Polypyrrole modified porous poly (ether sulfone) membranes with high performance for vanadium flow batteries

Zhizhang Yuan^{a,c}, Qing Dai^d, Yuyue Zhao^{a,c}, Wenjing Lu^{a,c}, Xianfeng

Li^{a,b*}, Huamin Zhang^{a,b*}

^a Division of Energy Storage, Dalian Institute of Chemical Physics,

Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023 (P.

R. China), Email: lixianfeng@dicp.ac.cn, zhanghm@dicp.ac.cn.

^b Collaborative Innovation Center of Chemistry for Energy Materials (iChEM), Dalian 116023 (P. R. China).

^c University of Chinese Academy of Sciences, Beijing 100039 (P. R. China).

^d Faculty of Chemical, Environmental and Biological Science and Technology, Dalian University of Technology.

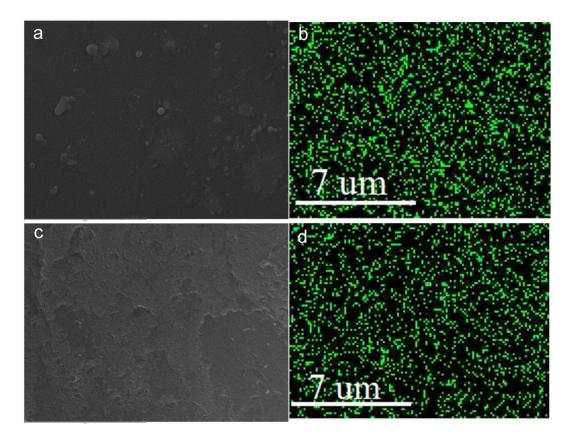


Fig. S1 Surface morphology of PPY/PES porous membranes (a) before and (c) after the cycling test; the corresponding EDS mapping of N distribution (b) before and (d) after the cycling test, confirming the uniform distribution and stability of the PPY in the membrane.

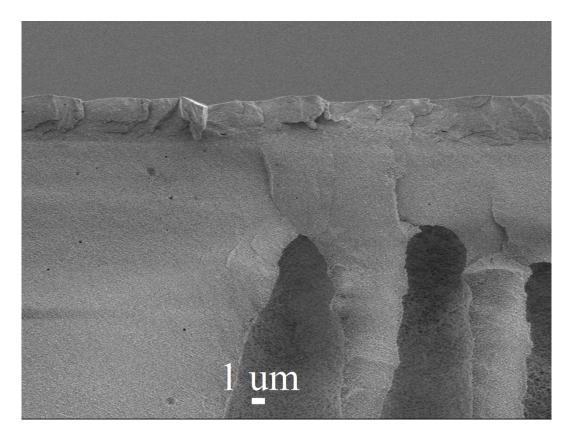


Fig. S2 Magnified cross section morphology of the substrate.

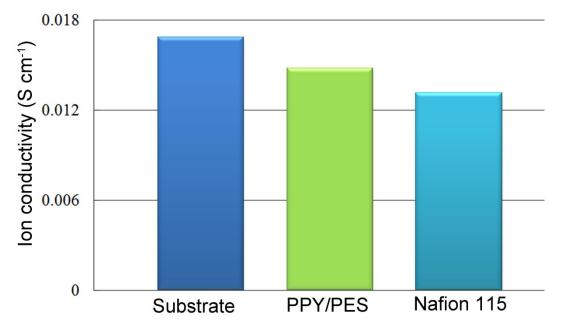


Fig. S3 Ion conductivity of the substrate, PPY/PES and Nafion 115 membranes.

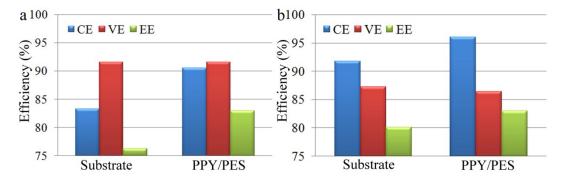


Fig. S4 The efficiency of a vanadium flow battery installed with substrate and PPY/PES with different membrane thickness (a: $85 \pm 5 \mu m$, b: $140 \pm 5 \mu m$) at 80 mA cm⁻².