

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A

This journal is © The Royal Society of Chemistry 2016

Hierarchically Structured Layered-Double- Hydroxide@Zeolitic- Imidazolate-Framework Derivatives for High-Performance Electrochemical Energy Storage

Yibo Dou,^{a,b} Jian Zhou,^a Fan Yang,^a Min-Jian Zhao,^a Zuoren Nie,^{b,*} and Jian-Rong Li^{a,*}

^a Beijing Key Laboratory for Green Catalysis and Separation and Department of Chemistry and Chemical Engineering, College of Environmental and Energy Engineering, Beijing University of Technology, Beijing 100124, P. R. China.

E-mail: jrli@bjut.edu.cn. Tel: +86-10-67392332

^b College of Materials Science and Engineering, Beijing University of Technology, Beijing 100124, P. R. China.

E-mail: zrnjie@bjut.edu.cn. Tel: +86-10-67391536

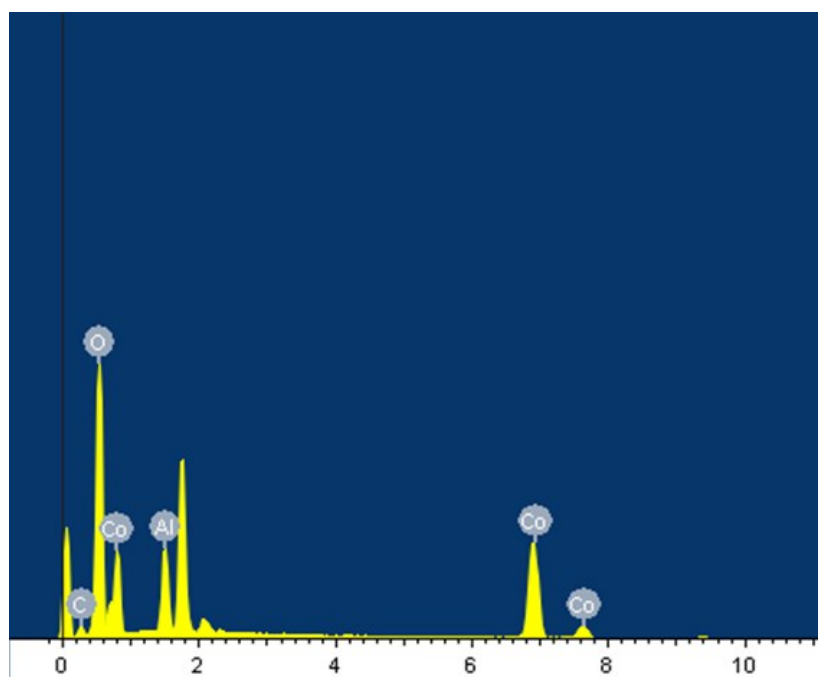


Figure S1. EDX spectrum of CoAl-LDH film.

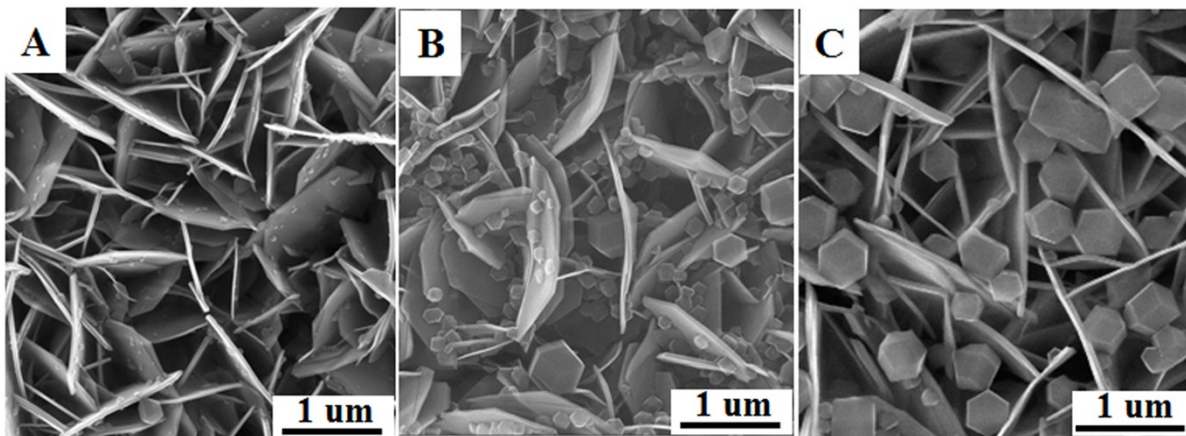


Figure S2. SEM images of the CoAl-LDH@ZIF-67 film as a function of reaction time (A) 10 h, (B) 18 h, and (C) 24 h.

