

Supporting Information for

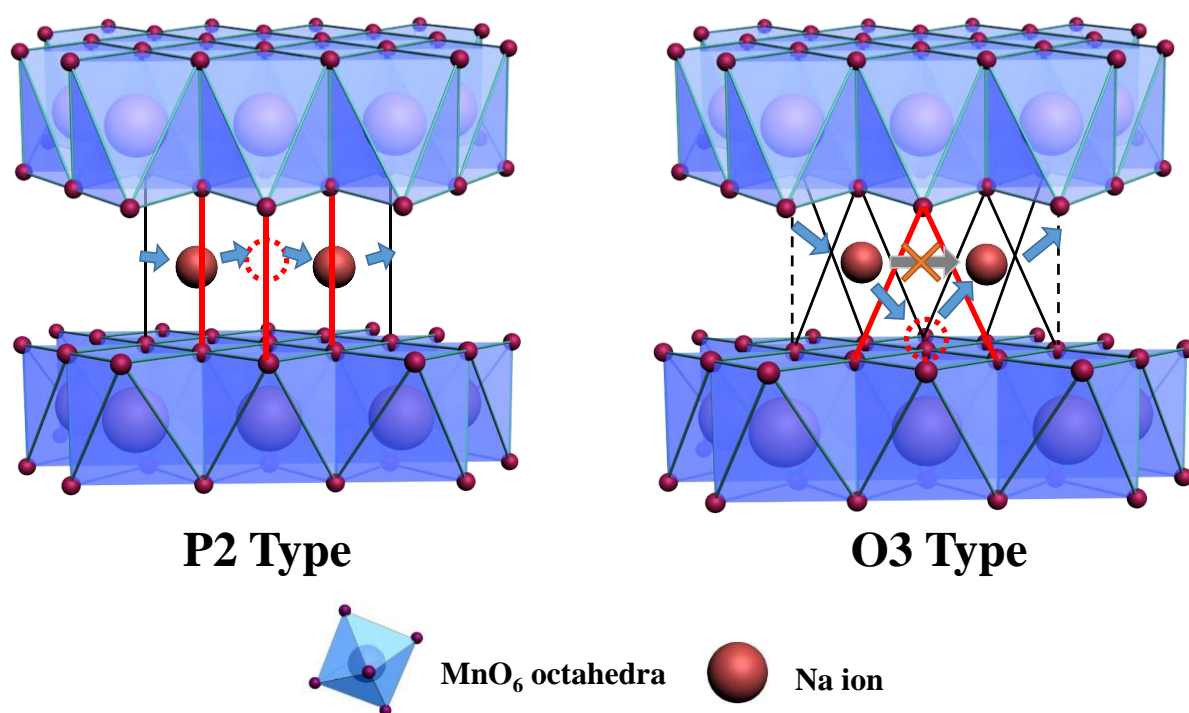
**CNTs Cocoon on Sodium Manganate Nanotubes Forming Core/branch cathode coupled with helical carbon nanofibres anode for enhanced sodium ion Batteries**

