

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A.
This journal is © The Royal Society of Chemistry 2016

Electronic Supplementary Information

Solution-processable hypercrosslinked polymers by low cost strategies: a promising platform for gas storage and separation

Yuwan Yang,^a Bien Tan^{*a} and Colin D Wood^{*b}

*^a School of Chemistry and Chemical Engineering, Huazhong University of
Science and Technology, Wuhan, 430074, China.*

Email: bien.tan@mail.hust.edu.cn

*^b CSIRO, Energy, Australian Resources Research Centre, Kensington, 6151
Western Australia, Australia.*

Email: colin.wood@csiro.au

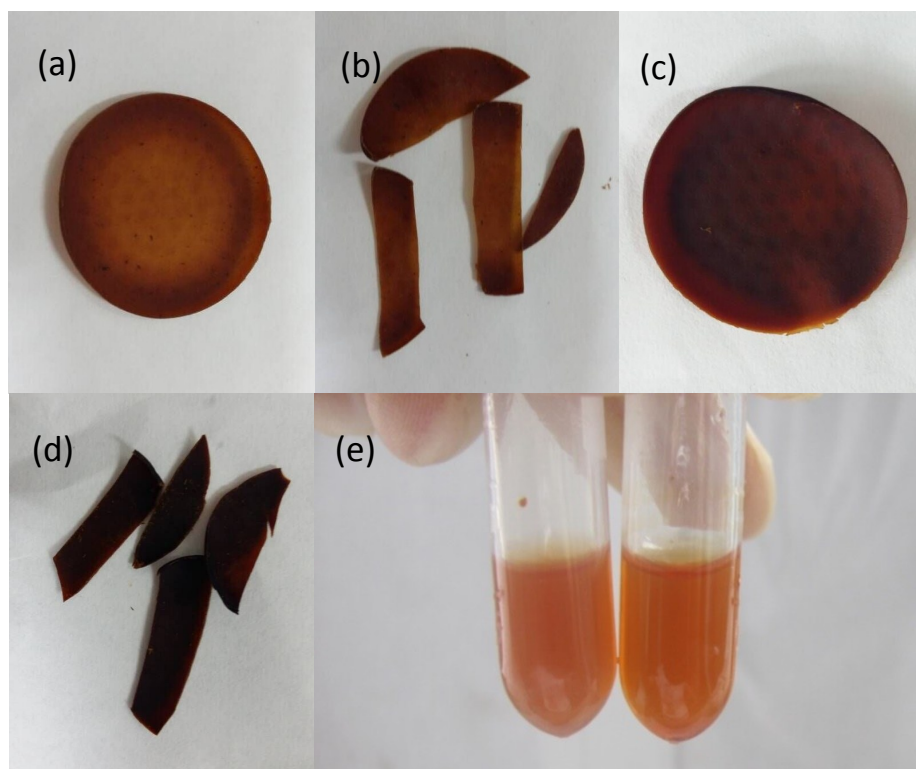


Figure S1 Photographs of insoluble HCP-1 (a) and HCP-2 (c) after filtrating, and small pieces of HCP-1 (b) and HCP-2 (d). Both materials swell in THF (e).