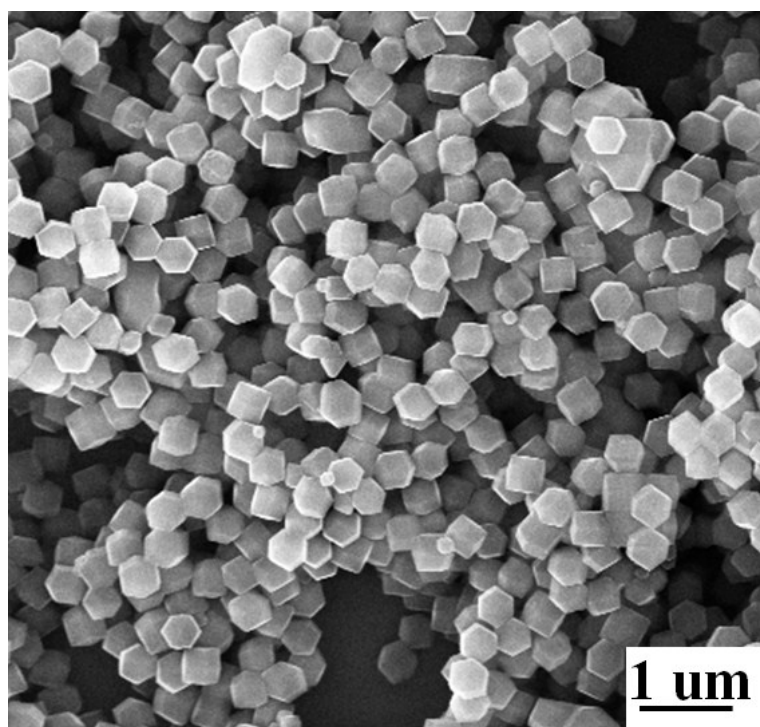


Figure S3. SEM image of ZIF-67.

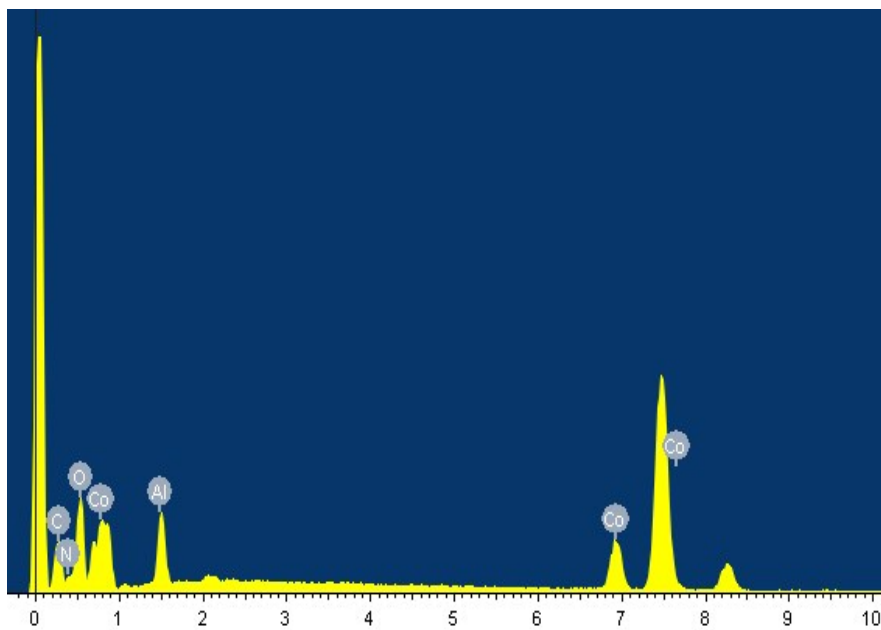


Figure S4. EDX spectrum of CoAl-LDH@ZIF-67 film.

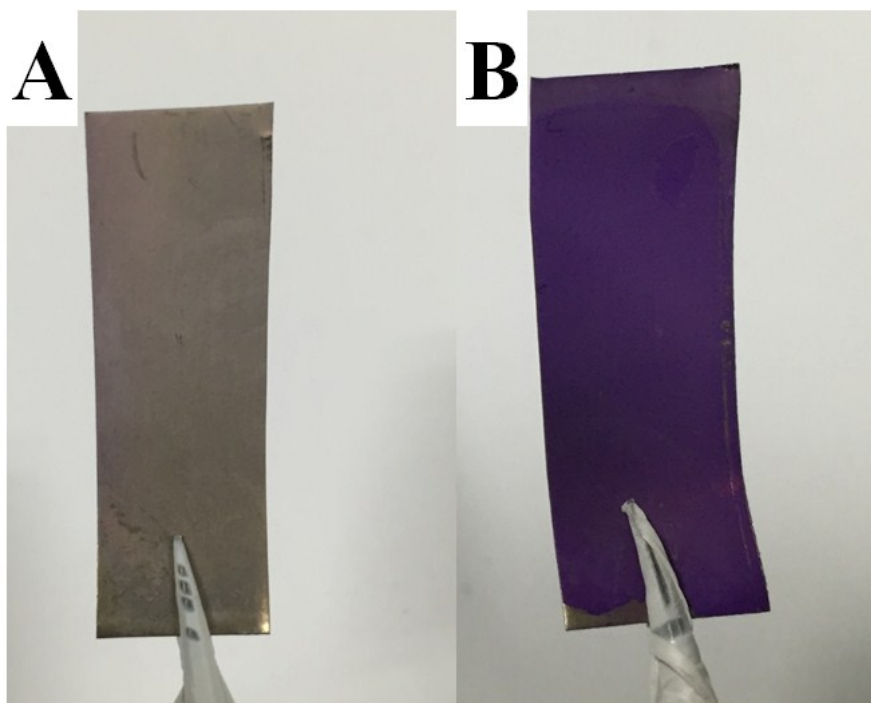


Figure S5. Digital photographs of (A) CoAl-LDH and (B) CoAl-LDH@ZIF-67 films.

