

Supporting Information for

Facile fabrication of Co₂CuS₄ nanoparticle anchored N-doped graphene for high-performance asymmetric supercapacitors

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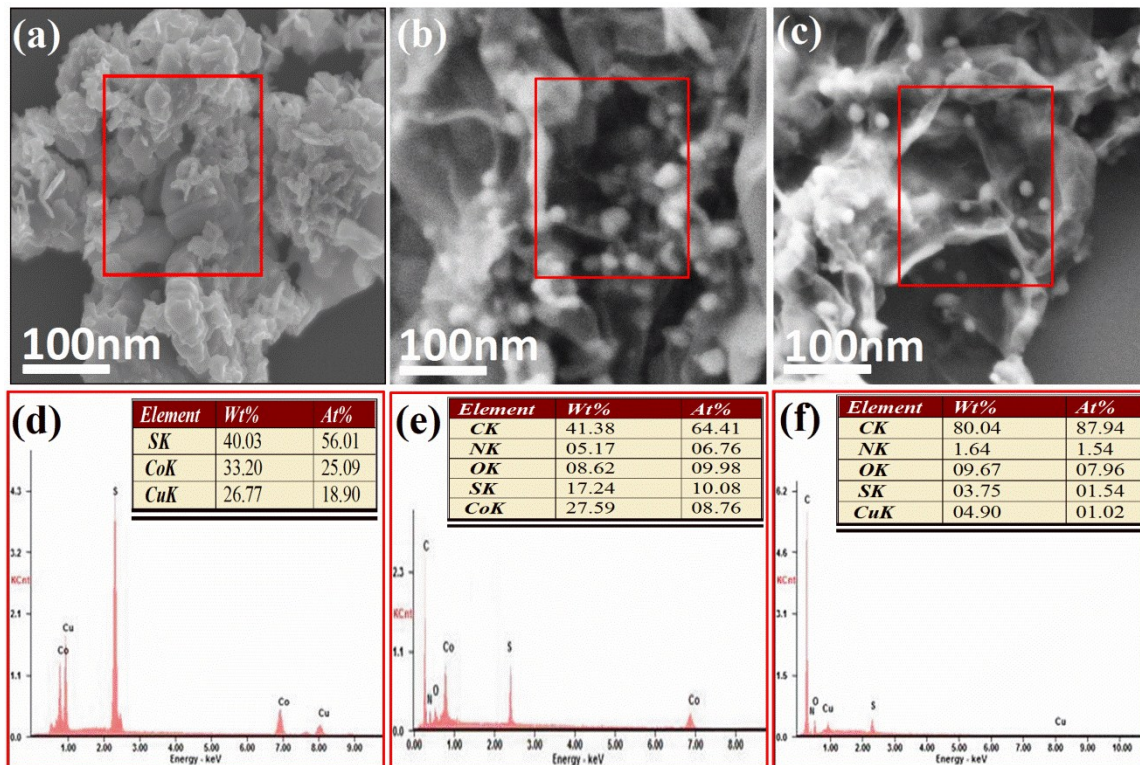


Fig. S1 (a-c) FE-SEM images, and (d-f) EDAX spectrum of pure Co_2CuS_4 , CoS/NG and $\text{Cu}_2\text{S/NG}$, respectively.

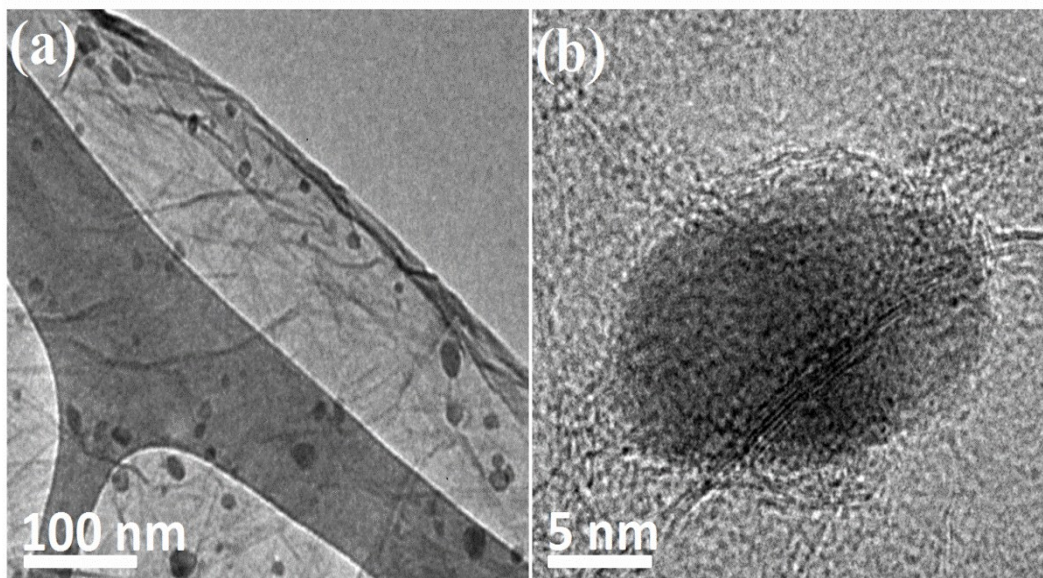


Fig. S2. TEM image with different magnification of Co₂CuS₄/NG composite (after 5 h strong ultrasonication and 12 h stirring in ethanol).

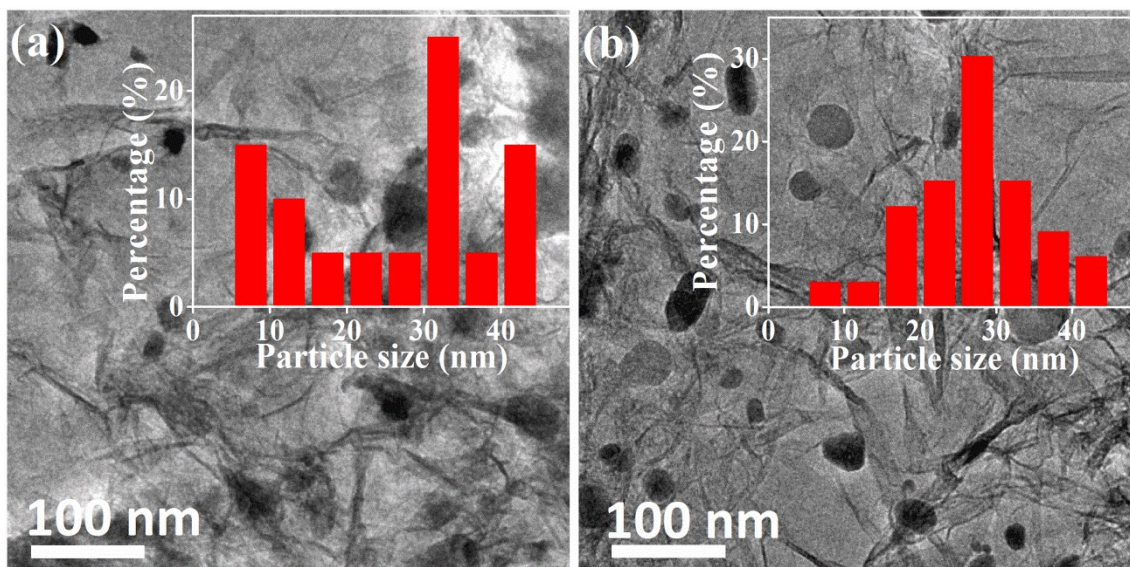


Fig. S3 TEM images of (a) CoS/NG, and (b) Cu₂S/NG composites.

