

Supporting Information

Two-step Synthesis of Binary Ni-Fe Sulfides Supported on Nickel Foam as Highly Efficient Electrocatalysts for Oxygen Evolution Reaction

Bin Dong^{*a, b}, Xin Zhao^a, Guan-Qun Han^{a, b}, Xiao Li^a, Xiao Shang^a,

Yan-Ru Liu^a, Wen-Hui Hu^a, Yong-Ming Chai^a, Hui Zhao^{*a}, Chen-Guang Liu^{*a}

a State Key Laboratory of Heavy Oil Processing, China University of Petroleum (East China), Qingdao 266580, PR China

b College of Science, China University of Petroleum (East China), Qingdao 266580, PR China

* Corresponding author. Email: dongbin@upc.edu.cn (B. Dong), zhaohui@upc.edu.cn (H. Zhao), cgliu@upc.edu.cn (C.-G. Liu) Tel: +86-532-86981376, Fax: +86-532-86981787

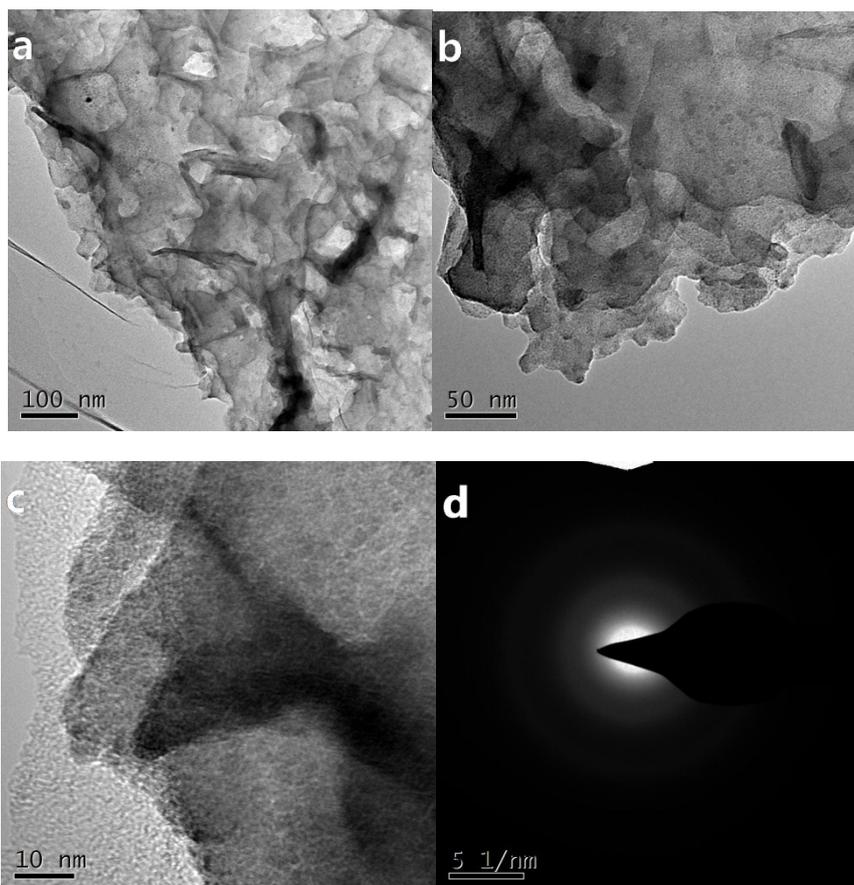


Fig. S1 (a, b, c) HR-TEM images of NiFe/NF; (d) Selected area electron diffraction image of NiFe/NF.