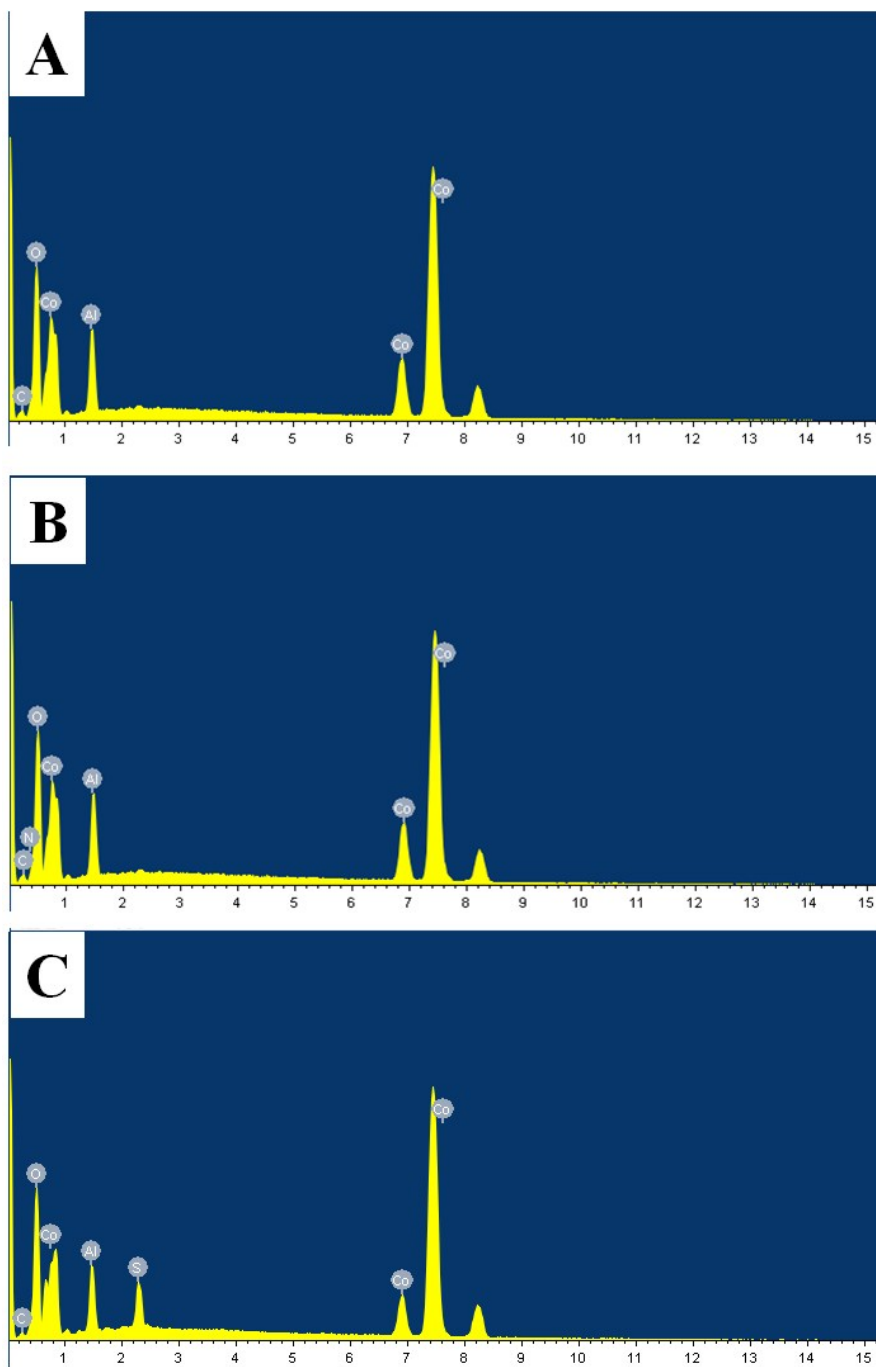


Figure S6. EDX spectrum of (A) MMO@Co₃O₄, (B) Spinelle@C, and (C) LDH@CoS, respectively.

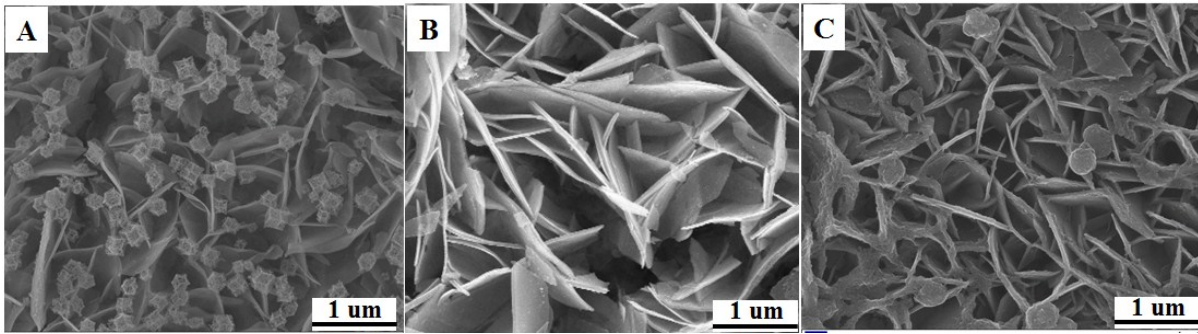


Figure S7. SEM images of (A) MMO@Co₃O₄, (B) Spinel@C, and (C) LDH@CoS obtained by oxidation at 300 °C under air for 5 h, carbonization at 800 °C under N₂ for 5 h, and sulfurization with thioacetamide for 3 h, respectively.

