

Supporting Information

Biotemplated hierarchical polyaniline composite electrode with high performance for flexible supercapacitors

Cheng-Ming Chang, Zh-Hao Hu, Ting-Yin Lee, Yi-An Huang, Wei-Fu Ji, Wei-Ren Liu,

Jui-Ming Yeh and Yen Wei*

Dr. C. M. Chang, Z. H. Hu, T. Y. Lee, Y. A. Huang, Prof. J. M. Yeh
Department of Chemistry and Center for Nanotechnology, Chung-Yuan Christian
University
Chung Li 32023, Taiwan, R.O.C.

*E-mail: juiming@cycu.edu.tw

Prof. W. R. Liu
Department of Chemical Engineering, R&D Center for Membrane Center, Chung-
Yuan Christian University
Chung Li 32023, Taiwan, R.O.C.

Prof. Y. Wei
Department of Chemistry and Key Lab of Organic Optoelectronic & Molecular
Engineering of Ministry of Education, Tsinghua University
Beijing, 100084, China

Materials needed:

Aniline (Sigma-Aldrich) was distilled prior to use. 4,4'-Diaminodiphenylamine sulfate (APS, Aldrich), N,N-dimethylacetamide (DMAc) and N-methyl-2-pyrrolidinone (NMP, Aldrich) and polydimethylsiloxane (PDMS, Dow Corning, Sylgard 184) were used as received without further purification. All the reagents were reagent grade unless otherwise stated.

The RGO purify process:

GO derived from SFG44 synthetic graphite powders (TIMCAL) was synthesized by a modified Hummers' method. 8.0 g synthetic graphite powder and 4.0 g NaNO_3 were put into 560 ml concentrated H_2SO_4 solution with stirring for 2 h. Then 32 g KMnO_4 was slowly added into the flask with ice bath for 2 h. The mixture was diluted by 800 ml deionized water. After that, 5% H_2O_2 was added into the solution until the color of the mixture changed to brown to ensure that KMnO_4 was fully reduced. The as-prepared GO slurry was re-dispersed in de-ionized water. Then, the mixture was washed with 0.1 M HCl solution to remove SO_4^{2-} ions. Finally, the GO solution was washed with distilled water to remove the residual acid until the solution pH reached ~ 5 . Reduction processes were adopted for GO to form RGO. Thermally reduced graphene oxides (RGO) were prepared by calcination at different temperatures under 15% H_2/N_2 atmosphere for 2 h with a heating rate of $0.5^\circ\text{C min}^{-1}$.

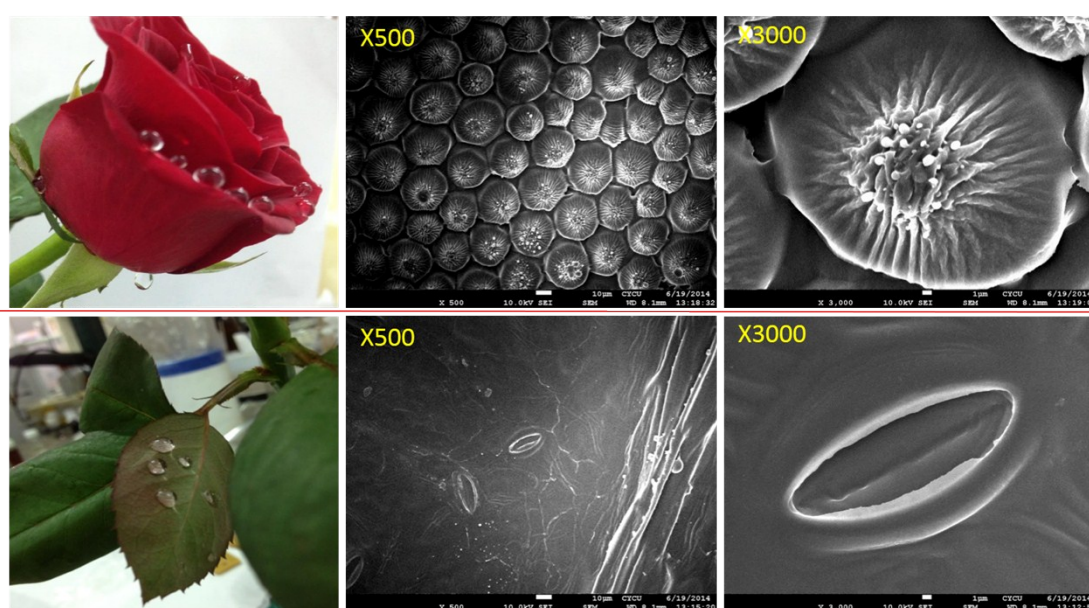
PDMS template method:

