

Electronic Supplementary Information

General strategy for coating metal-organic frameworks on diverse components and architectures

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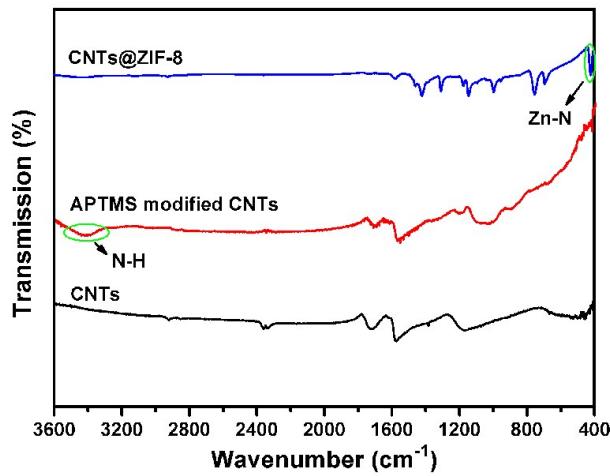


Fig. S1 FTIR spectra of the untreated CNTs, APTMS modified CNTs and core@shell CNTs@ZIF-8 (fractured) nanowires.

As can be seen in the FTIR spectra (Fig. S1, ESI), a new broad band in the range of 3200–3600 cm⁻¹ emerges after APTMS modification, which is ascribed to the N–H stretching and confirms the amination of core components with APTMS treatment.

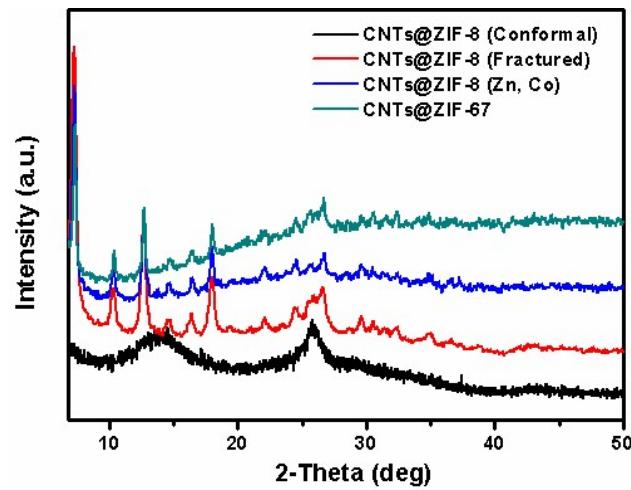


Fig. S2 XRD patterns of the as-synthesized core@shell products.

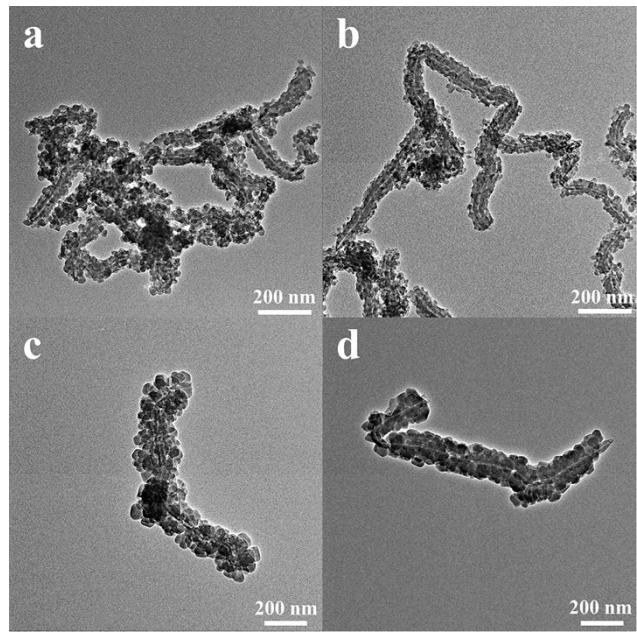


Fig. S3 TEM images of (a and b) CNTs@ZIF-8-0.5 (fractured) and (c and d) CNTs@ZIF-8-2 (fractured) nanowires.