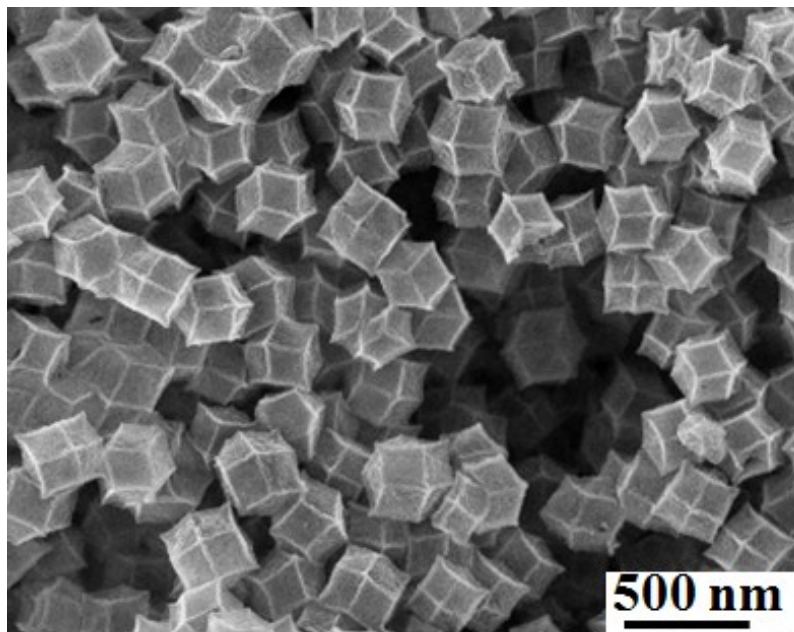


Figure S8. SEM image of Co₃O₄ powder derived from ZIF-67 powder.

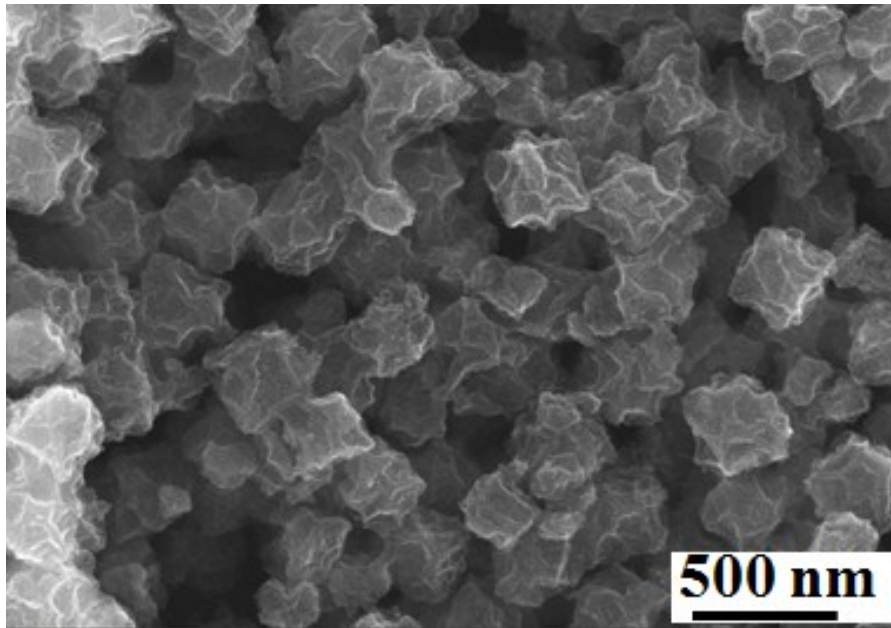


Figure S9. SEM image of nitrogen-doped carbon powder derived from ZIF-67 powder.

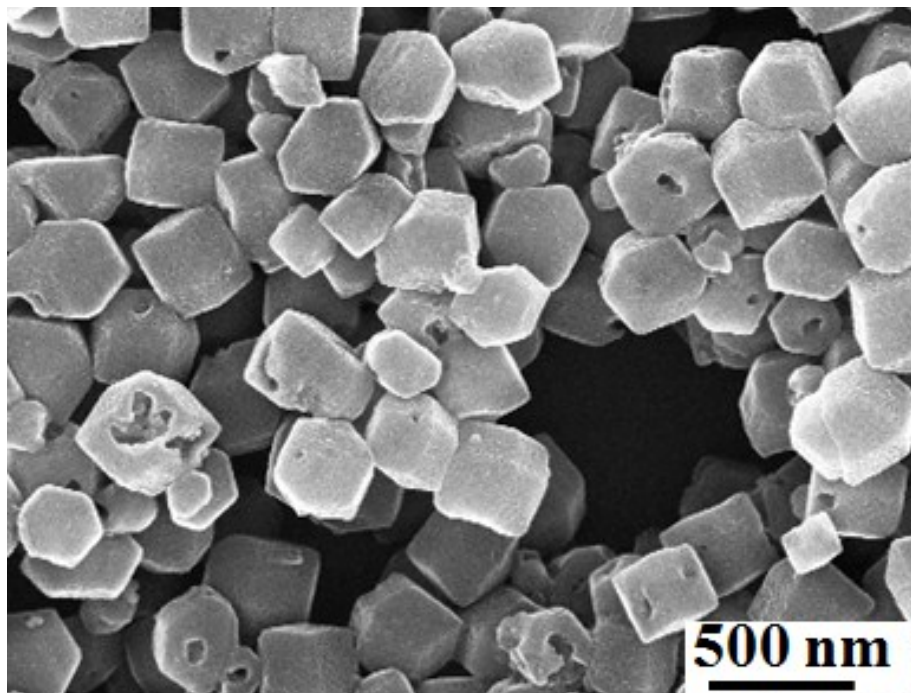


Figure S10. SEM image of CoS powder derived from ZIF-67 powder.

