.06°
Index ranges	-6 ≤ h ≤ 7 -18 ≤ k ≤ 18 -10 ≤ l ≤ 11	-7 ≤ h ≤ 4 -19 ≤ k ≤ 19 -11 ≤ l ≤ 9	-4 ≤ h ≤ 7 -19 ≤ k ≤ 17 -9 ≤ l ≤ 10
Reflections collected	8925	5568	5074
Independent reflections	1589[R(int) = 0.0528 (inf-0.9Å)]	1676[R(int) = 0.0519 (inf-0.9Å)]	1601[R(int) = 0.0472 (inf-0.9Å)]
Goodness-of-fit on F ²	1.020	1.027	1.030
Final R indexes [I>2σ(I) i.e. Fo>4σ(Fo)]	R1 = 0.0424 wR2 = 0.0762	R1 = 0.0508 wR2 = 0.0931	R1 = 0.0458 wR2 = 0.0680
Final R indexes [all data]	R1 = 0.0763 wR2 = 0.0865	R1 = 0.1081 wR2 = 0.1107	R1 = 0.0941 wR2 = 0.0787

	7a	7b	7c
CCDC	1442352	1442349	1442347
Empirical formula	C ₄ H ₆ N ₆ O ₆	C ₄ H ₇ N ₇ O ₆	C ₄ H ₆ N ₆ O ₇
Formula mass	234.15	249.17	250.03
Temperature (K)	101.7	101.7	101.7
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /c	P2 ₁ /c
a / Å, b / Å, c / Å	4.9037(3) 25.6452(14) 7.3737(4)	4.8027(2) 27.5834(14) 7.2531(4)	4.9109(4) 27.172(2) 6.9470(6)
$\alpha / ^\circ, \beta / ^\circ, \gamma / ^\circ$	90.00, 106.189(6), 90.00	90.00, 104.940(6), 90.00	90.00, 106.073(8), 90.00
Z	4	4	4
Volume (Å ³)	890.53(9)	928.37(8)	890.77(13)
ρ_{calc} (g·cm ⁻³)	1.746	1.783	1.806
μ (mm ⁻¹)	0.163	0.165	0.170
F(000)	480	512	495
Crystal size / mm ³	0.22 × 0.14 × 0.05	0.50 × 0.18 × 0.15	0.70 × 0.45 × 0.30
2 θ range for data collection	5.96 to 51.98°	6 to 52°	6.28 to 52°
Index ranges	-5 ≤ h ≤ 6 -22 ≤ k ≤ 31 -6 ≤ l ≤ 9	-5 ≤ h ≤ 5 -34 ≤ k ≤ 33 -8 ≤ l ≤ 8	-6 ≤ h ≤ 4 -33 ≤ k ≤ 31 -8 ≤ l ≤ 8
Reflections collected	4027	5667	3676
Independent reflections	1732[R(int)=0.0335 (inf-0.9Å)]	1814[R(int)=0.0199 (inf-0.9Å)]	1736[R(int)=0.0284 (inf-0.9Å)]
Goodness-of-fit on F ²	1.085	1.054	1.090
Final R indexes [I>2 σ (I) i.e. Fo>4 σ (Fo)]	R1 = 0.0397 wR2 = 0.0807	R1 = 0.0771 wR2 = 0.1669	R1 = 0.0436 wR2 = 0.0979
Final R indexes [all data]	R1 = 0.0542 wR2 = 0.0884	R1 = 0.0849 wR2 = 0.1726	R1 = 0.0542 wR2 = 0.1038

	6d	7d
CCDC	1442342	1442353
Empirical formula	C ₄ H ₇ N ₉ O ₆	C ₅ H ₈ N ₈ O ₆
Formula mass	227.19	276.19
Temperature (K)	173	101.8
Crystal system	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /c
a / Å, b / Å, c / Å	7.0482(7) 15.5415(18) 9.6301(9)	8.1268(5) 9.3064(4) 14.0786(9)
$\alpha / ^\circ, \beta / ^\circ, \gamma / ^\circ$	90.00, 94.891(3), 90.00	90.00, 92.296(7), 90.00
Z	4	4
Volume (Å ³)	1051.04(19)	1063.92(10)
ρ_{calc} (g·cm ⁻³)	1.752	1.724
μ (mm ⁻¹)	0.161	0.156
F(000)	568	568
Crystal size / mm ³	0.25 × 0.23 × 0.18	0.34 × 0.25 × 0.15
2 θ range for data collection	6.36 to 46.56°	6.66 to 52°
Index ranges	-6 ≤ h ≤ 8	-10 ≤ h ≤ 9
	-18 ≤ k ≤ 14	-11 ≤ k ≤ 11
	-11 ≤ l ≤ 9	-17 ≤ l ≤ 16
Reflections collected	5981	4871
Independent reflections	1887[R(int) = 0.0375 (inf-0.9Å)]	2082[R(int) = 0.0344 (inf-0.9Å)]
Goodness-of-fit on F ²	1.028	1.087
Final R indexes [I > 2σ(I) i.e. Fo > 4σ(Fo)]	R1 = 0.0392 wR2 = 0.0881	R1 = 0.0403 wR2 = 0.0943
Final R indexes [all data]	R1 = 0.0786 wR2 = 0.0881	R1 = 0.0512 wR2 = 0.1023

