Supplementary Information

Side-chain-type anion exchange membranes bearing pendant quaternary ammonium groups via flexible spacer for fuel cells

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Chemical structure	Scale	Bubbling time	Reaction time	Yield
	(mmol)	(h)	(h)	(%)
Br-(CH ₂) ₃ -Br	100	4	24	68
Br-(CH ₂) ₄ -Br	100	4	24	60
Br-(CH ₂) ₆ -Br	100	4	24	56
Br-(CH ₂) ₈ -Br	50	2	24	65
Br-(CH ₂) ₁₂ -Br	50	2	24	61

Table S1 Quaternization reaction of 1, ω -dibromoalkanes in THF.



Fig. S1 ¹H NMR spectra of Br-x-QA (x=3, 4, 6, 8 and 12).



Fig. S2 ¹H NMR spectra of PES-OCH₃ and PES-OH.



Fig. S3 1 H NMR spectra of PES-n-QA (n=3, 4, 6, 8 and 12).



Fig. S4 The appearance (a, b), SEM images of (c) cross-section and (d) surface of PES-6-QA.



Fig. S5 Small angle X-ray scattering (SAXS) of the AEMs.



Fig. S6 The number of absorbed water molecules around each QA group (λ) as a function of flexible spacer length for the PES-n-QA membranes at 30, 60 and 80 °C.



Fig. S7 Thermogram of water fusion enthalpy of the AEMs.



Fig. S8 Water state and hydration number of the AEMs.

	Tensile strength		Young's modulus		Elongation at break	
Membranes	(MPa)		(MPa)		(%)	
	Before	After	Before	After	Before	After
PES-3-QA	13.6	8.5	285	196	28.0	14.2
PES-4-QA	11.5	9.6	219	189	42.1	38.5
PES-6-QA	9.2	8.5	158	131	57.6	55.4
PES-8-QA	11.2	10.5	199	166	38.6	35.3
PES-12-QA	13.1	11.9	270	249	23.8	22.4



Table S2 The mechanical properties of the membranes before and after the alkaline stability test.



Fig. S9 ¹H NMR spectra of (a) PES-4-QA, (b) PES-6-QA and (c) PES-8-QA stored in a 1 M aqueous KOH solution at 60 °C for 0 and 720 h, respectively.



Scheme S1 The proposed degradation pathways of QA groups in PES-3-QA in alkaline media.¹

References

1 K. M. Meek, Y. A. Elabd. *Macromolecules*, 2015, 48, 7071-7084.