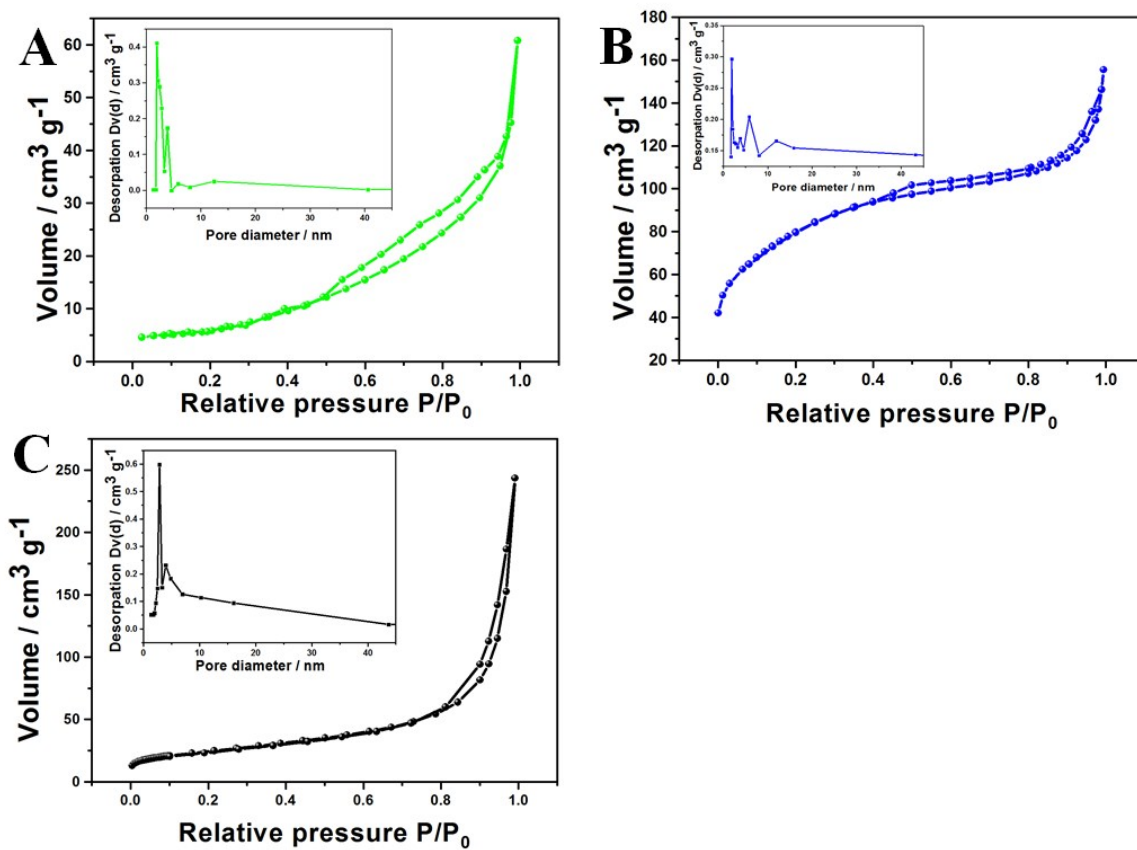


Figure S11. N_2 adsorption-desorption isotherms and pore size distributions (inset) of (A) $MMO@Co_3O_4$, (B) $Spinelle@C$, and (C) $LDH@CoS$, respectively.