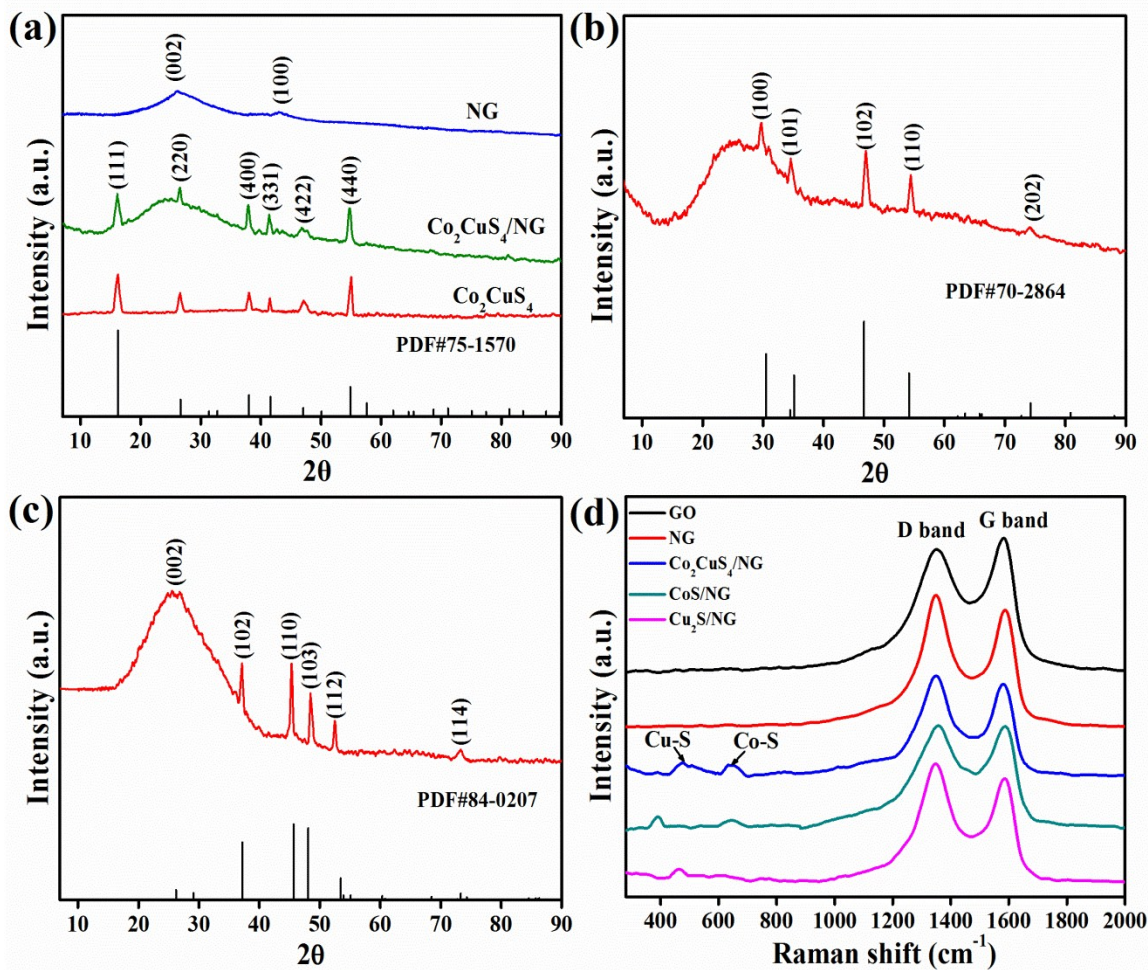


Fig. S4 XRD pattern of (a) NG, pure Co_2CuS_4 and $\text{Co}_2\text{CuS}_4/\text{NG}$ composite, (b) CoS/NG and (c) $\text{Cu}_2\text{S}/\text{NG}$, (d) Raman spectra of GO, NG, CoS/NG , $\text{Cu}_2\text{S}/\text{NG}$ and $\text{Co}_2\text{CuS}_4/\text{NG}$.

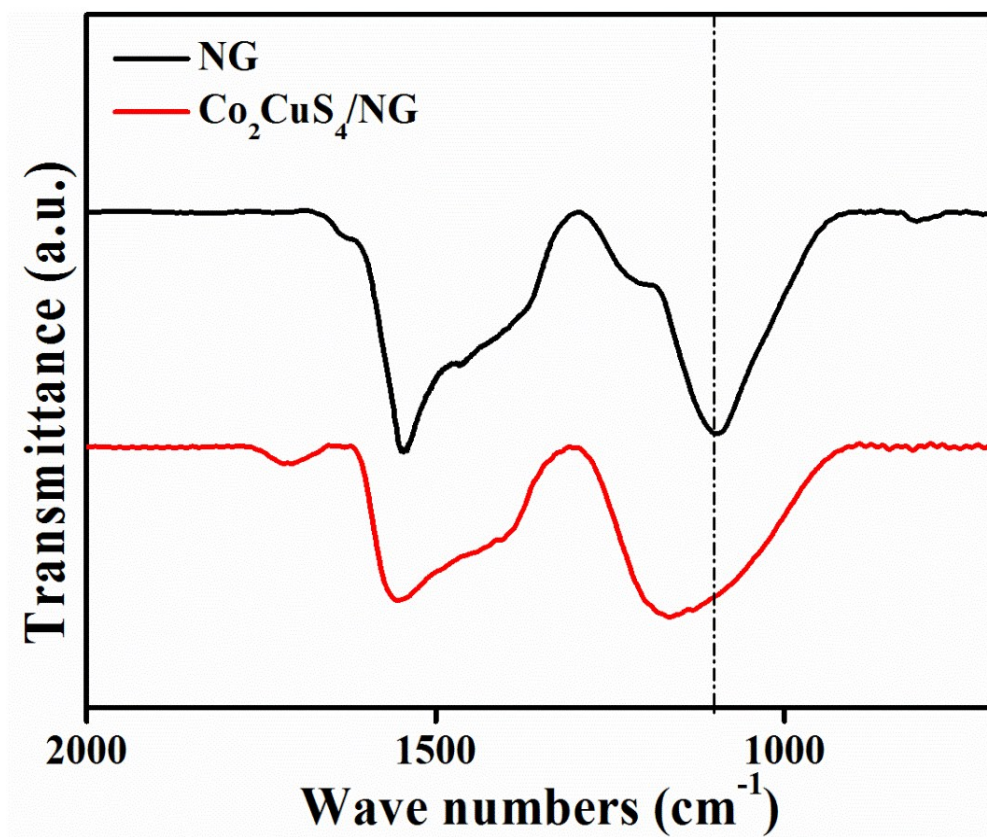


Fig. S5 FT-IR spectra of NG and Co₂CuS₄/NG composite.

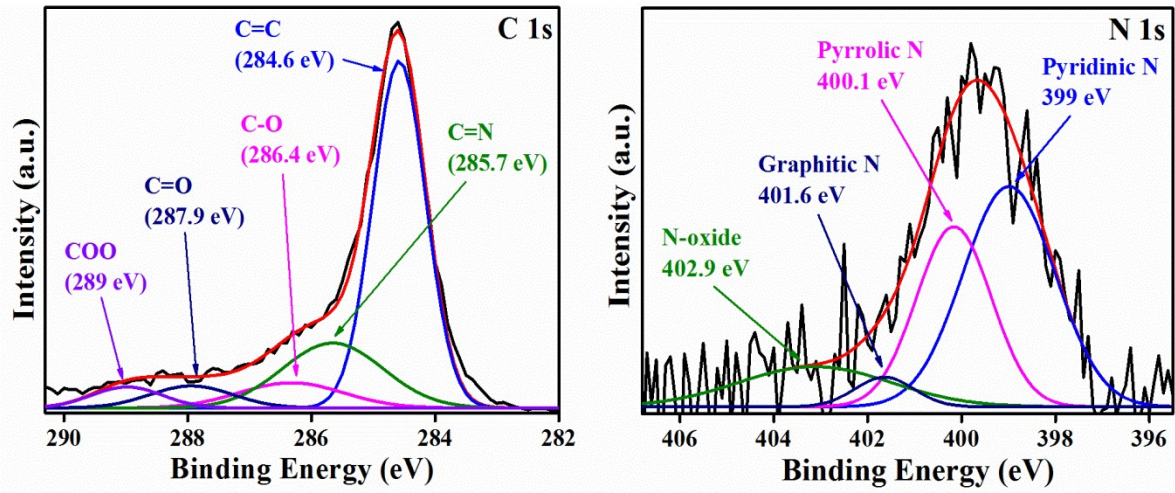


Fig. S6 High-resolution (a) C 1s spectrum, (b) N 1s spectrum of NG.

