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Supplementary information

Effect of coating on the electrochemical performance of LiCoPO₄ for green electrode processing

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Table S 1.	Output of Rietveld	Refinement of	uncoated and	coated LCP	usina Topas 4.2.	

	LCP	LCP-C1	LCP-C2	LCP-C3	LCP-C4	LCP-V		
R _{exp}	4.80	5.36	3.89	4.25	4.46	4.96		
R _{wp}	5.30	5.65	4.35	4.60	4.59	5.34		
R _p	4.20	4.48	3.44	3.66	3.64	4.26		
GOF	1.10	1.05	1.12	1.08	1.03	1.08		
DW	1.70	1.82	1.67	1.75	1.89	1.72		
Refinement range (20): 10-80°; LP factor: 0; Order of background parameter (Chebichev polynomial): 15								
Specimen	-0.243(2)	-0.154(2)	-0.232(2)	-0.297(3)	-0.339(3)	-0.127(1)		
displacement (mm)								
Phase 1	LiCoPO₄							
B _{eg} ** (Å ²)	4.4(1)	5.6(1)	6.3(1)	5.8(1)	3.8(1)	5.9(1)		
R-Bragg	1.861	2.164	2.544	1.674	1.206	2.914		
Space group	Pnma (62)							
Lattice parameter								
a (Å)	10.2069(3)	10.2064(4)	10.2052(3)	10.2005(5)	10.2061(6)	10.2054(2)		
b (Å)	5.9231(2)	5.9229(2)	5.9242(2)	5.9328(3)	5.9313(3)	5.9221(1)		
c (Å)	4.7011(2)	4.7006(2)	4.7011(2)	4.7043(3)	4.7049(3)	4.7004(1)		
Crystallite size								
Lorentzian (nm)	190(7)	127(5)	120(3)	96(3)	131(10)	210(6)		
Gaussian (nm)	145(5)	109(5)	136(6)	108(6)	59(2)	279(17)		
Phase 2	Co ₂ P							
B _{eq} **(Å ²)	=B _{eq} (Phase 1)							
R-Bragg	0.310	0.607	0.514	1.139	0.746	0.296		
Space group	Pnma (62)							
Lattice parameter	Lattice parameter							
a (Å)	5.63(7)	5.64(5)	5.652(6)	5.659(6)	5.666(5)	5.65(9)		
b (Å)	3.52(4)	3.52(3)	3.511(4)	3.511(4)	3.510(3)	3.50(6)		
c (Å)	6.66(8)	6.61(6)	6.610(7)	6.609(7)	6.611(6)	6.6(1)		
Crystallite size	Crystallite size							
Lorentzian (nm)*	50							
Gaussian (nm)*	50							
Phase 1 (%)	99.7(2)	99.6(2)	97.4(2)	96.6(2)	95.9(2)	99.8(2)		
Phase 2 (%)	0.3(2)	0.4(2)	2.6(2)	3.4(2)	4.1(2)	0.2(2)		
*fixed parameter								
**B _{eq} was simultaneously refined for all atoms set to the same value of B _{eq} The initial structural models used for the Rietveld refinement were taken from the data of F. Kubel, Zeitschrift für Kristallographie - Crystalline Materials 209 (1994), 755 (ICSD 400625) for phase 1 (LiCoPO ₄) and S. Rundqvist, Acta Chem. Scan, 14 (1960) 1961-1979 (ICSD 43685) for phase 2 (Co ₂ P).								

(Co₂P).

	LCP	LCP/C1	LCP/C2	LCP/V		
112 th cycle						
χ²	0.0064398	0.009176	0.007447	0.000409		
Rs	9.80	7.99	7.04	2.286		
R _s (error)	0.329	0.465		0.066		
R ₁	26.93	10.54	10.56	23.31		
R₁ (error)	0.558	0.583	0.165	5.574		
CPE₁-T	5.5E-06	5.0E-06	4.0E-06	3.9E-05		
CPE₁-T	0.021	0.046	0.021	0.028		
(error)						
CPE ₁ -P	0.74	0.75	0.81	0.66		
CPE₁-P	2.8	6.1	2.5	4.2		
(error)						
R ₂				241.4		
R ₂ (error)				4.9		
CPE ₂ -T				0.00011		
CPE ₂ -T				0.000008		
(error)						
CPE ₂ -P				0.59		
CPE ₂ -P				0.019		
(error)						
CPE _w -T	0.005	0.007	0.009	0.0016		
CPE _w -T	0.000048	0.000067	0.000087	0.000017		
(error)						
CPE _w -P	0.59	0.53	0.57	0.45		
CPE _w -P	0.004	0.004	0 004	0.004		
(error)						

Table S2. Results of impedance spectroscopy fitting for uncoated and coated LCP electrodes at 4.8 V of discharge during the 112th cycle using ZView 2 applying an equivalent circuit model.



Figure S1. TGA of Tween 80® in N₂ atmosphere.



Figure S2. EDX line scan of LCP with different coatings: A LCP/C1 (sucrose), B LCP/C2 (sucrose), C LCP/C3 (PAA), D LCP/C4 (Tween 80), E LCP/V (V_2O_5).



Figure S3. SEM images of electrodes with pristine and coated LCP A: 20 wt.% sucrose, B: 10 wt.% sucrose, C: 20 wt.% polyacrylic acid, D: 40 wt.% Tween 80®, E: 3 wt.% V_2O_5 , F: pristine LCP.

The morphology of pristine and coated LCP electrodes made using the aqueous binder (CMC), subsequently pressed to reduce the porosity was also investigated by SEM (Figure S3). Pristine LCP, LCP/C1 and LCP/C4 (A, D, F) showed very smooth and homogeneous electrode surfaces. On the other hand, some agglomerates and cracks were detected for LCP/C2 (B) while LCP/C3 (C) showed a very inhomogeneous distribution of active material and conductive carbon along with a large amount of agglomerates. LCP/V (E) showed some cracks, but an overall smooth electrode surface and no major inhomogeneities due to active material particles' agglomeration.[14]



Figure S4. a) Voltage profile and b) differential capacity plot of uncoated and coated LCP (selected materials) during the 5^{th} cycle (C/10).