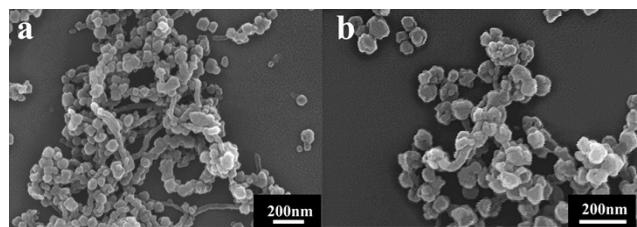


Fig. S4 SEM images of CNTs@ZIF-8 prepared without PSS treatment.

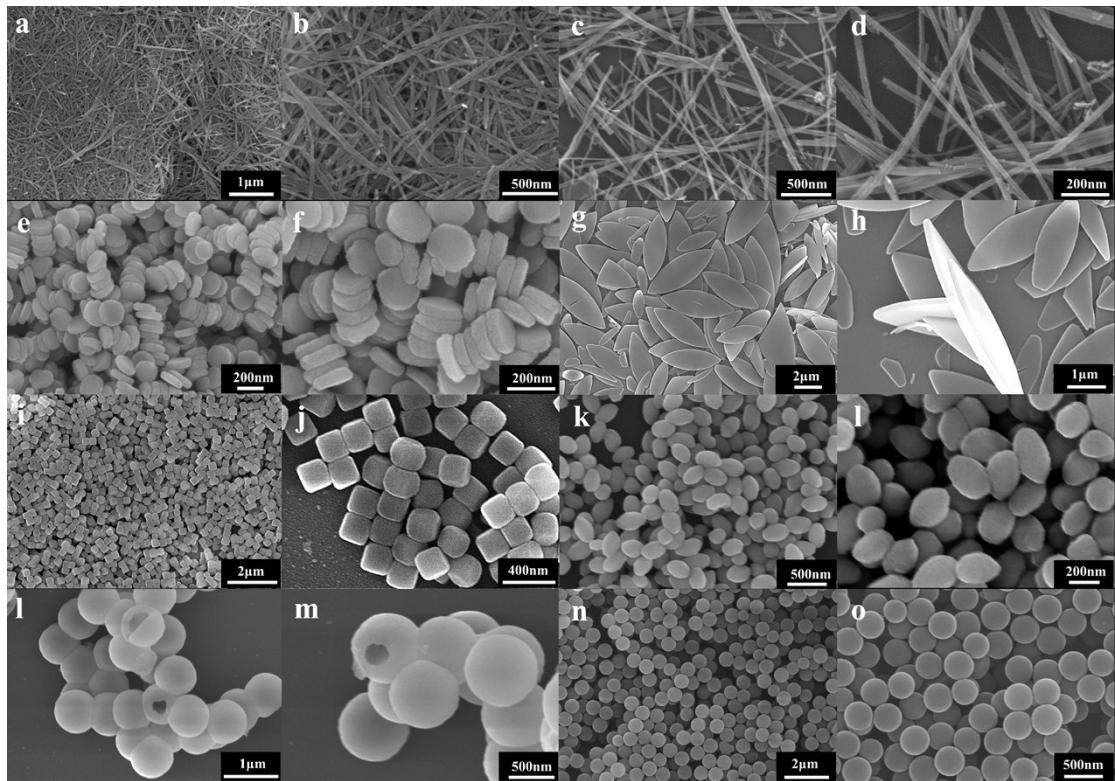


Fig. S5 SEM images of the diverse core components. (a and b) Ni(OH)₂ nanowires; (c and d) LiMn₂O₄ nanowires; (e and f) Fe₂O₃ nanoplates; (g and h) ZIF-L leaf-like microplates; (I and j) Ni₃[Co(CN)₆]₂ nanocubes; (k and l) Fe₂O₃ nanospindles; (l and m) SiO₂ hollow nanospheres and (n and o) RF nanospheres.