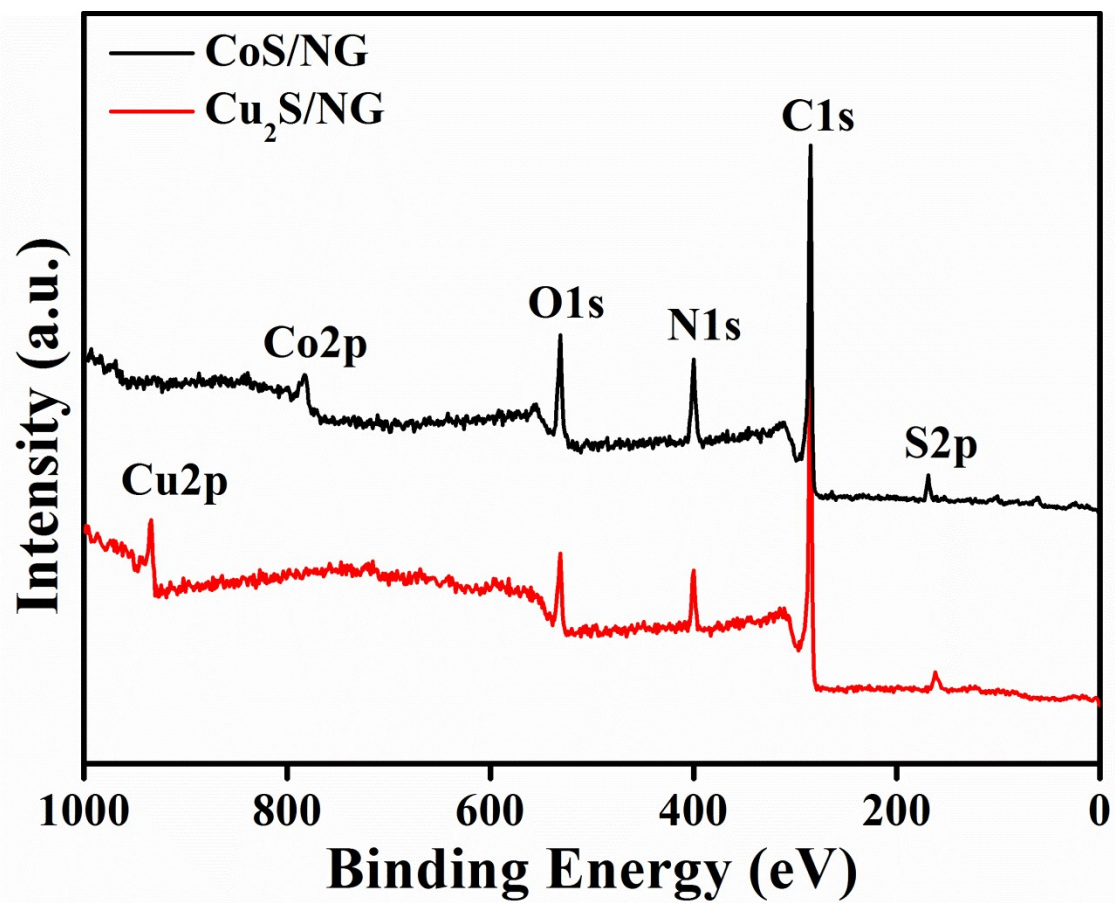


Fig. S7 XPS survey for CoS/NG and Cu₂S/NG.

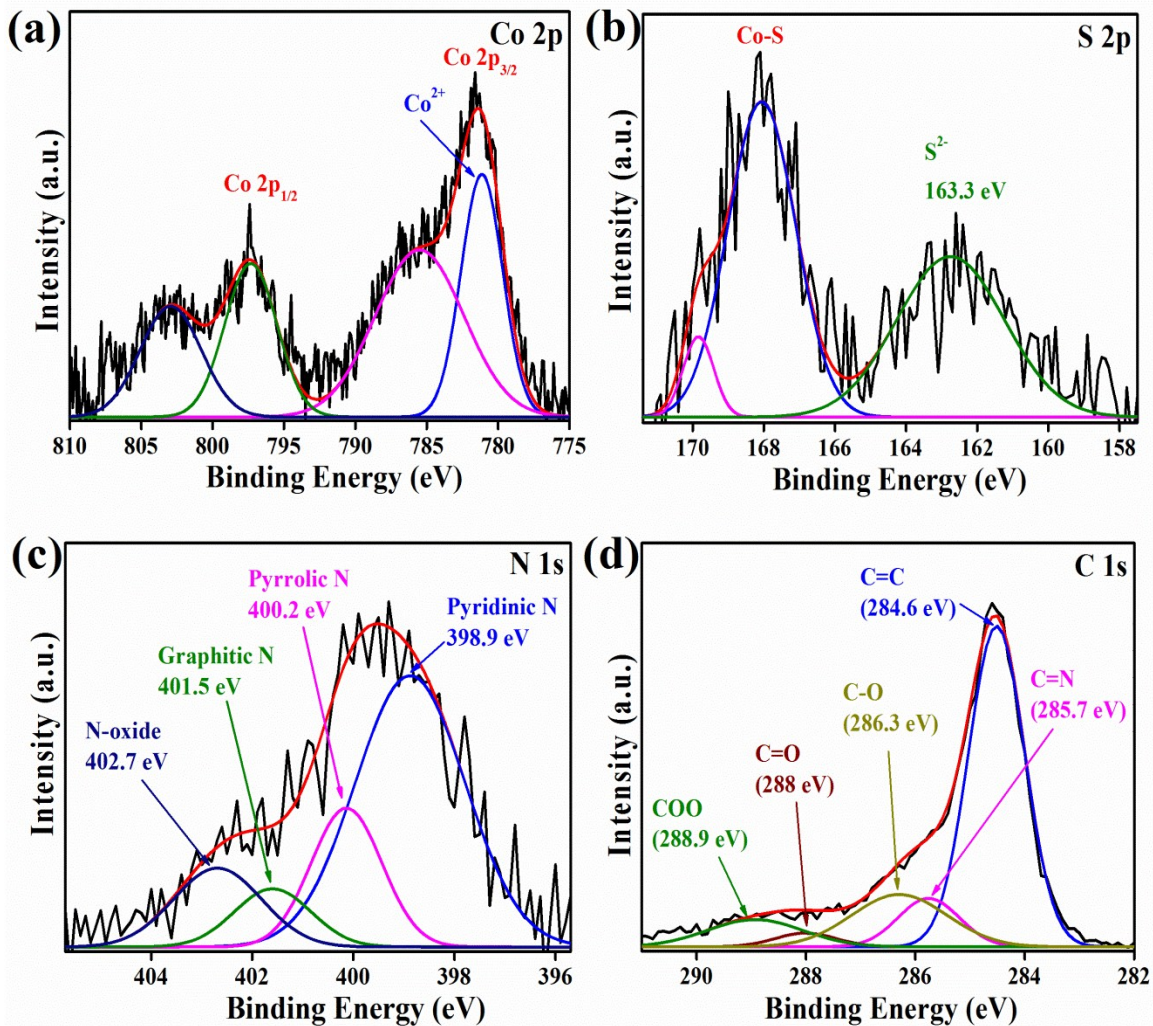


Fig. S8 High-resolution (a) Co 2p spectrum, (b) S 2p spectrum, (c) N 1s spectrum and (d) C 1s spectrum of CoS/NG composite.

