Supporting Information

Nickel-titanium Oxide as a Novel Anode Material for Rechargeable Sodium-ion Batteries

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Figure S1. a) XRD pattern and b) FE-SEM image of as-prepared NiTiO₃ powder samples that were synthesised through hydrothermal process.



Figure S2. a) XRD pattern and b) FE-SEM image of sputter-deposited NTO thin films on Cu substrate.



Figure S3. SEM-EDS elemental mapping of sputter-deposited NTO thin film.



Figure S4. Raman spectra for the NTO thin film, sputter-deposited on Cu sheet.



Figure S5. Core-level XPS spectra for: a) Ni 2p, b) Ti 2p, and c) O 1s in NTO thin film, sputter-deposited on Cu sheet.



Figure S6. a) Cyclic voltammograms measured at 0.1 mV s⁻¹ and b) charge-discharge-potential profiles at the specific current of 50 mA g⁻¹ for the sputter-deposited NTO-thin-film electrode.



Figure S7. FE-SEM images of the: a-c) NTO-powder and d-f) thin-film electrodes. (a, d) pristine, (b, e) sodiated (after 1st discharge), and (c, f) desodiated (after 1st charge).



Figure S8. a) Rate capability and b) cyclability of the NTO-thin-film-based electrode.



Figure S9. Nyquist-impedance spectra for the NTO-thin-film-based electrode: a) Before and after all discharge-charge tests, and b) after 1st discharge and 1st charge (symbols indicate the measured data, and lines indicate the simulated results).

	Before test	1 st discharge	1 st charge	After 200 cycles
$R_{e}(\Omega)$	2.28	2.8	9.47	18.44
$R_{f}\left(\Omega\right)$	1030	9.48	29.13	49.5
$R_{ct}(k\Omega)$	165.7	0.342	0.465	1.407
$Q_1(\mu F)$	1.87	1.25	3.21	5.93
$Q_2(\mu F)$	0.313	0.111	0.101	0.727

Table S1. Electrolyte resistance (R_e) , film resistance (R_f) , and charge-transfer resistance (R_{ct}) calculated from the Nyquist-impedance spectra that were recorded before, after, and during the charge-discharge tests for the NTO-thin-film electrode.