

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A.
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Electronic Supporting Information

Integrating photosystem II into porous TiO₂ nanotube network toward highly-efficient photo-bioelectrochemical cells

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Incident Photon-to-current Conversion Efficiency Measurements

The incident photon-to-current conversion efficiencies (IPCE) of the PSII/(TiO₂)₅-NTs-modified ITO electrode were calculated utilizing the following equation:

$$\text{IPCE} = jhc/eP\lambda$$

where h is Planck's constant, c is the vacuum velocity of light, λ is the central wavelength of the incident light and e is the charge of an electron. The generated photocurrent density (j) and the light power density (P) were recorded in the corresponding photoelectrochemical measurement. It was supposed that all incident photons were absorbed by the hybrid photoanode.

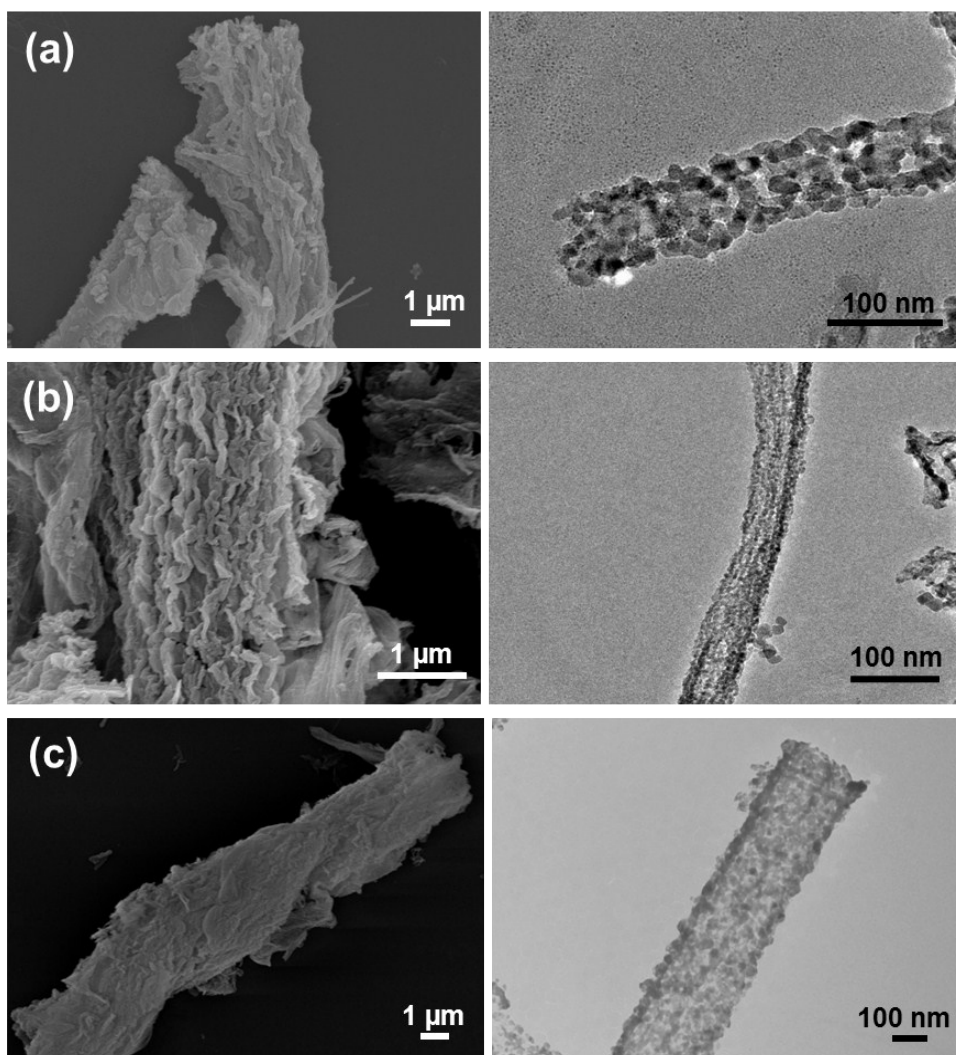


Fig. S1 Left column shows overview FE-SEM images of $(\text{TiO}_2)_5$ -NTs (a), $(\text{TiO}_2)_{10}$ -NTs (b) and $(\text{TiO}_2)_{20}$ -NTs (c). Right column displays TEM images of the corresponding individual titania nanotube.