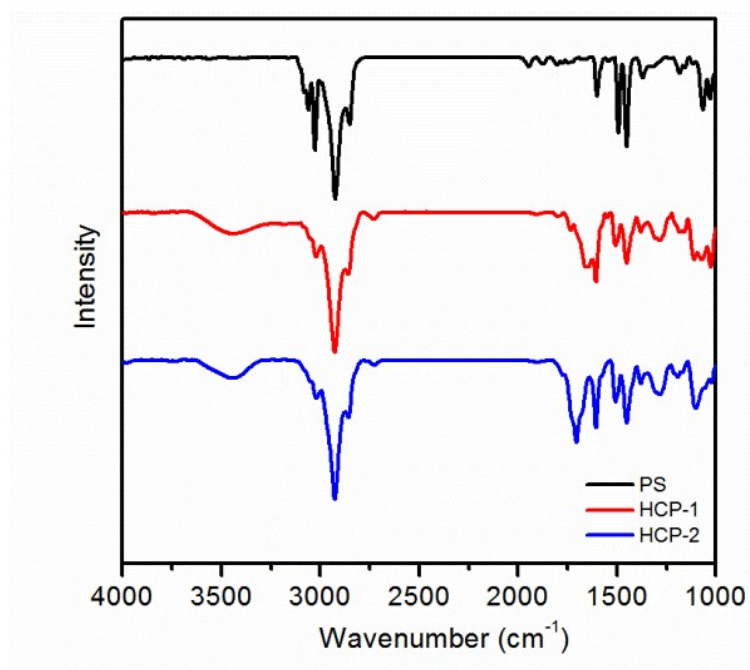


Figure S2 FTIR spectra of PS, HCP-1 and HCP-2.

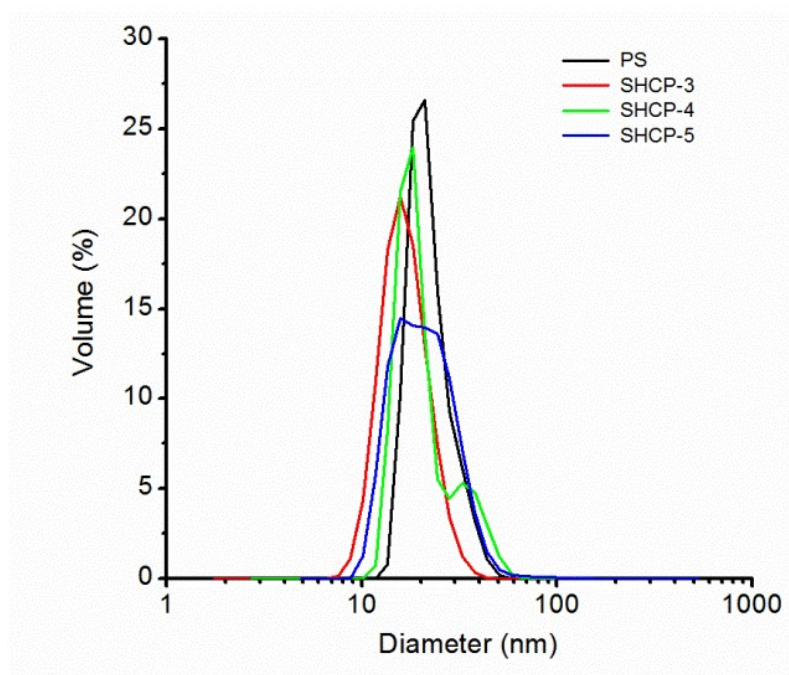


Figure S3 DLS curves of PS and SHCPs. Solvent: tetrahydrofuran.

