

Supporting information

Pd/Ag and Pd/Au bimetallic nanocatalysts on mesoporous silica for plasmon-mediated enhanced catalytic activity under visible light irradiation

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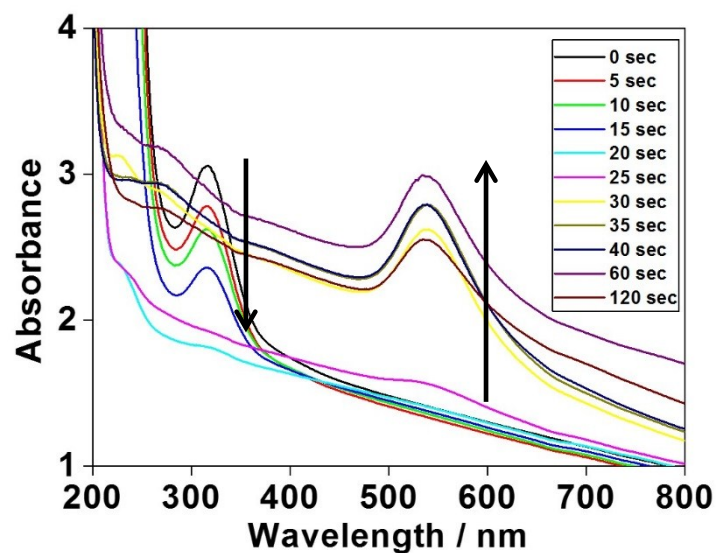


Figure S1. Time evolution of MW irradiation for the reduction of HAuCl₄ to Au NPs on a mesoporous silica support

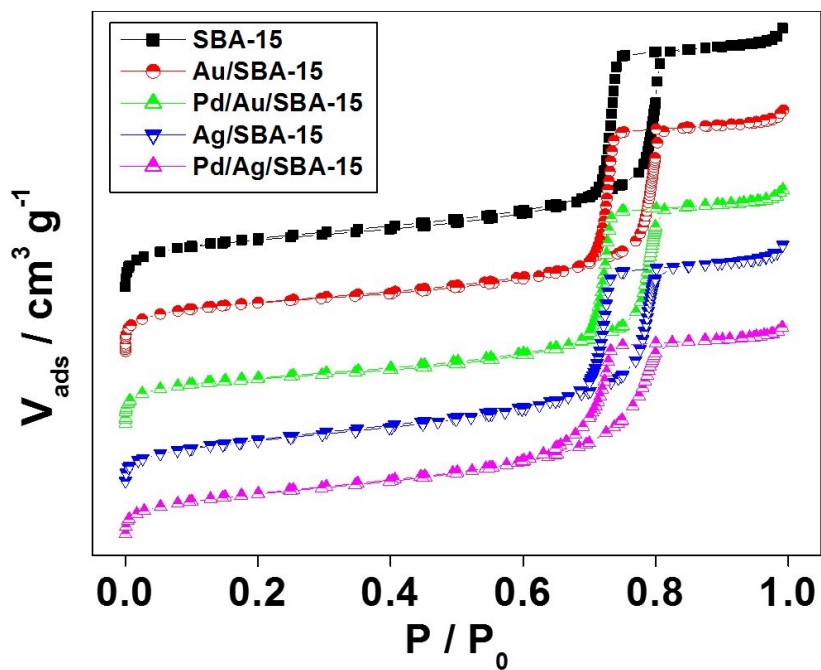


Figure S2. N₂ adsorption-desorption isotherms performed at 77 K for SBA-15, M/SBA-15 and Pd/M/SBA-15 catalysts (M=Au, Ag).