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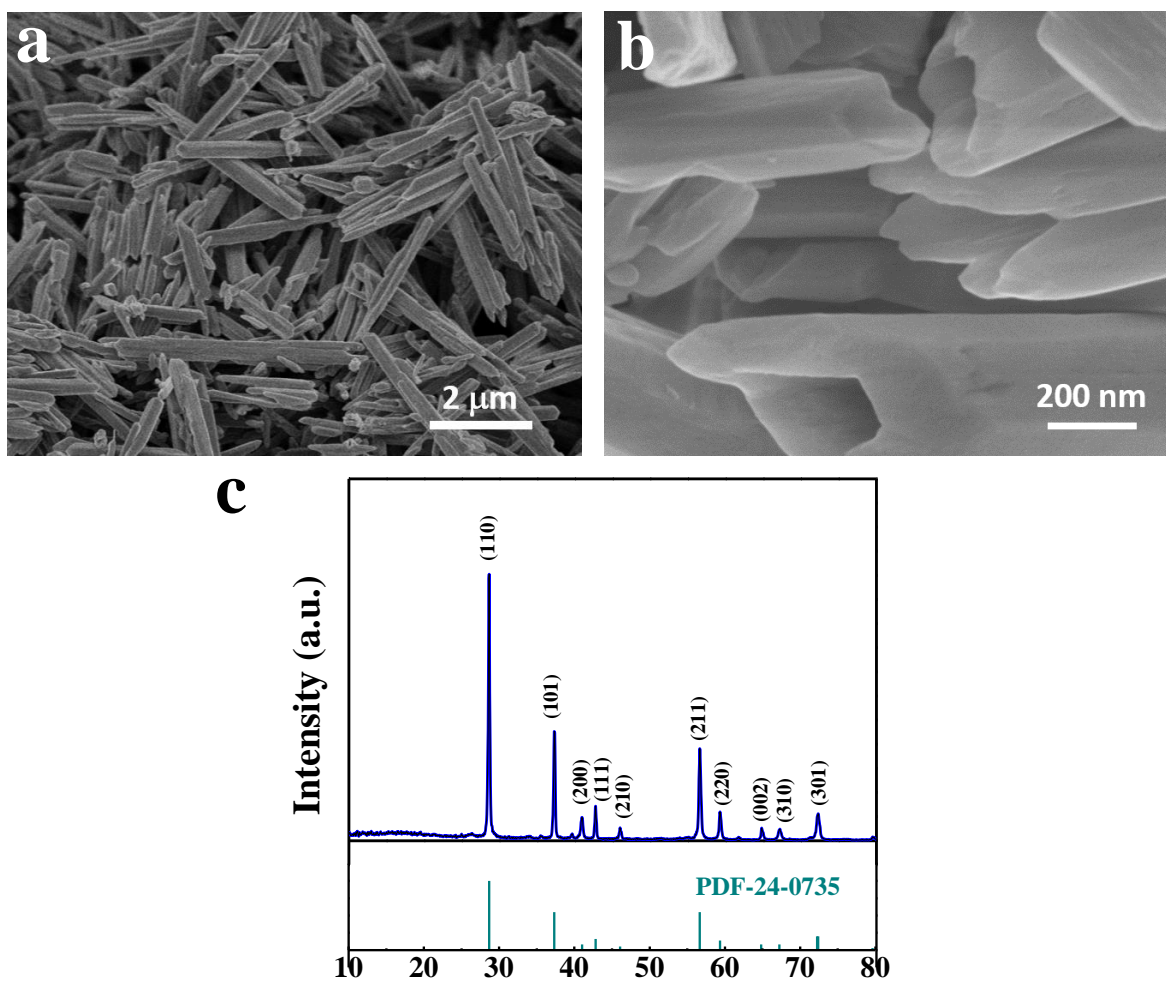
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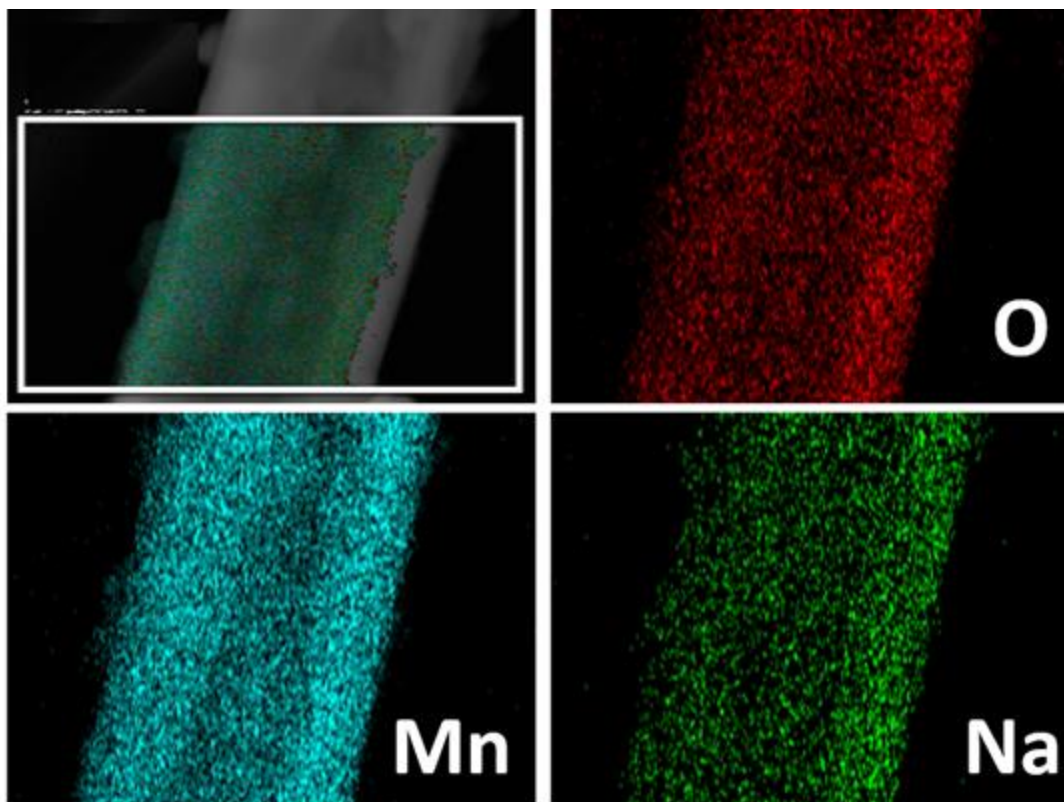
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**Figure S1.** Schematic of the P2-type and O3-type layered sodium manganese oxides and the mobility path of sodium ions in the interlayers