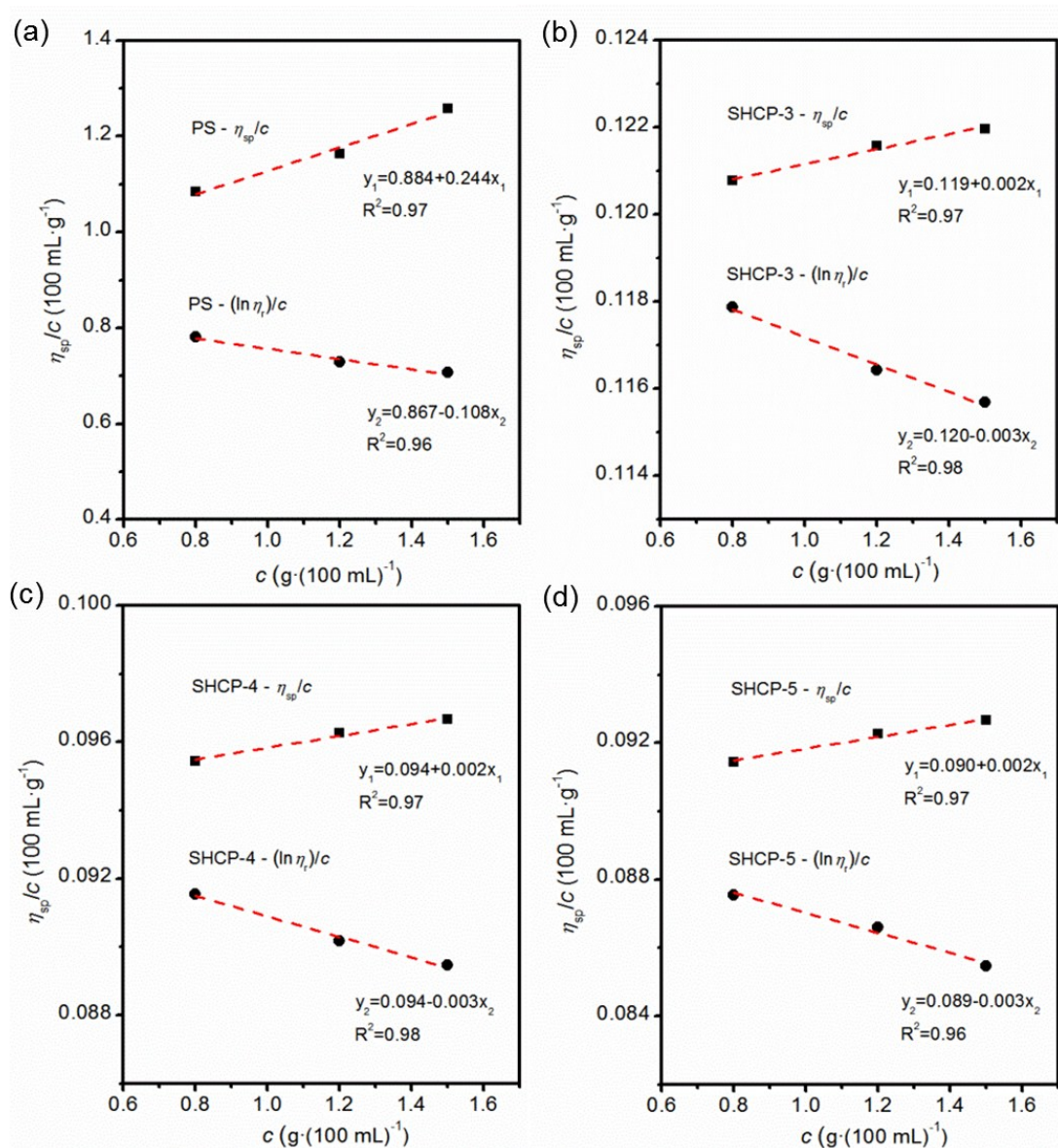


Figure S4 The intrinsic viscosity of PS and SHCPs. Solvent: tetrahydrofuran.

