

## Supplementary Information

Improved performance in micron-sized silicon anodes

by *in-situ* polymerization of its acrylic acid-based slurry

Chao Li,<sup>a,b</sup> Tongfei Shi,<sup>c</sup> Hideya Yoshitake,<sup>d</sup> and Hongyu Wang<sup>\*a</sup>

<sup>a</sup>*State Key Laboratory of Electroanalytical Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, 5625 Renmin Street, Changchun 130022, China*

<sup>b</sup>*University of Chinese Academy of Sciences, Beijing 100049, China*

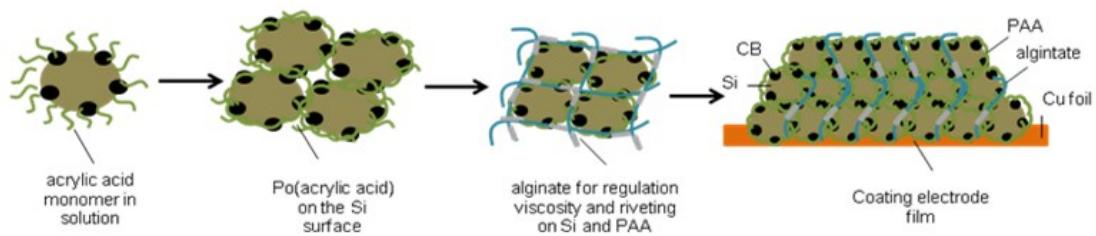
<sup>c</sup>*State Key Laboratory of Polymer Physics and Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences*

<sup>d</sup>*Faculty of Engineering, Yamagata University, Jyonan 4-3-16, Yonezawa 992-8510, Japan*

\* Corresponding author. Tel/Fax: 86-431-85262287. Email address: [hongyuwang@ciac.ac.cn](mailto:hongyuwang@ciac.ac.cn) (H. Wang)

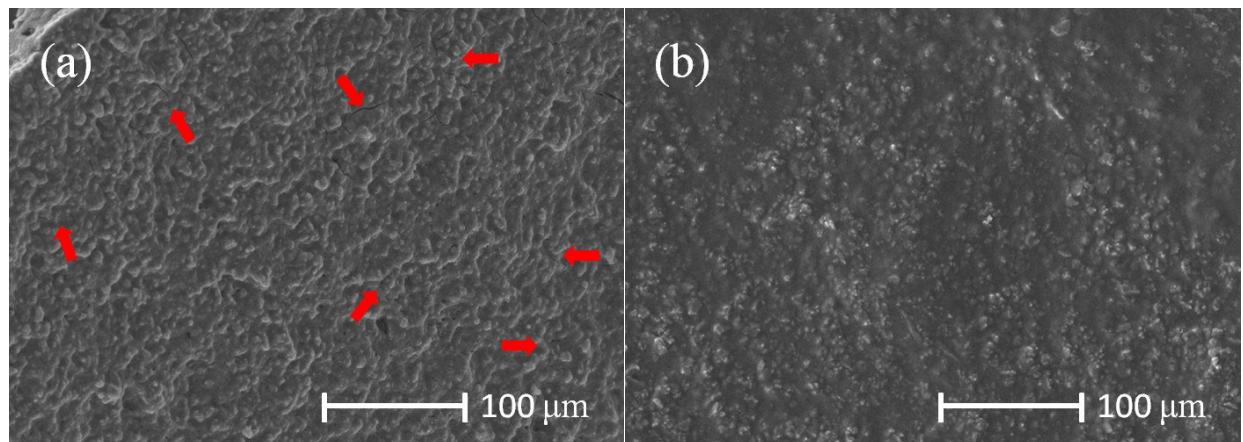
## Fig. S1

Reaction process: the acrylic acid monomer of solution can be polymerized at 72°C, and the C=C double bond in acrylic acid is broken. At the same time, the long chain is connected with other monomers. The dissolved-alginate in water plays the function as a thickener and increase the viscosity of solution. While the acrylic acid is added in the solution, it can form weak interaction with alginate and Si-O on the silicon surface owing to the electrostatic force. After polymerization, the long chain of poly(acrylic acid) realize the function of adhesive each others.



**Fig. S2**

SEM images of two kinds of Si electrodes, including the PA-Si electrodes of traditional mechanically-mixed slurry (a) and the cPA-Si electrodes obtained from slurry copolymerization method (b).



**Fig. S3**

AFM images of cPA-Si and PA-Si electrodes at a lithiation state, containing (a) 2D and (c) 3D images of PA-Si, (b) 2D and (d) 3D images of cPA-Si electrodes.