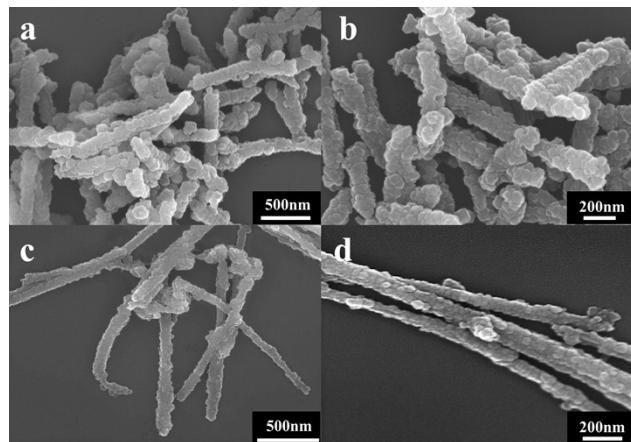


Fig. S6 SEM images of (a and b) Ni(OH)₂@ZIF-8 and (c and d) LiMn₂O₄@ZIF-8 nanowires.

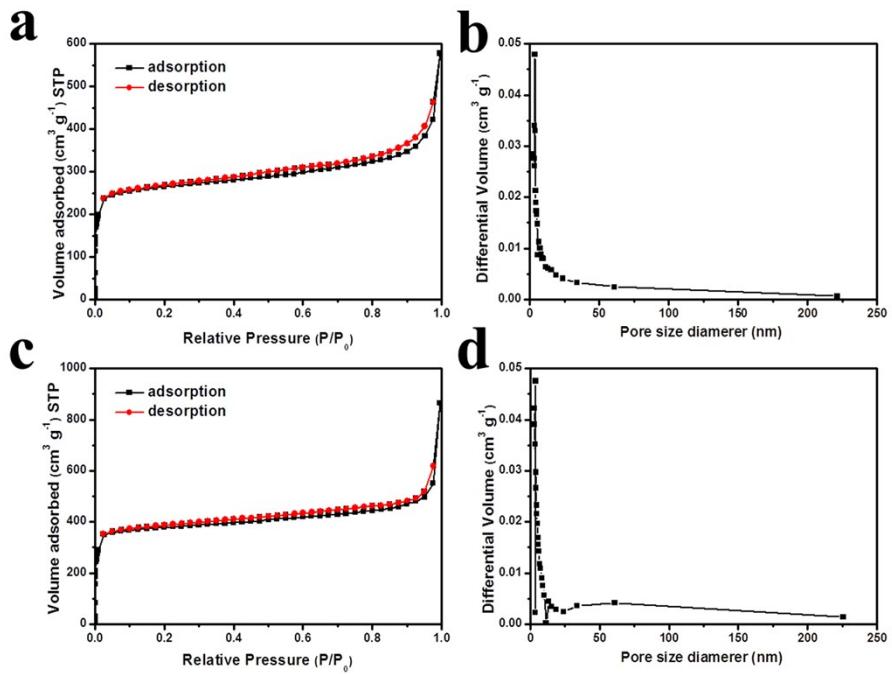


Fig. S7 N₂ adsorption-desorption isotherms and the corresponding pore size distributions of (a and b) CNTs@ZIF-8 (fractured) nanowires and (c and d) Ni(OH)₂@ZIF-8 nanowires.

N₂ sorption isotherms of the CNTs@ZIF-8 (fractured) and Ni(OH)₂@ZIF-8 nanowires are shown in Fig. S7. The observed features of the isotherms resemble type I and IV isotherms with a H3 type hysteresis loop. The sharp uptakes at low pressure imply the presence of micropores from ZIF-8. The distinct hysteresis in the relative pressure region of P/P₀>0.3 underscores the mesopores characteristics, which may arise from the void space between the interconnected nanosized building blocks of fractured shell. The BET surface areas of CNTs@ZIF-8 (fractured) and Ni(OH)₂@ZIF-8 are 1055.8 and 1592.7 m² g⁻¹, respectively. Calculated from the desorption branch using the BJH modal, the average pore size distributions of CNTs@ZIF-8 (fractured) and Ni(OH)₂@ZIF-8 center at 3.705 and 3.715 nm.