The PDMS pre-polymer was prepared from a 10:1 mixture (by weight) of the base and curing agent. After thoroughly mixing the base and curing agent, the pre-polymer was poured into the template of fresh Rose flower petal and leaves and then cured at 50°C for 4 h and 90°C overnight to ensure the completion of the PDMS cross-linking process. After curing, the resulting PDMS was separated from the template of the

petal and leaf and used as a negative template for preparing tri-dimensional construction PANI coatings.

Support information Fig. S1

The petal and leaves structure from *Rose flower* was characterized by photograph and SEM image.



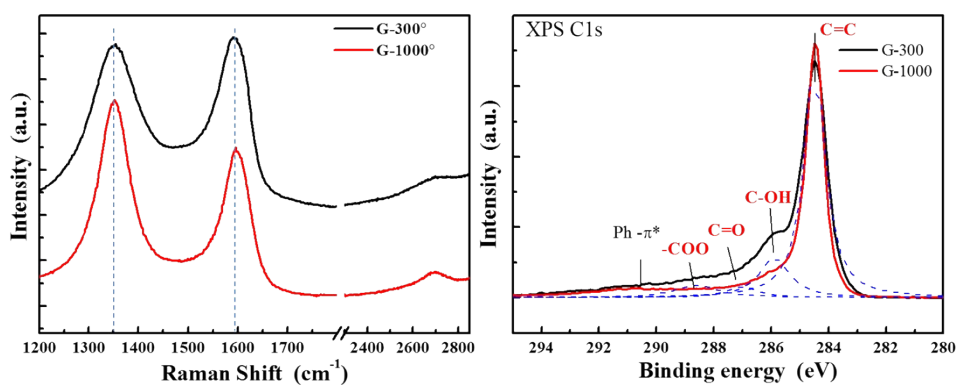
Support information Fig. S2

RGO structure change undergo in thermally calcination process was characterized by