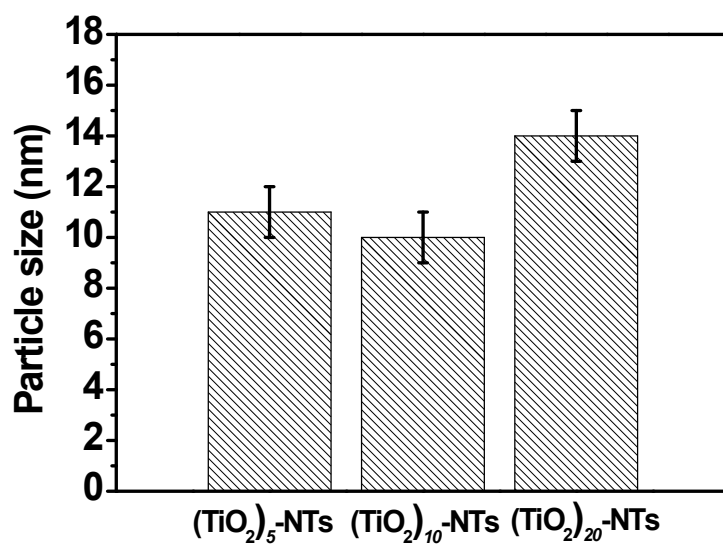


Fig. S2 Average sizes of the nanoparticles in $(\text{TiO}_2)_5\text{-NTs}$, $(\text{TiO}_2)_{10}\text{-NTs}$ and $(\text{TiO}_2)_{20}\text{-NTs}$. Error bars represent standard error of the mean, for $N = 5$ independent experiments, and fifty nanoparticles were taken in each experiment.

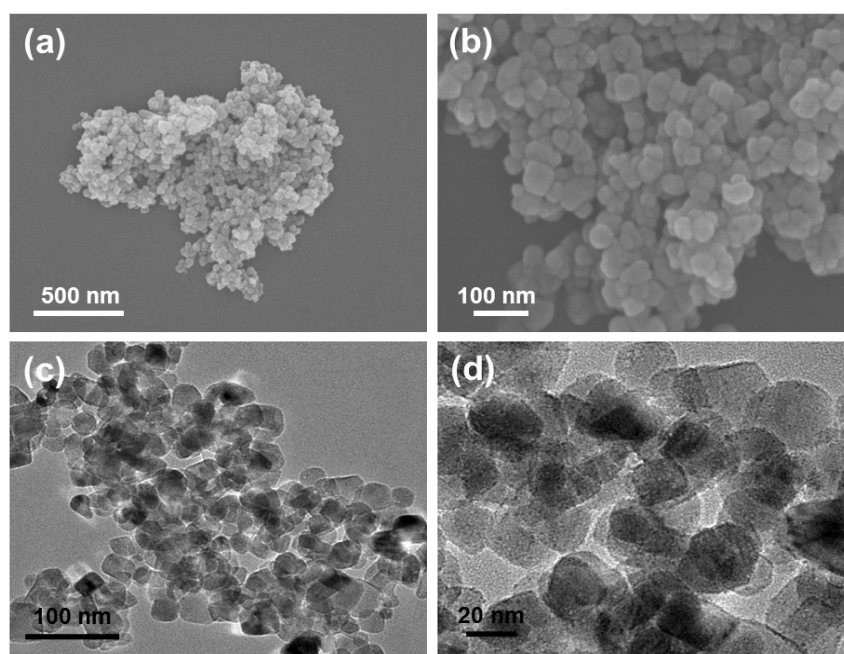


Fig. S3 Overview (a) and enlarged (b) FE-SEM images of the commercial titania nanopowder ($\text{TiO}_2(\text{C})$). Overview (c) and enlarged (d) TEM images of the commercial titania nanopowder ($\text{TiO}_2(\text{C})$).

