

Enhanced Preferential CO Oxidation on Zn₂SnO₄ Supported Au Nanoparticles: Support and H₂ Effects

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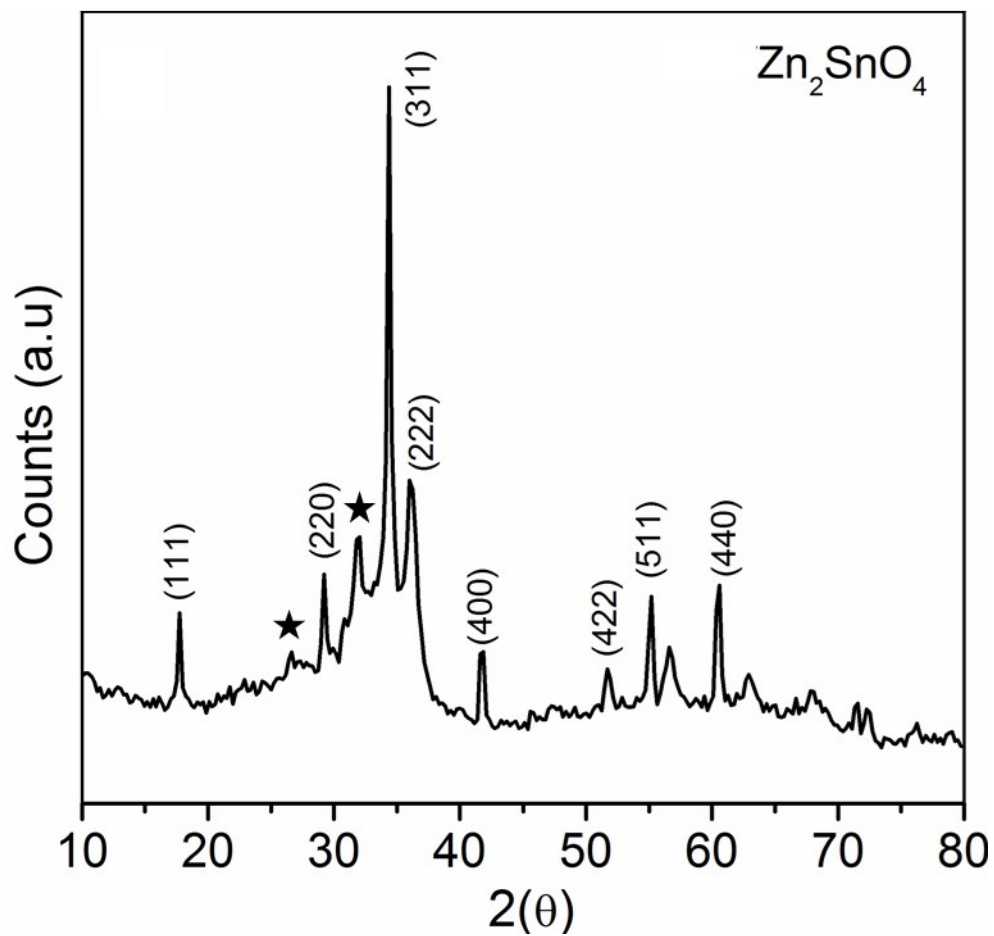


Fig. S1 XRD pattern of Zn₂SnO₄ inverse spinel structure, marked peaks corresponds to impurity phase