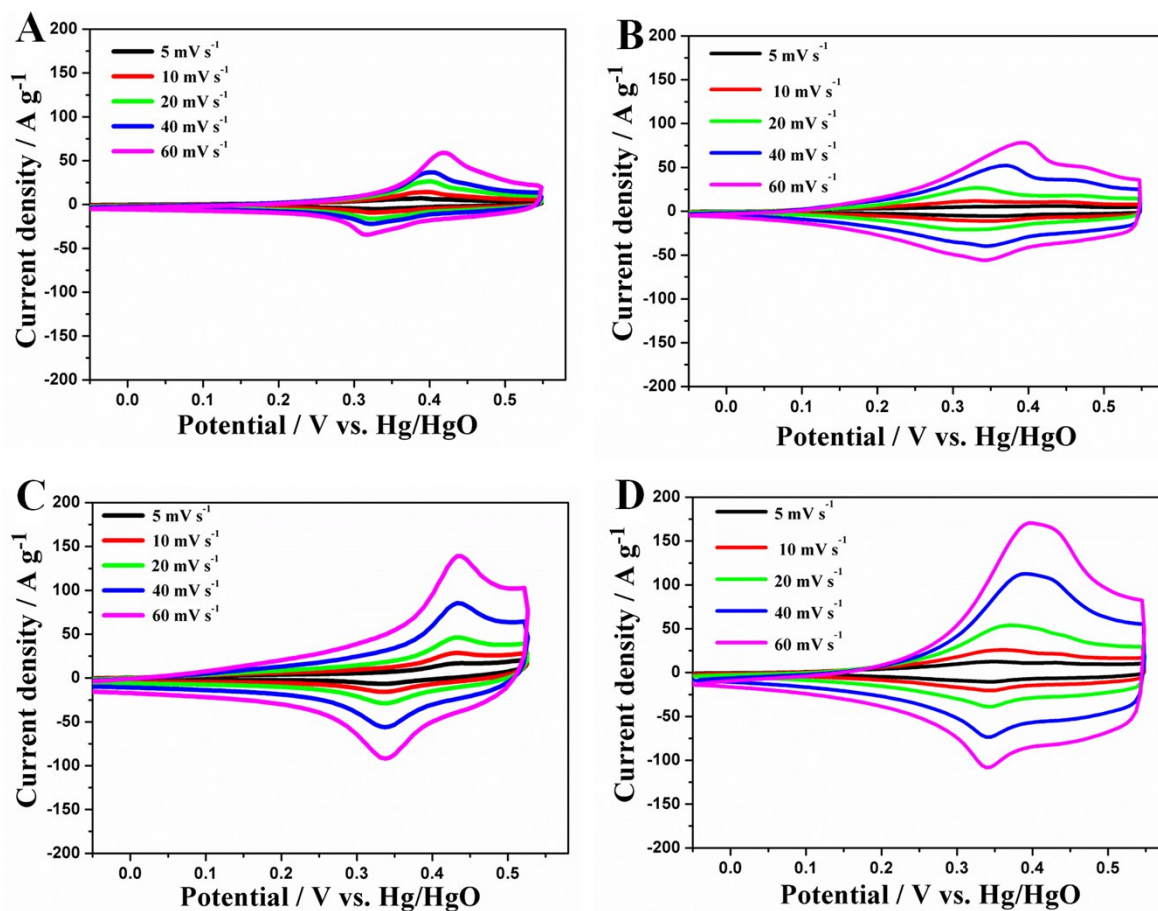


Figure S12. CV curves of (A) LDH, (B) MMO@Co₃O₄, (C) Spinelle@C, and (D) LDH@CoS at varied scan rates from 5 to 60 mV/s.

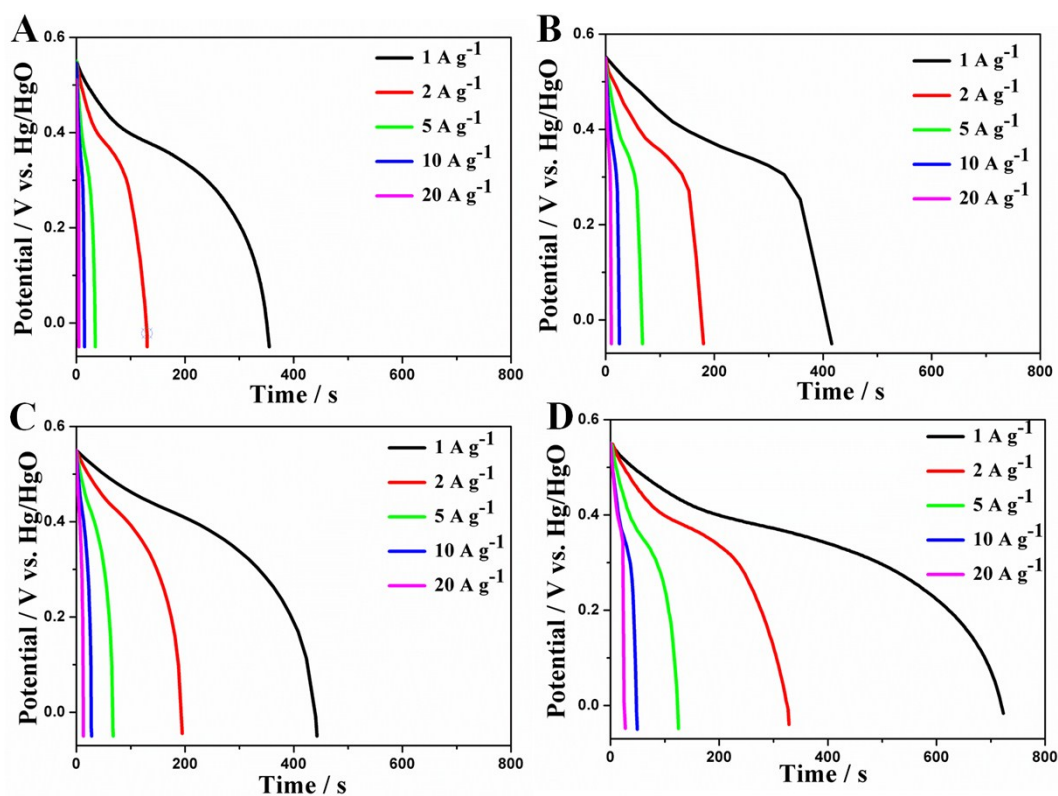


Figure S13. Galvanostatic discharge curves of (A) LDH, (B) $\text{MMO@Co}_3\text{O}_4$, (C) Spinelle@C, and (D) LDH@CoS at varied current densities from 1 to 20 A/g.

