

Defining a performance map of porous carbon sorbents for high-pressure carbon dioxide uptake and carbon dioxide-methane selectivity

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Table S1. Summary of elemental analysis, physical properties and CO₂ uptake for NPC and SPC precursors.

Sample ^a	C (wt%) ^b	O (wt%) ^b	N (wt%) ^b	S (wt%) ^b	Surface area $S_{\text{BET}} (\text{m}^2 \text{g}^{-1})$	Total pore volume (cm ³ g ⁻¹)	CO ₂ uptake at 30 bar (mmol.g ⁻¹)
Polypyrrole	65.75	18.26	15.98	0.00	62.00	0.03	3.09
PTh	63.35	11.84	0.00	24.81	75.00	0.04	2.85

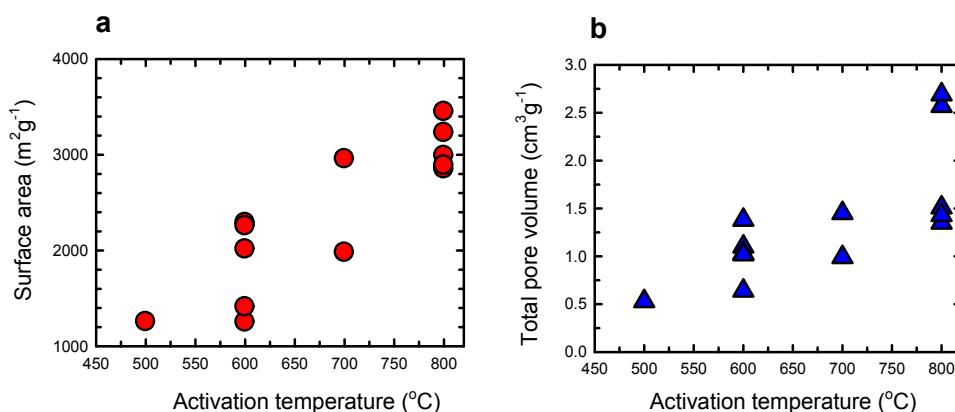


Fig. S1 Estimated (a) surface area and (b) total pore volume as a function of activation temperature for PC, NPC and SPC samples.

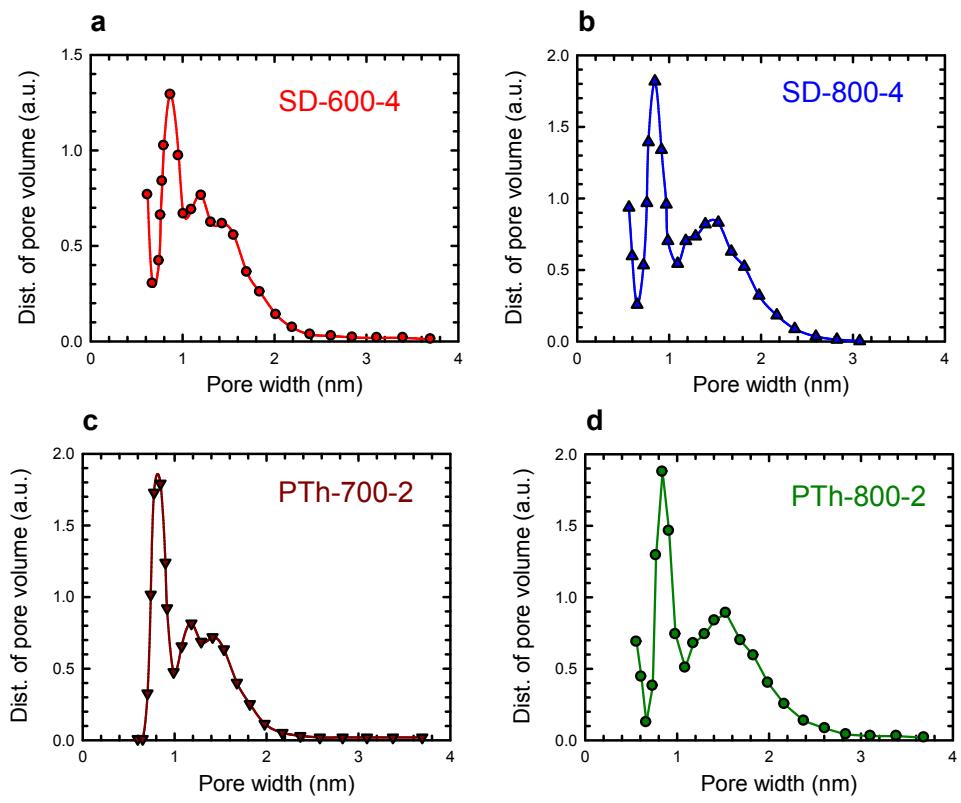


Fig. S2 Distribution of pore volumes as a function of pore width for (a) SD-600-4 (b) SD-800-4 (c) PTh-700-2 and (d) PTh-800-2 samples.