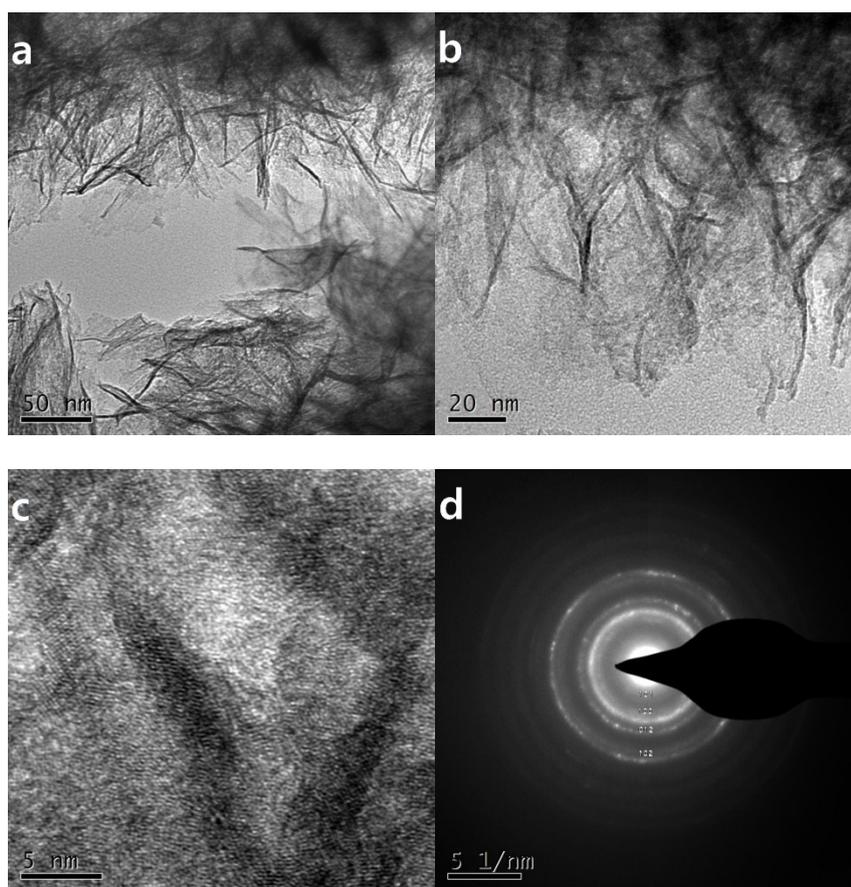


Fig. S2 (a, b, c) HR-TEM images of NiFeS-Fe/NF; (d) Selected area electron diffraction image of NiFeS-Fe/NF.

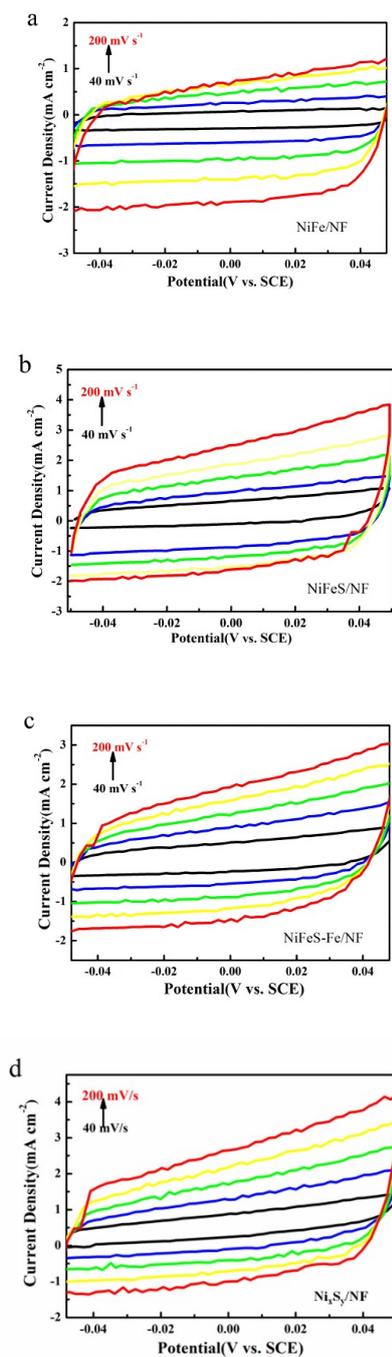


Fig. S3 CVs of the double-layer capacitance measurement for the four different samples in 1 M KOH in the non-Faradaic region of -0.05-0.05 V vs. SCE with different scan rates, varying from 40 mV s^{-1} to 200 mV s^{-1} : (a) NiFe/NF; (b) NiFeS/NF; (c) NiFeS-Fe/NF and (d) $\text{Ni}_x\text{S}_y/\text{NF}$.