Table S1. Textural properties of the prepared catalysts

Catalyst	Mesopore volume (cm ³ g ⁻¹)	BET surface area (m ² g ⁻¹)
SBA-15	1.32	756
Au/SBA-15	1.21	668
Pd/Au/SBA-15	1.18	642
Ag/SBA-15	1.24	717
Pd/Ag/SBA-15	1.10	692

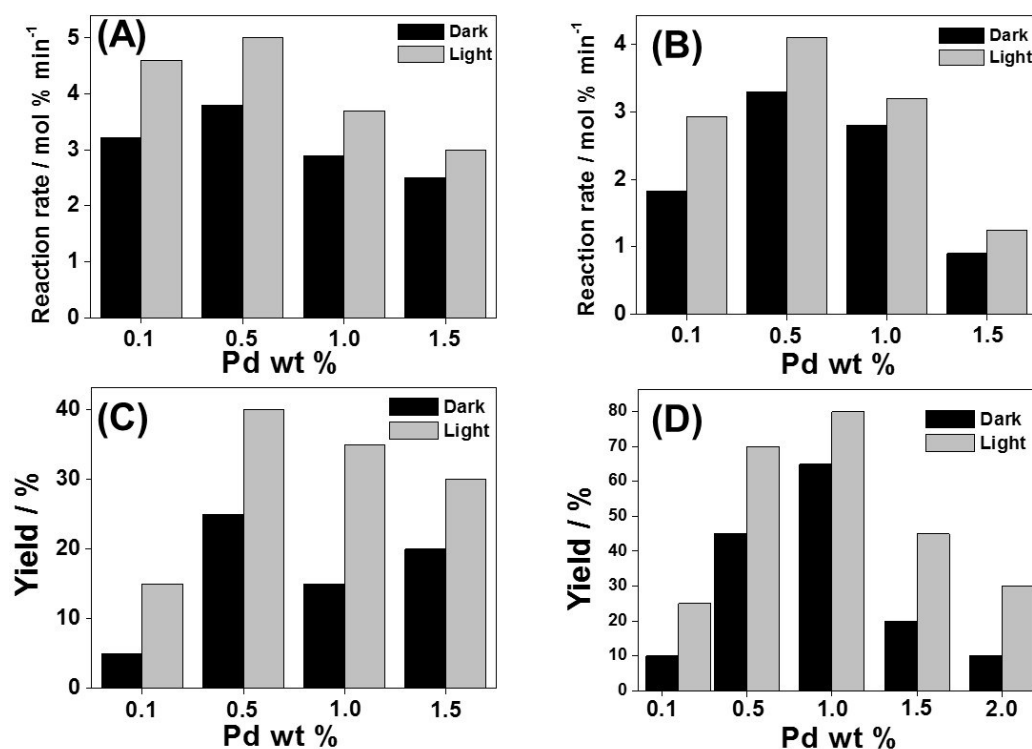


Figure S3. Optimization reactions for AB dehydrogenation (A) Pd on Ag, (B) Pd on Au and Suzuki coupling reaction (C) Pd on Ag, (D) Pd on Au.