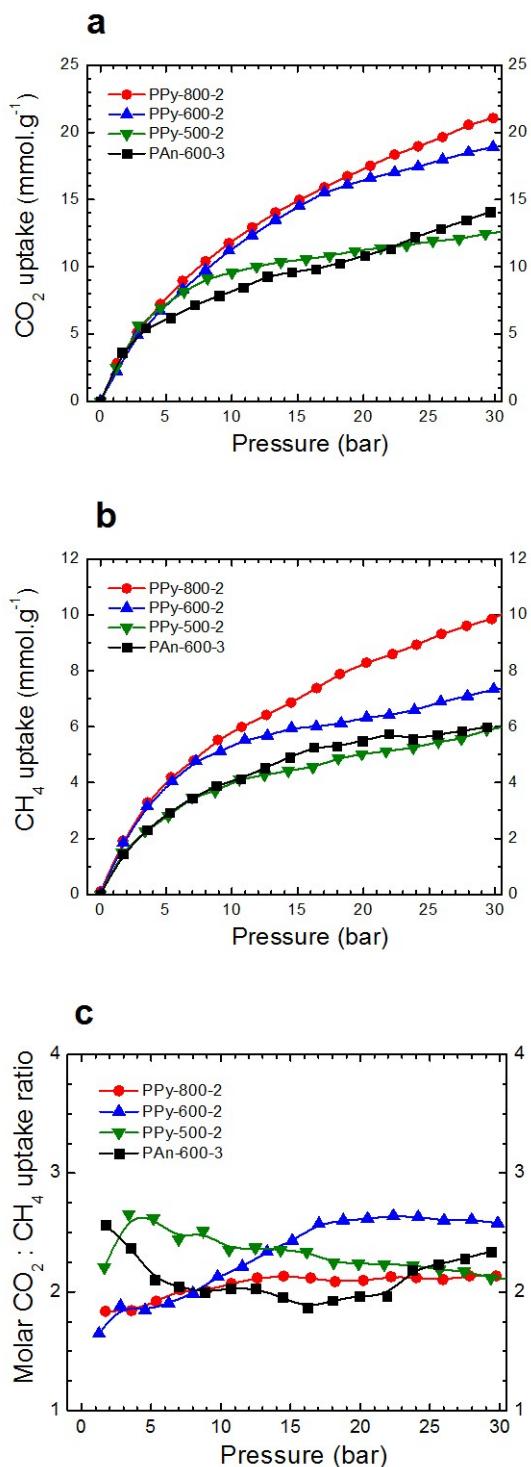


Fig. S3 Room temperature volumetric (a) CO_2 and (b) CH_4 adsorption uptake measurements for other NPC samples. (c) The molar $\text{CO}_2:\text{CH}_4$ uptake ratio as a function of gas pressure for PC, NPC, and SPC samples. Experiments were performed at 24 °C.

