

Supporting Information

Single-layered Platinum Nanocage: A Highly Selective and Efficient Catalyst for Fuel Cell

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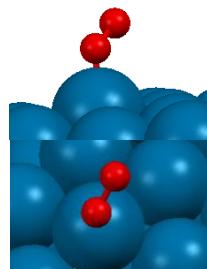
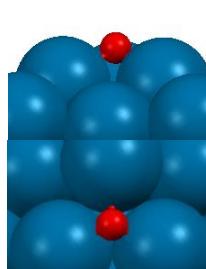
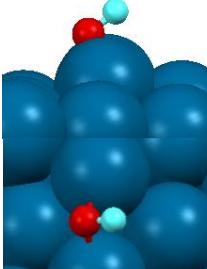
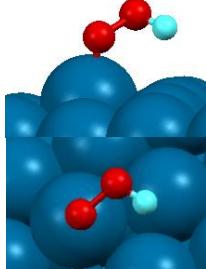
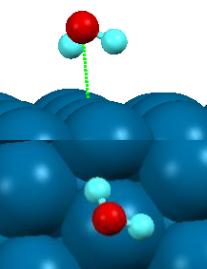
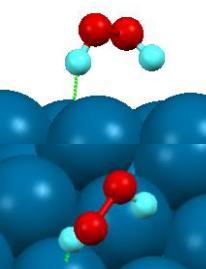
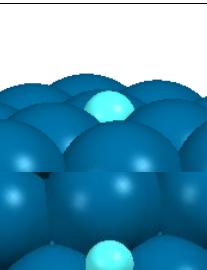
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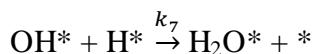
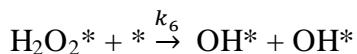
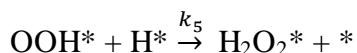
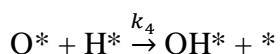
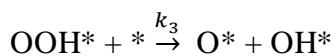
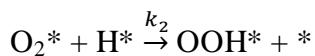
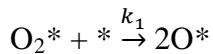
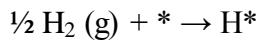
S5: Values of phonon frequencies.

S1: Adsorption site of the most stable adsorbates on the (111) facet of the Pt₆₆ nanocage

Adsorbate	Optimized structure	Adsorbate	Optimized structure
*O ₂		*O	
*OH		*OOH	
*H ₂ O		*H ₂ O ₂	
*H			

S2: Microkinetic Analysis

A microkinetic analysis is developed from the DFT calculated values to give more insight into the coverage, reaction rate and the selectivity of different products. Elementary steps considered in the microkinetic model are summarized as following.



The adsorption of O₂ and H₂ are highly exothermic and the process are assumed to be in equilibrium and are expected to be strongly adsorbed. The equilibrium constants are calculated using the following equation $K_{\text{eq}} = \exp[-(\Delta E_{\text{ads}} - T\Delta S)/k_B T]$, where ΔE_{ads} is the adsorption energy of O₂ and H₂, and ΔS is the entropy change of O₂ and H₂ upon adsorption, k_B is the Boltzmann's Constant and T is the temperature in Kelvin. The gas phase entropy is obtained from NIST database.¹ The other steps are described by forward and backward reaction rate. Rate constants are calculated using the following equation.

$$k_{i/-i} = \left(\frac{k_B T}{h} \right) \left(\frac{q_F}{q_I} \right) e^{-\Delta G^\ddagger / k_B T}$$

Where k_B is the Boltzmann constant, T is the temperature, h is the Plank constant. Here q_I and q_F are the vibrational partition functions for the initial and final state structures and ΔG^\ddagger is the

Gibbs free energy barrier for the initial and final state of the elementary reaction. The vibrational partition functions were calculated using the following equation:

$$q = \sum_i \frac{1}{1 - e^{-hv_i/k_B T}}$$

where v_i are the vibrational frequencies.

The backward reactions are not considered for all the exothermic forward reactions assuming the exothermic reactions are irreversible. The site balance of intermediate species included in the reaction is written in terms of coverage.

$$\Theta_{O_2} + \Theta_{OOH} + \Theta_O + \Theta_{OH} + \Theta_{H_2O} + \Theta_{H_2O_2} + \Theta_* = 1$$

The coverage of O₂ and H₂ are obtained by the following equations.

