

Supporting information

Ni₂P-CoP Hybrid Nanosheet Arrays Supported on Carbon Cloth as an Efficient Flexible Cathode for Hydrogen Evolution

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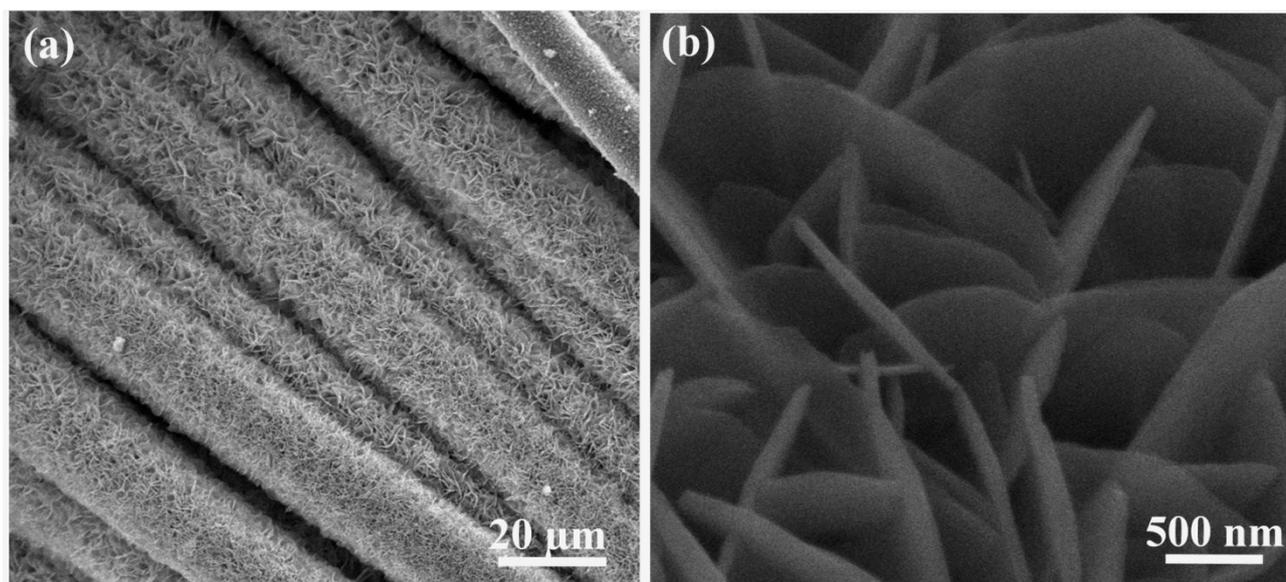


Figure S1. SEM images of the Ni(OH)₂-Co(OH)₂ HNSAs/CC with different magnifications.

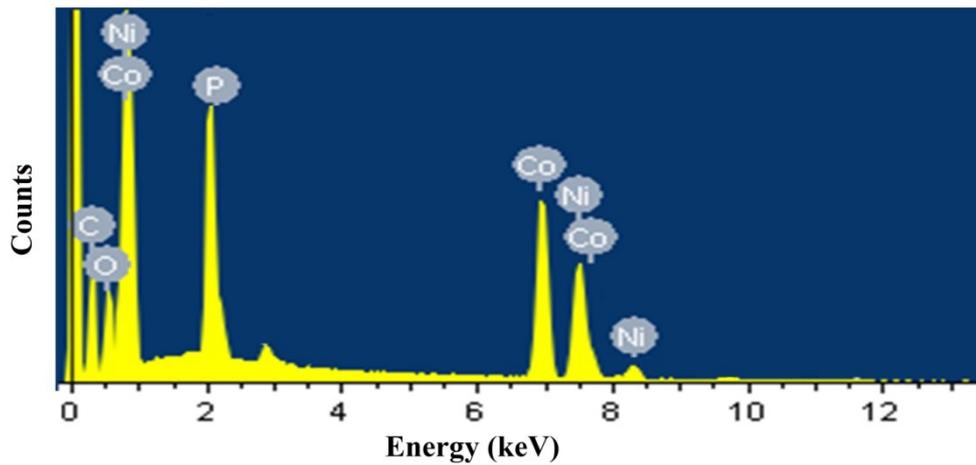


Figure S2. EDS spectrum of the Ni₂P-CoP HNSAs/CC.

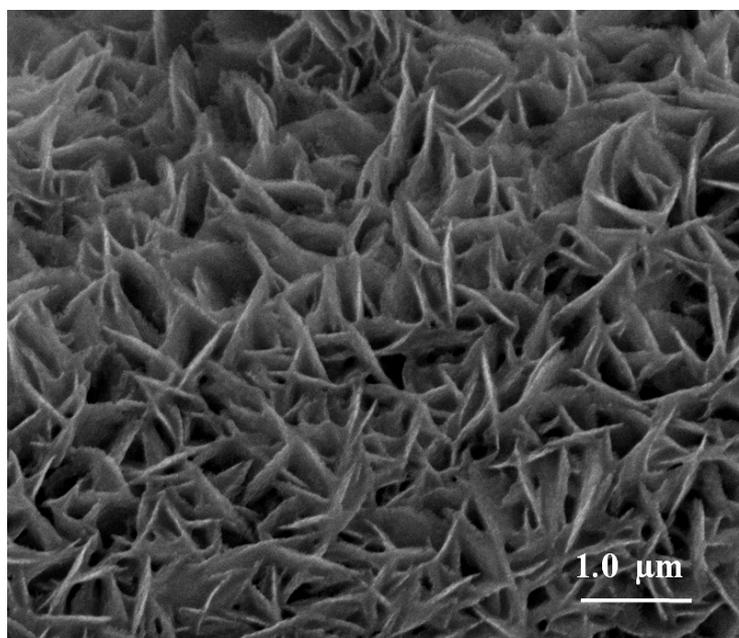


Figure S3. SEM image of Ni₂P NSs/CC with different magnifications.