Calculation of electrochemical surface area (ECSA) and roughness factor (RF)^{1,2}

According to the previous literature, the electrochemical surface area (ECSA) and the roughness factor (RF) can be calculated based on following equations.

$$\text{ECSA} = C_{\text{dl}} / C_s$$

$$\text{RF} = \text{ECSA} / \text{GSA}$$

C_{dl} : the measured double layer capacitance of NiFe/NF, NiFeS/NF and NiFeS-Fe/NF electrode in 1 M KOH (in mF);

C_s : the specific capacitance, according to the literature, the value is 0.040 mF cm⁻² in 1M NaOH;

GSA: the geometric surface area of NiFe/NF, NiFeS/NF and NiFeS-Fe/NF electrode, which is 1 cm²

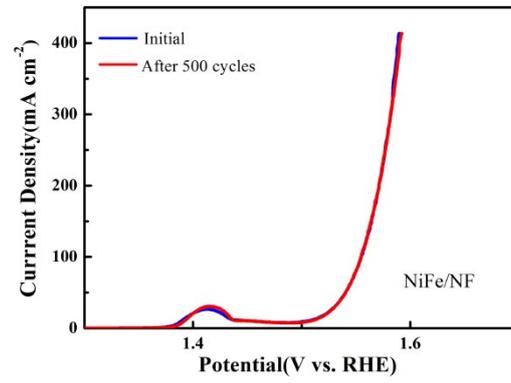


Fig. S4 Stability test of NiFe/NF after 500 cycles.