$$\Theta_{O_2} = p_{O_2} K_{O_2} \Theta_*$$

$$\Theta_H = p_{H_2}^{1/2} K_{H_2} \Theta_*$$

The coverage of the other species is obtained by considering the steady-state approximation, where the rate of formation and rate of consumption of each species are assumed to be equal:

$$d\Theta_O/dt = 2k_1 \Theta_{O_2} \Theta_* - k_4 \Theta_O \Theta_H + k_3 \Theta_{OOH} \Theta_* = 0 \quad (S1)$$

$$d\Theta_{OOH}/dt = k_2 \Theta_{O_2} \Theta_H - k_3 \Theta_{OOH} \Theta_* - k_5 \Theta_H \Theta_{OOH} = 0 \quad (S2)$$

$$d\Theta_{OH}/dt = k_4 \Theta_O \Theta_H + k_3 \Theta_{OOH} \Theta_* - k_7 \Theta_{OH} \Theta_H + 2k_6 \Theta_{H_2O_2} \Theta_* = 0 \quad (S3)$$

$$d\Theta_{H_2O_2}/dt = k_5 \Theta_H \Theta_{OOH} - k_6 \Theta_{H_2O_2} \Theta_* = 0 \quad (S4)$$

$$d\Theta_{H_2O}/dt = k_7 \Theta_{OH} \Theta_H = 0 \quad (S5)$$

Now, adding the S1 and S3, we get

$$2k_1 \Theta_{O_2} \Theta_* + 2k_3 \Theta_{OOH} \Theta_* - k_7 \Theta_{OH} \Theta_H + 2k_6 \Theta_{H_2O_2} \Theta_* = 0$$

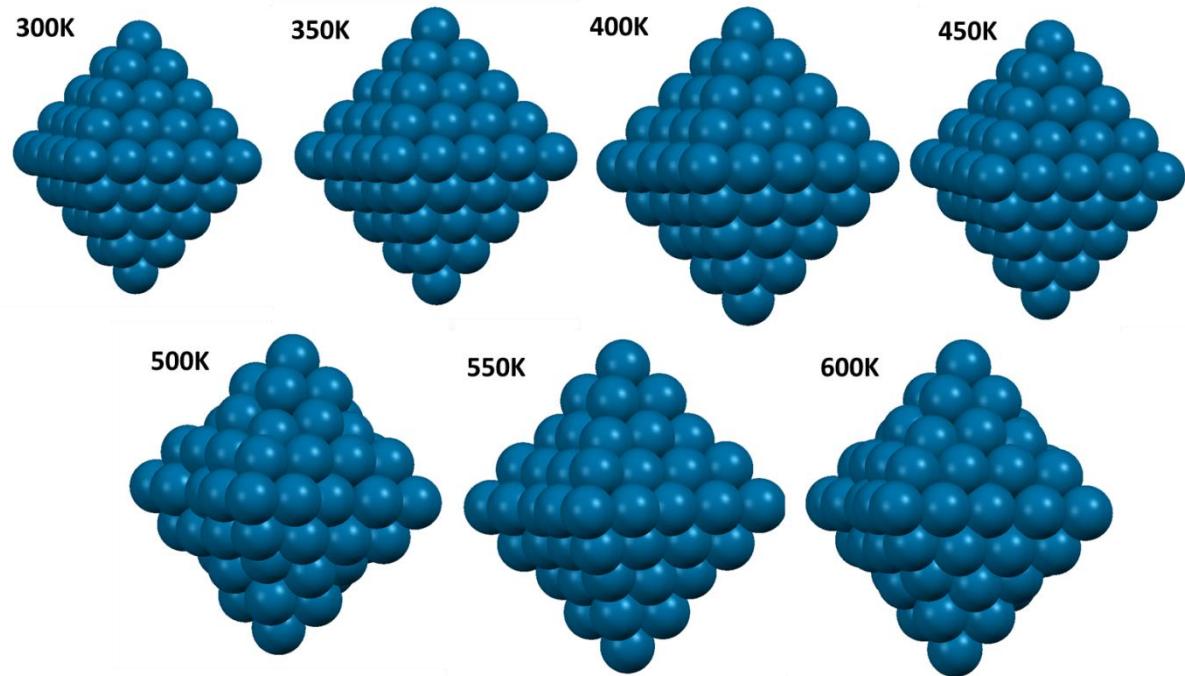
Using the value of Θ_{H₂O₂} in the above equation, we get

$$\Theta_{OH} = \frac{2k_1 \Theta_{O_2} \Theta_* + (2k_3 \Theta_* + 2k_5 \Theta_H) \Theta_{OOH}}{k_7 \Theta_H}$$