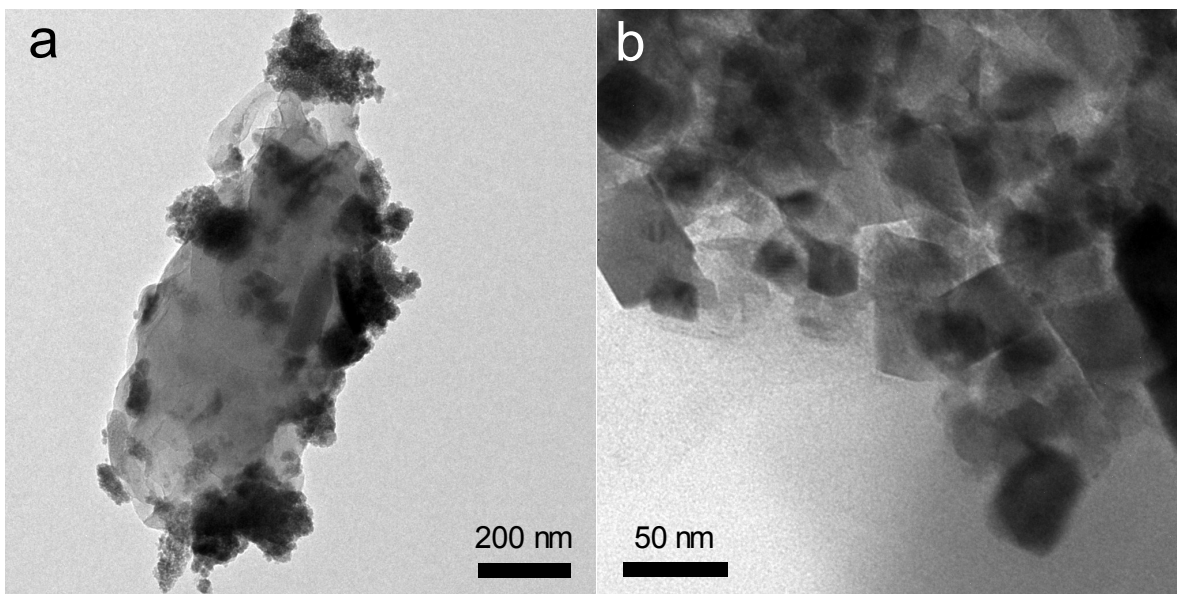


Fig. S2 Bright field TEM micrographs of combustion synthesized Zn_2SnO_4 at different magnifications.

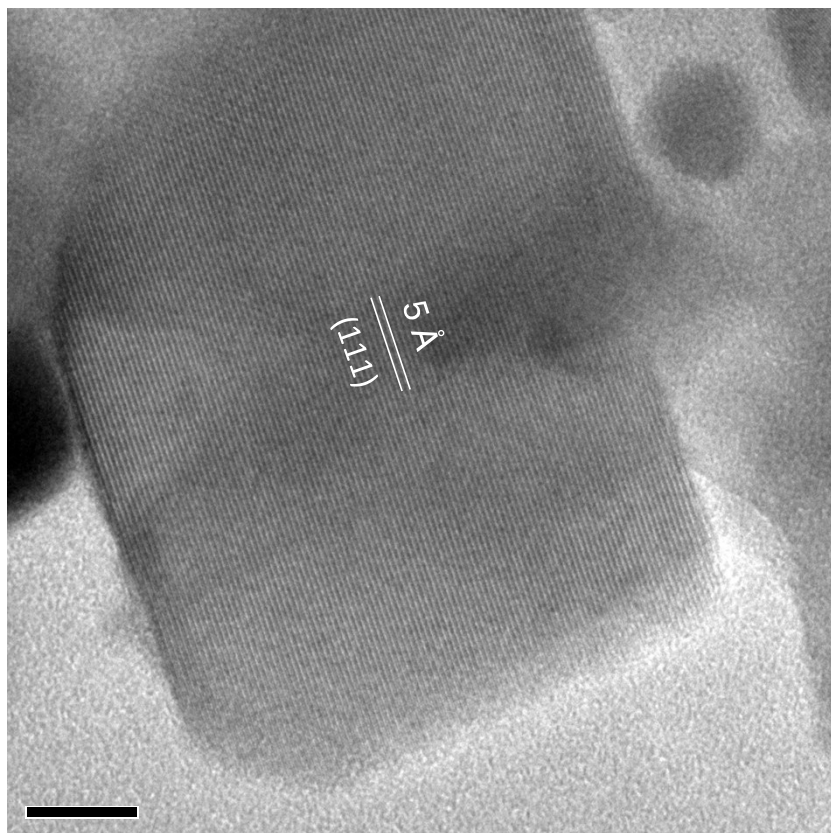


Fig. S3 High resolution TEM image of Zn₂SnO₄, illustrated d-spacing corresponds to crystallographic (111) plane of Zn₂SnO₄.