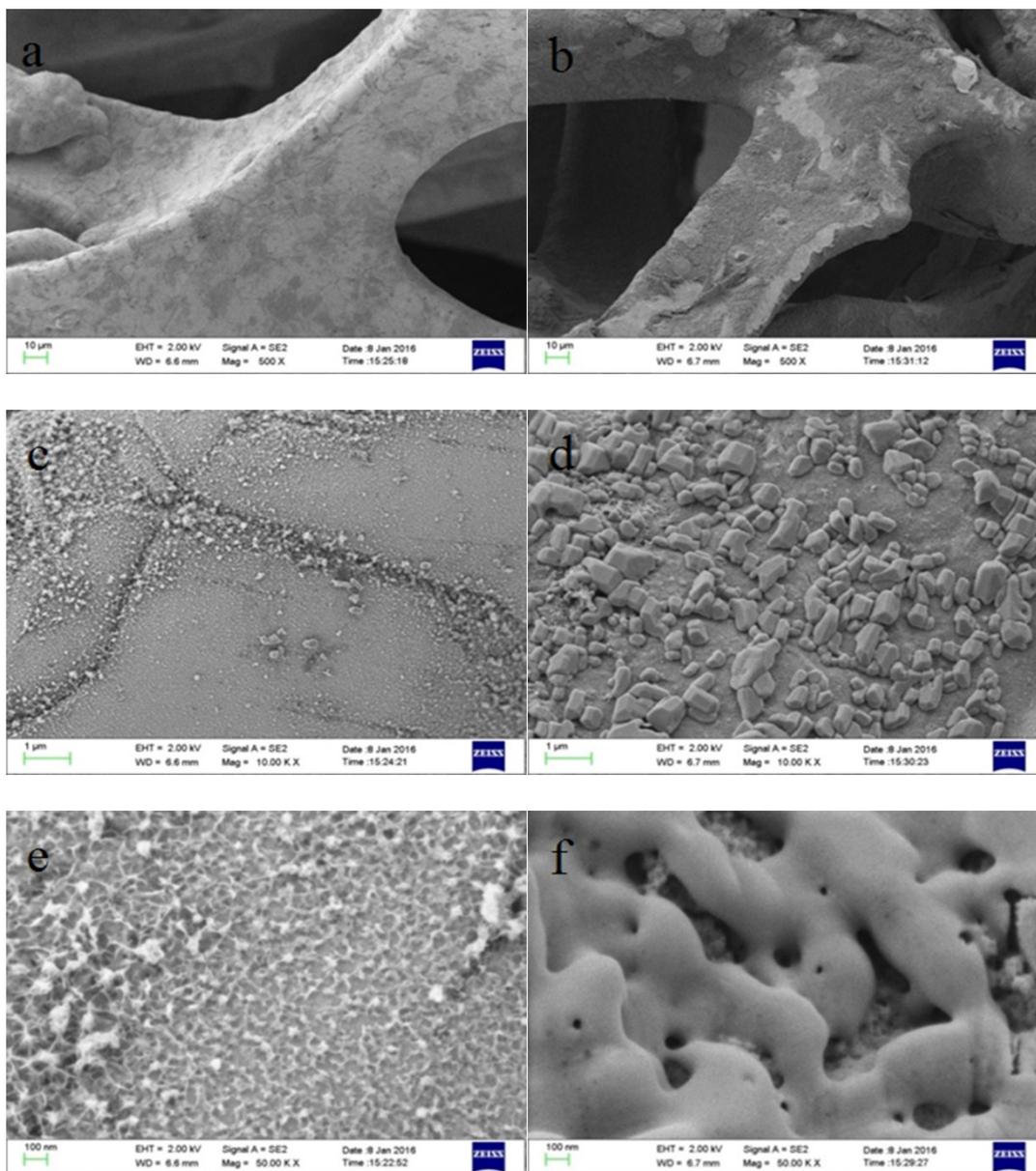


Fig. S5 (a, c, e) SEM image of NiFe/NF before stability test;
(b, d, f) SEM image of NiFe/NF after stability test.

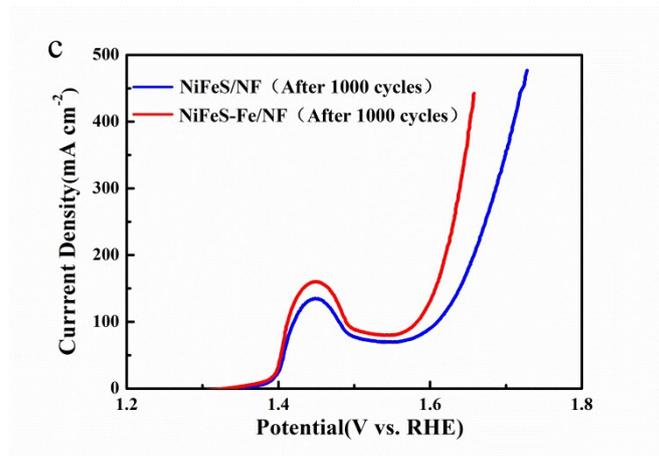
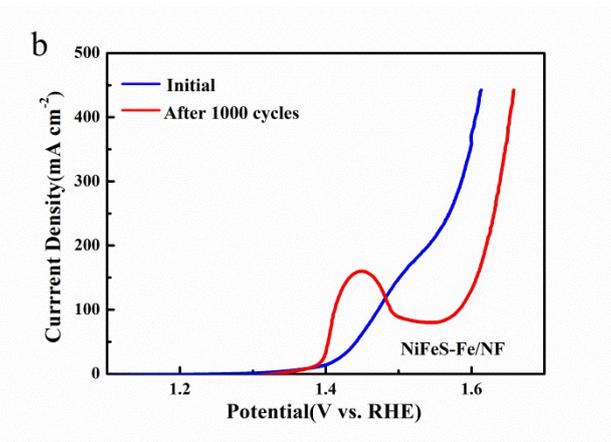
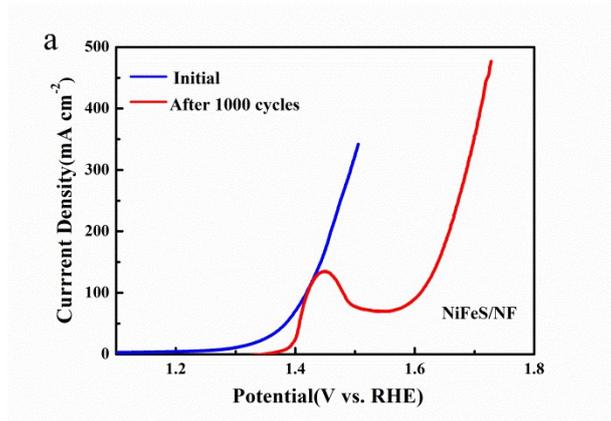


Fig. S6 Stability test of NiFeS/NF and NiFeS-Fe/NF after 1000 cycles.