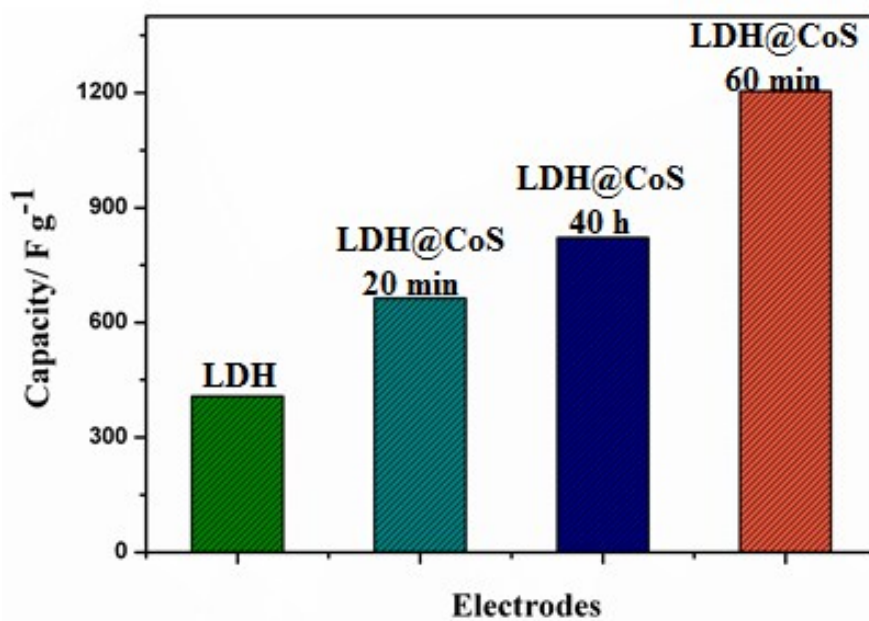


Figure S14. Specific capacity of LDH and LDH@CoS with varied amounts of CoS that derived based on different sulfurization time of 20, 40, and 60 min, respectively.

Table S1. Comparison study of LDH@ZIF-67 derivatives in this work and previously reported MOFs derivatives toward electrochemical energy storage.

| MOFs | MOFs derivatives | Specific capacitance (F g ⁻¹) | Current density (A g ⁻¹) | Reference |
|------------|------------------------------------|---|--------------------------------------|-----------|
| MOF-5 | NPC | 258 | 0.25 | 1 |
| Bi/MOF-5 | HPCs | 241 | 0.1 | 2 |
| Al-PCP | Carbon | 232.8 | 0.1 | 3 |
| ZIF-11 | Nitrogen-doped porous carbon | 307 | 1 | 4 |
| ZIF-8 | PCPs | 245 | 1 | 5 |
| IRMOF-3 | N doped Carbon | 213 | 0.5 | 6 |
| HKUST-1 | Cu _{1.96} S-C | 200 | 0.5 | 7 |
| ZIF-67 | Co ₃ O ₄ //C | 101 | 2 | 8 |
| Co-MOF | Co ₃ O ₄ | 150 | 1 | 9 |
| LDH@ZIF-67 | MMO@Co ₃ O ₄ | 692 | 1 | This work |
| LDH@ZIF-67 | Spinelle@C | 781 | 1 | This work |
| LDH@ZIF-67 | LDH@CoS | 1205 | 1 | This work |

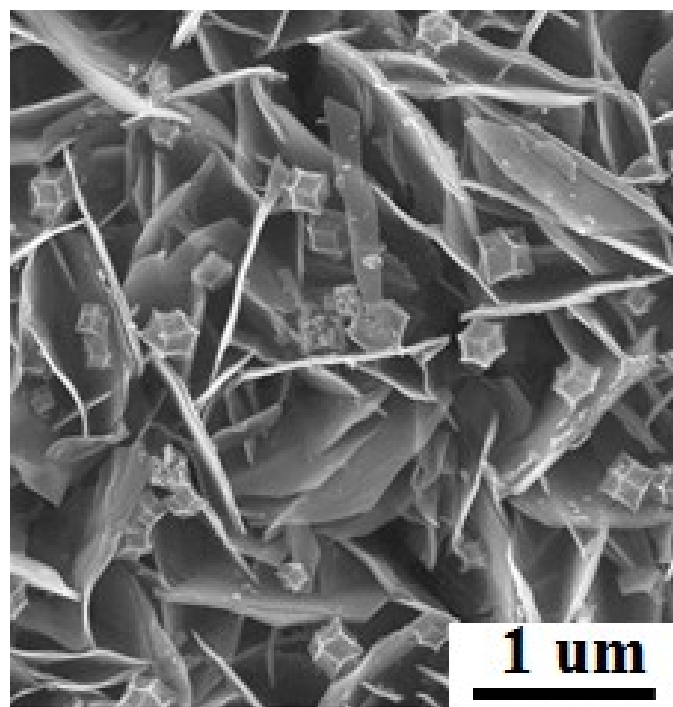


Figure S15. SEM image of MMO@Co₃O₄ after 2000 galvanostatic charge-discharge cycles.