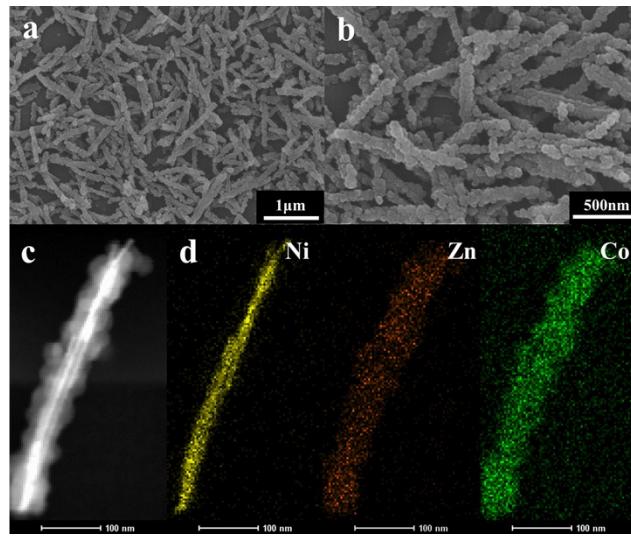


Fig. S8 (a and b) SEM images, (c) HAADF-STEM image and (d) elemental mapping of $\text{Ni}(\text{OH})_2@\text{ZIF-8}$ (Zn, Co) nanowires.

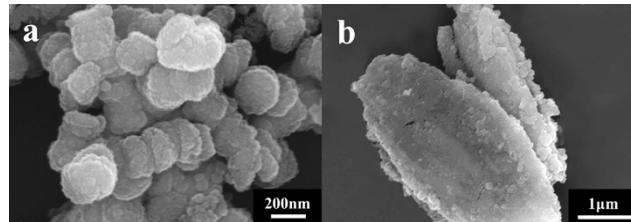


Fig. S9 SEM images of (a) $\text{Fe}_2\text{O}_3@\text{ZIF-8}$ nanoplates and (b) ZIF-L@ZIF-8 microplates.

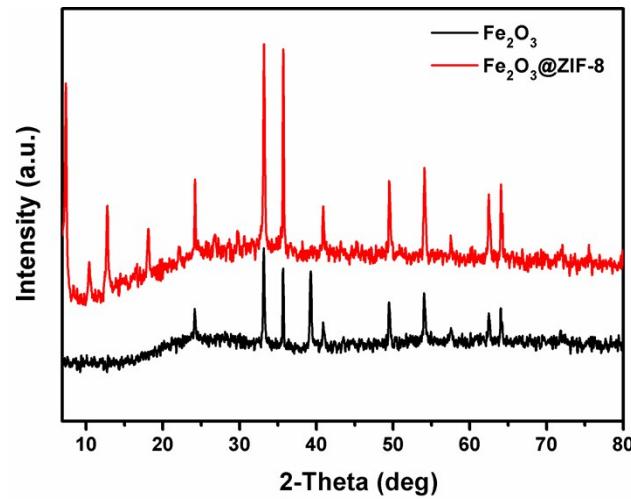


Fig. S10 XRD patterns of Fe_2O_3 and $\text{Fe}_2\text{O}_3@\text{ZIF-8}$ nanoplates.