	7	8
CCDC	1442344	1442348
Empirical formula	C ₄ H ₃ N ₅ O ₆	C ₃ H ₃ N ₇ O ₇
Formula mass	217.11	249.01
Temperature (K)	101.7	153(2)
Crystal system	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /c
a / Å, b / Å, c / Å	12.7070(8) 11.6741(8) 10.8273(8)	11.481(3) 6.6275(16) 12.722(3)
$\alpha / ^\circ, \beta / ^\circ, \gamma / ^\circ$	90.00, 99.727(7), 90.00	90.00, 110.433(3), 90.00
Z	8	4
Volume (Å ³)	1583.06(19)	907.1(4)
ρ_{calc} (g·cm ⁻³)	1.822	1.824
μ (mm ⁻¹)	0.172	0.177
F(000)	880	504
Crystal size / mm ³	0.50 × 0.45 × 0.45	0.42 × 0.38 × 0.28
2 θ range for data collection	6.44 to 52°	6.56 to 61.02°
Index ranges	-15 ≤ h ≤ 15	-16 ≤ h ≤ 16
	-14 ≤ k ≤ 13	-8 ≤ l ≤ 31
	-13 ≤ l ≤ 13	-18 ≤ l ≤ 18
Reflections collected	7014	11121
Independent reflections	3111[R(int) = 0.0446 (inf-0.9Å)]	3017[R(int) = 0.0260 (inf-0.9Å)]
Goodness-of-fit on F ²	1.054	0.999
Final R indexes [I > 2 σ (I) i.e. Fo > 4 σ (Fo)]	R1 = 0.0587 wR2 = 0.1484	R1 = 0.0402 wR2 = 0.0853
Final R indexes [all data]	R1 = 0.0748 wR2 = 0.1637	R1 = 0.0418 wR2 = 0.862

Heat of formation

Based on the Born-Haber energy cycle, heats of formation of ionic salts can be simplified by eqn. (1):

$$\Delta H_f^\circ (\text{ionic salts, 298 K}) = \sum \Delta H_f^\circ (\text{cation, 298 K}) + \sum \Delta H_f^\circ (\text{anion, 298 K}) - \Delta \text{HL} \quad (1)$$

in which ΔHL is the lattice energy of the ionic salts, which could be predicted by using the formula suggested by Jenkins et al.¹

$$\Delta \text{HL} = U_{\text{POT}} + [p(n_{\text{M}}/2 - 2) + q(n_{\text{X}}/2 - 2)]RT \quad (2)$$

in which n_{M} and n_{X} depend on the nature of the ions $\text{M}_{\text{p}+}$ and $\text{X}_{\text{q}-}$, respectively, and are equal to three for monoatomic ions, five for linear polyatomic ions, and six for nonlinear polyatomic ions. The equation for lattice potential energy U_{POT} has the form [eqn. (3)]:

$$U_{\text{POT}} (\text{kJ mol}^{-1}) = \gamma(\rho_{\text{m}}/M_{\text{m}})^{1/3} + \delta \quad (3)$$

in which ρ_{m} is the density (g cm^{-3}) and M_{m} is the chemical formula mass of the ionic material (g mol^{-1}), and the coefficients γ ($\text{kJ mol}^{-1} \text{cm}$) and δ (kJ mol^{-1}) are assigned literature values.

Table S1 Calculated total energy (E_0), zero-point energy (ZPE), thermal correction (HT), and heat of formation (HOF) of reference compounds.

Compd.	$E_0/\text{a.u.}$	ZPE (kJ/mol)	HT (kJ/mol)	HOF (kJ/mol)
5	-911.003790	215.27	35.12	301.3728
6	-895.005868	248.63	35.36	234.8022
7	-878.972332	280.05	35.56	175.75872
8	-1025.530974	301.75	41.02	98.22167
5 anion	-462.065951	83.93	20.31	8.7352
6 anion	-894.509467	215.23	34.60	-49.2789
7 anion	-878.459887	245.77	34.88	-67.0128
NH₄⁺				645.8 ²
N₂H₅⁺				774.1 ²
NH₄O⁺				687.2 ²
G⁺				571.9 ²

References

1. H. D. B. Jenkins, D. Tudeal and L. Glasser, L. *Inorg. Chem.*, 2002, **41**, 2364–2367.
2. D. Fischer, T. M. Klapçtke, M. Reymann and J. Stierstorfer, *Chem. Eur. J.*, 2014, **20**, 6401–6411.