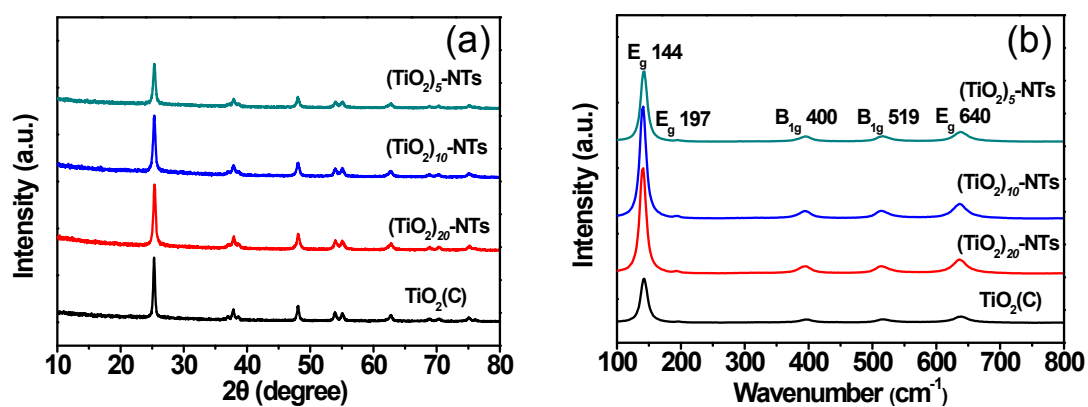


Fig. S4 (a) XRD patterns and (b) Raman spectra of $(\text{TiO}_2)_5\text{-NTs}$, $(\text{TiO}_2)_{10}\text{-NTs}$, $(\text{TiO}_2)_{20}\text{-NTs}$ and $\text{TiO}_2(\text{C})$.