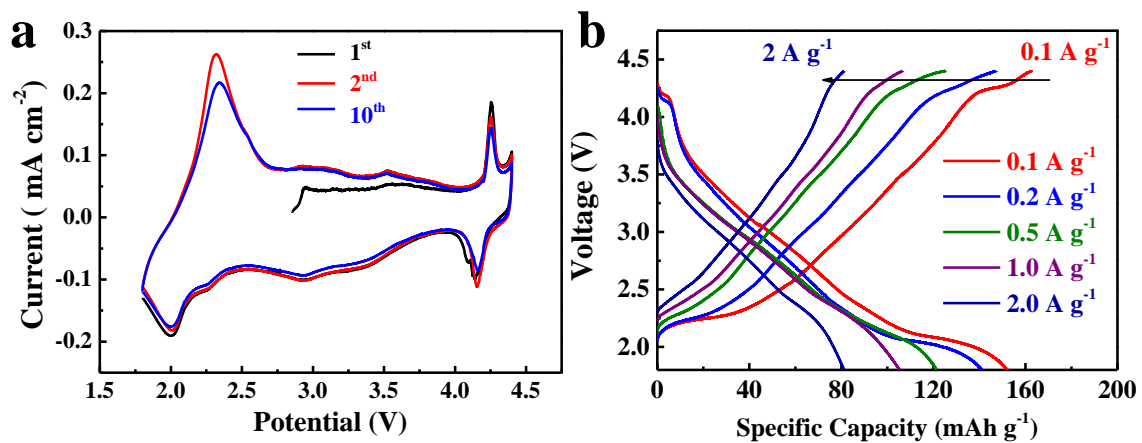
**Figure S2.** a,b) SEM images of as-prepared-MnO<sub>2</sub>; c) XRD pattern of as-prepared β-MnO<sub>2</sub>.