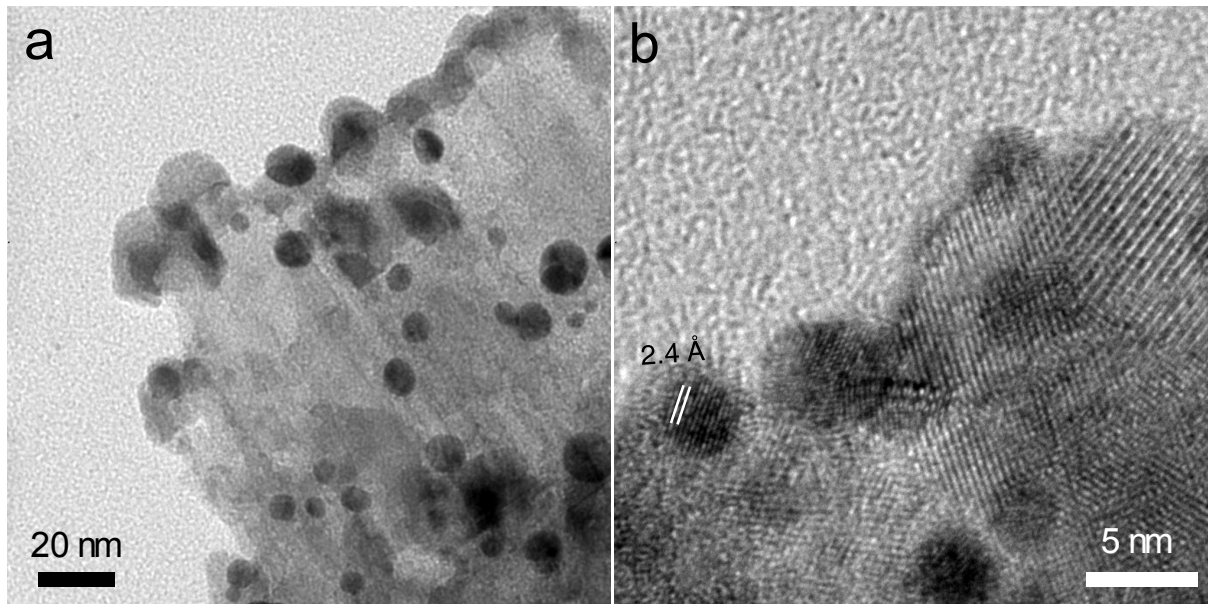


Fig. S4 (a) Bright field and (b) corresponding high-resolution TEM micrographs of Au/Zn₂SnO₄ the marked lattice spacing corresponds to Au (111) plane.

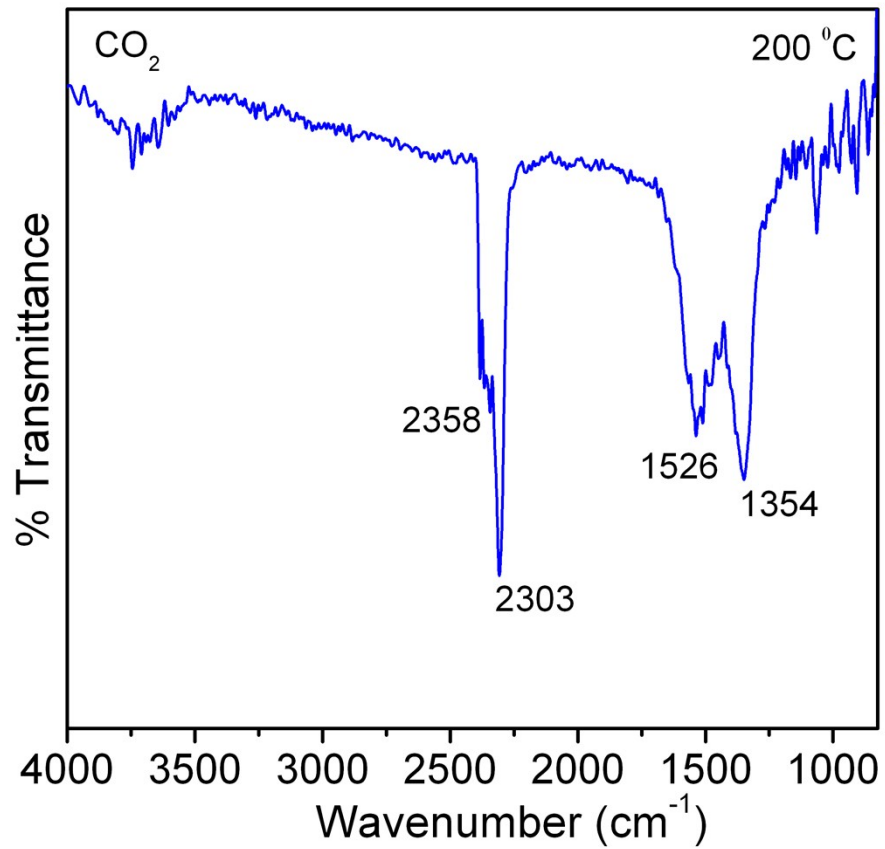


Fig. S5 DRIFT spectra recorded during flow of CO₂ on Zn₂SnO₄ at 200 °C. Carbonates signals along with gases CO₂ peaks are observed.