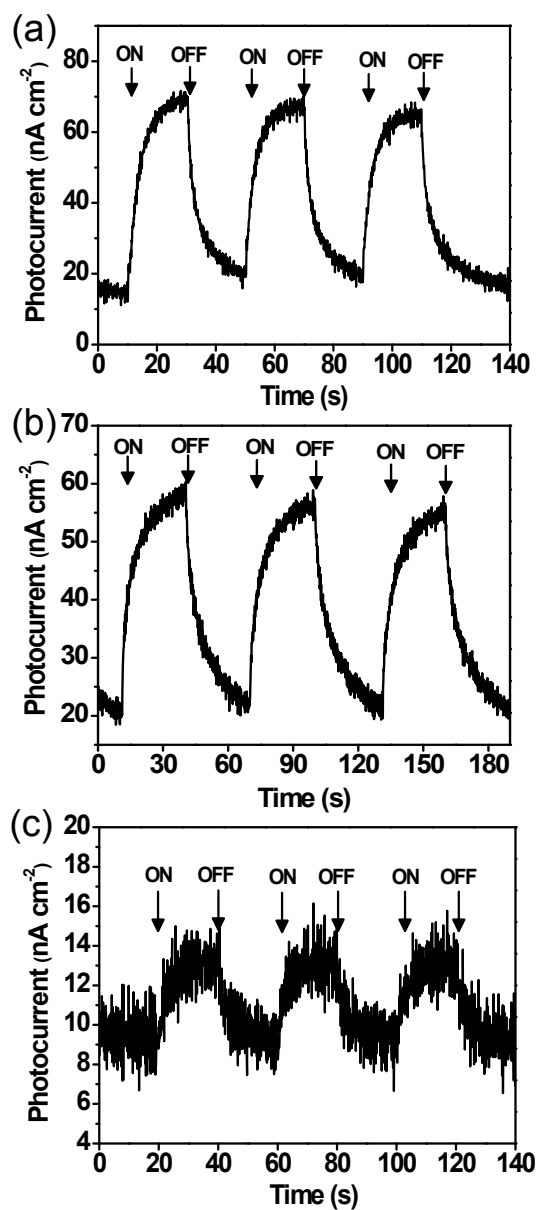


Fig. S5 DET photocurrent responses of (a) PSII/(TiO₂)₁₀-NTs, (b) PSII/(TiO₂)₂₀-NTs and (c) PSII/TiO₂(C)-modified ITO electrodes to repetitive illumination by red light ($550 < \lambda < 800$ nm, 10 mW cm^{-2}) and darkness cycles. A bias potential of +0.25 V vs. SCE was applied at 25 °C.