**Figure S3.** Element mapping images of O, Mn and Na.

**Table S1.** Na:Mn stoichiometric ratio of the as-prepared NMO analyzed by ICP-AES.

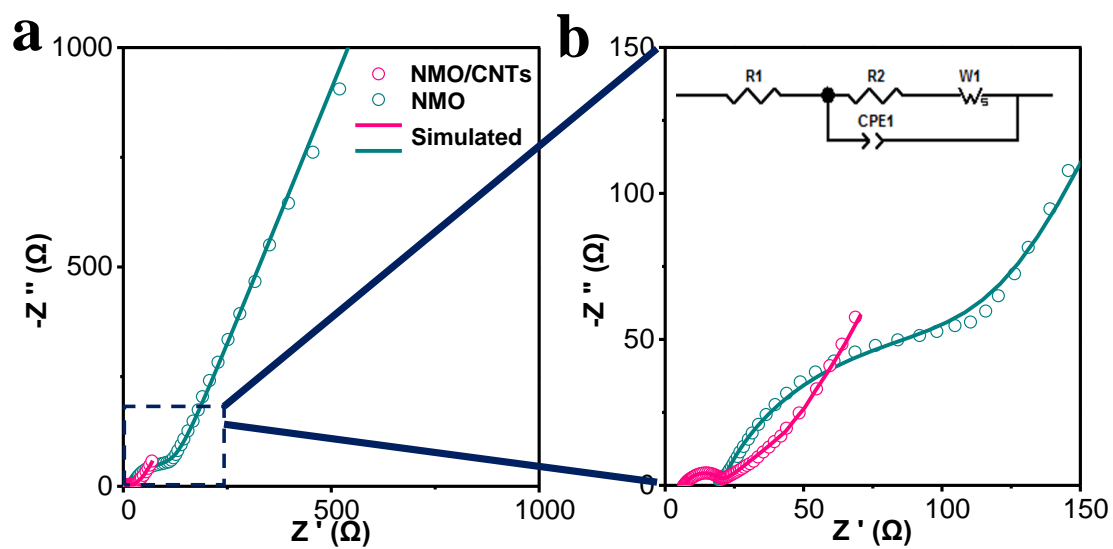
Element	Na	Mn
Molar ratio	0.7	1



**Figure S4.** a) CV curves of NMO at a scan rate of  $0.1 \text{ mVs}^{-1}$  of the 1<sup>st</sup>, 2<sup>nd</sup> and 10<sup>th</sup> cycle between 1.8 and 4.4 V. b) Discharge/charge voltage profiles at the various current densities from  $0.1 \text{ A g}^{-1}$  to  $2 \text{ A g}^{-1}$  at the 2<sup>th</sup> cycle.

**Table S2:** Literature overview: Sodium storage capacities for different sodium manganese oxide materials.

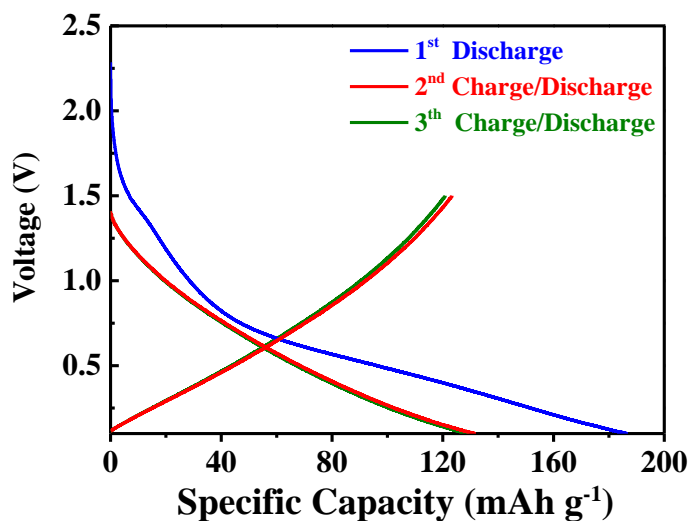
Type of NMO	Current Density/ A g <sup>-1</sup>	Capacity/ mAh g <sup>-1</sup>	Cycling performance	Potential Window /V	Reference
Na <sub>4</sub> Mn <sub>9</sub> O <sub>18</sub>	0.12	128	77% after 1000 cycles	1.5 - 4.1	[1]
	1.2	94			
	2.4	82			
Na <sub>0.7</sub> MnO <sub>2</sub>	0.04	163	93.7% after 10 cycles	2 - 4.5	[2]
	0.08	150			
	0.18	125			
Na <sub>0.67</sub> MnO <sub>2</sub>	0.01	174	83% after 60 cycles	2 – 4.3	[3]
	0.05	120			
	0.25	88			
Na <sub>0.71</sub> MnO <sub>2</sub> ·0.25H <sub>2</sub> O	0.01	149.5	87% after 100 cycles	1.8 – 3.7	[4]
	0.02	135.3			
	0.4	89.7			
Na <sub>0.6</sub> MnO <sub>2</sub>	Not reported	156	36% after 10 cycles	2.0 – 3.8	[5]
Na <sub>0.7</sub> MnO <sub>2.05</sub>	0.1	162.3	88% after 100 cycles at 0.1 A g <sup>-1</sup>	1.8 – 4.4	This work
	0.5	133.9			
	2.0	101.9			