Au/Zn₂SnO₄	Relative Intensity Ratio of Au		
	Au⁰	Au¹⁺	Au³⁺
As-synthesized	1	0.34	0.29
After COX	1	0.30	0.15
After PROX	1	0.27	0.09
After CO TPR	1	0	0

Table S1 XPS relative intensity ratios of Au oxidation state (AU 4f) in Au/Zn₂SnO₄ at different experimental conditions.

Au/Zn₂SnO₄	Zn-carbonates (Binding Energy in eV)	Zn-V_o (Binding Energy in eV)	Sn-V_o and carbonates (Binding Energy in eV)
As-synthesized	93.7	91.2	89.8
	92.6	88.7	87.0
After COX	94.1	91.2	90.0
	92.7	88.7	87.3
After PROX	94.5	91.3	88.9
	92.7	88.7	87.4
After CO TPR	94.2	91.2	89.3
	92.8	89.0	87.2

Table S2 XPS peak position of Zn 3p and Sn 4p in Zn/Au₂SnO₄ catalyst treated at different experimental conditions.

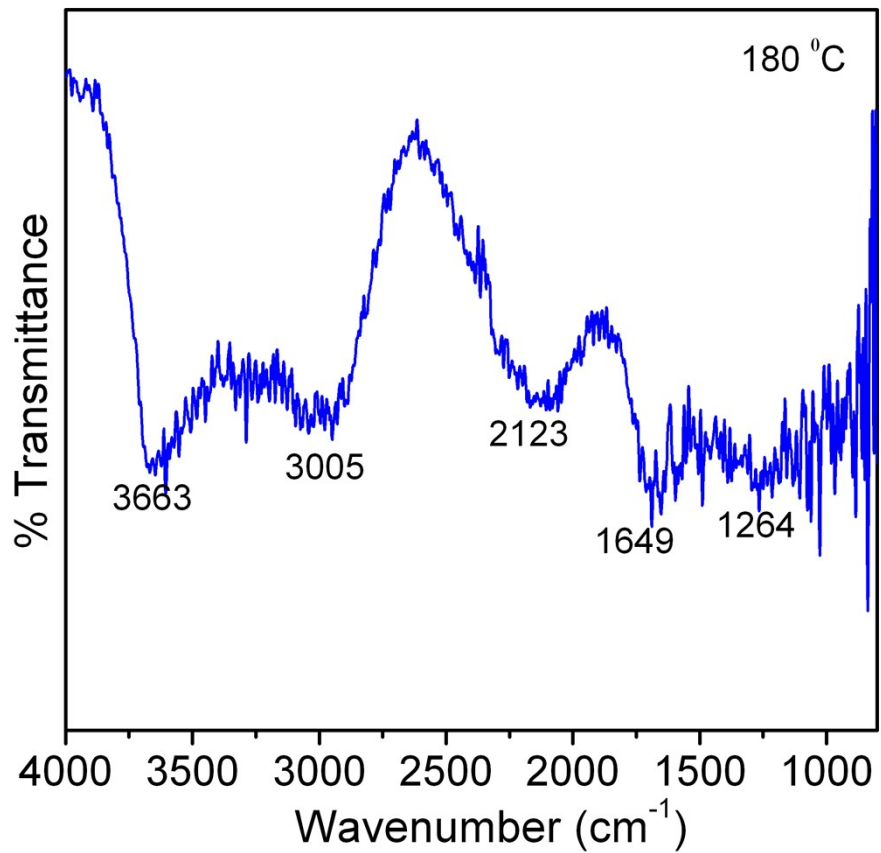


Fig. S6 DRIFT spectra recorded during PROX at 180 °C for Au/Zn₂SnO₄. Prominent, HOH vibrations are observed.