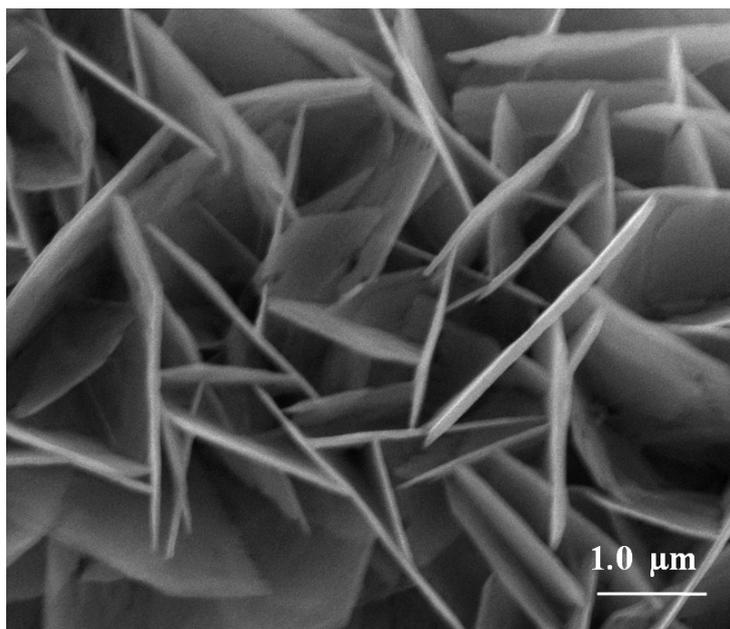


Figure S4. SEM images of CoP NSs/CC with different magnifications.

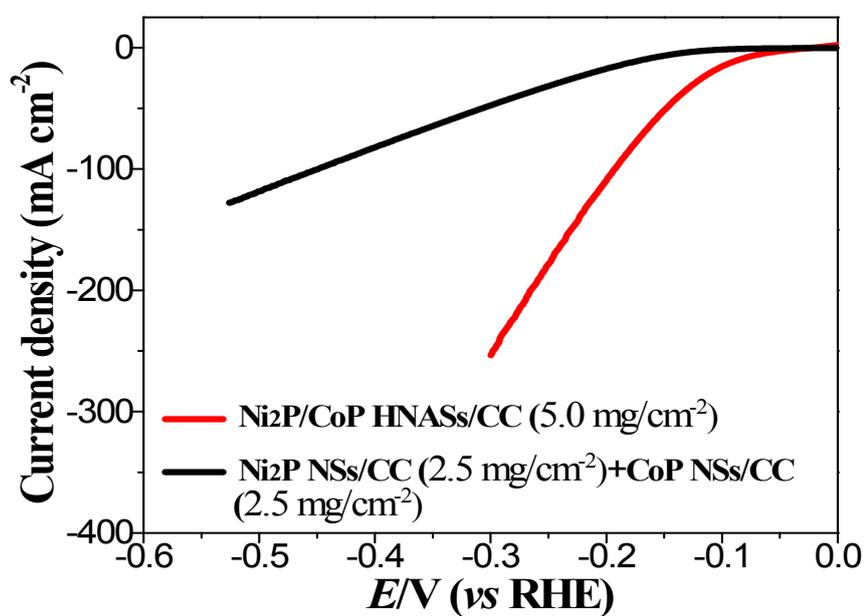


Figure S5. Polarization curves of Ni₂P-CoP HNSAs/CC (the mole ratio of Ni₂P:CoP is 1:2 and loading is 5 mg cm⁻²) and the electroactivity superimposition of Ni₂P NSs/CC (2.5 mg cm⁻²) + CoP NSs/CC (2.5 mg cm⁻²) in solution of 0.5 M H₂SO₄ at a scan rate of 2 mV s⁻¹.

Table S1. Comparisons of HER performance for Ni₂P-CoP HNSAs/CC in acid media with other non-noble-metal HER electrocatalysts. (The references are listed in paper)

Catalyst	Overpotential (η , mV)	Current density (j , mA cm ⁻²)	Reference
Ni₂P/CoP HNS/CC	85	10	This work
	192	100	
	266	200	
MoP nanoparticles	125	10	28
CoP/CNT	122	10	30
Ni ₂ P hollow nanoparticles	116	10	34
CoP/Ti	90	10	42
NiP ₂ NS/CC	75	10	46
CoSe ₂ nanoparticles /CC	137	10	48
MoS ₂ /rGO	150	10	49
Amorphous WP nanoparticles	120	10	50
CoP nanoparticles	95	20	25
Ni ₂ P nanoparticles	140	20	47
Mo ₂ C nanowire	200	60	51

*The above references have been listed in paper.

Table S2. Comparison of HER performance for Ni₂P-CoP HNSAs/CC in alkaline media with other non- noble metal HER electrocatalysts. (The references are listed in paper)

Catalyst	Overpotential (η , mV)	Current density (j , mA cm ⁻²)	Reference
Ni₂P/CoP HNS/CC	73	30	This work
	163	100	
Ni ₂ P NS/CC	102	10	46
Ni-Mo alloy/Ti	80	10	52
Ni nanowire	350	10	52
bulk MoB	225	10	53
CoP/CC	209	10	54

*The above references have been listed in paper.