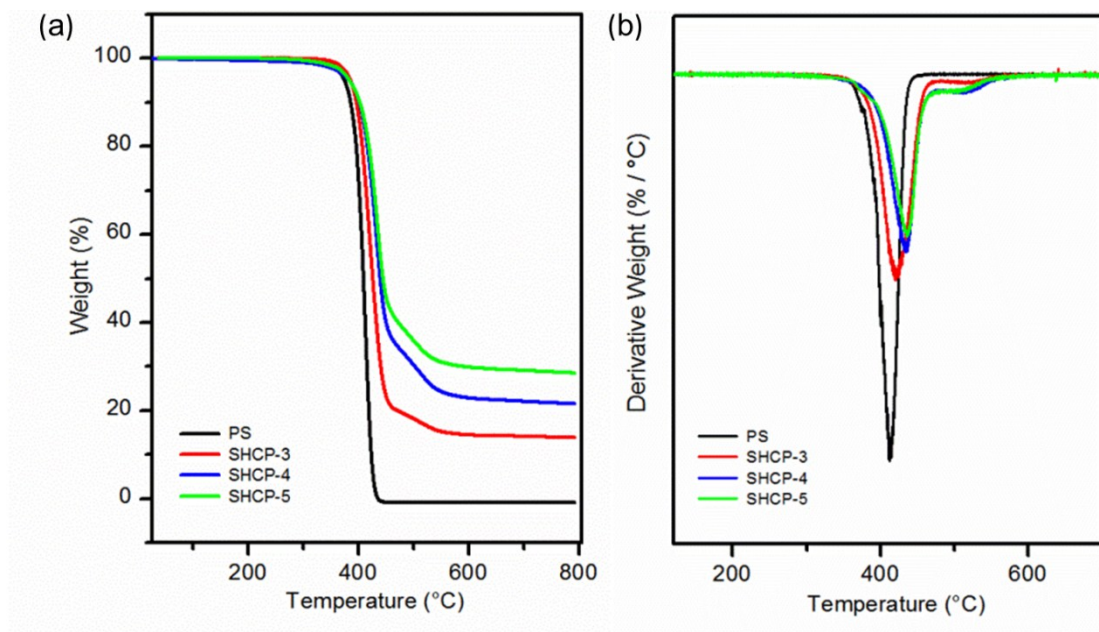


Figure S5 TG (a) and DTA (b) curves of PS and SHCPs at a heating rate of 10 °C/min under nitrogen atmosphere.

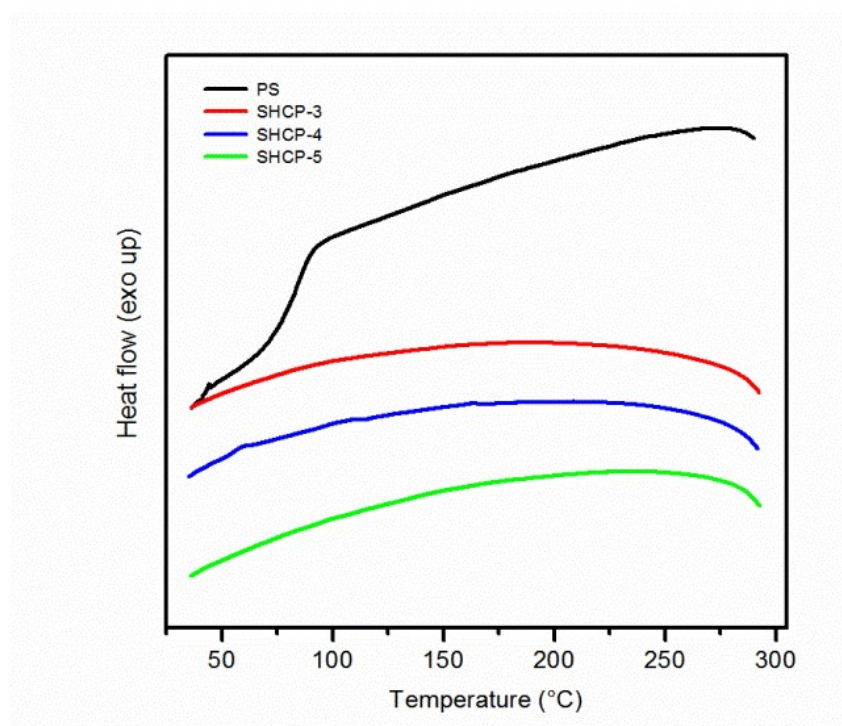


Figure S6 DSC curves of PS and SHCPs at a heating rate of 10 °C/min under nitrogen atmosphere.