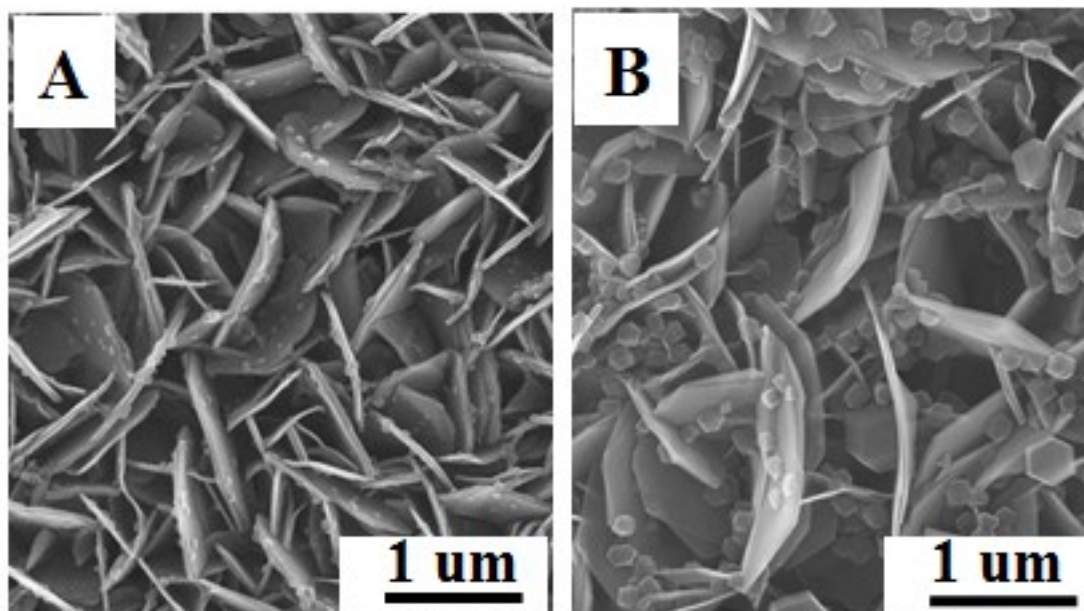


Figure S16. SEM images of (A) Spinelle@C and (B) LDH@CoS after 2000 galvanostatic charge-discharge cycles.

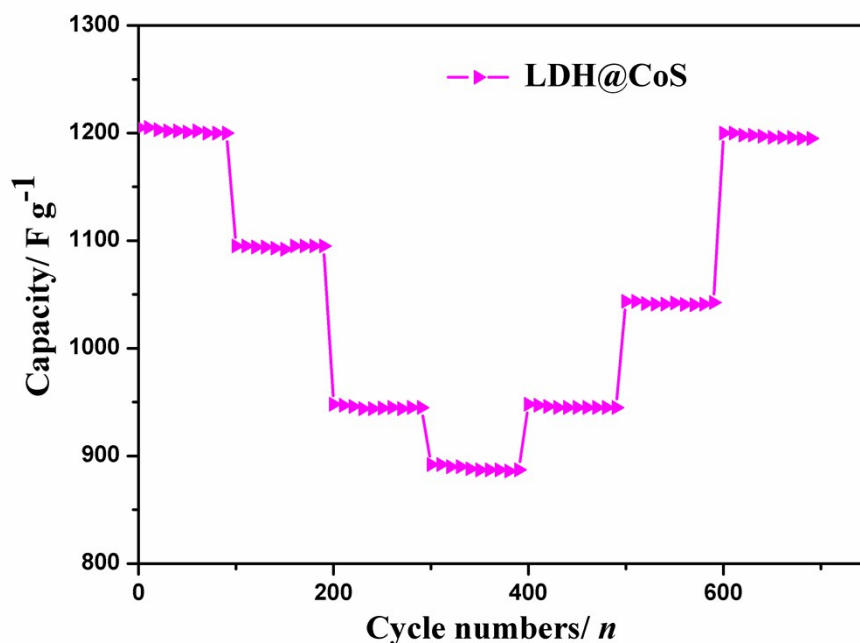


Figure S17. Cycling stability of the LDH@CoS at consecutively varied current densities.

Reference

- 1 B. Liu, H. Shioyama, T. Akita and Q. Xu, *J. Am. Chem. Soc.*, 2008, **130**, 5390–5391.
- 2 S. Mo, Z. Sun, X. Huang, W. Zou, J. Chen and D. Yuan, *Synthetic Met.*, 2012, **162**, 85–88.
- 3 X. Yan, X. Li, Z. Yan and S. Komarneni, *Appl. Surf. Sci.*, 2014, **308**, 306–310.
- 4 F. Hao, L. Li, X. Zhang and J. Chen, *Mater. Res. Bull.*, 2015, **66**, 88–95.
- 5 H. Yi, H. Wang, Y. Jing, T. Peng and X. Wang, *J. Power Sources*, 2015, **285**, 281–290.
- 6 J. Jeon, R. Sharma, P. Meduri, B. W. Arey, H. T. Schaefer, J. L. Lutkenhaus, J. P. Lemmon, P. K. Thallapally, M. I. Nandasiri, B. P. McGrail and S. K. Nune, *ACS Appl. Mater. Inter.*, 2014, **6**, 7214–7222.
- 7 R. Wu, D. P. Wang, V. Kumar, K. Zhou, A. W. K. Law, P. S. Lee, J. Lou and Z. Chen, *Chem. Commun.*, 2015, **51**, 3109–3112.
- 8 R. R. Salunkhe, J. Tang, Y. Kamachi, T. Nakato, J. H. Kim and Y. Yamauchi, *ACS Nano*, 2015, **9**, 6288–6296.
- 9 F. Meng, Z. Fang, Z. Li, W. Xu, M. Wang, Y. Liu, J. Zhang, W. Wang and X. Guo, *J. Mater. Chem. A*, 2013, **1**, 7235–7241.