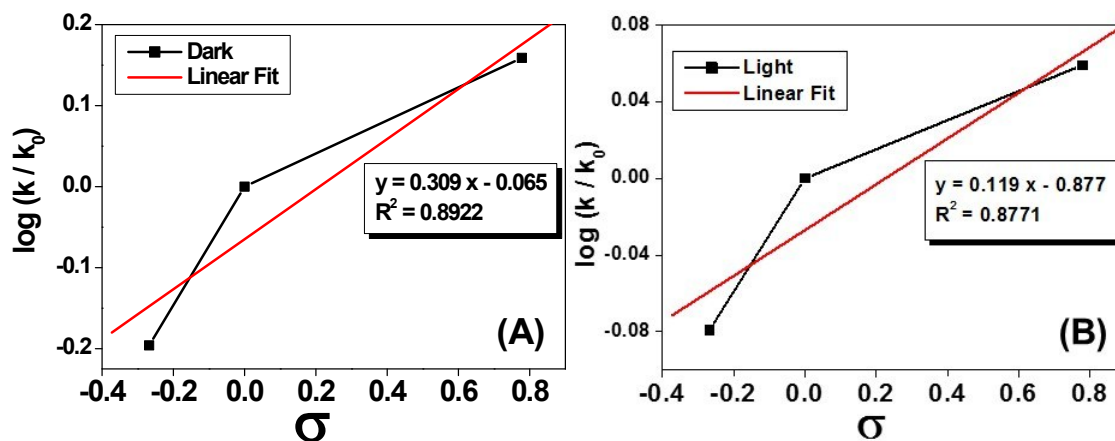


Figure S4. Hammett plot corresponding to Suzuki Miyaura coupling reaction for substituted iodobenzene under (A) Dark and (B) visible light irradiation

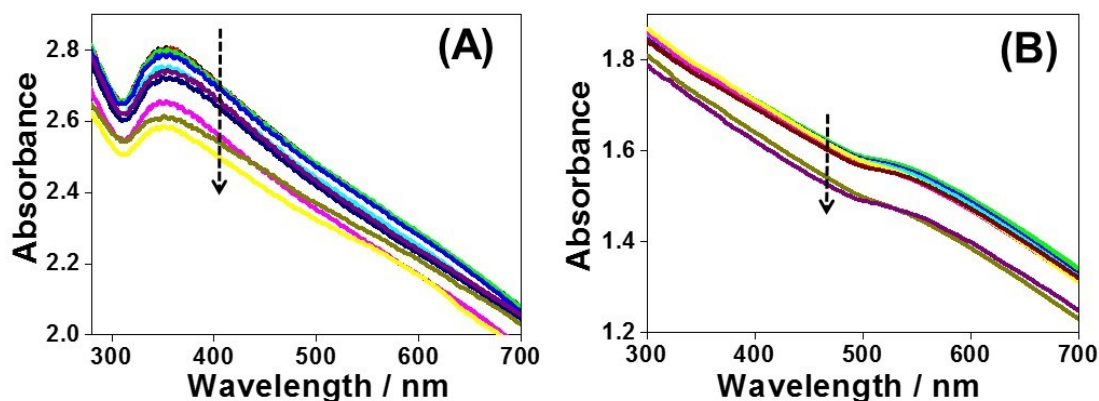


Figure S5. UV-vis spectra of (A) Pd/Ag/SBA-15 and (B) Pd/Au/SBA-15 as a result of electron injection experiments.

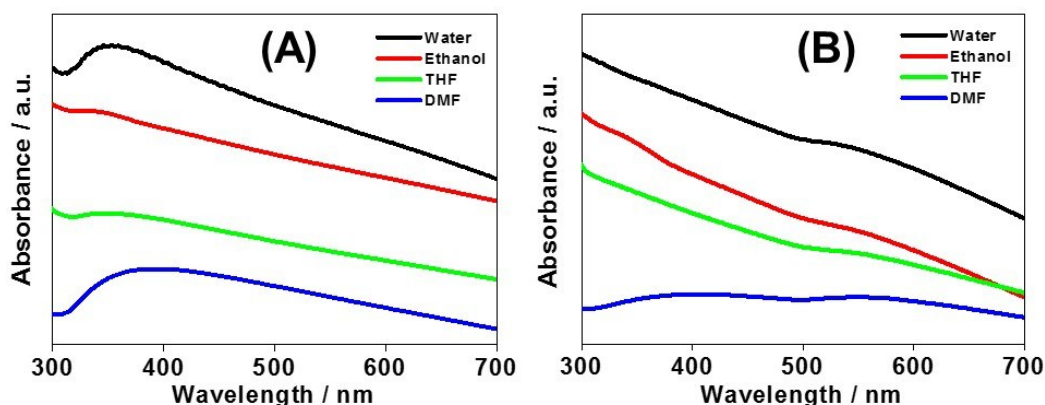


Figure S6. Effect of refractive index of solvent on plasmonic wavelength for (A) Pd/Ag/SBA-15 and (B) Pd/Au/SBA-15

