

Wire-on-flake Heterostructured Ternary $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$: An Efficient Hydrogen Evolution Electrocatalyst

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KEYWORDS: Wire-on-flake Heterostructure, Ternary $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$, Self-supported electrode,
Hydrogen evolution

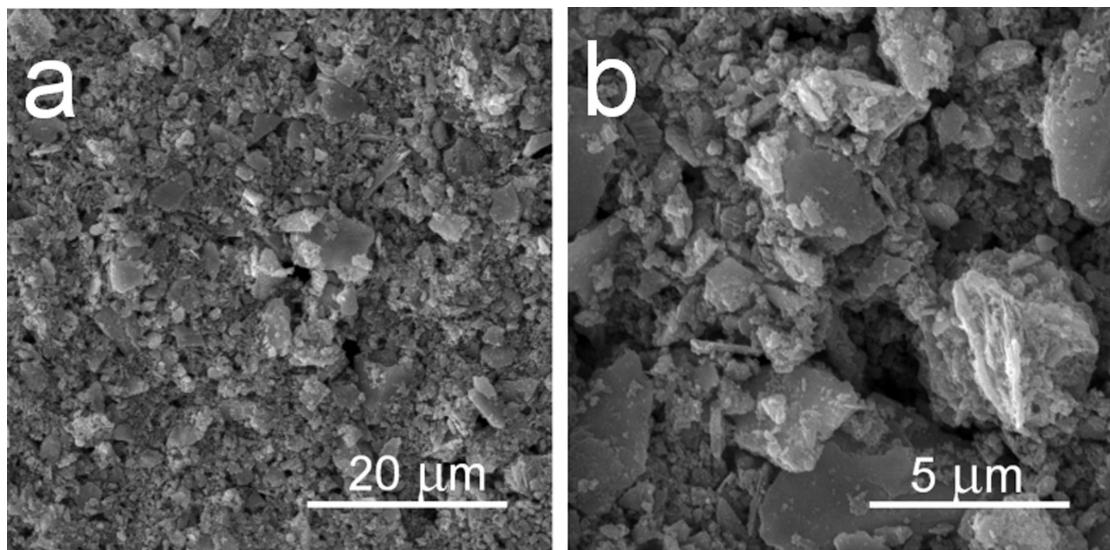


Figure S1. SEM images of $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}$ scratched from CC with different magnifications.

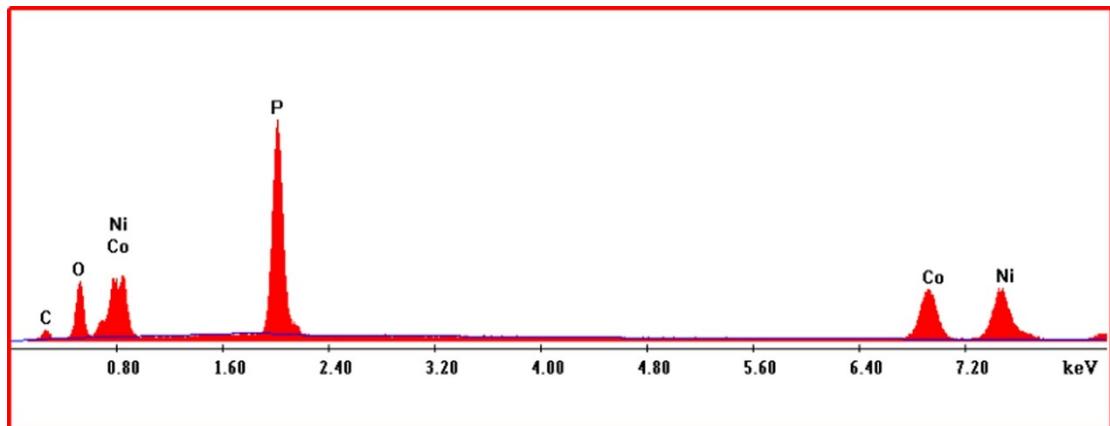


Figure S2. EDX spectrum for $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$.

Table S1. Elements percentage of $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$ obtained from EDX and ICP-MS.

Elements	Weight% (EDX)	Atomic% (EDX)	Atomic% (ICP-MS)
Co	33.13	25.48	25.91
Ni	33.73	26.04	25.35
P	33.14	48.48	48.74
Total	100	100	100
Co:Ni:P		1:1.02:1.90	1.02:1:1.92