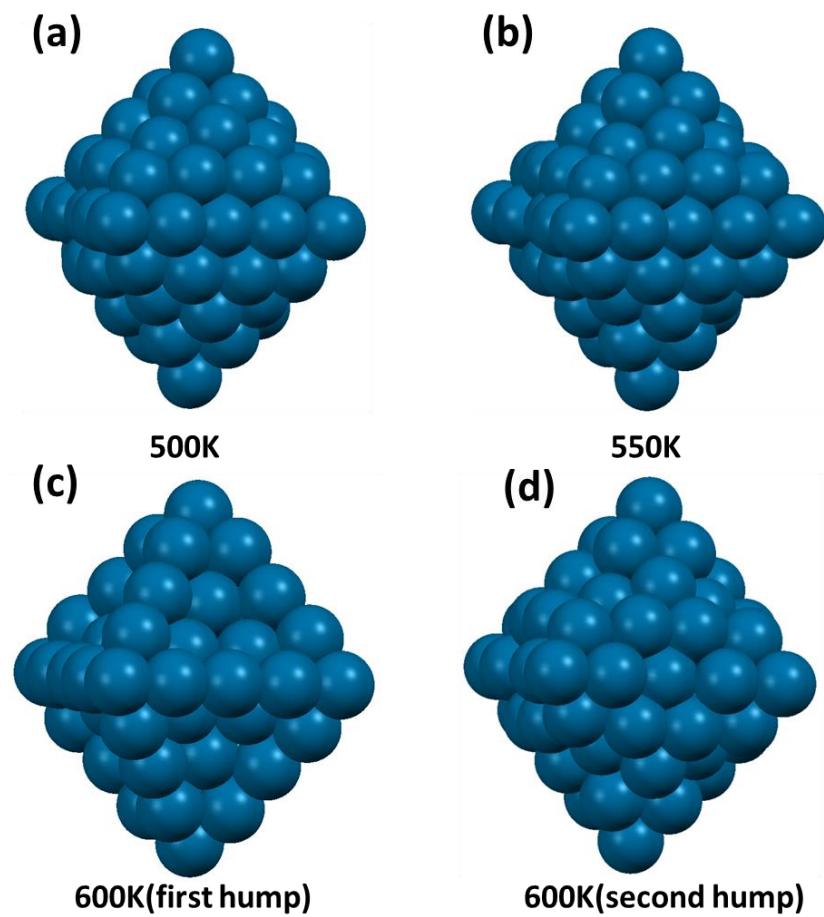
Now, using the value of Θ_{H₂O₂} and Θ_{OH}

$$\frac{d\theta_{H_2O_2}/dt}{d\theta_{H_2O}/dt} = \frac{2k_5\theta_H}{2k_1\theta_{O_2}\theta^* + 2k_3\theta^* + 2k_5\theta_H}$$

S3: Snapshots of the nanocage after the simulation at different temperatures



S4: Snapshots of the nanocage during the simulation at different temperatures



S5: Values of phonon frequencies

1 f	216.397269 cm-1
2 f	216.390613 cm-1
3 f	216.386107 cm-1
4 f	212.498069 cm-1
5 f	212.492794 cm-1
6 f	211.898624 cm-1
7 f	211.893677 cm-1
8 f	211.885754 cm-1
9 f	207.194281 cm-1
10 f	206.298681 cm-1
11 f	206.288807 cm-1
12 f	206.288406 cm-1
13 f	204.809370 cm-1
14 f	204.024513 cm-1
15 f	203.140960 cm-1
16 f	203.139597 cm-1
17 f	203.139471 cm-1
18 f	202.342601 cm-1
19 f	202.342100 cm-1
20 f	202.341279 cm-1
21 f	199.064115 cm-1
22 f	199.055852 cm-1
23 f	199.036579 cm-1
24 f	198.946030 cm-1
25 f	198.940236 cm-1
26 f	196.898465 cm-1

27 f	196.875841 cm-1
28 f	191.696890 cm-1
29 f	191.563261 cm-1
30 f	191.534444 cm-1
31 f	191.533104 cm-1
32 f	191.357522 cm-1
33 f	191.348837 cm-1
34 f	191.344277 cm-1
35 f	186.567979 cm-1
36 f	186.559462 cm-1
37 f	186.541792 cm-1
38 f	174.351973 cm-1
39 f	174.326767 cm-1
40 f	174.325227 cm-1
41 f	170.056933 cm-1
42 f	169.145360 cm-1
43 f	169.134252 cm-1
44 f	169.100871 cm-1
45 f	167.585393 cm-1
46 f	167.578682 cm-1
47 f	159.796396 cm-1
48 f	159.748679 cm-1
49 f	159.745521 cm-1
50 f	158.597013 cm-1
51 f	158.591112 cm-1
52 f	158.296059 cm-1
53 f	158.293344 cm-1