Raman and XPS

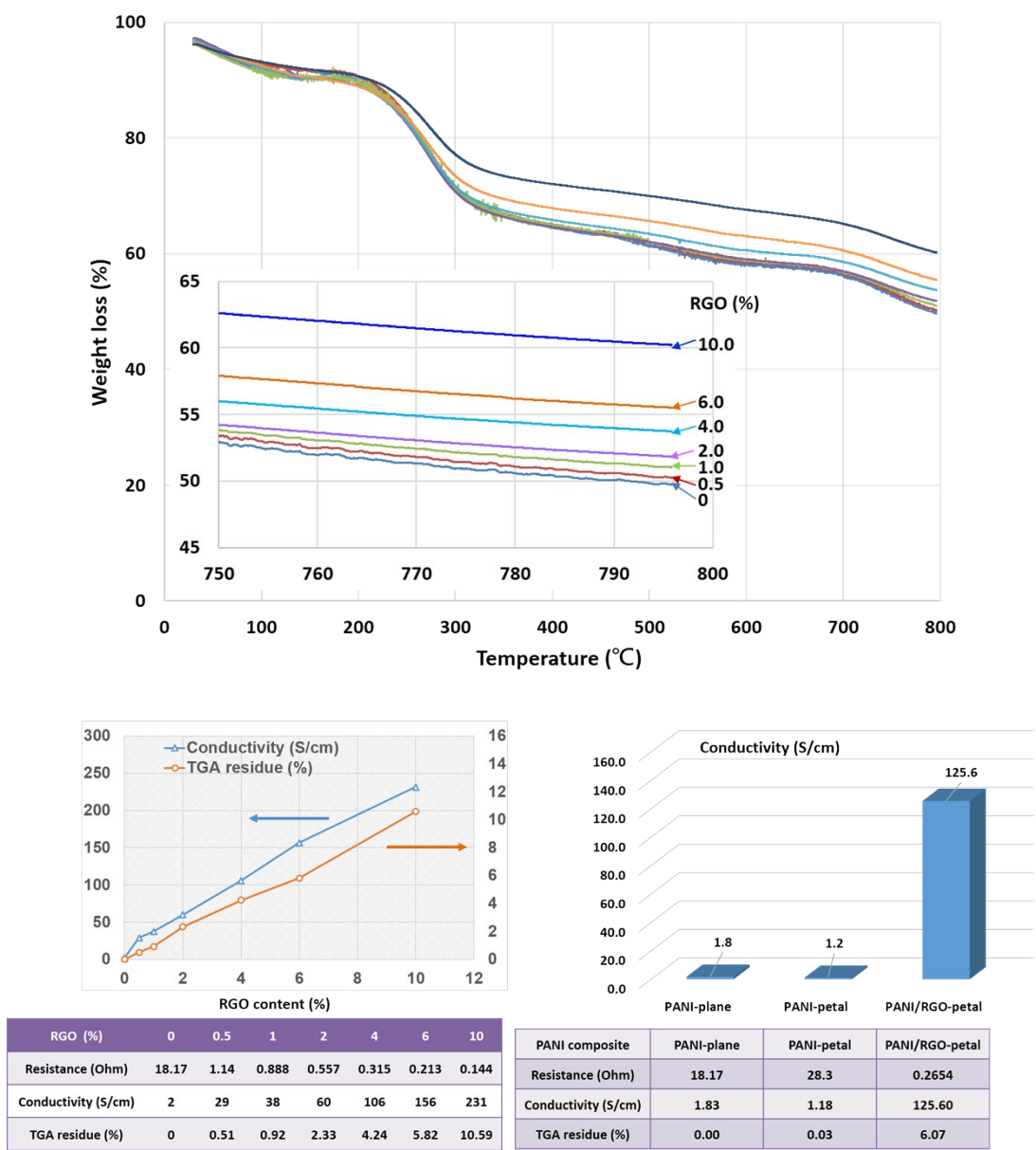
G-300°: thermally calcination temperature at 300°

G-1000°: thermally calcination temperature at 1000°



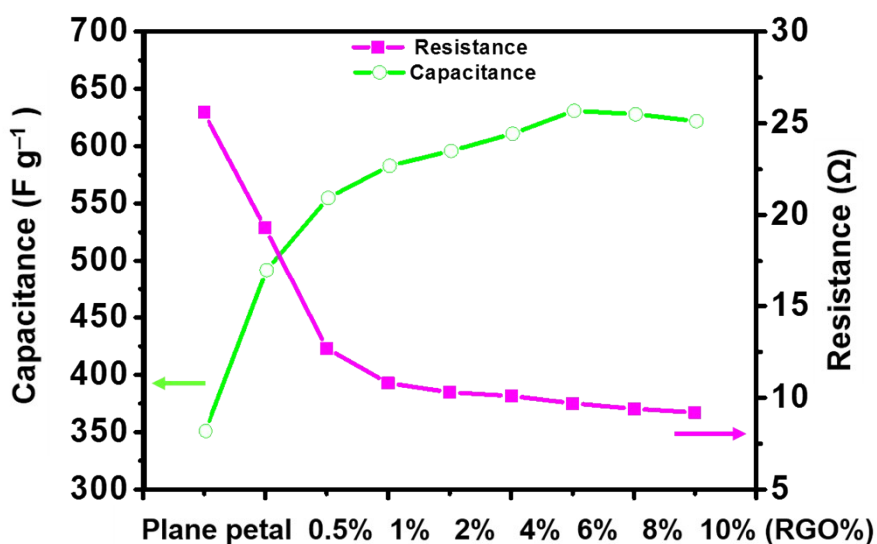
Support information Fig. S3

The RGO real content and conductivity of PANI composite film as shown in follow



Support information Fig. S4

The electrochemical capacities of reduced graphene oxide (RGO) content in the PANI/RGO-petal composite.



The conductive resistance of the PANI/RGO-petal hybrid decreased as the RGO loading content increased from 0.5 to 10 wt% (The petal-like hybrid film was easily cracked by peeling from template substrate at RGO content over 10 wt%);

Support information Fig. S5

The coulombic efficiency of PANI/RGO-petal composite and inset graphic is charge discharge curves between start and the end in the period of cycle testing