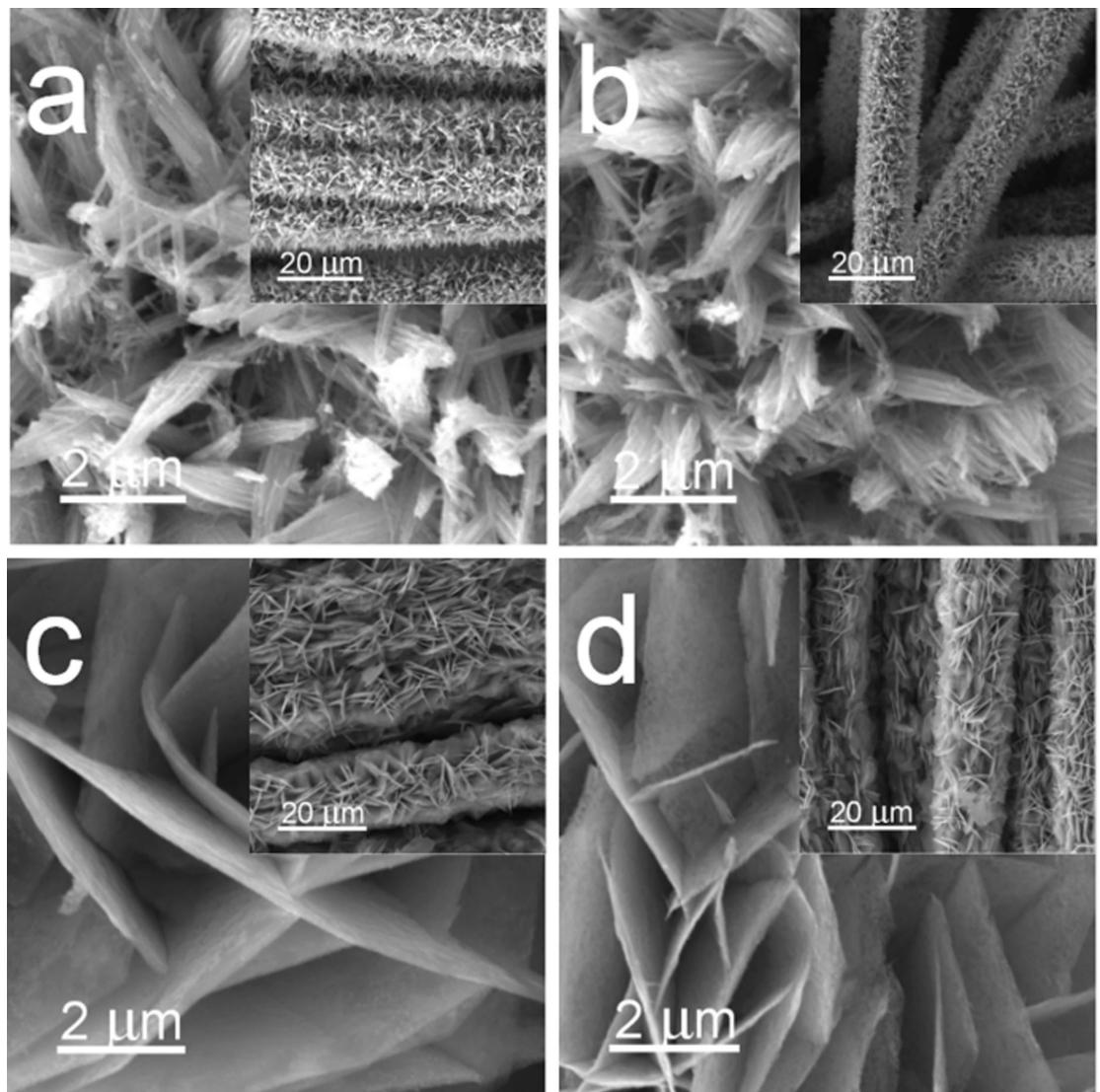


Figure S3. SEM images for (a) $\text{Co}_{0.75}\text{Ni}_{0.25}\text{P}/\text{CC}$, (b) $\text{Co}_{0.66}\text{Ni}_{0.33}\text{P}/\text{CC}$, (c) $\text{Co}_{0.33}\text{Ni}_{0.66}\text{P}/\text{CC}$ and (d) $\text{Co}_{0.25}\text{Ni}_{0.75}\text{P}/\text{CC}$. Inset in (a-d): SEM images of $\text{Co}_{0.75}\text{Ni}_{0.25}\text{P}/\text{CC}$, $\text{Co}_{0.66}\text{Ni}_{0.33}\text{P}/\text{CC}$, $\text{Co}_{0.33}\text{Ni}_{0.66}\text{P}/\text{CC}$ and $\text{Co}_{0.25}\text{Ni}_{0.75}\text{P}/\text{CC}$ with a larger magnification, respectively.