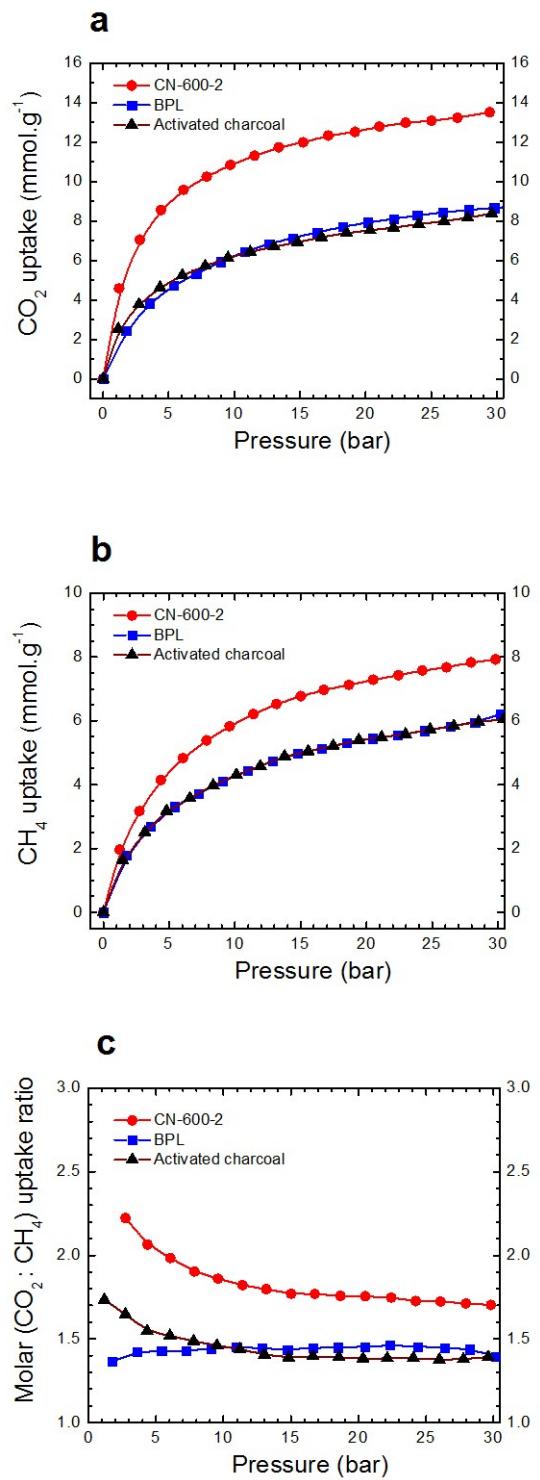


Fig. S4 Room temperature volumetric (a) CO₂ and (b) CH₄ adsorption uptake measurements for other PC samples. (c) The molar CO₂:CH₄ uptake ratio as a function of gas pressure for PC, NPC, and SPC samples. Experiments were performed at 24 °C.

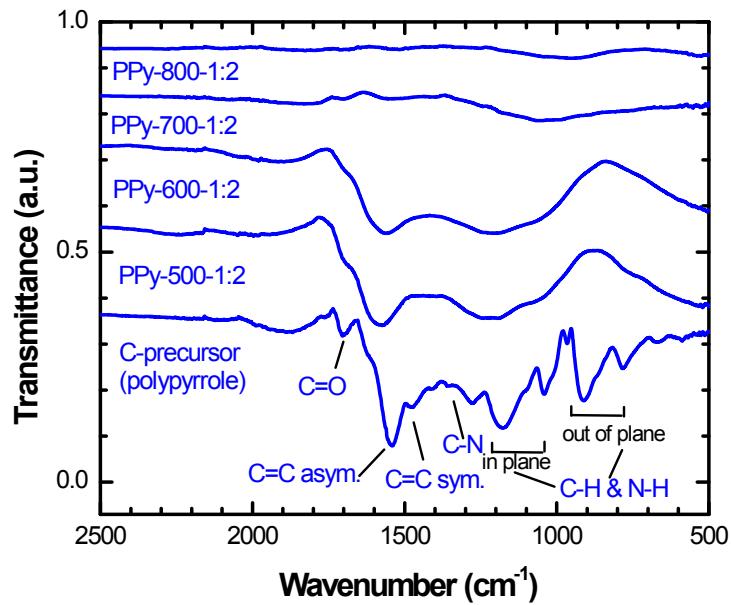


Fig. S5. Characterization of chemical composition of N-containing polymer precursor and porous carbon samples activated at increasing temperatures by FTIR spectroscopy.

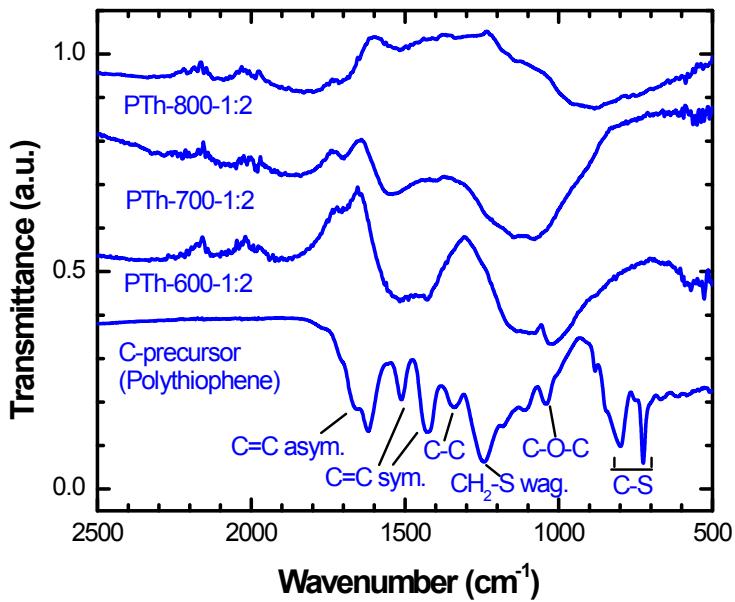


Fig. S6 Characterization of chemical composition of S-containing polymer precursor and porous carbon samples activated at increasing temperatures by FTIR spectroscopy.