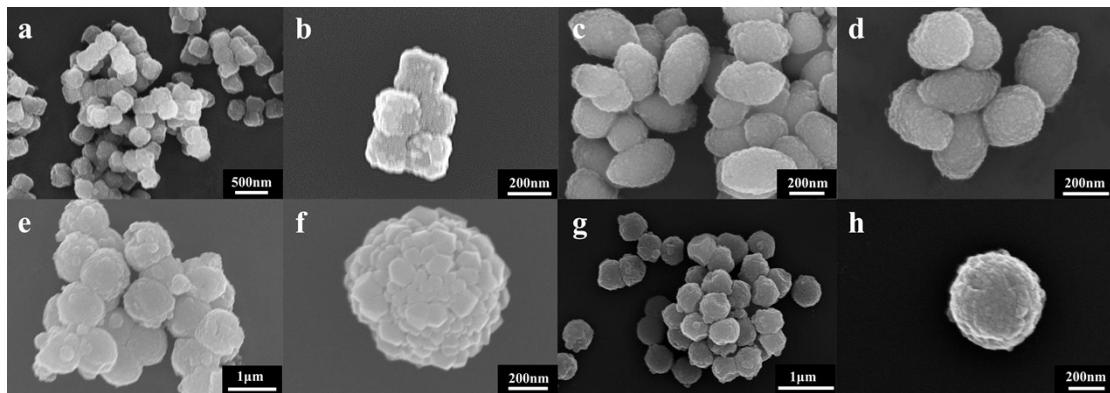


Fig. S11 SEM images of 3D core@shell composites. (a and b) Ni₃[Co(CN)₆]₂@ZIF-8 nanocubes; (c and d) Fe₂O₃@ZIF-8 nanospindles; (e and f) SiO₂@ZIF-8 hollow nanospheres and (g and h) RF@ZIF-8 nanospheres.

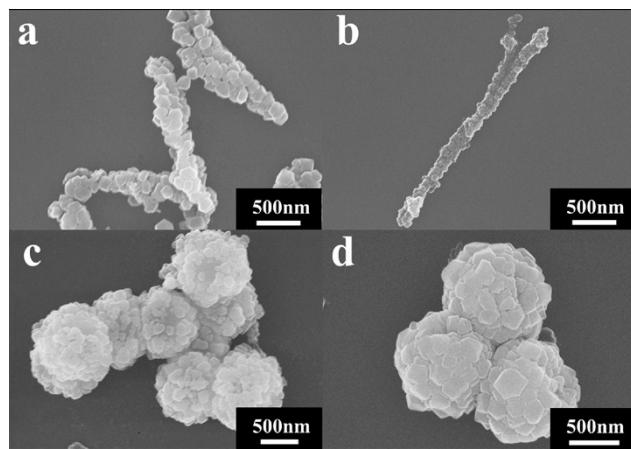


Fig. S12 SEM images of (a) Ni(OH)₂@ZIF-67 nanowires; (b) LiMn₂O₄@ZIF-67 nanowires; (c) SiO₂@ZIF-67 hollow nanospheres and (d) RF@ZIF-67 nanospheres.

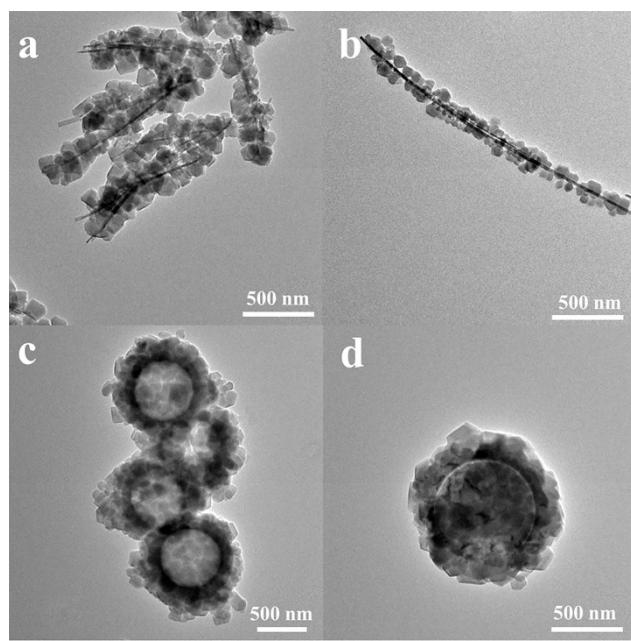


Fig. S13 TEM images of (a) $\text{Ni}(\text{OH})_2@\text{ZIF-67}$ nanowires; (b) $\text{LiMn}_2\text{O}_4@\text{ZIF-67}$ nanowires; (c) $\text{SiO}_2@\text{ZIF-67}$ hollow nanospheres and (d) $\text{RF}@\text{ZIF-67}$ nanospheres.