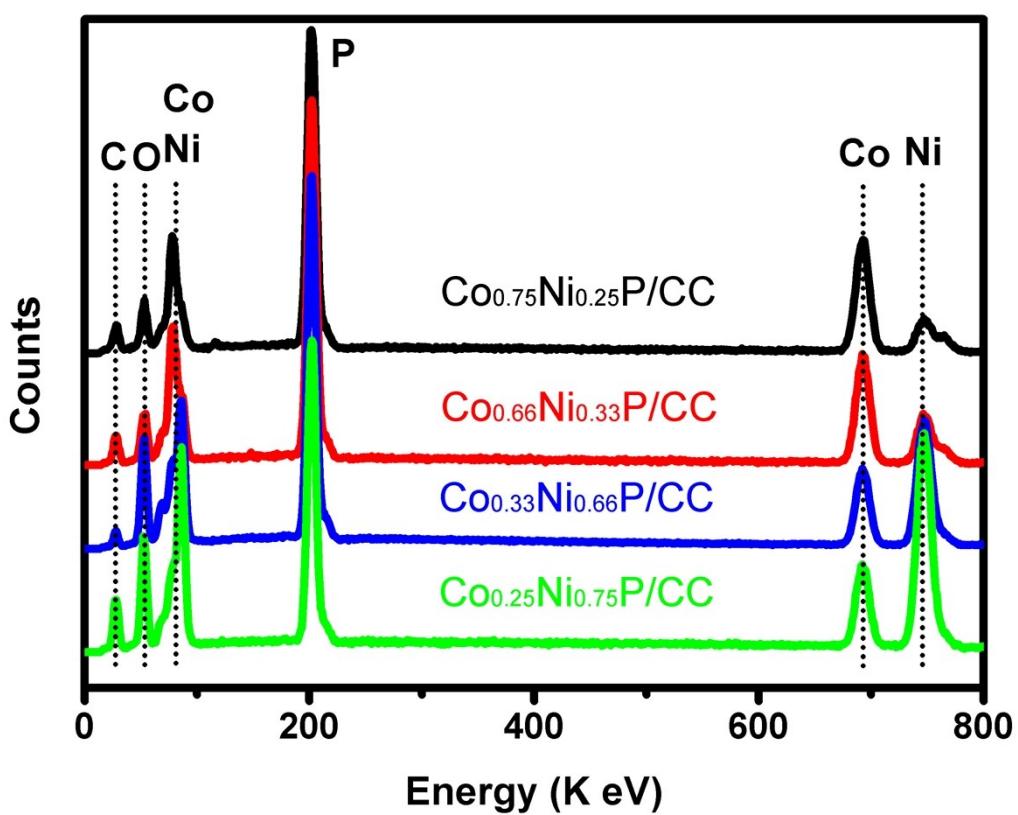


Figure S4. EDX spectra for $\text{Co}_{0.75}\text{Ni}_{0.25}\text{P/CC}$, $\text{Co}_{0.66}\text{Ni}_{0.33}\text{P/CC}$, $\text{Co}_{0.33}\text{Ni}_{0.66}\text{P/CC}$ and $\text{Co}_{0.25}\text{Ni}_{0.75}\text{P/CC}$, respectively.

Table S2. Elements percentage of $\text{Co}_{0.75}\text{Ni}_{0.25}\text{P}/\text{CC}$, $\text{Co}_{0.66}\text{Ni}_{0.33}\text{P}/\text{CC}$, $\text{Co}_{0.33}\text{Ni}_{0.66}\text{P}/\text{CC}$ and $\text{Co}_{0.25}\text{Ni}_{0.75}\text{P}/\text{CC}$ obtained from EDX.

Elements	Atomic% (EDX)			
	$\text{Co}_{0.75}\text{Ni}_{0.25}\text{P/C}$	$\text{Co}_{0.66}\text{Ni}_{0.33}\text{P/C}$	$\text{Co}_{0.33}\text{Ni}_{0.66}\text{P/C}$	$\text{Co}_{0.25}\text{Ni}_{0.75}\text{P/C}$
	C	C	C	C
Co	50.33	50.7	6.27	3.48
Ni	36.86	32.57	21.14	14.71
P	12.81	16.73	72.59	81.81
Total	100	100	100	100
Co:Ni:P	2.88:1:4	1.95:1:3	3.4:11.6:1	4.23:23.5:1