54 f	158.272506 cm-1
55 f	156.429550 cm-1
56 f	156.429417 cm-1
57 f	156.414640 cm-1
58 f	155.370191 cm-1
59 f	155.355359 cm-1
60 f	155.352346 cm-1
61 f	154.650994 cm-1
62 f	154.633116 cm-1
63 f	154.630503 cm-1
64 f	154.448433 cm-1
65 f	153.554207 cm-1
66 f	153.553562 cm-1
67 f	153.539671 cm-1
68 f	153.099550 cm-1
69 f	153.095081 cm-1
70 f	153.088415 cm-1
71 f	145.224251 cm-1
72 f	145.222899 cm-1
73 f	145.213292 cm-1
74 f	141.044103 cm-1
75 f	141.036874 cm-1
76 f	141.032107 cm-1
77 f	140.349207 cm-1
78 f	139.358300 cm-1
79 f	139.341066 cm-1
80 f	136.801158 cm-1

81 f	135.126733 cm-1
82 f	135.118000 cm-1
83 f	135.116471 cm-1
84 f	132.469803 cm-1
85 f	132.453610 cm-1
86 f	126.792780 cm-1
87 f	126.785384 cm-1
88 f	126.748204 cm-1
89 f	124.572329 cm-1
90 f	124.539914 cm-1
91 f	124.530342 cm-1
92 f	124.519404 cm-1
93 f	124.517932 cm-1
94 f	124.490562 cm-1
95 f	120.283321 cm-1
96 f	120.231270 cm-1
97 f	120.230184 cm-1
98 f	119.687426 cm-1
99 f	119.686831 cm-1
100 f	119.664871 cm-1
101 f	118.705245 cm-1
102 f	118.701471 cm-1
103 f	118.672999 cm-1
104 f	117.977674 cm-1
105 f	117.475657 cm-1
106 f	117.474674 cm-1
107 f	117.449522 cm-1

108 f	116.303550 cm-1
109 f	116.293997 cm-1
110 f	115.712217 cm-1
111 f	115.636521 cm-1
112 f	115.622368 cm-1
113 f	115.620605 cm-1
114 f	114.215915 cm-1
115 f	114.204048 cm-1
116 f	114.203932 cm-1
117 f	105.390923 cm-1
118 f	105.336583 cm-1
119 f	105.336224 cm-1
120 f	103.255597 cm-1
121 f	103.214325 cm-1
122 f	103.213635 cm-1
123 f	102.576424 cm-1
124 f	102.562288 cm-1
125 f	102.547928 cm-1
126 f	101.481127 cm-1
127 f	101.415193 cm-1
128 f	99.480960 cm-1
129 f	99.466128 cm-1
130 f	99.418246 cm-1
131 f	98.463143 cm-1
132 f	98.462015 cm-1
133 f	98.430052 cm-1
134 f	97.955717 cm-1

135 f	97.949681 cm-1
136 f	95.669067 cm-1
137 f	95.651752 cm-1
138 f	93.736848 cm-1
139 f	93.670527 cm-1
140 f	93.669284 cm-1
141 f	90.275764 cm-1
142 f	90.275181 cm-1
143 f	90.239315 cm-1
144 f	89.500314 cm-1
145 f	89.451227 cm-1
146 f	89.449778 cm-1
147 f	88.557440 cm-1
148 f	88.509162 cm-1
149 f	87.971942 cm-1
150 f	85.719509 cm-1
151 f	85.656995 cm-1
152 f	85.650446 cm-1
153 f	81.352762 cm-1
154 f	81.350543 cm-1
155 f	81.314130 cm-1
156 f	80.957805 cm-1
157 f	77.653238 cm-1
158 f	77.635871 cm-1
159 f	77.634017 cm-1
160 f	76.391176 cm-1
161 f	74.473961 cm-1

162 f	74.467352 cm-1
163 f	73.787004 cm-1
164 f	73.758942 cm-1
165 f	73.757655 cm-1
166 f	72.693832 cm-1
167 f	72.684093 cm-1
168 f	72.670422 cm-1
169 f	72.591312 cm-1
170 f	68.986312 cm-1
171 f	68.981394 cm-1
172 f	68.942350 cm-1
173 f	65.527997 cm-1
174 f	65.454709 cm-1
175 f	65.453659 cm-1
176 f	64.642665 cm-1
177 f	64.640316 cm-1
178 f	64.598151 cm-1
179 f	58.743402 cm-1
180 f	58.669549 cm-1
181 f	52.802558 cm-1
182 f	52.766415 cm-1
183 f	52.761236 cm-1
184 f	52.697560 cm-1
185 f	49.057866 cm-1
186 f	49.056609 cm-1
187 f	49.021709 cm-1
188 f	34.243464 cm-1

189 f	34.126837 cm-1
190 f	34.124727 cm-1
191 f	16.613894 cm-1
192 f	16.602548 cm-1
193 f	16.397284 cm-1
194 f/i	17.026834 cm-1
195 f/i	17.039677 cm-1
196 f/i	17.043370 cm-1
197 f/i	35.377571 cm-1
198 f/i	35.394129 cm-1