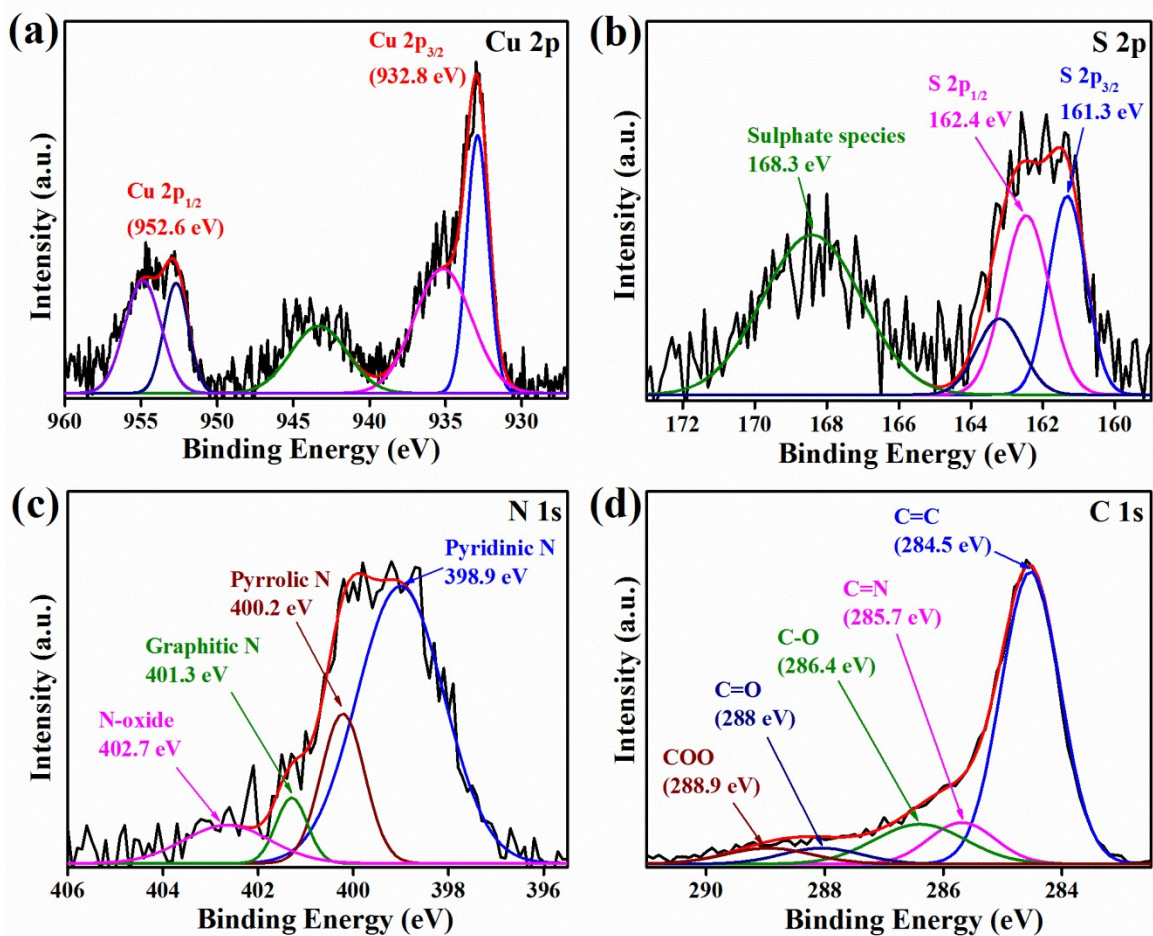


Fig. S9 High-resolution (a) Cu 2p spectrum, (b) S 2p spectrum, (c) N 1s spectrum and (d) C 1s spectrum of Cu₂S/NG composite.

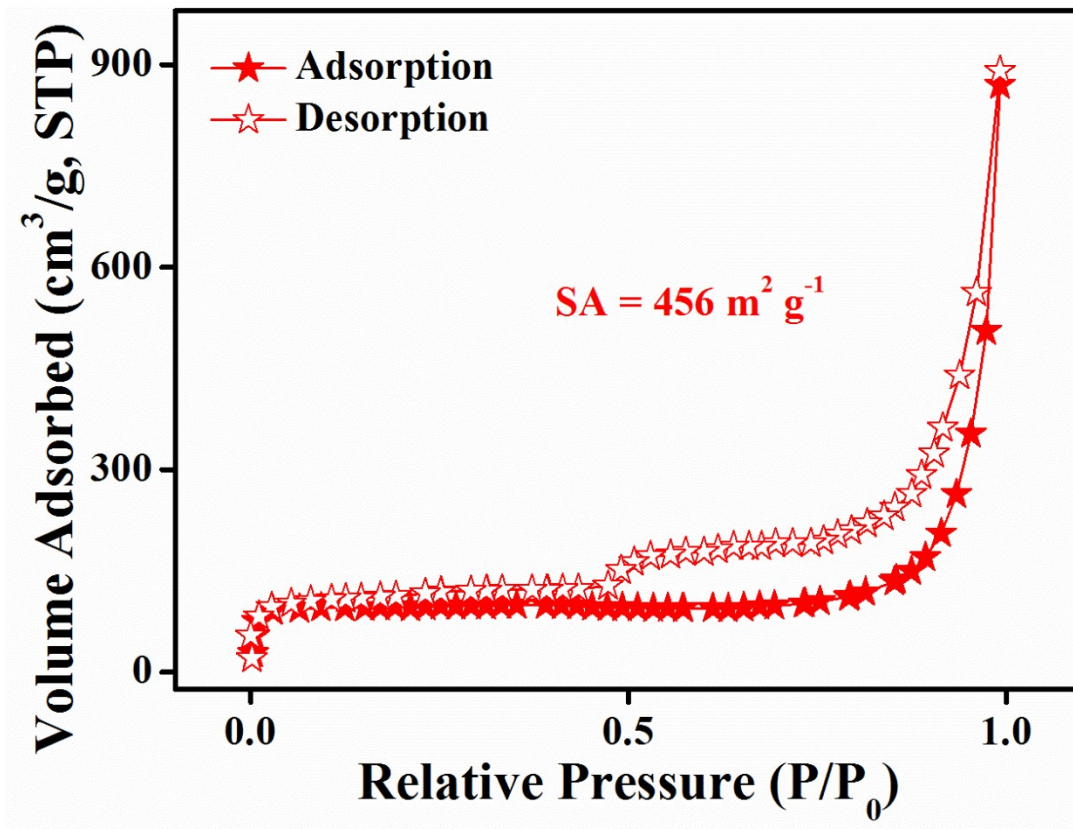


Fig. S10 N₂ adsorption–desorption isotherms of Co₂CuS₄/NG composite.

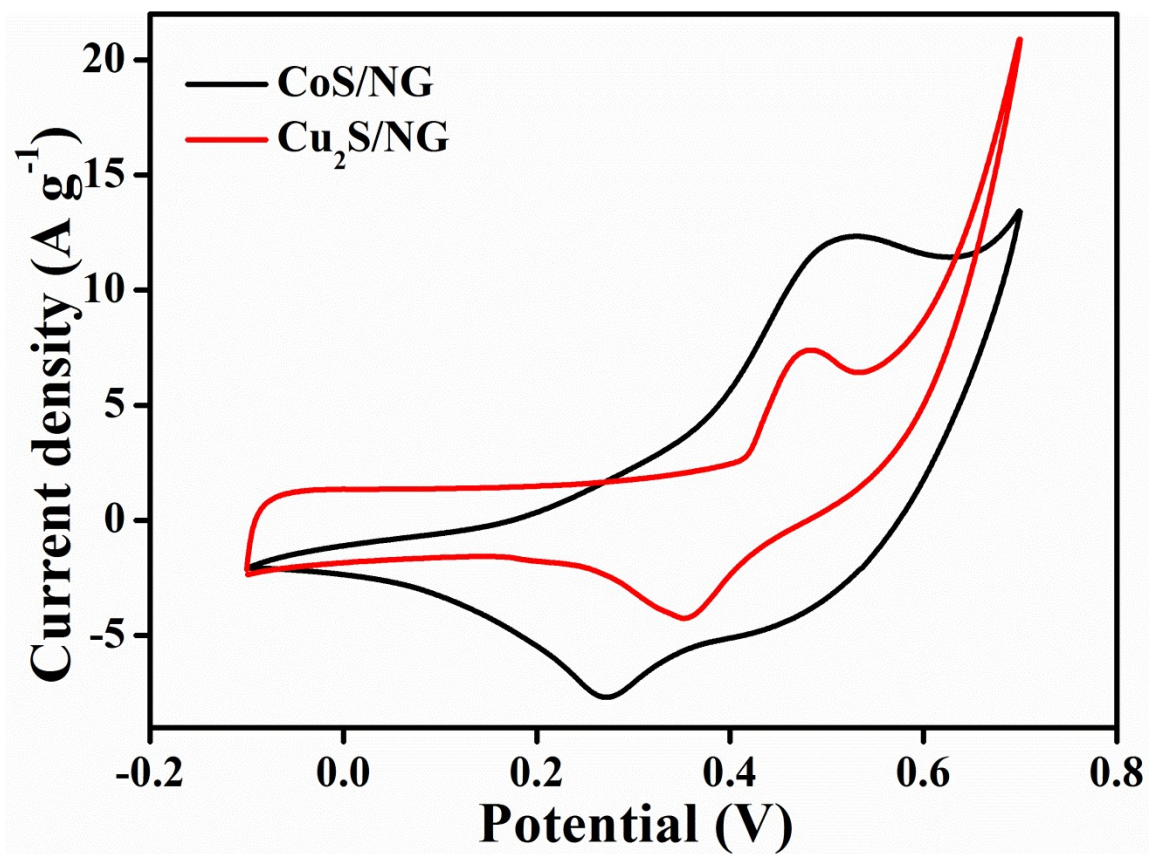


Fig. S11 CV curves of CoS/NG and Cu₂S/NG composite at 10 mV s⁻¹.