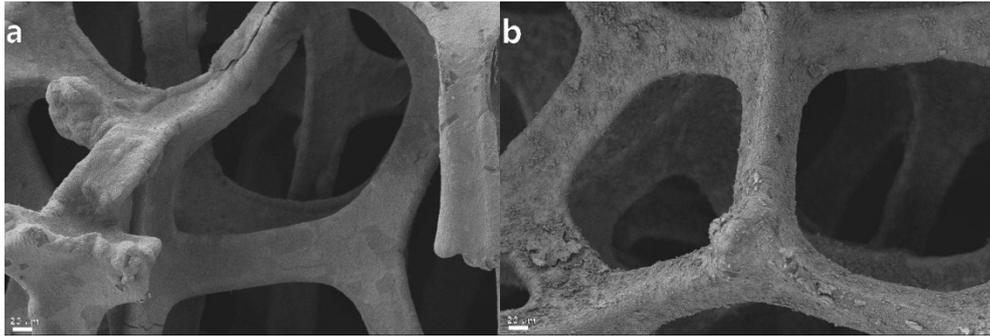


Fig. S7 (a) SEM image of NiFeS/NF before stability test of 1000 cycles;
(b) SEM image of NiFeS/NF after stability test of 1000 cycles.

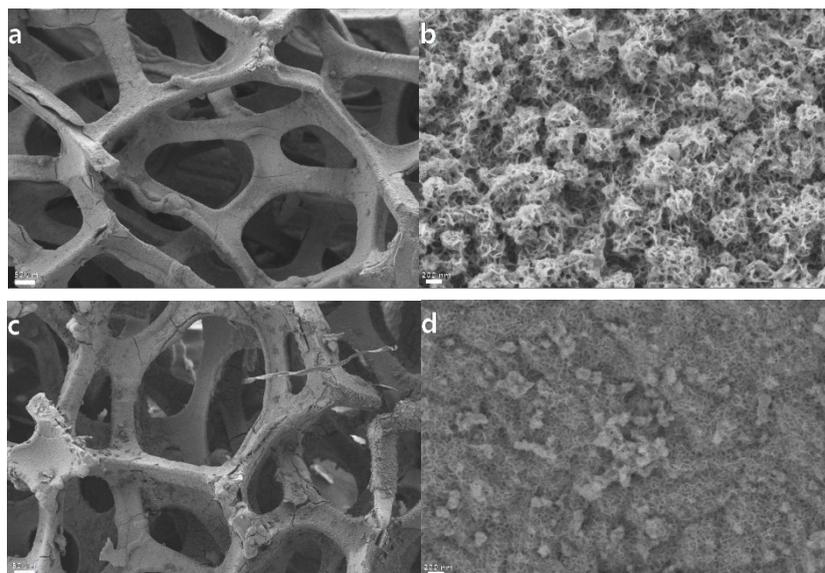


Fig. S8 (a, b) SEM image of NiFeS-Fe/NF before stability test of 1000 cycles; (c, d) SEM image of NiFeS-Fe/NF after stability test of 1000 cycles.

Table S1. The calculated ECSA and RF of the as-prepared NiFe/NF, Ni_xS_y/NF, NiFeS/NF and NiFeS-Fe/NF samples ^{1,2}.

Sample	C_{dl} (mF)	C_s (mF cm⁻²)	ECSA (cm²)	GSA (cm²)	RF
NiFe/NF	6.95	0.04	173.75	1	173.75
Ni _x S _y /NF	9.37	0.04	234.25	1	234.25
NiFeS/NF	10.05	0.04	251.25	1	251.25
NiFeS-Fe/NF	8.20	0.04	205.00	1	205.00

Reference

- 1 C. C. L. McCrory, S. Jung, J. C. Peters, T. F. Jaramillo, *J. Am. Chem. Soc.*, 2013, **135**, 16977.
- 2 X. Y. Lu, C. Zhao, *Nat. Commun.*, 2015, 6, 6616.