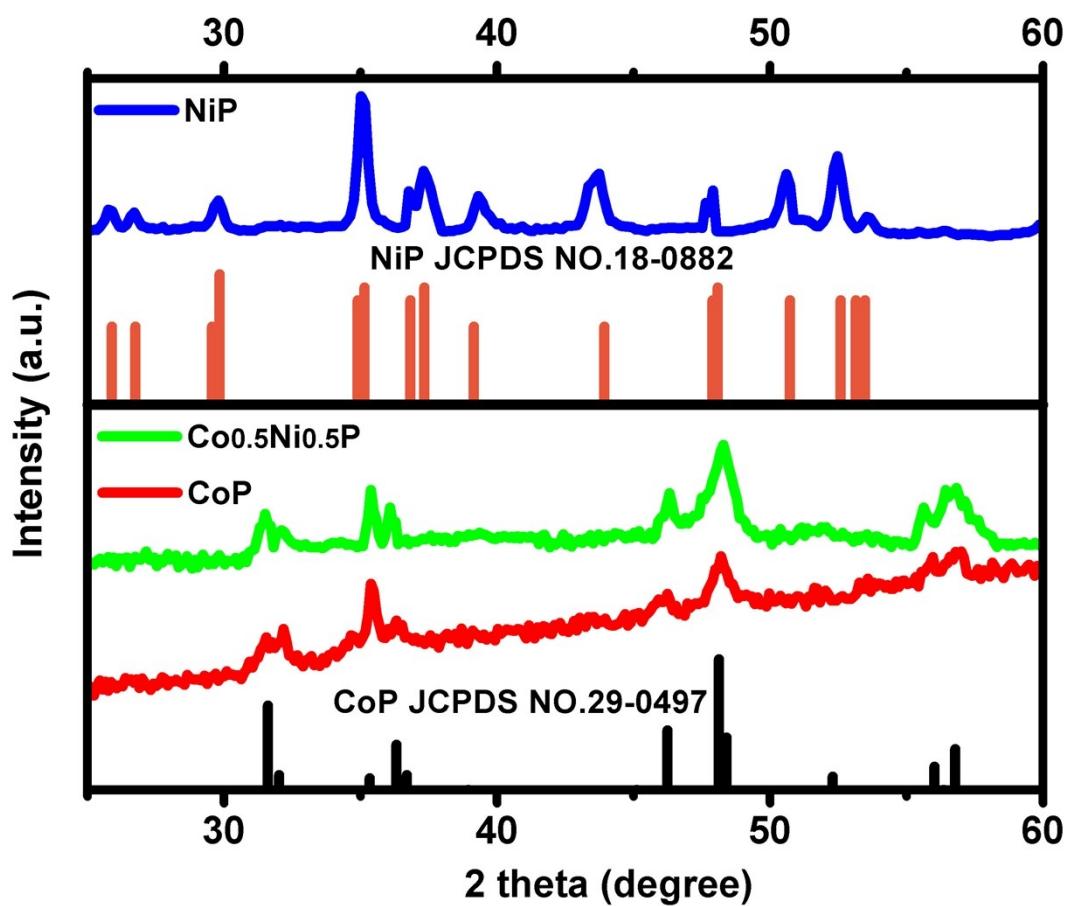


Figure S5. XRD spectra for NiP, CoP and $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}$ scratched from CC.

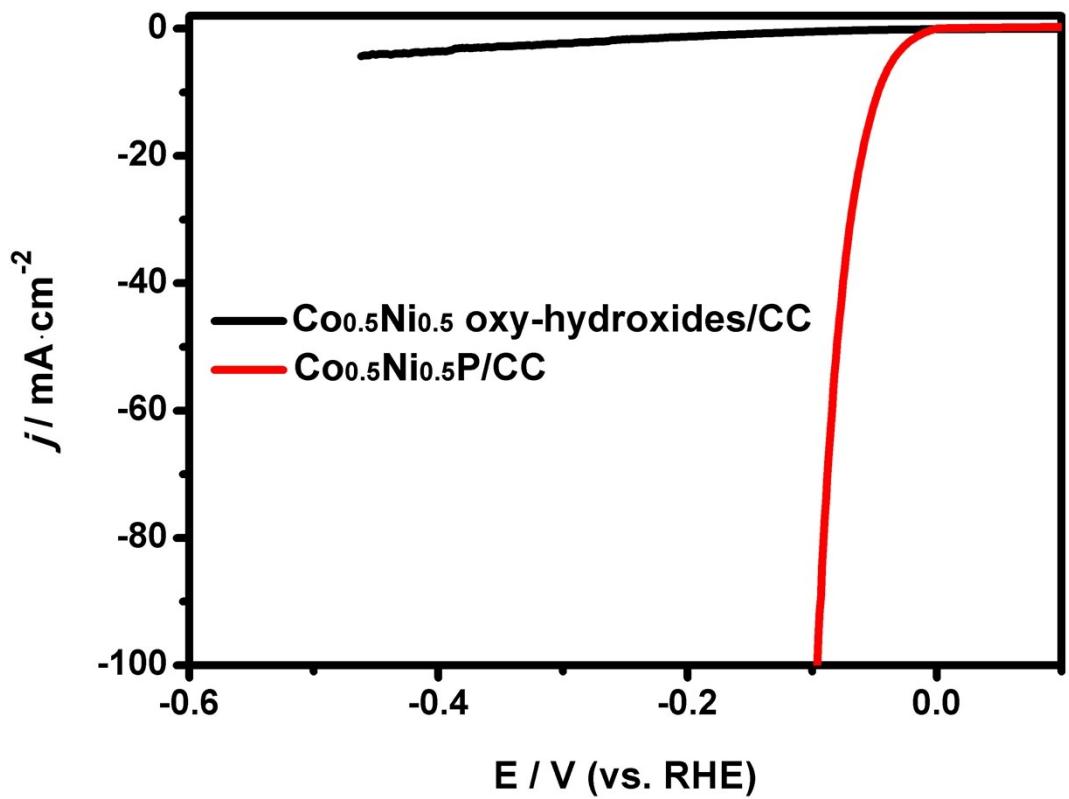


Figure S6. LSV curves for $\text{Co}_{0.5}\text{Ni}_{0.5}$ oxy-hydroxides/CC and $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$.