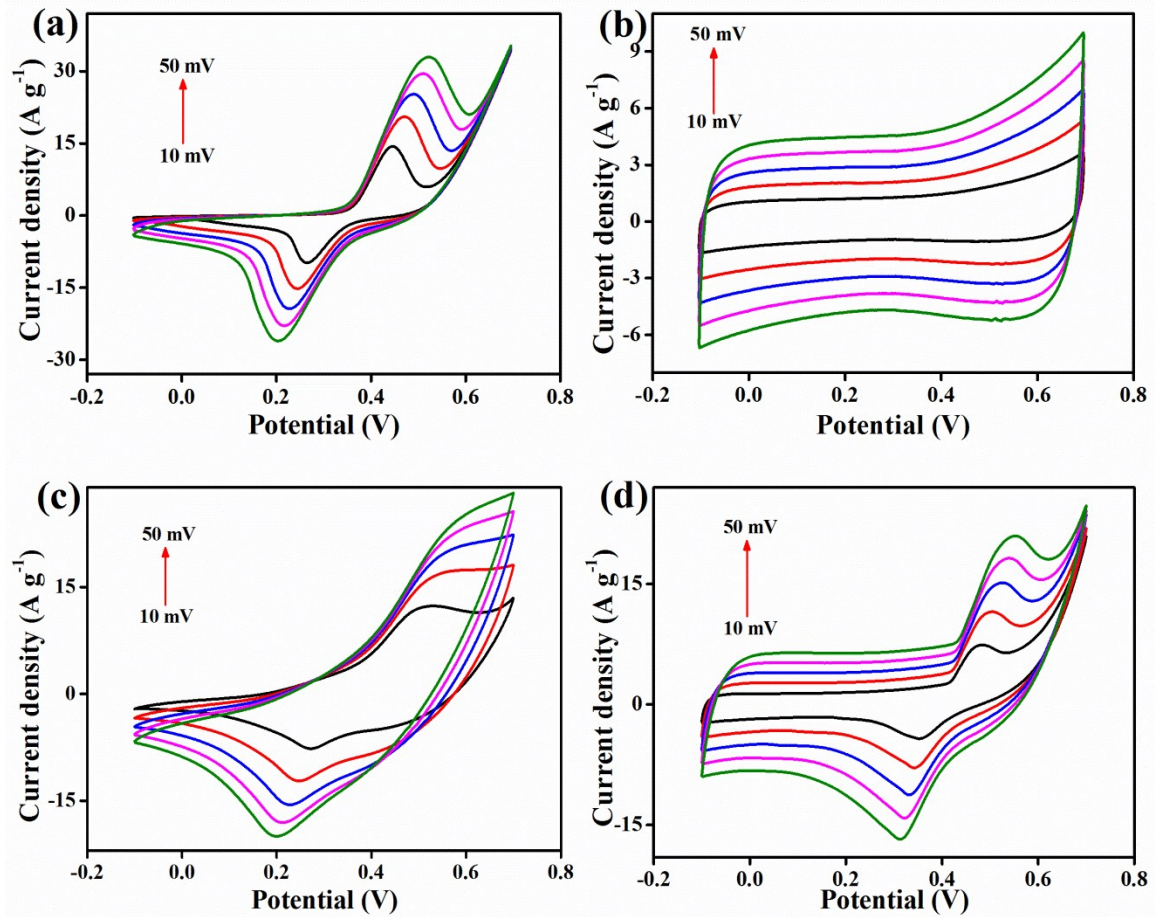


Fig. S12 CV curves at different sweep rates (10-50 $mV\ s^{-1}$) of (a) pure Co_2CuS_4 , (b) NG, (c) CoS/NG and (d) Cu_2S/NG .

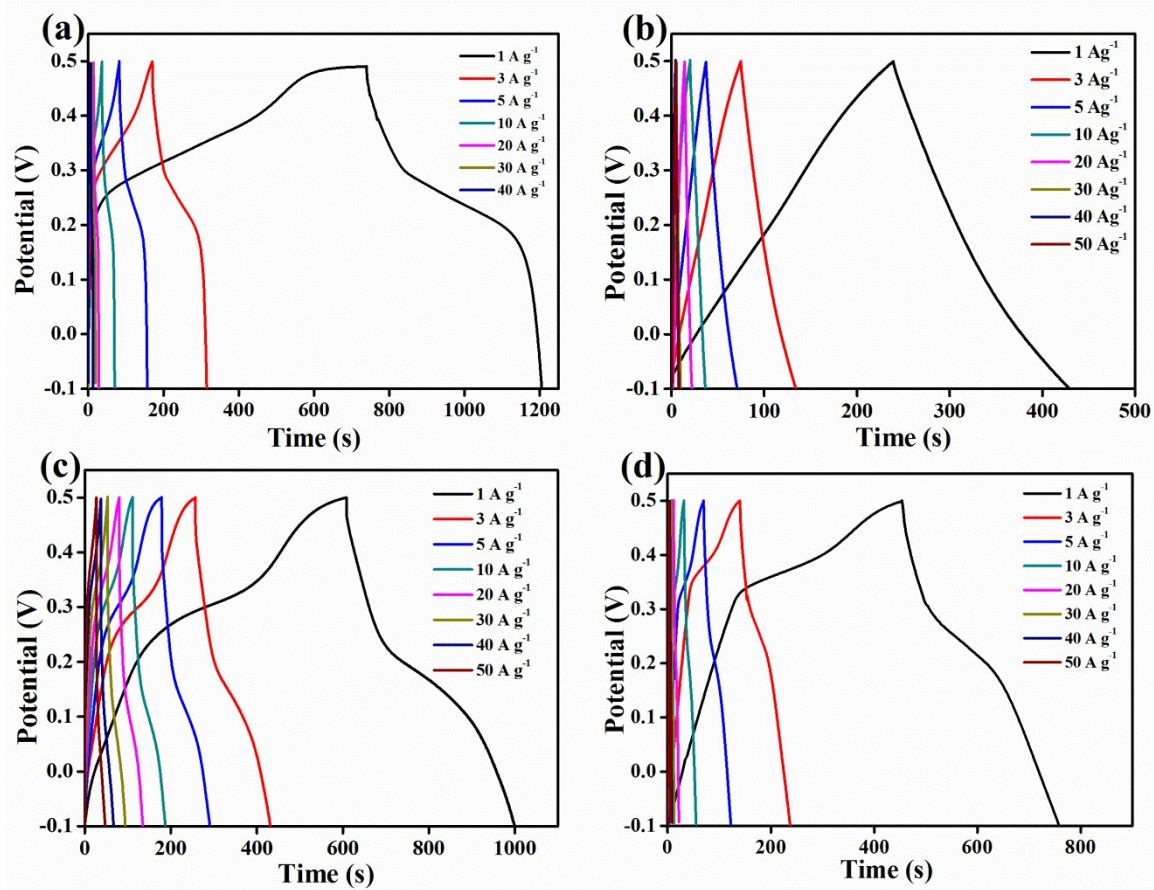


Fig. S13 Galvanostatic charge–discharge curves at different current densities of (a) pure Co_2CuS_4 , (b) NG, (c) CoS/NG and (d) $\text{Cu}_2\text{S}/\text{NG}$.

