

Electronic Supplementary Information (ESI)

Facile hydrothermal synthesis of SnCoS₄/graphene composites with excellent electrochemical performances for reversible lithium ion storage

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TGA was carried out to quantify the composition of SnS₂/GNS, CoS₂/GNS and SnCoS₄/GNS composites. As shown in Fig. S1(a), TGA curve of SnS₂/GNS exhibits three regions of weight loss in the air atmosphere. The initial weight loss below 200 °C is due to the evaporation of adsorbed water in the composite.^{1,2} The second weight loss between 200 °C to 425 °C is attributed to the transformation from SnS₂ to SnO₂¹ and oxidation of oxygen-contained functional group of reduced graphene oxide.^{3,4} The third weight loss in the temperature ranging from 430 °C to 600 °C is assigned to the removal of graphene.^{1,3} Thus, the remaining residue is SnO₂ for SnS₂/GNS composite. As shown in Fig. S1(b), there are several stages of weight loss for CoS₂/GNS, which is similar to that of CoS₂/carbon nanotubes and CoS₂/graphene nanocomposite reported elsewhere.^{5,6} Those several weight losses for CoS₂/GNS in the air atmosphere should include the evaporation of adsorbed water, oxidation of oxygen-contained functional group, reduced graphene oxide and CoS₂. According to the literatures,^{5,6} the weight loss from 600 °C to 800 °C should be ascribed to the oxidation of CoS₂ to Co₃O₄. Thus, the remaining residue should be Co₃O₄ for CoS₂/GNS composites. Fig. S1(c) shows that the TGA curve of SnCoS₄/GNS composite also displays several regions of weight loss in the air atmosphere. According to the literatures,^{1,5,6} SnS₂ and CoS₂ should be transferred to SnO₂ and Co₃O₄ over 800°C, respectively. The reduced graphene oxide (or graphene) also should be completely oxidized to CO₂ over 800 °C.³⁻⁵ Therefore, the

remaining residue was SnO₂ and Co₃O₄ for SnCoS₄/GNS composite.

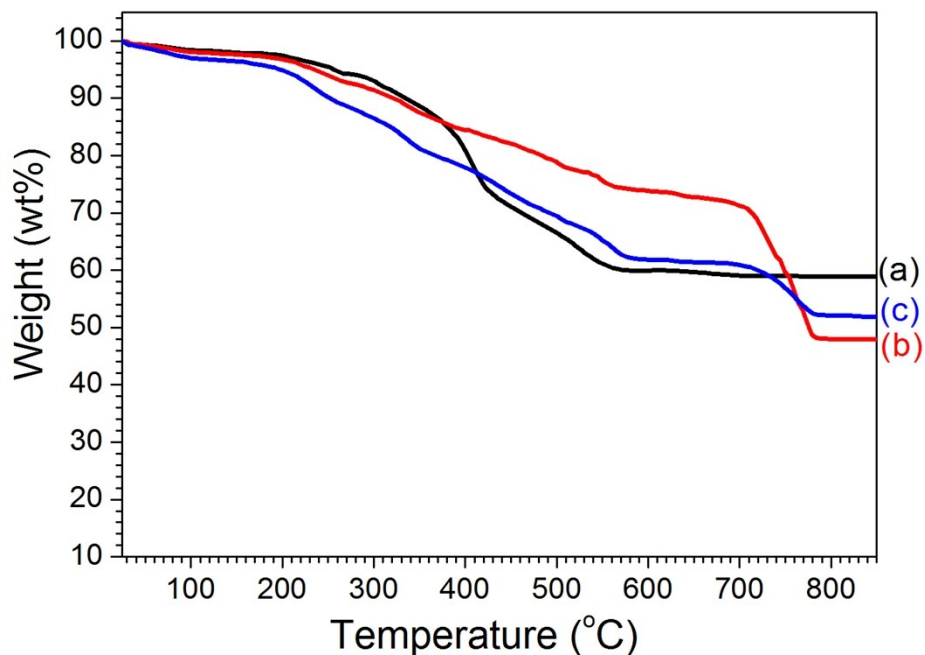


Fig. S1. The TGA curves of (a) SnS₂/GNS, (b) CoS₂/GNS and (c) SnCoS₄/GNS.

Based on the above investigations, the remaining residue should be SnO₂ for SnS₂/GNS composites, Co₃O₄ for CoS₂/GNS composites, SnO₂ and Co₃O₄ for SnCoS₄/GNS composite. Therefore, the SnS₂ and graphene content of SnS₂/GNS composite can be calculated to be 74.6 wt% and 25.4 wt%, respectively. The CoS₂ and graphene content of CoS₂/GNS composite can be calculated to be 75.8 wt% and 24.2 wt%, respectively. The SnCoS₄ and graphene content of SnCoS₄/GNS composite can be calculated to be 72.7 wt% and 27.3 wt%, respectively.

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