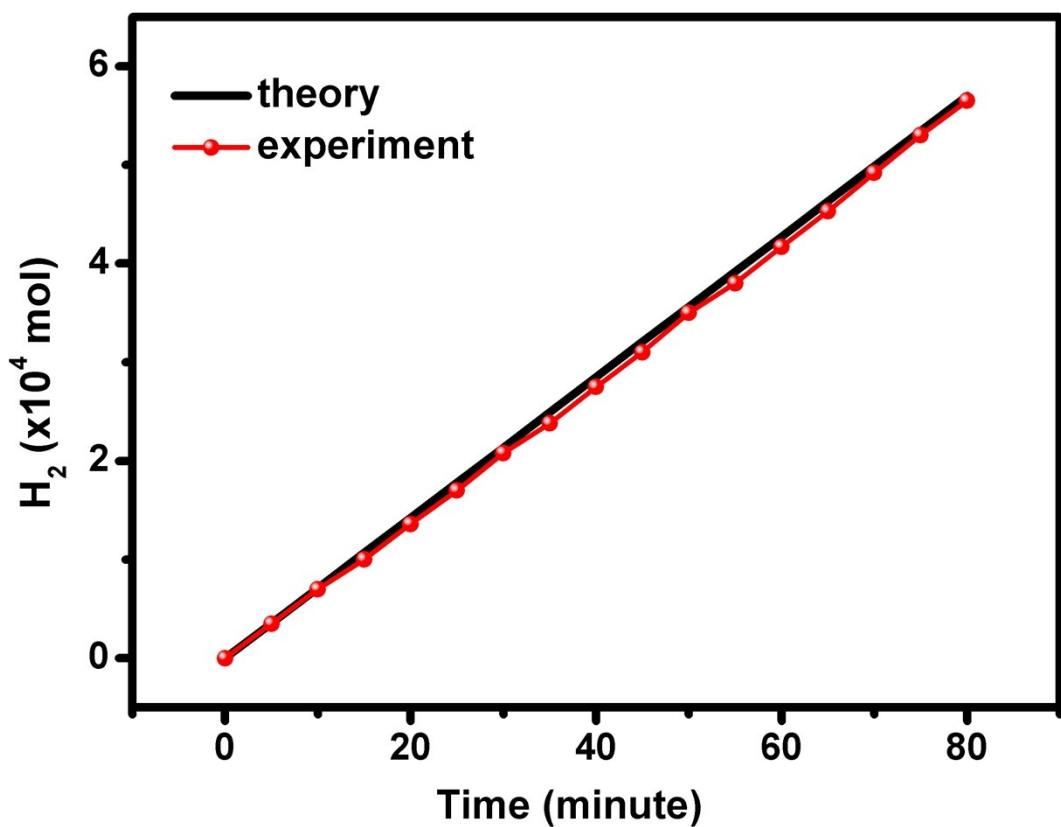


Figure S7. The amount of H_2 calculated by theory and measured from experiment versus time for $Co_{0.5}Ni_{0.5}P/CC$ in 0.5 M H_2SO_4 .

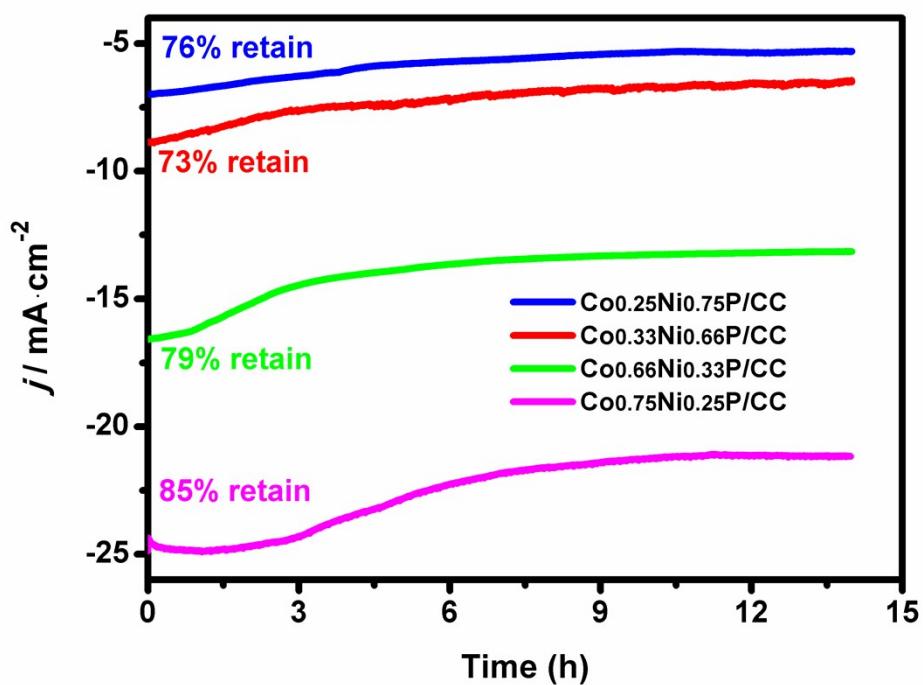


Figure S8. Time-dependent current density curves of Co_{0.25}Ni_{0.75}P/CC, Co_{0.33}Ni_{0.66}P/CC, Co_{0.66}Ni_{0.33}P/CC and Co_{0.75}Ni_{0.25}P/CC under a fixed overpotential of 90 mV for 14 h.