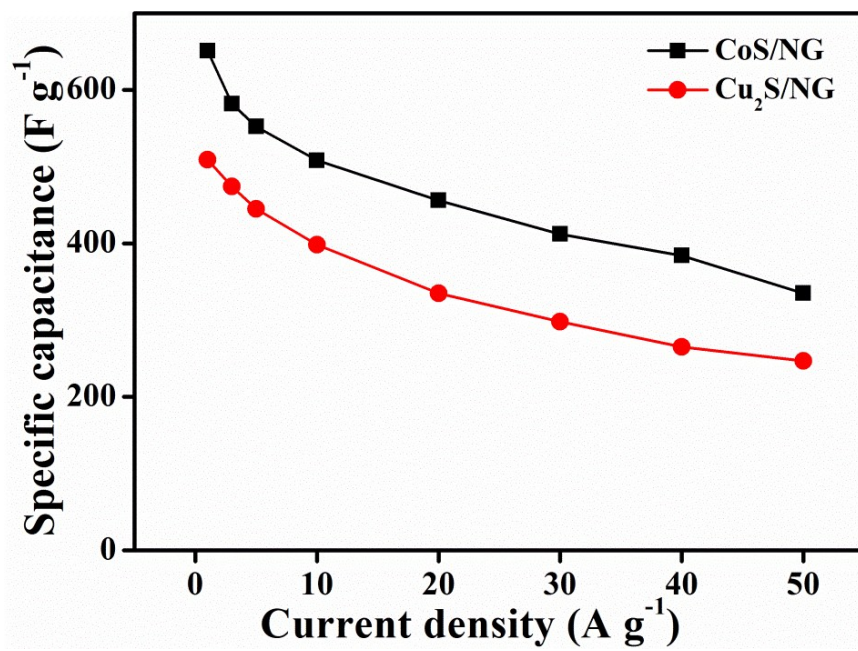


Fig. S14 Specific capacitance vs. current density of CoS/NG and Cu₂S/NG.

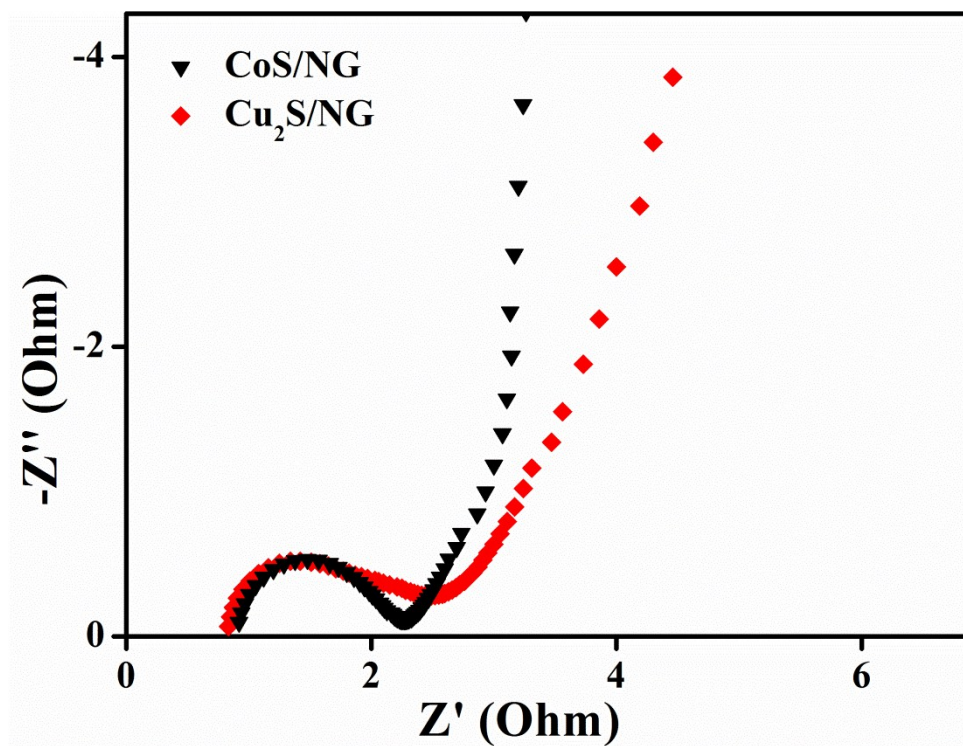


Fig. S15 Nyquist plots of CoS/NG and Cu₂S/NG.

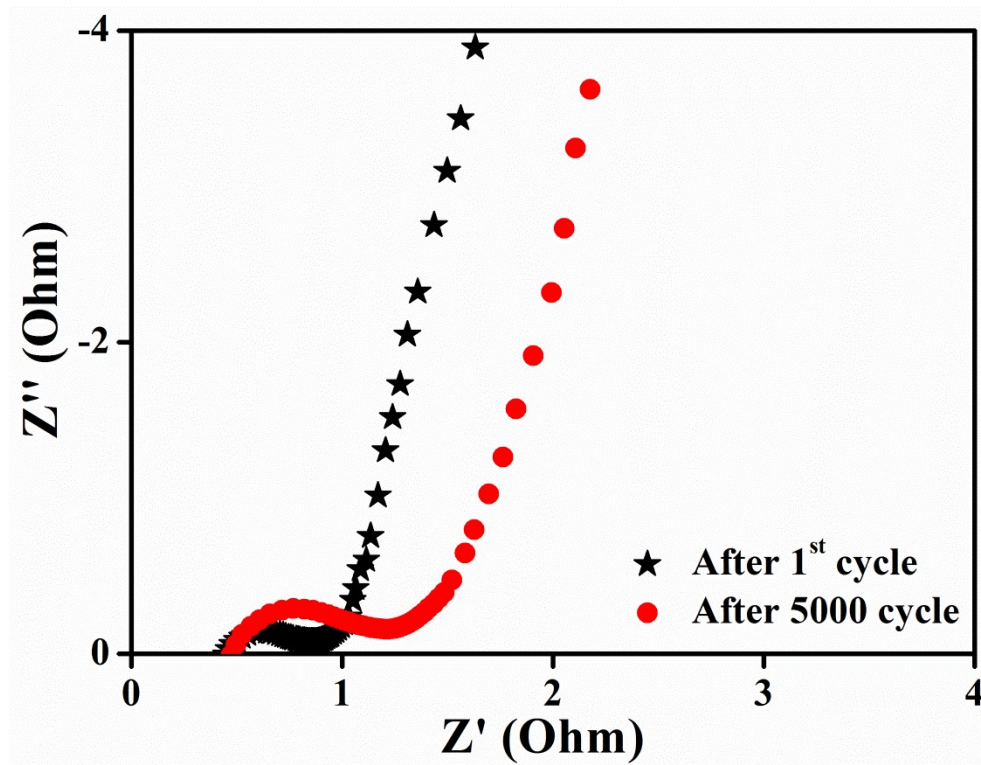


Fig. S16 Nyquist plots of $\text{Co}_2\text{CuS}_4/\text{NG}$ composite (measured during the cycle life test).

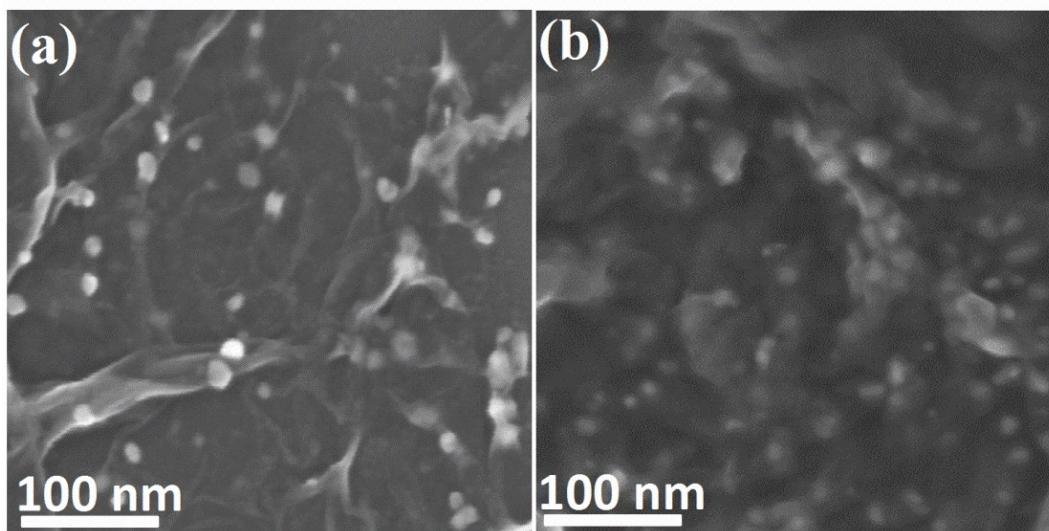


Fig. S17 FE-SEM image of Co₂CuS₄/NG composite (a) before and (b) after 5000 cycles test.