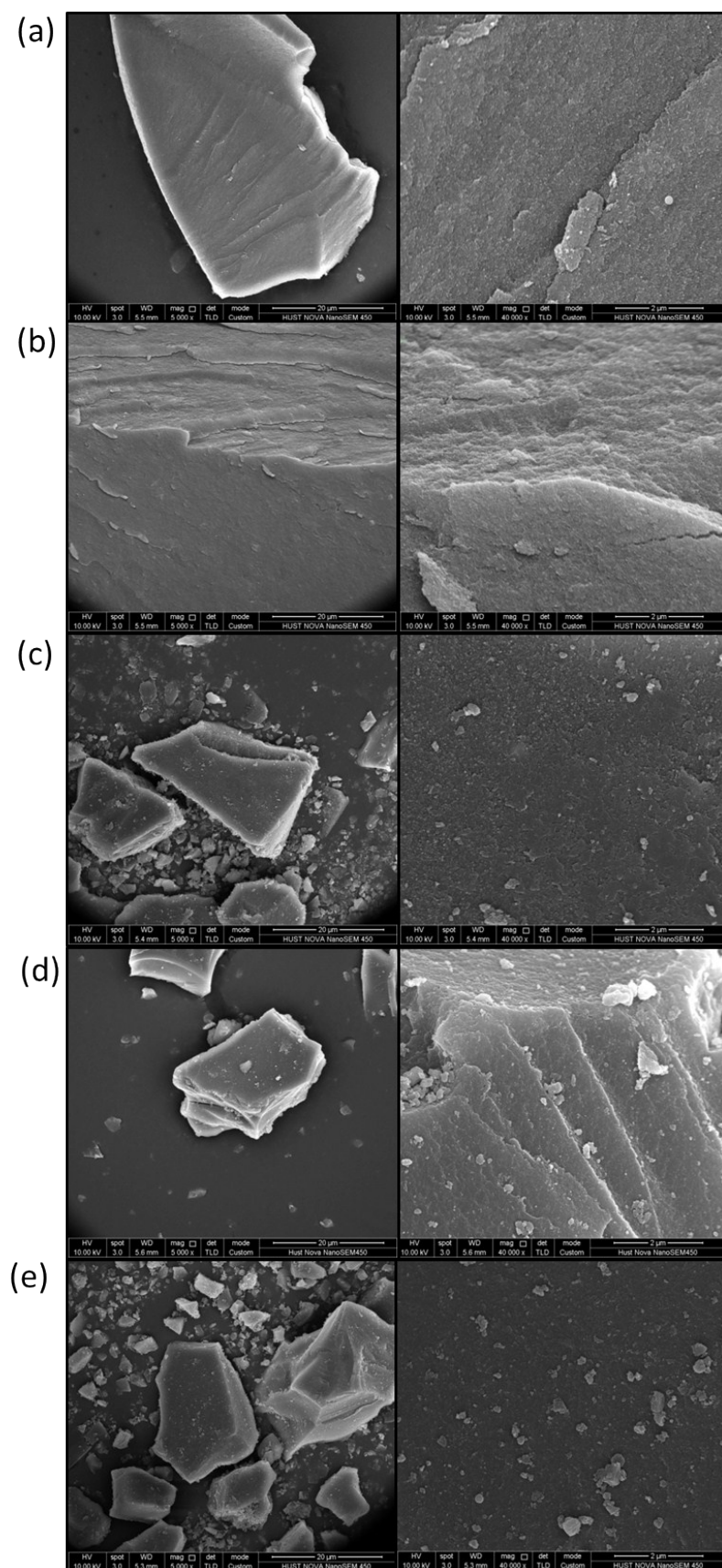


Figure S7 SEM image of the HCP-1 (a), HCP-2 (b), SHCP-3 (c), SHCP-4 (d) and SHCP-5 (e), scale bar: 20 μm (left), 2 μm (right).

Table S1 Reaction conditions for HCPs and SHCPs. The total volume of solvent was 450 mL in each reaction, for samples 3, 4 and 5 the cross-linker (FDA) was slowly added in a 50 mL solution.

Sample	PS (g)	FDA (μ L)	FeCl ₃ (g)	S _{BET} ^{a)} (m ² g ⁻¹)
HCP-1	0.10	270	1.22	981
HCP-2	0.22	580	1.08	1125
SHCP-3	0.05	270	0.60	187
SHCP-4	0.52	270	0.60	480
SHCP-5	0.52	500	0.60	724

^{a)}surface area for the complete sample, the surface area for the soluble component is shown in table S2.

Table S2 The weight of sedimentations of SHCPs.

Sample	M ₁ ^{a)} (g)	M ₂ ^{b)} (g)	f ^{c)} (%)	S _{BET} ^{d)} (m ² g ⁻¹)
SHCP-3	0	0.0500	100	158
SHCP-4	0.0110	0.0390	78	355
SHCP-5	0.0353	0.0147	29	530

^{a)}The weight of sedimentations; ^{b)}The weight which dissolved in chloroform; ^{c)}M₂/(M₁+M₂)*100%, weight fractions of soluble part; ^{d)}Apparent surface area of soluble fractions calculated from nitrogen adsorption isotherms at 77.3 K using BET equation.

Table S3 Molecular weight and polydispersity index of SHCPs.

Sample	S _{BET} (m ² g ⁻¹)	M _w	M _n	PDI
PS	0	167700	65600	2.56
SHCP-3	187	145000	65600	2.21
SHCP-4	480	177500	98000	1.81
SHCP-5	724	173300	106500	1.63