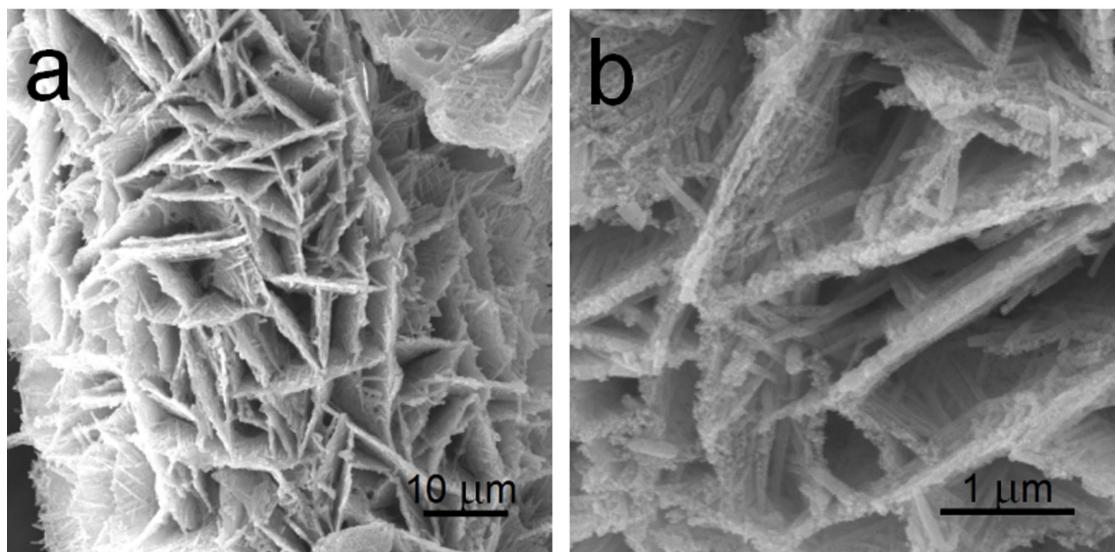


Figure S9. (a) and (b) SEM images of $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$ after time-dependent current density test under different magnifications.

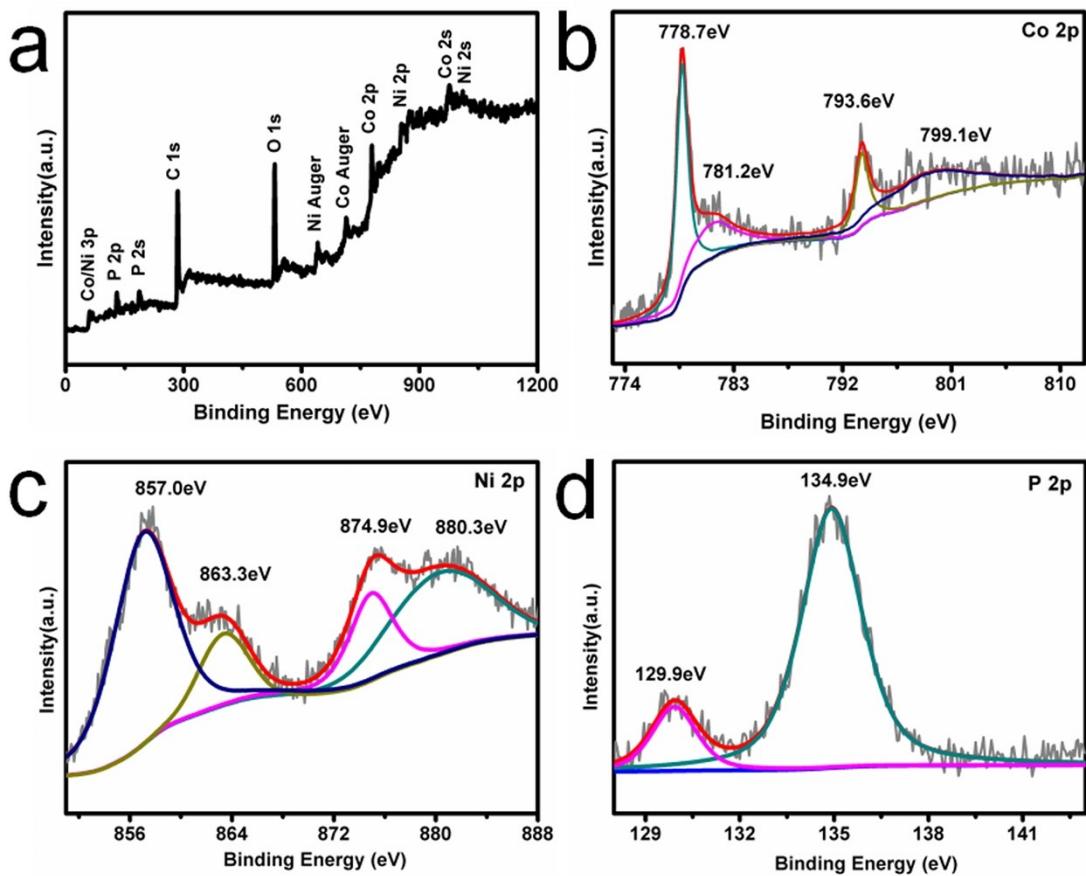


Figure S10. (a) XPS characterization of $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$ and its corresponding (b) Co 2p, (c) Ni 2p and (d) P 2p spectra after time-dependent current density test.