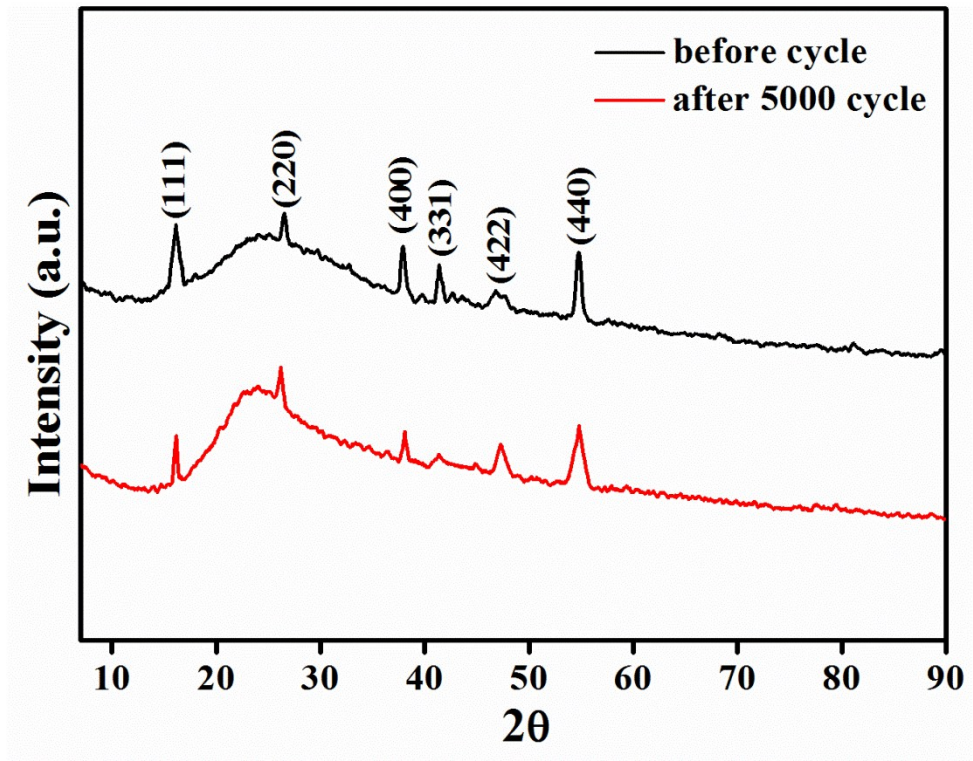


Fig. S18 XRD pattern of $\text{Co}_2\text{CuS}_4/\text{NG}$ composite before and after stability test.

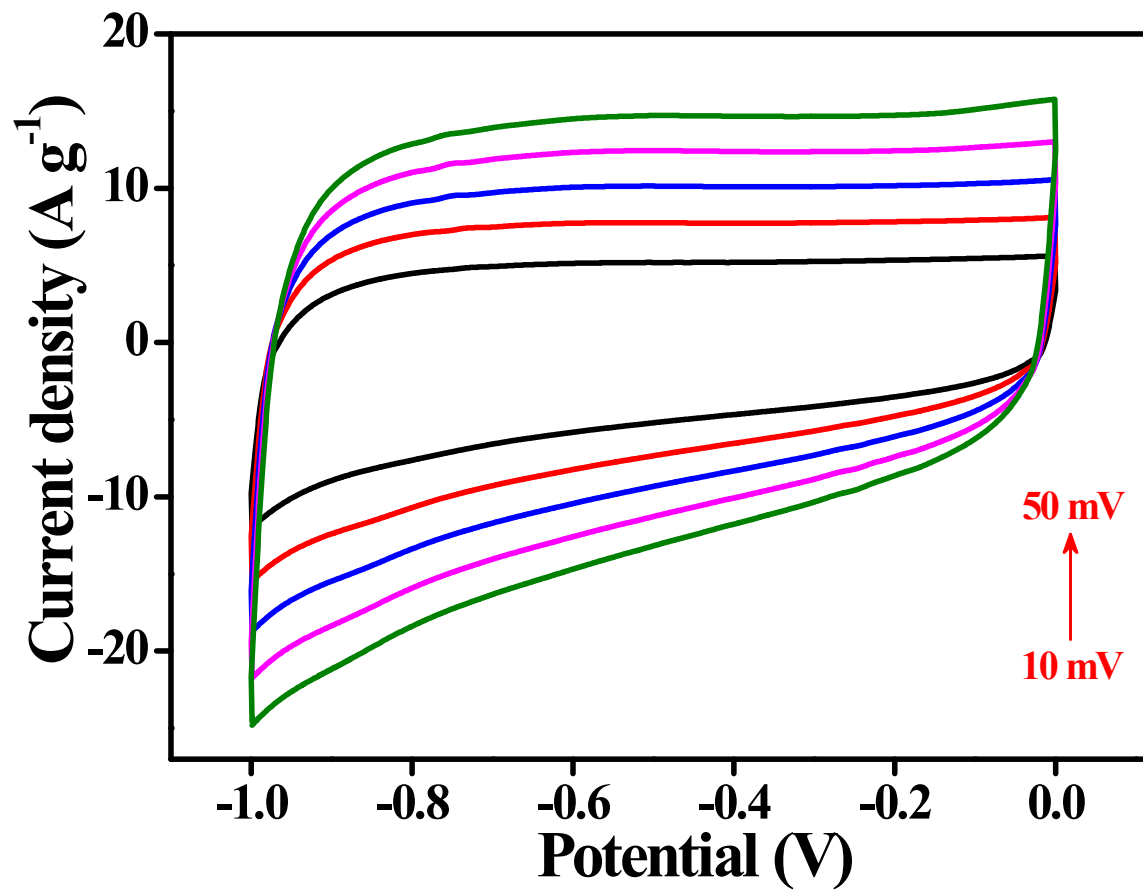


Fig. S19 CV curves of NG at different sweep rates (10-50 mV s⁻¹)

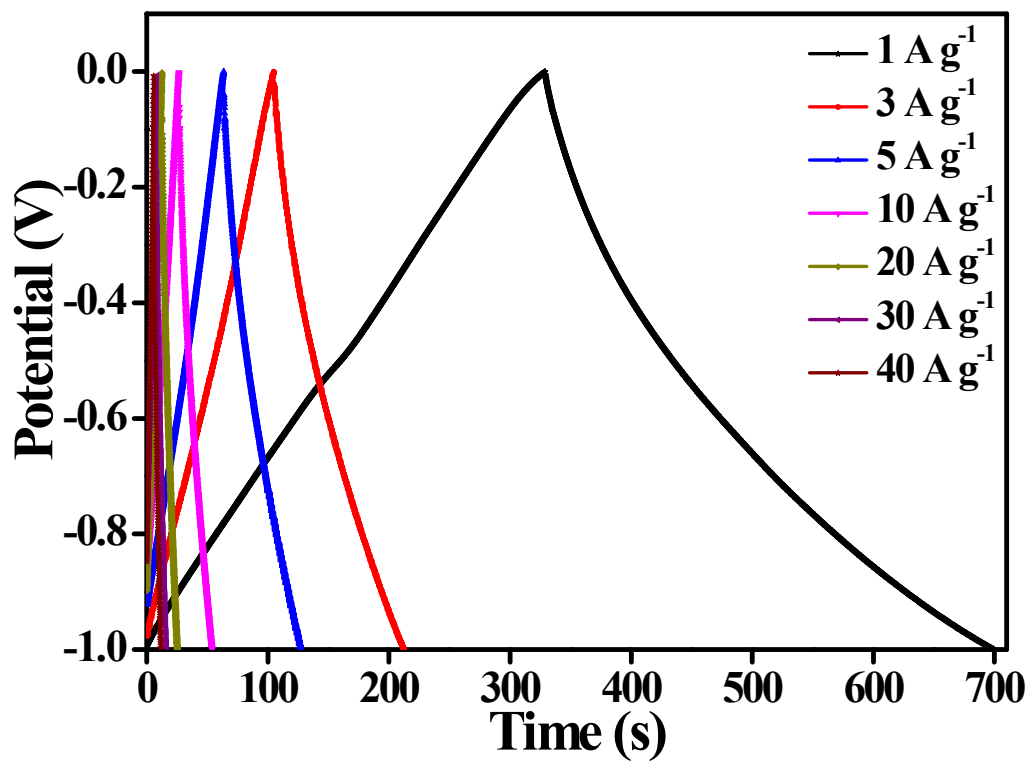


Fig. S20 Galvanostatic charge-discharge curves of NG (negative electrode) at different current densities.