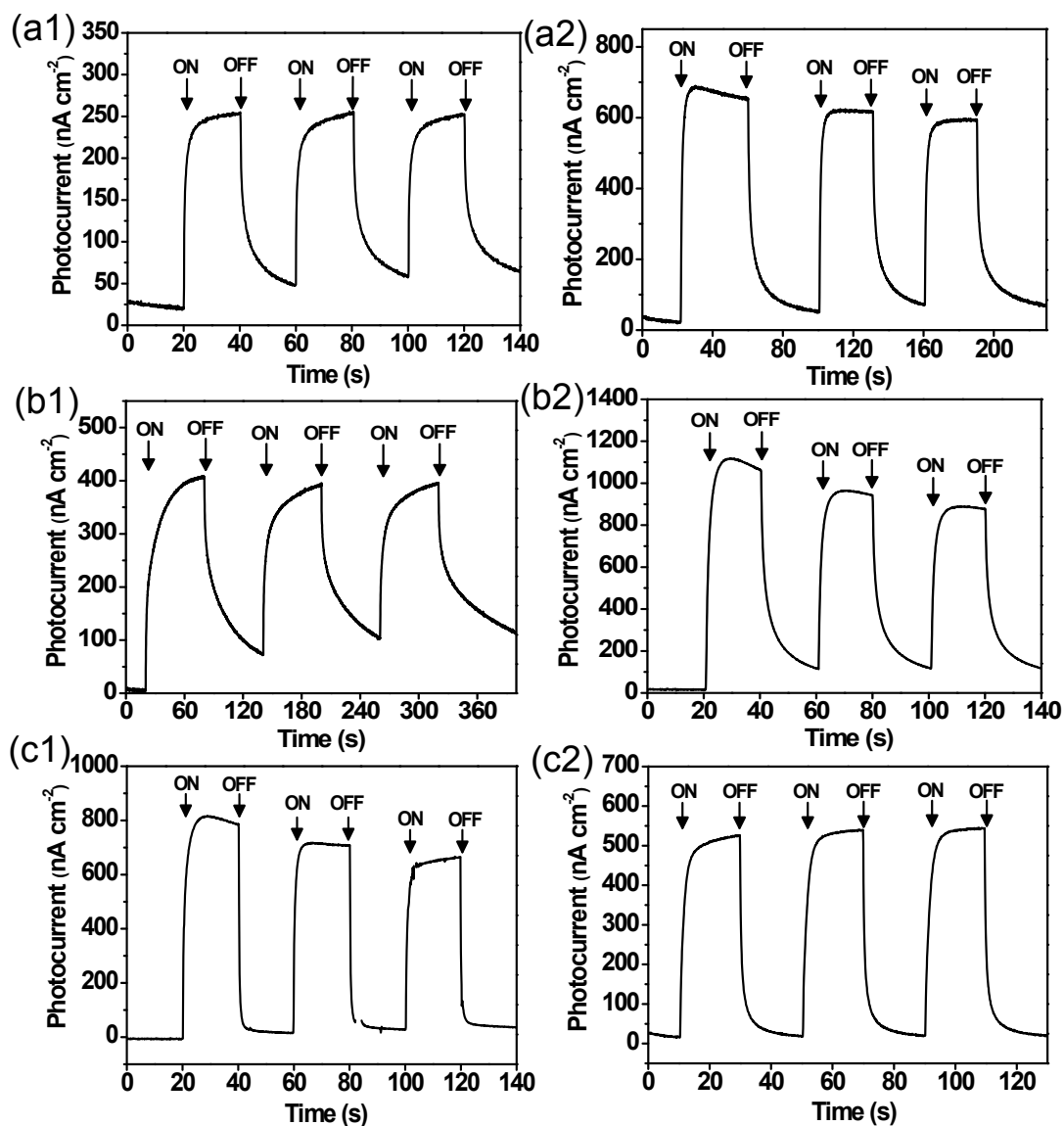


Fig. S6 DET photocurrent responses of bare titania (a1) (TiO₂)₁₀-NTs, (b1) (TiO₂)₂₀-NTs and (c1) TiO₂(C)-modified ITO electrodes to repetitive illumination by white light and darkness cycles. DET photocurrent responses of (a2) PSII/(TiO₂)₁₀-NTs, (b2) PSII/(TiO₂)₂₀-NTs and (c2) PSII/TiO₂(C)-modified ITO electrodes to repetitive illumination by white light and darkness cycles. A bias potential of +0.25 V vs. SCE was applied at 25 °C.