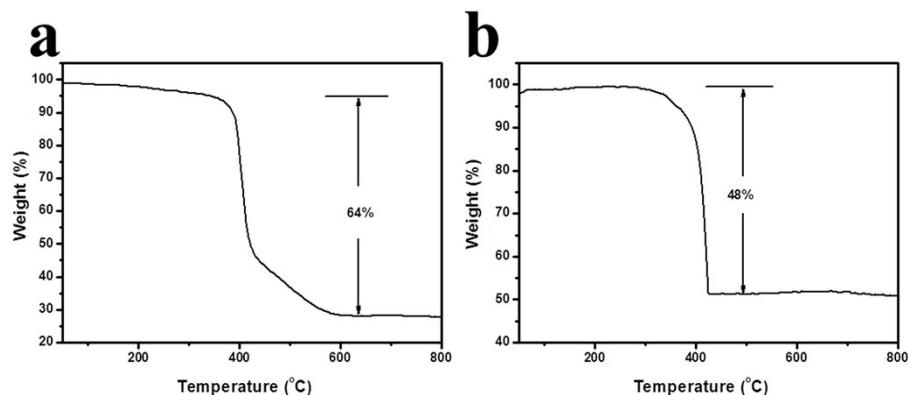


Fig. S14 Thermogravimetric analysis curves of (a) CNTs@ZIF-8 (Zn, Co) and (b) $\text{Ni}(\text{OH})_2@\text{ZIF-8}$ (Zn, Co) nanowires.

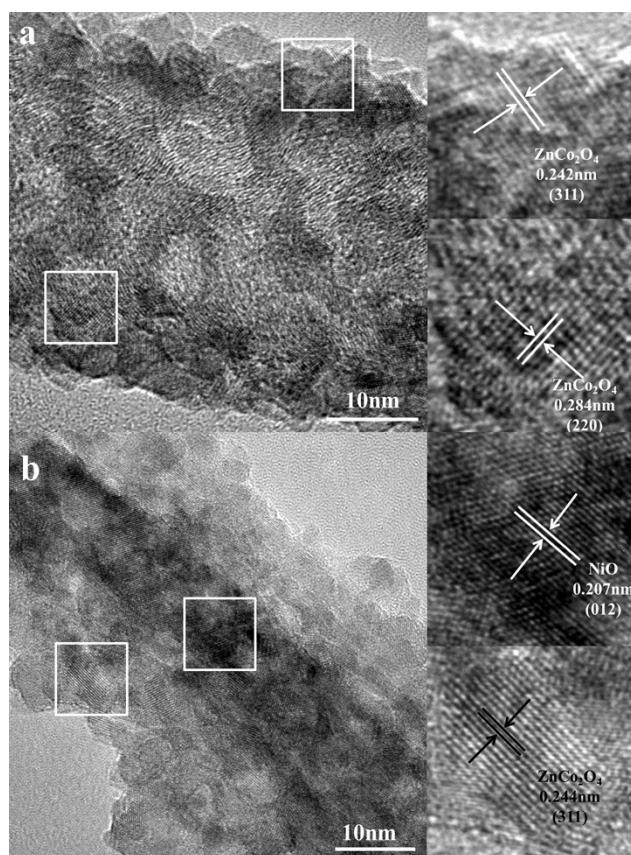


Fig. S15 HRTEM images of (a) CNTs@ZnCo₂O₄ and (b) NiO@ZnCo₂O₄ nanowires.