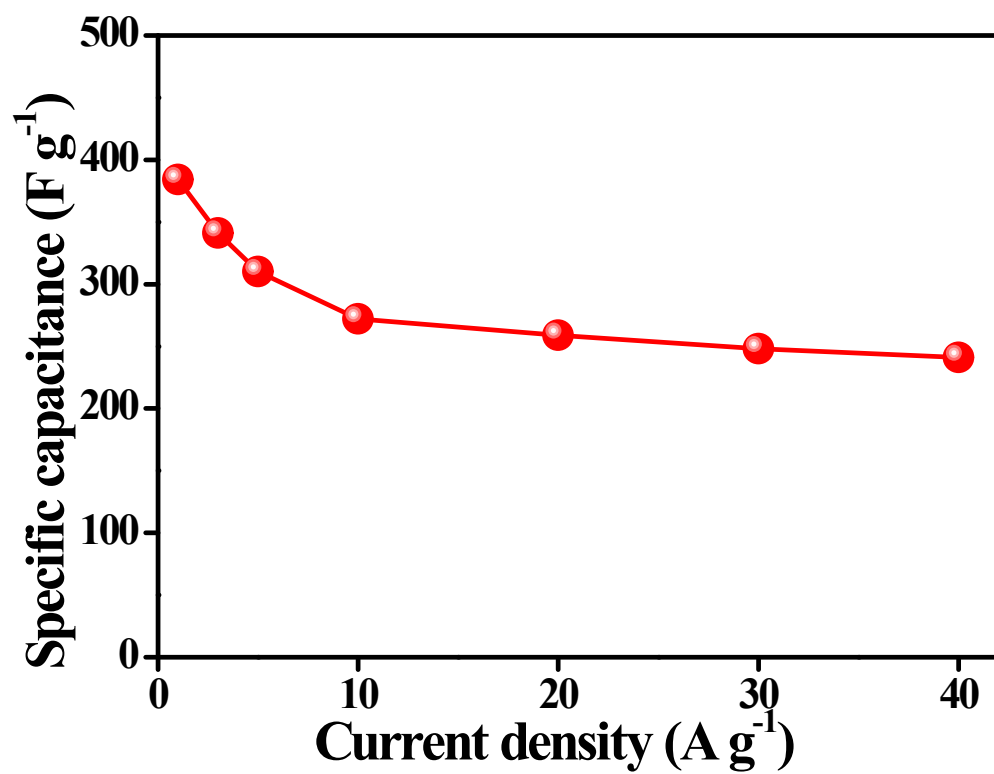


Fig. S21 specific capacitance vs the current density of NG (negative electrode).

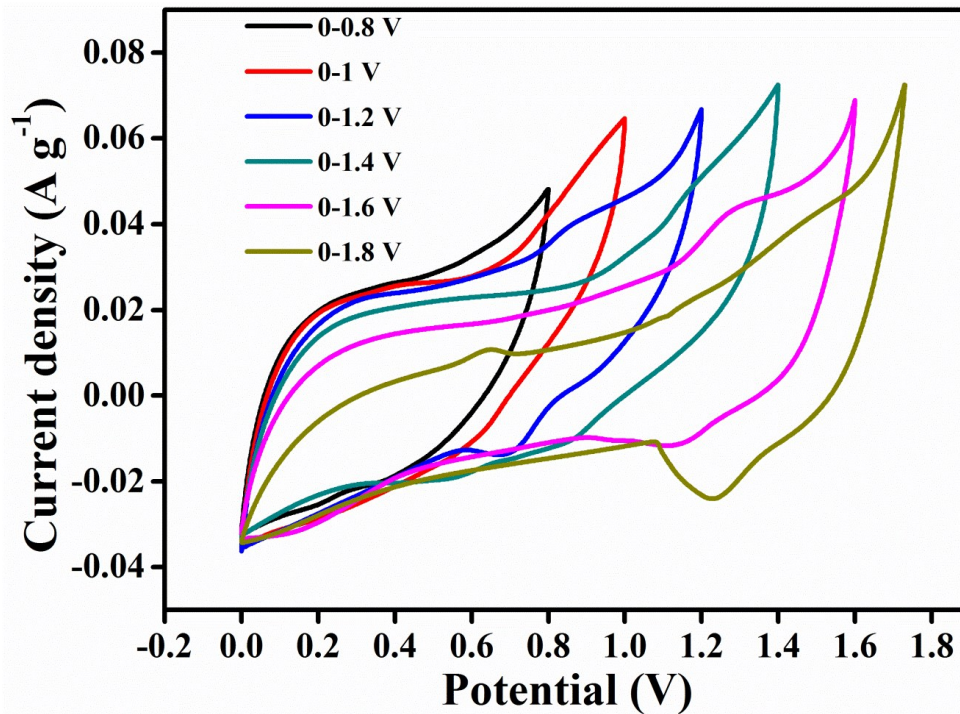


Fig. S22 CV curves (50 mV s^{-1}) of $\text{Co}_2\text{CuS}_4/\text{NG//NG}$ asymmetric supercapacitors at different potential windows.

Table S1. Chemical composition of NG, Co₂CuS₄/NG, CoS/NG and Cu₂S/NG measured from XPS analysis.

Sample	C (at. %)	N (at. %)	O (at. %)	S (at. %)	Co (at. %)	Cu (at. %)
NG	85.22	7.63	7.16	-	-	-
Co ₂ CuS ₄ /NG	69.9	9.39	10.13	5.06	1.27	4.25
CoS/NG	74.36	7.83	13.22	2.17	2.43	-
Cu ₂ S/NG	78.72	6.56	9.09	1.63	-	3.99

Table S2. Comparison of the electrochemical performance of the reported materials for supercapacitors.

Materials	Specific capacitance (F g ⁻¹)	Current density	Electrolyte	Voltage window (V)	Stability	References
NiS	845	1 A g ⁻¹	6 M KOH	0-0.55	-	1
MCs@GNS@NiS	775	0.5 A g ⁻¹	6 M KOH	0-0.4	88.1% (1000 cycles)	2
TNT/cobalt sulfide	400	5 mA cm ⁻²	Na ₂ SO ₃	-1-0	80% (1000 cycles)	3
NiCo ₂ S ₄ -carbon fiber	1154	1 A g ⁻¹	2 M KOH	-0.1-0.5	92.8% (8000 cycles)	4
Bi ₂ S ₃ -rGO	396	1 A g ⁻¹	2 M KOH	-0.2-0.45	75%	5
CNF@NiCo ₂ O ₄	906	2 A g ⁻¹	2 M KOH	0-0.55	96.4% (2400 cycles)	6
NiCo ₂ S ₄ nanotube	933	1 A g ⁻¹	6 M KOH	-0.1-0.5	99% (1000 cycles)	7
Ni _x Co _{1-x} S _{1.097} microsphere	1152	0.5 A g ⁻¹	2 M KOH	0-0.5	-	8
CoS ₂ ellipsoid	1040	0.5 A g ⁻¹	2 M KOH	-0.1-0.4	66% (1000 cycles)	9
PPy/MoS ₂	553.7	1 A g ⁻¹	1 M KCl	-0.5-0.3	90% (500 cycles)	10
Co ₂ CuS ₄ /NG	1005	1 A g ⁻¹	6 M KOH	-0.1-0.5	96.3% (5000 cycles)	This work

MCs@GNS@NiS - graphene nanosheets/nickel sulfide based on make-up cottons; TNT – titania nanotube

Notes and References

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