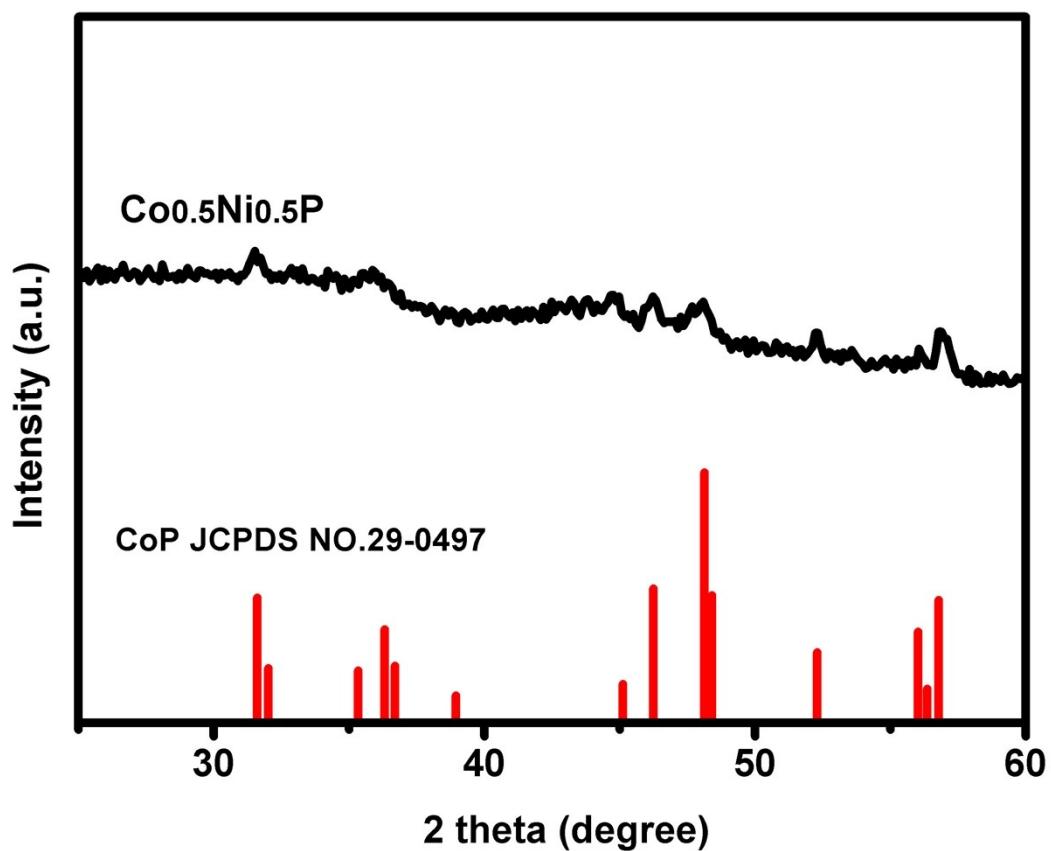


Figure S11. XRD spectrum of $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}$ scratched from CC after time-dependent current density test.

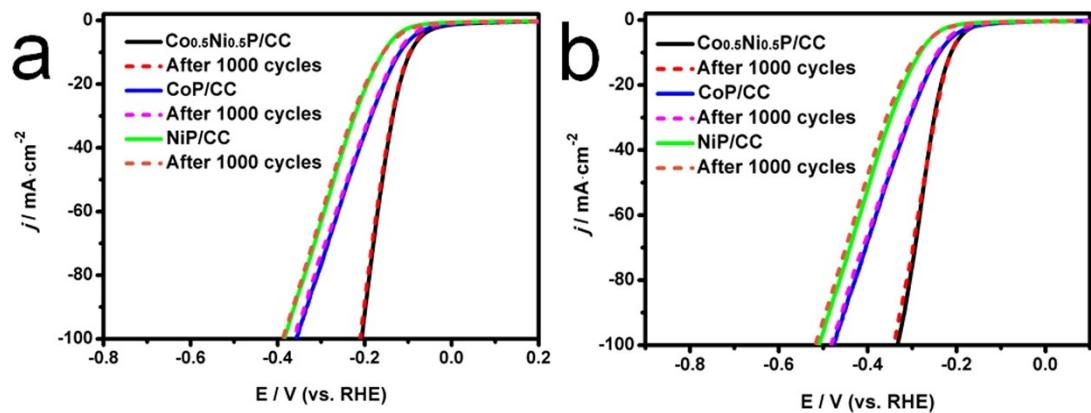


Figure S12. (a) Polarization curves of $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$, CoP/CC and NiP/CC before and after 1000 CV cycles in (a) 0.5 M PBS and (b) 1.0 M KOH.