Section S1. Tunability Experiments

- a) **Tunability by electron injection:** Tunability by electron injection was measured by in-situ UV-vis spectroscopy measurements. The reaction was carried out in quartz cell by adding 5 mg of plasmonic catalyst (M/SBA-15) and 2.5 mL of water. The process was monitored by UV after the successive addition (10 μ L) of 0.132 M sodium borohydride (used as electron injector in this case). The spectra obtained from Ag and Au after electron injection are shown in Fig. 8 (A) and (B).
- b) **Tunability by varying refractive index:** In a quartz cell, 5 mg of the catalyst powder was dispersed in solvents of different refractive index and their interaction with incoming light was again monitored by UV-vis spectroscopy. Different plasmonic peak showed in different solvents and found to be a linear relationship of

the refractive index and plasmonic shift. Corresponding absorbance shift and linear graph is shown in Fig. 9 (A) and (B) for Ag and Au.

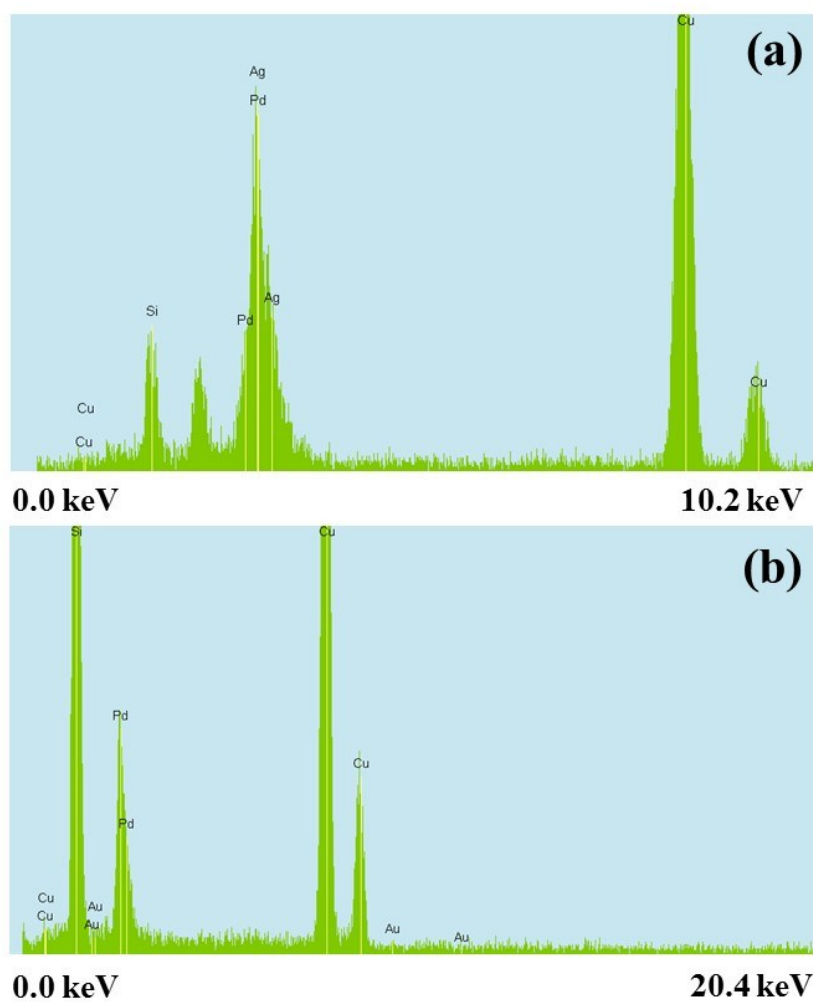


Figure S7. TEM-EDX analysis for (a) Pd/Ag/SBA-15 and (b) Pd/Au/SBA-15