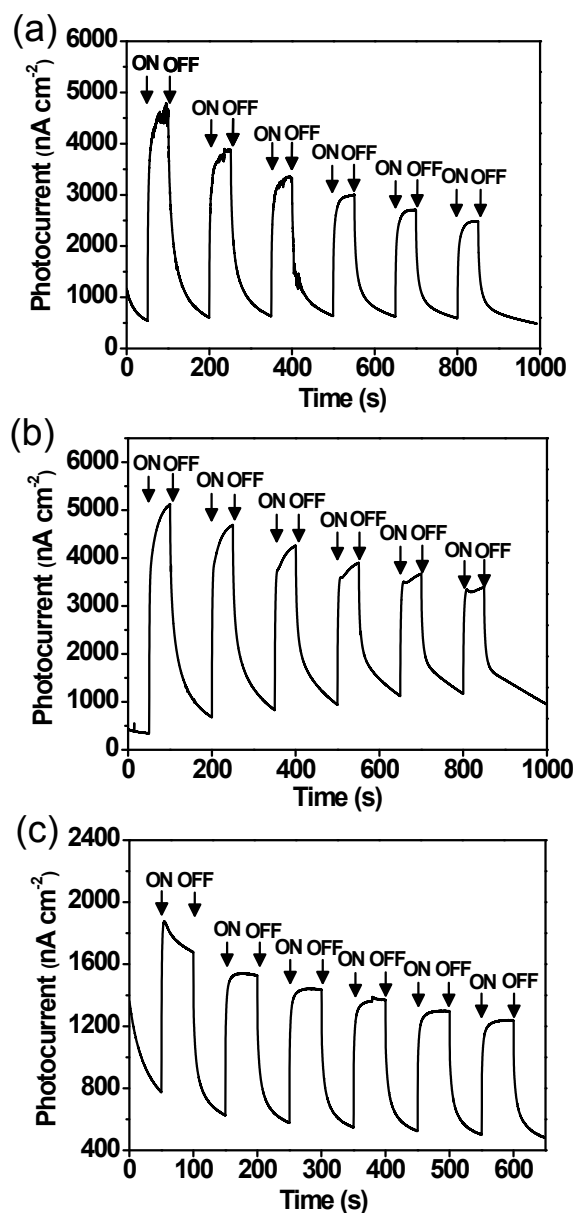


Fig. S7 MET photocurrent responses of (a) PSII/(TiO₂)₁₀-NTs, (b) PSII/(TiO₂)₂₀-NTs and (c) PSII/TiO₂(C)-modified ITO electrodes with 1 mM DCBQ in the electrolyte to repetitive illumination by red light ($550 < \lambda < 800$ nm, 10 mW cm^{-2}) and darkness cycles. A bias potential of +0.25 V vs. SCE was applied at 25 °C.