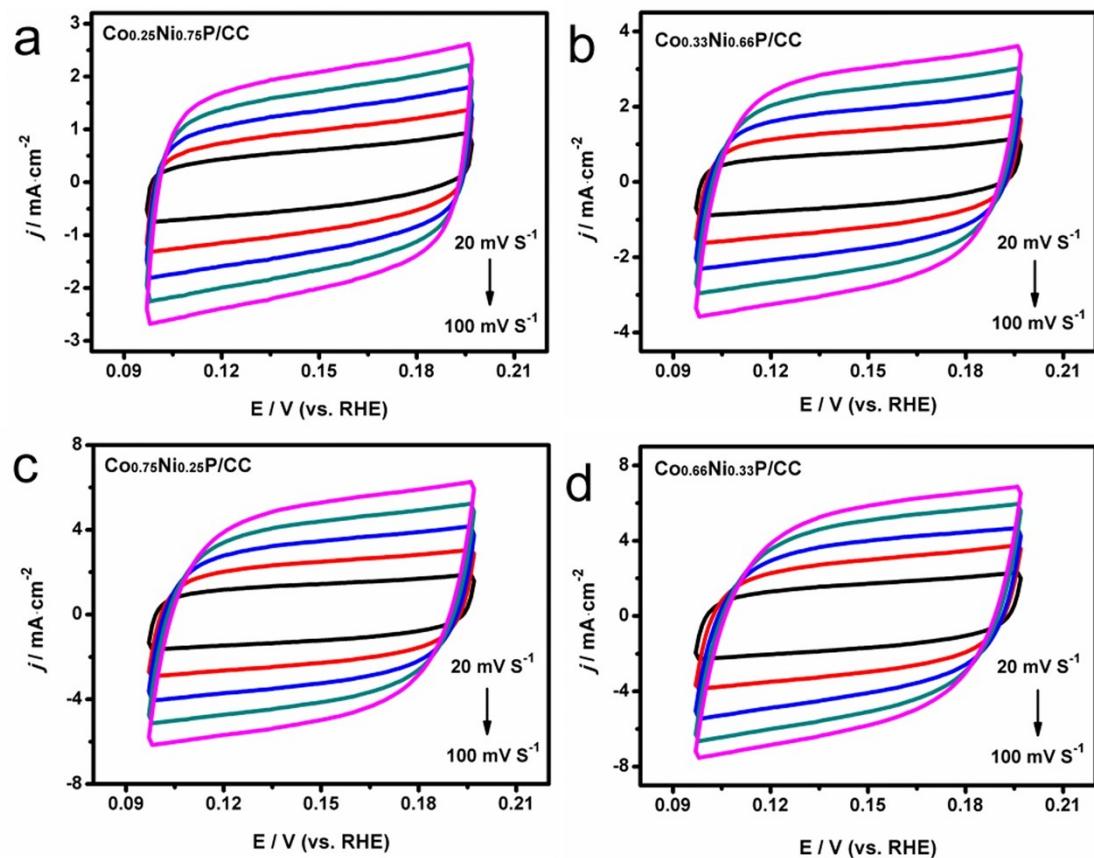


Figure S13. Cyclic voltammetry curves for (a) $\text{Co}_{0.25}\text{Ni}_{0.75}\text{P}/\text{CC}$, (b) $\text{Co}_{0.33}\text{Ni}_{0.66}\text{P}/\text{CC}$, (c) $\text{Co}_{0.75}\text{Ni}_{0.25}\text{P}/\text{CC}$ and (d) $\text{Co}_{0.66}\text{Ni}_{0.33}\text{P}/\text{CC}$ at different scan rates.

Table S3. Comparison of HER activity in acidic media for $\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$ with other existed non-noble-metal electrocatalysts.

Catalyst	Tafel slope (mV dec ⁻¹)	C_{dl} (mF cm ⁻²)	Current density (mA cm ⁻²)	Corresponding Overpotential (mV)	Ref
$\text{Co}_{0.5}\text{Ni}_{0.5}\text{P}/\text{CC}$	34.1	74.7	10 100	47 96	This work
P-1T-MoS ₂	43	63.1	10	153	S1
Co ₂ P@NPG	58	66.8	10 20	103 129	S2
Cu ₇ S ₄ @MoS ₂	48		10	133	S3
Co@BCN	63.7	83	10	96	S4
Se-enriched NiSe ₂	32	10.93	10	117	S5
CoPS	56	99.6	10	48	S6
Mo ₂ C@NPC/NPRGO	30	17.9	10	34	S7
Co-C-N complex	55	400	10 100	138 212	S8
CoSe ₂ nanoparticle/CP	40	14.1	10 100	137 181	S9
α -INS nanosheets	40		10	105	S10

References

1. Y. Yin, J. Han, Y. Zhang, X. Zhang, P. Xu, Q. Yuan, L. Samad, X. Wang, Y. Wang, Z. Zhang, P. Zhang, X. Cao, B. Song and S. Jin, *J. Am. Chem. Soc.*, 2016, **138**, 7965-7972.
2. M. Zhuang, X. Ou, Y. Dou, L. Zhang, Q. Zhang, R. Wu, Y. Ding, M. Shao and Z. Luo, *Nano Lett.*, 2016, **16**, 4691-4698.
3. J. Xu, J. Cui, C. Guo, Z. Zhao, R. Jiang, S. Xu, Z. Zhuang, Y. Huang, L. Wang and Y. Li, *Angew. Chem. Int. Ed.*, 2016, **55**, 6502-6505.
4. H. Zhang, Z. Ma, J. Duan, H. Liu, G. Liu, T. Wang, K. Chang, M. Li, L. Shi, X. Meng, K. Wu and J. Ye, *ACS Nano*, 2016, **10**, 684-694.
5. F. M. Wang, Y. C. Li, Tofik A. Shifa, K. L. Liu, F. Wang, Z. X. Wang, P. Xu, Q. S. Wang and J. He, *Angew. Chem. Int. Ed.*, 2016, **55**, 6919-6924.
6. M. Caban-Acevedo, M. L. Stone, J. R. Schmidt, J. G. Thomas, Q. Ding, H. C. Chang, M. L. Tsai, J. H. He and S. Jin, *Nat. Mater.*, 2015, **14**, 1245-1251.
7. J. S. Li, Y. Wang, C. H. Liu, S. L. Li, Y. G. Wang, L. Z. Dong, Z. H. Dai, Y. F. Li and Y. Q. Lan, *Nat. Commun.*, 2016, **7**, 11204.
8. Z. L. Wang, X. F. Hao, Z. Jiang, X. P. Sun, D. Xu, J. Wang, H. X. Zhong, F. L. Meng and X. B. Zhang, *J. Am. Chem. Soc.*, 2015, **137**, 15070-15073.
9. D. Kong, H. Wang, Z. Lu and Y. Cui, *J. Am. Chem. Soc.*, 2014, **136**, 4897-4900.
10. X. Long, G. Li, Z. Wang, H. Zhu, T. Zhang, S. Xiao, W. Guo and S. Yang, *J. Am. Chem. Soc.*, 2015, **137**, 11900-11903.