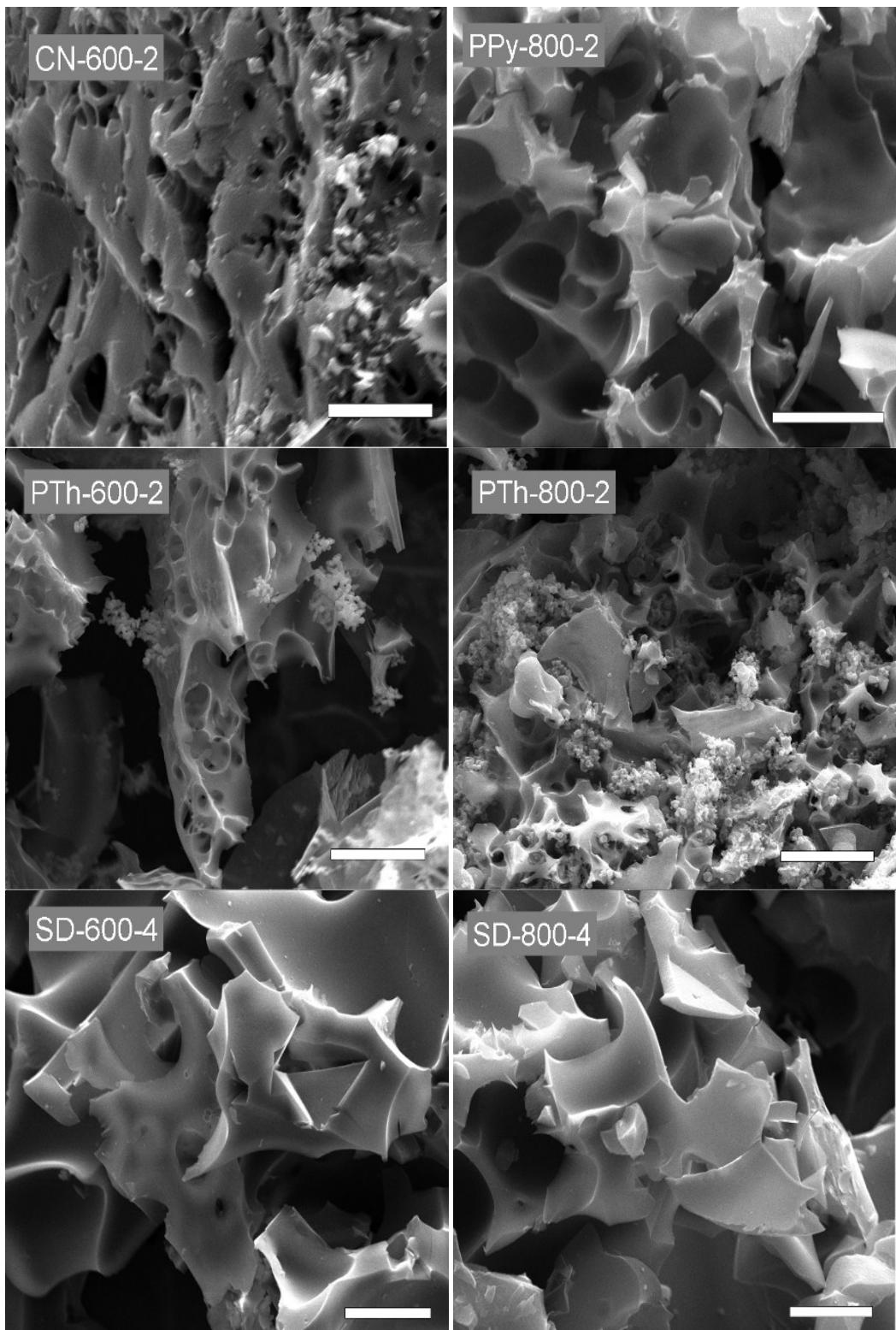


Fig S4. Scanning electron microscope (SEM) images of different activated PC samples synthesized from coconut shell, polypyrrole, polythiophene and sawdust respectively.