**Figure S5.** a) Nyquist plots of NMO and NMO/CNTs at potential of 4.3 V after 5th cycle. b) Enlarged Nyquist plots of the marked area in (a).

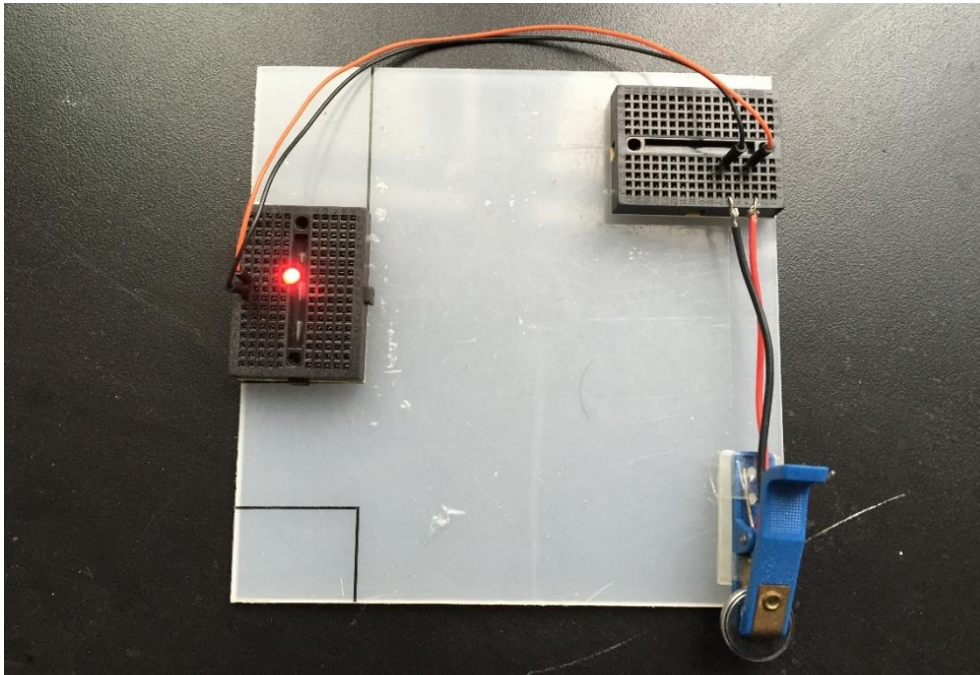
**Table S3:** Literature overview: sodium storage capacities for different carbon-based materials.

Type of carbon	Temp./ °C	Capacity/ mAh g <sup>-1</sup>	Rate/ Current density	Potential Window /V	Reference
Templated carbon	RT	130	C/5	0.01-1.6	[6]
Carbon black	RT	121	C/75	Not reported	[7]
Biomass-derived carbon	RT	122	0.1 A g <sup>-1</sup>	0.001-2.8	[8]
Carbon black	RT	120	0.1 A g <sup>-1</sup>	0.15-2.0	[9]
Sphere carbon	RT	105	0.075 A g <sup>-1</sup>	0.1-2.0	[10]
<b>HCNFs</b>	<b>RT</b>	<b>136.1</b>	<b>0.1 A g<sup>-1</sup></b>	<b>0.01-1.5</b>	<b>This work</b>



**Figure S6.** Discharge/charge voltage profiles of HCNFs electrode at 0.1 A g<sup>-1</sup> at the 1<sup>st</sup> cycle, 2<sup>nd</sup> cycle and 3<sup>rd</sup> cycle.





**Figure S7.** Optical image showing a LED (2.0 V) powered by NMO/CNTs//HCNFs SIB full cell.

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