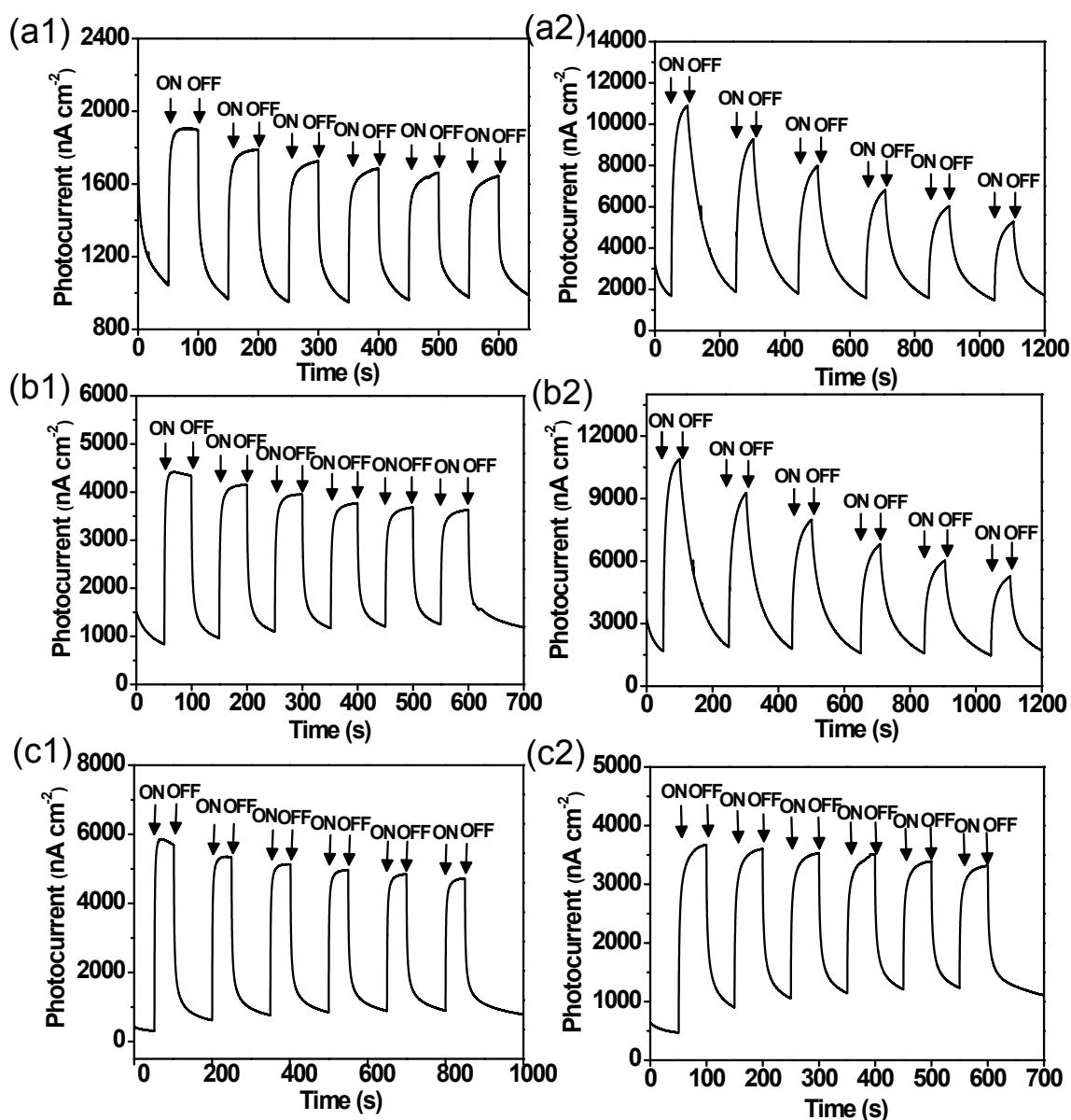


Fig. S8 MET photocurrent responses of bare titania (a1) $(\text{TiO}_2)_{10}$ -NTs, (b1) $(\text{TiO}_2)_{20}$ -NTs and (c1) $\text{TiO}_2(\text{C})$ -modified ITO electrodes with 1 mM DCBQ in the electrolyte to repetitive illumination by white light and darkness cycles. MET photocurrent responses of (a2) PSII/ $(\text{TiO}_2)_{10}$ -NTs, (b2) PSII/ $(\text{TiO}_2)_{20}$ -NTs and (c2) PSII/ $\text{TiO}_2(\text{C})$ -modified ITO electrodes with 1 mM DCBQ in the electrolyte to repetitive illumination by white light and darkness cycles. A bias potential of +0.25 V vs. SCE was applied at 25 °C.

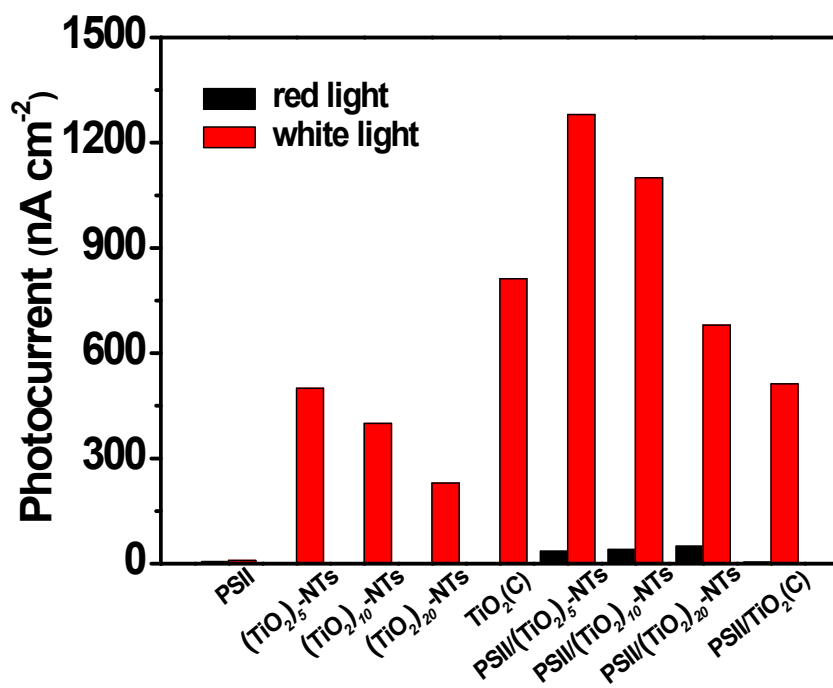


Fig. S9 Comparisons of DET photocurrent responses of PSII, (TiO₂)₅-NTs, (TiO₂)₁₀-NTs, (TiO₂)₂₀-NTs, TiO₂(C), PSII/(TiO₂)₅-NTs, PSII/(TiO₂)₁₀-NTs, PSII/(TiO₂)₂₀-NTs and PSII/TiO₂(C)-modified ITO electrodes under red light and white light irradiations.