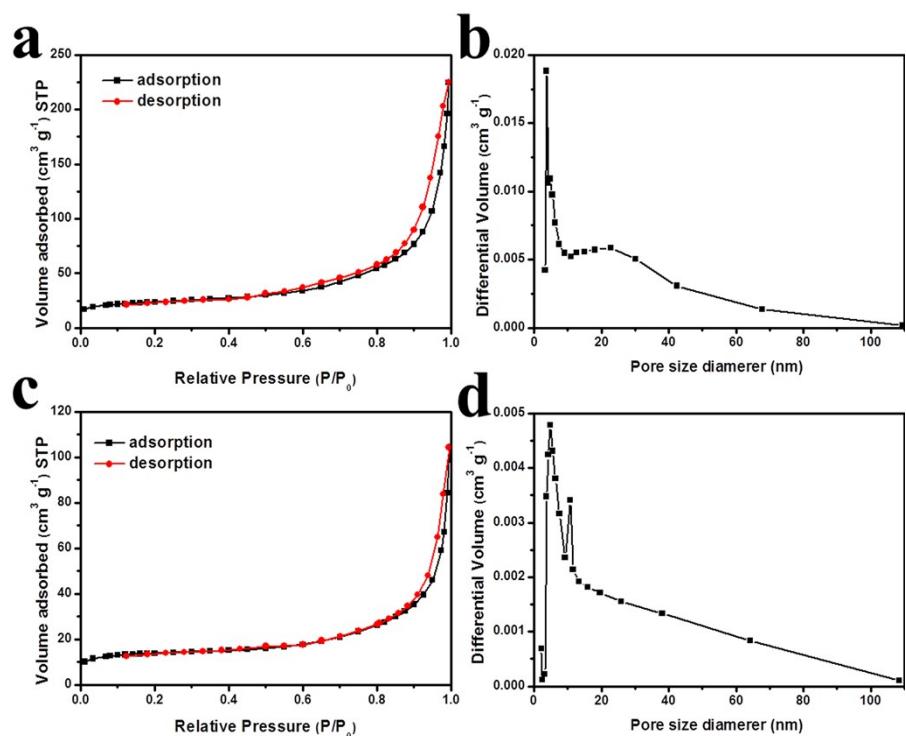


Fig. S16 N₂ adsorption-desorption isotherms and the corresponding pore size distributions of (a and b) CNTs@ZnCo₂O₄ and (c and d) NiO@ZnCo₂O₄ nanowires.

Table S1 The diameter and shell thickness of core@shell CNTs@ZIF-8 (fractured) nanowires synthesized at different precursor's concentrations

	CNTs@ZIF-8-0.5 (fractured)	CNTs@ZIF-8 (fractured)	CNTs@ZIF-8-2 (fractured)
Diameter	66 nm	76 nm	156 nm
Shell thickness	20 nm	26 nm	65 nm

Table S2 Summary of representative anode materials for Li-ion batteries

Anode materials	Cycling performance	Rate capability	Ref.
Peapod-like Co ₃ O ₄ @Carbon nanotube arrays	700 mA h g ⁻¹ after 100 cycles at 100 mA g ⁻¹	453 mA h g ⁻¹ at 1000 mA g ⁻¹	1
Two-dimensional porous micro/nano NiO	568 mA h g ⁻¹ after 50 cycles at 200 mA g ⁻¹	300 mA h g ⁻¹ at 1000 mA g ⁻¹	2
ZnO-loaded/porous carbon composite	654 mA h g ⁻¹ after 100 cycles at 100 mA g ⁻¹	497 mA h g ⁻¹ at 1000 mA g ⁻¹	3
Porous Co ₃ O ₄ /CuO composite	839 mA h g ⁻¹ after 150 cycles at 100 mA g ⁻¹	700 mA h g ⁻¹ at 1000 mA g ⁻¹	4
Carbon nanofibers anchored with Zn _x Co _{3-x} O ₄ nanocubes	600 mA h g ⁻¹ after 300 cycles at 500 mA g ⁻¹	337 mA h g ⁻¹ at 1000 mA g ⁻¹	5
Porous polyhedral ZnCo ₂ O ₄ /CNTs composites	864.6 mA h g ⁻¹ after 150 cycles at 100 mA g ⁻¹	606 mA h g ⁻¹ at 2000 mA g ⁻¹	6
Hierarchical ZnCo ₂ O ₄ /NiO core/shell nanowire arrays	357 mA h g ⁻¹ after 50 cycles at 100 mA g ⁻¹	200 mA h g ⁻¹ at 800 mA g ⁻¹	7
CNTs@ZnCo ₂ O ₄ nanowires	750 mA h g ⁻¹ after 100 cycles at 100 mA g ⁻¹	460 mA h g ⁻¹ at 1000 mA g ⁻¹	This work
NiO@ZnCo ₂ O ₄ nanowires	1002 mA h g ⁻¹ after 100 cycles at 100 mA g ⁻¹	462 mA h g ⁻¹ at 1000 mA g ⁻¹	

Reference

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