

**Supplementary information**

**Unexpected highly reversible topotactic CO<sub>2</sub>  
sorption/desorption capacity for potassium dititanate**

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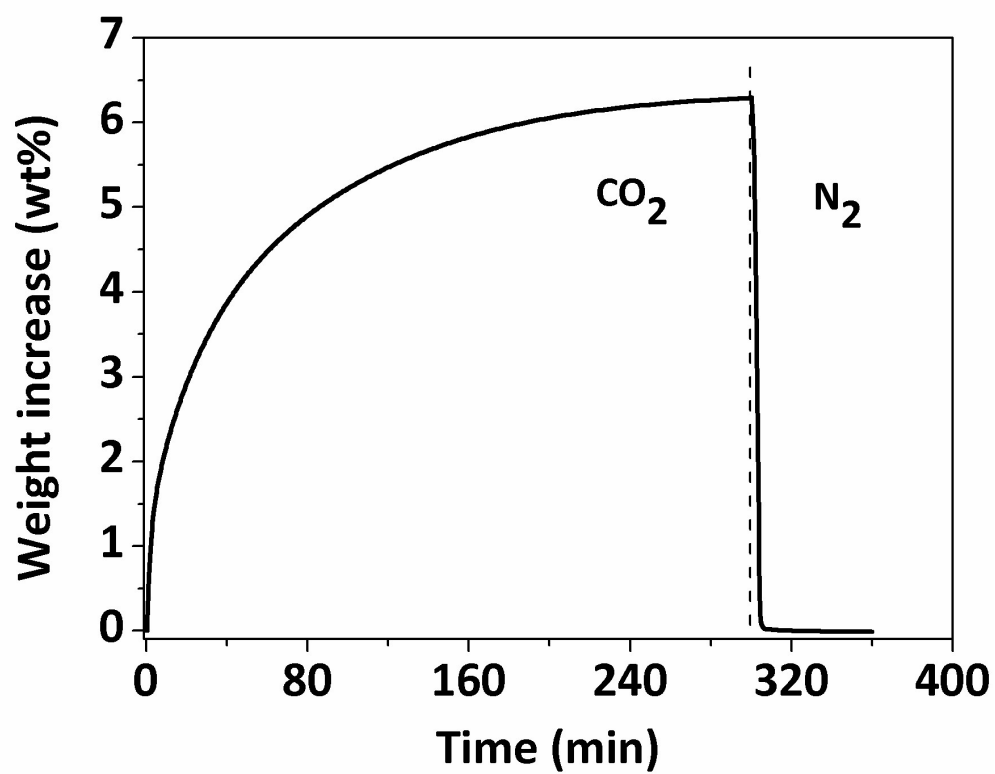
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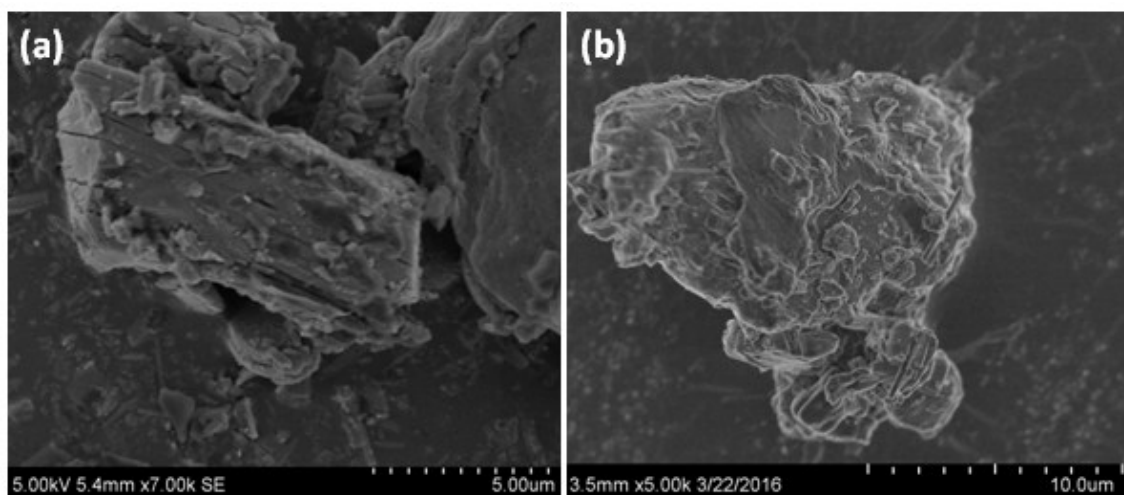
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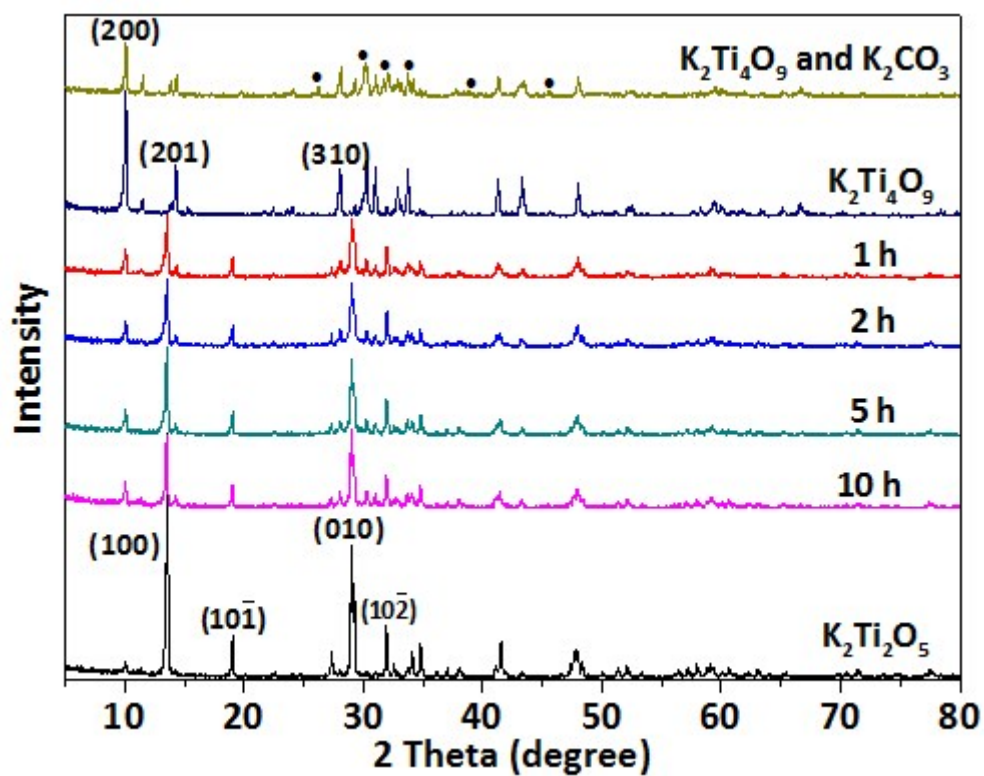
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**Figure S1.** One CO<sub>2</sub> sorption/desorption cycle over K<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub> both tested at 750 °C, which clearly indicates that the regeneration of the adsorbent K<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub> in N<sub>2</sub> is very rapid (< 6 min).



**Figure S2.** SEM images of (a) fresh  $K_2Ti_2O_5$ , and (b)  $K_2Ti_2O_5$  thermally treated at 750 °C in  $N_2$  for 10.0 h.



**Figure S3.** XRD patterns of fresh  $K_2Ti_2O_5$ , the mixture of  $K_2Ti_4O_9$  and  $K_2CO_3$ , and the thermally treated mixture of  $K_2Ti_4O_9$  and  $K_2CO_3$  with a ratio of 1:1 at 750 °C for 1.0, 2.0, 5.0, and 10.0 h, respectively, (●)  $K_2CO_3$ .