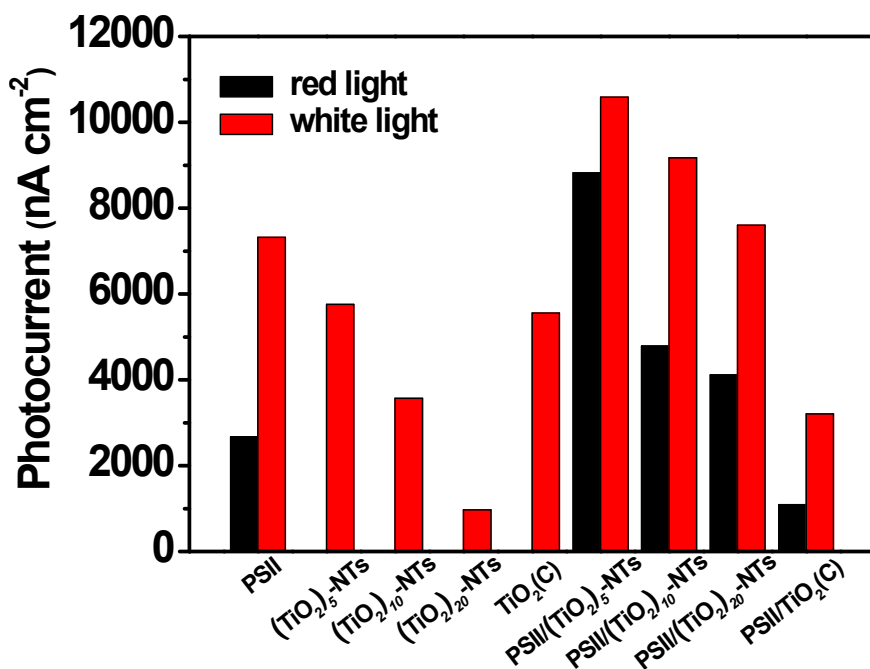


Fig. S10 Comparisons of MET photocurrent responses of PSII, (TiO₂)₅-NTs, (TiO₂)₁₀-NTs, (TiO₂)₂₀-NTs, TiO₂(C), PSII/(TiO₂)₅-NTs, PSII/(TiO₂)₁₀-NTs, PSII/(TiO₂)₂₀-NTs and PSII/TiO₂(C)-modified ITO electrodes under red light and white light irradiations.

Table S1 Summary of photoelectrochemical performance of PSII PF-PEC cells.

Reference	Electrode	origin	Light (nm)	Bias potential (V vs. SCE)	j ($\mu\text{A cm}^{-2}$)		Half-life time (min)	IPCE (%)	
					DET	MET			
23	PSII/Ni-NTA/Au	<i>T. elongatus</i>	675	0.26		14 ^a	18		
24	PSII/Ni-NTA/Au	<i>T. vulcanus</i>	680	0.16	0.13				
	PSII/Ni-NTA/Au (nanoparticles)	<i>T. vulcanus</i>	680	0.16	2.4				
26	PSII/pMBQ/Au	<i>M. laminosus</i>	> 400	0.15		2.7		1.0	
27	PSII/Os redox polymer/Au	<i>T. elongatus</i>	675	0.26		45	~ 55		
28	PSII/PBV ²⁺ /PSII/PBV ²⁺	<i>M. laminosus</i>	675	-0.04	0.5			~ 0.3	
31	PSII/mesoITO	<i>T. elongatus</i>	635	0.26	1.6		< 5		
								12 ^b	0.3 ^b
32	PSII/C ₂ CO ₂ ⁻ /mesoITO (electrostatic)	<i>T. elongates</i>	679	0.26	0.28		< 5	0.005	
								1.4 ^a	0.03 ^a
	PSII/C ₂ CO ₂ ⁻ /mesoITO (covalent)	<i>T. elongates</i>	679	0.26	0.43		~ 12	0.01	
This work	PSII/(TiO ₂) ₅ -NTs/ITO	<i>spinach</i>	550 ~ 800	0.25	0.035			~ 0.0006	
								8.8 ^a	~ 0.16 ^a
						White light		1.3	~ 0.02
						10.6 ^a	~ 0.19 ^a		

^a Mediated electron transfer using DCBQ. ^b Mediated electron transfer using potassium 1,4-naphthoquinone-2-sulfonate (NQS).